Stereocontrolled Total Synthesis of **Pancratistatin**

Sanghee Kim,^{*,†,†} Hyojin Ko,[†] Eunkyung Kim,[†] and Deukjoon Kim[‡]

Natural Products Research Institute, Seoul National University, 28 Yungun, Jongro, Seoul 110-460, Korea, and College of Pharmacy, Seoul National University, San 56-1, Shilim, Kwanak, Seoul 151-742, Korea

pennkim@snu.ac.kr

Received January 30, 2002

ORGANIC LETTERS 2002Vol. 4, No. 8 1343-1345





A new total synthesis of the antitumor alkaloids, pancratistatin (1), has been accomplished from readily available staring materials. The Claisen rearrangement of dihydropyranethylene 5 was employed to construct the A and C rings. Stereo- and regiocontrolled functional group interchange, such as iodolactonization, dihydroxylations, and a cyclic sulfate elimination reaction, allows for the production of the target natural product.

Pancratistatin (1, Figure 1) is a highly oxygenated phenanthridone alkaloid, which was isolated from the roots of





Pancratium littorale by Pettit and co-workers in 1984.¹ This alkaloid exhibits a high level of in vitro and in vivo cancer cell growth inhibitory activity and antiviral activity.² The significant synthetic interest in pancratistatin stems from its

promising pharmacological profile, low natural abundance, and unique structural features, as it contains six contiguous stereogenic centers in the C ring of a phenanthridone skeleton. The first total synthesis of the racemate was reported by Danishefsky in 1989,3 and the first enantioselective synthesis of the natural enantiomer was recorded by Hudlicky in 1995.⁴ In the same year, Trost presented an enantioselective synthesis with a high overall yield.⁵ Since then, Haseltine,⁶ Magnus,⁷ and Rigby⁸ have also presented

Natural Products Research Institute.

[‡] College of Pharmacy.

^{(1) (}a) Pettit, G. R.; Gaddamidi V.; Cragg G. M.; Herald D. L.; Sagawa Y. J. Chem. Soc., Chem. Commun. 1984, 1693. (b) Pettit, G. R.; Gaddamidi V.; Cragg G. M. J. Nat. Prod. 1984, 47, 1018.

^{(2) (}a) Pettit, G. R.; Gaddamidi, V.; Herald, D. L.; Singh, S. B.; Cragg, G. M.; Schmidt, J. M.; Boettner, F. E.; Williams, M.; Sagawa, Y. J. Nat. Prod. 1986, 49, 995. (b) Gabrielsen, B.; Monath, T. P.; Huggins, J. W.; Kevauver, D. F.; Pettit, G. R.; Groszek, G.; Holingshead, M.; Kirsi, J. J.; Shannon, W. M.; Schubert, E. M.; DaRe, J.; Ugarkar, B.; Ussery, M. A.; Phelan, M. J. J. Nat. Prod. 1992, 55, 1569. (c) Gabrielsen, B.; Monath, T. P.; Huggins, J. W.; Kirsi, J. J.; Holingshead, M.; Shannon, W. M.; Pettit. G. R. In Natural Products as Antiviral Agents; Chu, C. K., Cutler, H. G., Eds.; Plenum: New York, 1992; pp 121–135. (3) Danishefsky, S.; Lee, J. Y. J. Am. Chem. Soc. **1989**, 111, 4829.

⁽⁴⁾ Tian, X.; Hudlicky, T.; Königsberger, K. J. Am. Chem. Soc. 1995, 117, 3643.

⁽⁵⁾ Trost, B. M.; Pulley, S. R. J. Am. Chem. Soc. 1995, 117, 10143.

⁽⁶⁾ Doyle, T. J.; Hendrix, M.; VanDerveer, D.; Javanmard, S.; Haseltine, J. Tetrahedron 1997, 53, 11153.

^{(7) (}a) Magnus, P.; Sebhat, I. K. J. Am. Chem. Soc. 1998, 120, 5341. (b) Magnus, P.; Sebhat, I. K. Tetrahedron 1998, 54, 15509.

^{(8) (}a) Rigby, J. H.; Mateo, M. E. J. Am. Chem. Soc. 1997, 119, 12655. (b) Rigby, J. H.; Maharoof, U. S. M.; Mateo, M. E. J. Am. Chem. Soc. 2000, 122, 6624.



a new synthesis of (+)-pancratistatin. Recently, Pettit achieved the synthesis of (+)-1 from the more abundant alkaloid (+)-narciclasine (2, Figure 1).⁹ In this Letter, we wish to report our successful approach to the synthesis of (\pm)-pancratistatin.

The strategy of our synthesis is presented in Scheme 1. The B ring of the phenanthridone skeleton would be constructed at a relatively late stage of the synthesis by employing the Bischler–Napieralski reaction.^{7,10} The requisite cyclization precursor **3**, which contains the six stereocenters in the C ring, could be stereoselectively synthesized from the *cis*-disubstituted cyclohexene **4**. The presence of a γ , δ -unsaturated carbonyl unit in compound **4** suggested the use of a Claisen rearrangement of 3,4-dihydro-2*H*-pyranylethylene **5**.¹¹

The synthesis began by preparing the known bromide 6^{12} from the commercially available methyl gallate via a conventional four-step sequence. Treatment of 6 with excess trimethyl phosphite provided phosphonate 7 in 97% yield (Scheme 2).¹³ Employing the Honer–Wadsworth–Emmons reaction between 7 and commercially available acrolein dimer 8 (1.1 equiv) in the presence of LHMDS in THF afforded the desired (E)-olefin 5 with very high stereoselectivity in 60% yield (92% yield based on the recovered starting material).¹⁴ Only trace amounts (<1%) of the corresponding (Z)-olefin were detected in the crude NMR spectra. The Claisen rearrangement of dihydropyranethylene 5 (250 °C in a sealed tube) provided the cis-disubstituted cyclohexene 4 as a single isomer in 78% yield. As discussed by Büchi,¹¹ this rearrangement must proceed through a boatlike transition state.

(12) (a) Lalami, K.; Dhal, R.; Brown, E. *Heterocycles* 1988, 27, 1131.
(b) For an efficient preparation of the corresponding alcohol from methyl gallate, see: Pettit, G. R.; Singh, S. B. *Can. J. Chem.* 1987, 65, 2390.

(13) Jawad, A.; Jacques, C.; Ingrid, H.-K.; Christian H.; Hugues, M.;
 Robert, G. R. J. Med. Chem. 2000, 43, 560.

(14) Wittig reaction between the corresponding phosphonium bromide and acrolein dimer 8 in the presence of KOH and 18-crown-6 ether in CH₂-Cl₂ provided a mixture of olefins with a Z/E ratio of 5:1 in 85% yield.

(15) Kobayashi, S.; Kamiyama, K. Ohno, M. J. Org. Chem. 1990, 55, 1169.



^{*a*} (a) P(OMe)₃, toluene, sealed tube, 180 °C, 2 h, 97%; (b) **8**, LHMDS, THF, 0 °C to rt, 22 h, 60% (92% based on the recovered starting material); (c) toluene, sealed tube, 250 °C, 20 h, 78%; (d) NaClO₂, NaH₂PO₄·2H₂O, 2-methyl-2-butene, THF, *t*BuOH, H₂O, rt, 18 h, 90%; (e) (i) KI₃, aqueous NaHCO₃, CH₂Cl₂, rt, 20 h, (ii) DBU, benzene, reflux, 8 h, 78%; (f) NaOMe, MeOH, reflux, 20 h, 93%; (g) 1N LiOH, THF, rt, 18 h, 99%; (h) (i) DPPA, Et₃N, toluene, reflux, 15 h, (ii) NaOMe, MeOH, reflux, 0.5 h, 82%.

With the appropriately functionalized cyclohexene 4 in hand, our study focused on the selective introduction of the stereocenters in the C ring. First, the aldehyde group of 4 was oxidized with NaClO₂ to the corresponding carboxylic acid 9 in 90% yield. Iodolactonization of 9 under two-phase conditions followed by treatment of the resulting iodolactone with DBU in refluxing benzene led to the formation of the bicyclic lactone 10 with an overall yield of 78%.¹⁵ Methanolysis of the lactone 10 with NaOMe at room temperature for 18 h afforded an inseparable equilibrium mixture (ca. 1:1 ratio) of hydroxy ester 11 and its C-4a epimer (pancratistatin numbering). However, when the methanolysis was carried out in refluxing methanol for 20 h, epimerization of the methoxycarbonyl group was accomplished very cleanly to give 11 as the only identifiable product in 93% yield. Saponification of the methyl ester 11 with LiOH was followed by a modified Curtius rearrangement¹⁶ of the resulting acid 12 with diphenylphosphoryl azide in refluxing toluene to give a rather stable isocyanate intermediate that required further treatment with NaOMe/MeOH to generate the corresponding carbamate 13 in 82% overall yield.

The final C-ring functional group processing of **13** proceeded as follows (Scheme 3). At this stage, it was necessary to protect the free hydroxyl group of **13** to

⁽⁹⁾ Pettit, G. R.; Melody, N.; Herald, D. J. Org. Chem. 2001, 66, 2583. (10) (a) Banwell, M. G.; Cowden, C. J.; Gable, R. W. J. Chem. Soc., Perkin Trans. 1 1994, 3515. (b) Fodor, G.; Nagibandi, S. Tetrahedron 1980, 36, 1279.

⁽¹¹⁾ Büchi, G.; Powell, J. E., Jr. J. Am. Chem. Soc. 1970, 92, 3126.

^{(16) (}a) Shin, K. J.; Moon, H. R.; George, C.; Marquez, V. E.J. Org. Chem. **2000**, 65, 2172. (b) Evans, D. A.; Wu, L. D.; Wiener, J. J. M.; Johnson, J. S.; Ripin, D. H. B.; Tedrow, J. S. J. Org. Chem. **1999**, 64, 6411.



^{*a*} (a) BzCl, Et₃N, DMAP, CH₂Cl₂, rt, 15 h, 99%; (b) OsO₄, NMO, THF/H₂O, rt, 20 h, 96%; (c) (i) SOCl₂, Et₃N, CH₂Cl₂, 0 °C, 0.5 h, (ii) Oxone, RuCl₃·3H₂O, EtOAc/CH₃CN/H₂O, rt, 2 h, 83%; (d) DBU, toluene, reflux, 2 h, then H₂SO₄, H₂O/THF, rt, 4 h, 67%; (e) OsO₄, NMO, THF/H₂O, rt, 27 h, 88%; (f) (i) Ac₂O, DMAP, pyridine, CH₂Cl₂, rt, 1 h, 77%, (ii) Tf₂O, DMAP, CH₂Cl₂, 0 to 5 °C, 22 h, 78%; (g) BBr₃, CH₂Cl₂, -78 to 0 °C, 1 h, 65%; (h) NaOMe, MeOH, THF, rt, 4 h, 83%.

selectively install the C-2 hydroxyl group on the α -face. This was accomplished by reacting compound **13** with benzoyl chloride to furnish **14** in 99% yield. Attempted epoxidation of **14** to **15** with various regents such as mCPBA and dioxiranes was not successful, providing only decomposed materials. However, dihydroxylation of the $\Delta^{2,3}$ -olefin with OsO₄ did occur on the α -face of the molecule to produce diol **16** in 96% yield. The stereochemistry of **16** was tentatively assigned as α , on the basis of steric considerations.

The regioselective elimination of the C-3 hydroxyl group to generate the requisite $\Delta^{3,4}$ unsaturation was achieved by employing the cyclic sulfate elimination reaction.¹⁷ Treatment of diol **16** with thionyl chloride followed by oxidation with RuCl₃·3H₂O/Oxone¹⁸ provided the corresponding cyclic sulfate **17** in 83% yield. The reaction of cyclic sulfate **17** with DBU in refluxing toluene^{17a} led, after acidic workup, to the formation of the desired allylic alcohol **18** (67% yield). Routine *cis*-dihydroxylation of **18** with OsO₄ afforded the single isomer **19** in 88% yield, thereby completing the functionalization of the C ring of pancratistatin. The structural assignment made for this compound was strongly supported by its relevant ¹H NMR coupling patterns and by comparing the ¹H NMR spectral data of the derived tetraacetate with those reported by Magnus.⁷

The remaining steps to pancratistatin required protection of the hydroxyl groups, formation of the final lactam B ring, and protecting group removal and were accomplished by employing reaction conditions analogous to those of Magnus et al.⁷ Peracetylation of **19** (77%) was followed by a Banwell's modified Bischler–Napieralski cyclization,^{7,10} which provided predominantly the desired product **20**, along with a minor amount of the regioisomer **21** in 78% combined yield and 7:1 regioselectivity. Treatment of an inseparable mixture of **20** and **21** with BBr₃ to remove the C-7 methyl group protection yielded **22** (65%) and unreacted **21**, which were now separable.¹⁹ Finally, simple removal of protecting groups with NaOMe/MeOH afforded (\pm)-**1** in 83% yield, of which ¹H and ¹³C NMR spectral data were in good agreement with those reported.^{1,3–10}

In conclusion, we have accomplished the stereoselective synthesis of (\pm) -pancratistatin from readily available starting materials. We utilized the Claisen rearrangement of dihydropyranethylene **5** to construct the A and C rings, and subsequent iodolactonization, dihydroxylations, and cyclic sulfate elimination reactions to install six contiguous stereogenic centers in the C ring.

Acknowledgment. This work was supported by a grant (1999-2-21500-001-3) from the Basic Research Program of the Korea Science & Engineering Foundation.

Supporting Information Available: Full experimental procedures and spectral data of new compounds and ¹H NMR and ¹³C NMR spectra of compounds **1**, **4**, **5**, **12**, **14**, **19**, and **22**. This material is available free of charge via the Internet at http://pubs.acs.org.

OL0256419

⁽¹⁷⁾ For an elimination reaction of α-nonactivated cyclic sulfate, see: (a) Winkler, J. D.; Kim, S.; Harrison, S.; Lewin, N. E.; Blumberg, P. M. J. Am. Chem. Soc. **1999**, *121*, 296. (b) Kim, C. U.; Lew, W.; Williams, M. A.; Wu, H.; Zhang, L.; Chen, X.; Escarpe, P. A.; Mendel, D. B.; Laver, W. G.; Stevens, R. C. J. Med. Chem. **1998**, *41*, 2451. (c) Schaub, C.; Müller, B.; Schmidt, R. R. Eur. J. Org. Chem. **2000**, 1745.

⁽¹⁸⁾ Robins, M. J.; Lewandowska, E.; Wnuk, S. F. J. Org. Chem. 1998, 63, 7375.

⁽¹⁹⁾ This reaction patterned after a similar step in the Magnus synthesis.⁷