

Dearomatization Strategy and Palladium-Catalyzed Domino Reaction: Construction of Azepino[5,4,3-*cd*]indoles from 2-Alkynylanilines

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Supporting Information

ABSTRACT: A facile approach to construct 3,4-fused tricyclic azepino[5,4,3-*cd*]indoles from 2-alkynyl anilines, isocyanides, and α,β -unsaturated acids is reported. This synthetic process involves a regioselective meta-functionalization of 2-alkynylanilines using a dearomatization strategy and a palladium(II)-catalyzed domino heterocyclization/Heck reaction.



3,4-Fused tricyclic indoles constitute a very important class of compounds because of their remarkable medicinal and biological activities.¹ For example, the natural products lysergic acid diethylamide (LSD),² hapalindole U,³ arcyroxocin B,⁴ communesin B,⁵ decursivine,⁶ welwistatin,⁷ dehydrobufotenine,⁸ and indolactam V^9 have a 3,4-fused tricyclic indole unit as their core structure. Therefore, a number of synthetic approaches toward the construction of 3,4-fused tricyclic indole systems have been developed such as the intramolecular Fischer indole syntheses,¹⁰ Pictet–Spengler reactions,¹¹ electrocyclizations,¹² photocyclizations,¹³ and Diels-Alder reactions.¹⁴ Normally, these strategies involve an introduction of functional groups to the C-3 and the C-4 positions of indoles and a subsequent cyclization. A big challenge in these methods is the selective functionalization of the C-4 position of indoles, which is not a preferred site for the electrophilic substitution reaction. Recently, Jia and co-workers reported an intramolecular Larock indolization reaction, and 2,3-difunctionalized anilines proved to be potential precursors to a variety of 3,4-fused tricyclic indoles.¹⁵ Herein, we report an approach to construct 3,4-fused tricyclic azepino[5,4,3-cd]indoles from 2-alkynylanilines, isocyanides, and α_{β} -unsaturated acids. This process involves a regioselective meta-functionalization of 2-alkynylanilines using a dearomatization strategy and a palladium-catalyzed domino heterocyclization/Heck reaction (Scheme 1).

Under oxidizing conditions, the dearomatization¹⁶ of parasubstituted 2-alkynylanilines forms 2-alkynylcyclohexadienimines. The reactivity of these products has been shown to allow the regioselective 1,4-addition (Michael-type) of a variety of carbon and heteronucleophiles at the C-3 position.^{17,18} We conceived that this dearomatization strategy could be used to achieve the meta-functionalization of 2-alkynylanilines by isocycanides and α,β -unsaturated acids (Scheme 2).





First, we evaluated the reactions of *N*-protected 4-methyl-2-(2-phenylethynyl)benzenamines 1-5 with *tert*-butyl isocyanide and cinnamic acid in methanol with (diacetoxyiodo)benzene as oxidant (Scheme 3). The results revealed that the protecting group played a significant role in the transformation. When the

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Scheme 2. Meta-Functionalization of 2-Alkynylanilines Using Dearomatization Strategy







N-COCF₃-, N-Boc-, or N-Bz-protected substrate was used, the oxidative dearomatization proceeded well, but further conversion did not occur. When the N-Ms- or N-Ts-protected substrate was used, the desired 3-functionalized product 9 or 10 was formed in 31% or 35% yield, respectively.

To avoid the competition between cinnamic acid and acetic acid metabolized from (diacetoxyiodo)benzene, iodosylbenzene (PhIO) was used as oxidant instead.¹⁹ Further optimization showed that the dearomatization had to be conducted in methanol, but acetonitrile was the best reaction medium for the Michael addition and the aromatization steps. Therefore, the meta-functionalization reaction was carried out in a stepwise way: compound 5 was treated with 1.1 equiv of iodosylbenzene in methanol (0.1 M) at room temperature for 10 min. After methanol was removed under a pressure-reducing condition, acetonitrile, 1.5 equiv of tert-butyl isocyanide, and 1.2 equiv of cinnamic acid were added, and the resulting mixture was stirred at 50 °C for 48 h. This reaction gave rise to compound 10 in 84% yield. Under the optimized conditions, the reactions of 2alkynylanilines bearing a range of different substitutions with a variety of isocyanides and $\alpha_{,\beta}$ -unsaturated acids proceeded smoothly leading to the corresponding products in moderate to good yields (Table 1).

We initially investigated the possibility of constructing azepino[5,4,3-*cd*]indole via an iodocyclization²⁰ and an intramolecular Heck reaction.²¹ The desired product **31** was formed, but in a very low yield (Scheme 4). Therefore, we focused our attention on the palladium(II)-catalyzed domino heterocyclization/Heck reaction.²² We were pleased to observe the formation of compound **31** in the palladium diacetate catalyzed reaction, various phosphine ligands, palladium salts, bases, and oxidants were examined (Table 2). When a catalytic amount of palladium diacetate was used together with tris(3-methoxyphenyl)phosphine and a stoichiometric amount

Table 1. Meta-Functionalization of 2-Alkynylanilines

R ¹		R ² 1) (1.1 equi 2) (1.5 equi	v) PhIO, MeOH, v)R ³ NC, CH ₃ CN	rt, 10 min O	R^4			
	Ťs	(1.2 equi	исоон	Ļ	MH NH			
		(1.2 640)	R ⁴		PG			
entry	\mathbb{R}^1	R ²	R ³	\mathbb{R}^4	product ^a (%)			
1	Me	Ph	<i>t</i> -Bu	Ph	10 (84)			
2	Et	Ph	t-Bu	Ph	11 (65)			
3	<i>n</i> -Bu	Ph	t-Bu	Ph	12 (67)			
4	Me	4-MeC ₆ H ₄	t-Bu	Ph	13 (66)			
5	Me	4-MeOC ₆ H ₄	t-Bu	Ph	14 (56)			
6	Me	4-ClC ₆ H ₄	t-Bu	Ph	15 (63)			
7	Me	<i>t</i> -Bu	t-Bu	Ph	16 (62)			
8	Me	TMS	t-Bu	Ph	17 (74)			
9	Me	Н	t-Bu	Ph	18 (63)			
10	Me	Ph	cyclo-Hex	Ph	19 (63)			
11	Me	Ph	t-Bu	4-MeC ₆ H ₄	20 (82)			
12	Me	Ph	t-Bu	4-MeOC ₆ H ₄	21 (66)			
13	Me	Ph	t-Bu	4-ClC ₆ H ₄	22 (82)			
14	Me	Ph	t-Bu	$4-FC_6H_4$	23 (81)			
15	Me	Ph	t-Bu	2-furyl	24 (78)			
16	Me	Ph	t-Bu	Me	25 (83)			
17	Me	Ph	t-Bu	Н	26 (61)			
18	Me	TMS	t-Bu	Н	27 (63)			
19	Me	4-ClC ₆ H ₄	t-Bu	Н	28 (50)			
20	Et	Ph	t-Bu	Н	29 (46)			
21	<i>n</i> -Bu	Ph	t-Bu	Н	30 (48)			
^a Isolated yield based on 2-alkynylanilines.								



of potassium phosphate, the isolated yield of compound 31 increased to 73%. Molecular oxygen proved to be the best oxidant. The formation of compound 31 was not observed when copper(II) bromide, silver acetate, or *meta*-chloroperbenzoic acid was used.

Azepino [5,4,3-*cd*] indoles with a range of different substitutions were obtained in moderate yields (Scheme 5). The configuration of the double bond of compound **31** was confirmed by its NOESY spectrum. When the double bond in substrates was a terminal alkene, the corresponding reaction was conducted in the absence of phosphine ligand and base, and 2 equiv of acetic acid was added to promote the reaction. The structure of compound **36** was confirmed by its singlecrystal diffraction analysis.²³ The treatment of compound **31** with 5 equiv of Bu₄NF in THF at reflux led to the selective deprotection of the Ts group (eq 1).²⁴



Table 2. Evaluation of Conditions for Palladium(II)-Catalyzed Domino Heterocyclization/Heck Coupling Reaction

	0 Ph			t-Bu, _O	
	O N. Bu-t Ph	(0.1 equiv) PdX ₂ , (0.2 equ	iv) ligand	0₹ ^N ~~	Ph _/
Me		(2 equiv) LiCl, (2 equiv) base, oxidant THF, 70 °C		Me Ph	
	Щ. NH				
	Ťs			Ts	
	10			31	
entry	Pd(II)	R ₃ P	base	oxidant	31 (%)
1	$Pd(OAc)_2$	Ph ₃ P	K_2CO_3	O ₂	15
2	$Pd(OAc)_2$	$(2-MeC_6H_4)_3P$	K_2CO_3	O ₂	31
3	$Pd(OAc)_2$	$(3-MeC_6H_4)_3P$	K_2CO_3	O ₂	32
4	$Pd(OAc)_2$	(2- MeOC ₆ H ₄) ₃ P	K ₂ CO ₃	O ₂	<5
5	$Pd(OAc)_2$	(3- MeOC ₆ H ₄) ₃ P	K ₂ CO ₃	O ₂	48
6	$Pd(OAc)_2$	(4- MeOC ₆ H ₄) ₃ P	K ₂ CO ₃	O ₂	18
7	$Pd(OAc)_2$	$(C_6F_5)_3P$	K ₂ CO ₃	O ₂	23
8	$Pd(OAc)_2$	(cyclo-C ₆ H ₁₁) ₃ P	K ₂ CO ₃	O ₂	20
9	$Pd(OAc)_2$	(2-furyl) ₃ P	K ₂ CO ₃	O ₂	15
10	$Pd(OAc)_2$	$(n-Bu)_{3}P$	K ₂ CO ₃	O ₂	<5
11	$Pd(OAc)_2$	$(t-Bu)_3P$	K ₂ CO ₃	O ₂	<5
12	PdCl ₂	(3- MeOC ₆ H ₄) ₃ P	K_2CO_3	O ₂	<5
13	PdBr ₂	(3- MeOC ₆ H ₄) ₂ P	K ₂ CO ₃	O ₂	16
14	$Pd(OCOCF_3)$	$_{2}$ (3- MeOC ₆ H ₄) ₂ P	K_2CO_3	O ₂	43
15	$Pd(OAc)_2$	(3- MeOC₄H₄)₂P	Li_2CO_3	O ₂	<5
16	$Pd(OAc)_2$	(3- MeOC₄H₄)₂P	Cs_2CO_3	O ₂	67
17	$Pd(OAc)_2$	(3- MeOC₄H₄)₂P	NaOMe	O ₂	30
18	$Pd(OAc)_2$	(3- MeOC ₆ H ₄) ₂ P	KOBu-t	O ₂	<5
19	$Pd(OAc)_2$	(3- MeOC₄H₄)₂P	KOAc	O ₂	<5
20	$Pd(OAc)_2$	(3- MeOC₄H₄)₂P	K ₃ PO ₄	O ₂	73
21	$Pd(OAc)_2$	(3- MeOC ₆ H ₄) ₂ P	K ₃ PO ₄	BQ	25
22	$Pd(OAc)_2$	(3- MeOC ₆ H ₄) ₃ P	K ₃ PO ₄	m-CPBA	0
23	$Pd(OAc)_2$	(3- MeOC ₆ H ₄) ₃ P	K ₃ PO ₄	CuBr ₂	0
24	$Pd(OAc)_2$	(3- MeOC ₆ H ₄) ₃ P	K ₃ PO ₄	AgOAc	0

A plausible reaction pathway for the domino reaction is depicted in Scheme 6. The formation of π -alkynepalladium complex induces the intramolecular nucleophilic attack of the nitrogen to the activated carbon–carbon triple bond. The generated σ -indolylpalladium intermediate is trapped by the electron-poor double bond via *syn* insertion. After C–C bond rotation and *syn* β -hydride elimination, azepino[5,4,3-*cd*]indole structure is built, and palladium(0) is formed. The catalytically active palladium(II) is regenerated by oxidation of palladium(0).

In conclusion, we have developed a method to construct 3,4fused tricyclic azepino[5,4,3-*cd*]indole derivatives form 2alkynylanilines, isocycanides, and α , β -unsaturated acids. This synthetic process involves a regioselective meta-functionalization of 2-alkynylanilines using a dearomatization strategy, and a

Scheme 5



Scheme 6. Plausible Reaction Pathway of Domino Reaction



palladium-catalyzed domino heterocyclization/Heck reaction. A current effort has also been made to extend its scope and to explore its reaction mechanism and possible synthetic applications, and these results will be reported in due course.

ASSOCIATED CONTENT

Supporting Information

Experimental procedures, characterization data, NOESY spectrum of compound **31**, copies of ¹H NMR and ¹³C NMR of new compounds, and crystallographic data of compound **36** (CIF). This material is available free of charge via the Internet at http://pubs.acs.org.

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Notes

The authors declare no competing financial interest.

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REFERENCES

(1) (a) Humphrey, G. R.; Kuethe, J. T. Chem. Rev. 2006, 106, 2875.
(b) Cacchi, S.; Fabrizi, G. Chem. Rev. 2005, 105, 2873. (c) Gribble, G. W. J. Chem. Soc., Perkin Trans. 1 2000, 1045. (d) Saxton, J. E. In The Chemistry of Heterocyclic Compounds; Wiley: New York, 1983; Vol. 25, Part IV.

(2) Mukherjee, J.; Menge, M. Adv. Biochem. Eng. Biotechnol. 2000, 68, 1.

(3) (a) Muratake, H.; Natsume, M. *Tetrahedron* 1990, 46, 6331.
(b) Baran, P. S.; Thomas, J.; Maimone, T. J.; Richter, J. M. *Nature* 2007, 446, 404. (C) Rafferty, R. J.; Williams, R. M. J. Org. Chem. 2012, 77, 519.

(4) (a) Kamata, K.; Suetsugu, T.; Yamamoto, Y.; Hayashi, M.; Komiyama, K.; Ishibashi, M. *J. Nat. Prod.* **2006**, *69*, 1252. (b) Mayer, G.; Wille, G.; Brenner, M.; Zeitler, K.; Steglich, W. Synthesis **2011**, *2*, 330.

(5) (a) Numata, A.; Takahashi, C.; Ito, Y.; Takada, T.; Kawai, K.; Usami, Y.; Matsumura, E.; Imachi, M.; Ito, T.; Hasegawa, T. *Tetrahedron Lett.* **1993**, *34*, 2355. (b) Zuo, Z.; Ma, D. *Angew. Chem, Int. Ed.* **2011**, *50*, 12008.

(6) (a) Mascal, M.; Modes, K. V.; Durmus, A. Angew. Chem., Int. Ed.
2011, 50, 4445. (b) Qin, H.; Xu, Z.; Cui, Y.; Jia, Y. Angew. Chem., Int. Ed. 2011, 50, 4447. (c) Hu, W.; Qin, H.; Cui, Y.; Jia, Y. Chem.—Eur. J.
2013, 19, 3139. (d) Sun, D.; Zhao, Q.; Li, C. Org. Lett. 2011, 13, 5302.
(e) Leduc, A. B.; Kerr, M. A. Eur. J. Org. Chem. 2007, 237. (f) Koizumi, Y.; Kobayashi, H.; Wakimoto, T.; Furuta, T.; Fukuyama, T.; Kan, T. J. Am. Chem. Soc. 2008, 130, 16854.

(7) (a) Huters, A. D.; Quasdorf, K. W.; Styduhar, E. D.; Garg, N. K. J. Am. Chem. Soc. 2011, 133, 15797. (b) Bhat, V.; Allan, K. M.; Rawal, V. H. J. Am. Chem. Soc. 2011, 133, 5798.

(8) (a) Peat, A. J.; Buchwald, S. L. J. Am. Chem. Soc. 1996, 118, 1028.
(b) Jackson, B.; Hester, J. J. Org. Chem. 1964, 29, 1158.

(9) (a) Xu, Z.; Zhang, F.; Zhang, L.; Jia, Y. Org. Biomol. Chem. 2011,
9, 2512. (b) Bronner, S. M.; Goetz, A. E.; Garg, N. K. Synlett 2011,
2599. (c) Bronner, S. M.; Goetz, A. E.; Garg, N. K. J. Am. Chem. Soc.
2011, 133, 3832. (d) Meseguer, B.; Alonso-Diaz, D.; Griebenow, N.;
Herget, T.; Waldmann, H. Angew. Chem., Int. Ed. 1999, 38, 2902.
(e) Moreno, O. A.; Kishi, Y. J. Am. Chem. Soc. 1996, 118, 8180.

(10) Park, I.-K.; Park, J.; Cho, C.-G. Angew. Chem., Int. Ed. 2012, 51, 2496.

(11) (a) Schçnherr, H.; Leighton, J. L. Org. Lett. 2012, 14, 2610.
(b) Cheng, D.-J.; Wu, H.-B.; Tian, S.-K. Org. Lett. 2011, 13, 5636.
(c) Yamada, K.; Namerikawa, Y.; Haruyama, T.; Miwa, Y.; Yanada, R.; Ishikura, M. Eur. J. Org. Chem. 2009, 5752.

(12) Greshock, T. J.; Funk, R. L. J. Am. Chem. Soc. 2006, 128, 4946.
(13) (a) Yonemitsu, O.; Cerutti, P.; Witkop, B. J. Am. Chem. Soc. 1966, 88, 3941. (b) Kobayashi, T.; Spande, T. F.; Aoyagi, H.; Witkop, B. J. Med. Chem. 1969, 12, 636. (c) Mascal, M.; Moody, C. J.; Slawin, A. M. Z.; Williams, D. J. J. Chem. Soc., Perkin Trans.1 1992, 823.

(14) (a) Trost, B. M.; McDougall, P. J. Org. Lett. 2009, 11, 3782.
(b) Lauchli, R.; Shea, K. J. Org. Lett. 2006, 8, 5287. (c) Bur, S. K.; Padwa, A. Org. Lett. 2002, 4, 4135.

(15) (a) Breazzano, S. P.; Poudel, Y. B.; Boger, D. L. J. Am. Chem. Soc. 2013, 135, 1600. (b) Shan, D.; Gao, Y.; Jia, Y. Angew. Chem., Int. Ed. 2013, 52, 4902.

(16) For reviews on dearomatization, see: (a) Bach, T. Angew. Chem., Int. Ed. 1996, 35, 729. (b) Quideau, S.; Pouységu, L. Org. Prep. Proced. Int. 1999, 31, 617. (c) Pape, A. R.; Kaliappan, K. P.; Kündig, E. P. Chem. Rev. 2000, 100, 2917. (d) Magdziak, D.; Meek, S. J.; Pettus, T. R. R. Chem. Rev. 2004, 104, 1383. (e) Keane, J. M.; Harman, W. D. Organometallics 2005, 24, 1786. (f) López Ortiz, F.; Iglesias, M. J.; Fernández, I.; Andújar Sánchez, C. M.; Gómez, G. R. Chem. Rev. 2007, 107, 1580. (g) Quideau, S.; Pouységu, L.; Deffieux, D. Synlett 2008, 467. (h) Pouységu, L.; Deffieux, D.; Quideau, S. Tetrahedron 2010, 66, 2235. (i) Roche, S. P.; Porco, J. A., Jr. Angew. Chem., Int. Ed. 2011, 50, 4068. (j) Zhuo, C.-X.; Zhang, W.; You, S.-L. Angew. Chem., Int. Ed. 2012, 51, 12662. (k) Ding, Q.; Ye, Y.; Fan, R. Synthesis 2013, 45, 1. (17) (a) Yang, M.; Tang, J.; Fan, R. Chem. Commun. 2012, 48, 11775. (b) Zhang, L.; Li, Z.; Fan, R. Org. Lett. 2012, 14, 6076. (c) Zhang, L.; Wang, W.; Fan, R. Org. Lett. 2013, 15, 2018. (d) Yang, M.; Tang, J.; Fan, R. Org. Lett. 2013, 15, 3464.

(18) (a) Banfield, S. C.; England, D. B.; Kerr, M. A. Org. Lett. 2001,
3, 3325. (b) Jackson, S. K.; Banfield, S. C.; Kerr, M. A. Org. Lett. 2005,
7, 1215. (c) Quideau, S.; Pouységu, L.; Ozanne, A.; Gagnepain, J. Molecules. 2005, 10, 201. (d) Jean, A.; Cantat, J.; Bérard, D.; Bouchu,
D.; Canesi, S. Org. Lett. 2007, 9, 2553. (e) Sabot, C.; Bérard, D.;
Canesi, S. Org. Lett. 2008, 10, 4629. (f) Giroux, M.-A.; Guérard, K. C.;
Beaulieu, M.-A.; Sabot, C.; Canesi, S. Eur. J. Org. Chem. 2009, 3871.

(19) For reviews on hypervalent iodines, see: (a) Zhdankin, V. V.; Stang, P. J. Chem. Rev. 2002, 102, 2523. (b) Wirth, T. Hypervalent Iodine Chemistry: Modern Developments in Organic Synthesis. Top. Curr. Chem. 2003, 224. (c) Ciufolini, M. A.; Braun, N. A.; Canesi, S.; Ousmer, M.; Chang, J.; Chai, D. Synthesis 2007, 3759. (d) Zhdankin, V. V.; Stang, P. J. Chem. Rev. 2008, 108, 5299.

(20) Amjad, M.; Knight, D. W. Tetrahedron Lett. 2004, 45, 539.

(21) (a) Mitsudo, K.; Thansandote, P.; Wilhelm, T.; Mariampillai, B.; Lautens, M. Org. Lett. 2006, 8, 3939. (b) Yue, D.; Larock, R. C. Org. Lett. 2004, 6, 1037. (c) Yue, D.; Yao, T.; Larock, R. C. J. Org. Chem. 2006, 71, 62.

(22) (a) Isono, N.; Lautens, M. Org. Lett. 2009, 11, 1329.
(b) Martínez, C.; Álvarez, R.; Aurrecoechea, J. M. Org. Lett. 2009, 11, 1083. (c) Álvarez, R.; Martínez, C.; Madich, Y.; Denis, J. G.; Aurrecoechea, J. M.; de Lera, Á. R. Chem.—Eur. J. 2010, 16, 12746.
(d) Boyer, A.; Isono, N.; Lackner, S.; Lautens, M. Tetrahedron 2010, 66, 6468. (e) Martínez, C.; Aurrecoechea, J. M.; Madich, Y.; Denis, J. G.; de Lera, A. R.; Álvarez, R. Eur. J. Org. Chem. 2012, 99.

(23) Crystallographic data for compound **36** is available free of charge from the Cambridge Crystallographic Data Centre, accession no. CCDC 975321.

(24) Witulski, B.; Alayrac, C. Angew. Chem., Int. Ed. 2002, 41, 3281.

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