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Sulfonamide derivatives of bridgehead substituted bicyclo[4.2.1]nonanes as γ -secretase inhibitors

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Abstract—Bridgehead substituted derivatives of bicyclo[4.2.1]nonanes were synthesized and shown to be potent inhibitors of γ -secretase. Two related series were synthesized to explore the SARs. More potent compounds were found in the non-benzofused series compared with the benzofused series. One compound from each series showed good exposure in the hepatic portal vein (HPV) following oral dosing to rats.

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Alzheimer's disease (AD) is the most common cause of dementia in the elderly, and with life expectancy increasing, more patients than ever before will suffer progressive memory impairment, cognitive deficits and behavioural problems as a result of AD. Current therapies are palliative. Approaches towards a disease modifying therapy include reducing the production of amyloid- β 42 (A β 42) through inhibition of β -secretase or γ -secretase. Such approaches have been of interest to the pharmaceutical industry, and as a result, a number of γ -secretase inhibitors/modulators are in clinical development. 1

Recently we described the use of sulfonamide derivatives of substituted bicyclo[4.2.1]nonanes and substituted cyclic sulfamides represented by **A** and **B** as γ -secretase inhibitors with potential for the treatment of Alzheimer's disease.^{2,3}

Here we present new γ -secretase inhibitors, represented by \mathbf{C} and \mathbf{D} which, given the substitution pattern on the respective cores, represent synthetically challenging types of compounds for which pharmacological properties are unknown. However, related compounds have been proposed for use as analgesics and anti-inflammatories.⁴ Furthermore, analogues of such bicyclo[4.2.1]nonanes were prepared to aid in the re-definition of Bredt's rule.⁵

The preparation of bridgehead substituted bicyclo[4.2.1]nonanes possessing a quaternary carbon inherent within this structural motif has attracted two main synthetic strategies (Scheme 1). One involves alkylating the carbon bearing the most acidic proton prior to cyclisation onto the methylene adjacent to the carbonyl (method A)^{5a} and the second, generation of the bicyclic [4.2.1]nonane skeleton through reduction of an enol lactone (method B).^{5b}

We planned to utilize the bridgehead position as a site from which a range of groups (R in C and D) could be installed to build upon the SAR of previously disclosed compounds² by exploring the effects of introducing polar, basic and lipophilic groups on the potency and pharmacokinetic properties of these molecules. Herein we describe the synthesis of bridgehead substituted benzofused and non-benzofused bicyclo[4,2,1]nonanes, and derivatisation to sulfonamides as γ -secretase inhibitors.

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Scheme 1. Reagents and conditions: (a) 1,4-dibromobutane, K₂CO₃, acetone, reflux, o/n; (b) NaH, DMF, toluene, rt, 3 h (32% over 2 steps); (c) i—TiCl₄, Et₃N, 5-chlorothiophene-2-sulfonamide, THF; ii—NaBH₄, THF (67% over 2 steps); (d) lithium aluminium hydride, ether (99%); (e) SO₃·pyridine, DMSO (92%); (f) methyl diethylphosphonoacetate, LiOH·H₂O, THF (95%); (g) DIBAL, toluene (79%); (h) i—NaOH, THF (99%); ii—diphenylphosphoryl azide, Et₃N, toluene; iii—t-BuOH (49% over 2 steps); (i) HCl, ether (99%).

For the preparation of non-benzofused bicyclo[4.2.1] nonanes we used method A (Scheme 1). Thus, a combination of 1,4-dibromobutane as bis-electrophile, potassium carbonate as base and 1 followed by cyclisation with NaH as base in a mixture of DMF and toluene gave ketone 2 in modest overall yield. Attempts to improve the yield such as using different bases (e.g., LDA) and solvents (e.g., ether) failed. One by-product previously reported was the *O*-alkylated compound 3,5a although no attempt was made to isolate it here. Though low yielding, sufficient quantities of ketone 2 were isolated which was converted to sulfonamide 4 using a one-pot imine formation/reduction protocol.

The stereochemistry of the C–N bond in 4 was axial as proven by 1H NMR nOe experiments. Subsequently, reduction with lithium aluminium hydride produced alcohol 5 which was oxidised to the aldehyde, then a Horner–Emmons reaction to the α,β -unsaturated ester followed by further reduction with DIBAL furnished allylic alcohol 6. For the introduction of a bridgehead amine, ester 4 was converted to the acid then BOC-amine 7 via a Curtius reaction, and deprotected with acid to provide amine 8.

Heterocyclic groups were appended from the core by converting ester 4 to the amide 9 via saponification,

followed by generation of the acid chloride and reaction with ammonia (Scheme 2). Subsequently, the amide was dehydrated to nitrile 10 and imidazoline 11 formed by reaction with ethylenediamine. Oxidation with barium manganate gave the imidazole 12.

Benzofused analogues were prepared by analogy with method A. Thus, xylidene dibromide was reacted with

Scheme 2. Reagents and conditions: (a) i—NaOH, THF (93%); ii—oxalyl chloride, THF; iii—ammonia, THF (79% over 2 steps); (b) POCl₃, NaHCO₃ (81%); (c) i—gaseous HCl, MeOH, DCM; ii—ethylene diamine, THF (39% over 2 steps); (d) BaMnO₄, THF (51%).

ethyl 2-carboxycyclopentanone 1 and sodium ethoxide as base to give bromide 13 in low yield (Scheme 3). Simple attempts to improve the yield (stronger base, e.g., LDA) failed. The cyclisation again proved low yielding, in our hands using LDA as base gave 37% of ketone 14. Subsequently, 14 was converted to amine 15 using a two-step procedure via the oxime and reduction with zinc in acetic acid. The reduction was highly stereoselective giving >95% of a single diastereoisomer (by NMR of the crude product).

Capping of the amine with 5-chlorothiophene-2-sulfonyl chloride afforded sulfonamide 16, which acted as the central intermediate for subsequent manipulations. Improved yields were found for the preparation of 16 using these conditions than those used for 4. The stereochemistry of the C-N bond was again found to be axial as shown by nOe studies. For incorporating a bridgehead amine, intermediate 16 was saponified and the acid exposed to Curtius conditions to provide BOC-protected amine 17, which was deprotected with acid to give 18. Also, ester 16 was reduced with LAH to give primary alcohol 19.

To further scope out the SAR, bridgehead hydroxyl compound 22 was synthesized via a Baeyer–Villiger oxidation of ketone 21, produced from a Grignard reaction of the Weinreb amide of acid 20, followed by hydrolysis (Scheme 4).

Compounds were assayed using a whole cell γ -secretase inhibition assay using SHSY5Y cells⁶ (Table 1). Compounds **23** and **24**, as previously reported by our group, demonstrated reasonable potency. Overall, no

Scheme 3. Reagents and conditions: (a) xylidene dibromide, NaOEt, DMF, rt; (b) LDA, THF (37% over 2 steps); (c) hydroxylamine·HCl, NaOAc, THF; (d) zinc, AcOH; (e) 5-chlorothiophene-2-sulfonyl chloride, Et₃N, DCM (65% over 3 steps); (f) NaOH, THF (99%); (g) i—diphenyl phosphoryl azide, Et₃N, toluene; ii—t-BuOH (48% over 2 steps); (h) HCl, ether (99%); (i) lithium aluminium hydride, THF (90%).

Scheme 4. Reagents and conditions: (a) i—oxalyl chloride, *N-O*-dimethylhydroxylamine·HCl, pyridine, THF (46%); ii—MeMgBr, THF (73%); (b) i—*m*-CPBA, Na₂CO₃, DCM (32%); ii—lithium aluminium hydride, THF (76%).

significant improvement in potency was observed with the new compounds reported here. The non-benzofused series generally showed improved potency compared with benzofused compounds, notably for bridgehead ester (4 and 16) and primary alcohol substituents (5 and 19). This could be due to a size restriction in the active site, for which the presence of the benzofused moiety limits the space available to the bridgehead substituent. Additionally, basic groups were not tolerated in either series (8, 18, 11 and 12). Within both series, only hydroxyl containing groups 5 and 6 displayed comparable potency to the parent compound 23, although other small groups were moderately well tolerated (9 and 10).

In the benzofused series, a hydroxyl group was well tolerated (22), showing improved potency compared to 24, however introducing a methylene spacer (19) resulted in

Table 2. Rat HPV levels for compounds 5, 19 and 24

Compound	Dose (mg/kg)	HPV (ng h/ml)
24	30	8
5 ⁷	1	55
198	5	425

a significant loss in potency. In addition, analogue 18 was inactive.

Pharmacokinetic studies using oral dosing to rats showed that compound **24** was very poorly absorbed into the hepatic portal vein (HPV). To assess the effect on HPV levels of compounds containing polar bridgehead substituents, compounds **5** and **19** were dosed orally to rat. Gratifyingly, compound levels in the HPV were significantly improved (Table 2).

We have shown that the bridgehead position of compounds 23 and 24 could be substituted to produce two new classes of compounds. The non-benzofused compounds generally showed better γ -secretase potency than the benzofused series, in particular hydroxyl group containing substituents gave compounds equipotent (5 and 6) to parent (23). However, one compound from the benzofused series (22) showed potency comparable to parent (24). Furthermore, one compound from each series showed reasonable absorption into the HPV after oral dosing to rats (5 and 19). This represented an improvement compared with compound 24.

Table 1. Inhibitory activity of sulfonamides 4–12 and 16–24⁶

R	Compound	IC ₅₀ (nM) (n)	Compound	IC ₅₀ (nM) (n)
Н	23	$62 \pm 23 (23)$	24	63 ± 14 (4)
CO ₂ Et	4	$212 \pm 57 (4)$	16	>10,000 (3)
CH ₂ OH	5	$105 \pm 22 \ (4)$	19	$491 \pm 182 (3)$
ОН	6	81 ± 46 (4)	_	_
NHBOC	7	$416 \pm 183 (2)$	17	>10,000 (2)
NH_2	8	>10,000 (2)	18	>10,000 (2)
$CONH_2$	9	$685 \pm 26 (3)$	_	_
CN	10	$257 \pm 18 (3)$	_	_
N N	11	>10,000 (2)	_	_
N N N	12	>10,000 (2)	_	_
ОН	_	_	22	23 (2)

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- 7. For compound **5**: (360 MHz ¹H NMR, CDCl₃) 1.25–1.93 (12H, m), 2.19–2.27 (2H, m), 3.28 (1H, d, *J* = 11.3), 3.48 (1H, d, *J* = 11.3), 3.61–3.65 (1H, m), 4.90 (1H, d, *J* = 8.6), 6.92 (1H, d, *J* = 4.0), 7.42 (1H, d, *J* = 4.0); (¹³C NMR, CDCl₃) 25.9, 27.3, 29.9, 31.6, 32.1, 35.6, 40.0, 50.3, 62.1, 69.5, 128.6, 133.5, 139.2, 141.0. *m/z* (ES+): 332 [M–OH]⁺.
- 8. For compound **19**: (360 MHz ¹H NMR, CDCl₃) 1.13 (2H, m), 1.58 (2H, m), 2.20 (1H, d, *J* = 15.7), 2.35 (1H, m), 2.58 (1H, dd, *J* = 7.7, 16.1), 2.78 (1H, d, *J* = 15.9), 2.87 (1H, m), 3.04 (1H, d, *J* = 16.2), 3.42 (1H, d, *J* = 11.2), 3.63 (1H, d, *J* = 11.2), 3.70 (1H, dd, *J* = 6.7, 6.8), 5.47 (1H, d, *J* = 7.6), 6.95 (1H, d, *J* = 3.7), 7.07 (4H, m), 7.46 (1H, d, *J* = 3.6); *m/z* (ES+): 398 MH⁺.