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Organocatalysed synthesis of isoxazolines initiated by a chemoselective oxa-Michael reaction of *N*-BocNHOH[†]

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An organocatalysed and chemoselective one-pot oxa-Michael-cyclocondensation reaction of N-BocNHOH to unsaturated α -ketoesters is reported which affords an original entry to enantioenriched 3-isoxazoline carboxylate derivatives as biorelevant heterocyclic frameworks.

Introduction

The organocatalysed hetero-Michael reaction emerged as a powerful synthetic methodology to create enantioselectively C–N or C–O bonds.^{1,2} By means of bisnucleophiles such as hydroxylamines, a subsequent cyclisation step is allowed to construct, in a straightforward fashion, non-racemic isoxazolidine derivatives,³ important heterocyclic frameworks frequently found in biologically active compounds (Fig. 1).^{4,5}

In this context, few research groups have achieved an asymmetric organocatalysed domino hetero-Michael-cyclisation reaction towards the elaboration of isoxazolidine derivatives by addressing the NH versus OH chemoselectivity of hydroxylamine bisnucleophiles (Scheme 1).6 Shibata and colleagues reported that native hydroxylamine led to a domino oxa-Michael-cyclocondensation reaction onto activated fluorinatedenones upon the asymmetric phase transfer catalytic regime with strong bases (Scheme 1a).⁷ Pihko and collaborators elegantly used oximes as transient-nitrogen protected NH2OH nucleophiles to form 3-unsubstituted 2-isoxazolines via an iminium catalysed oxa-Michael reaction followed by an acid promoted trans-imination cyclisation (Scheme 1a).8 On the other hand, the group of Córdova performed an efficient iminium catalysed domino aza-Michael-hemiacetal formation reaction starting from commercially available N-BocNHOH 1a (Scheme 1b).⁹

Alternatively, we would like to report here on an original domino oxa-Michael-hemiaminal formation reaction of N-BocNHOH **1a** making use of unsaturated α -ketoesters **2**,

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Fig. 1 Bioactive 3-isoxazolecarboxylate derivatives.



Scheme 1 Domino aza- versus oxa-Michael-cyclisation reaction.

namely the 4-substituted 2-oxo-3-butenoates, upon Brønsted base catalysis (Scheme 1c).¹⁰ This sequence not only reveals an uncommon O-chemoselectivity during the conjugated addition reaction of *N*-BocNHOH **1a** bisnucleophile, giving rise to



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intermediate 3, but eventually leads to a concise elaboration of biologically important isoxazolines 4 flanked by a carboxylate functional group at C3 after a simple acid promoted Boc removal (Scheme 1c *versus* Fig. 1).¹¹ Despite some seminal experimental observations concerning the oxa-Michael outcome of hydroxyurea NH₂CONHOH with enals by Petrus and co-workers,¹² the asymmetric developments remain to be achieved. Towards this goal, we are pleased to describe the preliminary results towards an original asymmetric organocatalytic synthesis of 3-isoxazoline carboxylate derivatives. It is important to note that some of these chiral structures 4 were previously only directly accessible in a non-racemic form through enzymatic resolution or diastereoselective reaction.¹³⁻¹⁵

Results and discussion

At the onset of this project, it was shown that an *a priori* chemoselective oxa-Michael reaction of N-BocNHOH 1a smoothly took place with unsaturated keto-ester 2a in the presence of quinuclidine after 24 hours in toluene (Table 1, entry 2). Indeed, the starting materials 1a and 2a were recovered without Brønsted base (entry 1). The analyses of the crude reaction products mainly revealed two compounds which were assumed to be a mixture of cyclized hemi-aminal epimers 3a (Scheme 1c, $R = (CH_2)_2Ph$) due to the absence of a ketone functional group by ¹³C NMR (see ESI[†]). Next, the one-pot addition of trifluoroacetic acid (TFA) in excess easily provided the corresponding isoxazoline 4a through a formal sequence involving both dehydration and Boc-removal events with an overall 63% yield. It is worthy of note that no isoxazoline 4a formation was noticed in the presence of TFA alone (entry 3). This rules out an acid catalyzed formation of isoxazolidine 3a from 2a (Scheme 1c). With this original one-pot sequential oxa-Michael-cyclocondensation reaction in hand, we sought an asymmetric version but a screening of a large array of chiral Brønsted bases was required (see ESI† for an overview of screened conditions). Neither quinine derivatives 5, 6 (C9-OH or C9-OBz) nor the 9-epi-amino homologues 7, 8 (C9-NH₂ or C9-NHAc) were able to achieve a significant enantioselective transformation, although these organocatalysts furnished 4a in good yields (Table 1; entries 4–7). To our delight, the bifunctional 9-epi-aminoquinines 9-10 led to improved enantioselective excesses ranging from 34% ee with C9-thiourea derived catalysts 9 to 55% ee with the C9-squaramide one QN-SQ 10 with good (82%) overall yields (entries 8 and 9).¹⁶ Interestingly, increased enantiomeric excesses were measured by adding N-BocNHOH 1a dropwise in solution to give up to 73% ee (entry 10). The origin of this phenomenon has not been fully understood so far, but constitutes a reproducible procedure in any subsequent reactions. The N-CbzNHOH 1b and NH₂CONHOH 1c nucleophiles were also evaluated in the presence of squaramide catalyst 10 but lower selectivities of 35% and 58% ee respectively were measured (entries 11 and 12). But the most tedious issues with these N-functionalized

Table 1 Towards an enantioselective approach^a



^{*a*} *N*-protected hydroxylamines **1** (0.11 mmol, 1.1 equiv.) as a solid were added in one-portion into a solution of unsaturated ketoesters **2a-b** (0.10 mmol, 1 equiv.), catalyst (0.01 mmol, 0.1 equiv.) in toluene (0.05 M); then TFA (1.0 mmol, 10 equiv.). ^{*b*} Isolated yield after column chromatography. ^{*c*} Determined by chiral HPLC. ^{*d*} In the presence 0.1 or 30 equivalents of TFA. ^{*e*} The *N*-Boc hydroxylamine **1a** was added into a solution in one-portion. ^{*f*} The *N*-Boc hydroxylamine **1a** was added dropwise into a solution over 4 hours. ^{*g*} TFA was replaced by TMSI. ^{*h*} TFA was replaced by HCl in MeOH.

hydroxylamine derivatives **1b** and **1c** were related to the delicate acid-promoted isoxazoline **4a** formation requiring adapted reaction conditions which eventually led to modest isolated yields lower than 16% (entries **11** and **12**). Importantly, it was shown that no reaction occurred with enal **2b** in the presence of organocatalyst **10**. Accordingly, the required ester moiety on the 2-oxo-3-butenoate **4a** structure is probably an activated EWG for both unsaturated enone towards the oxa-Michael process with catalyst **10** and the ketone functional group during the hemi-aminal **3** formation.¹⁰ In all likelihood a kinetically driven **O**-alkylation is occurring, evidently of the less sterically hindered moiety of *N*-BocNHOH **1a**,^{17,18} and this C-**O** bond formation is secured by the subsequent cyclization to stable intermediate **3** (Scheme 1b *versus* **1**c).^{9a}

Next, we investigated the reaction conditions through the direct addition of *N*-BocNHOH **1a** in solution for practical reasons (Table 2). It was found that quinidine derived squaramide organocatalyst **11** (QD-SQ) (entry 1) could be employed instead of quinine QN-SQ with a slightly better ee but a reverse enantioselectivity (see Table 1, entry 10). The use of various aromatic solvents (entries 2 and 3, Table 2) did not improve

Table 2 Screening of conditions^a



^{*a*} Unsaturated ketoesters **2a** (0.10 mmol, 1 equiv.), *N*-Boc hydroxylamines **1a** (0.11 mmol, 1.1 equiv.) were added in one-portion into a solution, catalysts **10** or **11** (0.01 mmol, 0.1 equiv.), toluene (0.05 M); and then TFA (1.0 mmol, 10 equiv.). ^{*b*} Isolated yield after column chromatograph. ^{*c*} Determined by chiral HPLC.

20

82

70

the induction but less polar methylcyclohexane was detrimental and led to poor solubility issues (entry 4). On the other hand, moving from dichloromethane to more polar solvents decreased markedly the enantioselectivities (entries 5–8). The asymmetric outcome was insensitive to the decrease of temperature (entry 9) but the concentration has to be maintained at 0.05 M; otherwise lower enantiomeric excess was observed (entries 10–12).

Then, we evaluated this one-pot oxa-Michael-cyclocondensation sequence on various unsaturated α -ketoester derivatives 2 (Table 3). At that stage, it was found that 30 equivalents of TFA

 Table 3
 Scope of the reaction^a

HO´ 1a	R ¹ CC 2a-k	$p_2 R^2 = \frac{1.0}{2.00}$	Cat. (10 mol%) luene, 20°C, 24 TFA) 4 h ►	R ¹	CO ₂ R ²
Entry	R ¹	R^2	Catalyst ^a	Prod.	$\operatorname{Yield}^{b}(\%)$	ee ^c (%)
1	$Ph(CH_2)_2$	Et (2a)	QN-SQ (10)	4a	87	74 (S)
2	$Ph(CH_2)_2$	Et (2a)	QD-SQ (11)	4a	82	76 (R)
3	$Ph(CH_2)_2$	Me(2c)	QN-SQ (10)	4c	87	72 (S)
4	iso-Butyl	Et (2d)	QN-SQ (10)	4d	82	68
5	Cyclohexyl	Et (2e)	QN-SQ (10)	4e	73	31
6	<i>n</i> -Heptyl	Et (2f)	QN-SQ (10)	4f	86	77
7	$EtO(CH_2)_2$	Et (2 g)	QN-SQ (10)	4g	80	67
8	BnOCH ₂	Et (2 h)	QN-SQ (10)	4h	68	72
9	$H_2CCH(CH_2)_2$	Et (2i)	QN-SQ (10)	4i	79	78
10	Ph	Me (2j)	QN-SQ (10)	4j	21^d	5
11	3,4-DiClPh	Me (2k)	QN-SQ (10)	4k	42^d	8

^{*a*} Unsaturated ketoesters (0.250 mmol, 1 equiv.), *N*-Boc hydroxylamine **1a** (0.275 mmol, 1.1 equiv.) were added dropwise into a solution over 4 hours, catalyst (0.025 mmol, 0.1 equiv.), toluene (0.05 M); then TFA (7.5 mmol, 30 equiv.). ^{*b*} Isolated yield after column chromatography. ^{*c*} Determined by chiral HPLC. ^{*d*} Yield after 72 h.

provided more general conditions for the formation of isoxazolidines with various enones 2. First of all, with this optimized protocol, it was found that quinine 10 or quinidine squaramide pseudo-enantiomer catalysts 11 could be used with a similar efficiency in order to furnish both S- or R-enantiomer of isoxazoline 4a with 74–76% ee (entries 1 and 2).¹⁹ Both methyl and ethyl ester isoxazolines 4a and 4c were synthesized with rather similar enantiomeric excesses (entries 1 versus 3). Although 4-iso-butyl 2-oxo-3-butenoate 2d led to respectable 68% ee, the α-branched cyclohexyl pendant 2e underwent a smooth transformation into 4e in 73% yield albeit with lower 31% ee (entries 4 and 5). However, a longer *n*-heptyl alkyl chain was well tolerated giving rise to 77% ee (entry 6). Moreover, enones 2g-i with an alkyl chain flanked by ethers or alkene functional groups furnished the corresponding isoxazolines 4g-i with good yields and ee ranging from 67% to 78%. Although 4-alkyl substituted 2-oxo-3-butenoates 2a-i allowed the formation of the corresponding isoxazolines 4a-i with high yield and moderate to good enantiomeric excesses (entries 1-9), 4-aryl substituted derivatives 2j-k revealed a slow oxa-Michael reaction rate with low enantiomeric excess (entries 10 and 11). This outcome has not been fully studied at the moment, but the more stable 4-aryl substituted enones 4j-k might undergo an oxa-Michael addition under thermodynamic equilibrated conditions together with featuring a steric issue as seen with the lower enantiomeric excess obtained with the 4-cyclohexyl derivative 4e.

Most of the 5-alkyl substituted 3-isoxazoline carboxylate derivatives being novel products, we sought to both determine their absolute configurations and probe useful chemical transformations (Scheme 2). Inspired by a previous study of Nishiyama and Pihko, we attempted the fragmentation of the oxazole ring.²⁰ To our delight, after a facile saponification reaction into carboxylic acid **12**, a straightforward decarboxylative-fragmentation took place on the crude product **12** at 80 °C to form the β -hydroxy acetonitrile **13** with excellent 78% yield over the two steps. This *S*-featuring known alcohol **6** was obtained without racemization providing accordingly the absolute configuration of precursor **4a**.^{20b}

With this structure in hand, a plausible transition state is depicted in Fig. 2, stemming from previous proposals of Jørgensen and co-workers.^{16,21} The α -ketoester would be recognized and activated through a double hydrogen bonding network of the squaramide moiety while the R¹-pendant is pointing away from the bulky quinuclidine part. Then, the *Si* face approach of the incoming oxygen-nucleophile is assisted by the top-face position of the tertiary amine.



Scheme 2 Transformation and determination of the absolute configuration.

11

Toluene

0.025 M



Fig. 2 Proposed transition state.

Conclusions

In summary, although *N*-BocNHOH was previously used in the aza-Michael reaction upon iminium organocatalysis with enals,^{9*a*} we report in this paper an enantioselective oxa-Michael event of this bisnucleophile with unsaturated α -ketoesters upon Brønsted base catalysis. This original chemoselective process followed by a one-pot acid promoted cyclo-condensation reaction affords a unique entry to non-racemic bio-relevant 3-isoxazoline carboxylate derivatives. The search for more efficient squaramide derived organocatalysts is currently of interest.

Experimental section

Solvents were purchased as dehydrated ones, and toluene or CH₂Cl₂ was distilled on CaH₂. All reagents were used as received unless otherwise indicated. Chromatographic purification of compounds was achieved with silica gel 60 (40-63 µm). Thin layer chromatography was carried out on silica gel 60 F254 (1.1 mm) with spot detection under UV light or phosphomolybdic acid or ninhydrin or KMnO4 oxidation. ¹H NMR spectra were recorded on a Bruker AVANCE 300 at 300 MHz. Data appear in the following order: chemical shifts in ppm, number of protons, multiplicity (s, singlet; d, doublet; t, triplet; q, quartet; m, multiplet), and coupling constant *J* in Hz. ¹³C NMR spectra were acquired at 75.4 MHz operating with broad band ¹H decoupling. The hydrogen multiplicity was obtained by DEPT135. IR spectra were recorded on a Perkin Elmer IRTF 100 spectrometer using an ATR (Attenuated Total Reflectance) sampling with a solid dispersed or neat or on a Perkin Elmer IRFT 1650 with a solid dispersed on KBr pastille. Mp's were uncorrected. HRMS analyses were measured on a Q-TOF Micro WATERS spectrometer. HPLC analyses were performed with Daicel Chiralpack® or Daicel ChiralCel® columns (4.6 mm × 25 cm). A spectrosystem UV 1000 thermofisher detector and a chiral detector (polarimeter) JACSCO OR-1590 were used.

Representative procedure for compound 4a

To a solution of unsaturated ketoester 2a (0.25 mmol) and catalyst 10 (15.8 mg, 0.025 mmol, 0.1 equiv.) in toluene (5 mL,

0.05 M) in a Schlenk flask was added dropwise over 4 h a solution of N-Boc hydroxylamine (36.6 mg, 0.275 mmol, 1.1 equiv.) in toluene (750 µL). After stirring for 24 h at 20 °C, TFA (580 µL, 7.6 mmol, 30 equiv.) was added and the resulting solution was stirred for 1 h. The reaction mixture was partitioned between a saturated aqueous NaHCO3 solution (10 mL) and Et₂O (10 mL). The organic layer was washed with brine (10 mL), dried over anhydrous Na₂SO₄, and filtered and concentrated under reduced pressure. The crude residue was purified by silica gel chromatography to give the expected 4a as a pale yellow oil (54 mg, 87%). $R_{\rm f} = 0.25$ (petroleum ether-EtOAc: 9/1). $[\alpha]_{D}^{20} = -104.5$ (c 1.01, CHCl₃, 74% ee). IR (neat) $\nu_{\rm max}$ 3026, 2982, 2938, 2862, 1738, 1721, 1715, 1588, 1454, 1380, 1257, 1125, 1018, 934, 749 and 701 cm⁻¹. ¹H NMR (300 MHz; CDCl₃) δ_H 7.32-7.27 (2H, m), 7.23-7.18 (3H, m), 4.85-4.74 (1H, m), 4.35 (2H, q, J = 7.1 Hz), 3.25 (1H, dd, J = 11.0, 17.5 Hz), 2.85 (1H, dd, J = 8.3, 17.5 Hz), 2.84-2.67 (2H, m), 2.17-2.03 (1H, m), 1.96-1.84 (1H, m), 1.37 (3H, t, J = 7.1 Hz). ¹³C NMR (75.4 MHz; CDCl₃) $\delta_{\rm C}$ 161.0 (C), 151.5 (C), 140.8 (C), 128.7 (2 × CH), 128.5 (2 × CH), 126.3 (CH), 83.2 (CH), 62.2 (CH2), 38.6 (CH2), 36.9 (CH2), 31.5 (CH2), 14.3 (CH3). HRMS (ESI⁺): calcd for $C_{14}H_{18}NO_3$ [M + H]⁺: 248.1281; found: 248.1278. HPLC analysis: Daicel Chiralpak® AD-H (heptane-iPrOH = 98:2, flow rate 1.0 mL min⁻¹, UV 254 nm, $t_{minor} = 15.0$ min for R enantiomer; $t_{\text{major}} = 19.9 \text{ min for } S \text{ enantiomer, } 74\% \text{ ee}$).

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