Palladium-Catalyzed Highly Chemo-, Regio- and Stereoselective Synthesis of Eight- to Ten-Membered Lactones from Allenyl 3-Oxoalkanoates and Organic Halides

Baoqiang Wan,^a Guochen Jia,^{b,*} and Shengming Ma^{a,c,*}

- ^a State Key Laboratory of Organometallic Chemistry, Shanghai Institute of Organic Chemistry, Chinese Academy of Sciences, 345 Linglin Lu, Shanghai 200032, People's Republic of China Fax: (+86)-21-6416-7510; e-mail: masm@sioc.ac.cn
- ^b Department of Chemistry, the Hong Kong University of Science and Technology, Clear Water Bay, Kowloon, Hong Kong, People's Republic of China Fax: (+852)-2358-1594; e-mail: chjiag@ust.hk
- ^c Shanghai Key Laboratory of Green Chemistry and Chemical Process, Department of Chemistry, East China Normal University, 3663 North Zhongshan Lu, Shanghai 200062, People's Republic of China

Received: January 28, 2011; Revised: April 14, 2011; Published online: June 30, 2011

Supporting information for this article is available on the WWW under http://dx.doi.org/10.1002/adcs.201100075.

Abstract: A highly chemo-, regio-, and stereoselective synthesis of eight- to ten-membered lactones *via* the coupling cyclization of readily available allenyl 3-oxoalkanoates and organic halides through an *anti*- π -

Introduction

Medium-sized lactones are extremely important compounds since they may be found in many natural products with biological potential,^[1] including octalactin B,^[2a] obtained from the surface of the Sea of Cortez gorgonian octocoral *Pacifigorgia* sp., and halicholactone,^[2b] a novel fatty acid metabolite from the marine sponge, and ascidiatrienolide A^[2c] (Figure 1). It is well-known that medium-sized ring compounds



Figure 1. Natural eight- to ten-membered lactones.

Adv. Synth. Catal. 2011, 353, 1763-1774

allylic palladium intermediate is reported. The yields ranged from moderate to good.

Keywords: allenes; π -allylic intermediates; carbopalladation; cyclization; lactones

are difficult to synthesize not only by biosynthesis but also by artificial synthetic technology due to entropic and enthalpic factors.^[3] Up to now, the most useful methodologies used to construct medium-sized lactones are lactonization^[1a,4] and ring-closing metathesis (RCM),^[5] usually requiring high diluted conditions, thus, the synthesis of medium-sized lactones is still a formidable challenge.

Recently, transition metal-catalyzed cyclization reactions of functionalized allenes in the presence of organic halides have become powerful tools for the synthesis of carbo- and heterocyclic compounds.^[6,7] Cyclic carbopalladation has been reported to afford 8- to 12-membered rings.^[8] An intermolecular carbopalladation-allylation protocol of alkenyl halides bearing a nucleophilic moiety afforded 8-membered rings with 62:38 to 92:8 stereoselectivity.^[9] In addition, Trost and co-workers reported the synthesis of 9- to 17-membered rings via hydropalladation of allenes under highly diluted conditions.^[10] On the basis of our previous work on the carbopalladation of allenes^[7] forming cyclic compounds with malonate or amine as the nucleophilic moiety,^[11] we envisioned that the carbopalladation of allenyl 3-oxoalkanoate 1a [from the reaction of readily available β -methylene- β -lactone

© 2011 Wiley-VCH Verlag GmbH & Co. KGaA, Weinheim	WILEY I	1763



Scheme 1. Synthesis of the starting materials 1a-1m.



Scheme 2. Possible products of the reaction of allenyl 3-oxoalkanoate 1a with iodobenzene.

and the allenic alcohol or ester exchange reaction of 3-oxoalkanoates with 3,4-pentadienol (Scheme 1)]^[12] may form six-, eight-, or even ten-membered lactones **4a**, **2a**, **5a**, and **3a**, respectively with the *C*-attack^[13] or *O*-attack^[14] of the π -allylic palladium intermediate due to the presence of the keto ester unit (Scheme 2). It has been reported that in the absence of the substituent at the 2-position *syn*- π -allylic palladium is favored.^[15] Herein, we wish to describe a highly efficient chemo-, regio-, and stereoselective protocol to construct eight- to ten-membered lactones *via* carbopalladation of readily available allenyl 3-oxoalkanoates with organic halides. The most interesting feature is the unique selectivity for the formation of the 8-

membered ring over the 6-membered ring with the exclusive *C*-attack.

Results and Discussion

When we heated allene **1a**, PhI (1.5 equiv.), and K_2CO_3 (2.0 equiv.) in the presence of Pd(PPh₃)₄ (5 mol%) in CH₃CN at 85 °C for 46 h (Table 1, entry 1), interestingly, the formation of the most favored six-membered *C*-attack lactone **4a** should be <2% by NMR, if any. In addition, it should be noted that the *O*-attack products **3a** and **5a** were also not observed. Quite unexpectedly, the more difficult-to-

Table 1. The effects of catalyst and temperature on the Pdcatalyzed coupling-cyclization of 1a with PhI.^[a]



Entry	Catalyst (mol%)	Т [°С]	Time [h]	Yield of 2a [%] ^[b]
1	$Pd(PPh_3)_4$ (5)	85	46	66
2	$Pd(OAc)_{2}$ (5), PPh_{3} (10)	85	6	81 (75) ^[c]
3	$Pd(dba)_{2}(5), PPh_{3}(15)$	85	6	64 ^[d]
4	$Pd(OAc)_{2}$ (5), PPh_{3} (10)	50	21.7	69 ^[e]

- ^[a] Under argon, a mixture of **1a** (0.250 mmol), PhI (0.375 mmol), K_2CO_3 (2 equiv.), the palladium catalyst (5 mol%), and ligand in 2 mL of CH₃CN was stirred at the indicated temperature.
- ^[b] The yield was determined by ¹H NMR analysis using 1,3,5-trimethylbenzene as the internal standard.
- ^[c] The number shown in the parenthesis is the isolated yield.
- ^[d] 5% of the starting material **1a** remained.
- ^[e] 14% of the starting material **1a** remained.

Table 2. Solvent and base effects in the Pd-catalyzed coupling-cyclization of 1a with PhI.^[a]



Entry	Base	Solvent	Time [h]	Recovery of 1a [%] ^[b]	Yield of 2a [%] ^[b]
1	K ₂ CO ₃	CH ₃ CN	6	0	81
2	K_2CO_3	toluene	26	25	42
3	K_2CO_3	THF	26	22	59
4	K_2CO_3	DCE	30.5	28	42
5	K_2CO_3	CH ₃ NO ₂	24	0	68
6	Na ₂ CO ₃	CH ₃ CN	34	55	27
7	Cs_2CO_3	CH ₃ CN	11	0	8
8	Et ₃ N	CH ₃ CN	46.5	39	16
9	KOH	CH ₃ CN	15.5	0	20
10	K_3PO_4	CH ₃ CN	15.5	0	28
11	CaH ₂	CH ₃ CN	45.5	16	trace

^[a] The reaction was carried out with **1a** (0.250 mmol), PhI (0.375 mmol), Pd(OAc)₂ (5 mol%), PPh₃ (10 mol%), and base (2 equiv.) in solvent (2 mL). THF=tetrahydrofuran, DCE=1,2-dichloroethane.

^[b] Determined by ¹H NMR analysis using 1,3,5-trimethylbenzene as the internal standard.

form C-attack eight-membered lactone **2a** was formed and isolated as the only product with a very high chemo- and regioselectivity.

A screening of palladium catalysts was then performed, and it was found that the reaction proceeded smoothly with $Pd(OAc)_2$ (5 mol%) and PPh_3 (10 mol%) within a short period of time affording **2a** in 75% isolated yield and the formation of **4a** should be <2.3% by NMR, if any. (Table 1, entry 2). Shortening the reaction time to 90 min, we only detected the *C*-attack product **2a** in 59% yield and 29% recovery of **1a** by NMR. The *O*-attack product was not observed. Lowering the temperature to 50°C, the reaction became sluggish with 14% of **1a** remaining.

A subsequent comprehensive study of the solvent effect indicated that the reaction can afford product **2a** in all the tested solvents (Table 2, entries 1–5) with CH₃CN (Table 1, entry 1) being the best. The base effect using CH₃CN as solvent was then investigated: The reaction afforded the desired product **2a** in relatively lower yields with Na₂CO₃, Cs₂CO₃, Et₃N, KOH, K₃PO₄, and CaH₂ (Table 2, entries 6–11). Thus, the following reaction conditions [organic halide (1.5 equiv.), Pd(OAc)₂ (5 mol%), PPh₃ (10 mol%), K₂CO₃ (2.0 equiv.), MeCN, 85 °C] were defined as the standard for the coupling cyclization.

With the optimal conditions in hand, firstly, the scope of functionalized allenes was investigated. Changing the R^1 group, the reaction proceeded to afford the desired lactones in good yields (Table 3, entries 2 and 3). Notably, the reaction of more sterically hindered allenyl 2-substituted-3-oxoalkanoates also afforded the corresponding lactones in good to moderate yields, in which a quaternary carbon was constructed. It is interesting to observe that the R^2 substituent has a dramatic effect on the reaction time. From methyl to Bn to allyl to ethyl or propyl, the reaction requires a much longer time (Table 3, entries 4–7).

Then the standard conditions were applied to investigate the reaction of different organic halides with **1c**. The results are summarized in Table 4. Electronrich (Table 4, entries 1–5), electron-deficient (Table 4, entry 6), Br-substituted (Table 4, entry 7) aryl iodides and 3-thienyl iodide (Table 4, entry 8) all afforded the corresponding eight-membered lactones in moderate to good yields. In addition, 1-hexenyl iodide (Table 4, entry 9) and styryl iodide (Table 4, entry 10) may also be used. The structures of all the eight-membered lactones were assigned by analogy to that of **2m**, which was determined by a single crystal X-ray diffraction study (Figure 2).^[16] **Table 3.** Substrate variation of compounds 1.^[a]



Entry	1		Time [h]	Yield of 2 [%] ^[b]
	\mathbf{R}^1	\mathbf{R}^2		
1	Me	Н (1а)	4.5	73 (2a)
2	<i>i</i> -Pr	H (1b)	6	83 (2b)
3	Ph	H (1c)	1.5	79 (2 c)
4	Me	Me (1d)	10.5	81 (2d) ^[c]
4	Me	Bn (1 j)	22	$75 (2w)^{[c]}$
5	Me	Et (1e)	81.5	$63 (2e)^{[c,d]}$
6	Me	<i>n</i> -Pr (1i)	82	$67 (2v)^{[c,d]}$
7	Me	$CH_2 = CHCH_2$ (1f)	29	69 $(2f)^{[c]}$

^[a] The reaction was carried out with **1** (0.50 mmol), PhI (0.75 mmol), Pd(OAc)₂ (5 mol%), PPh₃ (10 mol%), and K_2CO_3 (2 equiv.) in 4 mL of CH₃CN.

^[b] Isolated yield.

^[c] The reaction was carried out at 100 °C.

^[d] 2.5 equiv. of PhI were used.

The efficiency by which eight-membered lactones formed prompted us to explore the possibility of synthesizing other lactones. To our delight, this set of standard conditions has been proven to be operative for the efficient synthesis of nine- and ten-membered lactones with highly chemo-, regio- and stereoselectivity (Scheme 3). The reaction of **1g** with 1-iodo-4-nitrobenzene gave the expected product *E*-**2s** in a satisfactory yield of 74%. The structure of *E*-**2s** was further established by single crystal X-ray diffraction study (Figure 3).^[17] In addition, the ten-membered lactone *E*-**2u**^[18] was obtained in 69% yield as a single stereoisomer and the configuration of C=C bond in this compound was established by single crystal X-ray dif
 Table 4. Coupling cyclization of 1c with different organic halides.^[a]



Entry	R	Time [h]	Yield of 2 [%] ^[b]
1	$4-MeC_6H_4$	2	75 (2 g)
2	$4 - MeOC_6H_4$	5	69 (2h)
3	$3-\text{MeC}_6\text{H}_4$	2	78 (2i)
4	$3,4-Me_2C_6H_3$	1.6	72 (2j)
5	$3,4-(OCH_2)_2C_6H_3$	2.5	66 (2k)
6	$4-EtO_2CC_6H_4$	3	58 (2I)
7	$4-BrC_6H_4$	3.5	63 (2m)
8	3-thienyl	2	70 (2n)
9	1-(E)-hexenyl	2.5	59 (2o)
10	(E)-styryl	1.5	58 (2p)

^[a] The reaction was carried out with **1c** (0.50 mmol), RI (0.75 mmol), Pd(OAc)₂ (5 mol%), PPh₃ (10 mol%), and K_2CO_3 (2 equiv.) in 4 mL of CH₃CN.

^[b] Isolated yield.



Figure 2. ORTEP representation of 2m.





1766 asc.wiley-vch.de

© 2011 Wiley-VCH Verlag GmbH & Co. KGaA, Weinheim

Adv. Synth. Catal. 2011, 353, 1763-1774



Figure 3. ORTEP representations of *E*-2s and *E*-2u.

fraction study (Figure 3). The reaction of 1g with iodobenzene afforded the product 2q on a one-gram scale in 82% yield.

In addition, we investigated some other allenyl 3oxoalkanoates. To our delight, the reaction of **1k** with 1-iodo-4-nitrobenzene gave lactone **2x** in 65% yield and >8:1 dr (Scheme 4).^[19] The structure of *trans*-**2x** was further established by a single crystal X-ray diffraction study (Figure 4). However, the reactions of **11** and **1m** bearing substitutents at the allene moiety with iodobenzene failed to afford the expected corresponding eight-membered lactones (Scheme 4).

A rationale is proposed to explain the high regioand stereoselectivity using the reaction of **1a** and iodobenzene as the example.^[11] As illustrated in Scheme 5, obviously, the existence of the phenyl group strongly favors the formation of the π -allylic palladium intermediate *anti*-**6** rather than *syn*-**6** to avoid the steric congestion between the phenyl group and the chain bearing the nucleophilic functionality.^[15,20] Subsequent highly chemoselective carbon nucleophilic attack^[13,14] to the π -allylic species at the less substituted terminal affords the stereodefined eightmembered lactone **2a**.

Conclusions

In conclusion, we have developed a highly chemo-, regio-, and stereoselective methodology for the syn-



Scheme 4. Coupling cyclization of some other substituted allenyl 3-oxoalkanoates.

Adv. Synth. Catal. 2011, 353, 1763-1774

© 2011 Wiley-VCH Verlag GmbH & Co. KGaA, Weinheim



Figure 4. ORTEP representation of 2x.

thesis of normally difficult-to-form eight- to ten-membered lactones by palladium-catalyzed carbon-carbon bond formation *via* the coupling cyclization of allenyl alkanoates with organic halides. Owing to the readily available nature of β -methylene- β -lactone,^[12] the allenols^[21] and organic halides (Scheme 1) and the importance of the lactone skeleton, the reaction will be potentially useful in organic synthesis and medicinal chemistry. In addition, the observed exclusive *C*attack is also quite interesting. Further studies in this area are on going in our laboratory.

Experimental Section

General Information

All reactions were carried out in oven-dried Schlenk tubes. CH_3CN was dried over calcium hydride before distillation. All the temperatures are referred to the oil baths used. The petroleum ether was distilled before use.

Typical Procedure for the Preparation of 2

To a flame-dried Schlenk tube were added $Pd(OAc)_2$ (5.7 mg, 0.025 mmol), PPh₃ (13.3 mg, 0.050 mmol), K₂CO₃ (138.3 mg, 1.0 mmol), CH_3CN (2 mL), iodobenzene (153.6 mg, 0.75 mmol), CH₃CN (1 mL), **1a** (83.2 mg, 0.50 mmol), and 1 mL of CH₃CN sequentially under argon. The mixture was stirred at 85°C in a preheated oil bath and monitored by TLC. Upon completion, the resulting mixture was filtered through a short column of silica gel and concentrated. The residue was purified by chromatography on silica gel [eluent: petroleum ether (b.p. 30-60 °C)/ethyl ether = 10/1] to afford 3-acetyl-5-phenyl-5(E)-3,4,7,8-tetrahydro-2H-oxocin-2-one (2a) as an oil; yield: 88.6 mg (73%). ¹H NMR (300 MHz, CDCl₃): $\delta = 7.39 - 7.18$ (m, 5H, Ar-H), 5.95 (t, J = 8.0 Hz, 1H, =CH), 4.54 (t, J = 11.3 Hz, 1H, one proton from OCH₂), 4.25-4.10 (m, 1H, one proton from OCH₂), 3.55 (dd, J₁=11.7 Hz, J₂=3.9 Hz, 1 H, O₂CCH), 3.38 [t, J=12.3 Hz, 1 H, one proton from C=C(Ph)CH₂], 2.91– 2.68 [m, 2H, one proton from $C=C(Ph)CH_2$ and one proton from PhC=CCH₂], 2.34–2.12 (m, 4H, one proton from PhC= CCH₂ and CH₃); ¹³C NMR (75.4 MHz, CDCl₃): $\delta = 200.7$, 173.0, 142.5, 142.1, 128.3, 127.4, 126.4, 126.3, 68.7, 63.4, 30.8, 29.8, 28.7; IR (neat): $\nu = 2958$, 1750, 1718, 1491, 1460, 1363, 1317, 1257, 1205, 1145, 1103, 1046, 1028, 1004 cm⁻¹; MS (70 eV, EI): m/z (%)=244 (M⁺, 10.87), 43 (100); HR-MS: m/z = 244.1098, calcd. for C₁₅H₁₆O₃ (M⁺): 244.1099.

Synthesis of 3-Isobutyryl-5-phenyl-5(*E*)-3,4,7,8-tetrahydro-2*H*-oxocin-2-one (2b)

The reaction of $Pd(OAc)_2$ (5.4 mg, 0.025 mmol), PPh_3 (13.3 mg, 0.050 mmol), K_2CO_3 (138.1 mg, 1.0 mmol), iodobenzene (152.9 mg, 0.75 mmol), and **1b** (97.6 mg, 0.50 mmol)



Scheme 5. Rational explanation for the regio- and stereoselectivity.

1768 asc.wiley-vch.de

© 2011 Wiley-VCH Verlag GmbH & Co. KGaA, Weinheim

in 4 mL of CH₃CN afforded 2b (eluent: petroleum ether/ ethyl ether = 10/1) as a liquid; yield: 112.5 mg (83%). ¹H NMR (300 MHz, CDCl₃): $\delta = 7.37 - 7.21$ (m, 5H, Ar-H), 5.95 (t, J = 8.0 Hz, 1H, =CH), 4.57 (t, J = 11.3 Hz, 1H, one proton from OCH₂), 4.21-4.09 (m, 1H, one proton from OCH_2), 3.70 (dd, $J_1 = 11.7$ Hz, $J_2 = 4.2$ Hz, 1 H, O_2CCH), 3.45 [t, J=12.5 Hz, 1H, one proton from C=C(Ph)CH₂], 2.90-2.65 [m, 3H, one proton from C=C(Ph)CH₂, one proton from PhC=CCH₂, and COCH(C)C], 2.30–2.17 (m, 1H, one proton from PhC=CCH₂), 1.09 [t, J=6.5 Hz, 6 H, C(CH₃)₂]; ¹³C NMR (75.4 MHz, CDCl₃): $\delta = 206.8$, 173.1, 142.7, 142.3, 128.4, 127.5, 126.5, 126.2, 68.7, 61.2, 40.3, 31.1, 29.9, 18.8, 17.9; IR (neat): v = 2928, 1744, 1705, 1497, 1456, 1379, 1366, 1261, 1206, 1151, 1105, 1068, 1037, 1015 cm⁻¹; MS (70 eV, EI): m/z (%)=272 (M⁺, 3.52), 43 (100); HR-MS: m/z= 272.1413, calcd. for C₁₇H₂₀O₃ (M⁺): 272.1412.

Synthesis of 3-Benzoyl-5-phenyl-5(*E*)-3,4,7,8-tetrahydro-2*H*-oxocin-2-one (2c)

The reaction of Pd(OAc)₂ (5.5 mg, 0.025 mmol), PPh₃ (13.4 mg, 0.050 mmol), K₂CO₃ (137.7 mg, 1.0 mmol), iodobenzene (152.0 mg, 0.75 mmol), and **1c** (114.9 mg, 0.50 mmol) in 4 mL of CH₃CN afforded 2c (eluent: petroleum ether/ethyl ether = 10/1) as an oil; yield: 120.5 mg (79%). ¹H NMR (300 MHz, CDCl₃): $\delta = 8.00$ (d, J = 7.8 Hz, 2H, Ar-H), 7.53 (t, J=7.1 Hz, 1H, Ar-H), 7.40 (t, J=7.7 Hz, 2H, Ar-H), 7.36–7.20 (m, 5H, Ar-H), 6.01 (t, J=8.1 Hz, 1H, =CH), 4.58 (t, J=11.4 Hz, 1H, one proton from OCH₂), 4.36 (dd, $J_1 = 11.6$ Hz, $J_2 = 4.1$ Hz, 1H, O₂CCH), 4.15–4.03 (m, 1H, one proton from OCH₂), 3.75 [t, J = 12.5 Hz, 1H, one proton from C=C(Ph)CH₂], 2.95-2.67 [m, 2H, one proton from C=C(Ph)CH₂ and one proton from PhC= CCH_2 , 2.30–2.17 (m, 1H, one proton from PhC=CCH₂); ¹³C NMR (75.4 MHz, CDCl₃): $\delta = 192.1$, 173.1, 142.6, 142.4, 135.7, 133.7, 128.7, 128.6, 128.5, 127.5, 126.5, 126.4, 68.7, 59.1, 31.4, 28.9; IR (neat): $\nu = 1743$, 1684, 1597, 1580, 1491, 1448, 1365, 1326, 1279, 1212, 1194, 1145, 1103, 1076, 1045, 1014 cm⁻¹; MS (70 eV, EI): m/z (%) = 306 (M⁺, 0.4), 105 (100); anal. calcd. for C₂₀H₁₈O₃ (%): C 78.41, H 5.92; found: C 78.41, H 5.97.

Synthesis of 3-Acetyl-3-methyl-5-phenyl-5(*E*)-3,4,7,8-tetrahydro-2*H*-oxocin-2-one (2d)

The reaction of $Pd(OAc)_2$ (5.7 mg, 0.025 mmol), PPh_3 (13.4 mg, 0.050 mmol), K₂CO₃ (138.5 mg, 1.0 mmol), iodobenzene (152.9 mg, 0.75 mmol), and 1d (90.1 mg, 0.50 mmol) in 4 mL of CH₃CN afforded 2d [eluent: petroleum ether (b.p. 30-60 °C)/ethyl ether = 10/1] as an oil; yield: 103.4 mg (81%). ¹H NMR (300 MHz, CDCl₃): $\delta = 7.36-7.19$ (m, 5H, Ar-H), 5.93 (t, J=8.0 Hz, 1H, =CH), 4.62 (t, J=11.4 Hz, 1H, one proton from OCH₂), 4.12-3.98 (m, 1H, one proton from OCH₂), 3.52 [d, J=13.2 Hz, 1H, C=C(Ph)CH₂], 2.82-2.65 [m, 2H, one proton from $C=C(Ph)CH_2$ and one proton from PhC=CCH₂], 2.30–2.19 (m, 1 H, one proton from PhC= CCH₂), 2.16 (s, 3H, COCH₃), 0.96 (s, 3H, CCH₃); ¹³C NMR $(75.4 \text{ MHz}, \text{ CDCl}_3): \delta = 205.0, 176.0, 144.8, 143.0, 128.4,$ 127.5, 127.2, 126.3, 67.4, 65.5, 37.1, 30.0, 26.0, 18.5; IR (neat): $\nu = 1748$, 1708, 1488, 1451, 1380, 1361, 1319, 1295, 1266, 1215, 1162, 1101, 1088 cm⁻¹; MS (70 eV, EI): m/z (%)=258 (M⁺, 4.18), 43 (100); HR-MS: m/z=258.1255, calcd. for C₁₆H₁₈O₃ (M⁺): 258.1256.

Synthesis of 3-Acetyl-3-ethyl-5-phenyl-5(*E*)-3,4,7,8-tetrahydro-2*H*-oxocin-2-one (2e)

The reaction of $Pd(OAc)_2$ (5.5 mg, 0.025 mmol), PPh_3 (13.2 mg, 0.050 mmol), K₂CO₃ (138.4 mg, 1.0 mmol), iodobenzene (255.6 mg, 1.25 mmol), and 1e (98.5 mg, 0.50 mmol) in 4 mL of CH₃CN afforded 2e (eluent: petroleum ether/ ethyl acetate = 20/1) as a white solid; yield: 86.2 mg (63%); mp 124-125°C (ethyl ether/petroleum ether). ¹H NMR (400 MHz, CDCl₃): $\delta = 7.35 - 7.18$ (m, 5 H, Ar-H), 5.90 (t, J =8.0 Hz, 1 H, =CH), 4.63 (t, J=11.4 Hz, 1 H, one proton from OCH₂), 4.12–3.97 (m, 1H, one proton from OCH₂), 3.37 [d, J = 13.2 Hz, 1 H, one proton from C=C(Ph)CH₂], 2.88 [d, J =13.6 Hz, 1 H, one proton from C=C(Ph)CH₂], 2.81–2.67 (m, 1H, one proton from PhC=CCH₂), 2.29-2.17 (m, 1H, one proton from PhC=CCH2), 2.12 (s, 3H, COCH3), 1.83-1.67 (m, 1H, one proton from O₂CCCH₂C), 1.39-1.25 (m, 1H, one proton from O₂CCCH₂C), 0.48 (t, J=7.6 Hz, 3H, CCH₃); ¹³C NMR (100.5 MHz, CDCl₃): δ =204.4, 175.7, 144.5, 143.1, 128.3, 127.7, 127.1, 126.3, 69.9, 67.3, 32.8, 29.9, 26.7, 23.3, 7.75; IR (neat): v = 1750, 1706, 1491, 1446, 1365, 1352, 1331, 1280, 1257, 1238, 1215, 1196, 1157, 1112, 1087, 1022 cm^{-1} ; MS (70 eV, EI): m/z (%)=272 (M⁺, 1.56), 43 (100); anal. calcd. for $C_{17}H_{20}O_3$ (%): C 74.97, H 7.40; found: C 74.91, H 7.24.

Synthesis of 3-Acetyl-3-allyl-5-phenyl-5(*E*)-3,4,7,8-tetrahydro-2*H*-oxocin-2-one (2f)

The reaction of Pd(OAc)₂ (5.7 mg, 0.025 mmol), PPh₃ (13.1 mg, 0.050 mmol), K₂CO₃ (138.7 mg, 1.0 mmol), iodobenzene (153.5 mg, 0.75 mmol), and **1f** (104.8 mg, 0.50 mmol) in 4 mL of CH₃CN afforded 2f [eluent: petroleum ether (bp 30-60 °C)/ethyl acetate = 20/1] as a white solid; yield: 99.2 mg (69%); mp 77-78°C (ethyl ether/petroleum ether). ¹H NMR (400 MHz, CDCl₃): $\delta = 7.33-7.14$ (m, 5H, Ar-H), 5.90 (t, J=8.0 Hz, 1H, C=CH), 5.40-5.25 (m, 1H, C=CHC), 4.87 (d, J = 10.0 Hz, 1H, one proton from CH₂= C), 4.65 (t, J=11.4 Hz, 1 H, one proton from OCH₂), 4.49 (d, J = 17.2 Hz, 1 H, one proton from CH₂=C), 4.10-4.00 (m, 1H, one proton from OCH₂), 3.40 [d, J = 13.2 Hz, 1H, one proton from C=C(Ph)CH₂], 2.83 [d, J=13.2 Hz, 1H, one proton from C=C(Ph)CH₂], 2.79-2.67 (m, 1H, one proton from PhC=CCH₂), 2.46 (dd, J₁=14.4 Hz, J₂=6.4 Hz, 1H, one proton from CH₂C=C), 2.30-2.19 (m, 1H, one proton from PhC=CCH₂), 2.13 (s, 3H, CH₃), 2.46 (dd, J_1 = 14.8 Hz, $J_2 = 8.2$ Hz, 1 H, one proton from CH₂C=C); ¹³C NMR (100.5 MHz, CDCl₃): $\delta = 203.7$, 175.2, 144.6, 143.2, 131.2, 128.3, 128.0, 127.2, 126.6, 119.2, 69.4, 67.4, 35.1, 33.7, 30.0, 27.1; IR (neat): $\nu = 1748$, 1709, 1490, 1465, 1443, 1366, 1272, 1254, 1231, 1187, 1160, 1129, 1110, 1069, 1012 cm^{-1} ; MS (70 eV, EI): m/z (%)=284 (M⁺, 1.44), 43 (100); anal. calcd. for C₁₈H₂₀O₃ (%): C 76.03, H 7.09; found: C 76.03, H 7.02.

Synthesis of 3-Benzoyl-5-*p*-tolyl-5(*E*)-3,4,7,8-tetrahydro-2*H*-oxocin-2-one (2g)

The reaction of $Pd(OAc)_2$ (5.8 mg, 0.025 mmol), PPh_3 (13.3 mg, 0.050 mmol), K_2CO_3 (138.0 mg, 1.0 mmol), *p*-meth-

ylphenyl iodide (164.1 mg, 0.75 mmol), and 1c (115.0 mg, 0.50 mmol) in 4 mL of CH₃CN afforded 2g (eluent: petroleum ether/ethyl ether = 10/1) as a white solid; yield: 120.1 mg (75%); mp 110-111 °C (ethyl ether/petroleum ether). ¹H NMR (300 MHz, CDCl₃): $\delta = 8.01$ (d, J = 7.5 Hz, 2H, Ar-H), 7.55 (t, J = 7.4 Hz, 1H, Ar-H), 7.43 (t, J = 7.7 Hz, 2H, Ar-H), 7.24 (d, J=7.5 Hz, 2H, Ar-H), 7.15 (d, J=8.1 Hz, 2H, Ar-H), 6.00 (t, J=8.1 Hz, 1H, =CH), 4.59 (t, J= 11.4 Hz, 1 H, one proton from OCH₂), 4.36 (dd, $J_1 = 11.7$ Hz, $J_2 = 4.2$ Hz, 1H, O₂CCH), 4.15–4.05 (m, 1H, one proton from OCH₂), 3.73 [t, J=12.5 Hz, 1H, one proton from C= C(Ar)CH₂], 2.95–2.77 [m, 2H, one proton from C= C(Ar)CH₂ and one proton from ArC=CCH₂], 2.35 (s, 3H, CH_3), 2.30–2.17 (m, 1H, one proton from $ArC=CCH_2$); ¹³C NMR (75.4 MHz, CDCl₃): $\delta = 192.1$, 173.0, 142.4, 139.4, 137.3, 135.7, 133.7, 129.1, 128.7, 128.6, 126.3, 125.6, 68.8, 59.1, 31.3, 29.9, 21.0; IR (neat): $\nu = 1738$, 1683, 1596, 1514, 1465, 1447, 1360, 1324, 1278, 1254, 1205, 1192, 1177, 1144, 1057, 1014 cm⁻¹; MS (70 eV, EI): m/z (%)=320 (M⁺, 2.57), 105 (100); anal. calcd. for $C_{21}H_{20}O_3$ (%): C 78.73, H 6.29; found: C 78.44, H 6.29.

Synthesis of 3-Benzoyl-5-(4-methoxyphenyl)-5(*E*)-3,4,7,8-tetrahydro-2*H*-oxocin-2-one (2h)

The reaction of Pd(OAc)₂ (5.5 mg, 0.025 mmol), PPh₃ (13.4 mg, 0.050 mmol), K₂CO₃ (138.5 mg, 1.0 mmol), p-methoxyphenyl iodide (176.1 mg, 0.75 mmol), and 1c (115.3 mg, 0.50 mmol) in 4 mL of CH₃CN afforded 2h [eluent: petroleum ether (b.p. 30-60 °C)/ethyl ether = 10/1] as a white solid; yield 116.8 mg (69%); mp 132-133 °C (ethyl ether/petroleum ether). ¹H NMR (300 MHz, CDCl₃): $\delta = 8.00$ (d, J = 7.8 Hz, 2H, Ar-H), 7.54 (t, J=7.4 Hz, 1H, Ar-H), 7.42 (t, J=7.5 Hz, 2H, Ar-H), 7.28 (d, J=8.7 Hz, 2H, Ar-H), 6.87 (d, J=8.7 Hz, 2H, Ar-H), 5.96 (t, J=8.0 Hz, 1H, =CH), 4.58 (t, J= 11.4 Hz, 1 H, one proton from OCH₂), 4.35 (dd, $J_1 = 11.6$ Hz, $J_2 = 4.1 \text{ Hz}, 1 \text{ H}, O_2 \text{CCH}), 4.15-4.05 \text{ (m, 1 H, one proton)}$ from OCH₂), 3.80 (s, 3H, OMe), 3.72 [t, J=12.3 Hz, 1H, one proton from C=C(Ar)CH₂], 2.95-2.77 [m, 2H, one proton from C=C(Ar)CH₂ and one proton from ArC= CCH_2], 2.30–2.15 (m, 1H, ArC= CCH_2); ¹³C NMR $(75.4 \text{ MHz}, \text{ CDCl}_3): \delta = 192.2, 173.1, 159.2, 142.0, 135.7,$ 134.7, 133.7, 128.70, 128.65, 127.6, 125.0, 113.8, 69.0, 59.2, 55.2, 31.3, 29.0; IR (neat): $\nu = 1742$, 1684, 1600, 1575, 1513, 1462, 1426, 1330, 1293, 1275, 1238, 1205, 1190, 1139, 1057, 1028, 1012 cm⁻¹; MS (70 eV, EI): m/z (%) = 336 (M⁺, 10.39), 77 (100); anal. calcd. for $C_{21}H_{20}O_4$ (%): C 74.98, H 5.99; found: C 75.02, H 6.04.

Synthesis of 3-Benzoyl-5-(*m*-tolyl)-5(*E*)-3,4,7,8-tetrahydro-2*H*-oxocin-2-one (2i)

The reaction of Pd(OAc)₂ (5.5 mg, 0.025 mmol), PPh₃ (13.3 mg, 0.050 mmol), K₂CO₃ (138.4 mg, 1.0 mmol), *m*-methylphenyl iodide (163.0 mg, 0.75 mmol), and **1c** (115.6 mg, 0.50 mmol) in 4 mL of CH₃CN afforded **2i** [eluent: petroleum ether (bp 30–60 °C)/ethyl ether = 10/1] as an oil; yield: 124.8 mg (78%). ¹H NMR (300 MHz, CDCl₃): δ =8.00 (d, *J*=7.8 Hz, 2 H, Ar-H), 7.52 (t, *J*=7.4 Hz, 1 H, Ar-H), 7.40 (t, *J*=7.7 Hz, 2 H, Ar-H), 7.22 (t, *J*=7.4 Hz, 1 H, Ar-H), 7.17–7.05 (m, 3 H, Ar-H), 5.99 (t, *J*=8.0 Hz, 1 H, = CH), 4.57 (t, *J*=11.4 Hz, 1 H, one proton from OCH₂), 4.35

(dd, J_1 =11.7 Hz, J_2 =4.2 Hz, 1H, O₂CCH), 4.15–4.00 (m, 1H, one proton from OCH₂), 3.73 [t, J=12.5 Hz, 1H, one proton from C=C(Ar)CH₂], 2.92–2.75 [m, 2H, one proton from C=C(Ar)CH₂ and one proton from ArC=CCH₂], 2.35 (s, 3H, CH₃), 2.28–2.14 (m, 1H, one proton from ArC=CCH₂); ¹³C NMR (75.4 MHz, CDCl₃): δ =192.0, 173.0, 142.6, 142.3, 137.9, 135.6, 133.6, 128.59, 128.55, 128.3, 128.2, 127.2, 126.1, 123.5, 68.6, 59.1, 31.3, 29.8, 21.3; IR (neat): ν = 1742, 1684, 1597, 1580, 1486, 1448, 1363, 1325, 1282, 1249, 1212, 1182, 1141, 1103, 1047, 1017 cm⁻¹; MS (70 eV, EI): m/z (%)=320 (M⁺, 0.74), 105 (100); anal. calcd. for C₂₁H₂₀O₃ (%): C 78.73, H 6.29; found: C 78.94, H 6.64.

Synthesis of 3-Benzoyl-5-(3,4-dimethylphenyl)-5(*E*)-3,4,7,8-tetrahydro-2*H*-oxocin-2-one (2j)

The reaction of Pd(OAc)₂ (5.7 mg, 0.025 mmol), PPh₃ (13.3 mg, 0.050 mmol), K₂CO₃ (138.5 mg, 1.0 mmol), 3,4-dimethylphenyl iodide (174.5 mg, 0.75 mmol), and 1c (113.9 mg, 0.50 mmol) in 4 mL of CH₃CN afforded 2j [eluent: petroleum ether (b.p. 30-60 °C)/ethyl ether = 10/1] as an oil; yield: 119.2 mg (72%). ¹H NMR (300 MHz, CDCl₃): $\delta = 8.01$ (d, J = 7.8 Hz, 2H, Ar-H), 7.54 (t, J =7.2 Hz, 1H, Ar-H), 7.42 (t, J=7.7 Hz, 2H, Ar-H), 7.16–7.04 (m, 3H, Ar-H), 6.00 (t, J = 8.0 Hz, 1H, =CH), 4.58 (t, J =11.4 Hz, 1 H, one proton from OCH_2), 4.36 (dd, $J_1 = 11.6$ Hz, $J_2 = 4.1$ Hz, 1H, O₂CCH), 4.15–4.05 (m, 1H, one proton from OCH₂), 3.72 [t, J=12.3 Hz, 1H, one proton from C= C(Ar)CH₂], 2.95–2.78 [m, 2H, one proton from C= C(Ar)CH₂ and one proton from ArC=CCH₂], 2.35–2.17 (m, 7 H, one proton from $ArC=CCH_2$ and $2 \times Ar-CH_3$; ¹³C NMR (75.4 MHz, CDCl₃): $\delta = 192.3$, 173.1, 142.5, 139.9, 136.6, 136.0, 135.8, 133.7, 129.7, 128.7, 127.8, 125.5, 123.9, 68.9, 59.2, 31.4, 29.9, 19.8, 19.3; IR (neat): $\nu = 1743$, 1685, 1598, 1580, 1501, 1448, 1362, 1326, 1278, 1248, 1213, 1195, 1176, 1143, 1127, 1100, 1048, 1018 cm⁻¹; MS (70 eV, EI); m/ z (%)=334 (M⁺, 3.22), 105 (100); HR-MS: m/z=334.1567, calcd. for C₂₂H₂₂O₃ (M⁺): 334.1569.

Synthesis of 3-Benzoyl-5-(2,3 dihydrobenzo[b][1,4]dioxin-6-yl)-5(E)-3,4,7,8-tetrahydro-2H-oxocin-2-one (2k)

The reaction of $Pd(OAc)_2$ (5.5 mg, 0.025 mmol), PPh_3 (13.4 mg, 0.050 mmol), K₂CO₃ (138.9 mg, 1.0 mmol), 6-iodo-1,4-benzodioxane (196.5 mg, 0.75 mmol), and 1c (116.0 mg, 0.50 mmol) in 4 mL of CH₃CN afforded **2k** [eluent: petroleum ether (bp 30-60 °C)/ethyl ether=10/1] as a white solid; yield: 121.3 mg (66%); mp 142-143 °C (ethyl ether/petroleum ether). ¹H NMR (300 MHz, CDCl₃): $\delta = 8.02$ (d, J =7.8 Hz, 2H, Ar-H), 7.55 (t, J=7.4 Hz, 1H, Ar-H), 7.43 (t, J = 7.4 Hz, 2H, Ar-H), 6.93–6.77 (m, 3H, Ar-H), 5.98 (t, J =8.1 Hz, 1 H, =CH), 4.56 (t, J=11.6 Hz, 1 H, one proton from OCH₂), 4.37 (dd, J₁=11.6 Hz, J₂=4.1 Hz, 1 H, O₂CCH), 4.25 (s, 4H, OCH₂CH₂O), 4.15-4.02 (m, 1H, one proton from OCH_2), 3.70 [t, J=12.5 Hz, 1 H, one proton from C= C(Ar)CH₂], 2.93–2.70 [m, 2H, one proton from C= C(Ar)CH₂ and one proton from ArC=CCH₂], 2.29–2.15 (m, 1H, one proton from ArC=CCH₂); ¹³C NMR (75.4 MHz, CDCl₃): $\delta = 192.2, 173.0, 143.3, 143.2, 141.7, 135.8, 135.7,$ 133.7, 128.71, 128.69, 125.2, 119.5, 117.2, 115.3, 68.9, 64.4, 64.3, 59.2, 31.2, 29.9; IR (neat): $\nu = 1742$, 1688, 1596, 1579,

1510, 1452, 1429, 1360, 1329, 1300, 1249, 1206, 1131, 1101, 1068, 1051, 1014 cm⁻¹; MS (70 eV, EI): m/z (%)=364 (M⁺, 8.38), 77 (100); anal. calcd. for C₂₂H₂₀O₅ (%): C 72.51, H 5.53; found: C 72.48, H 5.30.

Synthesis of 3-Benzoyl-5-(4-ethoxycarbonylphenyl)-5(*E*)-3,4,7,8-tetrahydro-2*H*-oxocin-2-one (21)

The reaction of Pd(OAc)₂ (5.7 mg, 0.025 mmol), PPh₃ (13.3 mg, 0.050 mmol), K₂CO₃ (138.7 mg, 1.0 mmol), ethyl 4iodobenzoate (206.9 mg, 0.75 mmol), and 1c (115.9 mg, 0.50 mmol) in 4 mL of CH₃CN afforded 2l [eluent: petroleum ether (bp 30-60 °C)/dichloromethane = 1/2] as a white solid; yield: 111.3 mg (58%); mp 123-124°C (ethyl ether/petroleum ether). ¹H NMR (300 MHz, CDCl₃): $\delta = 8.07-7.92$ (m, 4H, Ar-H), 7.56 (t, J=7.4 Hz, 1H, Ar-H), 7.48–7.32 (m, 4H, Ar-H), 6.12 (t, J=8.0 Hz, 1H, =CH), 4.60 (t, J=11.3 Hz, 1H, one proton from OCH₂CC), 4.45-4.25 (m, 3H, p-Ar-CO₂CH₂ and O₂CCH), 4.20-4.05 (m, 1H, one proton from OCH₂CC), 3.78 [t, J = 12.5 Hz, 1H, one proton from C=C(Ar)CH₂], 2.95-2.75 [m, 2H, one proton from C= C(Ar)CH₂ and one proton from ArC=CCH₂], 2.35-2.12 (m, 1 H, one proton from ArC=CCH₂), 1.39 (t, J = 7.2 Hz, 3 H, CH₃); ¹³C NMR (75.4 MHz, CDCl₃): $\delta = 191.9$, 173.0, 166.3, 146.8, 142.0, 135.7, 133.8, 129.8, 129.6, 128.8, 128.7, 128.2, 126.5, 68.4, 60.9, 59.1, 31.2, 30.0, 14.3; IR (neat): $\nu = 1743$, 1716, 1687, 1607, 1597, 1459, 1449, 1406, 1365, 1316, 1276, 1255, 1207, 1193, 1179, 1144, 1105, 1013 cm⁻¹; MS (70 ev, EI) m/z (%): 378 (M⁺, 0.16), 105 (100); anal. calcd. for C₂₃H₂₂O₅ (%): C 73.00, H 5.86; found: C 72.98, H 6.08.

Synthesis of 3-Benzoyl-5-(4-bromophenyl)-5(*E*)-3,4,7,8-tetrahydro-2*H*-oxocin-2-one (2m)

The reaction of Pd(OAc)₂ (5.4 mg, 0.025 mmol), PPh₃ (13.5 mg, 0.050 mmol), K₂CO₃ (137.4 mg, 1.0 mmol), 1bromo-4-iodobenzene (211.7 mg, 0.75 mmol), and 1c (113.8 mg, 0.50 mmol) in 4 mL of CH₃CN afforded 2m [eluent: petroleum ether (bp 30-60 °C)/ethyl ether = 15/1] as a solid; yield: 120.8 mg (63%); mp 136–137 °C (ethyl ether/ petroleum ether). ¹H NMR (300 MHz, CDCl₃): $\delta = 8.01$ (d, J = 7.5 Hz, 2H, Ar-H), 7.56 (t, J = 7.4 Hz, 1H, Ar-H), 7.52– 7.37 (m, 4H, Ar-H), 7.21 (d, J=8.7 Hz, 2H, Ar-H), 6.02 (t, J = 8.0 Hz, 1H, =CH), 4.59 (t, J = 11.4 Hz, 1H, one proton from OCH₂), 4.33 (dd, $J_1 = 11.6$ Hz, $J_2 = 4.4$ Hz, 1 H, O₂CCH), 4.15–4.05 (m, 1H, one proton from OCH₂), 3.75 [t, J = 12.5 Hz, 1H, one proton from C=C(Ar)CH₂], 2.95–2.75 [m, 2H, one proton from $C=C(Ar)CH_2$ and one proton from ArC=CCH₂], 2.32-2.13 (m, 1H, one proton from ArC=CCH₂); ¹³C NMR (75.4 MHz, CDCl₃): δ =191.9, 173.0, 141.7, 141.3, 135.6, 133.8, 131.6, 128.8, 128.6, 128.2, 127.1, 121.6, 68.5, 59.0, 31.3, 29.9; IR (neat): $\nu = 1748$, 1681, 1596, 1580, 1487, 1446, 1424, 1394, 1364, 1324, 1310, 1275, 1253, 1208, 1194, 1177, 1145, 1108, 1074, 1057, 1010 cm⁻¹; MS (70 eV, EI): m/z (%)=386 [M(⁸¹Br)⁺, 0.53], 384 [M(⁷⁹Br)⁺, 0.56], 105 (100); anal. calcd. for C₂₀H₁₇BrO₃ (%): C 62.35, H 4.45; found: C 62.15, H 4.61.

Synthesis of 3-Benzoyl-5-(3-thienyl)-5(*E*)-3,4,7,8-tetrahydro-2*H*-oxocin-2-one (2n)

The reaction of $Pd(OAc)_2$ (5.7 mg, 0.025 mmol), PPh_3 (13.3 mg, 0.050 mmol), K_2CO_3 (138.1 mg, 1.0 mmol), 3-iodo-

thiophene (156.5 mg, 0.75 mmol), and 1c (114.5 mg, 0.50 mmol) in 4 mL of CH₃CN afforded 2n [eluent: petroleum ether (bp 30-60 °C)/ethyl ether = 10/1] as a white solid; yield: 108.1 mg (70%); mp 120-122 °C (ethyl ether/petroleum ether). ¹H NMR (300 MHz, CDCl₃): $\delta = 8.04$ (d, J =7.8 Hz, 2H, Ar-H), 7.56 (t, J=7.4 Hz, 1H, Ar-H), 7.44 (t, J=7.7 Hz, 2H, Ar-H), 7.35–7.23 (m, 1H, Ar-H), 7.22–7.12 (m, 2H, Ar-H), 6.22 (t, J=8.1 Hz, 1H, =CH), 4.57 (t, J=11.4 Hz, 1 H, one proton from OCH₂), 4.48 (dd, $J_1 = 11.7$ Hz, $J_2 = 4.2$ Hz, 1H, O_2 CCH), 4.20–4.05 (m, 1H, one proton from OCH₂), 3.71 [t, J=12.5 Hz, 1H, one proton from C= C(Ar)CH₂], 2.95-2.75 [m, 2H, one proton from C= C(Ar)CH₂ and one proton from ArC=CCH₂], 2.30-2.15 (m, 1 H, one proton from ArC=CCH₂); ¹³C NMR (75.4 MHz, $CDCl_3$): $\delta = 192.2, 172.8, 142.8, 136.9, 135.7, 133.8, 128.75,$ 128.67, 126.1, 125.5, 124.9, 120.2, 69.1, 59.2, 30.7, 29.7; IR (neat): $\nu = 1742$, 1685, 1598, 1579, 1461, 1448, 1424, 1358, 1324, 1286, 1268, 1249, 1204, 1174, 1133, 1102, 1040, 1013 cm⁻¹; MS (70 eV, EI): m/z (%)=312 (M⁺, 2.01), 77 (100); anal. calcd. for $C_{18}H_{16}O_3S$ (%): C 69.21, H 5.16, S 10.26; found: C 69.49, H 5.02, S 10.25.

Synthesis of 3-Benzoyl-5-[(*E*)-hex-1-enyl]-5(*E*)-3,4,7,8-tetrahydro-2*H*-oxocin-2-one (20)

The reaction of Pd(OAc)₂ (5.6 mg, 0.025 mmol), PPh₃ (13.2 mg, 0.050 mmol), K₂CO₃ (138.7 mg, 1.0 mmol), (E)-1iodohex-1-ene (157.3 mg, 0.75 mmol), and 1c (114.9 mg, 0.50 mmol) in 4 mL of CH₃CN afforded 20 [eluent: petroleum ether (b.p. 30-60 °C)/ethyl ether = 15/1] as a white solid; yield: 92.0 mg (59%); mp 70-71 °C (ethyl ether/petroleum ether). ¹H NMR (300 MHz, CDCl₃): $\delta = 8.07$ (d, J = 8.1 Hz, 2H, Ar-H), 7.58 (t, J=7.4 Hz, 1H, Ar-H), 7.47 (t, J=7.5 Hz, 2H, Ar-H), 6.04 (d, J=15.9 Hz, 1H, C=CCH=C), 5.75 [t, J = 8.1 Hz, 1 H, CH=C(C)C=C], 5.71-5.58 (m, 1 H, C=CC=CHC), 4.53 (t, J=11.4 Hz, 1H, one proton from OCH₂), 4.44 (dd, $J_1 = 11.4$ Hz, $J_2 = 3.9$ Hz, 1H, O₂CCH), 4.15–4.00 (m, 1H, one proton from OCH₂), 3.42 [t, J = 12.6 Hz, 1H, one proton from C=C(C=C)CH₂], 2.87-2.63 [m, 2H, one proton from C=C(C=C)CH₂ and one proton from CH₂C= CC=C], 2.25-2.01 (m, 3H, CH₂CC, and one proton from CH₂C=CC=C), 1.47-1.20 (m, 4H, CCH₂CH₂C), 0.90 (t, J= 6.8 Hz, 3H, CH₃); ¹³C NMR (75.4 MHz, CDCl₃): $\delta = 192.4$, 172.8, 140.2, 135.8, 133.7, 132.5, 129.5, 128.74, 128.65, 127.3, 69.8, 58.9, 32.4, 31.5, 29.6, 26.8, 22.2, 13.9; IR (neat): v =1744, 1686, 1597, 1581, 1453, 1422, 1380, 1353, 1325, 1284, 1264, 1248, 1206, 1188, 1173, 1140, 1120, 1056, 1010 cm^{-1} ; MS (70 eV, EI): m/z (%)=312 (M⁺, 0.18), 105 (100); anal. calcd. for C₂₀H₂₄O₃ (%): C 76.89, H 7.74; found: C 77.16, H 7.99.

Synthesis of 3-Benzoyl-5-styryl-5(*E*)-3,4,7,8-tetrahydro-2*H*-oxocin-2-one (2p)

The reaction of Pd(OAc)₂ (5.7 mg, 0.025 mmol), PPh₃ (13.3 mg, 0.050 mmol), K₂CO₃ (139.0 mg, 1.0 mmol), (*E*)-(2-iodovinyl)benzene (173.3 mg, 0.75 mmol), and **1c** (115.0 mg, 0.50 mmol) in 4 mL of CH₃CN afforded **2p** [eluent: petroleum ether (b.p. 30–60 °C)/ethyl ether = 15/1] as a white solid; yield: 95.8 mg (58%); mp 116–117 °C (ethyl ether/petroleum ether); ¹H NMR (300 MHz, CDCl₃): δ =8.09 (d, *J*=8.1 Hz, 2H, Ar-H), 7.58 (t, *J*=7.1 Hz, 1H, Ar-H), 7.52–7.37 (m, 4H,

Ar-H), 7.31 (t, J=7.4 Hz, 2H, Ar-H), 7.22 (t, J=7.2 Hz, 1H, Ar-H), 6.78 [d, J=16.2 Hz, 1H, C=CCH=C(Ph)], 6.51 [d, J = 16.2 Hz, 1H, C=CC=CCH(Ph)], 6.01 [t, J = 8.1 Hz, 1H, CH=CC=C(Ph)], 4.62-4.45 (m, 2H, one proton from OCH₂ and O₂CCH), 4.20–4.05 (m, 1H, one proton from OCH₂), 3.54 [t, J=12.8 Hz, 1H, one proton from C=C(C=C)CH₂], 2.95–2.75 [m, 2H, one proton from C=C(C=C)CH₂ and one proton from (C=C)C=CCH₂], 2.20 [dd, J_1 =13.8 Hz, J_2 = 8.4 Hz, 1 H, one proton from $(C=C)C=CCH_2$; ¹³C NMR $(75.4 \text{ MHz}, \text{ CDCl}_3): \delta = 192.2, 172.7, 140.3, 137.0, 135.8,$ 133.8, 131.4, 130.7, 128.8, 128.7, 128.6, 127.5, 127.2, 126.4, 69.5, 58.9, 29.9, 26.7; IR (neat): $\nu = 1742$, 1684, 1597, 1580, 1495, 1448, 1356, 1325, 1275, 1249, 1212, 1182, 1129, 1075, 1015 cm⁻¹; MS (70 eV, EI): m/z (%)=332 (M⁺, 6.13), 105 (100); anal. calcd. for C₂₂H₂₀O₃ (%): C 79.50, H 6.06; found: C 79.56, H 6.27.

Synthesis of 3-Acetyl-5-phenyl-5(*E*)-3,4,8,9-tetrahydrooxonin-2(3*H*)-one (2q)

To a three-necked flask were added $Pd(OAc)_2$ (67.3 mg, 0.3 mmol), PPh₃ (158.0 mg, 0.6 mmol), K₂CO₃ (1.6581 g, 12.0 mmol), CH₃CN (28 mL), iodobenzene (1.8366 g, 9.0 mmol), CH₃CN (1 mL), 1g (1.0947 g, 6.0 mmol), and CH₃CN (1 mL) sequentially under argon. The mixture was stirred at 85°C with a preheated oil bath and monitored by TLC. Upon completion, the resulting mixture was filtered to remove the solid and the filtrate was concentrated. The residue was purified by chromatography on silica gel to afford **2q** (eluent: petroleum ether/ethyl acetate = 10/1) as a liquid; yield: 1.2690 g (82%). ¹H NMR (300 MHz, CDCl₃): $\delta =$ 7.39–7.17 (m, 5H, Ar-H), 5.77 (dd, $J_1 = 11.1$ Hz, $J_2 = 6.6$ Hz, 1H, =CH), 4.60-4.46 (m, 1H, one proton from OCH₂), 4.25-4.12 (m, 1H, one proton from OCH₂), 3.33-3.17 [m, 2H, O₂CCH, and one proton from C=C(Ph)CH₂], 2.86 [dd, $J_1 = 19.7 \text{ Hz}, J_2 = 11.1 \text{ Hz}, 1 \text{ H}, \text{ one proton from}$ C =C(Ph)CH₂], 2.47–2.27 [m, 1H, one proton from (Ph)C= CCH₂], 2.27–2.12 [m, 4H, one proton from (Ph)C=CCH₂ and CH₃], 2.06–1.85 (m, 2H, CCH₂C); ¹³C NMR (75.4 MHz, CDCl₃): $\delta = 201.6$, 170.7, 142.4, 136.6, 131.9, 128.3, 127.1, 126.7, 63.3, 58.2, 30.0, 28.8, 26.9, 23.3; IR (neat): $\nu = 1714$, 1598, 1492, 1462, 1445, 1377, 1358, 1275, 1218, 1150, 1064, 1042, 1028 cm⁻¹; MS (70 eV, EI): m/z (%) = 258 (M⁺, 16.80), 43 (100); HR-MS: m/z = 258.1258, calcd. for C₁₆H₁₈O₃ (M⁺): 258.1256.

Synthesis of 3-Acetyl-5-(4-methoxyphenyl)-5(*E*)-3,4,8,9-tetrahydrooxonin-2(3*H*)-one (2r)

The reaction of Pd(OAc)₂ (5.8 mg, 0.025 mmol), PPh₃ (13.3 mg, 0.05 mmol), K₂CO₃ (138.5 mg, 1.0 mmol), *p*-methoxyphenyl iodide (174.6 mg, 0.75 mmol), and **1g** (92.1 mg, 0.5 mmol) in 4 mL of CH₃CN afforded **2r** (eluent: petroleum ether/ethyl acetate = 10/1) as a liquid; yield: 114.4 mg (78%). ¹H NMR (300 MHz, CDCl₃): δ =7.21 (d, *J*=8.7 Hz, 2H, Ar-H), 6.84 (d, *J*=8.7 Hz, 2H, Ar-H), 5.71 (dd, *J*₁ = 11.1 Hz, *J*₂=6.3 Hz, 1H, =CH), 4.58–4.46 (m, 1H, one proton from OCH₂), 4.25–4.12 (m, 1H, one proton from OCH₂), 3.79 (s, 3H, OCH₃), 3.30–3.16 [m, 2H, O₂CCH and one proton from C=C(Ar)CH₂], 2.88–2.75 [m, 1H, one proton from (Ar)C=CCH₂], 2.27–2.12 [m, 4H, one proton from (Ar)C=CCH₂ and COCH₃], 2.06–1.85 (m, 2H, CCH₂C); ¹³C NMR (75.4 MHz, CDCl₃): $\delta = 201.7$, 170.7, 158.8, 136.0, 134.7, 130.5, 127.7, 113.7, 63.3, 58.3, 55.2, 30.0, 28.8, 26.9, 23.2; IR (neat): $\nu = 1714$, 1607, 1573, 1510, 1462, 1358, 1291, 1244, 1218, 1178, 1150, 1064, 1030 cm⁻¹; MS (70 eV, EI): m/z(%) = 288 (M⁺, 27.16), 43 (100); HR-MS: m/z = 288.1361, calcd. for C₁₇H₂₀O₄ (M⁺): 288.1362.

Synthesis of 3-Acetyl-5-(4-nitrophenyl)-5(*E*)-3,4,8,9tetrahydrooxonin-2(3*H*)-one (2s)

The reaction of Pd(OAc)₂ (5.4 mg, 0.025 mmol), PPh₃ (13.1 mg, 0.05 mmol), K₂CO₃ (138.0 mg, 1.0 mmol), 1-iodo-4-nitrobenzene (186.6 mg, 0.75 mmol), and 1g (91.5 mg, 0.5 mmol) in 4 mL of CH₃CN afforded 2s (eluent: petroleum ether/ethyl ether = 5/1) as a yellow solid; yield: 112.8 mg (74%); mp 115–116°C (ethyl ether/petroleum ether). ¹H NMR (300 MHz, CDCl₃): $\delta = 8.15$ (d, J = 8.7 Hz, 2H, Ar-H), 7.43 (d, J=8.7 Hz, 2H, Ar-H), 5.92 (dd, $J_1=11.1$ Hz, $J_2 = 6.6$ Hz, 1H, =CH), 4.57-4.46 (m, 1H, one proton from OCH₂), 4.31-4.17 (m, 1H, one proton from OCH₂), 3.37-3.15 [m, 2H, O₂CCH and one proton from C=C(Ar)CH₂], 2.86 [dd, $J_1 = 13.2$ Hz, $J_2 = 3.3$ Hz, 1 H, one proton from C= C(Ar)CH₂], 2.48-2.33 [m, 1H, one proton from (Ar)C= CCH₂], 2.33–2.17 [m, 4H, one proton from (Ar)C=CCH₂ and CH₃], 2.11–1.88 (m, 2H, CCH₂C); ¹³C NMR (75.4 MHz, CDCl₃): $\delta = 201.0, 170.4, 149.0, 146.8, 135.3, 135.2, 127.4,$ 123.7, 63.3, 57.9, 29.5, 28.8, 26.8, 23.5; IR (neat): $\nu = 1710$, 1592, 1511, 1460, 1441, 1343, 1317, 1284, 1263, 1209, 1159, 1149, 1100, 1060, 1025 cm⁻¹; MS (70 eV, EI): m/z (%)=303 $(M^+, 3.07), 43 (100);$ anal. calcd. for $C_{16}H_{17}NO_5$ (%): C 63.36, H 5.65, N 4.62; found: C 63.33, H 5.64, N 4.61.

Synthesis of 3-Acetyl-5-phenyl-5(*E*)-3,4,7,8,9,10hexahydro-2*H*-oxecin-2-one (2t)

The reaction of $Pd(OAc)_2$ (5.7 mg, 0.025 mmol), PPh_3 (13.1 mg, 0.05 mmol), K₂CO₃ (137.4 mg, 1.0 mmol), iodobenzene (152.2 mg, 0.75 mmol), and **1h** (97.8 mg, 0.5 mmol) in 4 mL of CH₃CN afforded 2t [eluent: petroleum ether (b.p. 30-60 °C)/ethyl acetate = 15/1] as a solid; yield: 96.4 mg (71%); m.p.: 100-101 °C (ethyl ether/petroleum ether). ¹H NMR (300 MHz, CDCl₃): $\delta = 7.40-7.17$ (m, 5H, Ar-H), 5.56 (dd, $J_1 = 12.3$ Hz, $J_2 = 3.9$ Hz, 1H, =CH), 4.93–4.81 (m, 1 H, one proton from OCH₂), 3.79 (t, J=11.4 Hz, 1 H, one proton from OCH₂), 3.40-3.25 [m, 2H, O₂CCH and one proton from C=C(Ph)CH₂], 2.93-2.64 (m, 2H), 2.28-2.12 (m, 4H), 2.12-1.85 (m, 1H), 1.85-1.72 (m, 2H), 1.59-1.45 (m, 1H); ¹³C NMR (75.4 MHz, CDCl₃): $\delta = 202.0$, 168.1, 142.1, 136.9, 131.6, 128.4, 127.2, 126.7, 67.2, 59.4, 28.5, 27.8, 27.4, 27.2, 25.0; IR (neat): v=1712, 1573, 1489, 1456, 1442, 1384, 1354, 1315, 1292, 1257, 1220, 1206, 1178, 1149, 1049, 1028 cm⁻¹; MS (70 eV, EI): m/z (%)=272 (M⁺, 4.88), 43 (100); anal. calcd. for $C_{17}H_{20}O_3$ (%): C 74.97, H 7.40; found: C 74.65, H 7.29.

Synthesis of 3-Acetyl-5-(4-bromophenyl)-5(*E*)-3,4,7,8,9,10-hexahydro-2*H*-oxecin-2-one (2u)

The reaction of $Pd(OAc)_2$ (5.6 mg, 0.025 mmol), PPh_3 (13.1 mg, 0.05 mmol), K_2CO_3 (138.1 mg, 1.0 mmol), 1bromo-4-iodobenzene (212.6 mg, 0.75 mmol), and **1h** (98.9 mg, 0.5 mmol) in 4 mL of CH₃CN afforded **2u** (eluent:

petroleum ether/dichloromethane = $1/1 \rightarrow 2/3$) as a vellow solid; yield: 122.4 mg (69%); mp 95-96°C (ethyl ether/petroleum ether); ¹H NMR (300 MHz, CDCl₃): $\delta = 7.40$ (d, J =8.4 Hz, 2H, Ar-H), 7.12 (d, J=8.4 Hz, 2H, Ar-H), 5.54 (dd, $J_1 = 12.3$ Hz, $J_2 = 3.9$ Hz, 1H, =CH), 4.90-4.76 (m, 1H, one proton from OCH₂), 3.77 (t, J=11.1 Hz, 1H, one proton from OCH₂), 3.35–3.20 [m, 2H, O₂CCH and one proton from C=C(Ar)CH₂], 2.87-2.55 (m, 2H), 2.26-2.10 (m, 4H), 1.99–1.68 (m, 3H), 1.57–1.43 (m, 1H); ¹³C NMR (75.4 MHz, $CDCl_3$): $\delta = 201.5$, 167.9, 140.9, 135.8, 132.2, 131.5, 128.3, 121.0, 67.2, 59.2, 28.5, 27.5, 27.4, 27.1, 24.9; IR (neat): $\nu =$ 1736, 1705, 1482, 1455, 1432, 1392, 1362, 1315, 1291, 1264, 1223, 1204, 1182, 1154, 1100, 1071, 1046, 1020, 1009 cm^{-1} ; MS (70 eV, EI): m/z = (%): 353 [M(⁸¹Br)⁺, 0.87], 351 $[M(^{79}Br)^+, 0.79], 43 (100);$ anal. calcd. for $C_{17}H_{19}BrO_3$ (%): C 58.13, H 5.45; found: C 58.13, H 5.17.

Synthesis of 3-Acetyl-3-propyl-5-phenyl-5(*E*)-3,4,7,8-tetrahydro-2*H*-oxocin-2-one (2v)

The reaction of Pd(OAc)₂ (5.5 mg, 0.025 mmol), PPh₃ (13.2 mg, 0.050 mmol), K₂CO₃ (138.5 mg, 1.0 mmol), iodo-(255.1 mg, 1.25 mmol), and **1i** (106.0 mg, benzene 0.50 mmol) in 4 mL of CH₃CN afforded 2v [eluent: petroleum ether/ethyl ether = 10/1] as a white solid; yield: 96.1 mg (67%); mp 111-112°C (ethyl ether/petroleum ether). ¹H NMR (300 MHz, CDCl₃): $\delta = 7.33 - 7.19 \text{ (m, 5H, Ar-H)}$, 5.91 (t, J=7.8 Hz, 1 H, C=CH), 4.62 (t, J=11.1 Hz, 1 H, one proton from OCH₂), 4.11-4.00 (m, 1H, one proton from OCH_2), 3.37 [d, J=13.2 Hz, 1H, one proton from C= $C(Ph)CH_2$, 2.90 [d, J=13.2 Hz, 1H, one proton from C= C(Ph)CH₂], 2.81-2.65 (m, 1H, one proton from PhC= CCH_2), 2.29–2.17 (m, 1H, one proton from PhC=CCH₂), 2.11 (s, 3H, COCH₃), 1.73–1.59 (m, 1H, one proton from O₂CCCH₂C), 1.28–1.15 (m, 2H), 1.00–0.69 (m, 2H, one proton from O_2CCCCH_2 , 0.44 (t, J=7.2 Hz, 3H, CCH₃); ¹³C NMR (75.4 MHz, CDCl₃): $\delta = 204.3$, 175.7, 144.5, 143.2, 128.3, 127.4, 127.1, 126.3, 69.5, 67.4, 33.0, 32.3, 29.9, 26.6, 16.7, 13.8; IR (neat): $\nu = 1749$, 1709, 1598, 1490, 1462, 1442, 1360, 1320, 1301, 1268, 1248, 1193, 1159, 1106, 1078, 1051, 1032, 1008 cm⁻¹; MS (70 eV, EI): m/z (%)=286 (M⁺, 1.57), 43 (100); anal. calcd. for C₁₈H₂₂O₃ (%): C 75.50, H 7.74; found: C 75.44, H 7.82.

Synthesis of 3-Acetyl-3-benzyl-5-phenyl-5(*E*)-3,4,7,8-tetrahydro-2*H*-oxocin-2-one (2w)

The reaction of Pd(OAc)₂ (5.4 mg, 0.025 mmol), PPh₃ (13.0 mg, 0.050 mmol), K₂CO₃ (138.0 mg, 1.0 mmol), iodo-(155.0 mg, 0.75 mmol), and **1**j (128.8 mg, benzene 0.50 mmol) in 4 mL of CH₃CN afforded 2w [eluent: petroleum ether/ethyl ether = 10/1] as a white solid; yield: 124.9 mg (75%); m.p.114–115°C (ethyl ether/petroleum ether). ¹H NMR (300 MHz, CDCl₃): $\delta = 7.37 - 7.22$ (m, 5H, Ar-H), 7.15-7.05 (m, 3H, Ar-H), 6.70-6.65 (m, 2H, Ar-H), 5.90 (t, J = 8.0 Hz, 1H, C=CH), 4.69 (t, J = 11.0 Hz, 1H, one proton from OCH_2), 4.15–4.05 (m, 1H, one proton from OCH_2), 3.41 (d, J=12.9 Hz, 1 H), 3.17 (d, J=14.1 Hz, 1 H), 2.89 (d, J = 13.2 Hz, 1H), 2.78–2.62 (m, 1H, one proton from PhC= CCH_2), 2.57 (d, J = 14.4 Hz, 1H), 2.35–2.22 (m, 1H, one proton from PhC=CCH₂), 1.93 (s, 3H, COCH₃); 13 C NMR $(75.4 \text{ MHz}, \text{ CDCl}_3): \delta = 205.6, 175.4, 144.9, 143.3, 135.7,$ 129.8, 128.4, 128.3, 128.1, 127.3, 127.0, 126.7, 70.6, 67.0, 38.5, 36.5, 29.9, 28.8; IR (neat): ν =1746, 1709, 1600, 1491, 1459, 1437, 1360, 1321, 1298, 1262, 1227, 1210, 1172, 1154, 1111, 1067, 1050, 1006 cm⁻¹; MS (70 eV, EI): m/z (%)=334 (M⁺, 1.34), 43 (100); anal. calcd. for C₂₂H₂₂O₃ (%): C 79.02, H 6.63; found: C 79.09, H 6.64.

Synthesis of 3-acetyl-5-(4-nitrophenyl)-5(*E*)-3,4,7,8-tetrahydro-2*H*-oxocin-2-one (2x)

The reaction of $Pd(OAc)_2$ (5.6 mg, 0.025 mmol), PPh_3 (13.0 mg, 0.050 mmol), K₂CO₃ (138.7 mg, 1.0 mmol), 1-iodo-4-nitrobenzene (186.5 mg, 0.75 mmol), and 1k (92.1 mg, 0.50 mmol) in 4 mL of CH₃CN afforded 2x [eluent: petroleum ether/ethyl acetate = 9/1] as a solid; yield: 99.6 mg (65%); mp 144–145°C (CH₂Cl₂/petroleum ether). The major isomer: ¹H NMR (300 MHz, CDCl₃): $\delta = 8.15$ (d, J = 8.7 Hz, 2H, Ar-H), 7.41 (d, J=8.7 Hz, 2H, Ar-H), 5.72 (d, J=7.2 Hz, 1H, C=CH), 4.29 (t, J=11.0 Hz, 1H, one proton from OCH₂), 4.02 (dd, $J_1 = 10.7$ Hz, $J_2 = 4.4$ Hz, 1 H), 3.51 (dd, $J_1 = 11.9$ Hz, $J_2 = 4.1$ Hz, 1 H), 3.38 (t, J = 12.5 Hz, 1 H), 3.07–2.91 (m, 1H, CHCH₃), 2.79 (dd, $J_1 = 13.2$ Hz, $J_2 =$ 3.9 Hz, 1 H), 2.22 (s, 3 H, COCH₃), 1.06 (d, J=6.9 Hz, 3 H, CCH₃); ¹³C NMR (75.4 MHz, CDCl₃): $\delta = 200.1$, 172.5, 148.7, 147.1, 139.6, 137.5, 127.3, 123.7, 72.9, 63.8, 35.2, 31.3, 28.9, 15.8; IR (neat) $\nu = 2962$, 2926, 1754, 1705, 1592, 1511, 1456, 1342, 1259, 1207, 1175, 1143, 1112, 1072, 1038, 1000 cm⁻¹; MS (70 eV, EI): m/z (%)=303 (M⁺, 1.63), 43 (100); anal. calcd. for $C_{16}H_{17}NO_5$ (%): C 63.36, H 5.65, N 4.62; found: C 63.36, H 5.62, N 4.41.

Acknowledgements

Financial support from the State Key Basic Research & Development Program (2011CB808700) and National Natural Science Foundation of China (20732005) are greatly appreciated. We thank Mr. B. Lü in this group for reproducing the preparation of 2d in Table 3, 2i and 2o in Table 4, and E-2u in Scheme 3.

References

- For reviews, see: a) A. Parenty, X. Moreau, J.-M Campagne, *Chem. Rev.* 2006, 106, 911; b) I. Shiina, *Chem. Rev.* 2007, 107, 239.
- [2] a) D. M. Tapiolas, M. Roman, W. Fenical, J. Am. Chem. Soc. 1991, 113, 4682; b) H. Niwa, K. Wakamatsu, K. Yamada, Tetrahedron Lett. 1989, 30, 4543; c) M. S. Congreve, A. B. Holmes, A. B. Hughes, M. G. Looney, J. Am. Chem. Soc. 1993, 115, 5815.
- [3] a) G. Illuminati, L. Mandolini, Acc. Chem. Res. 1981, 14, 95; b) G. A. Molander, Acc. Chem. Res. 1998, 31, 603.
- [4] For selected examples of lactonization: a) K. C. Nico-laou, *Tetrahedron* 1977, 33, 683; b) S. Masamune, G. S. Bates, J. W. Corcoran, *Angew. Chem.* 1977, 89, 602; *Angew. Chem. Int. Ed. Engl.* 1977, 16, 585; c) G. Rousseau, *Tetrahedron* 1995, 51, 2777; d) R. D. Norcross, I. Paterson, *Chem. Rev.* 1995, 95, 2041.

- [5] J. C. Conrad, M. D. Eelman, J. A. Duarte Silva, S. Monfette, H. H. Parnas, J. L. Snelgrove, D. E. Fogg, J. Am. Chem. Soc. 2007, 129, 1024.
- [6] For reviews, see: a) R. Zimmer, C. U. Dinesh, E. Nandanan, F. A. Khan, Chem. Rev. 2000, 100, 3067; b) J. A. Marshall, Chem. Rev. 2000, 100, 3163; c) A. S. K. Hashmi, Angew. Chem. 2000, 112, 3737; Angew. Chem. Int. Ed. 2000, 39, 3590; d) R. W. Bates, V. Satcharoen, Chem. Soc. Rev. 2002, 31, 12; e) L. K. Sydnes, Chem. Rev. 2003, 103, 1133; f) S. Ma, Acc. Chem. Res. 2003, 36, 701; g) S. Ma, Chem. Rev. 2005, 105, 2829; h) S. Ma, Pure Appl. Chem. 2006, 78, 197; i) S. Ma, Aldrichimica Acta 2007, 40, 91.
- [7] For selected reports from this group, see: a) S. Ma, N. Jiao, S. Zhao, H. Hou, J. Org. Chem. 2002, 67, 2837;
 b) S. Ma, N. Jiao, Q. Yang, Z. Zheng, J. Org. Chem. 2004, 69, 6463; c) F. Yu, X. Lian, S. Ma, Org. Lett. 2007, 9, 1703; d) S. Ma, Z. Zheng, X. Jiang, Org. Lett. 2007, 9, 529; e) Y. Deng, Y. Yu, S. Ma, J. Org. Chem. 2008, 73, 585; f) X. Cheng, S. Ma, Angew. Chem. 2008, 120, 4657; Angew. Chem. Int. Ed. 2008, 47, 4581; g) X. Jiang, X. Ma, Z. Zheng, S. Ma, Chem. Eur. J. 2008, 14, 8572.
- [8] a) S. Ma, E. Negishi, J. Org. Chem. 1994, 59, 4730; b) S.
 Ma, E. Negishi, J. Am. Chem. Soc. 1995, 117, 6345.
- [9] R. C. Larock, C. Tu, P. Pace, J. Org. Chem. 1998, 63, 6859.
- [10] a) B. M. Trost, P.-Y. Michellys, V. J. Gerusz, Angew. Chem. 1997, 109, 1837; Angew. Chem. Int. Ed. Engl. 1997, 36, 1750; for some reports about hydropalladation of allenes, see: b) Y. Yamamoto, M. Al-Masum, N. Asao, J. Am. Chem. Soc. 1994, 116, 6019; c) M. Meguro, S. Kamijo, Y. Yamamoto, Tetrahedron Lett. 1996, 37, 7453.
- [11] For a recent report from this group, see: X. Jiang, Q. Yang, Y. Yu, C. Fu, S. Ma, *Chem. Eur. J.* 2009, 15, 7283.
- [12] S. R. Wilson, C. E. Augelli, Org. Synth. 1990, 68, 210.
- [13] For palladium-catalyzed allylations with stabilized nucleophiles via C-attack, see: a) J. Tsuji, I. Minami, I. Shimizu, Tetrahedron Lett. 1983, 24, 1793; b) Y. Huang, X. Lu, Tetrahedron Lett. 1988, 29, 5663; c) M. Ahmar, B. Zazes, J. Gore, Tetrahedron Lett. 1984, 25, 4505; d) M. Ahmar, J. J. Barieux, B. Zazes, J. Gore, Tetrahedron 1987, 43, 513; e) C. Kammerer, G. Prestat, D. Madec, G. Poli, Chem. Eur. J. 2009, 15, 4224.
- [14] For palladium-catalyzed allylations with stabilized nucleophiles *via O*-attack, see: a) J. Tsuji, Y. Kabayashi, H. Kataoka, T. Takahashi, *Tetrahedron Lett.* 1980, 21, 1475; b) ref.^[7g]; c) ref.^[7j]
- [15] For reviews, see: a) C. G. Frost, J. Howarth, J. M. Williams, *Tetrahedron: Asymmetry* **1992**, *3*, 1089; b) B. M. Trost, D. L. Van Vranken, *Chem. Rev.* **1996**, *96*, 395.
- [16] Crystal data for compound **2m**: $C_{80}H_{68}Br_4O_{12}$; MW = 1540.98, triclinic, space group P-1, final *R* indices $[I > 2\sigma(I)]$, R1 = 0.0759, wR2 = 0.1538, *R* indices (all data)

R1=0.1140, wR2=0.1688, a=12.2236(10) Å, b= 16.4269(13) Å, c=18.2484(15) Å, α =75°, β =70°, γ = 90°, V=3328.2(5) Å³, T=173(2) K, Z=2, reflections collected/unique 38333/11633 (R_{int} =0.0618), number of observations [> $2\sigma(I)$] 7873, parameters: 865. Supplementary crystallographic data have been deposited at the Cambridge Crystallographic Data Center under CCDC 801336. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif.

- Crystal data for compound **2s**: $C_{16}H_{17}NO_5$; MW =[17] 303.31, triclinic, space group P-1, final R indices [I> $2\sigma(I)$], R1=0.0372, wR2=0.0972, R indices (all data) R1 = 0.0468, wR2 = 0.1035, a = 8.1581(6) Å, b =9.5889(7) Å, c = 10.4998(8) Å, $\alpha = 109^{\circ}$, $\beta = 106^{\circ}$, $\gamma =$ 95°, V = 733.42(9) Å³, T = 296(2) K, Z = 2, reflections collected/unique 7843/2585 (R_{int} =0.0396), number of observations [> $2\sigma(I)$] 2135, parameters: 199. Supplementary crystallographic data have been deposited at the Cambridge Crystallographic Data Centre under CCDC 801335. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif.
- [18] Crystal data for compound **2u**: $C_{17}H_{19}BrO_3$; MW = 351.23, monoclinic, space group C2/c, final R indices $[I > 2\sigma(I)]$, R1 = 0.0295, wR2 = 0.0717, R indices (all data) R1 = 0.0386, wR2 = 0.0769, a = 27.5708(9) Å, b = 6.0338(2) Å, c = 19.6616(7) Å, $a = 90^{\circ}$, $\beta = 102^{\circ}$, $\gamma = 90^{\circ}$, V = 3200.57(19) Å³, T = 296(2) K, Z = 8, reflections collected/unique 17661/2809 ($R_{int} = 0.0285$), number of observations [$> 2\sigma(I)$] 2354, parameters: 190. Supplementary crystallographic data have been deposited at the Cambridge Crystallographic Data Centre under CCDC 801334. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif.
- [19] Crystal data for compound **2x**: $C_{16}H_{17}NO_5$; MW = 303.31, triclinic, space group P-1, final *R* indices $[I > 2\sigma(I)]$, R1 = 0.0357, wR2 = 0.0912, *R* indices (all data) R1 = 0.0406, wR2 = 0.0967, a = 8.8711(5) Å, b = 9.5519(5) Å, c = 10.3514(6) Å, $\alpha = 65^{\circ}$, $\beta = 70^{\circ}$, $\gamma = 81^{\circ}$, V = 743.99(7) Å³, T = 296(2) K, Z = 2, reflections collected/unique 8679/2618 ($R_{int} = 0.0155$), number of observations [> $2\sigma(I)$] 2316, parameters: 199. Supplementary crystallographic data have been deposited at the Cambridge Crystallographic Data Centre under CCDC 822025. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif.
- [20] a) F.-Y. Yang, M.-Y. Wu, C.-H. Cheng, J. Am. Chem. Soc. 2000, 122, 7122; b) Z. Gu, X. Wang, W. Shu, S. Ma, J. Am. Chem. Soc. 2007, 129, 10948.
- [21] The terminal allenols were prepared from terminal alkynols according to the known procedure: J. Kuang, S. Ma, J. Org. Chem. 2009, 74, 1763.

1774