

Prithwish Ghosh,[†] Na Yeon Kwon,[†] Sangil Han, Saegun Kim, Sang Hoon Han, Neeraj Kumar Mishra, Young Hoon Jung, Sang J. Chung,*[®] and In Su Kim*[®]

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School of Pharmacy, Sungkyunkwan University, Suwon 16419, Republic of Korea

Supporting Information

Phosphonium Ylides

ABSTRACT: The synthesis of alkylated diazine derivatives is important for their practical utilization as pharmaceuticals and for other purposes. Herein, we describe the metal-free siteselective C-H alkylation of diazine N-oxides using phosphonium ylides that affords a variety of alkylated diazine derivatives with broad functional group tolerance. The utility of this method is showcased by the late-stage functionalization of a commercially available drug such as varenicline. Notably, the sequential C-H alkylation of pyrazine N-oxides for the total synthesis of a pyrazine-containing natural product, paenibacillin A, highlights the importance of this method.

he development of effective synthetic methods is important for advancing medicinal, agrochemical, and other chemical industries. The ability to synthesize substituted diazines is of great significance because these motifs are commonly found in biologically relevant molecules.¹ In particular, alkylated diazines are of paramount interest in medicinal chemistry due to their remarkable therapeutic potential. For instance, methoxymethylated, cyclopropylated, and cyclobutylated pyrazine frameworks have been recognized as essential units with potent pharmacological activity (Figure $1).^{2}$





alvcine transporter inhibitor





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However, the direct installation of these alkyl groups on diazine rings has been rarely explored. The classical approach for the synthesis of alkylated diazines is the transition metalmediated cross-coupling reaction between functionalized diazines and organometallic reagents (Scheme 1).³

The metal-mediated Minisci-type alkylation of diazines in the presence of radical sources has been reported.^{4a,b} In addition, the metal-free Minisci-type alkylations of various Nheterocycles were also disclosed.^{4c-e} However, the generation of residual metal wastes and regioisomeric impurities remains

Scheme 1. Site-Selective Alkylation of Diazine Heterocycles previous work

cat. [Ni, Fe, Pd]

a) diazine alkylations using organometallic reagents

b) diazine alkylations by Minisci-type reaction



this work (reductive alkylation of diazine-N-oxides)



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an obstacle for the pharmaceutical application of alkylated diazines. Therefore, an efficient method for the formation of alkylated diazines under mild and metal-free conditions is highly desirable.

Diazine N-oxides have been used as precursors in the synthesis of functionalized diazine derivatives.⁵ Halogenation, cyanation, and olefination have been described previously.⁶ Additionally, direct lithiation and subsequent alkylation or acylation of pyrazine N-oxides were investigated.⁷ Phosphonium ylides are conventionally utilized for olefination reaction with carbonyl compounds.⁸ Moreover, phosphonium salts have been employed in Michael additions,⁹ organocatalytic Mannich reactions,¹⁰ and other alkylation reactions.¹¹ Recently, our group first demonstrated the metal-free site-selective C-H alkylation of pyridine and quinoline N-oxides using Wittig reagents.^{12a} Other alkyl sources were also utilized for the metal-free C-H alkylations of heterocyclic N-oxides.^{12b-d} Due to the widespread relevance of diazine molecules in recent drug discovery, we herein describe the metal-free site-selective alkylation of diazine N-oxides using phosphonium ylides to furnish alkylated diazines. Notably, the sequential transformation of diazine N-oxides leading to the formation of highly substituted diazines highlights the applicability of the developed methodology.

This investigation was initiated by examining the optimal reaction conditions for the coupling of pyrazine *N*-oxide **1a** with methoxymethyltriphenylphosphonium chloride (**2a**), as shown in Table 1. Screening of various reaction conditions showed that the use of KO^tBu (3 equiv) in a THF solvent at 80 °C with a reaction time of 7 h afforded C2-alkylated pyrazine **3a** in 78% yield (Table 1, entry 2). It should be noted that this reaction was scaled up to 1 g (8.06 mmol) to give 0.95 g of **3a** in 77% yield (Table 1, entry 14).

Table 1. Selected Optimization of Reaction Conditio

Me	N			MeN_
	•	—PPh ₃ CI b	oase, solvent	I]
Me	N° H T MeO ¦⊖	20	80 °C, 7 h	Me N
	1a	24		OMe 3a
entry	base (equiv)	2a (equiv)	solvent	yield ^b
1	$KO^{t}Bu$ (2)	2	THF	25
2	$KO^{t}Bu$ (3)	3.5	THF	78
3	$KO^{t}Bu$ (3)	5	THF	76
4	NaO ^t Bu (3)	3.5	THF	8
5	$LiO^{t}Bu$ (3)	3.5	THF	trace
6	KOMe (3)	3.5	THF	6
7	KHMDS (3)	3.5	THF	10
8	DBU (3)	3.5	THF	no reaction
9	$KO^{t}Bu$ (3)	3.5	1,4-dioxane	20
10	$KO^{t}Bu$ (3)	3.5	CPME	72
11	$KO^{t}Bu$ (3)	3.5	MTBE	70
12	$KO^{t}Bu$ (3)	3.5	toluene	74
13 ^c	$KO^{t}Bu$ (3)	3.5	THF	51
14 ^d	KO ^t Bu (3)	3.5	THF	77

^{*a*}Reaction conditions: **1a** (0.2 mmol), **2a** (quantity noted), base (quantity noted), solvent (2 mL) at 80 °C for 7 h under a N_2 atmosphere in pressure tubes. Abbreviations: KHMDS, potassium bis(trimethylsilyl)amide; CPME, cyclopentyl methyl ether; MTBE, *tert*-butylmethyl ether. ^{*b*}Isolated percent yield determined by flash column chromatography. ^{*c*}The reaction was carried out at 60 °C. ^{*d*}Gram-scale experiment.

Having determined the optimal reaction conditions, we examined the scope of the reaction with a variety of pyrazine and quinoxaline *N*-oxides and phosphonium salts (Scheme 2).





^{*a*}Reaction conditions: 1 (0.2 mmol), 2 (0.7 mmol, 3.5 equiv), KO^{*t*}Bu (3 equiv), THF (2 mL) at 80 °C for 7 h. ^{*b*}Reaction conditions: 1 (0.2 mmol), 2 (0.7 mmol, 3.5 equiv), KO^{*t*}Bu (3.5 equiv), CPME (2 mL) at 90 °C for 10 h. ^{*c*}Reaction conditions: 1 (0.2 mmol), 2 (0.6 mmol, 3 equiv), KO^{*t*}Bu (2.5 equiv), THF (2 mL) at 80 °C for 6 h. ^{*d*}Reaction conditions: 1 (0.2 mmol), 2 (0.7 mmol, 3.5 equiv), KO^{*t*}Bu (3.5 equiv), THF (2 mL) at 80 °C for 6 h. ^{*d*}Reaction conditions: 1 (0.2 mmol), 2 (0.7 mmol, 3.5 equiv), KO^{*t*}Bu (3.5 equiv), THF (2 mL) at 80 °C for 10 h. ^{*e*}Reaction conditions: 1 (0.2 mmol), 2 (0.6 mmol, 3 equiv), KO^{*t*}Bu (3.5 equiv), THF (2 mL) at 80 °C for 4 h. ^{*f*}Reaction conditions: 1 (0.2 mmol), 2 (0.7 mmol, 3.5 equiv), KO^{*t*}Bu (3.5 equiv), THF (2 mL) at 80 °C for 6 h.

As anticipated, 2,3-dimethylpyrazine N-oxide (1a) was smoothly coupled with $EtPPh_3Br$ (2b) and cyclopropyl PPh_3Br (2c) to afford alkylated pyrazines 3b (53%) and 3c (72%), respectively.

To evaluate the steric effect in this process, 2,5dimethylpyrazine *N*-oxide (1b) was coupled with branched or linear alkylphosphonium salts 2d-2f. Under the modified reaction conditions, the corresponding adducts 3d-3f were formed in moderate to good yields. In addition, C3-aryl- and C3-oxyaryl-substituted pyrazine *N*-oxides 1c-1h exclusively underwent C-H alkylation at the less hindered site, affording

the desired products 3g-3n. Moreover, C2-aryl-substituted pyrazine N-oxides 1i-1p also participated in the C6 alkylation reaction to furnish the corresponding products 3o-3v, respectively. The tolerance of this reaction system for NO₂ and CN groups was of interest, as these moieties provide versatile synthetic handles in the products. This reaction proceeded readily with unsubstituted pyrazine N-oxide (1g) to furnish 3w in 61% yield. The chloro-substituted pyrazine Noxide 1r was also compatible, affording the desired product 3x, which can be further functionalized by nucleophilic aromatic substitution reactions.¹³ Furthermore, quinoxaline N-oxide (1s) also reacted with MePPh₃I (2g) to give 3y in 78% yield. It is known that phosphonium ylide (CH₃CO₂CH=PPh₃) and phosphonium salts (phenyl PPh₃Br and pyridinyl PPh₃Br) were unsuccessful in this coupling reaction, presumably due to the lower nucleophilicity of the delocalized carbanion and the lack of formation of (hetero)aryl carbanions under the current reaction conditions. Notably, this protocol allows direct incorporation of the cyclobutyl moiety on the diazine frameworks. For example, pyrazine N-oxide 1a was smoothly coupled with 4-bromobutylPPh₃Br (2i) to provide the C2cyclobutylated pyrazine adduct 4a in 41% yield (Scheme 3).

Scheme 3. Synthesis of the C2-Cyclobutylated Pyrazine Adduct



To gain mechanistic insight into this process, we first performed a deuterium labeling experiment using **deuterio-1s** and **2g** for 2 and 6 h, respectively, under otherwise identical reaction conditions (Scheme 4). Partial deuterium incorporation (10% D) at the benzyl position of the product **deuterio-3y** was observed after reaction for 2 h. This result reveals that

Scheme 4. Deuterium Labeling Experiments and Plausible Reaction Mechanism



intramolecular deuterium migration can be excluded in the reaction pathway. Moreover, 3% incorporation of deuterium at the benzyl position was detected after 6 h, indicatve of rapid proton exchange of benzylic deuterium under basic reaction conditions. Furthermore, the deuterium labeling experiment was conducted by using deuterio-1s and 2i, resulting in 15% deuterium incorporation. It should be noted that no exchange of deuterium at the C3 position was observed in any experiment. On the basis of the deuterium labeling experiments, a plausible reaction mechanism was proposed, as outlined in Scheme 4. The initial formation of phosphonium ylide and subsequent intramolecular cyclization afforded cyclobutyl phosphonium salt A, which further generated the second ylide intermediate B in the presence of KO^tBu. The intermolecular [3+2] annulation reaction between quinoxaline N-oxide deuterio-1s and phosphonium ylide B furnished intermediate C, leading to the formation of the C2-cyclobutylated quinoxaline deuterio-4b through aromatization by an external base.

Pyrimidine and pyridazine *N*-oxides 5a-5d were also employed in the reaction with 2a and 2c, as shown in Scheme 5. To our delight, C2-substituted pyrimidine *N*-oxides 5a and

Scheme 5. Substrate Scope of Pyrimidine and Pyridazine *N*-Oxides



^{*a*}Reaction conditions: **5a**–**5c** (0.2 mmol), **2a** or **2c** (0.6 mmol, 3 equiv), KO^tBu (3 equiv), THF (2 mL) at 80 °C for 6 h under N₂ in reaction tubes. ^{*b*}Reaction conditions: **5d** (0.2 mmol), **2a** (0.4 mmol, 2 equiv), KHMDS (1.8 equiv), THF (2 mL) at room temperature for 4 h under N₂ in reaction tubes.

Sb reacted with phosphonium salts **2a** and **2c**, affording products **6a** and **6b**, respectively. However, the C2-unsubstituted substrate, 5-phenyl pyrimidine *N*-oxide, did not produce any coupling product under the current reaction conditions (data not shown). Moreover, pyridazine *N*-oxide **5c** also participated in the coupling reaction to give **6c**, albeit with slightly decreased reactivity. Phthalazine *N*-oxide **(5d)** was also coupled with **2a** in the presence of KHMDS as a base to furnish phthalazine derivative **6d** in 55% yield. To investigate the relative reactivity of pyrazine *N*-oxides versus pyrimidine *N*-oxides, a competitive intermolecular experiment was performed (see the Supporting Information for details). Exposure of **2a** to equimolar quantities of **1c** and **5a** under otherwise identical conditions provided a mixture of **3g** (60%) and **6a** (31%).

This protocol allows late-stage functionalization and sequential alkylation of complex diazine molecules bearing an N-oxide group (Scheme 6). For example, N-oxide derivative 7a, derived from varenicline, a commercial smoking cessation agent, was selectively reacted with 2a to provide desired product 7b (32%), with recovery of starting material 7a (40%).

Scheme 6. Late-Stage Functionalization and Sequential Alkylations



Meanwhile, the sequential alkylation of pyrazine N-oxide 8a furnished 2,3-bis-alkylated pyrazine adduct 8d.

To highlight the utility of this protocol, we first report the total synthesis of racemic paenibacillin A, isolated from *Paenibacillus* sp. XY-2 in 2015,¹⁴ based on the sequential C– H alkylation of pyrazine N-oxides (Scheme 7). Treatment of 2-

Scheme 7. Total Synthesis of Racemic Paenibacillin A



methoxypyrazine *N*-oxide $(9a)^{15}$ with 2d under the developed conditions provided C6-isopropylated pyrazine 9b in 43% yield, which was subsequently oxidized by *m*-CPBA to afford pyrazine *N*-oxide 9c in 72% yield. Site-selective alkylation of 9c was performed using *s*-butylPPh₃Br (2j), furnishing 3,6dialkylated pyrazine 9d (31%), with recovery of starting material 9c (41%). Hydrolysis of 9d with concentrated HCl gave racemic paenibacillin A in 80% yield. The spectroscopic data for paenibacillin A were in full agreement with the reported literature values.

To illustrate the applicability of the alkylated diazines, we performed a series of synthetic transformations, as shown in Scheme 8. Benzylic bromination of 3k and subsequent amination using imidazole afforded 10a in 70% yield, which is recognized as a pivotal scaffold of a Cushing's syndrome agent (eq 1).¹⁶ The Ni-catalyzed C–H functionalization of 3k with benzyl alcohol gave alkylated product 10b (49%) and olefinated product 10c (12%) (eq 2). Finally, direct C–H

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Scheme 8. Synthetic Transformations of Alkylated Diazines



arylation of methoxymethyl pyrazine **3g** was performed to furnish **10d** in 27% yield (eq 3).

In conclusion, a highly efficient protocol for the formation of alkylated diazine derivatives via the metal-free site-selective C– H alkylation of diazine N-oxides using phosphonium ylides was developed. The method affords a wide substrate scope, high site selectivity, and broad functional group tolerance. In particular, the sequential C–H alkylation of diazine N-oxides and gram-scale reaction demonstrate the utility of the developed method. Furthermore, the late-stage C–H functionalization of a complex molecule and total synthesis of a natural product highlight the great potential of this method for application in drug discovery.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.or-glett.9b02365.

Experimental procedures, characterization data, and ¹H, ¹³C, and ¹⁹F NMR spectra for all compounds (PDF)

AUTHOR INFORMATION

Corresponding Authors

*E-mail: sjchung@skku.edu.

*E-mail: insukim@skku.edu.

ORCID 💿

Sang J. Chung: 0000-0002-3501-212X In Su Kim: 0000-0002-2665-9431

Author Contributions

[†]P.G. and N.Y.K. contributed equally to this work.

Notes

The authors declare no competing financial interest.

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