

ANGULAR DISTRIBUTIONS OF SULFUR ATOMS IN THE $3p^3P_J$ and $3p^1D$ STATES FROM TWO-PHOTON DISSOCIATION OF CARBON DISULFIDE

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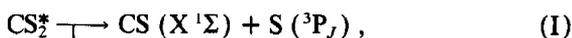
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Received 26 October 1987; in final form 8 January 1988

A molecular beam of carbon disulfide was photodissociated at 288–292 and 308–309 nm (a one-color, multiphoton dissociation). The angular distributions of sulfur atoms were measured in the S ($3p^3P_J$) and ($3p^1D_2$) states. Both S (1D) and (3P_J) fragments have anisotropic distributions for the two-photon dissociation.

1. Introduction

CS₂ is one of the triatomic molecules whose photodissociation dynamics have been studied by time-of-flight mass spectroscopy [1,2], laser-induced fluorescence [3], VUV absorption [4] and also diode laser absorption [5] techniques. It has been recognized that there are two primary processes for one-photon photodissociation of CS₂ in the UV region. One is the spin-allowed process from the optically prepared state; the other is the spin-forbidden process which becomes partially allowed by large spin-orbit interactions of two sulfur atoms in CS₂:



When the photon energy is increased, CS can be electronically excited. At ≈ 8 eV the intense Rydberg features of CS₂ lying to shorter wavelengths correlate directly with CS ($a^3\Pi$) + S (3P) and once the threshold at 158 nm has been crossed it is in this region that the quantum yield of CS ($a^3\Pi$) rises steeply

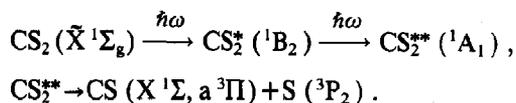
in the one-photon dissociation of CS₂ [6]. The energetics of CS₂ dissociation are shown in table 1. There are several dissociation pathways which generate S fragments in the 3P_J , 1D_2 , and 1S_0 states. These pathways can be attained by two-photon absorption of UV laser light via a real intermediate state of CS₂. The two-photon dissociation at 308 nm (4.0 eV) via the V system of CS₂ (1B_2) has been reported for the translational energy and angular distributions of the sulfur photofragments in the S (3P_2) state [8]:

Table 1
Threshold energies (eV) of primary photodissociative processes for CS₂^{a)}

CS	S		
	3P_2	1D_2	1S_0
X $^1\Sigma$	4.46(I)	5.61(II)	7.21(IV)
a $^3\Pi$	7.88(III)	9.03	10.63
a' $^3\Sigma^+$	8.32	9.47	
d $^3\Delta$	8.88	10.03	
e $^3\Sigma^-$	9.25	10.40	
A $^1\Pi$	9.27		
$\Delta E(^3P_1 - ^3P_2) = 0.049$ eV			
$\Delta E(^3P_0 - ^3P_2) = 0.071$ eV			

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^{a)} Ref. [7]. Numbers in parentheses denote the primary photodissociation processes, see text.



The angular distribution of S (1D) has not been reported although the formation of S (1D) is confirmed in the VUV one-photon dissociation [9]. If the dissociating species is the same for both S (3P_2) and S (1D) fragments, the angular distributions would have the same anisotropic parameters for both fragments. In this paper, we report on measurement of angular distributions of S (3P_J) and S (1D) generated in the two-photon dissociation via the 1B_2 state since the β values contain information on lifetimes, molecular structures and symmetries of dissociating states.

2. Experimental

The sulfur atomic photofragments were generated and ionized by one-color multiphoton dissociation and ionization. The angular distributions of ions were measured by a homemade mass spectrometer that has theoretically a perfect focusing property as has been described elsewhere [10,11]. Briefly, the perfect focusing property was realized by the combination of a uniform magnetic field with a three-dimensional quadrupole electrostatic field. The ions of a certain mass number generated in one focusing point of the electrostatic field reach the other focusing point irrespective of their initial velocity vectors. For mass analysis, the mass number was selected by changing the strength of the electric field. The detector has a slit for spatial resolution. The angular distribution of the photofragments was measured by changing the laboratory angle θ between the directions of the detector axis and the electric vector of the dye laser light (Lambda Physik FL 2002) with mass number fixed at $m/e=32$. The effusive molecular beam of CS_2 was introduced through a needle (0.3 mm diameter) near the focusing point where the laser beam was focused by a lens of $f=15$ cm. The maximum laser intensity is estimated ≈ 30 GW/cm 2 assuming a Gaussian profile of the laser beam. The polarization of the laser beam was changed from linearly polarized to circularly polarized by retardation plates.

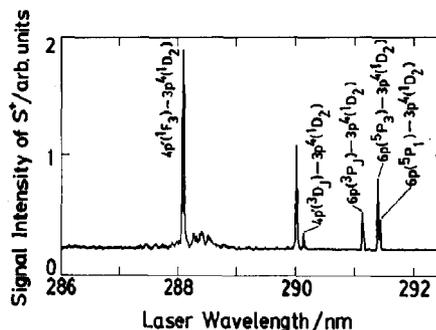
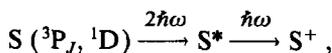


Fig. 1. Multiphoton-ionization spectrum of S atoms generated in multiphoton dissociation of CS_2 at 286–292 nm. Table 2 shows assignments of the two-photon transitions of sulfur atoms.

3. Results

A photon of wavelength longer than 277.7 nm contains insufficient energy to dissociate a CS_2 molecule. As the dye laser was scanned over the two wavelength ranges, $\lambda=288$ –292 nm and 308–311 nm, sulfur atoms were generated via two-photon dissociation of CS_2 . Successive resonance-enhanced photoionization (MPI) in sulfur photofragments occurred by the 2+1 three-photon (two to resonance) ionization in the 3^3P_J and 3^1D_2 states as shown in figs. 1 and 2. These photofragments can be generated by two-photon absorption of the same laser photons:



where S^* denotes an electronically excited sulfur atom. At $\lambda=288$ –292 nm, the MPI spectrum of fig.

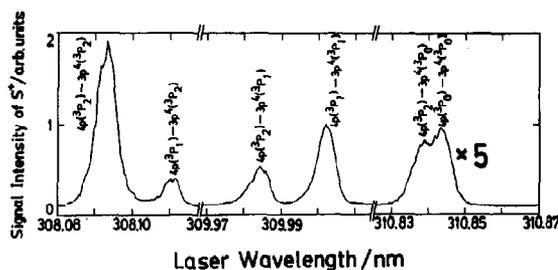


Fig. 2. Multiphoton-ionization spectrum of S atoms generated in multiphoton dissociation of CS_2 at 308–311 nm. Table 3 shows relative intensities of J'' levels in S ($3p^3P_{J''}$).

Table 2

Assignments of two-photon transitions of S atoms in two-photon dissociation of CS₂ with anisotropy parameters

λ (nm)		Two-photon transition	$\beta_2^{a)}$	$n^{b)}$	$\Omega^{c)}$
air	vacuum				
288.11	288.19	$4^1F_3-3^1D_2$	0.88 ± 0.07	3.1 ± 0.1	0.5
290.02	290.10	? -3^1S_0	1.13 ± 0.16	5.1 ± 0.1	0.2
291.14	291.22	$6^3P_0-3^1D_2$	0.81 ± 0.08	3.9 ± 0.1	0.6
291.39	291.47	$6^5P_3-3^1D_2$	0.88 ± 0.08	3.4 ± 0.1	0.6
291.40	291.49	$6^5P_1-3^1D_2$	0.88 ± 0.08	3.3 ± 0.1	0.3
308.10	308.20	$4^3P_2-3^3P_2$	0.54 ± 0.07	3.0 ± 0.1	-
310.01	310.10	$4^3P_2-3^3P_1$	0.52 ± 0.07	2.8 ± 0.1	-
310.84	310.94	$4^3P_2-3^3P_0$	0.52 ± 0.07	3.2 ± 0.1	-
308 ^{d)}		$4^3P_2-3^3P_2$	0.67 ± 0.12		

^{a)} Anisotropy parameters in eq. (1). β_4 values were found to be zero.

^{b)} Power index n is defined by $I_{\text{ion}} \propto (I_{\text{laser}})^n$.

^{c)} Intensity ratio I_c/I_l where I_c and I_l are the signal intensity for circularly and linearly polarized laser light respectively.

^{d)} Untuned XeCl excimer laser light [8]. S (3P_2) is generated by primary process (I).

I is assigned to the transitions from S (3^1D) except for the line at 290.02 nm. The assignments of the two-photon transitions of S atoms are summarized in table 2. The MPI excitation spectra at 308–311 nm were observed in more detail for the transition of $4^3P_2 \leftarrow 3^3P_J$. The relative population of the J'' levels were obtained using the two-photon absorption cross sections reported by Brewer et al. [12]. The results are summarized in table 3. The prior expectation for this branching ratio [13] is given by

$$g(3^3P_J)[E_{\text{AVL}}(3^3P_J)]^{5/2}/g(3^3P_{J'})[E_{\text{AVL}}(3^3P_{J'})]^{5/2},$$

which is approximated by $(2J+1)/(2J'+1)$, while the ratios obtained experimentally are only 1.2–1.4. The J'' population distributions obtained are not statistical. This result suggests that the formation of the J'' levels may not be a statistical process like the

predissociation via the hot CS₂ molecule.

By choosing appropriate wavelengths for S (3P_J , 1D), the intensity of S⁺ was measured as a function of θ . An example is shown in fig. 3. In these experiments, the laser power was reduced until the β values reached asymptotic values in order to avoid the reduction of the β values caused by saturation effects. The experimentally obtained angular distribution is close to the one obtained in the center-of-mass angular distribution $f(\theta)$ that is given by the following equation;

$$f(\theta) \propto 1 + \beta_2 P_2(\cos \theta) + \beta_4 P_4(\cos \theta), \quad (1)$$

where $P_l(\cos \theta)$ is the l th degree Legendre polynomial [14]. The curve drawn through the set of data of

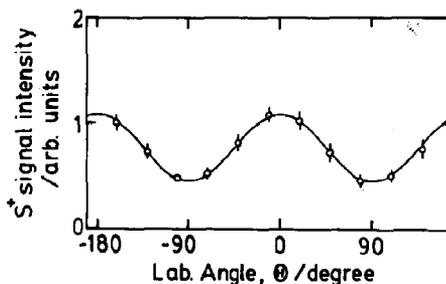


Fig. 3. Angular distribution of S (1D) photofragments from CS₂ multiphoton dissociation at 288.10 nm. The solid line is the least-squares fit of the data with eq. (1) for $\beta_2=0.88$ and $\beta_4=0$.

Table 3

Relative intensity of S (3P_J)

Two-photon transition	Relative signal intensity, I	$I/\sigma^{a)}$
$4^3P_2-3^3P_2$	13.6	1.2
$4^3P_2-3^3P_1$	3.3	1.4
$4^3P_2-3^3P_0$	1	1

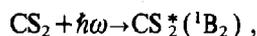
^{a)} Two-photon absorption cross section σ is adopted from ref. [12].

fig. 3 shows the least-squares fit of eq. (1) to the data. Other sets of data were also analyzed by this method. The β_2 values were positive while the β_4 values were found to be zero for all sets of data. β_2 values obtained are summarized in table 2. The β_2 value for S (3P_2) generated in the 308 nm photodissociation has been previously reported to be 0.67 by the MPI TOF method [8]. Compared with fig. 6 of ref. [8], fig. 3 of the present work is much improved in the S/N ratios because of the high sensitivity of our perfect focusing mass spectrometer. Although there is some discrepancy between the previously reported value of $\beta_2=0.67\pm 0.12$ and the present work of $\beta_2=(0.52-0.54)\pm 0.07$, it is safe to say that the β_2 values of S (3P_J) are ≈ 0.6 for the two-photon dissociation of CS₂ at 308–318 nm and less than those (≈ 0.9) of S (1D) obtained in the two-photon dissociation of CS₂ at 288–292 nm.

4. Discussion

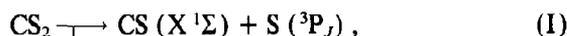
The intense and complicated absorption system of CS₂ in the region of 290–330 nm is called the V system. This V system is due to the transition from the ground $^1\Sigma_g^+$ state to the 1B_2 state ($\angle SCS=131^\circ$) that fluoresces with unit quantum yield [15]. Douglas and Milton [16] have studied the pronounced Zeeman effect of the absorption bands ($^1B_2 \leftarrow \bar{X}^1\Sigma_g^+$) of CS₂ and have shown that it can be understood if the observed 1B_2 state is a component of a 3A_2 state. The lowest 3A_2 state was detected by the technique of resonance-enhanced MPI at 361–371 nm [17]. This singlet-triplet transition is reported to have an oscillator strength of $\approx 10^{-7}$ [18]. Thus, the singlet-triplet coupling seems to be weak in CS₂ at ≈ 365 nm. However, Kanamori and Hirota [5] have reported that the spin-forbidden triplet channel (I) makes a significant contribution to the one-photon dissociation of CS₂ at 193 nm. The branching ratio of channels (I) and (II) is reported to be 1–0.25 [1,2,5] although the prior distribution [13] predicts that the ratio is 14.4. The intersystem crossing process seems to be a bottleneck one for yielding the S (3P_J) fragments. Although it was difficult to determine this ratio in the present work, the signal intensities of both S (3P) and S (1D) were comparable in the present experiment.

In the present multi-photon dissociations in the UV region, the β_4 values of eq. (1) were found to be zero as described above. This result implies that the dissociation is due to the two-photon absorption of CS₂ via the real intermediate 1B_2 state:



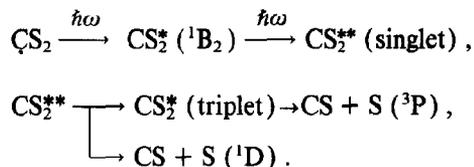
In the two-photon dissociation via a real and stable state the angular distribution must be represented by eq. (1) with $\beta_4=0$. If CS₂ absorbed three UV photons, β_4 would not be zero because the final two-photon absorption occurs via a real and dissociative intermediate state. Kawasaki et al. [8] have reported that β_4 was not zero or the $P_4(\cos\theta)$ term contributed to $f(\theta)$ for the two-photon dissociation of NO₂ via the real and dissociative intermediate state.

The two-photon energies for the photolysis at ≈ 290 and ≈ 308 nm are 8.5 and 8.0 eV, respectively. Table 1 shows that the following primary processes are energetically possible,



In the one-photon photolysis in the UV region, processes (I), (II), and (III) have been reported [1–6]. We have previously reported that the contribution of the spin-allowed process (III) is comparable to that of the spin-forbidden process (I), based on the TOF spectrum of the S (3P) fragments from two-photon dissociation at 308 nm [8]. The spin-forbidden process (I) contributes to the S (3P_2) formation through the bottlenecking intersystem crossing to the triplet repulsive state from the photoprepared singlet state as has been reported for the one-photon dissociation [1,2]. Elementary kinetics considerations will show that while a bottleneck such as that invoked will affect the yields of the various channels, the production rate of every species will be the same, determined by the sum of the rates through all channels. The rate can be estimated from the β values, molecular symmetries, and structures as will be described later. In brief, the smaller decay rate, the smaller is the β value if the molecular structure is not changed. In the pres-

ent work the β_2 values for S (3P_J) are smaller than those for S (1D_2) as shown in table 2. If a single photo-prepared state is the originating excited CS₂ state for S (3P) and S (1D) production, then the only way for the β values of S (3P) to differ from those of S (1D) is for the molecular structure of the triplet dissociating state to differ from that of the singlet one:



Concerning the symmetries of the dissociating states of CS₂, since the first one-photon transition is $^1\text{B}_2 \leftarrow \tilde{\text{X}}^1\text{A}_1$ in C_{2v} approximation, the symmetry-allowed states for the second one-photon absorption are A₁, A₂, and B₂. If the singlet dissociating state retains the structure of the intermediate $^1\text{B}_2$ state ($\angle \text{CSC} = 131^\circ$), only the $^1\text{A}_1 \leftarrow ^1\text{B}_2 \leftarrow \tilde{\text{X}}^1\text{A}_1$ transition gives a positive β_2 value according to the following equation [19]:

$$\beta_2 = 2P_2(\cos \chi) \frac{1 + \omega^2 \tau^2}{1 + 4\omega^2 \tau^2}, \quad (2)$$

where χ is the angle between the transition moment and the dissociation direction, ω the angular velocity of the parent molecule, and τ the dissociation lifetime. When the second transition moment lies in the S-S direction, χ is 24.5° and β_2 is 1.48. The lifetime estimated is ≈ 0.3 ps for the S (1D) channel using eq. (2) with β_2 values in table 2 and $\omega = 2 \times 10^{12} \text{ s}^{-1}$. The value of ω is estimated from the relation $\omega = (\pi kT/2I)^{1/2}$, where I is the average of the two large, nearly equal moments of inertia in the $^1\text{B}_2$ state of CS₂.

At 290.02 nm excitation, the β values are larger than those for S (1D). The power dependence is larger than others. In addition the intensity ratio I_c/I_q is also different from that for S (1D_2) where I_c is the MPI signal intensity for circularly polarized laser light and I_q for linearly polarized one. The atomic transition is not from S (1D_2) nor is clear even from the reported Grotrian energy level diagram [20]. The

transition at 290.02 nm may be from a singlet state, probably S (3^1S_0) generated from process (IV), because the β value is larger than for S (3P_J) and close to the singlet channel (II).

Acknowledgement

This work is partly supported by a Grant-in-Aid for Scientific Research from the Ministry of Education, Science, and Culture of Japan.

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