Total Synthesis of Luminacin D

J. Brad Shotwell,^{†,‡} Evan S. Krygowski,[†] John Hines,[‡] Brian Koh,[‡] Elliott W. D. Huntsman,[†] Hui Won Choi,[†] John S. Schneekloth, Jr.,^{†,‡} John L. Wood,^{*,†} and Craig M. Crews^{*,†,‡,§}

Sterling Chemistry Laboratory, Department of Chemistry, Department of Pharmacology, and Department of Molecular, Cellular, and Developmental Biology, Yale University, New Haven, Connecticut 06520

craig.crews@yale.edu

Received June 18, 2002

ORGANIC LETTERS

2002 Vol. 4, No. 18 3087–3089

ABSTRACT





A highly convergent synthesis of the angiogenesis inhibitor luminacin D has been achieved in 13 linear steps (19 steps total, 5.3% overall yield) utilizing a samarium(II) iodide-mediated mixed tandem aldol/Evans–Tishchenko reaction to construct the carbohydrate precursor. The modular synthetic design will allow derivatization at key positions necessary for biochemical mode of action studies.

A growing class of anticancer pharmaceuticals target angiogenesis, the process of new vascularization by which growing tumors establish a blood supply.¹ Despite an emerging understanding of the complex and highly regulated processes leading to angiogenesis, numerous small molecules currently under investigation have poorly understood mechanisms of action.² The luminacins (Figure 1), isolated recently by Wakabayashi and co-workers from the actinomycete *Streptomyces* sp., represent a potent class of antiangiogenic agents that elicit their biological response via an unconfirmed intracellular target.^{3,4} Herein, we report efficient synthetic access to luminacin D by means amenable to the future derivatization necessary for the identification of its cellular binding protein(s).

- [†] Department of Chemistry.
- [‡] Department of Molecular, Cellular, and Developmental Biology.

Interested in identifying any novel intracellular effectors that elicit an antiangiogenic response, we planned a versatile synthetic approach that could be readily extended to access luminacin-like biochemical probes and chemical analogues.⁵ Ultimately, a strategy involving construction of the C1'-C2' bond by nucleophilic addition of carbohydrate precursor **5** to aldehyde **4** evolved. We found that introduction of the





[§] Department of Pharmacology.

^{(1) (}a) Kerbel, R. S. Carcinogenesis, **2000**, 21, 505–515. (b) Oehler, MK.; Bicknell, R. Br. J. Cancer **2000**, 82, 749–752.

⁽²⁾ Deplanque, G.; Harris, A. L. Eur. J. Cancer 2000, 36, 1713–1724.
(3) (a) Naruse, N.; Kageyama-Kawase, R.; Funahashi, Y.; Wakabayashi, T. J. Antibiot. 2000, 53, 579–590. (b) Wakabayashi, T.; Kageyama-Kawase, R.; Naruse, N.; Funahashi, Y.; Yoshimatsu, K. J. Antibiot. 2000, 53, 591–596. (c) Hata-Sugi, N.; Kawase-Kageyama, K.; Wakabayashi, T. Biol. Pharm. Bull. 2002, 25(4), 446–451.

C2'-C11' bond as an unsaturated precursor eliminated problematic retro-aldol-type fragmentation of advanced intermediates saturated at C2'-C11' and delayed introduction of the C2' stereocenter.⁶ Our strategy called for late-stage introduction of the sensitive spiro-epoxypyranose functionality and allowed for installation of the alkyl residues at C2', C8', and C5 by means amenable to future derivatization.

The requisite aryl electrophile (4) was prepared efficiently from known triol **6** (Scheme 1).⁷ While exposure of **6** to



standard bromination conditions led invariably to overhalogenated species, iodination at the more activated position proceeded cleanly to **7**. Selective benzylation of the phenolic hydroxyls and silylation of the benzylic alcohol afforded **8** (82% yield, two steps). Stille cross-coupling with (tributyl)*iso*butenylstannane in the presence of catalytic palladium-(II) furnished **9**. Finally, two-step adjustment of the ester oxidation state gave **4**.

Construction of the carbohydrate precursor began with known vinyl iodide 10^8 , which was elaborated to α -bromo ketone 11 via the protection, acylation, α -bromination sequence illustrated in Scheme 2. Formation of the corresponding samarium enolate by treatment of 11 with 2 equiv of samarium iodide in THF followed by sequential addition

⁽⁵⁾ Tatsuta and co-workers have recently reported synthetic access to luminacin C1 (1) and C2 (2). Beginning from L-glucal, their strategy involves 36 linear steps (43 steps total) and has served to establish both the relative and absolute stereochemistry for this class of natural products: Tatsuta, K.; Nakano, S.; Narazaki, F.; Nakamura, Y. *Tetrahedron Lett.* 2001, 42, 7625–7628.



⁽⁷⁾ Saimoto, H.; Yoshida, K.; Murakomi, T.; Marimoto, M.; Sashiwa, H.; Shigemosa, Y. J. Org. Chem. **1996**, *61*, 6768–6769



of (*E*)-2-bromo-2-pentenal⁹ and acetaldehyde furnished, after basic workup, diol **13** as a single diastereomeric product in excellent yield. Analogy to previous work suggests that the high degree of stereo-induction arises from an organized eight-membered ring chelate (**12**).¹⁰ Protection of **13** as the corresponding acetonide furnished **5**.¹¹

Tandem aldol-Evans—Tishchenko-type additions catalyzed by divalent and/or trivalent samarium species have been previously studied,¹² and variants catalyzed by other metals are also known.¹³ However, the smooth condensation of two different aldehydes via sequential addition greatly enhances the scope of this samarium-mediated transformation.

Metal-halogen exchange of vinyl bromide **5** with *t*butyllithium at -78 °C in diethyl ether followed by rapid addition of a chilled solution of **4** in pentane gave a mixture of diastereomeric alcohols that were simultaneously oxidized by Dess-Martin periodinane. Subsequent bis-desilylation cleanly afforded **14** (Scheme 3). Initial plans called for the bis-allylic oxidation of **14** and subsequent acetal removal to afford compounds similar to **17**. We anticipated that epoxidation of **17** would be axially directed by the resident hydroxyl.¹⁴ Unfortunately, attempted cyclization of the requisite bis-aldehyde proved to be frustrating, with apparent scrambling of the C6'-C8' olefin geometry and dehydration always predominating. Fortunately, installation of the epoxide prior to cyclization effectively eliminated these problems.

Although highly diastereoselective epoxidations of primary allylic alcohols have been observed,¹⁵ a model for the observed diastereoselectivity has not been forthcoming. A careful screening of a variety of epoxidation conditions in the present case revealed that VO(acac)₂/TBHP afforded **15**

⁽⁴⁾ Recently, luminacin C2 was isolated in a screen for Src kinase inhibitors. In vitro experiments suggest that it elicits some of its biological effects via disruption of SH3-mediated association of any number of intracellular proteins with Src. See: (a) Sharma, S.; Oneyama, C.; Yamashita, Y.; Nakano, H.; Sugawara, K.; Hamada, M.; Kosaka, N.; Tamaoki, T. *Oncogene* **2001**, *20*, 2068–2079. (b) Oneyama, C.; Nakano, H.; Sharma, S. *Oncogene* **2002**, *21*, 2037–2050.

⁽⁸⁾ Rossi, R.; Carpita, A.; Cossi, P. Synth. Comm. 1993, 23(2), 143-152.

⁽⁹⁾ Lütjens, H.; Knochel, P. *Tetrahedron Lett.* **1994**, *5*(7), 1161–1162.
(10) Evans, D. A.; Hoveyda, A. H. J. Am. Chem. Soc. **1990**, *112*, 6447–6449.



in the best combined regio- and diastereoselectivity (50% isolated yield). The surprising ease with which the resulting diastereomers could be separated and the good efficiency with which **14** could be accessed (nine steps, 33% overall yield) allowed progress to continue toward luminacin D.

Bis-allylic oxidation of **15** with Dess–Martin periodinane and subsequent treatment with a warm solution of acetic acid in THF/H₂O gave the fully functionalized luminacin D spiroepoxypyranose (**18**) in good yield (Scheme 4). Global



compound	IC_{50}
(±)-luminacin D 19 18	$\begin{array}{l} {\rm 4.5 \pm 0.7 \ \mu M} \\ {\rm 4.9 \pm 0.6 \ \mu M} \\ {\rm 3.4 \pm 0.9 \ \mu M} \end{array}$

^{*a*} Proliferating BAE cells were incubated with vehicle or drug at 37.5 °C for 20 h; [³H]-thymidine was added and the incubation continued for an additional 4 h. Relative [³H]-thymidine incorporation into newly synthesized DNA was quantitated.

reduction of **18** proceeded through **19** under 1 atm of H_2 in the presence of 5% Pd/C to give luminacin D as an epimeric mixture at C2' favoring the natural configuration (ca. 1.5: 1). Luminacin D (**3**) and its C2' epimer (**20**) are separable by preparative thin-layer chromatography.

With a variety of synthetic intermediates in hand, we investigated their antiproliferative profiles. Luminacin D dose dependently inhibited bovine aortic endothelial (BAE) cell proliferation as determined by measuring the incorporation of [³H]-thymidine into cellular DNA. Our results revealed that luminacin D possessed an IC₅₀ = $4.5 \pm 0.7 \,\mu$ M, while its immediate synthetic precursors (**18** and **19**) seemed equally potent (Table 1).

An efficient and versatile total synthesis of luminacin D has been achieved (13 linear steps, 5.3% overall yield). The antiproliferative profiles of several synthetic intermediates provide evidence that the C2' alkyl chain and the free phenols may tolerate modification without a detrimental impact on the overall potency of the derivatives. Current synthetic efforts are aimed at generating biotinylated affinity reagents at these positions. Studies directed toward a molecular mode of action for this class of natural products remain our principle aim.

Acknowledgment. We thank Esai Co., Ltd., for providing us with natural luminacin D. J.B.S. thanks the NSF, The American Chemical Society Division of Medicinal Chemistry, and Pfizer, Inc., for graduate fellowships. C.M.C. acknowledges the NIH (GM 62120) for financial support. J.L.W. acknowledges Bristol-Myers Squibb, Eli Lilly, GlaxoSmithKline, Yamanouchi, and AstraZeneca for financial support through their Faculty Awards Programs and the Henry Dreyfus Foundation for a Teacher–Scholar Award.

Supporting Information Available: Experimental and spectral data pertaining to all new compounds (3-11, 13-16, 18-21). This material is available free of charge via the Internet at http://pubs.acs.org.

OL026382Q

^{(11) &}lt;sup>13</sup>C NMR for 5 is consistent with a twist-chair conformation characteristic of an anti-acetonide; see: Rychnovsky, S. D.; Rogers, B. N.; Richardson, T. I. Acc. Chem. Res. **1998**, *31*, 9–17.
(12) (a) Lu, L.; Chang, H.; Fang, J. J. Org. Chem. **1999**, *64*, 843–853.

^{(12) (}a) Lu, L.; Chang, H.; Fang, J. J. Org. Chem. 1999, 64, 843–853.
(b) Hsu, J.; Fang, J. J. Org. Chem. 2001, 66, 8573–8584.

⁽¹³⁾ For a recent example, see: Mascarehnah, C. M.; Miller, S. P.; White, P. S.; Morken, J. P. Angew. Chem., Int. Ed. 2001, 40(3), 601–603.

^{(14) (}a) Vedejs, E.; Dent, W. J. Am. Chem. Soc. 1989, 111, 6861–6862.
(b) Barton, D.; Bath, S.; Billington, D.; Gero, S.; Quiclet-Sire, B.; Samadi, M. J. Chem. Soc., Perkin Trans. 1. 1995, 1551–1558.

⁽¹⁵⁾ Hoveyda, A.; Evans, D.; Fu, G. *Chem. Rev.* **1993**, *93*, 1307–1370. (In the present case, application of *m*CPBA favors **16** by 11:1, while Ti- $(iPrO)_4$ favors **15** by 2:1.)