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3,8-Diazabicyclo[3.2.1]octan-2-one Peptide Mimetics: Synthesis of a Conformationally Restricted Inhibitor of Farnesyltransferase[†]

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ABSTRACT

A new synthesis of the 3,8-diazabicyclo[3.2.1]octan-2-one framework is described. Transannular enolate alkylation of piperazinone derivatives provides a flexible route to highly constrained bicyclic peptidomimetic synthons with substitution at the $C\alpha$ position. The chemistry was used to produce a conformationally constrained farnesyltransferase inhibitor, which aided the elucidation of enzyme-bound conformation.

The manipulation of amino acid conformation is an important strategy in the chemical investigation of biological systems. Conformationally constrained amino acid segments have been important tools for drug design in support of understanding and optimizing interactions between ligands and macromolecules. Methods of affixing desired dihedral angles within the backbone of a peptide segment are numerous, la-c and many employ the Freidinger strategy of incorporating ring-induced constraints.

Piperazinones 2 function as peptidomimetics wherein the N_i and N_{i-1} positions of an amino acid backbone fragment 1 are linked by an ethylene unit. This ring constraint results

in restriction of the ϕ_i , ψ_i , and ω_i torsion angles.³ Amino group acylation induces an A(1,3)-strain-enforced pseudo-axial position of the C α side chain (R²) substituent, further restricting the conformation.⁴ A more rigid variant of **2** with better defined structure is achieved by the incorporation of a transannular bridge. Thus, the 3,8-diazabicyclo[3.2.1]octan-2-one compound **3** is a piperazinone—proline hybrid in which the side chain (R²) substituent is forced to occupy a pseudoequatorial position with respect to the piperazinone ring by virtue of the diaxially disposed ethano bridge. Since chair—chair interconversion is prevented, the peptide backbone torsion angles are severely constrained. While methods

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exist for the construction of the unsubstituted scaffold (3, $R^2 = H$) from pyroglutamate derivatives,⁵ most of these syntheses have provided racemic material,^{5b-f} and none features the incorporation of a $C\alpha$ side chain substituent. In this report we describe a versatile synthetic approach to 3 that enables facile substitution at the $C\alpha$ position. We also discuss the synthesis and biological characterization of a novel 3,8-diazabicyclo-[3.2.1]octan-2-one farnesyltransferase inhibitor that provides insight into the enzyme-bound conformation of related flexible inhibitors (vide infra).

The synthesis is based on the preparation of an appropriately substituted piperazinone^{3,6} followed by a pivotal transannular alkylation reaction⁷ to install the [3.2.1] bicyclic ring system (Scheme 1). Boc-(S)-allylglycine **4** was con-

^a Reagents and conditions: (Dmb = 2,4-dimethoxybenzyl) (a) MeO(Me)NH·HCl, EDC, HOBt, DMF, 0 °C, 24 h, 67%. (b) LiAlH₄, Et₂O, $-50 \rightarrow 5$ °C, 3 h, 85%. (c) 2,4-(MeO)BnNH₂, Na(AcO)₃BH, 4 Å MS, ClCH₂CH₂Cl, 16 h, 90%. (d) ClCH₂COCl, NaHCO₃, EtOAc−H₂O, 0 °C, 30 min, 93%. (e) Cs₂CO₃, DMF, 65 °C, 16 h, 75%. (f) OsO₄, NMO, *t*-BuOH, THF, H₂O, 7 h; NaIO₄, NaHCO₃, 1.5 h, 96% (g) NaBH₄, EtOH, 0 °C → rt, 1 h, 92%. (h) PhSO₂Cl, Et₃N, CH₂Cl₂, 0 °C → rt, 1 h, 83%. (i) LiHMDS, THF, $-78 \rightarrow 0$ °C, 40 min, 86%.

verted to the corresponding aldehyde 5^8 and then used to reductively alkylate 2,4-dimethoxybenzylamine to give 6. Chloroacetylation and base-promoted cyclization provided the 5-allyl-2-piperazinone 7. Refunctionalization of the olefin in 7 was accomplished by oxidative cleavage, aldehyde reduction, and conversion to the benzenesulfonate 8. Finally, after treatment of 8 with LiHMDS in THF at -78 °C and warming the reaction to 0 °C, an intramolecular enolate alkylation occurred to afford the bicyclic framework 9a in 86% yield. 9,10

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An amino acid $C\alpha$ side chain substituent was easily introduced into the bicyclic framework from the piperazinone intermediate 7, providing a method for the preparation of a constrained phenylalanine isostere. ¹¹ Thus, alkylation of 7 with benzyl bromide (Scheme 2) provided the *trans*-

Scheme 2^a

^a Reagents and conditions: (a) LiHMDS, THF, -78 °C; BnBr, -78 °C, 3.5 h, 74% borsm. (b) OsO₄, NMO, *t*-BuOH, THF, H₂O, 2.5 h; NaIO₄, NaHCO₃, 2 h. (c) NaBH₄, EtOH, 0 °C, 20 min, 70% two steps. (d) PhSO₂Cl, Et₃N, CH₂Cl₂, 0 °C, 1 h, 90%. (e) LDA, THF, $-78 \rightarrow 0$ °C, 2 h, 63%.

disubstituted compound 10 (diastereoselectivity \geq 95:5). Smooth conversion as before to the benzenesulfonate 11, followed by cyclization with strong base (LDA, THF, -78 to 0 °C), ¹² afforded the constrained bicyclic analogue 9b.

The intermediate **9a**, prepared above, was used to synthesize a conformationally restricted inhibitor of farnesyltransferase (FTase). FTase is an important posttranslational processing enzyme that prenylates proteins and enables the participation of some in signal transduction during cell proliferation. ¹³ Inhibitors of this enzyme (FTI) are promising antitumor agents, and several are currently being evaluated in human clinical trials. ¹⁴ During our investigations of peptidomimetic 1-aryl-2-piperazinone FTIs (e.g., **12**, Figure

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(12) Attempted cyclization using LiHMDS (THF, -78 to 0 °C, 5 h) gave only recovered 11. The use of KHMDS under the same conditions resulted in gradual conversion to the dihydrooxazinone *iii*.

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866 Org. Lett., Vol. 3, No. 6, 2001

⁽⁹⁾ Compound **9a** was subjected to Boc-deprotection (HCl, EtOAc, 0 °C), conversion to the (+)- and (-)-10-camphorsulfonamides (10-camphorsulfonyl chloride, Et₃N, DMF), and HPLC analysis to confirm \geq 97% enantiomer purity.

⁽¹⁰⁾ Transannular alkylation of the methanesulfonate i under the same conditions provided an 8:1 mixture of 9a and the alcohol derived from cleavage of the methanesulfonyl group. Attempts to cyclize the sulfonium salt ii derived from L-methionine were unsuccessful.

1),¹⁵ we discovered that they adopt a folded conformation when bound to the enzyme and that a macrocyclic constraint (e.g., 13) enhances potency by stabilizing the bound con-

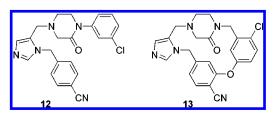
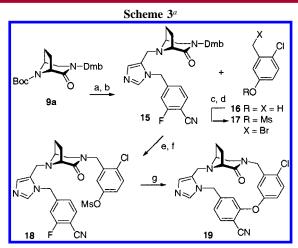


Figure 1. Piperazinone farnesyltransferase inhibitors 12 and 13.

formation. ¹⁶ The importance of the carbonyl group position for FTase inhibitory activity was also established. ¹⁷ We sought to further elucidate enzyme-bound conformations of these compounds with respect to the position of the carbonyl group. In principle, this could be accomplished by constraining the conformational flexibility of the piperazinone ring and inhibiting the positional mobility of the carbonyl group. On the basis of these considerations, a 3,8-diazabicyclo-[3.2.1]octan-2-one analogue of the macrocycle **13** was selected as a target for synthesis.

The intermediate **9a** was deprotected and reductively alkylated with 1-(3-fluoro-4-cyanobenzyl)imidazole-5-car-boxaldehyde (**14**)¹⁶ to provide amine **15** (Scheme 3). The



^a Reagents and conditions: (a) HCl, EtOAc, 0 °C, 0.5 h, 100%. (b) 1-(3-fluoro-4-cyanobenzyl)imidazole-5-carboxaldehyde (**14**), Na(AcO)₃BH, 4 Å MS, DCE, 24 h, 56%. (c) MsCl, Et₃N, CH₂Cl₂, 30 min. (d) NBS, AIBN, CCl₄, 80 °C, 14 h, 70%, two steps. (e) TfOH, Et₃SiH, CH₂Cl₂−CH₃CN, 14 d, 69% based on 75% conv. (f) NaH, DMF, **17**, 0 °C, 1 h, 55%. (g) Cs₂CO₃, DMSO, 80 °C, 5 h, 34%.

2,4-dimethoxybenzyl amide protecting group, normally sensitive to treatment with TFA, ¹⁸ required prolonged

exposure to triflic acid and triethylsilane in a solubilizing dichloromethane—acetonitrile mixture to effect deprotection. The robustness of this protecting group is likely due to protonation of the piperazinone amino group in 15, which impedes protonation of the nearby amide. Alkylation of the bridged piperazinone with bromide 17 gave 18, which was subjected to a tandem base-promoted methanesulfonate group deprotection and $S_{\rm N}Ar$ cyclization to provide the cyclophane 19.19

The calculated lowest-energy structure of **19** reveals the anticipated conformation with the ethylene bridge projecting to the exterior of the macrocyclic ring (Figure 2, gray).²⁰

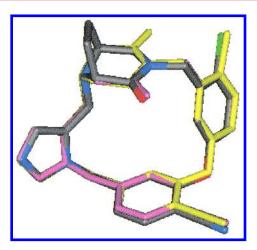


Figure 2. Superposition of the calculated lowest energy conformations of 19 (gray) and 13 (pink), and a conformation of 13 with the carbonyl group projecting back (yellow).

Variable temperature ^{1}H NMR studies revealed differential line broadening below 0 °C, with two sets of resonances at -50 °C (\sim 5:1 ratio), indicating slow conformational exchange. NOE data obtained at room temperature are consistent with a mixture of available species, most likely

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Org. Lett., Vol. 3, No. 6, 2001

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derived from mobility of the imidazole, cyanophenyl, and chlorophenyl rings relative to the 3,8-diazabicyclo[3.2.1]-octan-2-one ring.²¹ Compound 13, in its lowest energy state (Figure 2, pink),²⁰ adopts the same conformation as does 19. The corresponding orientation of 13 with the carbonyl group projected in the opposite direction²² is only 2.63 kcal/mol higher in energy (Figure 2, yellow). In comparison, a conformation of 19 with the alternative rearward orientation of the carbonyl group (not shown) is 39.2 kcal/mol higher in energy than the ground state, since it requires that the ethylene bridge reside in the congested interior of the macrocycle.

Compound 19 was assayed for inhibition of FTase-catalyzed incorporation of [³H]-farnesylpyrophosphate into recombinant Ras-CVIM²³ and was found to be more potent

than the parent unbridged compound 13 (IC₅₀ 19 = 0.13 nM, IC₅₀ 13 = 0.7 nM). Taken with the conformational analyses described above, this result suggests that the FTase-bound conformation of 13 is one in which the carbonyl group is projected in a forward orientation (*S*-configuration of the piperazinone plane), as in the constrained compound 19. This, in conjunction with earlier results, ¹⁶ implies that 1-aryl-2-piperazinone FTIs (e.g., 12) adopt a similar enzyme-bound conformation as well.

In summary, a new method for the synthesis of the 3,8-diazabicyclo[3.2.1]octan-2-one framework enables access to $C\alpha$ -substituted derivatives as constrained peptide mimetics. A bicyclic-constrained analogue of a piperazinone farnesyltransferase inhibitor was prepared and was useful for the elucidation of enzyme-bound conformation.

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Supporting Information Available: Experimental procedures and characterization for compounds. This material is available free of charge via the Internet at http://pubs.acs.org. OL015504W

868 Org. Lett., Vol. 3, No. 6, 2001

⁽²¹⁾ This interpretation is supported by computational analyses (ref 20), which revealed only moderate energy increases for the following conformational perturbations relative to lowest energy 19 (cf. Figure 2): front-to-back rotations of the imidazole, cyanophenyl rings and back-to-front rotation of the chlorophenyl ring (1.66 kcal/mol); axial—equatorial isomerization (2.94 kcal/mol).

⁽²²⁾ An alternative conformation of **13** with the carbonyl projected in the opposite direction relative to **19** is the enantiomer of lowest energy **13**. (23) For assay conditions, see: Graham, S. L.; deSolms, S. J.; Giuliani, E. A.; Kohl, N. E.; Mosser, S. D.; Oliff, A. I.; Pompliano, D. L.; Rands, E.; Breslin, M. J.; Deanna, A. A.; Garsky, V. M.; Scholz, T. H.; Gibbs, J. B.; Smith, R. L. *J. Med. Chem.* **1994**, *37*, 725.