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A route to selective functionalization of polyhydroxypyrrolidines

Carlos A. D. Sousa^a, Fabio Rizzo-Aguiar^a, M. Luísa C. Vale^a, Xerardo García-Mera^b, Olga Caamaño^b, José E. Rodríguez-Borges^{a,*}

^a Centro de Investigação em Química, Department of Chemistry and Biochemistry, Faculty of Science, University of Porto, Rua do Campo Alegre 687, 4169-007 Porto, Portugal ^b Departamento de Química Orgánica, Facultade de Farmacia, Universidade de Santiago de Compostela, E-15782 Santiago de Compostela, Spain

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ABSTRACT

A route to selective functionalization of polyhydroxypyrrolidines is described. The method is based on orthogonal protection/deprotection along the process of synthesis of the referred pyrrolidines, which consist in hydroxylation of the double bond of 2-azabicyclo[2.2.1]hept-5-enes followed by its oxidative cleavage and in situ reduction of the intermediate dialdehyde. The synthesis of a novel *N*-hydroxypyrrolidine is also described.

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Currently, the synthesis of iminosugars, also called azasugars, is an important and active field in synthetic organic chemistry.¹ Due to their structural resemblance to sugars and the consequent ability to act as glycosidase inhibitors,² these molecules (polyhydroxylated pyrrolidines/piperidines) often show potential in biological functions mediated by carbohydrates. In fact, these 'glycomimetics' have shown application in the control of diabetes,³ Gaucher's disease,⁴ cancer^{3c,5} and viral infections (including influenza)^{3c,6} or HIV.⁷ In particular, DAB 1 (1), fagomine (2), and (2S,3R,4R)-3,4-dihydroxyproline (3) have been screened as potential inhibitors of HIV replication.^{6d,e} 1-Deoxynojirimycin (4) has proved to be a powerful inhibitor of α -glycosidases⁸; and its *N*-alkylated form is used to treat Gaucher's disease (Migustat (5)).^{4e} Some *O*-alkylated *N*-hydroxypiperidines (6) have also been described as active against glycosidases.⁹

In this context, we have previously reported the synthesis of polyhydroxymethylpyrrolidines **7** and **8**.¹⁰ Herein we describe a route to the selective functionalization of the hydroxyl groups of this class of compounds, including novel polyhydroxymethyl *N*-hydroxypyrrolidines, whose synthesis is also described.



Selective protection of the hydroxyl groups of prolinate **9**, prepared from azabicycle **10** (*exo* isomer) by hydroxylation of the double bond and subsequent oxidative cleavage and in situ reduction of the intermediate dialdehyde,¹⁰ using *tert*-butylchlorodiphenylsilane was attempted (Scheme 1). However, the addition of equimolar amounts of *tert*-butylchlorodiphenylsilane resulted in a mixture of monoprotected pyrrolidines **11** and **12** (yield = 45%; ratio 1:3) and diprotected pyrrolidine **13**.¹¹

The absolute configuration of compounds **11** and **12** was unequivocally determined from crystallographic data of X-ray diffraction of the 3,5-dinitrobenzoylderivative of the isolated major monosilylated pyrrolidine **12** (Fig. 1).¹¹

Although compounds **11** and **12** can be isolated by column chromatography, this method is not effective for the selective protection of the hydroxyl groups of compound **9**, particularly if we aim to





^{*} Corresponding author. Tel.: +351 220402564; fax: +351 220402659. *E-mail address:* jrborges@fc.up.pt (J.E. Rodríguez-Borges).

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Scheme 1. Reagents and conditions: (i) OsO₄/*N*-methylmorpholine *N*-oxide, dioxane/acetone/H₂O, rt, 12 h (90%); (ii) NalO₄/SiO₂, CH₂Cl₂/H₂O, 30 min; (iii) NaBH₄, MeOH, 30 min (ii + iii, 84%); (iv) ClSi'BuPh₂, Et₃N, DMAP, CH₂Cl₂, 12 h, 0 °C [**11**, 11%; **12**, 34%, **13**, 15%].

obtain compound **11**. On the other hand, disilylated methyl prolinate **13** could readily be obtained in an 89% yield by using twice the equimolar amount of *tert*-butylchlorodiphenylsilane. Further application of a selective protection/deprotection methodology to compound **13**, yielded pyrrolidines **14–18** (Scheme 2), thus providing the possibility of regioselective functionalization at positions 1, 2, 3, and 5 of the (2*R*,3*R*,5*R*)-*tris*(hydroxymethyl)pyrrolidine.

N-debenzylation of **13** to afford **14** was accomplished by hydrogenolysis. Reduction of the ester group of **13** with borane dimethyl sulfide afforded compound **15**, whereas treatment of **13** with lithium aluminum hydride led to concomitant desilylation at position 3, yielding compound **16**. The structure of **16** was unequivocally established by X-ray crystallography of the monohydrated chloride **20** (Fig. 2).¹² *O*-acetylation of **16** followed by *O*-desilylation afforded compound **18**.

The *endo* stereoisomer of azabicycle **10** (**21**) was submitted to the same process of hydroxylation of the double bond described before, affording compound **22**; oxidative cleavage of **22** followed by in situ reduction of the intermediate dialdehyde afforded azabicycle **23** in a 13% overall yield (Scheme 3).¹³ Subsequent protection



Scheme 2. Reagents and conditions: (i) H_2 6 bar, Pd/C 10%, HCl/MeOH, 10 days (74%); (ii) BH₃:SMe₂ (5 M in Et₂O), 75 °C, 48 h (40%); (iii) LiAlH₄, Et₂O (89%); (iv) AcCl, Et₃N, DMAP, CH₂Cl₂, 24 h, 0 °C (70%); (v)TBAF/H₂O, acetone, 20 h (83%); (vi) H₂/Pd(OH)₂ 10% in MeOH, 40 h (70%); (vi) MeOH/H₂O/HCl/CH₂Cl₂ (slow evaporation).

of the hydroxyl group with *tert*-butylchlorodiphenylsilane, followed by *N*-debenzylation afforded compound **25**, which is suitable for *N*-functionalization.

In what concerns *N*-hydroxypyrrolidines, only a few reports of multi-step syntheses were found in the literature.¹⁴ *N*-hydroxy-pyrrolidines can be obtained by a similar methodology as described above, starting from the *exo* stereoisomer of the *N*-hydroxy-azabicycle **26** (Scheme 4).¹⁵ After protection of the hydroxyl group of **26** using *tert*-butylchlorodiphenylsilane, the resulting bicycle **27** was submitted to di-hydroxylation followed by oxidative cleavage and in situ reduction, affording pyrrolidine **29**.



Figure 1. X-ray single crystal structure of the 1,3-dinitrobenzoyl derivative of 12.



Figure 2. X-ray single crystal structure of 20.



Scheme 3. Reagents and conditions: (i) OsO_4/N -methylmorpholine *N*-oxide, diox-ane/acetone/H₂O, rt, 7 h (32%); (ii) $NaIO_4/SiO_2$, CH_2Cl_2/H_2O , 2 h; (iii) $NaBH_4$, MeOH, 2H (ii + iii, 42%); (iv) $CISi^{T}BuPh_2$, Et_3N , DMAP, CH_2Cl_2 , 12 h, 0 °C (57%); (v) H₂ 6 bar, Pd/C 10%, HCI/MeOH, 24 h (85%).



Scheme 4. Reagents and conditions: (i) $ClSi^{t}BuPh_{2}$, $Et_{3}N$, DMAP, $CH_{2}Cl_{2}$, 12 h, 0 °C (92%); (ii) OsO_{4}/N -methylmorpholine *N*-oxide, dioxane/acetone/H₂O, rt, 18 h (90%); (iii) $NalO_{4}/SiO_{2}$, $CH_{2}Cl_{2}/H_{2}O$, 3h; (iv) $NaBH_{4}$, MeOH, 5 h (iii + iv, 62%); (v) BnOC(NH)CCl_{3}, $CF_{3}SO_{3}H$, $CH_{2}Cl_{2}$, 1.5 h, rt (99%); (vi) TBAF/H₂O, acetone, 24 h, rt (75%).

Di-benzylation of the hydroxyl groups and subsequent desilylation afforded *N*-hydroxypyrrolidine **31**.

Attempts to perform the same reactions with the *endo* isomer of **28** (oxidative cleavage of the C_5-C_6 bond followed by in situ reduction of the intermediate dialdehyde) led to degradation of the

starting material with recovery of traces of bicyclic compounds, pointing to the occurrence of undesired rearrangements similar to those illustrated in Scheme 3 for compound **22**.

All new compounds presented in this work were fully characterized by ¹H and ¹³C NMR and mass spectrometry.

In conclusion, an efficient and straightforward orthogonal protection/deprotection methodology, allowing selective introduction of functional groups in polyhydroxymethylpyrrolidines, whose synthesis revealed to be effective and simple, has been developed.

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Supplementary data

Supplementary data (NMR data, NMR spectra and mass spectroscopy data for all compounds) associated with this article can be found, in the online version, at doi:10.1016/j.tetlet.2011.12.037.

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- 12. The crystallographic data for the compound **20** have been deposited at the Cambridge Crystallography Data Centre as supplementary publication number

UKCCDC805275. Copies of the data can be obtained free of charge on application to CCDC, 12 Union Road, Cambridge, CB2, 1EZ.

- 13. The low yield observed results from two main reasons: (1) endo adducts similar to 21 are usually more sensible to reaction conditions than their exo isomers; (2) oxidative cleavage and in situ reduction of endo adducts similar to 22 may result in material degradation and/or intramolecular rearrangements.
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