

Chiral, Densely Functionalized Cycloheptanes from Carbohydrates. I. The Nitron Route

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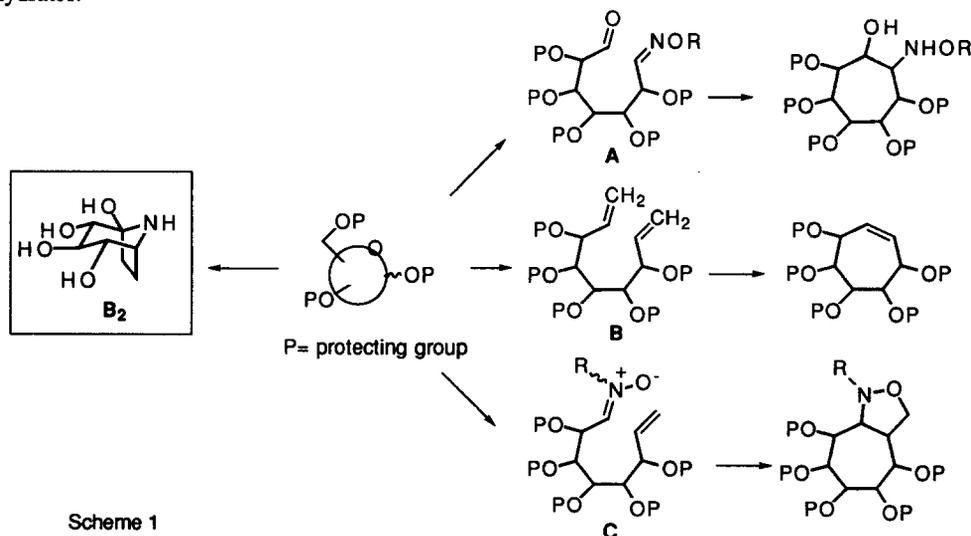
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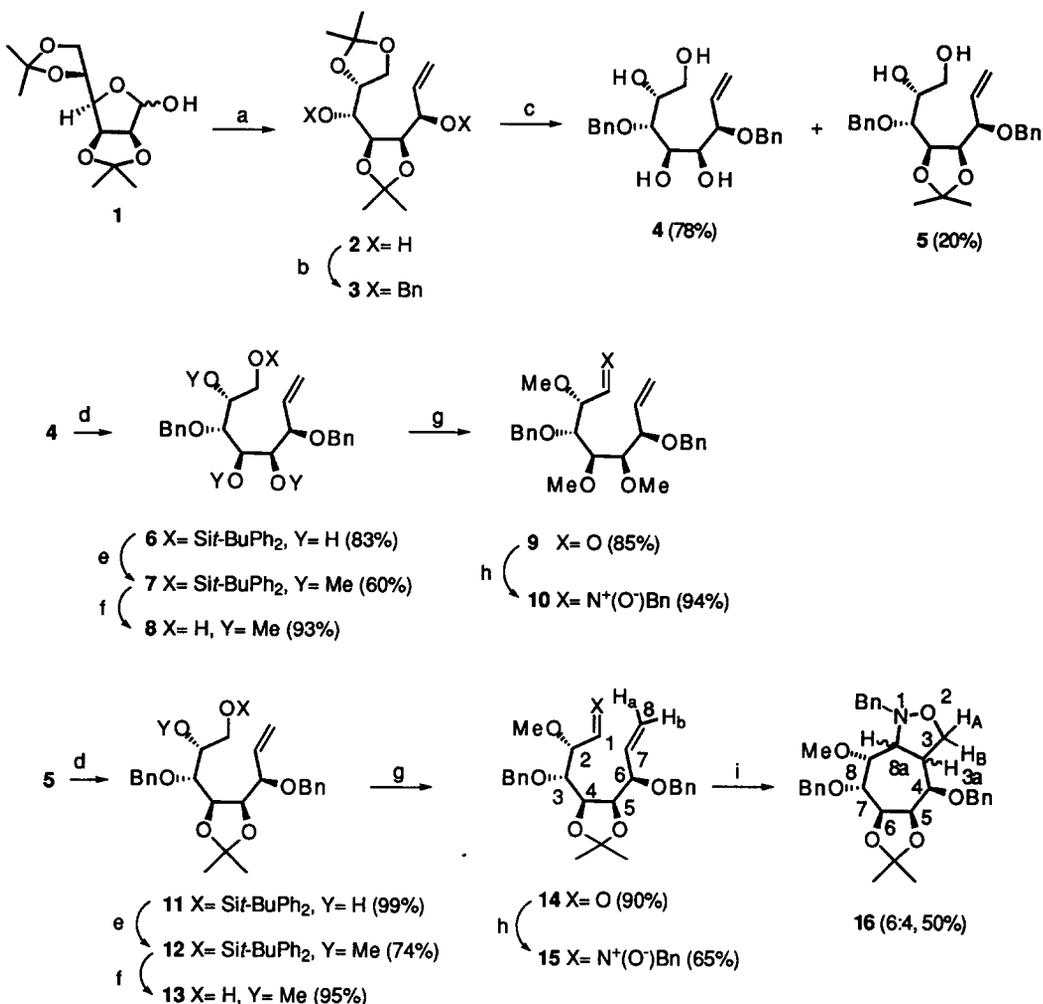
Abstract.— The first intramolecular 1,3-dipolar cycloaddition of an acyclic, chiral, polyfunctionalized 7-alkenyl tethered *N*-benzyl nitron is reported. This is a new and efficient approach for the synthesis of chiral, densely functionalized cycloheptanes from carbohydrates. © 1999 Elsevier Science Ltd. All rights reserved.

The asymmetric synthesis of cycloheptanes from carbohydrates is limited to some isolated examples: (a) nitro condensation of 1,6-dialdehydo sugar derivatives,¹ (b) ring expansion of enantiomerically pure cyclohexanones^{2,3} and (c) the intramolecular 1,3-dipolar cycloaddition (1,3-DC) of nitrile oxides.⁴ Our current interest in the synthesis of calystegine **B**₂ (Scheme 1)⁵ and the absence of a general and efficient methodology for the synthesis of chiral cycloheptanes⁶ from carbohydrates, prompted us to explore new synthetic possibilities for such a conversion. We have studied a free radical cyclization,⁷ a metathesis ring closing⁸ and the intramolecular 1,3-DC of 7-alkenyl nitrones,^{9,10} *via* intermediates **A**, **B** and **C**, respectively (Scheme 1).

In this communication we present our preliminary results on the last approach, showing the first example of an intramolecular 1,3-DC¹¹ of an acyclic, 7-alkenyl tethered *N*-benzyl nitron of type **C** (Scheme 1), that has resulted in a new and efficient asymmetric synthesis of chiral, densely functionalized cycloheptanes from carbohydrates.



Scheme 1



Scheme 2. Reagents: (a) BrMgCH=CH₂, THF (2:epi-(C-1)-2/ 4:1;75%; separate isomers); (b) BnBr, NaH, THF (85%); (c) AcOH:H₂O (4:1), 0.2 M, 55 °C; (d) ClSi^t-BuPh₂, py; (e) MeI, NaH, THF; (f) Bu₄NF, THF; (g) PCC, NaOAc, molecular sieves; (h) NaHCO₃, EtOAc, HONHBn. HCl; (i) Chlorobenzene, 130 °C.

For our studies directed to the synthesis of the required precursors, we selected the readily available 2,3:5,6-bis-*O*-isopropylidene- α -D-mannofuranose (**1**).¹² Vinylmagnesium bromide addition, gave the major isomer **2**¹³ that was isolated in good chemical yield (Scheme 2). Benzoylation and acid hydrolysis gave a mixture of tetrol **4** and monoacetone **5**.^{14,15} Compound **4** was submitted to standard protocols for the final assembly of the required functionality for nitron 1,3-cycloaddition (Scheme 2). Silylation followed by methylation under mildly basic conditions¹⁶ and desilylation gave compound **8** in good chemical yield. Product **8** was submitted to oxidation with PCC in the presence of molecular sieves and sodium acetate to give aldehyde **9** that was transformed into the nitron **10** by the usual methodology. To our great surprise,⁴ thermolysis of

compound **10** did not afford any of the expected cycloisoxazolidine adducts; only decomposition to the aldehyde **9** was observed and no cyclized product could be isolated.

In view of these results we turned our attention to the intermediate **5** and submitted it to the same sequence of synthetic transformations that gave compound **10** from **4** in good overall yield. Following these procedures, nitron **15**¹⁷ was obtained from compound **5**. To our satisfaction, thermolysis of this precursor, in chlorobenzene, afforded product **16** in 50% yield,¹⁷ as a mixture of isomers in 6:4 ratio, that we were unable to separate, and whose relative and absolute stereochemistry at the newly formed stereocenters could not be assigned. From these results it was clear that the 1,3-dioxolane ring at positions C-4 and C-5 in the precursor is a potent and critical structural element that controls the success of the intramolecular reaction leading to the cycloheptane.¹⁸ This structural motif obviously reduces the conformational degrees of freedom in the transition state, and as a result favours the carbocyclization.

The analytical [correct elemental analysis for C₃₃H₃₉NO₆; MS: *e/m* (70 eV): 545 (M⁺, 21), 530 (M⁺-15), 91 (C₇H₇⁺, 100)] and spectroscopic data (full set of NMR experiments: ¹H, ¹³C, DEPT, COSY HMQC) of compound **16** clearly supported the regiospecific formation of the bicyclo[5.3.0] type of isoxazolidine, instead of the alternative bicyclo[5.2.1] type of isoxazolidine¹⁹ (the absence of signals at ~26.0 ppm, typical for a methylene in the bridge position¹⁹ was significant and moved us to discard this hypothesis). Particularly diagnostic was the analysis of the ¹³C NMR spectrum of the mixture; in this spectrum we could observe coherent²⁰ and significant signals for compound **16** [in the major isomer: (δ) 42.4 (C3a), 67.0 (C-3), 67.6 (C-8a), 62.3 (N-CH₂C₆H₅), 58.1 (OCH₃); in the minor isomer: (δ) 44.4 (C3a), 68.3 (C-3), 67.5 (C-8a), 64.6 (N-CH₂C₆H₅), 61.1 (OCH₃)]. In agreement with this assumption, in the ¹H NMR spectrum of compounds **16**, we could analyze H-3A (3.53, dd, *J* = 10.8, 6.6 Hz), H-3a (2.75, dt, *J* = 10.8, 8.1 Hz), H-8a (2.90, m) for the major isomer, and H-3A (3.68, dd, *J* = 10.9, 7.6 Hz), H-3a (3.00, m), H-8a (3.19, m) for the minor isomer.

In summary, we have described the first successful intramolecular 1,3-DC of an acyclic, chiral, polyfunctionalized 7-alkenyl tethered *N*-nitron. We have found that the presence of an annulated ring in the precursor (i. e., an acetonide in positions O-C4 and O-C5 in compound **15**) was critical for the success of the process. These results pave the way for a new, simple and efficient synthesis of chiral, densely functionalized cycloheptanes from carbohydrates. Work is in progress to check the scope and limits of this methodology.

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References and Notes

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 - All new compounds showed excellent analytical and spectroscopic data.
 - After some experimentation, conditions were found (AcOH:H₂O/4:1, 0.2 M, 55 °C) to improve the yields of compounds **4** and **5** for the synthesis of the advanced intermediates **10** and **15**, in order to compare their respective reactivity in the 1,3-DC reaction.
 - The 60% yield in the synthesis of compound **7** from **6** was not optimized. In this reaction other more polar (probably mono and/or disubstituted methylated derivatives) were detected, but not isolated.
 - To a solution of aldehyde **14** (50 mg, 0.11 mmol) and *N*-benzylhydroxylamine hydrochloride (54 mg, 0.34 mmol) in dry methylene chloride (1 mL) pyridine (0.031 mL, 0.39 mmol) was added and the mixture was refluxed for 12 h. The solution was cooled, diluted with methylene chloride, washed with brine and dried. After evaporation, the residue was submitted to flash chromatography (hexane: ethyl acetate, 1:1) to give pure nitrono **15** (38 mg, 65%) [oil; ¹H NMR (CDCl₃, 300 MHz) δ 7.38-7.26 (m, 15 H, aromatic), 6.90 (d, *J*_{1,2} = 7.3 Hz, H-1), 5.80 (ddd, *J*_{7,8 cis} = 9.8 Hz, *J*_{7,8 trans} = 17.3 Hz, *J*_{7,6} = 8.5 Hz, 1 H, H-7), 5.41 (dd, *J*_{6,8 cis} = 1.6 Hz, 1 H, H-8a), 5.27 (dd, *J*_{6,8 trans} = 1.3 Hz, 1 H, H-8b), 4.85 (m, 1 H, H-2; s, 2 H, NCH₂C₆H₅), 4.80 and 4.62 (d, d, AB system, *J* = 12.6 Hz, 2 H, OCH₂C₆H₅), 4.53 and 4.12 (d, d, AB system, *J* = 11.3 Hz, 2 H, OCH₂C₆H₅), 4.22 (dd, *J*_{5,6} = 8.5, *J*_{5,4} = 5.4 Hz, 1 H, H-5), 4.11 (dd, *J*_{3,4} = 7.0 Hz, 1 H, H-4), 3.98 (dd, *J*_{2,3} = 1.7 Hz, 1 H, H-3), 3.83 (t, 1 H, H-6), 3.2 (s, 3 H, OCH₃), 1.40 and 1.26 (s, 3 H, 3 H, OC(CH₃)₂O). This product was dissolved in chlorobenzene (3 mL) and refluxed (bath at 130 °C) for 20 h. The solvent was removed and the residue was submitted to chromatography (hexane: ethyl acetate, 1:3) to give adduct **16** (16 mg, 50%) as an oil.
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