

Stereoselective Synthesis of the Monomeric Unit of Actin Binding Macrolide Rhizopodin[†]

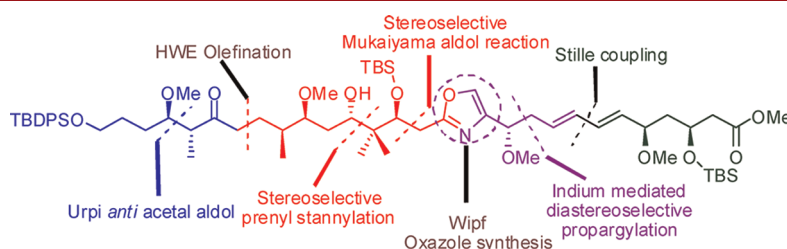
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ABSTRACT



An efficient, scalable, and stereocontrolled synthesis of the entire carbon framework of an actin binding dimeric macrolide rhizopodin has been accomplished in its protected form. The key features of our synthesis include a titanium catalyzed *anti* acetal aldol reaction, a substrate controlled diastereoselective prenyl stannylation, a Mukaiyama aldol reaction, an indium mediated diastereoselective propargylation, and an advanced stage Stille coupling reaction.

Myxobacteria were found to be a rich source of novel secondary metabolites, which includes several biologically active cytotoxic compounds such as epothilone, chondramide, tubulysin A, etc.¹ Rhizopodin (**1**) (Figure 1) is another such novel and unique polyketide isolated in 1993 from the culture broth of the *Myxococcus stipitatus*.² Initially, the structure of **1** was found to be a 19-membered macrolide, which was later revised as a C_2 -symmetric 38-membered dilactone exhibiting 18 stereogenic centers, two conjugated diene systems in combination with two disubstituted oxazoles, and two enamide side chains.³ Importantly, **1** shows potent cytotoxic activity against various cancer cell lines in low nanomolar concentration and also displays activity against certain fungi.⁴ The cytotoxic activity of **1** results from its ability to bind, thereby inhibiting the polymerization of G-actin. The effects of **1** resemble those of

latrunculin but are elicited at a 10-fold lower concentration. In contrast to latrunculin, the effects of **1** are irreversible on the cytoskeleton of actin and take a little longer to appear. The scarcity of **1** together with its intriguing biological activity and challenging architecture make it an attractive target for total synthesis. As part of our continuous interest in the development of C_2 -symmetric peptidomimetics, and the total synthesis of biologically active C_2 -symmetric diolides,⁵ we embarked on the total synthesis of rhizopodin **1**. The first total synthesis of **1** has just appeared.⁶ Earlier

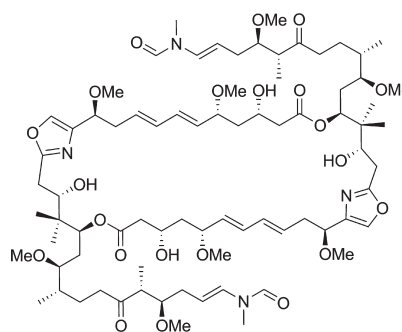


Figure 1. Structure of rhizopodin **1**.

[†] CDRI Communication no. 8250.

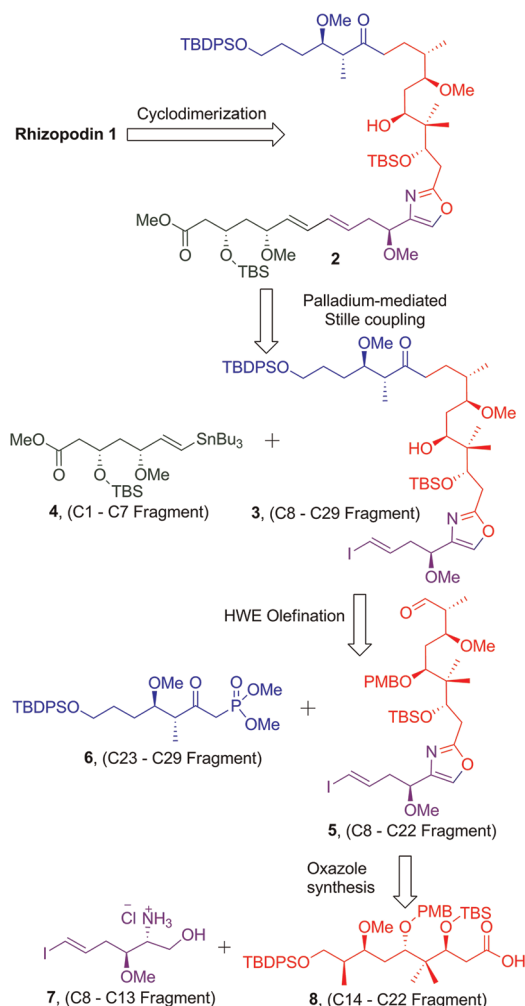
(1) Reichenbach, H. *J. Ind. Microbiol. Biotechnol.* **2001**, *27*, 149.

(2) Sasse, F.; Steinmetz, H.; Höfle, G.; Reichenbach, H. *J. Antibiot.* **1993**, *46*, 741.

(3) (a) Hagelueken, G.; Albrecht, S. C.; Steinmetz, H.; Jansen, R.; Heinz, D. W.; Kalesse, M.; Schubert, W.-D. *Angew. Chem., Int. Ed.* **2009**, *48*, 595. (b) Horstmann, N.; Menche, D. *Chem. Commun.* **2008**, *41*, 5173. (c) Jansen, R.; Steinmetz, H.; Sasse, F.; Schubert, W.-D.; Hagelueken, G.; Albrecht, S. C.; Müller, R. *Tetrahedron Lett.* **2008**, *49*, 5796.

(4) Gronewold, T. M. A.; Sasse, F.; Lunsdorf, H.; Reichenbach, H. *Cell Tissue Res.* **1999**, *295*, 121.

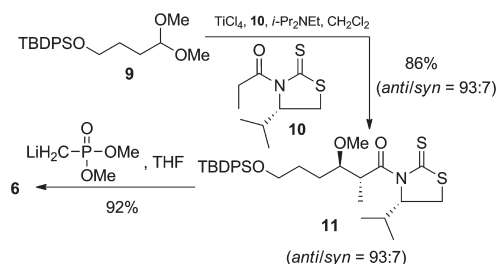
Scheme 1. Retrosynthesis of Rhizopodin 1



Nicolaou et al. reported the synthesis of mono rhizopodin,⁷ while others described fragment synthesis.⁸ We also reported the synthesis of C1–C15 and C16–C28 fragments of the molecule.⁹ Herein, we describe an efficient, stereoselective, and highly convergent synthesis of the entire monomeric unit of rhizopodin in its protected form **2**.

Our retrosynthetic analysis of **1** is depicted in Scheme 1. With our previous experience in the synthesis of C_2 -symmetric diolides, we envisaged that the 38-membered

Scheme 2. Synthesis of Ketophosphonate Fragment 6



dilactone could be synthesized from cyclodimerization of suitably protected monomer **2**, which could be obtained from a palladium catalyzed C–C bond formation between vinyl iodide **3** and vinyl stannane **4**. We previously reported the synthesis of vinyl stannane **4** via an iterative acetate aldol reaction.^{9a} Vinyl iodide **3** would arise from the Horner–Wadsworth–Emmons (HWE) olefination with keto phosphonate **6** and aldehyde **5** containing oxazole and vinyl iodide. Aldehyde **5**, in turn, would be obtained from oxazole assembly of amino alcohol **7** and carboxylic acid **8** via an amide bond formation, oxidation, and cyclodehydration strategy.

Our synthetic endeavor began from the synthesis of keto phosphonate **6** (Scheme 2) which started with the direct installation of *anti* β -methoxy- α -methyl stereocenters in a single operation by utilizing the Urpi diastereoselective *anti* acetal aldol reaction.¹⁰ Treatment of dimethyl acetal **9**¹¹ with the titanium enolate generated from (*S*)-valine derived *N*-propionylthiozolidine thione **10** provided the desired aldol product in 86% yield (*anti/syn* = 93:7, dr). Optically pure *anti* isomer **11** was isolated in 79% yield by simple column chromatography. Treatment of **11** with lithiated methyl dimethylphosphonate (generated using *n*-BuLi) directly furnished the β -keto phosphonate **6** in 92% yield.¹²

Next, the synthesis of amino alcohol **7** started from the (*R*)-Garner aldehyde **12** (Scheme 3) which was prepared from D-serine following a reported procedure.¹³ An asymmetric indium mediated propargylation of aldehyde **12**, using (1*S*,2*R*)-(+)-2-amino-1,2-diphenylethanol **13** as a chiral auxiliary developed by Singaram et al.,¹⁴ produced the homo propargyl alcohol **14** in 83% yield with good diastereoselectivity (10:1 dr, HPLC).¹⁵ Selective methylation of the secondary hydroxyl group followed by one-pot hydrozirconation–iodination using a Schwartz reagent (generated *in situ* from Cp_2ZrCl_2 and DIBAL-H),

(5) (a) Chakraborty, T. K.; Roy, S.; Koley, D.; Dutta, S. K.; Kunwar, A. C. *J. Org. Chem.* **2006**, *71*, 6240. (b) Chakraborty, T. K.; Ghosh, S.; Rao, M. H. V. R.; Kunwar, A. C. *ARKIVOC* **2005**, xi, 89. (c) Chakraborty, T. K.; Srinivasu, P.; Kumar, S. K.; Kunwar, A. C. *J. Org. Chem.* **2002**, *67*, 2093. (d) Chakraborty, T. K.; Reddy, V. R. *Tetrahedron Lett.* **2006**, *47*, 2099. (e) Chakraborty, T. K.; Reddy, V. R.; Chattopadhyay, A. K. *Tetrahedron Lett.* **2006**, *47*, 7435. (f) Chakraborty, T. K.; Reddy, V. R.; Gajula, P. K. *Tetrahedron* **2008**, *64*, 5162.

(6) Dieckmann, M.; Kretschmer, M.; Li, P.; Rudolph, S.; Herkommer, D.; Menche, D. *Angew. Chem., Int. Ed.* DOI: 10.1002/anie.201201946.

(7) Nicolaou, K. C.; Jiang, X.; Lindsay-Scott, P. J.; Corbu, A.; Yamashiro, S.; Bacconi, A.; Fowler, V. M. *Angew. Chem., Int. Ed.* **2011**, *50*, 1139.

(8) (a) Cheng, Z.; Song, L.; Xu, Z.; Ye, T. *Org. Lett.* **2010**, *12*, 2036. (b) Kretschmer, M.; Menche, D. *Org. Lett.* **2012**, *14*, 382.

(9) (a) Chakraborty, T. K.; Pulukuri, K. K.; Sreekanth, M. *Tetrahedron Lett.* **2010**, *51*, 6444. (b) Chakraborty, T. K.; Sreekanth, M.; Pulukuri, K. K. *Tetrahedron Lett.* **2011**, *52*, 59.

(10) (a) Cosp, A.; Romea, P.; Talavera, P.; Urpí, F.; Vilarrasa, J.; Font-Bardia, M.; Solans, X. *Org. Lett.* **2001**, *3*, 615. (b) Gálvez, E.; Parelló, R.; Romea, P.; Urpí, F. *Synlett* **2008**, 2951.

(11) Okada, H.; Mori, T.; Saikawa, Y.; Nakata, M. *Tetrahedron Lett.* **2009**, *50*, 1276.

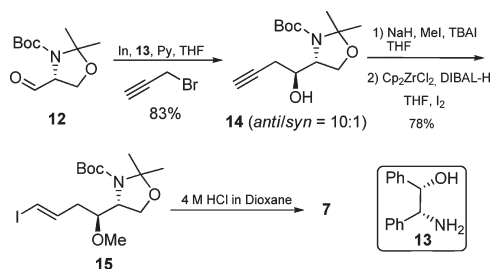
(12) Crimmins, M. T.; Siliphaivanh, P. *Org. Lett.* **2003**, *5*, 4641.

(13) Garner, P.; Park, J. M. *J. Org. Chem.* **1987**, *52*, 2361.

(14) (a) Hirayama, L. C.; Dunham, K. K.; Singaram, B. *Tetrahedron Lett.* **2006**, *47*, 5173. (b) Haddad, T. D.; Hirayama, L. C.; Buckley, J. J.; Singaram, B. *J. Org. Chem.* **2012**, *77*, 889.

(15) (a) Stereochemistry of the product was confirmed by comparing with reported data. See: Koviach, J. L.; Chappell, M. D.; Halcomb, R. L. *J. Org. Chem.* **2001**, *66*, 2318. (b) The diastereomeric ratio was determined by HPLC of the benzoate prepared by treatment with benzoyl chloride, NEt_3 , and DMAP.

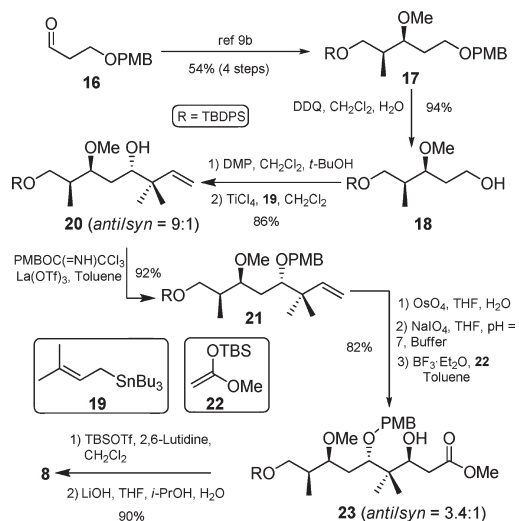
Scheme 3. Synthesis of Amino Alcohol Fragment 7



following Negishi's protocol,¹⁶ provided exclusively *E*-vinyl iodide **15** in 78% yield in two steps. Treatment of **15** with 4 M HCl in dioxane furnished the desired amino alcohol **7**.¹⁷ Due to the high sensitivity of **7** for workup as well as chromatography, the product was confirmed by ¹H NMR of the crude reaction mixture after removal of reaction volatiles and was used as such for the next reaction.

On the other hand, carboxylic acid **8** was synthesized (Scheme 4) in a stereocontrolled manner starting from protected compound **17**, which was prepared in four steps from PMB protected aldehyde **16** following a literature procedure.^{9b} Oxidative removal of the PMB group in **17** using DDQ gave the primary alcohol **18** in 94% yield. Dess–Martin periodinane oxidation¹⁸ of the resultant primary alcohol yielded the corresponding aldehyde, which was then subjected to methoxy directed¹⁹ chelation controlled prenyl stannylation using TiCl₄ to obtain the alcohol in 86% yield with a 9:1 ratio (determined by the ¹H NMR integration) of separable diastereomers in favor of the desired *anti* isomer **20**. Lower selectivities were observed with other Lewis acids such as MgBr₂·Et₂O, ZnBr₂, and BF₃·Et₂O. This transformation establishes the C17 geminal dimethyl and C18 stereocenter with a good level of 1,3-*anti* stereinduction. The C18 hydroxyl group was protected as *p*-methoxybenzyl ether to obtain **21** using PMBOC(=NH)CCl₃ and La(OTf)₃.²⁰ Dihydroxylation of olefin followed by oxidative cleavage of the resulting diol provided the corresponding aldehyde in good yield. To install the β-hydroxy ester, a Lewis acid mediated Mukaiyama aldol reaction²¹ between the aldehyde and commercially available silyl enol ether **22** was investigated. After various Lewis acids and conditions were screened,

Scheme 4. Synthesis of Carboxylic Acid Fragment 8



the monodentate Lewis acid BF₃·Et₂O was identified to promote the desired aldol addition in toluene to give the required β-hydroxyl ester **23** in 82% yield (three steps) with 3.4:1 diastereoselectivity in favor of the desired *anti* isomer.²² Noteworthy is that the use of bidentate Lewis acid TiCl₄ provided the aldol product in a reversal of the diastereoselectivity (1:1.7, dr).²³ Since the resultant diastereomers were not separable at this stage we continued further as such for the silyl protection of hydroxyl group followed by saponification of methyl ester to obtain the acid **8** in 90% yield (*anti/syn* = 3.4:1).

With both amino alcohol **7** and carboxylic acid **8** in hand, we endeavored to combine them through an oxazole synthesis (Scheme 5). First, an amide coupling between amino alcohol **7** and carboxylic acid **8** using EDCI, HOBT, and NEt₃²⁴ afforded a mixture of hydroxyl amide (*anti/syn* = 3.2:1) in 76% yield. At this stage both C16 epimers were separated by column chromatography and optically pure *anti* isomer **24** was isolated in 58% yield. Dess–Martin periodinane mediated oxidation of hydroxyl amide **24** followed by one-pot cyclodehydration of the resultant aldehyde and subsequent elimination of HBr from the bromooxazoline intermediate following Wipf conditions²⁵ afforded the oxazole **25** in 74% yield. Selective primary silyl deprotection using acetic acid buffered

(16) Huang, Z.; Negishi, E.-I. *Org. Lett.* **2006**, *8*, 3675.

(17) Han, G.; Tamaki, M.; Hruby, V. J. *J. Peptide Res.* **2001**, *58*, 338.

(18) (a) Dess, D. B.; Martin, J. C. *J. Org. Chem.* **1983**, *48*, 4155. (b) Dess, D. B.; Martin, J. C. *J. Am. Chem. Soc.* **1991**, *113*, 7277.

(19) (a) Keck, G. E.; Castellino, S.; Wiley, M. R. *J. Org. Chem.* **1986**, *51*, 5478. (b) Keck, G. E.; Castellino, S. *J. Am. Chem. Soc.* **1986**, *108*, 3847. (c) Keck, G. E.; Abbott, D. E. *Tetrahedron Lett.* **1984**, *25*, 1883. (d) For methoxy directed allylation, see: Paterson, I.; Coster, M. J.; Chen, D. Y.-K.; Gibson, K. R.; Wallace, D. J. *Org. Biomol. Chem.* **2005**, *3*, 2410.

(20) Rai, A. N.; Basu, A. *Tetrahedron Lett.* **2003**, *44*, 2267.

(21) (a) Mukaiyama, T.; Banno, K.; Narasaka, K. *J. Am. Chem. Soc.* **1974**, *96*, 7503. (b) Paterson, I.; Smith, J. D. *J. Org. Chem.* **1992**, *57*, 3261. (c) Paterson, I.; Smith, J. D.; Ward, R. A. *Tetrahedron* **1995**, *51*, 9413. (d) Mutou, T.; Suenaga, K.; Fujita, T.; Itoh, T.; Takada, N.; Hayamizu, K.; Kigoshi, H.; Yamada, K. *Synlett* **1997**, 199. (e) Evans, D. A.; Dart, M. J.; Duffy, J. L.; Yang, M. G. *J. Am. Chem. Soc.* **1996**, *118*, 4322. (f) Evans, D. A.; Allison, B. D.; Yang, M. G. *Tetrahedron Lett.* **1999**, *40*, 4457.

(22) (a) The absolute configuration of the major C16 stereogenic center was determined as (16S) by ¹H NMR analysis of the corresponding (*R*)- and (*S*)-MTPA esters. See: Supporting Information for details. (b) Ohtani, I.; Kusumi, T.; Kashman, Y.; Kakisawa, H. *J. Am. Chem. Soc.* **1991**, *113*, 4092.

(23) Other Lewis acids, i.e. TiCl₂(O-*i*Pr)₂, MgBr₂·Et₂O, ZnBr₂, and LiClO₄, were also investigated but were less effective. Under these conditions either no reaction or decomposition of aldehyde was observed.

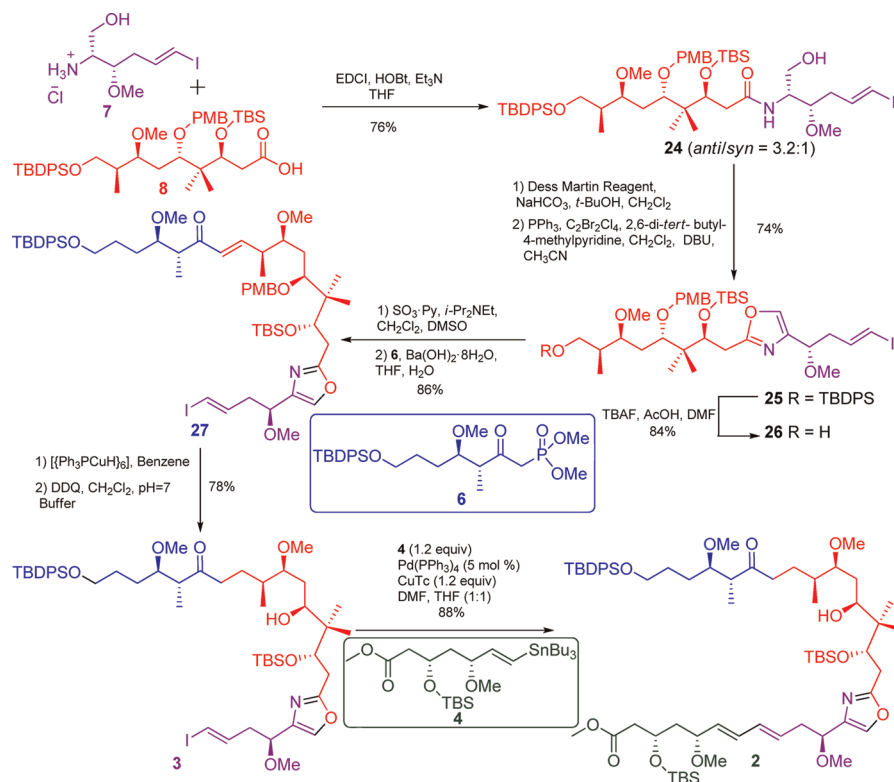
(24) Sheehan, J.; Cruickshank, P.; Boshart, G. *J. Org. Chem.* **1961**, *26*, 2525.

(25) (a) Wipf, P.; Miller, C. P. *J. Org. Chem.* **1993**, *58*, 3604. (b) Wipf, P.; Lim, S. *J. Am. Chem. Soc.* **1995**, *117*, 558.

(26) Higashibayashi, S.; Shinko, K.; Ishizu, T.; Hashimoto, K.; Shirahama, H.; Nakata, M. *Synlett* **2000**, 1306.

(27) Parikh, J. R.; Doering, W. V. E. *J. Am. Chem. Soc.* **1967**, *89*, 5505.

Scheme 5. Assembly of Fragments 7, 8, 6, and 4 To Complete the Synthesis of 2



TBAF²⁶ furnished the primary alcohol **26** in 84% yield. Parikh–Doering oxidation²⁷ of primary alcohol **26** furnished the corresponding aldehyde which was subjected to HWE olefination with keto phosphonate **6** in the presence of Ba(OH)₂²⁸ to achieve the enone **27** in 86% yield. Selective 1,4-reduction of enone using Stryker's reagent²⁹ followed by oxidative cleavage of PMB ether gave the alcohol **3** (C8–C29 fragment) in 78% yield.

With key fragment **3** in hand, the stage was now set for coupling with vinyl stannane **4** through C7–C8 bond formation using the Stille reaction.³⁰ Initial attempts to accomplish the Stille coupling of vinyl iodide **3** and vinyl stannane **4** with standard Stille conditions using 10 mol % of Pd (0, II) catalysts were less effective and gave the desired diene only in 30% yield, along with a byproduct resulting from the homocoupling of vinyl iodide **3**. After several conditions were examined by using various combinations of palladium catalysts (Pd(PPh₃)₄, PdCl₂(CH₃CN)₂, Pd₂(dba)₃, PdCl₂(PPh₃)₂), transmetalation ligands (As(PPh₃)₃, TFP), transmetalation catalysts (CuCl, CuI, CuTc), and solvents (NMP, DMF, THF), it was found, to our delight, that use of 5 mol % Pd(PPh₃)₄ and a stoichiometric amount

of copper thiophene-2-carboxylate (CuTc) in a DMF/THF (1:1) mixture³¹ gave the coupled product **2** in 88% yield.

In summary, a concise and scalable synthesis of the protected monomeric unit **2** of rhizopodin, suitable for subsequent steps and a late-stage introduction of the enamide side chains, was achieved in a highly convergent way in 21 steps (longest linear sequence) starting from **16**. Notable features of our synthesis include a diastereoselective *anti* acetal aldol reaction, a highly diastereoselective propargylation of the Garner aldehyde, a substrate controlled *anti* stereoselective prenyl stannylation, a Mukaiyama aldol reaction, an advanced oxazole synthesis, and Stille coupling. Further work toward the total synthesis of rhizopodin **1** and its simplified structural analogs is in progress.

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Supporting Information Available. Experimental procedures, spectral data, ¹H and ¹³C spectral data of compounds. This material is available free of charge via the Internet at <http://pubs.acs.org>.

(31) Both Pd and CuTc are required for this coupling.

The authors declare no competing financial interest.

(28) Paterson, I.; Yeung, K.-S.; Smaill, J. B. *Synlett* **1993**, 774.
(29) (a) Mahoney, W. S.; Brestensky, D. M.; Stryker, J. M. *J. Am. Chem. Soc.* **1988**, *110*, 291. (b) Terminal olefin resulting from the reduction of vinyl iodide appeared as a side product in 5–10% yield.
(30) (a) Stille, J. K.; Groh, B. L. *J. Am. Chem. Soc.* **1987**, *109*, 813. (b) Allred, G. D.; Liebeskind, L. S. *J. Am. Chem. Soc.* **1996**, *118*, 2748. (c) Fürstner, A.; Funel, J.-A.; Tremblay, M.; Bouchez, L. C.; Nevado, C.; Waser, M.; Ackerstaff, J.; Stimson, C. C. *Chem. Commun.* **2008**, 2873.