



SAR of *N*-phenyl piperidine based oral integrin $\alpha 5\beta 1$ antagonists

Gunther Zischinsky, Frank Osterkamp, Doerte Vossmeier, Grit Zahn, Dirk Scharn, Ariane Zwintscher, Roland Stragies*

Jerini AG, Invalidenstraße 130, Berlin D-10115, Germany

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ABSTRACT

Recently, a new class of selective integrin $\alpha 5\beta 1$ inhibitors consisting of a heterocyclic based scaffold was published. Herein the SAR and pharmacokinetic profiles of *N*-phenyl piperidine derivatives are described.

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Integrins are a family of transmembrane adhesion receptors that mediate cell contact to neighboring cells and proteins of the extracellular matrix.¹ They have been described to be involved in several pathological conditions including inflammatory diseases, autoimmune diseases, tumor progression and metastasis by regulating leukocyte trafficking, cell proliferation, migration, angiogenesis and lymphangiogenesis.^{2–5} The therapeutic efficacy of integrin antagonists like cilengitide ($\alpha v\beta 3$) and natalizumab ($\alpha 4\beta 1$) has been shown for the treatment of glioblastoma and multiple sclerosis, respectively.^{6,7}

Knockout studies of the fibronectin receptor $\alpha 5\beta 1$ integrin and tumor angiogenesis studies with $\alpha 5\beta 1$ integrin inhibitors have revealed a role in blood vessel formation for this integrin.^{8,9} Furthermore, systemic treatment with an $\alpha 5\beta 1$ integrin small molecule antagonist or an $\alpha 5\beta 1$ integrin antibody reduced tumor growth in proof-of-concept studies for different tumor types.^{10,11} Integrin $\alpha 5\beta 1$ antagonists are therefore promising drug candidates for the therapy of diseases that involve angiogenic and proliferative processes.

Recently, the discovery of integrin $\alpha 5\beta 1$ antagonists with low nanomolar affinities based on heterocyclic core structures (**2**) derived from a 3-hydroxypyrrolidine scaffold like in **1**¹² was reported (Fig. 1).¹³ Herein, the SAR and optimization of one of these new scaffolds with an *N*-phenyl piperidine core is disclosed.

Compounds **3a–i** were synthesized according to the convergent synthesis strategy shown in Schemes 1–3. The three piperidines **6a–c** were obtained from alkylation of intermediate **5**¹³ and subsequent deprotection of the piperidine nitrogen atom (Scheme 1).

The boronic acid derivatives **8** and **10** were synthesized from the corresponding commercially available starting materials (Scheme 2).

The key building blocks **11a–c** and (*S*)-**11a** were obtained by a copper mediated coupling reaction¹⁴ of **6a–c** with either the racemic boronic acid **8**¹⁵ or the optically pure potassium trifluoroborate **10**. After liberation of the primary amines, amide formation with different benzoic acids and saponification of the methyl ester gave the desired integrin antagonists **3a–i**.

Previously, the first piperidine based integrin $\alpha 5\beta 1$ antagonist **3b** was reported¹³ and was shown to have excellent integrin $\alpha 5\beta 1$ affinity (2.3 nM). Selectivity against integrin $\alpha v\beta 3$ was medium, whereas selectivities against integrins $\alpha v\beta 5$ and $\alpha IIb\beta 3$ were excellent (Table 1). Compound **3b** showed high total body clearance (CL = 70 ml/min/kg) after iv administration in rat. Therefore selectivity against integrin $\alpha v\beta 3$ and the pharmacokinetic profile should be improved.

As previously reported,¹² the electron accepting or donating nature of substituents in the 4-position of the 2-aminopyridine has a great influence on integrin binding (Table 1). As expected, the electron donating 4-OMe derivative **3c**, had the highest integrin $\alpha 5\beta 1$ binding activity in the ELISA and the cellular binding assay. Otherwise, the 4-H and 4-Me derivatives **3a** and **3b** showed similar IC₅₀ values in the ELISA, whereas the cellular IC₅₀ value of **3c** is significantly lower. However, as the cellular binding assay represents the more functional assay it was considered to be more relevant one.¹²

* Corresponding author. Correspondence should be addressed to: Pericles Calias, Shire HGT, 125 Spring Street Lexington, MA 02421, USA. Tel.: +1 781 482 0701; fax: +1 781 482 2961.

E-mail address: pcalias@shire.com (R. Stragies).

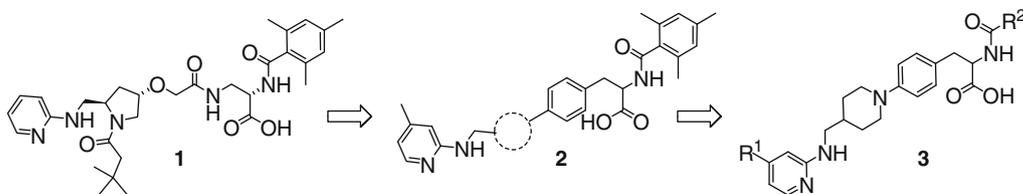
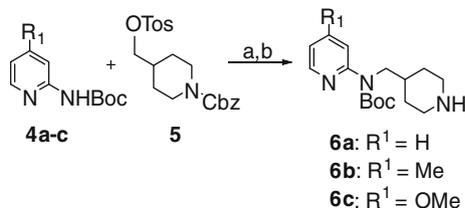


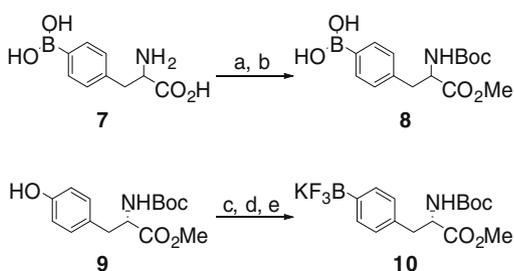
Figure 1. Generic structure **3** with an *N*-phenyl piperidine core was discovered during the optimization of the new integrin $\alpha 5\beta 1$ antagonists type **2** which were based on the 3-hydroxypyrrolidine lead **1**. The dotted cycle in **2** represents heterocyclic five- or six-membered ring systems.

In contrast to the increasing integrin $\alpha 5\beta 1$ binding affinity, an opposing trend was found for oral bioavailability and the in vitro permeability (Table 2). This observation can be rationalized by the different pK_a values caused by the increasing electron donating effect of the 4-H < 4-Me < 4-OMe substituents in **3a–c**. The different fractions of unprotonated base at physiological pH may explain the observed in vitro permeabilities as well as the oral bioavailabilities of **3a–c**.

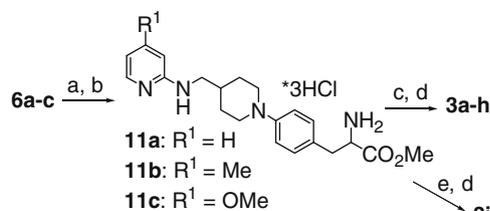
Furthermore, the same trend was found for the systemic clearance after iv administration and in the microsomal stability (MS rat). In favor of better pharmacokinetic properties, the unsubstituted 2-aminopyridine of derivative **3a** was therefore selected and kept fixed for further optimization.



Scheme 1. Reagents and conditions: (a) KHMDs, DMSO, 0 °C; (b) Pd/C, H₂, ethyl acetate, rt.



Scheme 2. Reagents and conditions: (a) HCl, MeOH, reflux; (b) Boc₂O, NaHCO₃, ^tBuOH; (c) Tf₂O, pyridine, DCM, –30 °C to rt; (d) PdCl₂(dppf), bis(pinacolato)diborane, KOAc, DMSO, 80 °C; (e) KHF₂, H₂O/MeOH, –5 °C.



Scheme 3. Reagents and conditions: (a) **8** or **10**, Cu(OAc)₂, DIPEA, DCE, 3 Å MS, 40 °C; (b) HCl, MeOH, reflux; (c) R²CO₂H, HATU, DIPEA, DMF, rt; (d) LiOH, THF/H₂O, rt; (e) 2-(bromomethyl)benzoyl bromide, DIPEA, DCM, 0 °C to rt.

Table 1

IC₅₀ values of **3a–c** for integrins $\alpha 5\beta 1$, $\alpha v\beta 3$, $\alpha v\beta 5$, $\alpha IIb\beta 3$ in competitive integrin binding assays

Comps	R	IC ₅₀ ^a , nM				
		$\alpha 5\beta 1$	$\alpha v\beta 3$	$\alpha v\beta 5$	$\alpha IIb\beta 3$	cell
3a	H	2.1	642	15,000	40,200	198
3b	Me	2.3	84	10,700	23,400	69
3c	OMe	0.92	89	3540	34,400	34

^a Values are means of three experiments.

It is well known that α -amino acid derived integrin $\alpha 5\beta 1$ inhibitors can gain selectivity against αv -integrins by switching from the unsubstituted α -benzoylamide to mesitylamide.^{12,16–18}

As previously discussed, the selectivity for integrin $\alpha v\beta 3$ for the *N*-phenyl piperidine scaffold had to be improved. For optimization the 2,6-dimethyl derivative **3d** was selected as starting point (Table 3). To increase selectivity a more bulky ethyl group was incorporated (**3e**). But both derivatives showed only medium stability to human microsomal degradation. To overcome this issue a 4-fluoro atom was introduced (**3f–g**), improving the microsomal stability. Specifically, the 2-ethyl-4-fluoro-6-methyl compound **3g** showed good integrin $\alpha 5\beta 1$ affinities in both the ELISA (2.0 nM) and the cellular assay (76 nM) in combination with excellent selectivities and human microsomal stability.

Analysis of the various stereo isomers of **3g** (Table 4), indicates that stereo chemistry has a profound impact on the activity as well as the selectivity toward other integrins.

The (*S*)-enantiomer exhibited the highest binding activity and selectivity. The (*R*)-enantiomer was found to be less potent than the (*S*)-enantiomer or the racemic mixture and the relative selectivity toward other integrins dropped by two orders of magnitude.

As previously described, the 2,6-disubstituted benzamide moiety represents the key for selectivity of integrin $\alpha 5\beta 1$ antagonists. To elucidate this, derivative **3i** with a plane isoindolin-1-one was synthesized and the integrin activities were compared with the more flexible benzamides **3h** and **3g** in the same position (Table 5).

The increasing activities for integrin $\alpha 5\beta 1$ are associated with a decrease of integrin $\alpha v\beta 3$ activities, which resulted in high integrin $\alpha 5\beta 1$ selectivities. The calculated torsion angle[†] (Fig. 2) between the amide and phenyl planes correlates positively with these activities. On the other hand, no direct correlation of the integrin $\alpha IIb\beta 3$

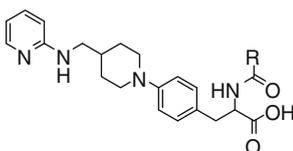
[†] For energy minimization the MM2 force field implemented in Chem3D Pro 10.0[®] (CambridgeSoft) was used.

Table 2
In vitro and in vivo pharmacokinetic properties of **3a–c**

Compds	pK _a	P _{app} , 10 ⁻⁶ cm/s	Efflux ratio	CL _r , ml/min/kg	t _{1/2} ^a , min	F ^a , %	MS rat ^b , %
3a	7.0	4.18	0.9	16	127	24	85
3b	7.5	1.42	2.0	70	66	16	77
3c	7.7	0.96	2.3	110	25	6	62

^a The compounds were administered into two groups of male rats (4 animals in each group) at 1 mg/kg iv (group 1) and 10 mg/kg po (group 2). Eight blood samples were taken from each animal. The aliquots of the plasma samples were transferred, precipitated with methanol containing a structural analogue of the analyte as internal standard and analyzed by HPLC-MS/MS for any taken time point to determine time/concentration values. The pharmacokinetic parameters were calculated by using Winnonlin version 5.1.

^b % Remaining after 1 h. The stability of the compounds at 1 μM in microsomal preparations was determined as described in Ref. 20. Microsomal preparations from different species were obtained from Tebu-bio, Offenbach, Germany.

Table 3
IC₅₀ values of **3d–g** for integrins α5β1, αvβ3, αvβ5, αIIbβ3 in competitive integrin binding assays

Compds	R	IC ₅₀ ^a (nM)				hMS ^b , %
		α5β1	αvβ3	αvβ5	αIIbβ3	
3d		1.6	239	1870	77,700	59
3e		2.2	648	11,500	87,200	52
3f		2.7	402	8290	105,000	69
3g		2.0	1050	11,700	71,700	105

^a Values are means of three experiments.

^b % Remaining after 1 h. The stability of the compounds at 1 μM in microsomal preparations was determined as described in Ref. 20. Microsomal preparations from different species were obtained from Tebu-bio, Offenbach, Germany.

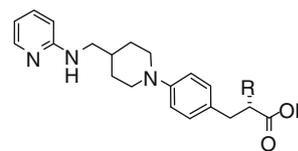
Table 4
IC₅₀ values of the stereoisomers of **3g** for integrins α5β1, αvβ3, αvβ5, αIIbβ3 in competitive integrin binding assays

Compds	IC ₅₀ ^a , nM				Cell
	α5β1	αvβ3	αvβ5	αIIbβ3	
3g	2.0	10,450	11,700	71,700	76
(S)- 3g	0.90	614	12,400	121,000	21
(R)- 3g	351	16,600	15,600	159,000	6912

^a Values are means of three experiments.

and αvβ5 binding and the torsion angles was found. It could be discussed that the missing NH-donor in **3i** is responsible for the low integrin α5β1 activity, but we assume that the twisted binding conformation of the 2,6-substituted benzamide of **3g** is directly correlated with the high activity on integrin α5β1 and the selectivity toward integrin αvβ3.

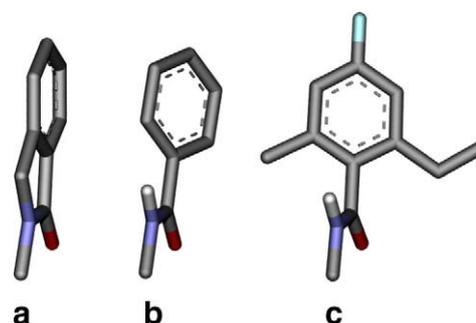
Compound (S)-**3g** was further characterized in male beagle dogs. Following an iv bolus injection of (S)-**3g** at 10 mg/kg, the systemic clearance was found to be 11 mL/min/kg, which corre-

Table 5
IC₅₀ values of (S)-**3g** and **3h–i** for integrins α5β1, αvβ3, αvβ5, αIIbβ3 in competitive integrin binding assays

Compds	R	IC ₅₀ ^a (nM)				Torsion ^b (°)
		α5β1	αvβ3	αvβ5	αIIbβ3	
(S)- 3g		0.90	614	12,400	121,000	57
3h		70	4.3	32.1	14,800	28
3i		234	13.4	3320	76,700	0

^a Values are means of three experiments.

^b Torsion angles between amid and phenyl plane (Fig. 2).

**Figure 2.** Torsion angles of N-methylamide and phenyl planes. (a) **3i** = 0°; (b) **3h** = 28°; (c) **3g** = 57°.

sponded to 37% of dog hepatic blood flow.¹⁹ The mean value for elimination half-life (t_{1/2}) was 215 min while the bioavailability after oral administration of an aqueous solution was found to be 11%. The observed kinetics of (S)-**3g** represents a solid foundation for further investigations.

In summary, the SAR and pharmacokinetic properties of N-phenyl piperidine scaffolds as integrin antagonists are described. Modulating pharmacokinetic properties with the substitution of the 2-aminopyridine and selectivities by benzamide substitution pattern

led to compound (S)-**3g** with overall acceptable properties for further development.

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