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ACS Catal., Just Accepted Manuscript • DOI: 10.1021/acscatal.8b01646 • Publication Date (Web): 09 Jul 2018

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Divergent Catalytic Approach from Cyclic Oxime Esters to Nitrogen-Containing Heterocycles with Group 9 Metal Catalysts

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Supporting Information Placeholder

ABSTRACT: We report the divergent catalytic transformation of alkene-tethered isoxazol-5(4*H*)-ones by using rhodium and cobalt catalysts to afford 2*H*-pyrroles (with Rh catalyst) and azabicyclic cyclopropanes (with Co catalyst). The rhodium-catalyzed 2*H*-pyrrole formation involving hydrogen shift is supported by deuterium-labeling experiments. The control experiments in the cobalt-catalyzed reaction indicate that the bicyclic aziridines as the primary product undergo a skeletal rearrangement assisted by metal iodide salts.

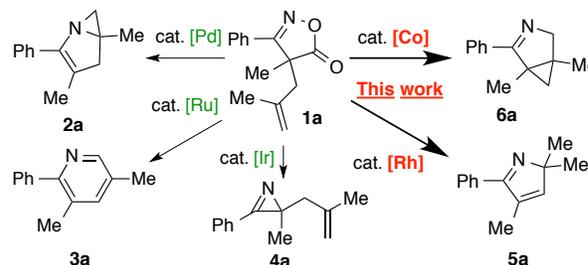
Keywords: divergent catalysis, rhodium, cobalt, N-heterocycles, decarboxylation

1. INTRODUCTION

Divergent catalytic transformation of common substrates toward various synthetic scaffolds through the common intermediates by changing transition metals^{1,2} or by using different ligands^{3,4} is an efficient and powerful strategy as well as a more challenging issue than the usual synthetic strategy aiming at an individual approach to each target. Such kinds of divergent approaches are considered to be effective in the synthesis of various nitrogen-containing heterocycles, which are privileged structural motifs in biologically active molecules, including synthetic drugs.^{1,3} Recently, the generation of nitrogen-centered active species generated by catalytic N–O bond cleavage of oxime esters has emerged as one of the most attractive methods for synthesizing various nitrogen-containing heterocycles.^{5,6} We have recently found that isoxazol-5(4*H*)-ones (isoxazolones) could act as the alkenylnitrene equivalent (Scheme 1), which is followed by palladium-catalyzed cyclization with the internal olefin to form a bicyclic aziridine.^{7,8} During the course of studies, we also found different cyclization reactions of the same isoxazolones, catalyzed by iridium and ruthenium, providing 2*H*-azirines and pyridines,

respectively.^{9,10} We report herein the divergence of these nitrene-transfer reactions by employing rhodium and cobalt catalysts, affording two different nitrogen-containing heterocycles, 2*H*-pyrroles and azabicyclo[3.1.0]hex-2-enes.

Scheme 1. Divergent synthesis of nitrogen-containing heterocycles by transition metal-catalyzed transformation of alkene-tethered isoxazol-5(4*H*)-ones.



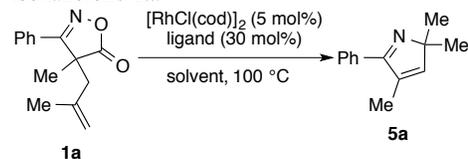
2. RESULTS AND DISCUSSION

2.1. Rhodium-catalyzed 2*H*-Pyrrole Formation

At the outset of this study, we examined the reaction of the model substrate **1a** with a catalyst precursor ($[\text{RhCl}(\text{cod})]_2$, 10 mol% per Rh) and various phosphine ligands in 1,4-dioxane at 100 °C for 5 h (Table 1). Trialkylphosphines $\text{P}(\text{alkyl})_3$ (alkyl = Me, Cy, *t*-Bu) did not work as ligands for 2*H*-pyrrole formation (entries 1–3). When we employed PPh_3 as a ligand, 2*H*-pyrrole **5a** was obtained as the corresponding decarboxylative cyclization product in 67% yield (entry 4). We next examined various triarylphosphine and bisphosphine ligands under similar conditions. Among the various bisphosphine ligands examined, dppf exhibited the highest selectivity for 2*H*-pyrrole formation, 2*H*-pyrrole **5a** being obtained in 74% yield (entries 5–8). When the electronic effect of the triarylphosphine ligands was examined, the use of triarylphosphine with electron-deficient trifluoromethyl group $\text{P}(\text{4-CF}_3\text{C}_6\text{H}_4)_3$ resulted in very low conversion (entry 9). This result makes a striking contrast to the palladium-catalyzed aziridination, in which $\text{P}(\text{4-CF}_3\text{C}_6\text{H}_4)_3$ is the most effective ligand for obtaining bicyclic aziridine **2a** in the highest yield.^{7a} Finally, we found the electron-rich

triarylphosphine, P(4-MeOC₆H₄)₃, was the best ligand in the selective formation of 2*H*-pyrroles (entry 10). Reducing the amount of catalyst loading to 5 mol% Rh diminished the yield of 2*H*-pyrrole significantly in 1,4-dioxane (entry 11). The yield was increased to 84% (73% isolated yield) when acetonitrile was used as the solvent (entries 12–14).

Table 1. Rhodium-catalyzed formation of 2*H*-pyrrole **5a** from isoxazolone **1a**.^a

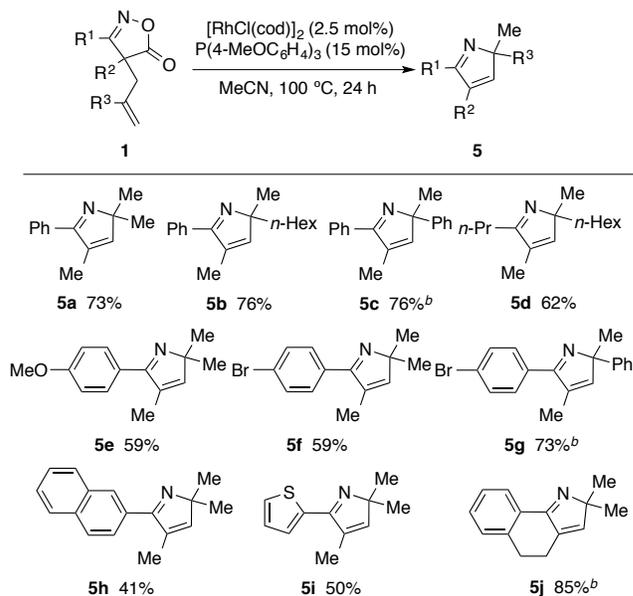


entry	ligand	solvent	time (h)	conv (%) ^b	yield (%) ^b
1	PMe ₃	1,4-dioxane	5	31	2
2	PCy ₃	1,4-dioxane	5	0	0
3	P(<i>t</i> -Bu) ₃	1,4-dioxane	5	0	0
4	PPh ₃	1,4-dioxane	5	99	67
5 ^c	dppe	1,4-dioxane	5	88	38
6 ^c	dppb	1,4-dioxane	5	86	36
7 ^c	dppf	1,4-dioxane	5	96	74
8 ^c	<i>rac</i> -binap	1,4-dioxane	5	100	33
9	P(4-CF ₃ C ₆ H ₄) ₃	1,4-dioxane	5	9	2
10	P(4-MeOC ₆ H ₄) ₃	1,4-dioxane	5	100	81
11 ^d	P(4-MeOC ₆ H ₄) ₃	1,4-dioxane	24	73	44
12 ^d	P(4-MeOC ₆ H ₄) ₃	THF	24	100	69
13 ^d	P(4-MeOC ₆ H ₄) ₃	DME	24	100	77
14 ^d	P(4-MeOC ₆ H ₄) ₃	MeCN	24	100	84 (73) ^e

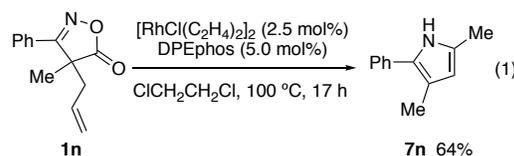
^a The reaction was carried out with isoxazolone **1a** (0.20 mmol), [RhCl(cod)]₂ (5 mol%), and ligand (30 mol%) in 1,4-dioxane (2.0 mL) at 100 °C. ^b Determined by ¹H NMR spectra of the crude products. ^c 10 mol% of the ligand was used. ^d [RhCl(cod)]₂ (2.5 mol%) and ligand (15 mol%) were used. ^e The yield of isolated product.

The rhodium-catalyzed 2*H*-pyrrole formation reaction can be applied to several substrates bearing a variety of substituents (Table 2). When methyl, *n*-hexyl, and phenyl groups were employed as R³ substituents, the corresponding 2*H*-pyrroles were obtained in good yields (**5a–c**). Isoxazolone **1d** having all alkyl substituents for R¹, R², and R³ afforded 2*H*-pyrrole **5d** in 62% yield. Isoxazolones bearing various aromatic substituents on R¹, such as *p*-methoxyphenyl, *p*-bromophenyl, naphthyl, and thienyl groups also afforded 2*H*-pyrroles **5e–i** in good yields. Tricyclic product **5j** was also obtained from tetralin-fused isoxazolone **1j**. The reaction of isoxazolone **1n**, which possesses an allyl group (R³ = H) gave 1*H*-pyrrole **7n** in place of 2*H*-pyrrole in good yield, when the reaction was carried out with [RhCl(C₂H₄)₂]₂ and DPEphos in dichloroethane at 100 °C, 17 h (eq 1).

Table 2. Rhodium-catalyzed reaction of isoxazol-5(4*H*)-one **1** giving 2*H*-pyrrole **5a**.^a

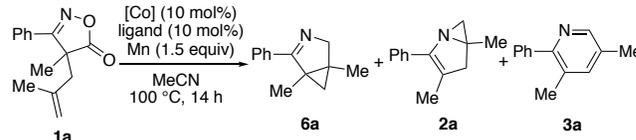


^a Reaction conditions: isoxazolone **1** (0.20 mmol), [RhCl(cod)]₂ (2.5 mol%), and P(4-MeOC₆H₄)₃ (15 mol%) in acetonitrile (2.0 mL). ^b [RhCl(cod)]₂ (5 mol%) and P(4-MeOC₆H₄)₃ (30 mol%) were used.



2.2. Cobalt-catalyzed Azabicyclic Cyclopropane Formation

Table 3. Cobalt-catalyzed decarboxylative transformation of isoxazolone **1a**.^a



entry	[Co]	ligand	conv (%) ^b	6a (%) ^b	2a (%) ^b	3a (%) ^b
1	CoI ₂	dppe	100	66 (60)	0	9
2	CoI ₂	dppbz	100	66	0	5
3	CoI ₂	dppp	57	14	0	1
4	CoI ₂	dppf	100	30	0	6
5	CoI ₂	DPEphos	58	24	0	4
6	CoI ₂	PPh ₃	56	10	0	3
7	CoI ₂ (dppe)	none	100	67	0	7
8	CoBr ₂	dppe	79	6	32	1
9 ^c	CoI ₂	dppe	100	0	43	11

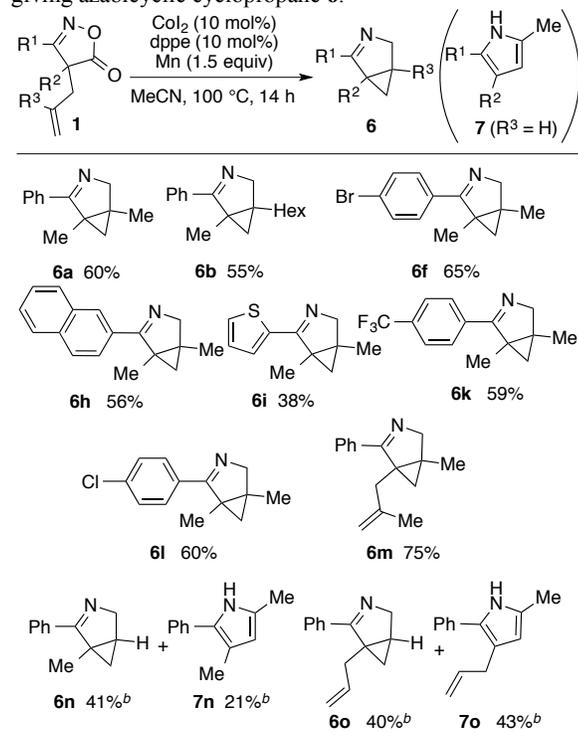
^a Reaction conditions: isoxazolone **1a** (0.20 mmol), CoI₂ (10 mol%), Mn (0.30 mmol), and ligand (10 mol%) in acetonitrile (2.0 mL) at 100 °C. ^b Determined by ¹H NMR analyses of the crude products. ^c Reaction at 60 °C.

During the exploration of catalysis with other metals, we found that a reaction of **1a** catalyzed by a low-valent cobalt-dppe complex, which was generated via reduction of CoI₂ with metallic Mn powder *in situ*,¹¹ afforded a new product **6a** with a small amount of pyridine **3a** (Table 3,

entry 1). Hence, the reaction conditions of the cobalt-catalyzed reaction of isoxazolone **1a** including ligands, temperature, and solvent were examined. Acetonitrile was also found to be the most effective solvent in the cobalt catalysis.¹² Dppe and dpbz gave the best yields of product **6a** as a result of the screening of various monophosphine and bisphosphine ligands (entries 1–6). The reaction also proceeded smoothly when a presynthesized cobalt–phosphine complex was used (entry 7). In sharp contrast, the employment of a cobalt dibromide complex instead of the iodide complex afforded the product **6a** only in 6% yield. Instead, bicyclic aziridine **2a** was obtained in moderate yield (entry 8). The selective formation of aziridine **2a** was also observed in the reaction at a lower temperature (entry 9).

With the optimized conditions for the selective formation of cyclopropanes established, a variety of **1** was tested (Table 4). Bromo, chloro, thienyl, and trifluoromethyl groups on the aromatic substituent in R¹ could be tolerated under the reaction conditions (**6f–l**). Isoxazolone **1n** and **1o** having one and two allyl groups afforded cyclopropanes **6n** and **6o**, respectively, together with 1*H*-pyrroles **7n** and **7o**.

Table 4. The cobalt-catalyzed reaction of isoxazol-5(4*H*)-one **1** giving azabicyclic cyclopropane **6**.^a



^a Reaction conditions: isoxazolone **1** (0.20 mmol), CoI_2 (10 mol%), Mn (0.30 mmol), and dppe (10 mol%) in acetonitrile (2.0 mL) at 100 °C.

2.3. Derivatization of the reaction products

2*H*-Pyrroles obtained from the rhodium-catalyzed reaction could be transformed into oxidized or reduced five-membered nitrogen-containing heterocycles by simple oxidation and reduction (Scheme 2). Heterogeneous palladium–carbon-catalyzed hydrogenation of 2*H*-pyrrole **5a** afforded pyrroline **8** selectively, while the reaction of 2*H*-pyrrole **5a** with DIBAL-H resulted in the reduction of the

C=N double bond to form pyrroline **9** selectively. Oxidation of 2*H*-pyrrole **5a** with *m*-CPBA afforded the corresponding *N*-oxide **10** in good yield. The 2*H*-pyrrole *N*-oxide derived from **5g**, like **5a**, was further converted to the platinum complex **11** in moderate yield. The structure of **11** was determined by X-ray crystallographic analysis (eq 2, Figure 1).

Azabicyclic cyclopropane **6a**, which was obtained in the cobalt-catalyzed reaction, was also applied to hydride reduction with NaBH_3CN to give the corresponding bicyclic amine **12** as a single diastereomer in good yield (eq 3).¹³

Scheme 2. Oxidation and reduction of 2*H*-pyrroles into five-membered heterocyclic compounds

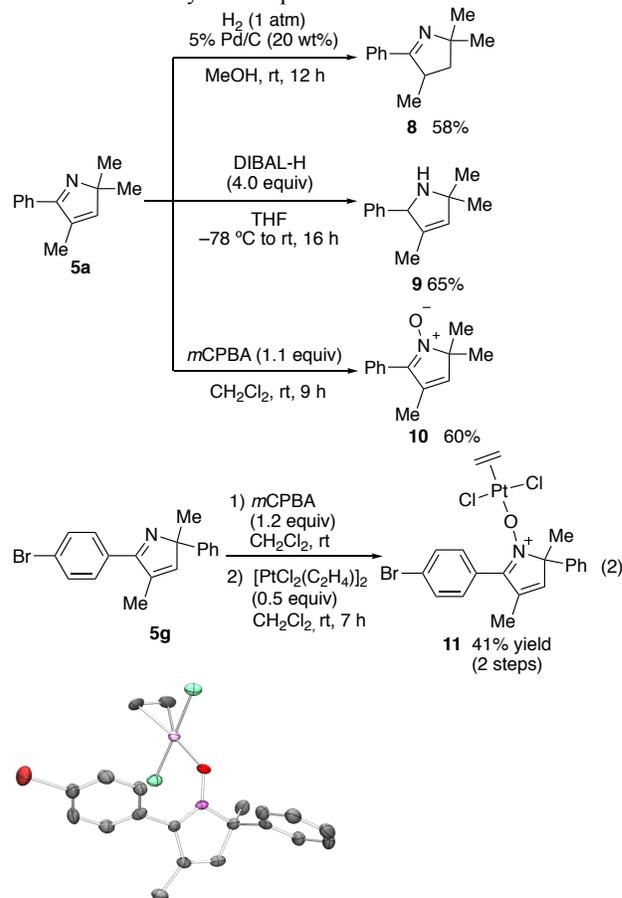
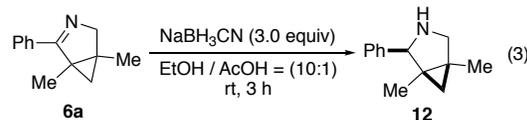


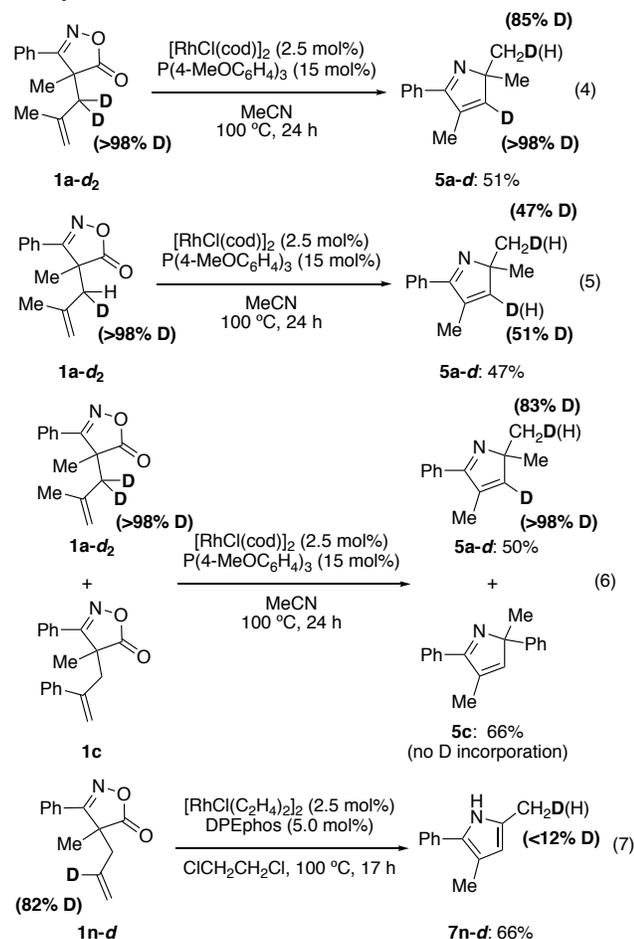
Figure 1. An ORTEP drawing of platinum complex **11**. Hydrogen atoms are omitted for clarity. CCDC 1839972.



2.4. Deuterium-labeling Experiments

Deuterium-labeling experiments were performed to gain insight into the mechanism of the rhodium-catalyzed reaction. Deuterated isoxazolone **1a-d₂** was reacted under the standard conditions to give 2*H*-pyrrole **5a-d**, in which one of the deuterium atoms was incorporated into the methyl group (85% incorporation) and the other deuterium remained attached to the original carbon (eq 4). The reaction

of monodeuterated isoxazolone **1a-d**₁ showed no significant intramolecular kinetic isotope effect (eq 5). A crossover reaction using **1a-d**₂ and **1c** in 1:1 ratio resulted in the formation of deuterated **5a-d** and **5c**, which indicates that no intermolecular exchange of hydrogen atoms occurred during the reaction (eq 6). Moreover, allyl-substituted isoxazolone **1n-d** with one deuterium at the 2-position of an allyl group was transformed into 1*H*-pyrrole **7n-d**, in which the deuterium incorporation ratio at the methyl position was less than 12% (eq 7). This result is in sharp contrast to the palladium-catalyzed formation of 1*H*-pyrroles.^{7c} The formation of 1*H*-pyrroles can be explained by assuming the prototropic shift of mechanism from that of the palladium catalyst.¹⁴



2.5. Control Experiments in Cobalt-catalyzed Reaction

Considering that the cobalt-catalyzed reaction at the lower temperature gave aziridine **2a** as a major product, generation of azabicyclic cyclopropane **6a** by cobalt catalysis would involve the skeletal rearrangement of aziridine **2a**. Therefore, we investigated several conditions of the reaction of isolated aziridine **2a** giving **6a** (Table 5). In the presence of CoI_2 , dppe, and metallic Mn, aziridine **2a** was isomerized to product **6a** in good yield, which indicates that aziridine **2a** was a key intermediate for the formation of **6a** (entry 1). The isomerization proceeded in the presence of a catalytic amount of metal iodides such as CoI_2 , ZnI_2 , SmI_2 , and LiI (entries 2–5). While Bu_4NI and ZnBr_2

did not show prominent catalytic activity toward the isomerization (entries 6 and 7), the efficacy of metal iodide salts in isomerization was obvious.

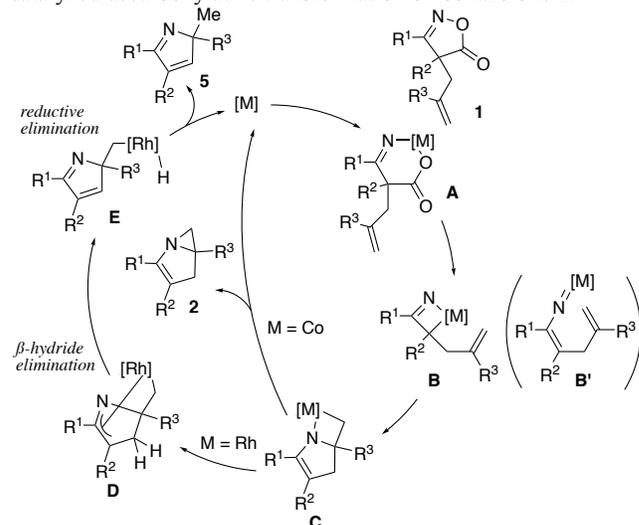
Table 5. Skeletal rearrangement of bicyclic aziridine **2a** into azabicyclic cyclopropane **6a**.^a

entry	reagents	6a (%) ^b
1	CoI_2 (10 mol%), dppe (10 mol%), Mn (1.5 equiv)	68
2	CoI_2 (10 mol%)	49
3	ZnI_2 (10 mol%)	61
4	SmI_2 (10 mol%)	72
5	LiI (10 mol%)	57
6	Bu_4NI (10 mol%)	13
7	ZnBr_2 (10 mol%)	0

^a Reaction conditions: aziridine **2a** (0.20 mmol) and reagents (as described above) in acetonitrile (2.0 mL) at 100 °C. ^b Determined by ¹H NMR analyses of the crude products.

2.6. Mechanistic Consideration

Scheme 3. A proposed catalytic cycle for rhodium- and cobalt-catalyzed decarboxylative transformation of isoxazolone **1**.

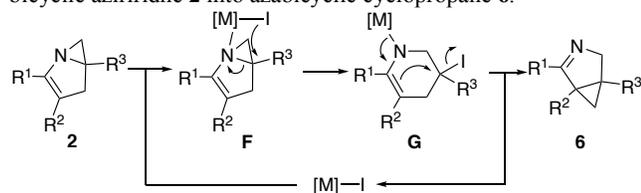


Based on the above experimental results, the proposed mechanism for rhodium-catalyzed 2*H*-pyrrole formation and cobalt-catalyzed bicyclic aziridine formation is shown in Scheme 3. Considering the catalytic cycles proposed for palladium-, ruthenium-, and iridium-catalyzed reactions, the present catalytic cycle also begins from the oxidative addition of isoxazolone **1** to the low-valent metal species to form intermediate **A** via N–O bond cleavage.^{7–10} Intermediate **A** rapidly undergoes decarboxylation to form azametallacyclobutene **B** rather than alkenylnitrene species **B'**,¹⁵ which then undergoes intramolecular cycloaddition or 1,2-insertion of olefinic moieties into the metal–nitrogen double bond to form bicyclic metallacycle **C**. In the cobalt-catalyzed reaction, C–N bond-forming reductive elimination from intermediate **C** affords bicyclic aziridine **2** as well as the regenerated catalyst, which is very similar to the palladium-catalyzed system.^{7a,16} In the rhodium-catalyzed reaction, on the other hand, intermediate **C** would isomerize into aza- η^3 -allylrhodium intermediate **D**. Then the β -hydride elimination followed by reductive elimination

would form 2*H*-pyrrole **5** as well as the regenerated catalyst. These reaction pathways are in good agreement with deuterium-labeling experiments as described above. 1*H*-Pyrrole **7** is considered as the prototropically isomerized product of 2*H*-pyrrole **5**, which has also been confirmed by the deuterium-labeling experiment described above.

It is assumed that the skeletal rearrangement of bicyclic aziridine **2** to azabicyclic cyclopropane **6** would be triggered by the coordination of the nitrogen atom to the Lewis acidic metal species,^{17,18} taking the control experiments as shown in Table 3 into consideration. Then complex **F** undergoes a C–N bond-cleaving ring expansion simultaneously with the formation of the C–I bond to form six-membered species **G**,¹⁹ which readily undergoes the C–C bond-forming recyclization to give bicyclic product **6**, regenerating the metal iodide species (Scheme 4).

Scheme 4. Proposed mechanism for skeletal rearrangement of bicyclic aziridine **2** into azabicyclic cyclopropane **6**.



3. CONCLUSION

We have explored divergence in the decarboxylative ring-reconstruction of alkene-tethered isoxazol-5(4*H*)-ones by employing rhodium and cobalt catalysts. 2*H*-Pyrroles²⁰ and azabicyclo[3.1.0]hexanes,^{13,21} both of which have been obtained in the present catalyst system, are both promising candidates of a framework for the synthesis of biologically active molecules, including drugs and natural products.^{20c,22,23} Mechanistic elucidations using deuterium-labeling experiments and control experiments have supported the mechanism for the formation of the target heterocycles. The divergence of the catalytic transformation of isoxazolones that can be switched by changing transition metals may provide a new guiding principle in the development of divergent methods for nitrogen-containing heterocyclic molecules by transition metal catalysis.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website.

Details on experimental procedures for the catalytic reactions, spectroscopic data for the products, and crystallographic data (PDF)

X-ray crystallographic structure of complex **11** (CIF)

ACKNOWLEDGMENTS

This work was financially supported by Grants-in-Aid for Scientific Research (B) (Grant No. 18H01976) and (C) (Grant No. 17K05861) from JSPS. We also thank Dr. Keisuke Nogi (Kyoto University) for his helpful advice on the formation of low-valent cobalt species.

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Notes

No competing financial interests have been declared.

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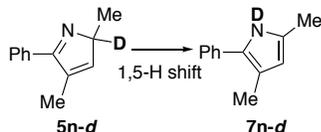
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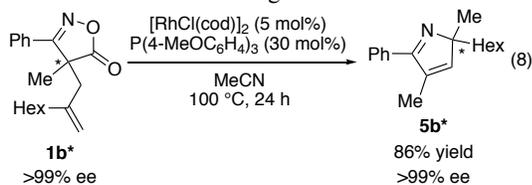
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