

Tetrahedron 54 (1998) 3833-3850

TETRAHEDRON

Synthesis of Allenic Diols by Samarium Diiodide-Promoted Coupling Between Alkynyloxiranes and Ketones

José M. Aurrecoechea*, Eva Alonso and Mónica Solay

Departamento de Química Orgánica, Facultad de Ciencias, Universidad del País Vasco, Apartado 644, 48080 Bilbao, Spain

Received 20 November 1997; revised 2 February 1998; accepted 5 February 1998

Abstract: The SmI₂-mediated reductive coupling between alkynyloxiranes and ketones provides a new route to 2,3-pentadiene-1,5-diols. The preferred stereochemistry observed in the coupling products is the result of the new C-C bond forming *anti* with respect to the opening epoxide ring. Yields and diastereoselectivities are dependent on the alkynyloxirane substitution pattern. © 1998 Elsevier Science Ltd. All rights reserved.

INTRODUCTION

The emergence of the one-electron reducing agent samarium diiodide (SmI₂) in the synthetic field has provided useful methodology for the formation of C-C bonds in processes involving radical and/or organometallic intermediates.¹

Particularly valuable are those reactions that preserve the unsaturation present in the starting materials, as the products can be further elaborated within a synthetic scheme. Thus, SmI₂ promotes the coupling of carbonyl compounds with allylic derivatives² and vinyloxiranes³ where these substrates behave, in a formal sense, as nucleophilic allylating agents. Similarly, a number of propargylic derivatives^{2b,4} behave as synthetic equivalents of the propargyl-allenyl anion synthon when treated with SmI₂ in the presence of ketones. By analogy, alkynyloxiranes, under the same reductive conditions, should provide an entry into alkoxy propargyl-allenyl anion synthons, thus significantly expanding the synthetic utility of those substrates, which are normally employed as electrophiles in C-C bond-forming processes.⁵

Therefore, it became of interest to study the possibility of using SmI_2 to effect the coupling of simple alkynyloxiranes 1 with carbonyl compounds (Scheme 1). This paper reports the stereoselective synthesis of allenic diols 2 in this fashion.⁶



Scheme 1

RESULTS AND DISCUSSION

Representative alkynyloxiranes 1a-j were selected for this study. They were obtained by epoxidation of commercial or readily available engues that were prepared using conventional procedures.



The addition of a mixture of a carbonyl compound and an alkynyloxirane 1a-j to SmI₂ in THF gave in general good yields of coupling products 2 (Scheme 1, Table 1) and also provided clean reaction mixtures, as well as a simple procedure since the end of the reaction was easily visualized by the turning of the SmI₂ characteristic blue solutions into a yellow-green suspension. A reaction temperature in the range 0-(-5°C) was found optimal. Reversing the order of addition or adding the carbonyl compound after reaction between the alkynyloxirane and SmI₂ was complete gave inferior results. In the latter case the addition of a solution of 1a to two equivalents of SmI₂ at 0°C resulted in a rapid (< 20 min) decoloration of the SmI₂ characteristic blue color but, after addition of 3-pentanone, only a 2 % yield of the allenic diol product 2e was obtained. Unreacted starting 1a and unidentified side products accounted for the remaining of the same epoxide to intractable mixtures where substantial amounts of unreacted 1a were detected by ¹H-NMR. In contrast to the related intermolecular reactions between propargylic esters and ketones, where products were found in this case. Small amounts of 1-ethynyl-2-iodocyclohexanol (3a), the result of epoxide ring opening on unreacted epoxide by iodide anion,⁸ usually accompanied the formation of allenes 2 derived from 1a.

Table 1 reveals that, while both terminal and substituted alkynes can be employed successfully, the degree of substitution at the epoxide ring appears to be crucial for the outcome of the reaction. Thus, yields from alkynyloxiranes further substituted at both epoxide termini were uniformly high, with the exception of the Ph-substituted alkyne 1c (entry 11). However, the absence of substitution at any of those positions resulted in much lower yields of coupling products and the competing reduction of the epoxide⁹ with formation of the enynes 4 (Scheme 2); in one case (entry 14), the aldol product 5 (Scheme 2) was also obtained.¹⁰

| Entry | Epoxide | R5 | R ₆ | T (°C) | t (min) | Product | Yield (%) | anti / syn ^b |
|-----------------|------------|---------------------------------|---|--------|---------|---------|-----------------|-------------------------|
| 1 | 1a | (CH ₂)5 | | -5 | 150 | 2a | 74 | 4.9 : 1 |
| 2 | 1 a | (CH ₂) ₆ | | 0 | 90 | 2 b | 77¢ | 2.3 : 1 |
| 3 | 1a | (CH ₂) ₄ | | 0 | 45 | 2c | 56 ^c | 3.3 : 1 |
| 4 | 1a | Me | Me | -5 | 180 | 2d | 81 | 5.7 : 1 |
| 5 | 1a | Et | Et | 0 | 120 | 2 e | 73¢ | 7.3 : 1 |
| 6 | 1a | Me | Ph(CH ₂) ₂ | -5 | 60 | 2f | 84 | 3.0:1 |
| 7 | 1a | Me | (CH ₂) ₂ CH=CH ₂ | 0 | 120 | 2 g | 64 ^c | d |
| 8 | 1a | Me | (CH ₂) ₃ CN | 0 | 80 | 2h | 64c | d |
| 9 | 1a | n-Hex | (CH ₂) ₁₀ CO ₂ Me | 0 | 90 | 2 i | 67° | 4.5 : 1 |
| 10 | 1 b | | (CH ₂) ₅ | -5 | 90 | 2j | 7 7 | > 50 : 1 ^e |
| 11 | 1 c | | (CH ₂) ₅ | -5 | 45 | 2 k | 38 | 2.2:1 |
| 12 ^f | 1 d | | (CH ₂)5 | -5 | 5 | 21 | 50g | |
| 13 | 1 e | | (CH ₂)5 | -5 | <1 | 4e | 76 ^h | |
| 14 | 1 f | Et | Et | -5 | 270 | 2 m | 28 ⁱ | 1.3 : 1 |
| 15 | 1 g | | (CH ₂)5 | -5 | 90 | 2n | 43j | 1 .2 : 1 |
| 16 | 1h | | (CH ₂) ₅ | -5 | 60 | 20 | 9 6 | 1.2 : 1 |
| 17 | 1i | | (CH ₂)5 | -5 | 120 | 2 p | 89 | 3.3 : 1 |
| 18 k | 1j | | (CH ₂)5 | -5 | 210 | 2q | 74 ¹ | 4.5 : 1 ^m |

Table 1. Reductive Coupling between Alkynyloxiranes and Ketones^a.

^a Unless otherwise specified, the General Procedure was used. See Experimental. ^b Values obtained from ¹H NMR or HPLC (entries 17,18) data taken on the purified products. ^c Also isolated was 3a in 3-7% yield. ^d Precise ratio could not be determined. ^e A single isomer to the limit of detection of NMR and GC. ^f Reaction run with three equivalents of SmI₂. ^gAlso isolated were 1d (5%), 3d (4%) and 4d (5%) (Scheme 2). ^h See Scheme 2. ⁱ Also isolated were 1f (21%), 4f (8%) and 5 (9%) (Scheme 2). ^j Also isolated was 6 (20%) (Scheme 3). ^k Reaction run with five equivalents of cyclohexanone. ¹ Combined yield of 2q and 8 (see Scheme 4). ^m Diastereomeric ratio is the result of assuming that diol 8 comes exclusively from *anti-*2q.





The coupling reaction is tolerant of some useful functionality, both in the carbonyl substrate (entries 7-9) and in the alkynyloxirane (entries 12, 16-18). One exception is the hydroxyl group of substrate 1g (entry 15; Scheme 3) which interferes with the coupling presumably by protonation of an organosamarium derived from the alkynyloxirane (*vide infra*). In marked contrast to ketones, aldehydes perform poorly in the coupling reaction. As previously noted,¹¹ the aldehyde carbonyl is comparatively very reactive towards SmI₂ and side reactions often preclude effective couplings.^{4a,d} Thus, the coupling between 1a and 3-phenylpropanal proceeded in very low yield and no other aldehyde was tested.



Scheme 3

The reaction of the silyl ether 1j with cyclohexanone has some mechanistic interest. When performed under the standard conditions, with equimolar amounts of 1j and ketone, a low yield of coupling product 2q was obtained, being the dimer 7 the major product of the reaction (Scheme 4; Table 2). Also isolated were the enyne 4j and some unreacted 1j (4%). The formation of the dimer was dependent on the 1j/cyclohexanone ratio and could be totally suppressed by increasing the amount of cyclohexanone to at least 4 equivalents relative to the oxirane 1j (entries 3,4; Table 2). In these cases, small amounts of the uncoupled allene 9 and the diol 8 (the result of the migration of the silyl group to the adjacent hydroxyl group)¹² were also formed.



Scheme 4

Table 2. Reductive Coupling of 1j. Effect of the Relative Amount of Cyclohexanone

| Entry | nª | 4j (%) | 7 (%) | 2q (%) | 9 (%) |
|-------|----|--------|-------|--------|-------|
| 1 | 1 | 10 | 32 | 27 | _ |
| 2 | 2 | 26 | 21 | 42 | _ |
| 3 | 4 | <7 | - | 69b | 10 |
| 4 | 5 | _ | - | 74b | - |

^a n = Molar ratio of cyclohexanone to 1j. ^b Combined yield 2q+8.

The formation of the dimer 7 and its suppression in the presence of excess cyclohexanone are interpreted as an indication that resonance-stabilized radicals 10 (Scheme 5), produced in the one-electron transfer from SmI₂ to the starting alkynyloxirane,¹³ are involved in the coupling. Thus, for substrate 1j, the formation of product 2q must proceed, at least partially, by coupling between a ketyl radical anion 12 and 10. In this scenario, dimerization of 10 leading to 7 is the dominant pathway unless sufficient amounts of the ketyl 12 are present. However, these results do not necessarily rule out an alternative pathway involving carbonyl addition of an organosamarium 11 derived from 10. The formation of 6 in the reaction of the hydroxy-substituted substrate 1g (Scheme 3) indicates that, to some extent, organosamarium species are also involved.¹⁴ Given these overall results and literature precedence in SmI₂-mediated couplings involving ketones,¹⁵ it is likely that both mechanisms could be operative in general.



Scheme 5

Stereochemistry of the Coupling.

In general, the SmI₂-promoted between alkynyloxiranes and ketones took place with low to moderate diastereoselectivity. Monocyclic substrates without bulky substituents gave coupling products with low diastereoselectivity (see entries 14-16 in Table 1) probably as a result of the conformational flexibility of the presumed intermediates 10 or 11 (Scheme 5). The rigidity imposed by the added ring in bicyclic substrates 1a-c results in moderate axial diastereoselectivities. Bicyclic substrate 1b, with a methyl substituent at the alkynyl terminus (entry 10, Table 1) was exceptional in that a single isomer, as judged by NMR and capillary GC, was obtained. With substrates 1g-j (entries 15-18, Table 1) the nature of the hydroxyl protecting group was also found to exert a significant effect on diastereoselectivity. Thus, no selectivity was found with hydroxy or benzyloxy substituents whereas the reactions of substrates with a bulkier sililoxy substituent were moderately selective. The couplings using a prochiral ketone (entries 6-9, Table 1) produced all the four possible isomers, with two of them predominating over the others in a ratio within the range of related couplings. This indicates that the carbonyl addition proceeds with no facial selectivity.

The stereochemistry of the major product is the result of the new C-C bond forming *anti* to the opening epoxide. Therefore, the formation of allenic diols with this new SmI₂-promoted coupling is stereochemically complementary to the alternative organocuprate S_N2' displacements on alkynyloxiranes bearing a propargylic hydroxyl group.^{5e} Stereochemical assignments followed from chemical correlations of the coupling products **2d**, **j**, **q** with the corresponding diols obtained by stereochemically well defined literature procedures. The rest of diols were given stereochemical assignments by analogy with these cases. Thus, the major product obtained in the LAH reduction of the epoxypropargyl alcohol **13** (Scheme 6) was identical to the diol **2d** predominating in the SmI₂-promoted coupling reaction. Since the LAH reduction of propargyl alcohols related to **13** has been reported to lead to products resulting from overall *syn* hydride attack,¹⁶ our major product is the result of the new group entering *anti* with respect to the opening epoxide. The methyl-substituted allenes **2j** and **2q** were similarly correlated with the products obtained by the cuprate S_N2' displacements on propargyl alcohols **14** and **16**. This known *anti* process^{5e} afforded the minor (*syn*) diastereomers of **2j** (not obtained in the SmI₂-

promoted coupling) and 2q, respectively, thus confirming the preferred *anti* stereocourse of the coupling.¹⁷ Also obtained in the cuprate reactions of 14 and 16 were the alkynyl diols 15 and 17, respectively, derived from direct cuprate addition to the epoxide, as well as the allenic diols 2a and 2p (diastereomeric mixtures), the result of alkynyloxirane reduction under the reaction conditions.^{5e}





In conclusion, the SmI₂-promoted reductive coupling between alkynyloxiranes and ketones provides a new strategy for the synthesis of penta-2,3-diene-1,5-diols with moderate *anti* diastereoselectivity. This stereochemical preference is complementary to that observed using alternative literature methodology. In general, effective couplings are realized when the alkynyloxirane substitution pattern favors electron transfer over iodide-promoted ring-opening.

Experimental.

General. All reactions involving air- and moisture-sensitive materials were conducted under an atmosphere of dry Ar. THF was freshly distilled from sodium/benzophenone and, for reactions with SmI₂, it was deoxygenated prior to use. Dichloromethane, and DMF were distilled from CaH₂ and stored over 4 Å molecular sieves. The carbonyl reagents used in the coupling reactions were distilled immediately prior to use. Organic extracts were dried over Na₂SO₄, filtered and evaporated under reduced pressure. Flash column chromatography²⁰ was performed on silica gel (230-400 mesh). HPLC purifications were carried out with either LiChrosorb Si60 (7µm, 25 x 2.5 cm) (column 1) or µPorasil (10µm, 19 x 1.5 cm) (column 2) columns. ¹H and ¹³C RMN spectra were obtained in CDCl₃ at 250 MHz and 62.9 MHz, respectively. IR data include only characteristic absorptions. Mass spectra were obtained at 70 eV. GC-MS analysis were performed at 70-

280 °C (20 °C/min) with a stationary phase of methylphenylsilicone (0.25 μm, 30 m x 0.25 mm). Epoxides 1a-c were prepared according to literature procedures.^{5a,21}

General Epoxidation Procedure. In a typical experiment, to a solution of the appropriate enyne (47.0 mmol) in CH₂Cl₂ (182 mL) at 0°C was added MCPBA (70.5 mmol). The solution was stirred for 30 min, allowed to warm to room temperature and further stirred for 13 h. 1M NaOH (100 mL) was added and the layers separated. The aqueous layer was extracted with CH₂Cl₂ (50 mL) and the combined organic extracts washed with 1M NaOH (2 x 50 mL). The residue after evaporation was purified as indicated for the individual cases.

4-(tert-Butyldiphenylsilyloxy)but-1-yne. Tert-Butyldiphenylsilyl chloride (4.31 g, 15.0 mmol) was added dropwise at room temperature to a solution of but-3-yn-1-ol (1.00 g, 14.0 mmol) and imidazole (2.13 g, 31.0 mmol) in DMF (8.4 mL) under Ar. After stirring the solution for 30 min water (10 mL) was added, the mixture was stirred further 20 min, diluted with CH₂Cl₂ (10 mL) and the aqueous layer extracted with CH₂Cl₂ (3 x 10 mL). The combined organic layers were extracted with H₂O (5 x 5 mL). The crude after evaporation was purified by flash chromatography (5:95 EtOAc/hexanes) to yield the title compound (4.40 g, 100%) as a colorless oil: ¹H NMR δ 1.10 (s, 9H, (CH₃)₃-C), 1.97 (t, J = 2.5 Hz, 1H, H-1), 2.50 (td, J = 7.0, 2.5 Hz, 2H, H-3), 3.82 (t, J = 7.0 Hz, 2H, H-4), 7.3-7.5 (m, 6H), 7.7 (m, 4H); ¹³C NMR δ 19.2, 22.6, 26.7, 62.3, 69.3, 81.4, 127.6, 129.7, 133.5, 135.5; IR (neat) v 3300, 2110, 1115 cm⁻¹; HRMS calcd for C₂₀H₂₄OSi 308.159644, found 308.159181.

6-(tert-butyldiphenylsilyloxy)-2-hexylhex-1-en-3-yne (4d). Cacchi's procedure²² was followed: Pd(PPh₃)₂(OAc)₂ (65 mg, 0.08 mmol) was added to a solution of 4-(tert-butyldiphenylsilyloxy)but-1-yne (1.75 g, 5.7 mmol) and oct-1-en-2-yl triflate²³ (93%, 1.00 g, 4.1 mmol) in DMF (14 mL) and tributylamine (20 mL) at 70°C under Ar and the solution was maintained at the same temperature for 1 h. After cooling to room temperature, H₂O (30 mL) was added and the resulting mixture was stirred for 20 min and the layers separated. The organic layer was washed with H₂O (5 x 10 mL). The residue after evaporation was purified by flash chromatography (2.5:97.5 EtOAc/hexanes) to yield **4d** (1.83 g, 59%) as an oil: ¹H NMR δ 0.89 (t, J = 6.5 Hz, 3H, CH₃), 1.08 (s, 9H, (CH₃)₃-C), 1.3 (m, 6H), 1.4-1.5 (m, 2H), 2.12 (t, J = 7.4 Hz, 2H, C₂-CH₂), 2.61 (t, J = 6.9 Hz, 2H, H-5), 3.81 (t, J = 6.9 Hz, 2H, H-6), 5.15 (t, J = 0.96 Hz, 1H, H-1), 5.23 (d, J = 1.7 Hz, 1H, H-1), 7.3-7.4 (m, 6H), 7.7-7.8 (m, 4H); ¹³C NMR δ 14.1, 19.2, 22.6, 23.5, 26.7, 28.0, 28.6, 31.6, 37.4, 62.5, 82.1, 86.7, 119.8, 127.6, 129.6, 132.2, 133.6, 135.5; IR (CHCl₃) v 2210, 1610, 1110 cm⁻¹. Anal. Calcd for C₂₈H₃₈OSi: C, 80.33; H, 9.16. Found: C, 80.08; H, 9.17.

6-(tert-Butyldiphenylsilyloxy)-1,2-epoxy-2-hexylhex-3-yne (1d). Enyne 4d (0.80 g, 1.91 mmol) was subjected to the General Epoxidation conditions to afford, after flash chromatography (5:95 EtOAc/hexanes) and HPLC (Column 1, 9 mL/min, 5:95 EtOAc/hexanes, $t_R = 16$ min) the epoxide 1d (0.65 g, 78%) as an oil: ¹H NMR δ 0.90 (distorted t, 3H, CH₃), 1.06 (s, 9H, (CH₃)₃-C), 1.3 (m, 6H), 1.5-1.7 (m, 4H), 2.48 (t, J = 6.9 Hz, 2H, H-5), 2.70 (d, J = 5.5 Hz, 1H, H-1), 2.93 (d, J = 5.5 Hz, 1H, H-1), 3.76 (t, J = 6.9 Hz, 2H, H-6), 7.3-7.5 (m, 6H), 7.7 (m, 4H); ¹³C NMR δ 14.0, 19.2, 22.5, 22.8, 25.5, 26.7, 29.0, 31.7, 36.5, 51.1, 54.6, 62.2, 79.7, 80.8, 127.7, 129.7, 133.5, 135.5; IR (neat) v 1115 cm⁻¹. Anal. Calcd for C₂₈H₃₈O₂Si: C, 77.37; H, 8.82. Found: C, 77.46; H, 8.97.

(Z)-7-(tert-butyldiphenylsilyloxy)hept-2-en-4-yne (Z-4e). A solution of 4-(tertbutyldiphenylsilyloxy)but-1-yne (1.80 g, 6.50 mmol) in Et₂NH (7 mL) was added to a solution of (Z)-1bromopropene (0.94 g, 7.8 mmol), Pd(PPh₃)₂Cl₂ (0.11 g, 1.62 mmol) and CuI (0.12 g, 6.50 mmol) in Et₂NH (20 mL) at room temperature. The resulting solution was stirred for 15 h. Diethyl ether (15 mL) and saturated

.

NH₄Cl (10 mL) were added, the mixture was stirred for 20 min and the layers separated. The aqueous layer was extracted with diethyl ether (10 mL). The residue after evaporation was purified by flash chromatography (2.5:97.5 EtOAc/hexanes) to give a 86:14 Z/E mixture of (4e) (1.40 g, 70%) as a colorless oil. A second chromatographic separation (1:99 EtOAc/hexanes) afforded a sample (0.70 g) enriched in the Z-isomer (93:7): ¹H NMR δ 1.06 (s, 9H, (CH₃)₃-C), 1.83 (dd, J = 6.8, 1.7 Hz, 3H, H-1), 2.62 (td, J = 7.0, 2.0 Hz, 2H, H-6), 3.81 (t, J = 7.0 Hz, 2H, H-7), 5.4-5.5 (m, 1H, H-3), 5.90 (dq, J = 10.6, 6.8 Hz, 1H, H-2), 7.3-7.5 (m, 6H), 7.7 (m, 4H); ¹³C NMR δ 15.7, 19.2, 23.7, 26.7, 62.7, 78.2, 91.5, 110.2, 127.6, 129.6, 133.6, 135.5, 137.3; IR (CHCl₃) v 2200 1665, 1110 cm⁻¹.

 $(5S^*, 6R^*)$ -1-(tert-Butyldiphenylsilyloxy)-5,6-epoxyhept-3-yne (1e). Enyne 4e (93% Z, 600 mg, 1.72 mmol) was epoxidized according to the General Procedure to afford, after flash chromatography (5:95 EtOAc/hexanes), the epoxide 1e (401 mg, 64%) as a colorless oil: ¹H NMR δ 1.06 (s, 9H, (CH₃)₃-C), 1.38 (d, J = 5.2 Hz, 3H, H-7), 2.50 (td, J = 6.9, 1.6 Hz, 2H, H-2), 3.11 (dq, J = 5.2, 4.0 Hz, 1H, H-6), 3.39 (dt, J = 4.0, 1.6 Hz, 1H, H-5), 3.76 (t, J = 6.9 Hz, 2H, H-1), 7.3-7.5 (m, 6H), 7.7 (m, 4H); ¹³C NMR δ 14.6, 19.2, 22.9, 26.7, 45.7, 53.9, 62.2, 76.0, 83.5, 127.7, 129.7, 133.5, 135.5; IR (neat) v 2230, 1110 cm⁻¹; HRMS calcd for C₁₉H₂₈O₂Si (M-⁴Bu) 307.11543, found 307.11399.

(E)-Dec-3-en-1-yne (E-4f). Potassium tert-butoxide (4.70 g, 41.0 mmol) was added to a suspension of iodomethyltriphenylphosphonium iodide²⁴ (11.00 g, 20.0 mmol) in THF (300 mL) at room temperature. The resulting solution was cooled to -78°C and (E)-non-2-enal (2.90 g, 20.0 mmol) in THF (43 mL) was added. The mixture was stirred at the same temperature for 2 h and then allowed to warm to room temperature over 90 min. After adding hexanes (50 mL) the mixture was extracted with brine (2 x 20 mL) and water (10 mL). The residue after evaporation was purified by flash chromatography in hexanes to afford the enyne E-4f (1.10 g, 41%) as a volatile yellowish oil: ¹H NMR δ 0.8-0.9 (m, 3H, H-10), 1.3-1.4 (m, 8H), 2.1 (m, 2H, H-5), 2.76 (d, J = 2.0 Hz, 1H, H-1), 5.4 (dm, J = 16.0 Hz, 1H, H-3), 6.24 (dt, J = 15.9 Hz, 7.0 Hz, 1H, H-4); ¹³C NMR δ 14.0, 22.5, 28.5, 28.7, 31.6, 33.0, 75.5, 82.6, 108.4, 147.0; IR (CHCl₃) v 3300, 2100 cm⁻¹.

 $(3S^*,4S^*)$ -3,4-Epoxydec-1-yne (1f).^{5a} Enyne E-4f (0.40 g, 2.94 mmol) was epoxidized according to the General Procedure to afford, after flash chromatography in hexanes, the epoxide 1f^{5a} (0.17 g, 39%) as a volatile colorless oil.

 $(2R^*, 3S^*)$ -2,3-Epoxy-3-methylpent-4-yn-1-ol (1g).²⁵ The General Epoxidation Procedure was applied to (Z)-3-methylpent-2-en-4-yn-1-ol (1.00 g, 10.4 mmol). Flash chromatography (25:75 EtOAc/hexanes) of the crude product afforded the epoxide 1g (0.72 g, 62%) as a colorless solid: mp 40-41°C.

 $(3S^*, 4S^*)$ -5-Benzyloxy-3,4-epoxy-3-methylpent-1-yne (1h). (E)-3-Methylpent-2-en-4-yn-1-ol (10 g, 0.1 mol) was added to NaH (3.74 g, 0.15 mol) in THF (210 mL) at 0°C and the mixture was stirred for 2 h with occasional cooling while hydrogen was evolved. Benzyl bromide (23.13 g, 0.13 mol) was added, stirring continued for 48 h and the whole poured over ice/water (~ 30 g). The mixture was extracted with diethyl ether (4 x 30 mL), the combined organic extracts washed with brine (3 x 20 mL) and dried (Na₂SO₄). The residue after evaporation of the solvents was partially purified by flash chromatography (5:95 EtOAc/hexanes) to yield (E)-benzyloxy-3-methyl-3-penten-1-yne (17.2 g) which contained ~ 14% of benzyl bromide: ¹H NMR δ 1.56 (s, 3H, CH₃), 2.87 (s, 1H, acetylenic), 4.1 (d, J = 6.5 Hz, 2H, CH₂-C=C), 4.52 (s, 2H, CH₂-Ph), 6.1 (m, 1H, H-C=C), 7.2-7.4 (m, 5H). Without further purification this material (2.1 g) was subjected to the General Epoxidation conditions to afford, after flash chromatography (10:90 EtOAc/hexanes) the epoxide 1h (1.45 g, 60% over two steps) as an oil: ¹H NMR δ 1.50 (s, 3H, CH₃), 2.32 (s, 1H, H-1), 3.42 (apparent t, J = 5.2 Hz, 1H, H-4), 3.56 (dd, J = 11.3, 5.5 Hz, 1H, H-5), 3.67 (dd, J = 11.3, 5.0 Hz, 1H, H-5), 4.53 (d, J = 11.3, 5.0 Hz, 1H, H-5), 4.53 (d, J = 11.3, 5.5 Hz, 1H, H-5), 3.67 (dd, J = 11.3, 5.0 Hz, 1H, H-5), 4.53 (d, J = 11.3, 5.5 Hz, 1H, H-5), 3.67 (dd, J = 11.3, 5.0 Hz, 1H, H-5), 4.53 (d, J = 11.3, 5.5 Hz, 1H, H-5), 3.67 (dd, J = 11.3, 5.0 Hz, 1H, H-5), 4.53 (d, J = 11.3, 5.5 Hz, 1H, H-5), 3.67 (dd, J = 11.3, 5.0 Hz, 1H, H-5), 4.53 (dd, J = 11.3, 5.5 Hz, 1H, H-5), 3.67 (dd, J = 11.3, 5.0 Hz, 1H, H-5), 4.53 (d, J = 11.3, 5.5 Hz, 1H, H-5), 3.67 (dd, J = 11.3, 5.0 Hz, 1H, H-5), 4.53 (d, J = 11.3, 5.5 Hz, 1H, H-5), 3.67 (dd, J = 11.3, 5.0 Hz, 1H, H-5), 4.53 (d, J = 11.3, 5.5 Hz, 1H, H-5), 3.67 (dd, J = 11.3, 5.0 Hz, 1H, H-5), 4.53 (d, J = 11.3, 5.5 Hz, 1H, H-5), 3.67 (dd, J = 11.3, 5.0 Hz, 1H, H-5), 4.53 (d, J = 11.3, 5.5 Hz, 1H, H-5), 3.67 (dd, J = 11.3,

11.8 Hz, 1H, CH-Ph), 4.63 (d, J = 11.8 Hz, 1H, CH-Ph), 7.3-7.4 (m, 5H); ¹³C NMR δ 18.3, 49.8, 62.1, 67.5, 70.3, 73.2, 83.6, 127.7, 127.8, 128.4, 137.5; IR (neat) v 3290, 2110 cm⁻¹; HRMS calcd for C₁₃H₁₄O₂ 202.09942, found 202.09860.

 $(3S^*,4S^*)$ -5-(*tert-Butyldiphenylsilyloxi*)-3,4-epoxy-3-methylpent-1-yne (1i). The silylation procedure described above was applied to (*E*)-methylpent-2-en-4-yn-1-ol (2.50 g, 26.0 mmol). The crude after evaporation was purified by flash chromatography (3:97 EtOAc/hexanes) to yield (*E*)-5-(*tert*-**butyldiphenylsilyloxy**)-3-methylpent-3-en-1-yne (8.02 g, 92%) as a yellowish oil: ¹H NMR δ 1.08 (s, 9H, (CH₃)₃-C), 1.65 (d, *J* = 1.1 Hz, 3H, C₃-CH₃), 2.81 (s, 1H, H-1), 4.29 (d, *J* = 6.2 Hz, 2H, H-5), 6.14 (td, *J* = 6.2, 1.3 Hz, 1H, H-4), 7.4-7.5 (m, 6H), 7.7 (m, 4H); ¹³C NMR δ 17.4, 19.1, 26.7, 60.6, 74.6, 86.1, 117.8, 127.7, 129.7, 133.4, 135.5, 138.1; IR (neat) v 3280, 1110 cm⁻¹. This enyne (1.00 g, 3.0 mmol) was epoxidized according to the General Procedure and the epoxide **1i** (0.90 g, 86%) was obtained after flash chromatography (5:95 EtOAc/hexanes) as a yellowish oil. The analytical sample was obtained after HPLC purification (Column 1, 8 mL/min, 5:95 EtOAc/hexanes, t_R = 23 min): ¹H NMR δ 1.07 (s, 9H, (CH₃)₃-C), 1.35 (s, 3H, C₃-CH₃), 2.31 (s, 1H, H-1), 3.41 (t, *J* = 5.3 Hz, 1H, H-4), 3.71 (dd, *J* = 11.7, 5.3 Hz, 1H, H-5), 3.79 (dd, *J* = 11.7, 5.3 Hz, 129.5, 129.8, 132.9, 133.1, 135.5; IR (neat) v 3280, 2090, 1110 cm⁻¹; HRMS calcd for C₂₂H₂₆O₂Si 350.170209, found 350.170229. Anal. Calcd for C₂₂H₂₆O₂Si: C, 75.40; H, 7.47. Found: C, 74.94; H, 7.43.

(E)-6-(tert-Butyldiphenylsilyloxy)-4-methylhex-4-en-2-yne (4j). n-BuLi (1.3 M in hexanes, 6.9 mL, 9.0 mmol) was added dropwise to a solution of (E)-5-(tert-butyldiphenylsilyloxy)-3-methylpent-3-en-1-yne (2.50 g, 7.5 mmol) in THF (26 mL) at -78°C and, after stirring the solution for 45 min, MeI (2.65 g, 18.7 mmol) in THF (8 mL) was added. The solution was stirred at the same temperature for 30 min and allowed to reach room temperature over 2 h. Water (10 mL) was added, the aqueous layer was extracted with ether (3 x 10 mL) and the combined organic layers were washed with brine (5 mL). The crude product was purified by flash chromatography (5:95 EtOAc/hexanes) to yield enyne 4j (2.40 g, 92%) as an oil: ¹H NMR δ 1.06 (s, 9H, (CH₃)₃-C), 1.61 (s, 3H, C4-CH₃), 1.95 (s, 3H, H-1), 4.26 (d, J = 6.2 Hz, 2H, H-6), 5.95 (t, J = 6.2 Hz, 1H, H-5), 7.3-7.4 (m, 6H), 7.7 (m, 4H); ¹³C NMR δ 4.1, 17.8, 19.1, 26.7, 60.7, 82.2, 83.2, 119.3, 127.6, 129.6, 133.6, 134.9, 135.5; IR (CHCl₃) v 2240, 1725, 1115 cm⁻¹; HRMS calcd for C₂₃H₂₈OSi 348.19094, found 348.19092.

 $(4S^*, 5S^*)$ -6-(*tert-Butyldiphenylsilyloxy*)-4,5-epoxy-4-methylhex-2-yne (1j). The General Epoxidation Procedure was followed with enyne 4j (2.25 g, 6.90 mmol). Purification by flash chromatography (5:95 EtOAc/hexanes) afforded the epoxide 1j (1.80 g, 77%) as a colorless oil. The analytical sample was obtained after HPLC purification (Column 1, 8 mL/min, 5:95 EtOAc/hexanes, t_R = 20.5 min): ¹H NMR δ 1.06 (s, 9H, (CH₃)₃-C), 1.31 (s, 3H, C₄-CH₃), 1.83 (s, 3H, H-1), 3.35 (t, J = 5.3 Hz, 1H, H-5), 3.70 (dd, J = 11.6, 5.3 Hz, 1H, H-6), 3.78 (dd, J = 11.6, 5.3 Hz, 1H, H-6), 7.3-7.5 (m, 6H), 7.7 (m, 4H); ¹³C NMR δ 3.5, 18.7, 19.2, 26.7, 51.1, 62.0, 64.1, 78.5, 79.5, 127.7, 129.8, 133.0, 133.2, 135.5; IR (CHCl₃) v 2240, 1115 cm⁻¹. Anal. Calcd for C₂₃H₂₈O₂Si: C, 75.78; H, 7.75. Found: C, 75.79; H, 7.76.

SmI₂-Mediated Coupling between Alkynyloxiranes and Ketones. General Procedure. SmI₂ (ca. 0.1M in THF) was prepared²⁶ as reported from diiodoethane,²⁷ diiodomethane^{1b} or iodine^{1b}. In a typical experiment the mixture of 1 (1.14 mmol) and a ketone (1.14 mmol) in THF (6 mL) was added dropwise under Ar to SmI₂ (ca. 0.1M in THF, 2.56 mmol) at the temperature indicated in Table 1. The mixture was stirred until disappearance of the epoxide as judged by TLC or until the blue solution turned yellow-green (Table 1). The reaction mixture was poured over saturated K₂CO₃ (10 mL) and the aqueous layer was extracted with EtOAc (3 x 25 mL). The

crude product after evaporation was purified by flash chromatography as specified for the individual cases listed below.

2-[2-(1-Hydroxycyclohexyl)ethenylidene]cyclohexanol (2a). Eluent: 40:60 EtOAc/hexanes. Colorless solid: mp 105-107°C; ¹H NMR δ 1.3-1.8 (m, 15H), 1.9-2.0 (m, 2H), 2.36 (br d, J = 12.7 Hz, 1H), 3.1 (br s, 2H, OH), 4.0 (m, 1H, H-1), 5.36 (t, J = 2.9 Hz, 1H, H-C=C=C, syn-isomer), 5.41 (t, J = 2.7 Hz, 1H, H-C=C=C, anti-isomer); ¹³C NMR δ 22.5, 22.6, 22.8, 23.8, 23.9, 25.5, 25.6, 26.5, 26.8, 30.1, 30.2, 35.9, 36.0, 37.4, 37.7, 38.6, 69.0, 69.2, 70.9, 71.1, 103.2, 103.4, 110.9, 111.1, 193.7, 193.9; IR (CHCl₃) v 3350, 1975 cm⁻¹. Anal. Calcd for C₁₄H₂₂O₂: C, 75.62; H, 9.98. Found: C, 75.49; H, 10.17.

2-[2-(1-Hydroxycycloheptyl)ethenylidene]cyclohexanol (2b). Elution with 10:90 EtOAc/hexanes provided iodide 3a (6%). Further elution with 25:75 EtOAct/hexanes afforded the diastereometric mixture of 2b as a yellowish oil. The diastereomers were separated by HPLC (Column 1, 10 mL/min, 35:65 EtOAc/hexanes). $(15^*, 1'S^*)$ -Isomer (anti-2b): t_R = 43 min; colorless solid; mp 98-99°C; ¹H NMR δ 1.2-2.1 (m, 19H), 2.4 (br d, J = 12.9 Hz, 1H, H-3), 3.0 (br s, $W_{1/2} = 43.8$ Hz, 2H, OH), 4.0 (m, 1H, H-1), 5.45 (t, J = 2.8Hz, 1H, H-C=C=); ¹³C NMR δ 22.0, 22.3, 23.8, 26.5, 29.3, 30.2, 35.9, 41.1, 41.7, 69.1, 74.9, 104.5, 111.1, 192.8; IR (neat, diastereomeric mixture) v 3650-3100, 1970 cm⁻¹. Anal. Calcd for C₁₅H₂₄O₂: C, 76.23; H, 10.23. Found: C, 75.92; H, 10.37. (15*,1' R^*)-Isomer (syn-2b): $t_R = 52$ min; oil; ¹H NMR δ 1.3-2.1 (m, 19H), 2.4 (br d, J = 13.5 Hz, 1H, H-3), 2.5 (br s, 2H, OH), 4.0 (m, 1H, H-1), 5.46 (t, J = 3.3 Hz, 1H, H-C=C=); 13 C NMR § 22.2, 22.5, 23.9, 27.0, 29.4, 30.2, 36.2, 41.3, 41.7, 69.2, 74.9, 105.2, 111.5, 192.6; HRMS calcd for C15H24O2 236.17763, found 236.17818. Data for 1-ethynyl-2-iodocyclohexanol (3a).: ¹H NMR & 1.2-1.4 (m, 1H), 1.5-1.9 (m, 4H), 2.1-2.4 (m, 3H), 2.61 (s, 1H), 2.74 (s, 1H), 4.17 (dd, J = 12.3, 4.3 Hz, 1H, H-2); ¹³C NMR δ 23.3, 28.1, 37.5, 37.9, 45.4, 72.6, 74.4, 84.7; IR (neat) v 3420, 3280, 2100 cm⁻¹; MS (EI) m/z (%) 250 (9, M), 207 (10), 128 (15), 127 (85), 123 (base, M-I), 105 (22), 103 (13), 95 (77), 94 (14), 93 (19), 91 (13), 81 (60), 79 (36), 78 (12), 77 (36), 69 (18), 68 (14), 67 (78), 66 (19), 65 (17), 63 (11), 55 (68), 54 (12), 53 (76), 51 (28).

2-[2-(1-Hydroxycyclopentyl)ethenylidene]cyclohexanol (2c). Elution with 10:90 EtOAc/hexanes provided iodide 3a (3%). Further elution with 25:75 EtOAc/hexanes afforded the diastereomeric mixture of 2c as a yellowish oil. The diastereomers were separated by HPLC (Column 1, 10 mL/min, 75:25 EtOAc/hexanes). (15*,1'5*)-Isomer (anti-2c): $t_R = 43$ min; pale yellow solid; mp 95-96°C; ¹H NMR & 1.2-1.5 (m, 3H), 1.6-2.0 (m, 10H), 2.0-2.1 (m, 2H), 2.4 (br d, J = 13.0 Hz, 1H, H-3), 2.7 (br s, $W_{1/2} = 34.3$ Hz, 2H, OH), 4.0-4.1 (m, 1H, H-1), 5.55 (t, J = 3.0 Hz, 1H, H-C=C=); ¹³C NMR & 23.5, 23.6, 23.9, 26.6, 30.3, 36.0, 40.0, 40.5, 69.2, 80.4, 103.0, 111.4, 192.6; IR (neat, diastereomeric mixture) v 3500-3200, 1970 cm⁻¹; HRMS calcd for C₁₃H₂₀O₂ 208.14633, found 208.14605. (15*,1'R*)-Isomer (syn-2c): $t_R = 51$ min, oil; ¹H NMR & 1.3-1.5 (m, 3H), 1.7-2.0 (m, 10H), 2.0-2.1 (m, 2H), 2.4 (br d, J = 13.3 Hz, 3H, OH, H-3), 4.0-4.1 (m, 1H, H-1), 5.55 (t, J = 3.3 Hz, 1H, H-C=C=); ¹³C NMR & 23.5, 23.6, 23.8, 26.7, 27.0, 30.1, 36.1, 39.9, 40.4, 40.5, 69.1, 80.2, 103.1, 103.1, 111.5, 192.6; HRMS calcd for C₁₃H₂₀O₂ 208.14633, found 208.14649.

2-(3-Hydroxy-3-methylbut-1-enylidene)cyclohexanol (2d). Elution with 15:85 EtOAc/hexanes provided iodide 3a (3%). Further elution with 60:40 EtOAc/hexanes afforded the diastereomeric mixture of 2d as a solid. The analytical sample was obtained by HPLC (Column 2, 6 mL/min, 65:35 EtOAc/hexanes, $t_R = 20.74$ min): mp 53-54°C; ¹H NMR δ 1.1-1.3 (m, 9H), 1.23 (s, CH₃, overlapped with mult. at 1.1-1.3, anti-isomer), 1.25 (s, CH₃, overlapped with mult. at 1.1-1.3, anti-isomer), 1.29 (s, CH₃, overlapped with mult. at 1.1-1.3, synisomer), 1.32 (s, CH₃, overlapped with mult. at 1.1-1.3, synisomer), 1.32 (s, CH₃, overlapped with mult. at 1.1-1.3, synisomer), 1.6-1.7 (m, 2H), 1.9-2.0 (m, 2H), 2.28 (br d, J = 13.7 Hz, 1H), 3.73 (s, 2H, OH), 3.9 (m, 1H, CH-OH), 5.42 (t, J = 2.9 Hz, 1H, H-C=C=C, anti-

isomer), 5.47 (t, J = 2.8 Hz, 1H, H-C=C=C, syn-isomer); ¹³C NMR δ 23.7, 23.9, 26.4, 28.9, 29.1, 30.1, 30.2, 35.7, 35.9, 69.0, 69.2, 69.4, 69.7, 103.7, 104.0, 110.5, 110.8, 192.7, 193.1; IR (CHCl₃) v 3300, 1960, 1160 cm⁻¹; HRMS calcd for C₁₁H₁₈O₂ 182.13068, found 182.12888. Anal. Calcd for C₁₁H₁₈O₂: C, 72.47; H, 9.96. Found: C, 72.06; H, 10.04.

2-(3-Ethyl-3-hydroxypent-1-enylidene)cyclohexanol (2e). Elution with 10:90 EtOAc/hexanes provided iodide 3a (7%). Further elution with 25:75 EtOAc/hexanes afforded the diastereomeric mixture of 2f as a yellowish oil. The diastereomers were separated by HPLC (Column 1, 10 mL/min, 65:35 EtOAc/hexanes). (15*,1'S*)-Isomer (anti-2e): $t_R = 31$ min; colorless solid; mp 80-81°C; ¹H NMR & 0.8-1.0 (m, 6H, CH₃), 1.3-2.3 (m, 11H), 2.4 (br d, J = 12.3 Hz, 1H, H-3), 2.6 (br s, 2H, OH), 4.0-4.1 (m, 1H, H-1), 5.31 (t, J = 3.0 Hz, 1H, H-C=C=); ¹³C NMR & 7.9, 8.0, 23.7, 26.6, 30.3, 32.4, 32.8, 35.9, 69.0, 74.3, 102.1, 111.8, 193.1; IR (neat, diastereomeric mixture) v 3600-3100, 1970 cm⁻¹. Anal. Calcd for C₁₃H₂₂O₂: C, 74.24; H, 10.54. Found: C, 74.04; H, 10.58. (1S*,1'R*)-Isomer (syn-2e): $t_R = 36$ min; oil; ¹H NMR & 0.9-1.0 (m, 6H, CH₃), 1.4-2.4 (m, 14H), 4.0-4.1 (m, 1H, H-1), 5.3-5.35 (m, 1H, H-C=C=); ¹³C NMR & 7.9, 8.0, 32.7, 32.8, 32.9, 36.0, 36.1, 69.0, 74.1, 74.3, 102.3, 102.4, 112.1, 192.9, 193.1; HRMS calcd for C₁₃H₂₂O₂ 210.16198, found 210.16203.

2-(3-Hydroxy-3-methyl-5-phenylpent-1-enylidene)cyclohexanol (2f). Eluent: 40/60 EtOAc/hexanes. Oil: ¹H NMR δ 1.3-1.4 (m, 6H), 1.35, 1.38 and 1.42 (s, CH₃, overlapped with mult. at 1.3-1.4), 1.7-1.8 (m, 1H), 1.8-2.0 (m, 3H), 2.0-2.1 (m, 2H), 2.41 (br d, J = 13.8 Hz, 1H), 2.6-2.7 (m, 2H), 3.31 (br s, 1H, OH), 4.1 (m, 1H, H-1), 5.4-5.5 (m, 1H, H-C=C=C), 7.1-7.3 (m, 5H); ¹³C NMR δ 23.6, 23.9, 26.5, 26.5, 26.9, 27.2, 28.4, 30.1, 30.4, 30.5, 30.7, 36.0, 36.0, 44.1, 44.6, 69.2, 69.3, 71.7, 71.8, 71.9, 102.8, 103.3, 103.3, 111.4, 111.9, 125.6, 128.3, 142.4, 142.5, 192.9, 193.3, 193.7; IR (neat) v 3300, 1975 cm⁻¹; HRMS calcd for C₁₈H₂₄O₂ 272.17763, found 272.17749.

2-(3-Hydroxy-3-methylhepta-1,6-dien-1-ylidene)cyclohexanol (2g). Elution with 10:90 EtOAc/hexanes provided iodide 3a (7%). Further elution with 25:75 EtOAc/hexanes afforded the diastereomeric mixture of 2g as an oil: ¹H NMR δ 1.2-1.4 (m, 6H), 1.5-1.8 (m, 4H), 2.0-2.1 (m, 4H), 2.3 (br d, J = 13.9 Hz, 1H, H-3), 3.1 (br s, $W_{1/2} = 23.8$ Hz, 2H, OH), 4.0 (m, 1H, H-1), 4.9-5.0 (m, 2H, H-7'), 5.3-5.4 (m, 1H, H-C=C=), 5.7-5.9 (m, 1H, H-6'); ¹³C NMR δ 23.4 , 23.6, 23.9, 26.5, 26.7, 26.8, 27.1, 27.2, 28.2, 28.6, 29.8, 30.1, 30.2, 30.4, 35.7, 35.8, 35.9, 41.0, 41.2, 41.7, 69.1, 69.1, 69.2, 71.6, 71.6, 71.8, 71.9, 102.8, 103.2, 103.2, 110.9, 111.2, 111.7, 114.2, 138.7, 138.9, 192.8, 193.2; IR (neat) v 3500-3200, 1970 cm⁻¹; HRMS calcd for C₁₄H₂₂O₂ 222.16198, found 222.16173.

5-Hydroxy-7-(2-hydroxycyclohexylidene)-5-methylhept-6-enenitrile (2h). Elution with 10:90 EtOAc/hexanes provided iodide 3a (7%). Further elution with 50:50 EtOAc/hexanes afforded the diastereomeric mixture of 2h as a colorless oil: ¹H NMR δ 1.2-1.5 (m, 6H), 1.5-1.8 (m, 6H), 1.9-2.1 (m, 2H), 2.3-2.7 (m, 5H), 4.1 (m, 1H, CH-OH), 5.4 (m, 1H, H-C=C=); ¹³C NMR δ 17.5, 20.5, 22.6, 23.5, 23.6, 23.8, 26.6, 26.8, 28.1, 28.8, 30.0, 30.2, 30.4, 31.9, 36.0, 36.3, 40.8, 41.3, 69.2, 69.3, 71.4, 71.5, 103.0, 103.2, 103.3, 112.5, 119.8, 192.6, 192.8; IR (neat) v 3600-3100, 2240, 1970, 1640 cm⁻¹; HRMS calcd for C_{14H19}NO (M-H₂O) 217.14666, found 217.14645.

Methyl 12-hexyl-12-hydroxy-14-(2-hydroxycyclohexylidene)tetradec-13-enoate (2i). Elution with 10:90 EtOAc/hexanes provided iodide 3a (5%). Further elution with 25:75 EtOAc/hexanes afforded the diastereomeric mixture of 2i as a colorless oil. The diastereomers were separated by HPLC (Column 1, 10 mL/min, 25:75 EtOAc/hexanes). ($1S^*, 1'R^*$)-Isomer (syn-2i): $t_R = 27$ min; colorless oil; ¹H NMR δ 0.8-0.9 (m, 3H, CH₃), 1.3-2.1 (m, 37H), 2.2-2.3 (m, 2H), 2.4 (br d, J = 13.0 Hz, 1H), 3.66 (s, 3H, CO₂-CH₃), 3.8-4.0 (m,

1H, CH-OH), 5.37 (t, J = 3.3 Hz, 1H, H-C=C=); ¹³C NMR δ 14.1, 22.6, 23.7, 24.9, 26.8, 29.1, 29.2, 29.4, 29.5, 29.6, 29.7, 30.0, 30.1, 31.8, 34.1, 36.3, 40.9, 51.4, 69.0, 73.6, 103.6, 112.5, 174.3, 192.3; HRMS calcd for C₂₇H₄₈O₄ 436.35526, found 436.35592. (1S*,1'S*)-Isomer (anti-2i): t_R = 31 min; colorless oil; ¹H NMR δ 0.86 (t, J = 5.95 Hz, 3H, CH₃), 1.3-2.0 (m, 37H), 2.3 (m, 2H), 2.4 (br d, J = 12.5 Hz, 1H), 3.65 (s, 3H, CO₂-CH₃), 4.0-4.1 (m, 1H, CH-OH), 5.36 (t, J = 2.7 Hz, 1H, H-C=C=); ¹³C NMR δ 14.1, 22.6, 23.6, 23.7, 23.8, 24.9, 26.7, 29.1, 29.2, 29.4, 29.5, 29.6, 29.7, 30.0, 30.1, 31.8, 34.1, 35.9, 41.0, 51.4, 69.0, 73.9, 103.3, 112.2, 174.3, 192.4; IR (neat, mixture of isomers) v 3600-3150, 1970, 1740 cm⁻¹. Anal. Calcd for C₂₇H₄₈O₄: C, 74.26; H, 11.08. Found: C, 74.40; H, 11.16.

 $(1S^*, 1'S^*)-2-[2-(1-Hydroxycyclohexyl)prop-1-en-1-ylidene]cyclohexanol (anti-2j). Eluent: 40:60 EtOAc/hexanes. Colorless solid: mp 97-99°C; ¹H NMR <math>\delta$ 1.2-1.7 (m, 15H), 1.76 (s, 3H, CH₃), 1.7-1.8 (m, 1H), 1.9-2.1 (m, 2H), 2.3-2.4 (m, 2H), 3.90 (dd, J = 8.8, 4.6 Hz, 1H, H-1); ¹³C NMR δ 15.1, 22.4, 22.5, 23.8, 25.7, 26.9, 30.2, 36.1, 36.3, 36.9, 69.3, 72.4, 109.1, 110.8, 191.7; IR (KBr) v 3300, 1960, 950 cm⁻¹. Anal. Calcd for C₁₅H₂₄O₂: C, 76.21; H, 10.24. Found: C, 76.32; H, 10.31.

2-[2-(1-Hydroxycyclohexy)-2-phenylethenylidene)]cyclohexanol (2k). Eluent: 30:70 EtOAc/hexanes. The isomers were separated by HPLC (Column 2, 6 mL/min, 35:65 EtOAc/hexanes). Data for the less polar isomer (68% of mixture): ¹H NMR δ 1.2-1.8 (m, 15H), 2.0-2.2 (m, 4H), 2.49 (br d, J = 12.6 Hz, 1H), 4.11 (br s, $W_{1/2} = 7.0$ Hz, 1H, OH), 7.2-7.3 (m, 3H), 7.51 (d, J = 6.5 Hz, 2H); ¹³C NMR δ 22.5, 22.7, 23.5, 25.6, 26.6, 29.8, 36.0, 37.0, 38.0, 69.5, 73.4, 109.9, 116.5, 126.8, 127.9, 129.2, 136.8, 195.4; IR (CHCl₃) v 3540, 3340, 1960 cm⁻¹; HRMS calcd for C₂₀H₂₆O₂ 298.19328, found 298.19392. Data for the more polar isomer (32% of mixture): ¹H NMR δ 1.3-2.1 (m, 15H), 2.5 (d, J = 11.6 Hz, 1H), 4.0 (m, 1H, CH-OH), 7.2-7.3 (m, 3H), 7.4-7.5 (m, 2H); ¹³C NMR δ 22.5, 22.8, 24.0, 25.7, 26.8, 30.3, 36.1, 36.9, 38.1, 69.4, 73.2, 110.3, 117.0, 126.9, 128.0, 129.3, 136.7, 194.7; IR (CHCl₃) v 3600, 3400, 1960 cm⁻¹; HRMS calcd for C₂₀H₂₆O₂ 298.19328.

6-(tert-Butyldiphenylsilyloxy)-4-(1-hydroxycyclohexyl)-2-hexylhexa-2,3-dien-1-ol (21). First eluted with 5:95 EtOAc/hexanes were enyne 4d and recovered 1d (5%). Further elution with 10:90 EtOAc/hexanes afforded iodide 3d. Finally, diol 2l was eluted with 20:80 EtOAc/hexanes as an oil: ¹H NMR & 0.87 (t, J = 6.8 Hz, 3H, CH₃), 1.05 (s, 9H, (CH₃)₃-C), 1.1-1.7 (m, 19H), 1.9-2.0 (m, 2H), 2.2-2.4 (m, 3H), 3.73 (t, J = 6.3 Hz, 2H, H-6), 3.96 (s, 2H, H-1), 7.4 (m, 6H), 7.6-7.7 (m, 4H); ¹³C NMR & 14.1, 19.0, 22.2, 22.3, 22.6, 25.6, 26.8, 27.9, 29.2, 29.5, 30.4, 31.6, 36.9, 37.3, 62.8, 64.3, 72.0, 109.7, 114.2, 127.7, 129.7, 133.3, 135.5, 194.9; IR (neat) v 3360, 1900, 1110 cm⁻¹; HRMS calcd for C₃₄H₅₀O₃Si 534.352924, found 534.353584. Data for 6-(tert-butyldiphenylsilyloxy)-2-hexyl-1-iodohex-3-yne-2-ol (3d): unstable oil; ¹H NMR & 0.87 (distorted t, 3H, CH₃), 1.06 (s, 9H, (CH₃)₃-C), 1.3 (m, 6H), 1.4-1.5 (m, 2H), 1.7-1.8 (m, 2H), 2.21 (s, 1H, OH), 2.49 (t, J = 6.8 Hz, 2H, H-5), 3.37 (d, J = 10.1 Hz, 1H, H-1), 3.46 (d, J = 10.0 Hz, 1H, H-1), 3.77 (t, J = 6.8 Hz, 2H, H-6), 7.3-7.4 (m, 6H), 7.6-7.7 (m, 4H); ¹³C NMR & 14.0, 19.1, 20.8, 22.5, 22.8, 24.8, 26.8, 29.2, 31.6, 40.6, 62.2, 69.3, 81.2, 83.1, 127.6, 129.7, 133.5, 135.5; IR (neat) v 3450, 2240, 1110 cm⁻¹.

3-Ethyltrideca-4,5-dien-3,7-diol (2m). Elution with 5:95 EtOAc/hexanes led to enynes 4f (73:27E/Z:) and recovered 1f (21%). Further elution with 10:90 EtOAc/hexanes afforded ketone 5^{28} followed by diols 2m as an oil: ¹H NMR δ 0.8-0.9 (m, 9H, CH₃), 1.2-1.4 (m, 8H), 1.4-1.6 (m, 6H), 2.10 (br s, $W_{1/2} = 18.2$ Hz, OH), 2.31 and 2.38 (br s, $W_{1/2} = 44.7$ Hz, OH), 2.90 (br s, $W_{1/2} = 21.4$ Hz, OH), 4.1 (m, 1H, H-7), 5.24 (dd, J = 6.2, 2.1 Hz, H-4), 5.30 (dd, J = 6.3, 2.3 Hz, H-4), 5.37 (t, J = 6.4 Hz, H-6), 5.44 (t, J = 6.1 Hz, H-6); ¹³C NMR δ 8.0, 8.0, 14.0, 22.6, 25.4, 25.5, 29.1, 31.8, 32.8, 32.9, 37.4, 69.8, 70.3, 74.2, 74.3, 99.5, 100.7, 101.4, 200.0, 200.2; IR (neat) v 3300, 1960 cm⁻¹; HRMS calcd for C₁₅H₂₆O (M-H₂O) 223.206162, found

223.206291.

5-(1-Hydroxycyclohexyl)-3-methylpenta-3,4-diene-1,2-diol (2n). Eluent: EtOAc. First eluted was diol 6 followed by the solid triol 2n that was further purified by HPLC (Column 2, 6 mL/min, EtOAc): mp 74-76°C; ¹H NMR δ 1.4-1.6 (m, 8H), 1.70 (d, J = 2.8 Hz, 3H, CH₃), 1.72 (d, J = 2.9 Hz, 3H, CH₃), 3.5-3.7 (m, 2H, H-1), 4.02 (br s, 4H, H-2, OH), 5.3 and 5.4 (t, J = 2.5 Hz, 1H, H-5); ¹³C NMR δ 15.7, 16.0, 22.3, 22.4, 22.5, 22.6, 25.4, 37.9, 37.9, 38.6, 38.7, 64.4, 64.5, 71.1, 72.6, 72.7, 101.7, 102.8, 103.1, 103.7, 198.3, 198.6; IR (CHCl₃) v 3400, 1965 cm⁻¹; HRMS calcd for C₁₂H₁₉O₃ (M-1) 211.13342, found 211.13273. Data for **3-methylpenta-3,4-dien-1,2-diol** (6): ¹H NMR δ 1.74 (t, J = 3.2 Hz, 3H), 1.9-2.0 (br s, 2H, OH), 3.59 (dd, J = 11.3, 6.7 Hz, 1H, H-1), 3.73 (dd, J = 11.3, 3.5 Hz, 1H, H-1), 4.05 (m, 1H, H-2), 4.85 (m, 2H, H-5); ¹³C NMR δ 15.2, 65.0, 72.2, 77.8, 99.3, 204.8; IR (CHCl₃) v 3300, 1960 cm⁻¹; HRMS calcd for C₆H₁₀O₂ 114.06808, found 114.06728.

1-Benzyloxy-5-(1-hydroxycyclohexyl)-3-methylpenta-3,4-dien-2-ol (20). Eluent: 35:65 EtOAc/hexanes. Oil: ¹H NMR δ 1.2-1.6 (m, 10H), 1.73 (t, J = 2.9 Hz, 3H, CH₃), 2.94 (br s, 1H, OH), 3.32 (br s, 1H, OH), 3.4-3.6 (m, 2H), 4.18 (br s, $W_{1/2} = 19.0$ Hz, 1H), 4.54 (s, 2H, O-CH₂-Ph), 5.3 and 5.4 (m, 1H, H-C=C=C), 7.2-7.3 (m, 5H); ¹³C NMR δ 15.6, 22.3, 22.4, 25.4, 37.9, 38.5, 38.6, 70.6, 71.0, 71.2, 72.3, 72.6, 73.3, 102.3, 102.7, 103.2, 103.5, 127.6, 127.7, 128.3, 137.6, 137.7, 198.2, 198.4; IR (neat) v 3380, 1970 cm⁻¹. Anal. calcd for C₁₉H₂₆O₃: C, 74.45; H, 8.67; found: C, 74.38; H, 8.54.

1-(tert-Butyldiphenylsilyloxy)-5-(1-hydroxycyclohexyl)-3-methylpenta-3,4-dien-2-ol (**2p**). Eluent: 25:75 EtOAc/hexanes. Oil: ¹H NMR δ 1.08 (s, 9H, (CH₃)₃-C), 1.2-1.6 (m, 10H), 1.67 (d, J = 2.9 Hz, 3H, C₃-CH₃, syn-isomer), 1.71 (d, J = 2.9 Hz, 3H, C₃-CH₃, anti-isomer), 2.50 (d, J = 4.1 Hz, 1H, OH), 2.70 (d, J = 5.2 Hz, 1H, OH), 3.64 (dd, J = 10.3, 7.2 Hz, 1H, H-1), 3.75 (dd, J = 10.3, 4.1 Hz, 1H, H-1), 4.1 (m, 1H, H-2), 5.3 (m, 1H, H-5), 7.3-7.5 (m, 6H), 7.7-7.8 (m, 4H); ¹³C NMR δ 15.4, 15.7, 19.2, 22.4, 25.4, 26.8, 38.1, 38.4, 66.4, 66.7, 70.1, 72.5, 102.2, 102.8, 103.2, 127.7, 129.8, 133.0, 133.5, 198.4; IR (CHCl₃) v 3380, 1930, 1430, 1115 cm⁻¹; HRMS calcd for C₂₈H₃₆O₂Si (M-H₂O) 432.248373, found 432.248459.

(2R*,4S*)-[1-(tert-Butyldiphenylsilyloxy)-5-(1-hydroxycyclohexyl)-3-methylhexa-3,4-dien-2-ol (2q). Elution with 5:95 EtOAc/hexanes afforded enyne 4j and allene 9. Further elution with 20:80 EtOAc/hexanes afforded dimer 7 followed by the mixture of diols 2q and 8, that were separated by HPLC (Column 1, 12 mL/min, 25:75 EtOAc/hexanes). (2 R^{+} ,4 S^{+})-Isomer (anti-2q): t_R = 22 min; oil; ¹H NMR δ 1.07 (s, 9H, $(CH_3)_3$ -C), 1.1-1.2 (m, 1H), 1.4-1.6 (m, 10H), 1.66 (s, 3H), 1.67 (s, 3H), 2.60 (d, J = 4.1 Hz, 1H, OH), 3.60 (dd, J = 10.3, 7.0 Hz, 1H, H-1), 3.71 (dd, J = 10.3, 3.6 Hz, 1H, H-1), 4.1 (m, 1H, H-2), 7.3-7.4 (m, 6H), 7.6-7.7 (m, 4H); ¹³C NMR δ 14.2, 15.8, 19.2, 22.2, 22.3, 25.6, 26.8, 36.2, 36.6, 66.8, 72.1, 72.9, 101.1, 109.5, 127.7, 129.8, 133.1, 135.6, 197.2; IR (CHCl₃) v 3400, 1970, 1120 cm⁻¹. Anal. calcd for C₂₉H₄₀O₃Si: C, 74.96; H, 8.68. Found: C, 74.62; H, 8.56. (2R*,4R*)-Isomer (syn-2q): t_R = 27 min; oil; mp 75-77°C; ¹H NMR § 1.07 (s, 9H, (CH₃)₃-C), 1.2-1.7 (m, 11H), 1.66 (s, 3H), 1.68 (s, 3H), 2.48 (br s, $W_{1/2}$ =10.9 Hz, 1H, OH), 3.60 (dd, J = 10.3, 7.4 Hz, 1H, H-1), 3.71 (dd, J = 10.3, 3.9 Hz, 1H, H-1), 4.1 (m, 1H, H-2), 7.3-7.5 (m, 6H), 7.6-7.7 (m, 4H); ¹³C NMR & 14.3, 15.7, 19.2, 22.2, 25.6, 26.8, 36.3, 36.6, 67.1, 72.1, 72.9, 101.1, 109.2, 127.7, 129.8, 133.1, 135.6, 196.9; IR (CHCl₃) v 3300, 1970, 1110 cm⁻¹. Anal. calcd for C₂₉H₄₀O₃Si: C, 74.96; H, 8.68. Found: C, 74.73; H, 8.73. Data for 1-(tert-Butyldiphenylsilyloxy)-3-methylhexa-3,4-dien-2-ol (9): Oil; ¹H NMR δ 1.07 (s, 9H, (CH₃)₃-C), 1.6-1.7 (m, 6H, H-6 and C₃-CH₃), 3.64 (dd, J = 10.2, 6.5 Hz, 1H, H-1), 3.73 (dd, J = 10.2, 3.8 Hz, 1H, H-1), 4.1 (m, 1H, H-2), 5.1-5.2 (m, 1H, H-5), 7.3-7.5 (m, 6H), 7.7 (m, 4H); ¹³C NMR & 14.5, 15.5, 15.7, 19.3, 26.8, 66.5, 72.6, 87.9, 99.1, 127.7, 129.6, 129.8, 133.2, 134.8, 135.6, 201.4; IR (neat) v 3400,

1975, 1115 cm⁻¹; HRMS calcd for $C_{23}H_{30}O_2Si$ 366.201509, found 366.201777. Data for 1,10-Di(tertbutyldiphenylsilyloxy)-3,5,6,8-tetramethyldeca-3,4,6,7-tetraene-2,9-diol (7): Oil; ¹H NMR δ 1.06 (s, 9H, (CH₃)₃-C), 1.6 (m, 3H, CH₃), 1.7 (m, 3H, CH₃), 2.4 (m, 1H, OH), 3.62 (dd, J = 10.1, 7.1 Hz, 1H), 3.72 (dd, J = 10.1, 3.9 Hz, 1H), 4.1-4.2 (m, 1H), 7.3-7.4 (m, 6H), 7.7 (m, 4H); ¹³C NMR δ 15.5, 15.7, 17.8, 19.2, 26.8, 66.8, 67.2, 73.0, 101.8, 102.3, 127.7, 129.8, 133.2, 135.6, 200.7, 200.8; IR (CHCl₃) v 3550, 3450, 1600, 1110 cm⁻¹; HRMS calcd for C₄₆H₅₈O₄Si₂ 730.38737, found 730.38742. Data for (2R*,4S*)-2-(tert-Butyldiphenylsilyloxy)-5-(1-hydroxycyclohexyl)-3-methylhexa-3,4dien-1-ol (8): t_R = 19 min; oil; ¹H NMR δ 1.08 (s, 9H, (CH₃)₃-C), 1.2-1.6 (m, 14H), 1.56 (s overlapped with m at 1.2-1.6, CH₃), 1.72 (s, 3H, CH₃), 3.50 (apparent d, 2H, H-1), 4.27 (apparent t, 1H, H-2), 7.3-7.4 (m, 6H), 7.6-7.7 (m, 4H); ¹³C NMR δ 13.4, 14.3, 19.4, 21.9, 22.0, 25.4, 27.0, 35.3, 36.3, 64.5, 72.0, 75.3, 100.6, 107.5, 127.5, 127.7, 129.7, 129.8, 133.5, 133.9, 135.7, 135.8, 197.5; IR (CHCl₃) v 3300, 1970, 1110 cm⁻¹; HRMS calcd for C₂₉H₄₀O₃Si (M-H₂O) 446.26411, found 446.26282.

4-(1,2-Epoxycyclohexyl)-2-methylbut-3-yn-2-ol (13). n-BuLi (1.42 M, 2 mL, 2.85 mmol) was added dropwise to a solution of 1-ethynylcyclohexene (0.300 g, 2.83 mmol) in THF (3.3 mL) at -78°C and the resulting solution stirred at this temperature for 15 min, allowed to warm to room temperature over 1 h and recooled to -78°C. A solution of acetone (0.164 g, 2.83 mmol) in THF (1.5 mL) was slowly added, the reaction mixture allowed to reach 25°C and kept at this temperature for 3 h. Brine (5 mL) was added and the aqueous layer was extracted with diethyl ether (3 x 15 mL). The residue after evaporation was purified by flash chromatography (12:88 EtOAc/hexanes) to afford **4-(1-cyclohexenyl)-2-methylbut-3-yn-2-ol** (0.350 g, 76%): ¹H NMR δ 1.48 (s, 6H, CH₃), 1.5-1.6 (m, 4H), 2.0-2.1 (m, 4H), 2.5 (br s, 1H, OH), 6.0 (m, 1H, H-C=C); ¹³C NMR δ 21.3, 22.1, 25.4, 29.0, 31.1, 31.4, 65.3, 83.6, 91.1, 120.0, 134.6. This material 0.260 g (1.83 mmol) was treated with MCPBA according to the General Epoxidation Procedure. The title epoxide **13** was obtained after purification by flash chromatography (10:90 EtOAc/hexanes): ¹H NMR δ 1.1-1.4 (m, 4H), 1.45 (s, 6H), 1.8-2.0 (m, 1H), 2.0-2.1 (m), 2.61 (br s, OH), 3.27 (t, J = 2.2 Hz, 1H); ¹³C NMR δ 18.7, 19.2, 23.9, 29.6, 31.2, 50.1, 59.9, 64.8, 82.1, 86.7; IR (neat) v 3400, 1170 cm⁻¹.

Preparation of anti-2d from 13. LAH (0.088 g, 2.33 mmol) was added to alcohol 13 (0.070 g, 0.388 mmol) in THF (18 mL) and the mixture was refluxed for 6 h. Water (5 mL) was added and the whole extracted with EtOAc (5 x 15 mL). The residue after evaporation was purified by flash chromatography (30:70 EtOAc/hexanes) to yield allene *anti-2d* (0.050 g) contaminated with an unknown impurity. Further purification under the same conditions yielded a pure sample of *anti-2d*.

l-[(1,2-Epoxycyclohexyl)ethynyl]cyclohexanol (14). The procedure previously described for 13 was applied to cyclohexanone to afford, after flash chromatography (10:90 EtOAc/hexanes), the epoxide 14 (64% over two steps) as a yellowish oil: ¹H NMR δ 1.1-1.6 (m, 12H), 1.7-2.2 (m, 6H), 2.30 (br s, $W_{1/2} = 50.0$ Hz, 1H, OH), 3.28 (s, 1H, CH-O); ¹³C NMR δ 18.8, 19.3, 23.1, 24.0, 25.0, 29.8, 39.7, 50.2, 60.0, 68.4, 84.5, 85.7; IR (CHCl₃) v 3400, 1070 cm⁻¹; HRMS calcd for C₁₄H₂₀O₂ 220.14633, found 220.14607.

Preparation of syn-2j from 14. MeLi (1.6 M, 4.3 mL, 6.8 mmol) was added dropwise to a suspension of CuI (0.65 g, 3.4 mmol) in THF (35 mL) at -25° C. After stirring at that temperature for 45 min a solution of 14 in THF (1 mL) was added, the mixture was stirred at -25° C further 30 min and allowed to reach room temperature overnight. A 1:1 mixture of sat NH₄Cl and 3% NH₄OH (10 mL) was added and the mixture was stirred 1 h. The aqueous layer was extracted with diethyl ether (3 x 15 mL) and the combined organic layers were washed with water (2 x 10 mL). Flash chromatography (35:65 EtOAc/hexanes) of the residue after evaporation afforded a 7:5:1 mixture of alkyne 15, syn-2j and diol 2a (0.160 g, 100%) that were separated by HPLC (Column 1, 12 mL/min, 70:30 EtOAc/hexanes). Data for 2-(1-Hydroxycyclohexyl)ethynyl-2-

methylcyclohexanol (15): t_R = 27 min; mp 104°C; ¹H NMR δ 1.2-1.3 (m, 6H), 1.29 (s, overlapped with m at 1.2-1.3, CH₃), 1.4-1.6 (m, 9H), 1.6-1.9 (m, 8H), 3.13 (dd, J = 10.8, 3.9 Hz, 1H, H-1); ¹³C NMR δ 22.7, 23.7, 23.7, 24.9, 25.2, 26.9, 32.6, 38.5, 40.0, 40.3, 40.4, 68.8, 76.5, 86.8, 88.5; IR (CHCl₃) v 3340 cm⁻¹; HRMS calcd for C₁₅H₂₄O₂: 236.177630, found 236.177115. Anal. calcd for C₁₅H₂₄O₂: C, 76.27; H, 10.17. Found: C, 75.85; H, 10.33. Data for (1S*,1'R*)-2-[2-(1-hydroxycyclohexyl)prop-1-en-1-ylidene]cyclohexanol (*syn-2j*): t_R = 34 min; mp 110-111°C; ¹H NMR δ 1.3-1.8 (m, 18H), 1.77 (s, overlapped with m at 1.3-1.8, CH₃), 1.9-2.1 (m, 2H), 2.34 (apparent d, J = 14.2 Hz, 1H), 3.9 (m, 1H, H-1); ¹³C NMR δ 14.7, 22.4, 22.6, 24.0, 25.7, 26.9, 30.5, 36.2, 36.3, 37.0, 69.2, 72.2, 109.5, 111.6, 191.0; HRMS calcd for C₁₅H₂₄O₂ 236.177630, found 236.178569.

 $(3'S^*,4'S^*)$ -1-[5-(tert-Butyldiphenylsilyloxy)-3,4-epoxy-3-methylpent-1-ynyl]cyclohexanol (16). The procedure used in the preparation of epoxide 13 was followed from (E)-5-(tert-butyldiphenylsilyloxy)-3-methylpent-3-en-1-yne and cyclohexanone to yield, after flash chromatography (15:85 EtOAc/hexanes), the epoxide 16 (26 %, two steps) as as oil: ¹H NMR δ 1.07 (s, 9H, (CH₃)₃-C), 1.34 (s, 3H, CH₃), 1.5-1.7 (m, 8H), 1.9 (m, 2H), 1.98 (s, 1H, OH), 3.37 (t, J = 5.2 Hz, 1H, CH-O), 3.76 (d, J = 5.2 Hz, 2H, CH₂-O), 7.4-7.5 (m, 6H), 7.7 (m, 4H); ¹³C NMR δ 18.6, 19.2, 23.2, 25.1, 26.7, 39.7, 50.7, 62.0, 64.2, 68.5, 84.3, 85.6, 127.7, 129.8, 132.9, 133.2, 135.5; IR (neat) v 3400, 2240, 1110 cm⁻¹. Anal. calcd for C₂₈H₃₈O₃Si: C, 74.96; H, 8.09. Found: C, 75.25; H, 8.16.

Preparation of syn-2q from 16. The reaction of alcohol 16 with Gilman's cuprate under conditions similar to those described above for alcohol 14 afforded, after flash chromatography (12:88 EtOAc/hexanes) and HPLC (Column 1, 10 mL/min, 25:75 EtOAc/hexanes), in order of elution, the alkyne 17 (33%), syn-2q (27%) and 2p (14%, mixture of diastereomers). Data for 1-(tert-butyldiphenylsilyloxy)-5-(1hydroxycyclohexyl)-3,3-dimethylpent-4-yn-2-ol (17): $t_R = 17$ min; ¹H NMR δ 1.06 (s, 9H, (CH₃)₃-C), 1.17 (s, 3H, CH₃), 1.18 (s, 3H, CH₃), 1.3-1.8 (m, 11H), 2.75 (br s, 1H, OH), 3.54 (dd, J = 8.3, 3.1 Hz, 1H, H-1), 3.72 (apparent t, 1H, H-1), 3.91 (dd, J = 10.0, 3.2 Hz, 1H, H-2), 7.4 (m, 6H), 7.6-7.7 (m, 4H); ¹³C NMR δ 19.2, 23.4, 24.4, 25.1, 26.7, 26.8, 34.1, 40.1, 40.1, 65.3, 65.8, 68.5, 77.2, 85.4, 89.0, 127.8, 129.8, 133.0, 135.5; IR (CHCl₃) v 3400, 1110 cm⁻¹. Anal. calcd for C₂₉H₄₀O₃Si: C, 74.95; H, 8.68. Found: C, 74.71; H, 8.56.

Acknowledgments. Financial support by the Dirección General de Investigación Científica y Técnica (DGICYT PB89-0412 and PB92-0449) and by the Universidad del País Vasco (UPV 170.310-0133/89, 170.310-EC021/92 and 170.310-EC201/94) is gratefully acknowledged. We also thank the Ministerio de Educación y Ciencia (Spain) for a Fellowship (to M.S.).

REFERENCES AND NOTES

- Reviews: (a) Molander, G. A. Chem. Rev. 1992, 92, 29-68. (b) Molander, G. A. In Organic Reactions, L. A. Paquette, Ed.; John Wiley & Sons: New York, 1994; Vol. 46; pp 211-367. (c) Molander, G. A.; Harris, C. R. Chem. Rev. 1996, 96, 307-338.
- (a) Girard, P.; Namy, J. L.; Kagan, H. B. J. Am. Chem. Soc. 1980, 102, 2693-2698. (b) Souppe, J.; Namy, J. L.; Kagan, H. B. Tetrahedron Lett. 1982, 23, 3497-3500. (c) Tabuchi, T.; Inanaga, J.; Yamaguchi, M. Tetrahedron Lett. 1986, 27, 1195-1196. (d) Ujikawa, O.; Inanaga, J.; Yamaguchi, M. Tetrahedron Lett. 1989, 23, 2837-2840. (e) Fukuzawa, S.; Fujinami, T.; Sakai, S. Chem. Lett. 1990, 927-930. (f) Molander, G. A.; Mckie, J. A. J. Org. Chem. 1994, 59, 3186-3192. (g) Molander, G. A.; McWilliams, J. C.; Noll, B. C. J. Am. Chem. Soc. 1997, 119, 1265-1276. (h) Kito, M.; Sakai, T.; Shirahama, H.; Miyashita, M.; Matsuda, F. Synlett 1997, 1997, 219-220.

- (a) Aurrecoechea, J. M.; Iztueta, E. Tetrahedron Lett. 1995, 36, 7129-7132. (b) Molander, G. A.; Shakya, S. R. J. Org. Chem. 1996, 61, 5885-5894.
- (a) Tabuchi, T.; Inanaga, J.; Yamaguchi, M. Chem. Lett. 1987, 2275-2278. (b) Inanaga, J.; Tabuchi, T.; Yamaguchi, M. Chem. Express 1991, 6, 419-422. (c) Sugimoto, Y.; Hanamoto, T.; Inanaga, J. Kidorui 1991, 18, 144-145. (d) Aurrecocchea, J. M.; Fañanás-San Antón, R. J. Org. Chem. 1994, 59, 702-704. (e) Sugimoto, Y.; Hanamoto, T.; Inanaga, J. Appl. Organometal. Chem. 1995, 9, 369-375. (f) Mikami, K.; Yoshida, A.; Matsumoto, S.; Feng, F.; Matsumoto, Y.; Sugino, A.; Hanamoto, T.; Inanaga, J. Tetrahedron Lett. 1995, 36, 907-908.
- Recent applications: (a) Alexakis, A.; Marek, I.; Mangeney, P.; Normant, J. F. Tetrahedron 1991, 47, 1677-1696. (b) Lee, N. H.; Jacobsen, E. N. Tetrahedron Lett. 1991, 32, 6533-6536. (c) Marshall, J. A.; DuBay, W. J. J. Am. Chem. Soc. 1992, 114, 1450-1456. (d) Nussbaumer, P.; Stutz, A. Tetrahedron Lett. 1992, 33, 7507-7508. (e) Marshall, J. A.; Pinney, K. G. J. Org. Chem. 1993, 58, 7180-7184. (f) Marshall, J. A.; Yu, B. C. J. Org. Chem. 1994, 59, 324-331. (g) Katritzky, A. R.; Li, J. Q. J. Org. Chem. 1995, 60, 638-643. (h) Katritzky, A. R.; Li, J. Org. Chem. 1996, 61, 1624-1628.
- 6. Preliminary communication: Aurrecoechea, J. M.; Solay, M. Tetrahedron Lett. 1995, 36, 2501-2504.
- (a) Hasegawa, E.; Curran, D. P. Tetrahedron Lett. 1993, 34, 1717-1720. (b) Shabangi, M.; Flowers, R. A. Tetrahedron Lett. 1997, 38, 1137-1140. (c) Hou, Z. M.; Zhang, Y. G.; Wakatsuki, Y. Bull. Chem. Soc. Jpn. 1997, 70, 149-153.
- (a) Tabuchi, T.; Inanaga, J.; Yamaguchi, M. Tetrahedron Lett. 1986, 27, 3891-3894. (b) Matsukawa, M.; Tabuchi, T.; Inanaga, J.; Yamaguchi, M. Chem. Lett. 1987, 2101-2102. (c) Bonini, C.; Federici, C.; Rossi, L.; Righi, G. J. Org. Chem. 1995, 60, 4803-4812.
- 9. Formation of the envnes probably takes place by ring-opening of epoxides with iodide anion associated to Sm(III), followed by reduction of the corresponding iodohydrins (*i.e.* 3d): See ref's 2a, 8a,b.
- The use of Sm(III) species in aldol reactions has been reported: (a) Sasai, H.; Suzuki, T.; Arai, S.; Arai, T.; Shibasaki, M. J. Am. Chem. Soc. 1992, 114, 4418-4420. (b) Miyoshi, N.; Takeuchi, S.; Ohgo, Y. Chem. Lett. 1993, 2129-2132. (c) Van de Weghe, P.; Collin, J. Tetrahedron Lett. 1993, 34, 3881-3884. (d) Sasai, H.; Arai, S.; Shibasaki, M. J. Org. Chem. 1994, 59, 2661-2664.
- 11. Souppe, J.; Danon, L.; Namy, J. L.; Kagan, H. B. J. Organomet. Chem. 1983, 250, 227-236.
- The migration of silyl groups between adjacent hydroxyl groups has been shown to proceed under basic conditions: (a) Mulzer, J.; Schöllhorn, B. Angew. Chem. Int. Ed. Engl. 1990, 29, 431-432. (b) Mulzer, J.; Greifenberg, S. Heterocycles 1995, 40, 93-96.
- A related SET from Ti(III) species to epoxides yielding α-alkoxy radicals has been reported: Nugent, W. A.; RajanBabu, T. V. J. Am. Chem. Soc. 1988, 110, 8561-8562.
- 14. See for example: Molander, G. A.; Harris, C. R. J. Org. Chem. 1997, 62, 7418-7429, and references cited therein.
- The coexistence of radical and organometallic pathways in SmI₂-promoted couplings has been demonstrated: Curran, D. P.; Fevig, T. L.; Jasperse, C. P.; Totleben, M. J. Synlett 1992, 943-961.
- 16. Hlubucek, J. R.; Hora, J.; Russell, S. W.; Toube, T. P.; Weedon, B. C. L. J. Chem. Soc., Perkin Trans. 1 1974, 848-852.
- 17. The determination of the stereochemistry of hydroxyalkyl-2,5-dihydrofurans obtained by stereospecific Ag⁺-promoted^{5e,18,19} cycloisomerization of diols **2a**, **d**, **j** further confirmed these stereochemical assignments. See Ref. 6 and following paper in this issue.
- 18. Marshall, J. A.; Wang, X. J. J. Org. Chem. 1991, 56, 4913-4918.
- 19 Marshall, J. A.; Bartley, G. S. J. Org. Chem. 1994, 59, 7169-7171.
- 20. Still, W. C.; Kahn, M.; Mitra, A. J. Org. Chem. 1978, 43, 2923-2925.
- 21. The literature procedure^{5a} for the preparation of 1b was modified as follows: THF was used as solvent

and n-BuLi at -78°C as base.

- 22. Cacchi, S. Synthesis 1986, 320-322.
- 23. Takai, K.; Sakogawa, K.; Kataoka, Y.; Oshima, K.; Utimoto, K. Org. Synth. 1993, 72, 180-188. Purification by distillation, as reported, resulted in isomerization to oct-2-en-2-yl triflate. Alternatively, flash chromatography in hexanes afforded a 93:7 mixture of oct-1-en-2-yl triflate and its isomeric internal triflate.
- 24. Seyferth, D.; Heeren, J. K.; Singh, G. J. Organometal. Chem. 1966, 5, 267-274.
- 25. Schmidt, U.; Respondek, M.; Lieberknecht, A.; Werner, J.; Fischer, P. Synthesis 1989, 256-261.
- 26. Excess samarium metal, customarily employed in the preparation of SmI₂, should be kept at about 2-3 mol% as the use of larger excesses in reactions of 1a was observed to lead occasionally to the formation of diene byproducts *i* and to loss of diastereoselectivity in the formation of diols 2.



- 27. Molander, G. A.; Etter, J. B.; Zinke, P. W. J. Am. Chem. Soc. 1987, 109, 453-463.
- 28. Colonge, J. Bull. Soc. Chim. Fr. 1934, 1101-1104.