



Synthesis of highly substituted dihydropyrrolophenanthridine derivatives by tandem reaction

Yimin Hu ^{*}, Tao Zhu, Xiaolong Mu, Quansheng Zhao, Tao Yu, Lei Wen, Yulong Zhang, Min Wu, Hao Zhang

Laboratory of Functional Molecular Solids, Ministry of Education, Anhui Key Laboratory of Molecular-Based Materials, School of Chemistry and Materials Science, Anhui Normal University, Wuhu, Anhui 241000, PR China

ARTICLE INFO

Article history:

Received 22 May 2011

Received in revised form 27 September 2011

Accepted 11 October 2011

Available online 19 October 2011

Keywords:

C–H activation

Palladium

Domino reaction

Pyrrolophenanthridine

Cyclization

ABSTRACT

The efficient cross-coupling of bromo(iso)quinoline to 1,6-diyne with the Pd-catalyzed system was established. Using unactivated simple diarynes with bromo(iso)quinoline in the presence of palladium catalytic system afforded different kinds of rare 7,11-diphenyl-9,10-dihydro-8*H*-pyrrolo[3,4-*j*]phenanthridine derivatives through regioselective C–H functionalization in one step. Different diarynes (**a–p**) and different bromo(iso)quinolines were shown to be very active in the reaction. Thus, the generality of this process makes the reaction highly valuable in view of the synthetic and medicinal importance of these pyrrolo[3,4-*j*]phenanthridine derivatives.

© 2011 Elsevier Ltd. All rights reserved.

1. Introduction

The pyrrolophenanthridine class of substances displays a variety of pharmacological properties. This class has been the subject of considerable interest as potential antitumor, antimicrobial, and antiviral agents.¹ A large number of naturally occurring alkaloids that contain a phenanthridine ring system and have a known scope of activity are mentioned in the literature.^{2–4} Lycorine is a phyrrolophenanthridine alkaloid, which displays an antiviral effect against poliovirus and measles and a high antiretroviral activity as well.⁵ Therefore, numerous synthetic strategies for the preparation of these scaffolds have been developed.^{6–8} Recently, Yu reported the intermolecular C–H functionalization of pyridine rings at the 3- and 4-positions using a Pd⁰/PR₃/ArBr catalytic system, providing a powerful method for the preparation of structure-related nicotinic and isonicotinic acids that are of great importance in drug discovery.^{9,10} Lautens et al. reported a new and general strategy for the synthesis of hexahydrobenzo[c]phenanthridine alkaloids with the novel and highly enantioselective palladium(II)-catalyzed ring-opening reaction of a mesoazabicyclic, with an aryl boronic acid as the key step.¹¹ Cronin developed facile routes for the synthesis and isolation of biologically active DNA intercalating framework tetrahydroimidazo[1,2-*f*]phenanthridines and pyrrolo[1,2-*f*]phenanthridine, which are

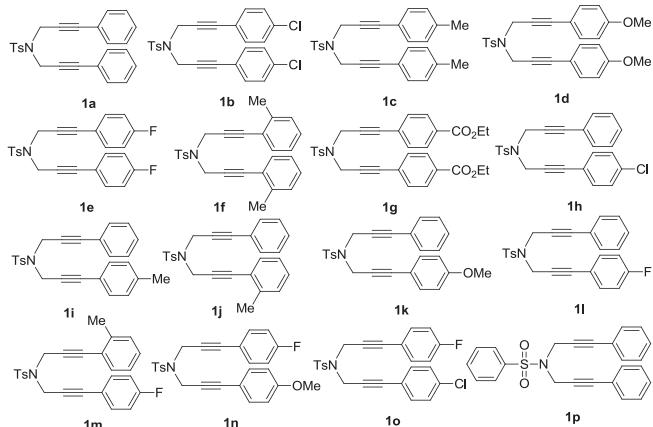
reactive intermediates in the three-step cascade synthesis of 2,3-dihydro-1*H*-imidazo[1,2-*f*]phenanthridinium cations.¹² Our group has also employed a domino strategy¹³ to devise alternative processes for the efficient construction of fused phenanthridine derivatives.¹⁴

2. Results and discussion

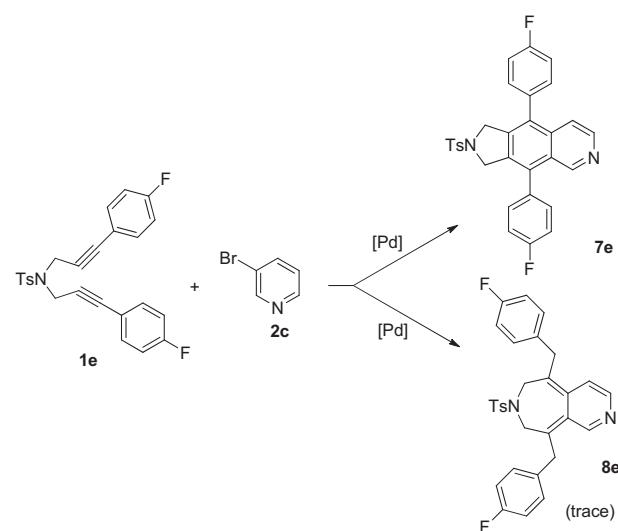
In this study, we introduce a diversity-oriented synthesis of highly substituted dihydropyrrolophenanthridine four-rings condensed ring system from 1,6-diyne and bromoquinoline. The cascade consists of inter-intramolecular Heck reactions and subsequent regioselectively directed arylation by C–H activation of the quinoline ring at the 2- or 4-positions. Herein, we report on the palladium-catalyzed novel domino reactions of **1a–p** with 3-bromo(iso)quinoline or 3-bromopyridine to provide a direct, efficient, and economic methodology for the construction of phenanthridine and quinoline through both C–C bond coupling and C–H bond activation.

In our previous research on palladium-catalyzed cyclization, we designed a substance, *N,N*-bis(3-phenylprop-2-yn-1-yl)-4-methylbenzenesulfonamide. A survey of the reaction conditions using 3-bromoquinoline (**2a**) and *N,N*-bis(3-(4-chlorophenyl)prop-2-yn-1-yl)-4-methylbenzenesulfonamide (**1b**) was performed as a test experiment (Scheme 1, Table 1). In a typical experiment, the reaction of **1b** and **2a** in *N,N*-dimethylformamide (DMF) in the presence of catalyst Pd(OAc)₂ produced 6-chloro-4-(4-chlorophenyl)-2-tosyl-2,3-

* Corresponding author. Tel.: +86 553 386 9310; fax: +86 553 388 3517; e-mail address: yiminhu@mail.ahnu.edu.cn (Y. Hu).



dihydro-1*H*-benzo[f]isoindole **9b** in 42% yield after 24 h at 140 °C. By controlling the experimental conditions, we discovered the following: (1) The efficiency of domino reaction to produce the product **3b** can be lowered the reaction temperature to 130 °C (Table 1, entry 6); (2) The additive bases play an important role in the overall efficiency of this domino reaction by simply varying the bases from tributylamine to silver carbonate (Table 1, entries 6, 11–12, 16), with the **3b** in 84%, 15%, and 5% yield when using tributylamine, potassium carbonate, and silver, respectively, under identical conditions; (3) Among the catalysts investigated (Table 1, entries 12–16), the palladium(II) acetate/(triphenylphosphine) catalytic system is most effective (Table 1, entries 6, 13–16). (4) DMF is a better solvent for this reaction than either toluene or MeCN (Table 1, entries 1–2, 7). As shown in Table 1, the tetracyclic compound **3b** is only isolated in yields of 11% and 39%, respectively, when toluene was employed as

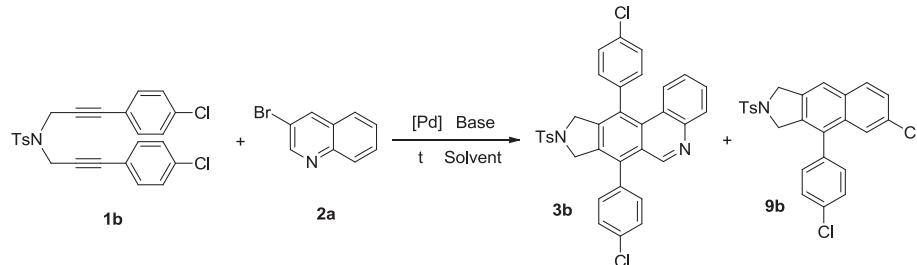


Scheme 1. Synthesis of substituted heterocycle **8e**.

a solvent, indicating temperature/solvent effects on the outputs of the reactions and on the formation of the final products. Thus, the following standard reaction conditions were used for carrying out the following studies: 1,6-diyne (1 equiv) was reacted with 3-bromoquinoline (1.1 equiv) in the presence of palladium acetate (2 mol %) and Ph₃P (4 mol %), with (*n*-Bu)₃N (1.2 equiv) as an additive in DMF at 130 °C.

To probe the scope of the tandem reaction, a range of substituted 3-bromoquinoline and diynes were examined (Table 2). Various diynes and 3-bromoquinoline or 3-bromoisoquinoline or

Table 1
Optimization of cyclization conditions^a



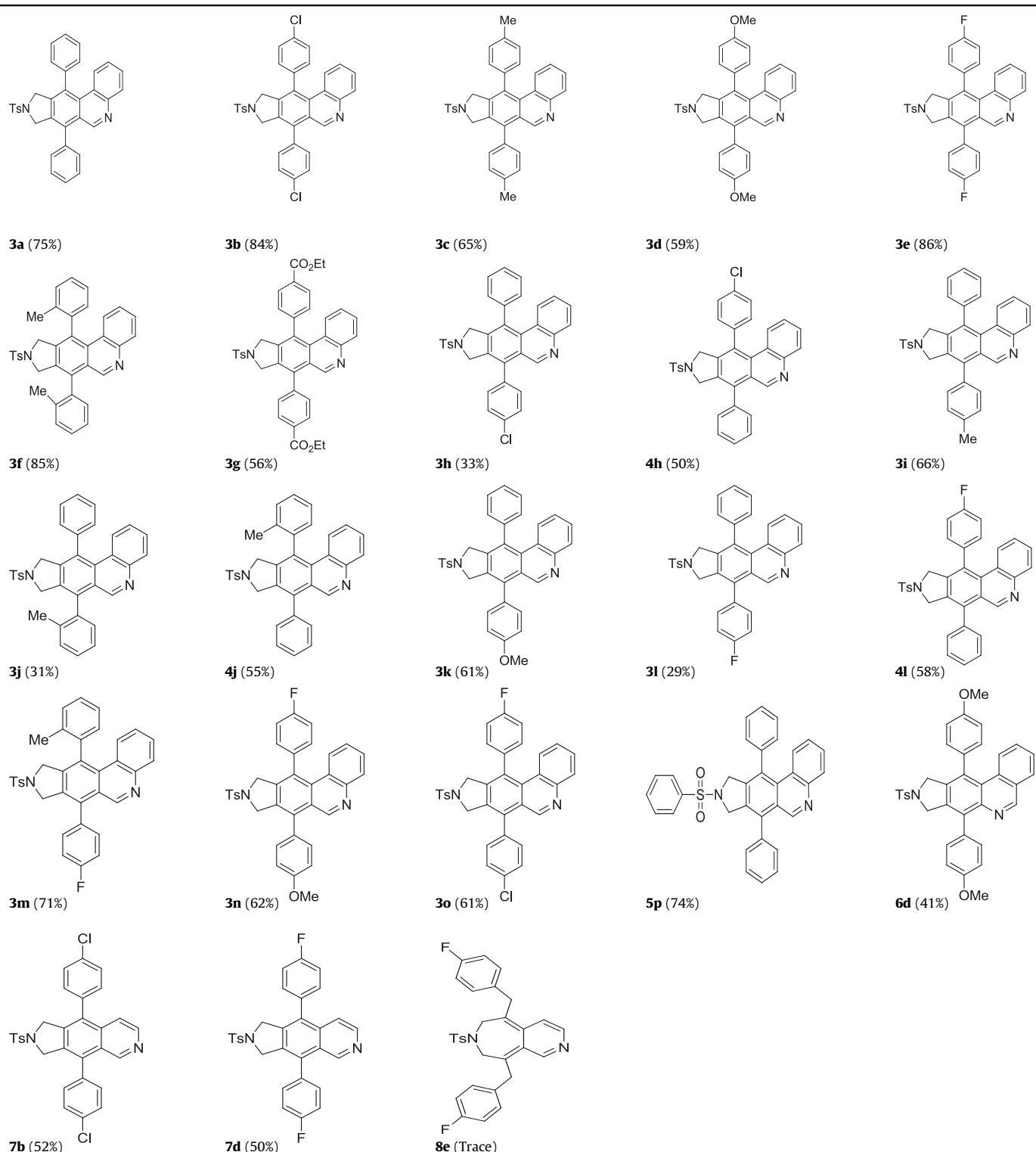
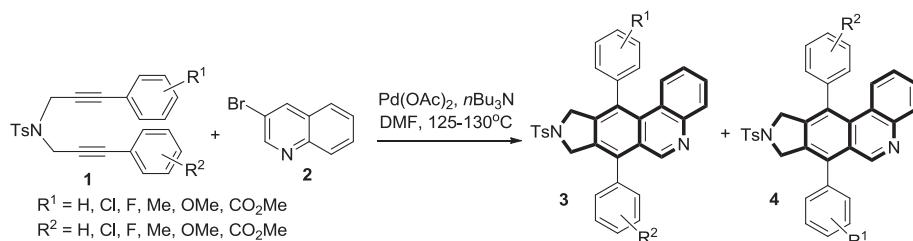
Entry ^a	[Pd]/PPh ₃ (mol %)	Base (equiv)	Solvent	<i>t</i> (h)	<i>T</i> ^b (°C)	Yield ^c (%)	
						3b	9b
1	Pd(OAc) ₂ /PPh ₃ (2:4)	(<i>n</i> -Bu) ₃ N	MeCN	10	100	—	—
2	Pd(OAc) ₂ /PPh ₃ (2:4)	(<i>n</i> -Bu) ₃ N	Toluene	10	120	11	Trace
3	Pd(OAc) ₂ /PPh ₃ (2:4)	(<i>n</i> -Bu) ₃ N	DMF	11	110	Trace	—
4	Pd(OAc) ₂ /PPh ₃ (2:4)	(<i>n</i> -Bu) ₃ N	DMF	11	120	25	7
5	Pd(OAc) ₂ /PPh ₃ (2:4)	(<i>n</i> -Bu) ₃ N	DMF	16	130	42	8
6	Pd(OAc) ₂ /PPh ₃ (2:4)	(<i>n</i> -Bu) ₃ N	DMF	24	130	84	10
7	Pd(OAc) ₂ /PPh ₃ (2:4)	(<i>n</i> -Bu) ₃ N	Toluene	24	130	39	11
8	Pd(OAc) ₂ /PPh ₃ (1:2)	(<i>n</i> -Bu) ₃ N	DMF	24	130	54	8
9	Pd(OAc) ₂ /PPh ₃ (2:4)	(<i>n</i> -Bu) ₃ N	DMF	15	130	61	9
10	Pd(OAc) ₂ /PPh ₃ (2:4)	(<i>n</i> -Bu) ₃ N	DMF	24	140	14	42
11	Pd(OAc) ₂ /PPh ₃ (2:4)	K ₂ CO ₃	DMF	24	130	15	—
12	Pd(OAc) ₂ /PPh ₃ (2:4)	NaHCO ₃	DMF	24	130	Trace	—
13	Pd(PPh ₃) ₄ (2)	(<i>n</i> -Bu) ₃ N	DMF	24	130	10	—
14	PdCl ₂ (2)	(<i>n</i> -Bu) ₃ N	DMF	24	130	—	—
15	Pd(dba) ₂ ^d (2)	(<i>n</i> -Bu) ₃ N	DMF	24	130	20	Trace
16	Pd(OAc) ₂ (2)	Ag ₂ CO ₃	DMF	24	130	5	—

^a All reactions were carried out under argon using **1b** (1.0 equiv), 3-bromoquinoline (1.1 equiv), [Pd(OAc)₂] (2 mol %), Ph₃P, base (2.0 equiv) and solvent (10 mL) at the indicated temperature.

^b Oil bath temperature.

^c Isolated yield.

^d dba: dibenzylidene acetone.

Table 2The palladium-catalyzed domino reaction for the formation of fused pyrrolophenanthridine^{a,b}^a Reaction conditions: **a–p** (1.0 equiv), bromo(iso)quinoline (1.2 equiv), $\text{Pd}(\text{OAc})_2$ (2 mol %), PPh_3 (4 mol %), $(n\text{-Bu})_3\text{N}$ (1.2 equiv), DMF 10 mL, 125–130 °C, 24 h.^b Isolated yield after flash column chromatography.

3-bromopyridine is compatible with this palladium-catalyzed domino reaction. Several 7,11-diphenyl-9,10-dihydro-8*H*-pyrrolo[3,4-*j*]phenanthridine compounds were readily isolated in good to excellent yields, except in the case of **6d**, where 3-bromoisoquinoline was employed. The *para*- or *ortho*- substituted groups on the benzene ring of 1,6-diynes could be chloro, fluoro, methyl, methoxyl, and ethoxycarbonyl. Using 3-bromoquinoline with diyne substrates (**b**, **e–f**, **h**, **j**, **l**), the reaction yielded 7,11-diphenyl-9,10-dihydro-8*H*-pyrrolo[3,4-*j*]phenanthridine **3b**, **3e**, **3f**, **3h+4h**, **3j+4j**, and **3l+4l**, respectively, in yields higher than 83%. The yield of compound (**3l+4l**) was the highest at 87%. When 3-bromoquinoline was used in the reaction with **a**, **c,d,g,i,k**, and **m–p**, the desired pyrrolo[3,4-*j*]phenanthridines were obtained in good yields ranging from 56% to 75% (**3a**, **3c**, **3d**, **3g**, **3i**, **3k**, **3m**, **3n**, **3o**, and **5p**). The reaction of 3-bromoquinoline with diyne having a *para*-chloro, *ortho*-methyl, or *para*-fluoro group (**3h**, **3j**, and **3l**) gave much lower yields of the products than that of **4h**, **4j**, and **4l**. Compound **6d** yield was low (41%), but it is a novel 7,11-diphenyl-9,10-dihydro-8*H*-pyrrolo[3,4-*b*]phenanthridine. Finally, with these conditions, the generality of the cyclization was studied using 3-bromopyridine in place of 3-bromoquinoline. Results showed that the 4,9-diphenyl-2-tosyl-2,3-dihydro-1*H*-pyrrolo[3,4-*g*]isoquinolines **7b** and **7e** (Scheme 1) gave moderate yields of 52% and 50%, respectively. We isolated traces of **8e** besides the product **7e** when *N,N*-bis(3-(4-fluorophenyl)prop-2-yn-1-yl)-4-methylbenzenesulfonamide and 3-bromopyridine were used in the reaction. The regioselectivity of the reaction has no clear superiority when either electron-donating or electron-withdrawing groups on the aryl halides. A proposed pathway for this tandem Heck/C–H functionalization reaction is shown in Scheme 1.

All the resulting fused pyrrolo[3,4-*b*]phenanthridine compounds were confirmed by various spectroscopic techniques (¹H, ¹³C NMR, and IR spectroscopy, and HRMS). The molecular structures and relative configurations of **3e** (Fig. 1) and **8e** (Fig. 2) were unambiguously confirmed by single-crystal X-ray analysis (Supplementary data).¹⁵

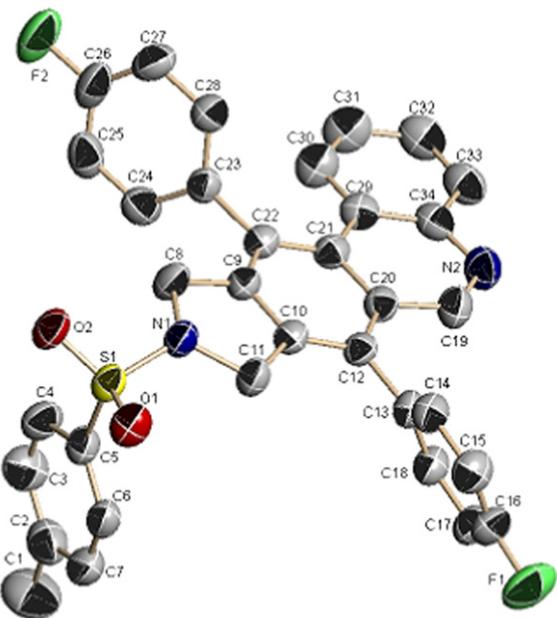


Fig. 1. ORTEP plot of **3e** showing ellipsoids at the 30% probability level. Selected bond lengths [Å] and angles [°]: C9–C10 1.396(5), C10–C12 1.384(5), C12–C20 1.417(5), C21–C22 1.418(5), C9–C22 1.375(5), C9–C10–C11 110.7(3), C11–C10–C12 129.3(3), C10–C12–C20 117.5(3), C9–C22–C21 118.2(3), C12–C20–C21 122.6(3).

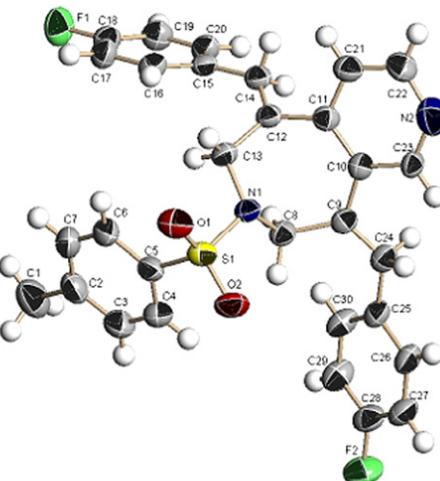
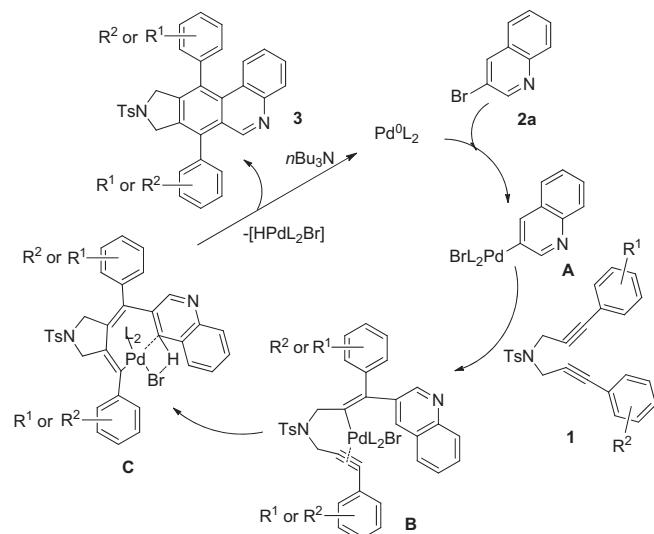


Fig. 2. ORTEP plot of **8e** showing ellipsoids at the 30% probability level. Selected bond lengths [Å] and angles [°]: C9–C10 1.487(3), C9–C24 1.336(3), C10–C11 1.402(3), C11–C12 1.491(3), C12–C14 1.337(3), N1–C8–C9 111.4(2), C8–C9–C10 119.0(2), C9–C10–C11 123.8(2), C10–C11–C12 124.4(2), C11–C12–C13 118.6(2), N1–C13–C12 111.3(2).

Simplified catalytic cycle mechanisms for selective formation pyrrolophenanthridines are proposed in Scheme 2. Oxidative addition of bromoquinoline (**2a**) would yield aryl-palladium intermediate **A**, which can subsequently undergo carbopalladation with the diyne moiety (**1**) to yield intermediate **B**. Intermediate **B** possesses a carbon–carbon double bond that can undergo carbopalladation and σ -bond metathesis onto the aryl group to yield intermediate **C**. Proton abstraction¹⁶ by the base yields pyrrolophenanthridine **3**.



Scheme 2. Proposed catalytic cycle.

3. Conclusions

We developed a tandem sequence method for the preparation of diversely substituted pyrrolo[3,4-*j*]phenanthridines from readily available precursors through multistep C–C bond formation and C–H activation of quinoline or isoquinoline ring. This series produced pyrrolo[3,4-*j*]phenanthridine derivatives in moderate to very good yields. This methodology will aid in generating more interesting structures for use in biological studies. Aside from the

4. (a) Ackermann, L. *Angew. Chem., Int. Ed.* **2011**, *50*, 3926–3928; (b) Ackermann, L. *Chem. Rev.* **2011**, *111*, 1315–1345; (c) Ackermann, L. *Pure Appl. Chem.* **2010**, *82*, 1403–1413; (d) Ackermann, L.; Kapdi, A. R.; Potukuchi, H. K.; Kozhushkov, S. I. *Syntheses via C–H Bond Functionalizations In Handbook of Green Chemistry*; Li, C.-J., Trost, B. M., Eds.; Wiley-VCH: Weinheim, 2011.
5. (a) Liegault, B.; Renaud, J.-L.; Bruneau, C. *Chem. Soc. Rev.* **2008**, *37*, 290–299; (b) Thirunavukkarasu, V. S.; Parthasarathy, K.; Cheng, C.-H. *Chem.—Eur. J.* **2010**, *16*, 1436–1440; (c) Donaldson, L. R.; Haigh, D.; Hulme, A. N. *Tetrahedron* **2008**, *64*, 4468–4477; (d) Lian, Y. J.; Davies, H. M. L. *J. Am. Chem. Soc.* **2010**, *132*, 440–441; (e) Della Ca', N.; Motti, E.; Mega, A.; Catellani, M. *Adv. Synth. Catal.* **2010**, *352*, 1451–1454.
6. (a) Pozharskii, A. F.; Soldatenkov, A. T.; Katritzky, A. R. *Heterocycles in Life and Society*; John Wiley: New York, NY, 1997; (b) Janin, Y. L.; Croisy, A.; Riou, J. F.; Bisagni, E. *J. Med. Chem.* **1993**, *36*, 3686–3692; (c) Nagashima, T.; Davies, H. M. L. *Org. Lett.* **2002**, *4*, 1989–1992; (d) Schultz, D. M.; Wolfe, J. P. *Org. Lett.* **2010**, *12*, 1028–1031; (e) Giampietro, N. C.; Wolfe, J. P. *Angew. Chem., Int. Ed.* **2010**, *49*, 2922–2924.
7. (a) Kim, K. H.; Lee, H. S.; Kim, S. H.; Kim, S. H.; Kim, J. N. *Chem.—Eur. J.* **2010**, *16*, 2375–2380; (b) Renú, O.; Lapointe, D.; Fagnou, K. *Org. Lett.* **2009**, *11*, 4560–4563; (c) Leibeling, M.; Koester, D. C.; Pawliczek, M.; Schild, S. C.; Werz, D. B. *Nat. Chem. Biol.* **2010**, *6*, 199–201; (d) Angelin, M.; Vongvilai, P.; Fischer, A.; Ramström, O. *Chem. Commun.* **2008**, 768–770.
8. (a) Harayama, T.; Hori, A.; Abe, H.; Takeuchi, Y. *Tetrahedron* **2004**, *60*, 1611–1616; (b) Ren, H.; Li, Z.; Knochel, P. *Chem.—Asian J.* **2007**, *2*, 416–433; (c) Makhey, D.; Gatto, B.; Yu, C.; Liu, A.; Liu, L. F.; LaVoie, E. J. *Bioorg. Med. Chem.* **1996**, *4*, 781–791; (d) Iwayama, T.; Sato, Y. *Chem. Commun.* **2009**, 5245–5247.
9. (a) Wasa, M.; Worrell, B. T.; Yu, J.-Q. *Angew. Chem., Int. Ed.* **2010**, *49*, 1275–1277; (b) Wang, D.-H.; Yu, J.-Q. *J. Am. Chem. Soc.* **2011**, *133*, 5767–5769; (c) Yoo, E. J.; Wasa, M.; Yu, J.-Q. *J. Am. Chem. Soc.* **2010**, *132*, 12378–12380; (d) Wasa, M.; Engle, K. M.; Yu, J.-Q. *J. Am. Chem. Soc.* **2009**, *131*, 9886–9887.
10. (a) Wang, D. H.; Engle, K. M.; Shi, B. F.; Yu, J.-Q. *Science* **2010**, *327*, 315–319; (b) Wasa, M.; Engle, K.; Yu, J.-Q. *Isr. J. Chem.* **2010**, *50*, 605–616; (c) Wasa, M.; Yu, J.-Q. *J. Am. Chem. Soc.* **2008**, *130*, 14058–14059.
11. (a) Candito, D. A.; Lautens, M. *Angew. Chem., Int. Ed.* **2009**, *48*, 6713–6716; (b) Blanchot, M.; Candito, D. A.; Larnaud, F.; Lautens, M. *Org. Lett.* **2011**, *13*, 1486–1489; (c) Alberico, D.; Scott, M. E.; Lautens, M. *Chem. Rev.* **2007**, *107*, 174–238; (d) Panteleev, J.; Geyer, K.; Aguilar-Aguilar, A.; Wang, L. T.; Lautens, M. *Org. Lett.* **2010**, *12*, 5092–5095; (e) Dockendorff, C.; Sahli, S.; Olsen, M.; Milhau, L.; Lautens, M. *J. Am. Chem. Soc.* **2005**, *127*, 15028–15029.
12. (a) Richmond, C. J.; Eadie, R. M.; Parenty, A. D. C.; Cronin, L. *J. Org. Chem.* **2009**, *74*, 8196–8202; (b) Ritchie, C.; Cooper, G. J. T.; Song, Y. F.; Streb, C.; Yin, H.; Parenty, A. D. C.; McLaren, D. A.; Cronin, L. *Nat. Chem.* **2009**, *1*, 47–52; (c) Richmond, C. J.; Parenty, A. D. C.; Song, Y. F.; Cooke, G.; Cronin, L. *J. Am. Chem. Soc.* **2008**, *130*, 13059–13065; (d) Kitson, P. J.; Parenty, A. D. C.; Richmond, C. J.; Long, D. L.; Cronin, L. *Chem. Commun.* **2009**, 4067–4069.
13. (a) Hu, Y. M.; Yu, C. L.; Ren, D.; Hu, Q.; Zhang, L. D.; Cheng, D. *Angew. Chem., Int. Ed.* **2009**, *48*, 5448–5451; (b) Hu, Y. M.; Ren, D.; Zhang, L. D.; Lin, X. G.; Wang, J. *Eur. J. Org. Chem.* **2010**, *23*, 4454–4459; (c) Hu, Y. M.; Qu, Y.; Wu, F. H.; Gui, J. H.; Wei, Y.; Hu, Q.; Wang, S. *Chem.—Asian. J.* **2010**, *5*, 309–314; (d) Hu, Y. M.; Ouyang, Y.; Qu, Y.; Hu, Q.; Yao, H. *Chem. Commun.* **2009**, 4575–4577.
14. (a) Hu, Y. M.; Sun, Y. J.; Hu, J. P.; Zhu, T.; Yu, T.; Zhao, Q. *S. Chem.—Asian. J.* **2011**, *6*, 797–800; (b) Sherden, N. H.; Behenna, D. C.; Virgil, S. C.; Stoltz, B. M. *Angew. Chem., Int. Ed.* **2009**, *48*, 6840–6843; (c) Hu, Y. M.; Yao, H.; Sun, Y. J.; Wang, J.; Lin, X. G.; Zhu, T. *Chem.—Eur. J.* **2010**, *16*, 7635–7641; (d) Meng, T. J.; Hu, Y. M.; Sun, Y. J.; Zhu, T.; Wang, S. *Tetrahedron* **2010**, *66*, 8648–8653.
15. Crystal structure determination: $C_{35}H_{22}Cl_3F_2N_2O_2S$ **3e**, $M=681.98$, triclinic, space group $P\bar{1}$, $a=11.301(5)$, $b=11.524(5)$, $c=12.526(5)$ Å, $U=1567.9(11)$ Å³, $T=293$ K, $Z=2$, 5555 reflections measured, 13,622 unique ($R_{\text{int}}=0.022$), which were used in all calculations. The final $wR(F^2)$ was 0.1238 (all data). Crystal structure determination: $C_{30}H_{26}F_2N_2O_2S$ **8e**, $M=2519.0(3)$, triclinic, space group $P\bar{1}$, $a=12.6080(2)$, $b=18.4828(12)$, $c=12.3061(8)$ Å, $U=2519.0(3)$ Å³, $T=293$ K, $Z=4$, 5787 reflections measured, 21,530 unique ($R_{\text{int}}=0.025$), which were used in all calculations. The final $wR(F^2)$ was 0.0762 (all data).
16. (a) García-Cuadrado, D.; Mendoza, P. D.; Braga, A. A. C.; Maseras, F.; Echavarren, A. M. *J. Am. Chem. Soc.* **2007**, *129*, 6880–6886; (b) GarcMa-Cuadrado, D.; Braga, A. A. C.; Maseras, F.; Echavarren, A. M. *J. Am. Chem. Soc.* **2006**, *128*, 1066–1067.
17. SAINT and SADABS Software Reference Manual; Bruker AXS: Madison, Wisconsin (USA), 2003.
18. (a) Sheldrick, G. M. *SHELXL97, Program for Crystal Structure Refinement*; University of Göttingen: Germany, 1997; (b) Sheldrick, G. M. *SHELXS97, Program for Crystal Structure Refinement*; University of Göttingen: Germany, 1997; (c) Sheldrick, G. M. *Acta Crystallogr., Sect. A: Found. Crystallogr.* **1990**, *46*, 467–473.
19. Molander, G. A.; Cormier, E. P. *J. Org. Chem.* **2005**, *70*, 2622–2626.