Synthesis of the spiroacetal-containing anti-*Helicobacter pylori* agents CJ-12,954 and CJ-13,014[†]

Margaret A. Brimble* and Christina J. Bryant

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The first synthesis of the spiroacetal-containing anti-*Helicobacter pylori* agents *ent*-CJ-12,954 and *ent*-CJ-13,014 is reported based on the union of a heterocycle-activated spiroacetal-containing sulfone fragment with a phthalidecontaining aldehyde fragment; comparison of the ¹H and ¹³C NMR data, optical rotations and HPLC retention times of the synthetic compounds (3S,2''S,5''S,7''S)-(1a) and (3S,2''S,5''S,7''S)-(2a) and the (3R)-diastereomers (3R,2''S,5''S,7''S)-(1b) and (3R,2''S,5''R,7''S)-(2b) with the naturally occurring compounds established that the synthetic isomers (1a) and (2a) were in fact enantiomeric to the natural products CJ-12,954 and CJ-13,014.

Helicobacter pylori is a Gram-negative micro-aerophilic spiral bacterium that resides in the mucus layer above the gastric epithelium¹ and can cause peptic ulcer disease and gastric cancer in humans.² The International Agency for Research on Cancer classified *H. pylori* as a class I carcinogen in 1994.³ A variety of effective drugs for the treatment and eradication of *H. pylori* infection are clinically useful, including antibiotics (β-lactams, macrolides and quinolones), bactericidal agents (bismuth salts), and antiprotozoal agents (metronidazole); however, drug resistance, side effects and non-compliance⁴ prompt development of more effective and selective anti-*H. pylori* agents.

Dekker *et al.*⁵ isolated seven new 5,7-dimethoxyphthalide antibiotics with specific anti-*H. pylori* activity from the basidiomycete *Phanerochaete velutina* CL6387. The two most potent compounds, CJ-12,954 **1** and its C-5" epimer CJ-13,014 **2**, contained a 5,5-spiroacetal ring joined through a polymethylene chain to the phthalide unit (Fig. 1). While changes in the stereochemistry associated with the spiroacetal have little effect on antibacterial activity, the diketone formed by ring opening exhibits a decreased potency of approximately 100-fold. Two structurally related helicobactericidal compounds, spirolaxine **3** and its methyl ether **4**, contain a 6,5-spiroacetal ring joined through a polymethylene chain to a phthalide unit.⁶ Thus, phthalidecontaining spiroacetal compounds provide promising new leads for the treatment of *H. pylori*-related diseases.

Whilst Dekker *et al.*⁵ were unable to assign the stereochemistry of the stereogenic centre at C-3 on the phthalide unit in CJ-12,954 **1** and CJ-13,014 **2**, they were able to assign the relative

Department of Chemistry, The University of Auckland, 23 Symonds St., Auckland, New Zealand. E-mail: m.brimble@auckland.ac.nz; Fax: +64 9 3737599; Tel: +64 9 3737599 ext. 88259 stereochemistry of the three stereogenic centres on the spiroacetal ring. CJ-12,954 **1** was assigned with 1,3-*syn* stereochemistry between the C2"-Me group and the C5"–O6" bond with the 6'-CH₂ group 1,3-*syn* to C5"–O1". In the case of CJ-13,014 **2** the C2"-Me group was assigned as 1,3-*anti* to the C5"–O6" bond with the 6'-CH₂ group 1,3-*anti* to the C5"–O1" bond. The structures of CJ-12,954 **1** and CJ-13,014 **2** were initially arbitrarily depicted with the (*S*)-configuration at both C2" and C7"; however, the assignment of absolute stereochemistry to these stereogenic centres and C-3 on the phthalide unit requires the synthesis of these natural products which has not been reported to date although several simpler non spiroacetal-containing phthalides have been prepared with lack of stereocontrol of the phthalide unit.^{7–9}

We have reported¹⁰ the synthesis of (+)-spirolaxine methyl ether by coupling a 6,5-spiroacetal moiety of defined stereochemistry with a stereochemically-defined phthalide moiety, thus establishing the absolute stereochemistry of the natural product to be (3R,2''R,5''R,7''R). An alternative synthesis in which the stereochemistry at C-3 in the phthalide unit was not controlled necessitated separation of (+)-spirolaxine methyl ether from its C-3 diastereomer by HPLC in the final step.¹¹

We herein report a flexible convergent synthesis of CJ-12,954 and CJ-13,014 initially focusing on the synthesis of (3S,2''S,5''S,7''S)-(1a) and (3S,2''S,5''R,7''S)-(2a) arbitrarily chosen with the (*S*)-configuration at C-3 on the phthalide unit and at C-2'' and C-7'' in the spiroacetal. The key step involves modified Julia olefination of phthalide-aldehyde 5a (Scheme 1) with heterocycleactivated sulfones 6 and 7 (Scheme 2).

The (S)-stereochemistry at C-3 in phthalide-aldehyde **5a** (Scheme 1) was established by asymmetric reduction of ketone 8^{12} using (*R*)-2-Me-CBS-oxazaborolidine¹³ and borane–dimethyl sulfide affording homoallylic alcohol **9** in 92% yield and 94% e.e.¹⁴



Spirolaxine 3: R= H; Spirolaxine methyl ether 4: R= Me



[†] Electronic supplementary information (ESI) available: Experimental section and Fig. S2 depicting key NMR data for **1a/2a** and **1b/2b**. See DOI: 10.1039/b612757f



Scheme 1 Reagents and conditions: (i) (R)-MeCBS, BH₃–SMe₂, 15 min, then THF, **8**, 2 h, 92%, 94% ee; (ii) NBS, NH₄OAc, Et₂O, 24 h, 90%; (iii) NaH, THF, 0 °C then N,N-diethylcarbamoyl chloride, 90%; (iv) t-BuLi, THF, -78 °C, 2 h then camphorsulfonic acid, 20 °C, 12 h, 70%; (v) 2-methyl-2-butene, BH₃–SMe₂, THF, 0 °C then MeOH, NaOH, 30% H₂O₂, 71%; (vi) TPAP, NMO, CH₂Cl₂, 4Å mol. sieves, 6 h, 20 °C, 72%

Regioselective bromination of the aromatic ring afforded bromide **10** and subsequent conversion to diethyl carbamate **11** facilitated subsequent lithium–halogen exchange and intramolecular acylation to phthalide **12**. Hydroboration of the allyl group followed by oxidation then provided the desired phthalide-aldehyde **5a** in higher optical purity than its antipode which was prepared *via* an asymmetric allylation strategy.¹⁰

Attention next focused on the synthesis of sulfones **6** and **7** which are epimeric at the spirocentre (Scheme 2). 1-Phenyl-1*H*-tetrazol-5-yl sulfones **6** and **7** were chosen in preference to the use of benzothiazol-2-yl sulfones due to their increased stability in heterocycle-modified Julia olefinations.^{15,16} Lithium (*S*)-acetylide 17^{17} provides access to the 5,5-spiroacetal ring system with (*S*)-stereochemistry at C-2" and the (7"*S*)-stereochemistry is derived from homoallylic alcohol **14**, available *via* asymmetric allylation of aldehyde **13**.¹⁸

Addition of allylmagnesium bromide to (+)- β -diisopinocampheylmethoxyborane followed by addition of aldehyde 13¹⁸ afforded (*S*)-alcohol 14¹⁹ in 82% yield and 94% ee (determined by chiral HPLC‡). Silyl ether formation followed by hydroboration and oxidation of the resultant primary alcohol 15 afforded aldehyde 16. Addition of aldehyde 16 to lithium acetylide 17 at -78 °C in the presence of lithium bromide²⁰ provided alcohol 18 as a mixture of diastereomers that was oxidized to ketone 19 using TPAP and NMO. Reduction of the acetylene over PtO₂ followed by spirocyclisation using camphorsulfonic acid in dichloromethane afforded an inseparable 1 : 1 mixture of spiroacetals 21 and 22 after cleavage of the *tert*-butyldiphenylsilyl ether. Lack of stereocontrol from the anomeric effect²¹ contributed to the observed formation of equal quantities of 5,5-spiroacetals 21 and 22.

Mitsunobu displacement of hydroxyspiroacetals **21** and **22** with 1-phenyl-1*H*-tetrazole-5-thiol, PPh₃ and DEAD afforded sulfides **23** and **24** that underwent oxidation to an inseparable mixture of sulfones **6** and **7**. Finally the key heterocycle-activated¹⁵ modified



Scheme 2 *Reagents and conditions:* (i) allyl bromide, Mg, (+)- β -diisopinocampheylmethoxyborane, Et₂O, -78 °C to 20 °C, 82%, 94% ee; (ii) *t*-BuMe₂SiCl, imidazole, DMAP, CH₂Cl₂, 20 °C, 12 h, 90%; (iii) 2-methyl-2-butene, BH₃–SMe₂, 0 °C, 76%; (iv) Dess-Martin periodinane, py, CH₂Cl₂, 20 °C, 77%; (v) 17, n-BuLi, LiBr, THF, -78 °C, then 16, 84%; (vi) TPAP, NMO, 4Å mol sieves, CH₂Cl₂, 20 °C, 94%; (vii) H₂, PtO₂, K₂CO₃, THF–MeOH (1 : 1), 94%; (viii) CSA, CH₂Cl₂, 20 °C, 4 h, 93%; (ix) TBAF, CH₂Cl₂, 20 °C, 3 h, 77%; (x) 1-phenyl-1H-tetrazole-5-thiol, Ph₃P, DEAD, 78%; (xi) *m*-CPBA, NaHCO₃, 71%; (xii) KHMDS, THF, -78 °C then 5a, 84%; (xiii) H₂, PtO₂, K₂CO₃, THF–MeOH (1 : 1), 85%.



Scheme 3 *Reagents and conditions:* (i) 6 and 7 (1 : 1), KHMDS, THF, -78 °C then 5b, 76%; (ii) H₂, PtO₂, K₂CO₃, THF–MeOH (1 : 1), 90%.

Julia olefination using KHMDS proceeded in excellent yield (84%) providing a 1 : 1 mixture of phthalide-spiroacetals (3S,2''S,5''S,7''S)-(1a) and (3S,2''S,5''R,7''S)-(2a) after hydrogenation over PtO₂.

The ¹H and ¹³C NMR data recorded for phthalide-spiroacetals **1a** and **2a** were compared with the data reported for the natural products.⁵ Notably the chemical shifts observed for the key resonances at the stereogenic centres in the spiroacetal unit (C2", C5", C7" and 2"-Me) were in good agreement with the natural products (Fig. S2†). However, the chemical shift reported for H3 ($\delta_{\rm H}$ 5.29) in both **1a** and **2a** was at variance with the chemical shift reported for the same resonance in natural CJ-12,954 and CJ-13,014 ($\delta_{\rm H}$ 5.27). Further clarification of the relative stereochemistry between C3 on the phthalide with the stereogenic centres in the 5,5-spiroacetal ring was clearly required.

Due to the ready availability of (3*R*)-phthalide-aldehyde $5b^{10}$ we also prepared a 1 : 1 mixture of (3*R*,2"*S*,5"*S*,7"*S*)-(1b) and (3*R*,2"*S*,5"*R*,7"*S*)-(2b) with (3*R*)-stereochemistry on the phthalide (Scheme 3) *via* olefination of (3*R*)-phthalide-aldehyde 5b with a 1 : 1 mixture of sulfones 6 and 7 followed by hydrogenation. Frustratingly, the ¹H and ¹³C NMR data obtained for these latter isomers were similar to those recorded for both synthetic isomers (1a) and (2a) and the respective natural products (Fig. S2[†]).

Gratifyingly, procurement of samples of natural CJ-12,954 and CJ-13,014 allowed direct comparison of the HPLC retention times for the synthetic compounds with the natural products. Using the reported HPLC conditions⁵§ the retention times for the 1 : 1 mixture of synthetic (3S,2''S,5''S,7''S)-(**1a**) and (3S,2''S,5''R,7''S)-(**2a**) were in agreement with those recorded for natural CJ-12,954 (**1**) and CJ-13,014 (**2**), and differed from the retention times recorded for the 1 : 1 mixture of synthetic (3R,2''S,5''S,7''S)-(**1b**) and (3R,2''S,5''R,7''S)-(**2b**). The [α]_D – 38.0 (*c*, 0.48, CHCl₃) for the 1 : 1 mixture of (3S,2''S,5''R,7''S)-(**1a**) and (3S,2''S,5''R,7''S)-(**2a**) was of opposite sign and an average of the values reported⁵ for CJ-12,954 (**1**), [α]_D +6.0 (*c*, 0.07, CHCl₃), and CJ-13,014 (**2**), [α]_D +71.2 (*c*, 0.11, CHCl₃), establishing that the synthetic isomers (3S,2''S,5''S,7''S)-(**1a**) and (3S,2''S,5''R,7''S)-(**2a**) were in fact enantiomeric to the natural products.

In summary, the first synthesis of the enantiomers of the helicobactericidal agents CJ-12,954 and CJ-13,014, namely (3S,2"S,5"S,7"S)-(1a) and (3S,2"S,5"R,7"S)-(2a), has been achieved *via* modified Julia olefination of (3S)-phthalide-aldehyde 5a with a 1 : 1 mixture of heterocyclic sulfones 6 and 7. Complementary synthesis of the diastereomers (3R,2"S,5"S,7"S)-(1b) and (3R,2"S,5"R,7"S)-(2b) facilitated confirmation of the relative stereochemistry between C-3 on the phthalide unit and C5"/C7" on the 5,5-spiroacetal moiety, establishing that the absolute configuration of the natural product CJ-12,954 is (3R,2"R,5"R,7"R) and that of CJ-13,014 is (3R,2"R,5"S,7"R).

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Notes and references

[‡] HPLC conditions: Chiracel[®] OD-H column, *i*-propanol : hexane 5 : 95, flow rate 0.5 mL min⁻¹, retention times: 7.5 min (minor, *R*-isomer) and 8.7 min (major, *S*-isomer).

 $\$ HPLC conditions: YMC-Pack ODS-AM column, methanol : water 3 : 1, flow rate 0.5 mL min^{-1}.

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