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N-Aryl-O-glycosyl Haloacetimidates as Glycosyl Donors

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Reaction of 1-O-unprotected tetra-O-acetyl- and tetra-O-benzyl-glucopyranose with N-aryl haloacetimidoyl chlorides in the presence of sodium hydride and 15-crown-5 afforded N-aryl-O-glucopyranosyl haloacetimidates. Mainly the β -anomers were obtained in this anomeric O-acylation-type reaction. The glycosyl donor properties of these haloacetimidates were investigated with 6-O- and 4-O-unprotected glucopyranosides as acceptors. The results were compared with those obtained with the corresponding O-glucopyranosyl trichloroacetimidates as glycosyl donors and the same acceptors. It was found that N-(2-chloro-6-methylphenyl)-O-glucopyranosyl trifluoroacetimidates (16Ad, 16Bd) exhibit glycosyl donor properties closely related to those of the corresponding N-unsubstituted O-glucopyranosyl trichloroacetimidates (12A, 12B).

Keywords Carbohydrates; Glycosidation; *N*-Aryl-*O*-glycosyl haloacetimidates; Synthesis; *O*-Imidoylation

INTRODUCTION

O-Glycosyl trichloroacetimidates have become highly useful glycosyl donors as they combine the most convenient preparation with excellent glycosyl donor properties under simple acid catalysis. [1–3] Replacement of the trichloromethyl group, for instance, by the trifluoromethyl group [4] or the dichloro-cyanomethyl group [5] was reported. So far, these modifications of the haloacetimidate leaving group did not exhibit advantages over the trichloroacetimidate group in glycosylation reactions.

Introduction of a substituent at the nitrogen of the imidate group can be reached by replacing the convenient base-catalyzed reversible addition of the anomeric hydroxy group to electron-poor nitriles permitting even anomeric

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Figure 1: Imidoylation of 1-O-unprotected sugars by N-substituted imidoyl halides.

stereocontrol for a reaction with *N*-substituted imidoyl halides (Fig. 1).^[6,7] Since this anomeric *O*-imidoylation reaction is generally irreversible, anomeric stereocontrol can hardly be achieved.^[7,8] However, the available modification of the leaving group character may compensate for the less convenient preparation of such *N*-substituted *O*-glycosyl haloacetimidate donors. As *O*-glycosyl-*N*-phenyl-trifluoroacetimidates (Fig. 1, Ar=Ph, X=Y=Z=F) have previously been investigated as glycosyl donors in various glycosylation reactions,^[8,9] earlier comparative studies by our group in this direction will be disclosed.^[7]

RESULTS AND DISCUSSION

At first the synthesis and the reactions of N-aryl O-glucopyranosyl trichloroacetimidates were investigated (Sch. 1). To this end, aniline (1a) was transformed via the N-trichloroacetyl intermediate 2a into known imidoyl chloride 3a. [10,11] Anomeric O-imidoylation of O-benzyl-protected glucose **5B** with **3a** in the presence of sodium hydride as base and 15-crown-5 as complexing agent for sodium afforded the desired O-glucopyranosyl-N-phenyl trichloroacetimidate **6Ba** in excellent yield as anomeric mixture ($\alpha/\beta = 3.7$). [12] Following reaction of this donor with a primary and a secondary hydroxy group as acceptor, as, for instance, is present in the 6-O- and 4-O-unprotected glucopyranosides 8 and 9, respectively, led under standard TMSOTf catalysis at rt to a mixture of the known anomers of $10B^{[13,14]}$ and $11B^{[15]}$ in good yield (Table 1, entry 1). Comparison of this glycosylation result with that for the corresponding N-unsubstituted β -configurated O-glucopyranosyl trichloroacetimidate 12B^[12,13] (that is most closely related to the anomeric mixture of **6Ba**) exhibited almost quantitative formation of **10B** and **11B** though in lower anomeric selectivity (Table 1, entry 7). Hence, the preference for the α -anomers in 10B and 11B with 6Ba as donor reflects the lower reactivity of 6Ba favoring an S_N 2-type reaction, as also was observed for **12B** at low temperatures

 $\label{eq:Scheme 1: N-Aryl-O-glucopyranosyl-haloacetimidate formation and glycosylation of acceptors \textbf{8} and \textbf{9}. Reagents and conditions: (a) $Cl_3C-COCl$, Pyr; (b) $POCl_3$, PCl_5; (c) PPh_3, CCl_4, CH_2Cl_2; (d) NaH, 15-crown-15, CH_2Cl_2, rt; (e) $TMSOTf$ (0.1 eq), CH_2Cl_2, rt (see Table 1). }$

Table 1: Glycosylation results with glycosyl donors 6, 7, 12, and 16 and acceptors 8 and 9 leading to disaccharides 11A, B and 11A, Ba

Entry	Glycosyl Donor $(\alpha/\beta)^b$	Product with Glycosyl Acceptor 8 ($\mathbb{R}^6 = \mathbb{H}$)	Product with Glycosyl Acceptor 9 ($R^4 = H$)
1	6Ba (3:7)	10B 82%, α : β = 4:1	11B 71%, α : β = 5:1
2	7Αα (β)	10A 80%, β΄	11A 71%, β΄
3	7Ab (2:3)	10A 70%, β	11A 62%, β
4	7Bb (1:7)	10B 77%, α : $\beta = 5$:1	11B 69%, α : β = 5:1
4 5	7Bc (1:9)	10B 87%, α : $\beta = 3$: 1	11B 78%, α : $\beta = 2:1$
6 7	$12A(\beta)$	10A 82%, β	11A 74%, β
7	$12B(\beta)$	10B 98%, α : $\beta = 3:2$	11B 85%, α : β = 1:1
8	16Aa (1:3)	10A 74%, β΄	11A 67%, β΄
8 9	16Ab (1:2)	10A 59%, β	11A 53%, β
10	16Ac (1:4)	10A 80%, β	11A 71%, β
11	16Ad (2:1)	10A 85%, β	11A 74%, β
12	16 B α (β)	10B 85%, α : β = 1:1	11B 74%, α : β = 1:1
13	16Bb (1:5)	10B 71%, α : $\beta = 1:1$	11B 63%, α : $\beta = 1:1$
14	16Bc (β)	10B 91%, α : $\beta = 1:1$	11B 82%, α : $\beta = 1:1$
15	16Bd (β)	10B 98%, α : β = 1:1	11B 89%, α : β = 1:1

^aAll reactions were carried out in dichloromethane as solvent and TMSOTf (0.1 equivalents) as catalyst at rt.

or by using a less polar solvent; alternatively, through intimate ion pair formation of the activated glycosyl donor, a memory effect could influence the stereocontrol.[1,2]

Thereafter, the reaction of O-acetyl-protected **5A** with imidoyl chloride 3a was investigated. However, presumably due to the lower reactivity of 5A and the steric bulk of the trichloromethyl group, the reaction was met with difficulties. Therefore, we took advantage of a dehalogenation procedure found for the transformation of 2a into dichloroacetimidoyl chloride 4a with the help of triphenylphosphine and carbon tetrachloride^[17]; this procedure was similarly applied to **2b-d**, furnishing the corresponding imidoyl chlorides **4b-d**. Reaction of 5A with 4a and 4b readily provided O-acetyl-protected glucopyranosyl dihaloacetimidates 7Aa and 7Ab; similarly, reaction of 5B with 4b and **4c** afforded the O-benzyl-protected O-glucopyranosyl dichloroacetimidates **7Bb** and **7Bc**. The glycosylation results found for **7Aa** with $8^{[18]}$ and $9^{[15]}$ as acceptors (Table 1, entry 2) were closely related to those found for the related β -configurated O-glucopyranosyl trichloroacetimidate $\mathbf{12A}^{[1,2,16]}$ with $\mathbf{8}$ and $\mathbf{9}$ as acceptors (Table 1, entry 6). Based on the neighboring group participation, as expected, in both cases only the β -linked disaccharides 10B and 11B were obtained. Introduction of an electron-withdrawing trifluoromethyl substituent into the N-phenyl group as in **7Ab** and **7Bb** did not improve the glycosylation yields with acceptors 8 and 9 (Table 1, entries 3 and 4). However, it is noteworthy that donor **7Bb** furnished mainly α -linked disaccharides (Table 1,

bThe anomeric ratio was taken from the ¹H NMR spectra.

entry 4). The electron-donating 2-methoxy substituent in the N-phenyl group as in **7Bc** afforded almost glycosylation yields (Table 1, entry 5) as those found for 12B (Table 1, entry 7). [1,2,16] This result is presumably due to bidentate complexation of the catalyst by the leaving group, thus exhibiting a remote activation effect, as previously found for the 3-methoxy-2-pyridyloxy (MOP) leaving group.[19]

In order to generate better and less bulky N-aryl-haloacetimidate leaving groups, the related trifluoroacetimidate analogs also were prepared (Sch. 2). To this end, anilines **1a-d** were transformed with N-trifluoroacetylbenzotriazole^[20] (13) into the trifluoroacetanilides 14a-d. Reaction with triphenylphosphine/carbon tetrachloride in dichloromethane readily gave the Naryl imidoyl chlorides 15a-d that led with 5A and 5B in the presence of sodium hydride and 15-crown-5 to the O-acetyl-protected glucopyranosyl trifluoroacetimidates **16Aa–16Ad** and to the *O*-benzyl-protected analogs **16Ba–16Bd**, respectively; generally, mainly or exclusively the β -anomers were formed. The glycosidation reaction with acceptors 8 and 9 under the aforementioned conditions clearly showed that donors 16 are superior to the trichloro- (6Ba) and dichloroacetimidate analogs 7 (Table 1, entries 8-15). However, only the compounds with 2,6-di-substitution in the N-phenyl ring (16Ad and 16Bd) led to glycosylation results (Table 1, entries 11 and 15) as found for the corresponding trichloroacetimidates 12A and 12B (Table 1, entries 6 and 7). Hence,

Scheme 2: N-Aryl-O-glucopyranosyl-trifluoroacetimidate formation and glycosylation of acceptors **8** and **9**. Reagents and conditions: (a) CH_2CI_2 ; (b) PPh_3 , CCI_4 , CH_2CI_2 ; (c) NaH, 15-crown-15, CH_2Cl_2 , rt; (d) TMSOTf (0.1 eq), CH_2Cl_2 , rt (see Table 1).

these compounds possess similar glycosyl donor properties as 12A and 12B, which is presumably due to additional release of steric strain in the glycosylation step. Catalysis of these glycosylation reactions with BF₃·OEt₂ was also possible. [7] With glycosyl donors **16Ba–16Bd**, the α,β -selectivities in the glycosidation reactions dropped (Table 1, entries 12-15) as was also found for trichloroacetimidate 12B when reacted in dichloromethane at rt.[1,2] Hence, it seems that the ease of formation and the reactivity potential of the O-glycosyl trichloroacetimidates (as, for instance, 12A, B) can be matched by structurally varied N-aryl-O-glycosyl haloacetimidates.

CONCLUSION

N-Aryl-O-glucopyransoyl haloacetimidates were prepared and their glycosyl donor properties were investigated. The best glycosyl donor properties were found for N-(2-chloro-6-methylphenyl)-O-glucopyranosyl trifluoroacetimidates (compounds 16Ad and 16Bd). Comparison of these glycosyl donors with the corresponding O-glucopyranosyl trichloroacetimidates (12A and 12B) exhibited very similar behavior. Interestingly, the less reactive N-aryl-O-(tetra-Obenzyl-glucopyranosyl) dichloroacetimidates (**7Bd** and **7Bc**), having mainly β configuration, furnished in the glycosylation reaction higher amounts of the inversion products than the more reactive glycosyl donors 12B or 16Bd. Yet, similar results were previously observed for 12B when the reactivity was decreased by lowering the reaction temperature.

EXPERIMENTAL

Solvents were purified by standard procedures: ¹H NMR spectra: Bruker AC-250 (250 MHz): internal standard tetramethylsilane (TMS); flash chromatography: silica gel 60 (Baker: 0.04–0.063 mm); thin-layer chromatography (TLC): foil plates, silica gel 60 F₂₅₄ (Merck: layer thickness 0.2 mm); optical rotations: Perkin-Elmer polarimeter 241 MC: 1-dm cell, 20°C; elemental analyses: Heraeus CHN-O-Rapid; MALDI-MS: Kratos (Kompac Maldi 1); ESI-MS: Bruker Esquire 3000 plus.

General Procedure A: Trichloroacetylation of Anilines (1a–d)

To a solution of aniline (2 eq) in dry dioxane (2 mL/1 mmol aniline) was added trichloroacetyl chloride (1 eq) under cooling with ice. After stirring for 30 min, water was added and the reaction mixture extracted three times with ethyl acetate (10 mL/1 mmol aniline). The organic phases were combined and dried with MgSO₄, the filtrate concentrated in vacuo, and the resulting product purified.

General Procedure B: Trifluoroacetylation of Anilines (1a–d)

The aniline (1 eq) was dissolved in dry dichloromethane (2 mL/1 mmol aniline). To this solution was added N-trifluoro-acetyl-benzotriazole (13, 1.1 eq) in portions and then the mixture was stirred for 3 h at rt. After concentration of this mixture, pure product was obtained by flash chromatography.

General Procedure C: Synthesis of the Imidoyl Chlorides 4a–d and 15a-d

The anilides (2a-d, 14a-d) (1 eq) and triphenylphosphane (1.1 eq) were dissolved in a 1:1 mixture of carbon tetrachloride and dichloromethane (2 mL/1 mmol anilide) and the reaction mixture stirred for 4 h at 60°C. Then half of the solvent was removed in vacuo and the residue purified by flash chromatography on ALOX (Aluminia N32–63 active) with dry toluene as solvent.

General Procedure D: Synthesis of N-aryl-O-glucopyranosyl Haloacetimidates 6Ba, 7Aa, 7Ab, 7Bb, 7Bc, 16Aa–Ad and 16Ba-Bd

5A or **5B** (1 mmol) was dissolved in dry dichloromethane (5 mL) under argon and then 15-crown-5 (220 μ L, 1.1 mmol) and sodium hydride (26 mg, 1.1 mmol) added under stirring at rt. After 5 min N-aryl-imidoyl chloride (1 to 1.1 mmol) was added and stirring continued for 30 min. The reaction mixture was neutralized with acetic acid and then concentrated in vacuo. The residue was purified by flash chromatography with dry solvents.

General Procedure E: Glycosylation Reactions Furnishing 10A, B and 11A, B

Under argon the glycosyl donor (0.26 mmol) and the acceptor 8 or 9 (0.2 mmol) were dissolved in dry dichloromethane (0.4 mL). A catalytic amount of TMSOTf (0.02 mmol) was then added at rt. After 1 h, the reaction mixture was neutralized with triethylamine and concentrated in vacuo. Flash chromatography with toluene/ethyl acetate (20:1) gave the disaccharides that were identical with independently synthesized material: 10A, ref. 13; 10B, ref. 14; 11A, ref. 15; **11B**, ref. 15.

2,2,2-Trichloroacetyl-anilide (2a) and 2,2,2-Trichloro-N-phenyl-acetimidoyl Chloride (3a)

These compounds were obtained as previously described. [10,11]

N-(2-Trifluoromethylphenyl)-trichloroacetamide (2b)

Following general procedure A, from 1b (2.51 mL, 20 mmol) 2b (3.05 g, qu.) was obtained as yellow crystals. m.p. 46°C; R_f (toluene/ethyl acetate, 4:1) 0.68. ¹H NMR (250 MHz, CDCl₃): δ 8.75 (1 H, NH), 8.12 (m_c, 1 H, 6-H), 7.85 (m_c, 1 H, 3-H), 7.71-7.55 (m, 2 H, 4-H, 5-H). Anal. Calcd for $C_9H_5Cl_3F_3NO$ (306.50): C, 35.27; H, 1.64; N, 4.57. Found C, 35.20; H, 1.71; N, 4.67.

N-(2-Methoxyphenyl)-trichloroacetamide (2c)

Following general procedure A, from 1c (12 g, 1 mol) 2c (9.51 g, 36%) was obtained as clear viscous mass. R_f (toluene) 0.31. ¹H NMR (250 MHz, CDCl₃): δ 9.15 (1 H, NH), 8.30 (m_c, 1 H, 6-H), 7.17 (m_c, 1 H, 4-H), 7.01 (m_c, 1 H, 5-H), 6.93 (m_c, 1 H, 3-H), 3.91 (s, 3 H, OCH₃). Anal. Calcd for C₉H₈Cl₃NO₂ (268.52): C, 40.26; H, 3.00; N, 5.22. Found C, 40.45; H, 3.08; N, 5.23.

N-(2-Chloro-6-methylphenyl)-trichloroacetamide (2d)

Following general procedure A, from 1d (6.15 mL, 0.05 mol) 2d (6.10 g, 42%) was obtained as white crystals. m.p. 141°C; R_f (toluene) 0.15. ¹H NMR $(250 \text{ MHz}, \text{CDCl}_3): \delta 7.23-6.98 \text{ (m, 3 H, 3-H, 4-H, 5-H)}, 2.11 \text{ (s, 3 H, CH}_3). Anal.$ Calcd for C₉H₇Cl₄NO (286.97): C, 37.67; H, 2.46; N, 4.88. Found C, 37.38; H, 2.28; N, 4.86.

N-Phenyl-dichloroacetimidoyl Chloride (4a)

Following general procedure C, from **2a** (8.3 g, 35 mmol) **4a** (2.72 g, 35%) was obtained as colorless fluid. R_f (toluene) 0.89. ¹H NMR (250 MHz, CDCl₃): δ 7.42–6.97 (m, 5 H, 2-H, 3-H, 4-H, 5-H, 6-H), 6.43 (s, 1 H, CHCl₂). Anal. Calcd for C₈H₆Cl₃N (222.50): C, 43.18; H, 2.72; N, 6.30. Found C, 43.40; H, 2.73; N, 6.32.

N-(2-Trifluoromethylphenyl)-dichloroacetimidoyl Chloride (4b)

Following general procedure C, from 2b (1.53 g, 5 mmol) 4b (485 mg, 33%) was obtained as yellow fluid. R_f (toluene/ethyl acetate, 4:1) 0.98. ¹H NMR (250 MHz, $CDCl_3$): δ 7.69 (m_c, 1 H, 6-H), 7.58 (m_c, 1 H, 4-H), 7.33 (m_c, 1 H, 5-H), $6.90 \, (m_c, 1 \, H, 3-H), 6.48 \, (s, 1 \, H, CHCl_2)$. Anal. Calcd for $C_9H_5Cl_3F_3N \, (290.50)$: C, 37.21; H, 1.73; N, 4.82. Found C, 37.56; H, 1.78; N, 5.01.

N-(2-Methoxyphenyl)-dichloroacetimidoyl Chloride (4c)

Following general procedure C, from **2c** (1.85 g, 7 mmol) **4c** (743 mg, 43%) was obtained as colorless oil. R_f (toluene) 0.81. ¹H NMR (250 MHz, CDCl₃): δ 7.28–6.87 (m, 4 H, 3-H, 4-H, 5-H, 6-H), 6.53 (s, 1 H, CHCl₂), 3.84 (s, 3 H, CH₃). Anal. Calcd for C₉H₈Cl₃NO (252.52): C, 42.81; H, 3.19; N, 5.55. Found C, 43.31; H, 3.29; N, 5.50.

N-(2-Chloro-6-methylphenyl)-dichloroacetimidoyl Chloride (4d)

Following general procedure C, from 2d (2.87 g, 10 mmol) 4d (1.19 g, 44%) was obtained as viscous yellowish oil. R_f (toluene) 0.85. ¹H NMR (250 MHz, CDCl₃): δ 7.29–6.99 (m, 3 H, 3-H, 4-H, 5-H), 6.53 (s, 1 H, CHCl₂), 2.08 (s, 3 H, CH₃). Anal. Calcd for C₉H₇Cl₄N (270.97): C, 39.89; H, 2.60; N, 5.18. Found C, 40.36; H, 2.69; N, 5.18.

N-Phenyl-O-(2,3,4,6-tetra-O-benzyl- α , β -D-glucopyranosyl) Trichloroacetimidate (6Ba)

Following general procedure D, from 5B (540 mg, 1 mmol) and 3a (250 mg, 1 mmol) **6Ba** (647 mg, 85%) was obtained as yellowish oil. R_f (toluene/ethyl acetate, 4:1) 0.98, α/β 3:7. ¹H NMR (250 MHz, CDCl₃): δ 7.01–6.73 (m, 25 H, 2'-H, 3'-H, 4'-H, 5'-H, 6'-H, 4 Ph), 6.67 (d, 0.3 H, $J_{1,2} = 3.6$ Hz, $1-H_{\alpha}$), 5.91 (d, $0.7 \text{ H}, J_{1,2} = 8.4 \text{ Hz}, 1\text{-H}_{\beta}, 5.12\text{--}4.51 \text{ (m, 8 H, 4 CH}_{2}\text{Ph)}, 4.22 \text{ (dd, } 0.3 \text{ H}, J_{3,2} = 0.00 \text{ H}$ $9.7, J_{3,4} = 9.8 \text{ Hz}, 3-H_{\alpha}), 3.90-3.62 \text{ (m, } 5.7 \text{ H, } 2-\text{H, } 3-\text{H}_{\beta}, 4-\text{H, } 5-\text{H, } 6_{\text{A}}-\text{H, } 6_{\text{B}}-\text{H}).$ $C_{42}H_{40}Cl_3NO_6$ (759): ESI-MS (negative mode): m/z = 792 (M-2 H+Cl)⁻, 758 $(M-H)^-$, 724 $(M-Cl)^-$.

N-Phenyl-O-(2,3,4,6-tetra-O-acetyl- α , β -D-glucopyranosyl) Dichloroacetimidate (7Aa)

Following general procedure D, from 5A (340 mg, 1 mmol) and 4a (230 mg, 1 mmol) **7Aa** (358 mg, 69%) was obtained as clear oil. R_f (toluene/ethyl acetate, 6:1) 0.18. ¹H NMR (250 MHz, CDCl₃): δ 7.40–6.79 (m, 5 H, NPh), 6.12 (s, 1 H, $CHCl_2$), 5.93 (d, 1 H, $J_{1,2} = 8.2$ Hz, 1-H_{β}), 5.41–5.17 (m, 3 H, 2-H, 3-H, 4-H), 4.38-4.12 (dd, dd, 2-H, $J_{\text{gem}} = 12.3$, $J_{6A,5} = 4.5$, $J_{6B,5} = 2.3$ Hz, 6_{A} -H, 6_{B} -H), $3.90~(\mathrm{ddd}, 1~\mathrm{H}, J_{5,4} = 9.5, J_{5,6\mathrm{A}} = 4.5, J_{5,6\mathrm{B}} = 2.3~\mathrm{Hz}, 5\mathrm{-H}), \, 2.02\mathrm{-}2.09~(4\mathrm{s}, 12~\mathrm{H}, 1.0\mathrm{Hz})$ 4 OAc). C₂₂H₂₅Cl₂NO₁₀ (534.43). FAB-MS (positive mode, matrix: NBOH with NaI): m/z = 533/535 (M)⁺.

N-(2-Trifluoromethyl-phenyl)-O-(2,3,4,6-tetra-O-acetyl- α , β -Dglucopyranosyl) Dichloroacetimidate (7Ab)

Following general procedure D, from **5A** (340 mg, 1 mmol) and **4b** (280 mg, 1 mmol) **7Ab** (186 mg, 32%) was obtained as white foam. R_f (toluene/ethyl acetate, 6:1) 0.24. ¹H NMR (250 MHz, CDCl₃): δ 7.68–7.14 (m, 3-H, 3'-H, 5'-H, 6'-H), 6.89-6.79 (m, 1 H, 4'-H), 6.60 (d, 0.4 H, $J_{1,2} = 3.7$ Hz, $1-H_{\alpha}$), 5.99 (s, 0.4H, CHCl₂), 5.97 (s, 0.6 H, CHCl₂), 5.92 (d, 0.6 H, $J_{1,2} = 7.9$ Hz, 1-H_{β}), 5.65 (dd, $0.4 \text{ H}, J_{3,2} = 10.3, J_{3,4} = 10.3 \text{ Hz}, 3\text{-H}_{\alpha}, 5.40\text{--}5.11 \text{ (m, } 2.6 \text{ H, } 2\text{-H, } 3\text{-H}_{\beta}, 4\text{-H)},$ 4.37-4.08 (m, 2.4 H, $5-H_{\alpha}$, 6_A-H , 6_B-H), 3.90-3.82 (ddd, 0.6 H, $J_{5,4}=9.8$, $J_{5,6A}=0.8$) = 4.5, $J_{5,6\rm B}$ = 2.4 Hz, 5-H $_{\beta}$), 2.11–1.99 (m, 12 H, 4 OAc). C₂₃H₂₄Cl₂F₃NO₁₀ (601). FAB-MS (positive mode, matrix: NBOH with NaI): m/z=601 (M) $^{+}$, 624 (M+Na) $^{+}$.

N-(2-Trifluoromethyl-phenyl)-O-(2,3,4,6-tetra-O-benzyl- α , β -D-glucopyranosyl) Dichloroacetimidate (7Bb)

Following general procedure D, from **5B** (540 mg, 1 mmol) and **4b** (280 mg, 1 mmol) **7Bb** (343 mg, 43%) was obtained as clear oil. R_f (toluene/ethyl acetate, 8:1) 0.64. ¹H NMR (250 MHz, CDCl₃): δ 7.68–7.12 (m, 23 H, 3′-H, 5′-H, 6′-H, 4 Ph), 6.80 (m_c, 0.8 H, 4′-H_{β}), 6.71 (m_c, 0.2 H, 4′-H_{α}), 6.56 (d, 0.2 H, $J_{1,2} = 3.8$ Hz, 1-H_{α}), 6.02 (s, 0.8 H, CHCl_{2 β}), 5.98 (s, 0.2 H, CHCl_{2 α}), 5.82 (d, 0.8 H, $J_{1,2} = 7.9$ Hz, 1-H_{β}), 5.08–3.52 (m, 8 H, 2-H, 3-H, 4-H, 5-H, 6_A-H, 6_B-H, 4 CH₂Ph).

 $C_{43}H_{40}Cl_2F_3NO_6$ (793). FAB-MS (positive mode, matrix: NBOH with NaI): $m/z = 816 \text{ (M+Na)}^+$.

N-(2-Methoxyphenyl)-O-(2,3,4,6-tetra-O-benzyl- α , β -D-glucopyranosyl) Dichloroacetimidate (7Bc)

Following general procedure D, from **5B** (540 mg, 1 mmol) and **4c** (269 mg, 1 mmol) **7Bc** (188 mg, 25%) was obtained as dark-yellow oil. R_f (toluene/ethyl acetate, 16:1) 0.68. ¹H NMR (250 MHz, CDCl₃): δ 7.35–6.63 (m, 24 H, 1-H_{α}, 3'-H, 4'-H, 5'-H, 6'-H, 4 Ph), 6.01 (s, 0.9 H, CHCl_{2 β}), 5.98 (s, 0.1 H, CHCl_{2 α}), 5.91 (d, 0.9 H, $J_{1,2}$ = 8.4 Hz, 1-H_{β}), 5.07–4.42 (m, 8 H, 4 CH₂Ph), 3.70–3.57 (m, 9 H, OCH₃, 2-H, 3-H, 4-H, 5-H, 6_A-H, 6_B-H).

N-Phenyl-trifluoroacetamide (14a)

Following general procedure B, from **1a** (0.93 g, 10 mmol) and **13** (2.32 g, 11 mmol) **14a** (1.61 g, 85%) was obtained as white solid. m.p. 87.7°C; R_f (toluene/ethyl acetate, 20:1) 0.63. 1 H NMR (250 MHz, CDCl₃): δ 8.22–8.01 (1 H, NH), 7.60 (m_c, 2 H, 2-H, 6-H), 7.26 (m_c, 1 H, 4 H). Anal. Calcd for C₈H₆F₃NO (189.13): C, 50.80; H, 3.20; N, 7.41. Found C, 50.61; H, 3.09; N, 7.34.

N-(2-Trifluoromethylphenyl)-trifluoroacetamide (14b)

Following general procedure B, from **1b** (0.63 mL, 5 mmol) and **13** (1.16 g, 5.5 mmol) **14b** (1.23 g, 48%) was obtained as white crystals. m.p. 66° C; R_f (toluene/ethyl acetate, 6:1) 0.82. 1 H NMR (250 MHz, CDCl₃): δ 8.22 (1 H, NH), 8.08 (m_c, 1 H, 6-H), 7.72–7.57 (m, 2 H, 4-H, 5-H), 7.47 (m_c, 1 H, 3-H). Anal. Calcd for C₉H₅F₆NO (257.16): C, 42.04; H, 1.96; N, 5.45. Found C, 42.52; H, 2.39; N, 5.45.

N-(2-Methoxyphenyl)-trifluoroacetamide (14c)

Following general procedure B, from 1c (6.00 g, 48.7 mmol) and 13 (11.6 g, 53.9 mmol) **14c** (1.35 g, 62%) was obtained as white crystals. m.p. 39.9° C; R_f (toluene) 0.58. ¹H NMR (250 MHz, CDCl₃): δ 8.59 (m_c, 1 H, NH), 8.34 (m_c, 1 H, 6-H), 7.26–6.91 (m, 3 H, 3-H, 4-H, 5-H), 3.93 (s, 3 H, OCH₃). Anal. Calcd for C₉H₈F₃NO₂ (219.16): C, 49.32; H, 3.68; N, 6.39. Found C, 49.29; H, 3.68; N, 6.41.

N-(2-Chloro-6-methylphenyl)-trifluoroacetamide (14d)

Following general procedure B, from 1d (0.62 mL, 5 mmol) and 13 (1.16 g, 5.5 mmol) **14d** (493 mg, 41%) was obtained as white crystals. m.p. 71° C; R_f (toluene/ethyl acetate, 2:1) 0.72. ¹H NMR (250 MHz, CDCl₃): δ 7.68 (1 H, NH), 7.33–7.13 (m, 3 H, 3-H, 4-H, 5-H), 2.28 (s, 3 H, CH₃). Anal. Calcd for C₉H₇ClF₃NO (237.69): C, 45.47; H, 2.96; N, 5.89. Found C, 45.65; H, 2.96; N, 5.91.

N-Phenyl-trifluoroacetimidoyl Chloride (15a)

Following general procedure C, from 14a (380 mg, 2 mmol) 15a (277 mg, 66%) was obtained as clear liquid. R_f (toluene) 0.90. ¹H NMR (250 MHz, $CDCl_3$): δ 7.45 (m_c, 2 H, 3-H, 5-H), 7.28 (m_c, 1 H, 4-H), 7.09 (m_c, 2 H, 2-H, 6-H). Anal. Calcd for C₈H₅ClF₃N (207.6): C, 46.28; H, 2.43; N, 6.75. Found C, 46.29; H, 2.43; N, 6.78.

N-(2-Trifluoromethylphenyl)-trifluoroacetimidoyl Chloride (15b)

Following general procedure C, from 14b (514 mg, 2 mmol) 15b (338 mg, 61%) was obtained as clear viscous liquid. R_f (toluene/ethyl acetate, 12:1) 0.91. 1 H NMR (250 MHz, CDCl₃): δ 7.72 (m_c, 1 H, 6-H), 7.58 (m_c, 1 H, 4-H), 7.36 (m_c, 1 H, 5-H), 6.93 (m_c, 1 H, 3-H). Anal. Calcd for C₉H₄ClF₆N (275.58): C, 39.22; H, 1.46; N, 5.08. Found C, 38.94; H, 1.50; N, 5.36.

N-(2-Methoxyphenyl)-trifluoroacetimidoyl Chloride (15c)

Following general procedure C, from 14c (660 mg, 3 mmol) 15c (473 mg, 66%) was obtained as clear liquid. R_f (toluene) 0.83. ¹H NMR (250 MHz, $CDCl_3$): δ 7.28 (m_c, 1 H, 4-H), 7.08–6.96 (m, 3 H, 3-H, 5-H, 6-H), 3.89 (s, 3 H, OCH₃). Anal. Calcd for C₉H₇ClF₃NO (237.61): C, 45.49; H, 2.96; N, 5.89. Found C, 45.68; H, 2.95; N, 6.28.

N-(2-Chloro-6-methylphenyl)-trifluoroacetimidoyl Chloride (15d)

Following general procedure C, from **14d** (474 mg, 2 mmol) **15d** (321 mg, 63%) was obtained as yellowish viscous oil. R_f (toluene/ethyl acetate, 6:1) 0.73. 1H NMR (250 MHz, CDCl₃): δ 7.30–7.04 (m, 3 H, 3-H, 4-H, 5-H), 2.08 (s, 3 H, CH₃). Anal. Calcd for C₉H₆Cl₂N (256.05): C, 42.22; H, 2.36; N, 5.46. Found C, 42.39; H, 2.52; N, 5.59.

N-Phenyl-O-(2,3,4,6-tetra-O-acetyl- α , β -D-glucopyranosyl) Trifluoroacetimidate (16Aa)

Following general procedure D, from **5A** (340 mg, 1 mmol) and **15a** (230 mg, 1.1 mmol) **16Aa** (197 mg, 38%) was obtained as colorless oil. R_f (toluene/ethyl acetate, 4:1) 0.46, α/β 1:3. ^1H NMR (250 MHz, CDCl₃): δ Rotamere. 7.35–6.74 (5 H, 2'-H, 3'-H, 4'-H, 5'-H, 6'-H), 6.50 (0.25 H, 1-H_{\alpha}), 5.75 (0.75 H, 1-H_{\beta}), 6.54 (0.25 H, $J_{3,2} = 10.3$, $J_{3,4} = 10.4$ Hz, 3-H_{\alpha}), 5.78–5.06 (2.75 H, 2-H, 3-H_{\beta}, 4-H), 4.34–4.04 (2.25 H, 5-H_{\alpha}, 6_A-H, 6_B-H), 3.74 (0.75 H, 5-H_{\beta}), 2.11–1.99 (m, 12 H, 4 OAc). C₂₂H₂₄F₃NO₁₀ (518). FAB-MS (positive mode, matrix: NBOH with NaI): m/z = 519 (M)⁺, 542 (M+Na)⁺.

N-(2-Trifluoromethylphenyl)-*O*-(2,3,4,6-tetra-*O*-acetyl- α , β -D-glucopyranosyl) Trifluoroacetimidate (16Ab)

Following general procedure D, from **5A** (340 mg, 1 mmol) and **15b** (300 mg, 1.1 mmol) **16Ab** (211 mg, 36%) was obtained as white foam. R_f (toluene/ethyl acetate, 6:1) 0.31, α/β 1:2. 1H NMR (250 MHz, CDCl₃): δ Rotamere. 7.33–6.77 (m, 4 H, 3'-H, 4'-H, 5'-H, 6'-H), 6.51 (m_c, 0.34 H, 1-H_{\alpha}), 5.76 (m_c, 0.66 H, 1-H_{\beta}), 5.31–5.07 (m, 3 H, 2-H, 3-H, 4-H), 4.33–4.03 (dddd, 2 H, $J_{\rm gem} = 12.4$, $J_{\rm 6A,5} = 4.5$, $J_{\rm 6B,5} = 2.4$ Hz, 6_A-H, 6_B-H), 3.83–3.61 (m, 1 H, 5-H), 2.10–2.03 (m, 12 H, 4 OAc). C₂₃H₂₃F₆NO₁₀ (587.43). FAB-MS (positive mode, matrix: NBOH with NaI): m/z = 587 (M)+, 610 (M+Na)+.

N-(2-Methoxyphenyl)-O-(2,3,4,6-tetra-O-acetyl- α , β -D-glucopyranosyl) Trifluoroacetimidate (16Ac)

Following general procedure D, from **5A** (340 mg, 1 mmol) and **15c** (245 mg, 1 mmol) **16Ac** (197 mg, 36%) was obtained as yellowish oil. R_f (toluene/ethyl acetate, 4:1) 0.36, α/β 1:5. 1H NMR (250 MHz, CDCl₃): δ Rotamere. 6.55 (0.2 H, 1-H $_{\alpha}$), 5.80 (0.8 H, 1-H $_{\beta}$), 5.57–5.08 (m, 3 H, 3 ring protons), 4.34–3.63 (m, 6 H, OCH $_3$, 4 ring protons), 2.11–1.98 (m, 12 H, 4 OAc). $C_{23}H_{26}F_3NO_{11}$ (549).

N-(2-Chloro-6-methylphenyl)-O-(2,3,4,6-tetra-O-acetyl- α , β -D-glucopyranosyl) Trifluoroacetimidate (16Ad)

Following general procedure D, from 5A (340 mg, 1 mmol) and 15d (265 mg, 1 mmol) 16Ad (173 mg, 32%) was obtained as clear oil. R_f (toluene/

ethyl acetate, 6:1) 0.36, α/β 3:2. ¹H NMR (250 MHz, CDCl₃): δ Rotamere. 7.36–7.02 (m, 3 H, 3'-H, 4'-H, 5'-H), 6.38 (0.65 H, 1-H $_{\alpha}$), 6.14 (0.35 H, 1-H $_{\beta}$), 5.49–4.98 (m, 3 H, 2-H, 3-H, 4-H), 4.27–3.97 (m, 3 H, 5-H, 6 $_{\rm A}$ -H, 6 $_{\rm B}$ -H), 2.08 (s, 3 H, CH $_{\rm 3}$), 2.05–1.91 (m, 12 H, 4 OAc). C $_{\rm 23}$ H $_{\rm 25}$ ClF $_{\rm 3}$ NO $_{\rm 10}$ (567.94). FAB-MS (positive mode, matrix: NBOH with NaI): m/z=590 (M+Na)⁺, 740 (M+NaI+Na)⁺.

N-Phenyl-O-(2,3,4,6-tetra-O-benzyl- α , β -D-glucopyranosyl) Trifluoroacetimidate (16Ba)

Following general procedure D, from **5B** (540 mg, 1 mmol) and **15a** (230 mg, 1.1 mmol) **16Ba** (320 mg, 45%) was obtained as clear syrup. R_f (toluene/ethyl acetate, 9:1) 0.69. 1H NMR (250 MHz, CDCl₃): δ 7.43–6.68 (m, 25 H, 2'-H, 3'-H, 4'-H, 5'-H, 6'-H, 4 Ph), 3.80–3.58 (m, 15 H, 1-H, 2-H; 3-H, 4-H, 5-H, 6_A-H, 6_B-H, 4 CH₂Ph). $C_{42}H_{40}F_3NO_6$ (711). FAB-MS (positive mode, matrix: NBOH with NaI): m/z = 711 (M)⁺, 734 (M+Na)⁺.

N-(2-Trifluoromethylphenyl)-O-(2,3,4,6-tetra-O-benzyl- α , β -D-glucopyranosyl) Trifluoroacetimidate (16Bb)

Following general procedure D, from **5B** (540 mg, 1 mmol) and **15b** (300 mg, 1.1 mmol) **16Bb** (405 mg, 52%) was obtained as clear syrup. R_f (toluene/ethyl acetate, 9:1) 0.61, α/β 1:5. 1H NMR (250 MHz, CDCl₃): δ 7.53–6.65 (m, 24 H, 3'-H, 4'-H, 5'-H, 6'-H, 4 Ph), 6.44 (d, 0.17 H, $J_{1,2}=3.8$ Hz, 1-H $_{\alpha}$), 5.67 (d, 0.83 H, $J_{1,2}=8.0$ Hz, 1-H $_{\beta}$), 5.02–4.43 (m, 8 H, 4 CH $_2$ Ph), 4.08–3.39 (m, 6-H, 2-H, 3-H, 4-H, 5-H, 6 $_4$ -H, 6 $_8$ -H). $C_{43}H_{39}F_6NO_6$ (779). FAB-MS (positive mode, matrix: NBOH with NaI): m/z=779 (M)+, 802 (M+Na)+.

N-(2-Methoxphenyl)-O-(2,3,4,6-tetra-O-benzyl- α , β -D-glucopyranosyl) Trifluoroacetimidate (16Bc)

Following general procedure D, from **5B** (540 mg, 1 mmol) and **15c** (245 mg, 1.1 mmol) **16Bc** (310 mg, 42%) was obtained as yellowish oil. R_f (toluene/ethyl acetate, 9:1) 0.60. 1H NMR (250 MHz, CDCl₃): δ 7.40–6.67 (m, 24 H, 3′-H, 4′-H, 5′-H, 6′-H, 4 Ph), 5.01–3.58 (m, 18 H, OCH₃, 1-H, 2-H, 3-H, 4-H, 5-H, 6_A-H, 6_B-H, 4 CH₂Ph). $C_{43}H_{42}F_3NO_7$ (742).

N-(2-Chloro-4-methylphenyl)-O-(2,3,4,6-tetra-O-benzyl- α , β -D-glucopyranosyl) Trifluoroacetimidate (16Bd)

Following general procedure D, from **5B** (540 mg, 1 mmol) and **15d** (265 mg, 1.1 mmol) **16Bd** (743 mg, 98%) was obtained as colorless oil. R_f (toluene/ethyl acetate, 2:1) 0.93. 1H NMR (250 MHz, CDCl₃): δ Rotamere. 7.37–6.85 (m, 23 H, 3'-H, 4'-H, 5'-H, 4 Ph), 5.54 (1 H, 1-H $_{\beta}$), 4.93–4.42 (m,

8 H, 4 CH₂Ph), 3.92–3.11 (m, 6 H, 2-H, 3-H, 4-H, 5-H, 6_A-H, 6_B-H), 2.02 (s, 3 H, CH₃). $C_{43}H_{41}ClF_{3}NO_{6}$ (760.24). EI-MS: m/z = 759 (M–H)⁺.

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