

# Asymmetric Synthesis of Spiro 2-Pyrrolidin-5-ones and 2-Piperidin-6-ones

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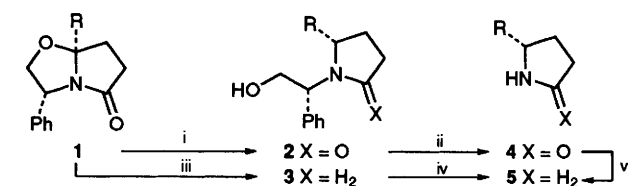
Bicyclic lactams **14–17** are isomerised on treatment with aluminium trichloride in 1,2-dichloroethane to give spiro lactams in high yield and >3 : 1 diastereoselectivity; from the structures of **19a** and **22b** determined by X-ray crystallography, it follows that the indenenes **19** and **21** are formed preferentially with retention of configuration at the spiro carbon atom and the naphthalenes **20** and **22** with inversion.

Bicyclic lactams incorporating  $\beta$ -amino alcohols as chiral auxiliary have been employed in various ways in asymmetric synthesis of tertiary and quaternary carbon centres.<sup>1</sup> The stereoselective conversion of bicyclic lactams **1** into 5-substituted 2-pyrrolidinones and pyrrolidines (Scheme 1)<sup>2</sup> suggested to us a new asymmetric approach to the synthesis of spiro lactams based on *N*-acyliminium ion chemistry.

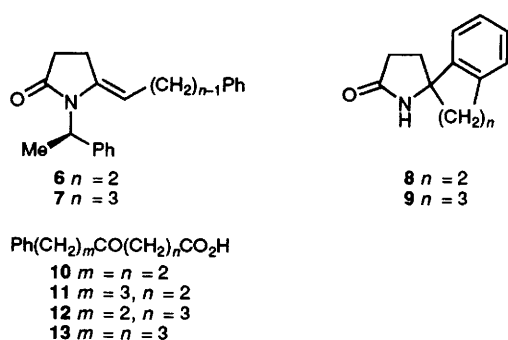
We have already seen *N*-acyliminium ion cyclisations involving an aromatic ring as  $\pi$ -nucleophile attached by a tether of variable length to the iminium carbon atom.<sup>3</sup> But this method applied to the chiral precursors **6** and **7** gave racemic spiro lactams **8** and **9**, respectively, as the major products through loss of the benzylic group. Therefore, we investigated the possibility of spiro cyclisations of fused oxazolidines **14–17** in which the bridgehead substituent  $R = (CH_2)_mPh$  ( $m = 2$  or  $3$ ) provides the  $\pi$ -nucleophile for intramolecular reaction with an *N*-acyliminium ion intermediate.

The 4-oxo acid **10** was heated with (*R*)-phenylglycinol to give a single product **14a**, in which the 3-phenyl and 7a-(3-phenylpropyl) substituents are *cis*, both on the convex face of the 5,5 bicyclic system, in line with the stereochemistry of related compounds.<sup>1,2</sup> The bicyclic lactam **14a** rearranged on treatment with aluminium trichloride to give a mixture of diastereoisomeric spiro lactams **19a,b**; the reaction was optimised with a 3 : 1 mole ratio of  $AlCl_3$  and **14a** in 1,2-dichloroethane at  $-5^\circ C$  to give the results shown in Table 1. The products were separated chromatographically and the stereochemistry assigned by X-ray crystallography of the major diastereoisomer **19a** (Fig. 1),<sup>†</sup> which shows retention of configuration at the spiro carbon centre. This is the same stereochemical result as that observed for reductive ring-opening of the oxazolidine ring in **1** (Scheme 1).<sup>2</sup>

The homologous 4-oxo acid **11** similarly afforded the bicyclic lactam **15a**, which on treatment with aluminium trichloride gave a similar mixture of diastereoisomeric spiro



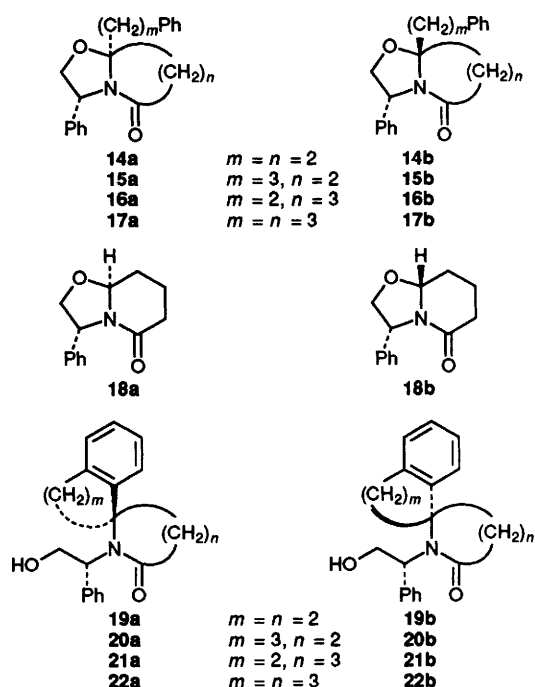
**Scheme 1** Reagents and conditions: i,  $Et_3SiH/TiCl_4$ ; ii,  $Na/NH_3$ ; iii,  $LiAlH_4/AlCl_3$ ; iv,  $HCO_2NH_4/Pd-C$ ; v,  $LiAlH_4$



lactams **20a,b**. However, in this case the minor product is the more polar (lower  $R_f$  value), whereas in the previous case the major product **19a** was more polar than the minor product **19b**. This and other evidence, in particular the crystal structure of another spiro naphthalene **22b** (Fig. 2), leads us to the surprising conclusion that the major product from **15a** is, in fact, **20b** in which the spiro centre has been formed with inversion of configuration at C-7a in structure **15a**.

The 5-oxo acids **12** and **13** reacted with (*R*)-phenylglycinol to give mixtures of diastereoisomeric bicyclic lactams **16a,b** and **17a,b**, respectively. These mixtures (*ca* 84 : 16 ratio from the  $^{13}C$  NMR spectra) were inseparable on a silica column. The major component in each case is the *cis* diastereoisomer, **16a** and **17a**, by analogy with stereochemical assignments to related compounds by Meyers *et al.*<sup>1</sup> However, it is noteworthy that the condensation of (*R*)-phenylglycinol with methyl 5-oxopentanoate gives the opposite stereochemical result, although the product **18b** (25% yield) is equilibrated to a 86 : 14 mixture of **18a,b** on treatment with acid.<sup>4</sup>

Isomerisation of **17a,b** on treatment with aluminium trichloride in 1,2-dichloroethane at room temperature afforded a mixture of spiro lactams **22a,b**, which was separated by



**Table 1** Bicyclic oxazolidines and spiro lactams

Oxo acid	Oxazolidine	Spiro lactams	Ring size <sup>a</sup>	Ratio a : b
<b>10</b>	<b>14a</b> 77%	<b>19a,b</b> 93%	5,5	3.6 : 1
<b>11</b>	<b>15a</b> 54%	<b>20a,b</b> 90%	5,6	1 : 3.9
<b>12</b>	<b>16a,b</b> 72%	<b>21a,b</b> 88%	6,5	3.2 : 1
<b>13</b>	<b>17a,b</b> 72%	<b>22a,b</b> 99%	6,6	1 : 3.0

<sup>a</sup> Ring size is lactam, carbocycle.

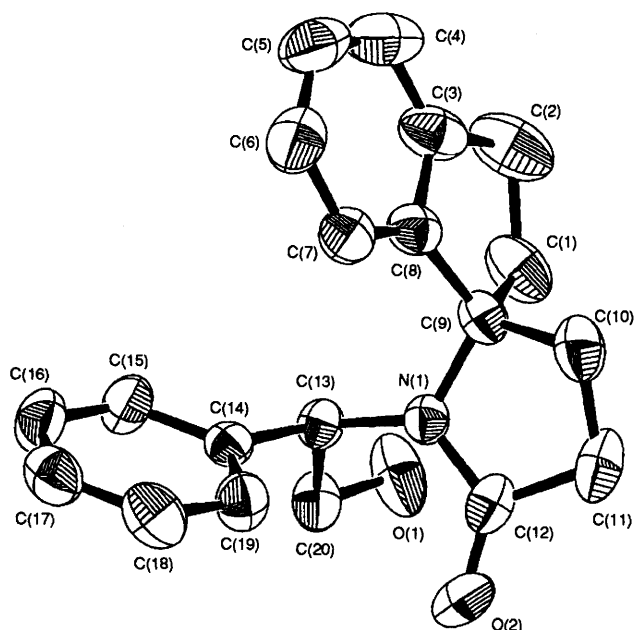


Fig. 1 ORTEP drawing of the structure of compound **19a** with crystallographic numbering scheme (hydrogen atoms omitted)

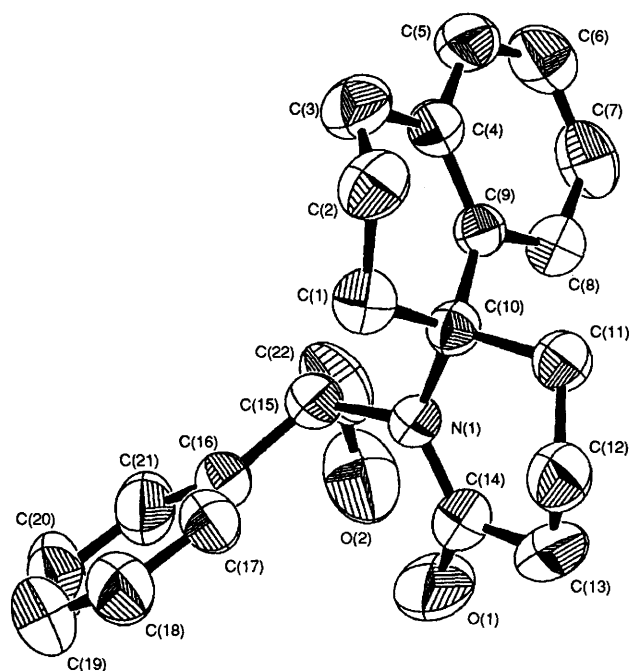


Fig. 2 ORTEP drawing of the structure of compound **22b** with crystallographic numbering scheme (hydrogen atoms omitted)

chromatography. The major diastereoisomer was the less polar component (eluted first) and its structure **22b** confirmed by X-ray crystallography,<sup>†</sup> which shows the spiro stereogenic centre has the (*S*)-configuration (Fig. 2). Analogous spiro lactams **21a,b** were obtained in the same way from **16a,b**, but in this case the major diastereoisomer was the more polar and it is therefore assigned the (*R,R*)-configuration **21a**, as for the

other spiro indene **19a**. These results are summarised in Table 1.

Our results show that it is possible to access spiro lactams with a range of ring sizes by this approach, and to separate diastereoisomeric products **19–22**. The usual reductive methods for debenzoylation are inappropriate for removal of the  $\alpha$ -hydroxymethylbenzyl group from nitrogen in **19–22** because of the likelihood of opening the lactam ring. However, an alternative method of debenzoylation in two steps used in a similar situation by Meyers *et al.* gave **5** (*R* = Ph, 51% yield) from **3** (*R* = Ph).<sup>2</sup> This will allow the further conversion of single diastereoisomers of **19–22** into single enantiomers of the corresponding 1-aza spirans.

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### Footnote

<sup>†</sup> *Crystal Data* for spiro compound **19a**.  $C_{20}H_{21}NO_2$ ,  $M = 307.4$ , monoclinic, space group  $P2_1$ ,  $a = 9.0974(9)$ ,  $b = 9.8532(11)$ ,  $c = 9.8766(10)$  Å,  $\beta = 108.101(8)^\circ$ ,  $U = 841.5(1)$  Å<sup>3</sup>,  $Z = 2$ ,  $D_c = 1.213$  g cm<sup>-3</sup>,  $F(000) = 328$ ,  $\mu(\text{Mo-K}\alpha) = 0.78$  cm<sup>-1</sup>, crystal size  $0.7 \times 0.6 \times 0.6$  mm.

For spiro compound **22b**.  $C_{22}H_{25}NO_2$ ,  $M = 335.4$ , orthorhombic, space group  $P2_12_12_1$ ,  $a = 10.486(3)$ ,  $b = 18.490(5)$ ,  $c = 9.459(3)$  Å,  $U = 1833.9(8)$  Å<sup>3</sup>,  $Z = 4$ ,  $D_c = 1.215$  g cm<sup>-3</sup>,  $F(000) = 720$ ,  $\mu(\text{Mo-K}\alpha) = 0.77$  cm<sup>-1</sup>, crystal size  $1.0 \times 0.6 \times 0.5$  mm.

Intensity data were collected at 293 K on a Rigaku four-circle diffractometer with graphite-monochromated Mo-K $\alpha$  X-radiation,  $\lambda = 0.7107$  Å. Equivalent reflections were merged and only Lorentz and polarisation corrections were applied. The structures were solved by direct methods using SHELX-76 and refined on  $F^2$  using SHELXL. Full-matrix least-squares refinement of 208 parameters for 1577 independent reflections [ $I \geq \sigma(I)$ ] in the range  $2.66 < \theta < 25^\circ$  gave  $R_F = 0.0373$  and  $wR_i = 0.0937$  [ $R_F = 0.0327$  on  $I \geq 2\sigma(I)$  data] for structure **19a**. Similar refinement of 226 parameters for 1858 independent reflections [ $I \geq \sigma(I)$ ] in the range  $3 < \theta < 25^\circ$  gave final values of  $R_F = 0.0738$  and  $wR_i = 0.2184$  [ $R_F = 0.0408$  on  $I \geq 2\sigma(I)$  data] for structure **22b**. The absolute configuration at the spiro centre in each compound [(*R*) in **19a**, (*S*) in **22b**] was established relative to the known absolute configuration of the (*R*)-phenylglycinol moiety. Atomic coordinates, bond lengths and angles, and thermal parameters have been deposited at the Cambridge Crystallographic Data Centre. See Information for Authors, Issue No. 1.

### References

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