

Ruthenium-Catalyzed Tandem Carbene/Alkyne Metathesis/N–H Insertion: Synthesis of Benzofused Six-Membered Azaheterocycles

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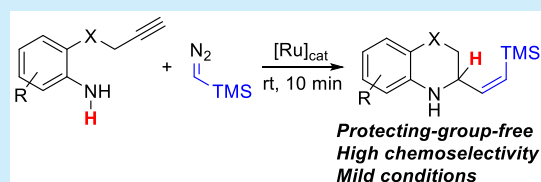


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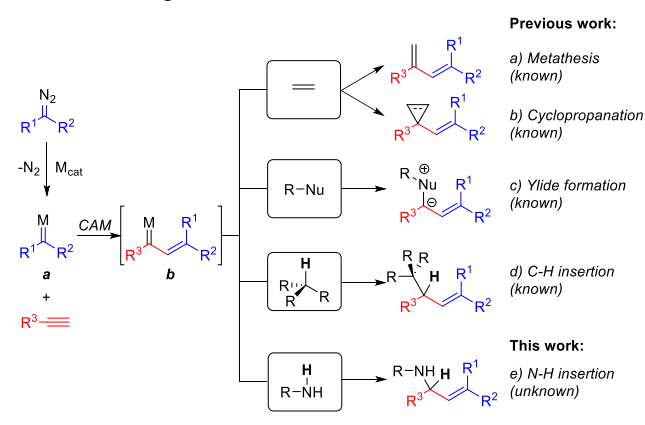
Supporting Information

ABSTRACT: The Cp^{*}RuCl-based catalyst enables expedient access to a variety of benzofused six-membered azaheterocycles from unprotected *o*-alkynylanilines and trimethylsilyldiazomethane through an unprecedented tandem carbene/alkyne metathesis/N–H insertion reaction. The transformation takes place under mild reaction conditions (room temperature, <15 min) and with excellent functional group tolerance. The synthetic utility of the final products and a mechanistic rationale are also discussed.



Tandem processes involving catalytic metal carbenes have proved to be useful strategies for the rapid generation of molecular complexity.¹ In particular, the in situ generation of metal vinyl carbenes through carbene/alkyne metathesis (CAM) represents a versatile route for alkyne bifunctionalization.² These intermediates are known to react with olefins to give dienes³ (Scheme 1a) or vinyl cyclopropa(nes⁴ (Scheme

Scheme 1. Reactivity Pattern of Metal Vinyl Carbenes Formed through CAM



1b) with nucleophiles to afford ylide intermediates⁵ (Scheme 1c) or with C–H bonds to give new C–C bonds⁶ (Scheme 1d). However, as far as we know, a tandem CAM process ending up in a N–H insertion reaction has never been reported (Scheme 1e).

The development of such a tandem process is challenging. The coexistence of two metal carbenes (a and b in Scheme 1) in the reaction media may lead to competitive processes such as dimerizations or unselective N–H insertions. In addition, current methodologies for intramolecular N–H insertions typically require the amine to be protected as an amide,

carbamate, or sulfonamide,^{7,8} thus leading to less atom-economical processes.

We now report our efforts in the development of the first tandem carbene/alkyne metathesis coupled to an intramolecular N–H insertion, leading to unprotected benzofused six-membered azaheterocycles,⁹ which are privileged scaffolds present in a myriad of bioactive compounds and natural products (Figure 1).^{10,11}

o-Alkynylaniline 1a, an unprotected primary aromatic amine, was synthesized and subjected to our previously reported

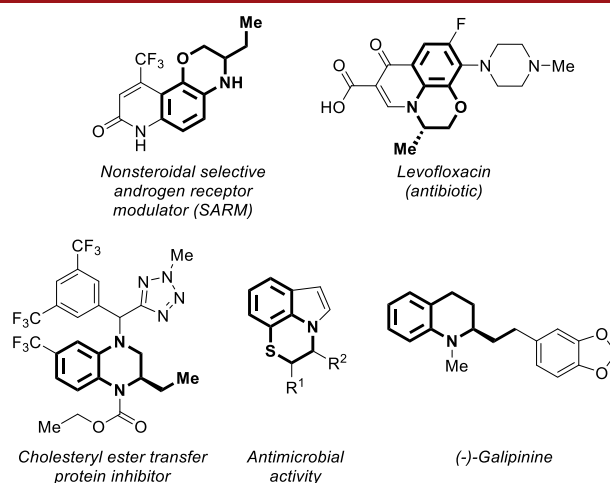
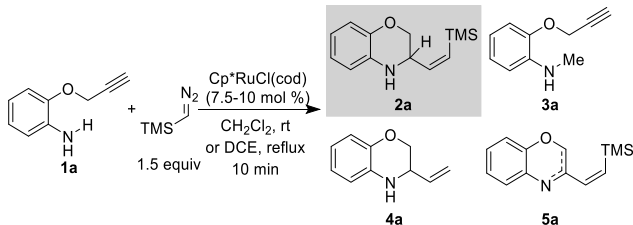


Figure 1. Selected bioactive compounds and natural products.

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conditions for aliphatic secondary amines (Table 1, entry 1).^{5f} Gratifyingly, 3-vinylidihydrobenzoxazine **2a** was selectively

Table 1. Optimization of the Reaction Conditions^a



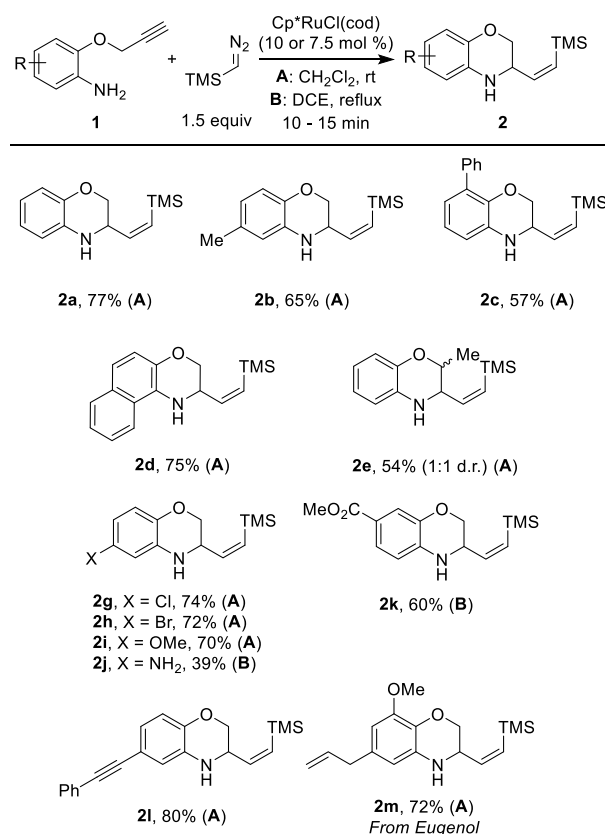
entry	deviation from standard conditions	product/yield (%) ^b
1	none	2a /77
2	5 mol % Rh ₂ (OAc) ₄ instead of Cp*RuCl(cod)	3a /15 ^{c,d}
3	5 mol % Rh ₂ (esp) ₂ instead of Cp*RuCl(cod)	3a /8 ^{c,d}
4	[Cp*Ru(CH ₃ CN) ₃]PF ₆ instead of Cp*RuCl(cod)	3a + 4a /n.d. ^c
5	EtO ₂ CCHN ₂ instead of TMSCHN ₂	complex mixture
6	[Cp*RuCl] ₄ instead of Cp*RuCl(cod)	2a /60 ^c
7	THF/MeOH/iPrOH/CH ₃ CN instead of DCM	2a + 5a /10–31 ^c
8	7.5 mol % of Cp*RuCl(cod)/DCE reflux/ 2 mmol scale	2a /73

^aReaction conditions: **1a** (0.2 mmol), TMSCHN₂ (1.5 equiv), and solvent (0.15 M) with the indicated catalyst at rt. ^bIsolated yields. ^cIncomplete consumption of **1a** was observed. ^dSlow addition of the diazo compound over 1 h.

formed in 77% yield as a single *Z* stereoisomer¹² in <10 min of reaction at room temperature. A direct comparison between the Cp*RuCl(cod) precatalyst and traditional Rh(II) catalysis (Rh₂(OAc)₄, entry 2 and Rh₂(esp)₂, entry 3) highlights the virtues of the half-sandwich ruthenium complex in promoting CAM rather than direct N–H insertion. In fact, the reaction proved to be very sensitive to the electronic nature of the ruthenium precatalyst and the diazo compound as the use of the cationic analog [Cp*Ru(CH₃CN)₃]PF₆ (entry 4) or ethyl diazoacetate (entry 5) gave rise to a mixture of the desilylated product **4a** together with minor amounts of the direct N–H insertion product **3a** and a complex mixture, respectively. The use of the tetranuclear complex [Cp*RuCl]₄ afforded a similar result as Cp*RuCl(cod), but an incomplete consumption of **1a** was observed (entry 6), probably due to a faster deactivation of the catalyst. The nature of the solvent also proved to be crucial as the employment of more polar (protic and aprotic) solvents led to low conversions (entry 7) and the formation of side products of type **5a**. Pleasingly, we discovered that it is possible to scale up the reaction to 2 mmol and diminish the catalyst loading from 10 to 7.5 mol % by using 1,2-dichloroethane as a solvent at reflux (entry 8).

Having established the optimal reaction conditions for the tandem CAM/N–H insertion reaction, we decided to explore the scope and limitations of our methodology. First, O-tethered *o*-alkynylanilines were tested (Scheme 2). The cascade reaction tolerates any substitution pattern on the aromatic ring, affording the corresponding 1,4-benzoxazines **2a–d** from moderate to good yields. Substitution at the propargylic position was also tolerated, albeit benzoxazine **2e** was obtained as a 1:1 mixture of diastereomers in 54% yield.¹³ Remarkably, the reaction proceeded with excellent chemoselectivity in the presence of a wide range of functional groups such as halides (**2g** and **2h**), ethers (**2i**), unprotected anilines (**2j**), esters (**2k**), internal alkynes (**2l**), or terminal olefins

Scheme 2. Scope and Functional Group Tolerance for the Tandem CAM/N–H Insertion of O-Tethered *o*-Alkynylanilines^a



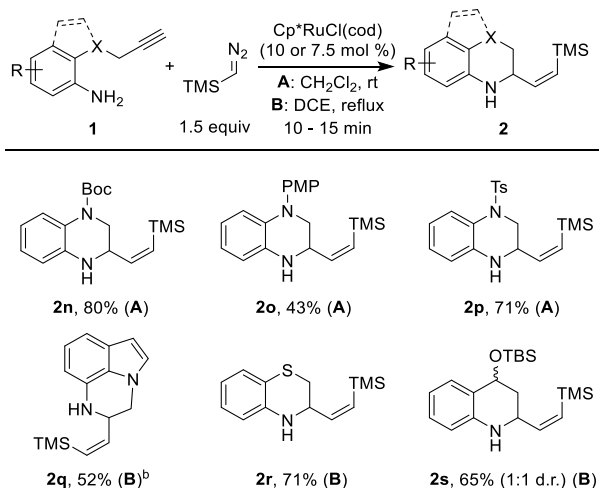
^aConditions: Method A: **1** (0.2 mmol), TMSCHN₂ (1.5 equiv), CH₂Cl₂ (0.15 M), and Cp*RuCl(cod) (10 mol %) at rt for 10–15 min. Method B: Same conditions as method A but using 7.5 mol % of Cp*RuCl(cod) and DCE as a solvent at reflux for 15 min.

(**2m**). Considering the slight excess of TMSCHN₂ used for this transformation, one might expect further evolution of the final products **2** through N–H insertion of the resulting secondary aniline, unselective N–H insertion with the primary aniline **2j**, CAM with the internal alkyne **2l**, or metathesis/cyclopropanation with the terminal olefin **2m**; however, none of these side reactions were detected in the analysis of the crude mixtures.

The extension of the tandem CAM/N–H insertion to the synthesis of other kinds of six-membered heterocycles was subsequently analyzed (Scheme 3). To our delight, the cyclization reaction allowed access to a variety of functionalized tetrahydroquinoxalines (**2n–p**) and indoloquinoxalines (**2q**), dihydrobenzothiazines (**2r**), or tetrahydroquinolines (**2s**) from moderate to good yield. These results further exemplify the excellent functional group tolerance toward carbamates, sulfonamides, heteroaromatic systems, thioethers, or silyl ethers. Curiously, these results are in striking contrast with our previous experience with secondary benzylamines in the tandem CAM/ylide rearrangement, where N-, S-, or C-tethered *o*-alkynylanilines were not tolerated.^{5f}

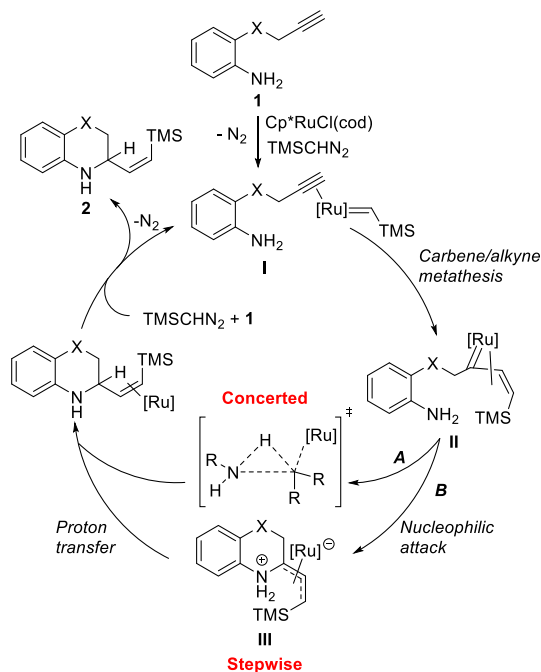
According to precedent literature and the experimental observations, a tentative mechanism was proposed (Scheme 4). The Cp*RuCl(cod) precatalyst would react with the diazo compound to generate a ruthenium carbene that readily coordinates to the *o*-alkynylaniline **1** (I). A chemo-

Scheme 3. Scope and Functional Group Tolerance for the Tandem CAM/N–H Insertion of Carbon- and Heteroatom-Tethered *o*-Alkynylanilines^a



^aConditions: Method A: **1** (0.2 mmol), TMSCHN_2 (1.5 equiv), CH_2Cl_2 (0.15 M), and $\text{Cp}^*\text{RuCl}(\text{cod})$ (10 mol %) at rt for 10–15 min. Method B: The same conditions as method A but using 7.5 mol % of $\text{Cp}^*\text{RuCl}(\text{cod})$ and DCE as the solvent at reflux for 15 min. ^bNo full conversion of *o*-alkynylaniline **1q** was observed.

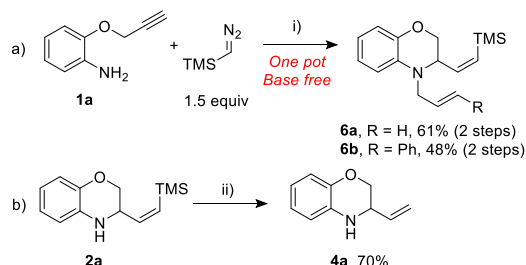
Scheme 4. Mechanistic Hypothesis



stereoselective CAM process would generate vinyl carbene **II** that would then react with the aniline through two alternative routes. In route A, a concerted N–H insertion process would directly give rise to the observed product **2**. In route B, the mild electrophilic ruthenium vinyl carbene would induce a nucleophilic attack by the aniline to give an ylide intermediate **III**, which after a regioselective proton transfer would release **2**. In this stage of our investigations, we were not able to unequivocally determine whether the N–H insertion step occurs in a concerted or stepwise manner.^{14,15}

The presence of a versatile unprotected allylaniline functionality in the cyclized products **2** led us to explore some manipulations to prove their synthetic utility as potential building blocks for organic synthesis (Scheme 5). First, the

Scheme 5. Derivatization of Benzoxazine 2a^a



^aConditions: (i) **1a** (0.2 mmol), TMSCHN_2 (1.5 equiv), $\text{Cp}^*\text{RuCl}(\text{cod})$ (10 mol %) in CH_2Cl_2 (0.15 M) at rt for 10 min, then, the corresponding allyl bromide ($\text{RCH}=\text{CH}-\text{CH}_2\text{Br}$) was added (1.5 equiv) and stirred for 6–12 h. (ii) **2a** (1 mmol), TBAF (1.5 equiv) in THF (0.5 M) at reflux for 15 h.

mild conditions required for the cyclization enabled the one-pot/base-free allylation of the secondary aniline **2a** to afford the corresponding bis-allylaniline **6a** and **6b** in good overall yield. On the contrary, the desilylation of **2a** could be performed to render the terminal olefin **4a** in 70% yield.

To conclude, we have developed the first tandem CAM/N–H insertion reaction to afford unprotected and functionalized benzofused six-membered azaheterocycles. The reaction proceeded under very mild conditions and high chemoselectivity thanks to a fast CAM process catalyzed by a half-sandwich ruthenium complex.

■ ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/acs.orglett.0c00596>.

Experimental procedures including characterization data (PDF)

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Author Contributions

All authors have given approval to the final version of the manuscript.

Notes

The authors declare no competing financial interest.

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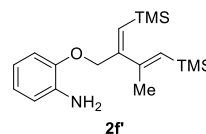
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