Synthetic Studies on Novel 1,4-Dihydro-2-methylthio-4,4,6-trisubstituted Pyrimidine-5-carboxylic Acid Esters and Their Tautomers

Yoshio Nishimura,^{*,a} Yasuko Okamoto,^b Masaya Ikunaka,^a and Yoshihiko Ohyama^a

^a Department of Pharmaceutical Chemistry, Faculty of Pharmacy, Yasuda Women's University; 6–13–1 Yasuhigashi, Asaminami-ku, Hiroshima 731–0153, Japan: and ^b Faculty of Pharmaceutical Sciences, Tokushima Bunri University; Yamashiro-cho, Tokushima 770–8514, Japan.

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A mixture of alkyl 1,4-dihydro-2-methylthio-4,4,6-trisubstituted pyrimidine-5-carboxylate 1 and its tautomeric isomer, alkyl 1,6-dihydro-2-methylthio-4,6,6-trisubstituted pyrimidine-5-carboxylate 2 is synthesized by the Atwal—Biginelli cyclocondensation reaction of S-methylisothiourea hemisulfate salt 3 with 2-(gem-disubstituted)methylene-3-oxoesters 4 that can be accessed by the Lehnert procedure for the Knoevenagel-type condensation. The structures of the tautomeric products of the Atwal–Biginelli cyclocondensation reaction, 1 and 2, which are inseparable from each other, are determined unambiguously by ¹H-NMR spectroscopy at various temperatures and nuclear Overhauser enhancement spectroscopy (NOESY) experiment. Because these dihydropyrimidine products are otherwise inaccessible and thus hitherto unavailable, the synthetic methods established in this study will help to expand the molecular diversity of their related derivatives.

Key words Atwal-Biginelli reaction; 2-(gem-disubstituted methylene)-3-oxoester; tautomer; alkyl 1,4-dihydro-2-methylthio-4,4,6-trisubstituted pyrimidine-5-carboxylate; alkyl 1,6-dihydro-2-methylthio-4,6,6-trisubstituted pyrimidine-5-carboxylate

This article deals with a general approach to synthesizing alkyl 1,4-dihydro-2-methylthio-4,4,6-trisubstituted pyrimidine-5-carboxylate **1** and its tautomeric isomer, alkyl 1,6-dihydro-2-methylthio-4,6,6-trisubstituted pyrimidine-5-carboxylate **2**, which was achieved by applying the Atwal–Biginelli cyclocondensation reaction^{1–3)} of *S*-methylisothiourea hemisulfate salt **3** to 2-(*gem*-disubstituted)methylene-3-oxoesters **4** available by the Lehnert procedure^{4,5)} for the Knoevenagel-type condensation (Fig. 1). This synthetic study was conducted as part of our medicinal chemistry program

to explore alkyl 3,4-dihydro-2-oxo-3,4,4,6-tetrasubstituted pyrimidine-5-carboxylates **5a** and their 2-thioxo congeners **5b** for novel therapeutic agents,⁶⁾ which was inspired by the two track records with the related 1,4-dihydro-4-monoaryl-6-methylpyrimidine derivatives that follow: (*R*)-SQ 32926 **6**,⁷⁾ a calcium channel blocker developed as an orally active anti-hypertensive agent, and monastrol **7**,⁸⁻¹⁰⁾ a possible anti-cancer agent that inhibits mitotic cell division by blocking the activity of kinesin Eg 5, a motor protein causing spindle bipolarity.



Fig. 1. Synthetic Approach to Alkyl 1,4-Dihydro-2-methylthio-4,4,6-trisubstituted Pyrimidine-5-carboxylate 1 and Its Tautomeric Isomer 2 and Structures of Related 2-Oxo and 2-Thioxo Analogs



Chart 2. Typical Atwal-Biginelli Cyclocondensation Reaction to Afford Alkyl 1,4-Dihydro-2-methylthio-4,6-disubstituted Pyrimidine-5-carboxylate 15 and Its Tautomeric Isomer 16

Results and Discussion

One of the ultimate synthetic targets being 3,4-dihydro-4,4,6-trisubstituted-2-thioxopyrimidine **5b** ($R^5=H$), a threecomponent Biginelli reaction was first attempted, in which ethyl acetoacetate (10 1.0 eq) was treated with thiourea 8 (1.2 eq) and acetone 9 (3-10 eq) in dimethylformamide (DMF) at 65 °C for 12 h in the presence of a Brønsted acid (1.0 eq), such as HCl, H₂SO₄, trifluoroacetic acid, 4-toluenesulfonic acid, or methanesulfonic acid (Chart 1).¹¹⁾ However, no desired ethyl 3,4-dihydro-4,4,6-trimethyl-2-thioxopyrimidine-5-carboxylate 11 was produced, and the starting material 10 was recovered intact. Furthermore, no such threecomponent reaction took place to afford ethyl 1,4-dihydro-4,6,6-trimethyl-2-methylthiopyrimidine-5-carboxylate 12 or its 1,6-dihydro tautomer 13 when 9 (3-10 eq) and 10 (1.0 eq) were exposed to 3 (1.2 eq) instead of 8 in DMF in the presence of NaHCO₃ (4.0 eq) at 65 °C for 12 h.¹²⁻¹⁴⁾

In contrast, when ethyl 2-benzylidene-3-oxobutanoate 14 [E/Z (1:3.0)],¹⁵⁾ a typical 2-(monosubstituted)methylene-3-oxoester prepared under the conventional Knoevenagel reaction conditions, was subjected to Atwal–Biginelli conditions

[3 (1.2 eq), NaHCO₃ (4.0 eq), DMF, 60 °C for 5 h], the cyclocondensation reaction proceeded uneventfully to provide ethyl 1,4-dihydro-2-methylthio-4,6-disubstituted pyrimidine-5-carboxylate 15 and its tautomeric isomer 16 as an inseparable mixture [15/16 (2.2:1)] in a combined yield of 67% (Chart 2).16,17) Structural assignments for 15 and 16 were made by the following nuclear Overhauser enhancement spectroscopy (NOESY) experiment. With the major component 15, its 6-methyl protons exhibited a significant nuclear Overhauser effect (NOE) when the 1-NH proton was irradiated; hence, its structure was determined to be 1,4-dihydropyrimidine 15 as depicted in Chart 2. Irradiation of the 1-NH proton in the minor component 16 caused the NOE on the 6-proton, which led to its structure being determined to be 1,6-dihydropyrimidine 16, a tautomeric isomer of 15 (Chart 2): for detail, see Experimental. Hence, it was envisioned that when prefabricated ethyl 3-oxo-2-(2-propylidene)-butanoate 17 was allowed to react with 8 and 3, the cyclocondensation in question should proceed to give 11 and a mixture of 12 and 13, respectively (Chart 1).

To test the above-mentioned proposition, 17 was required



Chart 3. Synthesis of Ethyl 1,4-Dihydro-4,4,6-trimethyl-2-methylthiopyrimidine-5-carboxylate 12 and Its 1,6-Dihydro Tautomer 13

in sufficient quantity. To our disappointment, it was obtained in less than 3% yield when ethyl acetoacetate **10** (1.0 eq) and acetone **9** (2.0 eq) were subjected to the conventional Knoevenagel reaction conditions [piperidine (0.25 eq), molecular sieves 4A (34% w/w based on **10**), toluene, 50 °C, 24 h] (Chart 3).¹⁸⁾ However, literature search in parallel with failed experimentation led to identifying the Lehnert's report,^{19–24)} which was demonstrated to be the most effective means of preparing **17** by the Knoevenagel-type condensation: when **10** (1.0 eq) and **9** (2.0 eq) were treated with TiCl₄ (1.0 eq) in the presence of pyridine (4.0 eq) in tetrahydrofuran (THF) at room temperature for 12 h, **17** was obtained in 44% yield (Chart 3).^{4,5)}

Having secured 17 in quantity, we turned our attention towards the synthesis of 11 (Chart 3). When 17 was treated with thiourea 8 in the presence of HCl according to the typical procedures of the three-component Biginelli reaction, the expected 2-thione product 11 was obtained in a low yield of 1%. In contrast, the Atwal–Biginelli cyclocondensation reaction proceeded successfully with 17 and S-methylisothiourea hemisulfate salt 3 to afford 1,4-dihydro-4,4,6-trimethyl-2methylthiopyrimidine-5-carboxylate 12 and its 1,6-dihydro tautomer 13 as an inseparable mixture [12/13 (1.5:1)] in a combined yield of 89% (Chart 3). The structures of **12** and **13** were determined to be as depicted in Chart 3 by the NOESY experiment similar to that conducted with the tautomeric mixture of **15** and **16**: for detail, see Experimental.

To see the concentration dependence of the tautomeric ratio of **12** to **13**, their combined concentration was changed from 1.1×10^{-2} to 8.8×10^{-2} M. The tautomeric ratio determined by ¹H-NMR at 25 °C remained unaffected at 1.5:1 throughout the concentrations varied. In contrast, the tautomeric ratio changed in a temperature-dependent manner. When the ¹H-NMR spectra were measured at temperatures increased from 25 to 105 °C, the ratio decreased from 1.5:1 to 1.3:1. When the mixture was cooled to 25 °C, the ratio was restored to the original value. Although the change in the composition was not large over the range of temperatures tested, it was such the temperature-dependent interconvertibility that corroborated the tautomeric relationship between **12** and **13**.

Now that the Knoevenagel-type condensation between the β -oxoester **10** and the ketone **9** had provided **17** in good yield under the Lehnert conditions,^{4,5)} we chose to apply those conditions to the synthesis of a range of 2-(*gem*-disubstituted)-methylene-3-oxoesters **4**. The Lehnert's procedures worked

Table 1. Synthesis of 2-(gem-Disubstituted)methylene-3-oxoesters 4 under Lehnert Conditions for the Knoevenagel-Type Reaction^a)

	R ² R ¹ O + 18 (2.0 equiv)	O O R ³ 19 (1.0 equiv)	TiCl₄ (1.0 equiv) pyridine (4.0 equiv) THF, rt, 12 h	R ¹ ∕ → C	² O OR ⁴ R ³ 4	
Entry		R^1	R ²	R ³	\mathbb{R}^4	Yield (%)
1	a	Me	Me	Me	Me	47
2	b	Me	Me	Me	CH_2Ph	37
3	c	-§-CH ₂ (CH ₂))₃CH₂-ѯ-	Me	Et	17
$4^{b,c)}$	d	Me	Me(CH ₂) ₄	Me	Et	16
$5^{b,c,d)}$	e	Me(CH ₂) ₂	Me(CH ₂) ₂	Me	Et	11
$6^{b,d,e)}$	f	Me	Me	Ph	Et	15
$7^{b,c,d)}$	g	Ph	Et	Me	Et	7
$8^{b,d,f)}$	h	er troit	ž.	Me	Et	7

a) General conditions: a mixture of **18** (80 mmol), **19** (40 mmol), TiCl₄ (40 mmol) and pyridine (160 mmol) in THF (40 ml) was stirred at room temperature for 12 h under an atmosphere of argon unless otherwise specified. b) For 24 h. c) **18** (1.0 eq) was used. d) TiCl₄ (2.0 eq) was used. e) **18** (3.0 eq) and pyridine (6.0 eq) were used at 40 °C. f) **18** (1.0 eq), **19** (1.5 eq), TiCl₄ (3.0 eq) and pyridine (6.0 eq) were used.

for the Knoevenagel-type condensation of the aliphatic ketones **18a**—**f** and the aromatic ketones **18g** and **h** to afford the expected products **4a**—**f**, as summarized in Table 1.

Lehnert reported that ketones could undergo the Knoevenagel-type condensation with dialkyl malonate in the presence of $TiCl_4$ and pyridine,⁴⁾ with the use of 3-oxoester being limited to the condensation reaction with aldehydes.⁵⁾ Thus, the experimental results in Table 1 are the first successful expansion of the Lehnert procedures to the Knoevenagel-type condensation between 3-oxoesters and ketones.

With a variety of 2-(*gem*-disubstituted)methylene-3-oxoesters 4 (17, 4a—h) in hand, the stage was set for assembling 1,4-dihydro-2-methylthio-4,4,6-trisubstituted pyrimidine 1 and its 1,6-dihydro tautomeric isomer 2, both of which had remained otherwise inaccessible. With all these substrates, 17 and 4a—h, the Atwal–Biginelli cyclocondensation reaction [3 (1.2 eq), NaHCO₃ (4.0 eq), DMF, 65 °C, 12 h)] proceeded uneventfully to give 1,4-dihydro-2-methylthio pyrimidine 1 and its 1,6-dihydro tautomer 2 in a fair to good yield, as listed in Table 2.

What is worth making comments with respect to the tabulated results are as follows. In addition to ethyl ester 17 (Chart 3), the methyl and benzyl esters 4a and 4b, respectively, were tolerated in the reaction, and the corresponding products 1a/2a and 1b/2b could be obtained without incident (entries 1, 2). Their alkylidene substituents being sterically demanding, 4c and 4e underwent olefin isomerization towards deconjunction under the basic conditions applied so that steric congestion can be relieved; however, when 3 was used in excess amounts, the cyclocondensation could compete with the isomerization to afford the respective products 1c/2c and 1e/2e in acceptable yields (entries 3, 5). When starting from ethyl 2-isopropylidene-3-phenyl-3-oxopropanoate 4f, prepared from ethyl benzoylacetate 19f, a phenyl group could be installed at the 6-position of 1f/2f although 3 needed to be used in 3.0 eq to gain an acceptable yield (entry 6). A phenyl group could also be accommodated at the 4-position, as illustrated by 1g/2g (entry 7). Furthermore, a fluorenylidene group could be appended to the dihydropyrimidine skeleton, as shown by 1h/2h (entry 8).

Conclusion

In summary, it was demonstrated that alkyl 1,4-dihydro-2methylthio-4,4,6-trisubstituted pyrimidine-5-carboxylate 1 (12, 1a—h) and its 1,6-dihydro tautomer 2 (13, 2a—h) could be assembled by applying the Atwal–Biginelli cyclocondensation reaction to 2-(*gem*-disubstituted)methylene-3-oxoesters 4 (17, 4a—h), which, in turn, was prepared from ketones, 9 and 18a—h, and 3-oxoesters, 10 and 19a—h, by resorting to the Lehnert conditions for the Knoevenagel-type condensation. In view of the fact that all the dihydropyrimidine derivatives reported in this article are new in spite of their existence as inseparable tautomeric mixtures, the synthetic procedures developed in our laboratory should help to expand the dihydropyrimidine-based molecular diversity, which would impact the drug discovery program.⁶

Experimental

All melting points were determined with an AS ONE melting point apparatus ATM-02 without correction. IR spectra were measured on a JASCO FT/IR-6100. ¹H-NMR spectra were recorded on a Bruker AVANCETM III 600 (600 MHz) with tetramethylsilane (0 ppm) or dimethylsulfoxide (2.49 ppm) as an internal standard. ¹³C-NMR spectra were recorded on a Bruker

	NH ∭ NH₂ ∙0.5 H₂SO₄	+ $R^2 \xrightarrow{0} 0$ 0 R^3	R ² O NaHCO ₃ (4.0 equiv) DMF 65 °C, 12 h		2 O OR ⁴ + R ³ MeS	$ \begin{array}{c} $	
	3 (1.2 equiv)	4 (1.0 equiv	()	1		2	
Entry		R^1	R ²	R ³	R^4	$\operatorname{Yield}^{b}(\%)$	1/2 ^{c)}
1	a	Me	Me	Me	Me	84	1.5:1
2	b	Me	Me	Me	CH_2Ph	68	1.6:1
3 ^{<i>d</i>})	с	-ફૂ–CH ₂ (Cł	H₂)₃CH₂−ξ−	Me	Et	63	8.4:1
4 ^{<i>e</i>)}	d	Me	Me(CH ₂) ₄	Me	Et	80	1.7:1
5 ^{<i>d</i>})	e	Me(CH ₂) ₂	Me(CH ₂) ₂	Me	Et	63	1.8:1
6 ^{<i>d</i>})	f	Me	Me	Ph	Et	73	1:3.7
7 ^{.f)}	g	Ph	Et	Me	Et	63	3.0:1
8	h	his	rine -	Me	Et	40	6.0:1

Table 2. Synthesis of Alkyl 1,4-Dihydro-2-methylthio-4,4,6-trisubstituted Pyrimidine-5-carboxylate 1 and Its 1,6-Dihydro Tautomer 2 under Atwal-Biginelli Conditions^a)

a) General conditions: a mixture of **3** (0.6 mmol), **4** (0.5 mmol), NaHCO₃ (2.0 mmol) and DMF (1.0 ml) was stirred and heated at 65 °C for 12 h under an atmosphere of argon unless otherwise specified. *b*) A combined yield of **1** and **2**. *c*) Determined by ¹H-NMR spectroscopy featuring NOE experiment. *d*) **3** (3.0 eq) was used. *e*) With **4d** was used a (2.5 : 1) mixture of the *E/Z* isomers. *f*) With **4g** was used a single geometric isomer although its configuration was unidentified.



Fig. 2. NOE and HSQC Observed with 15 and 16

AVANCETM III 600 (150 MHz) with chloroform (77.0 ppm) or dimethylsulfoxide (39.7 ppm) as an internal standard. Mass spectra were recorded on a JEOL JMS-700. High-resolution mass spectroscopy (HRMS) was performed using a JEOL JMS-700. Column chromatography was performed on silica gel 60 (nacalai tesque, 70–230 mesh) using the indicated solvents. TLC was performed using pre-coated silica gel 60 F₂₅₄ plates (Merck KGaA) using the indicated solvents.

Ethyl 1,4-Dihydro-6-methyl-2-methylthio-4-phenyl Pyrimidine-5-carboxylate (15) and Ethyl 1,6-Dihydro-4-methyl-2-methylthio-6-phenyl Pyrimidine-5-carboxylate (16) Under an atmosphere of argon, a mixture of S-methylisothiourea hemisulfate (3; 84 mg, 0.6 mmol), 14 [109 mg, 0.5 mmol, E/Z (1:3.0)],¹⁵⁾ NaHCO₃ (168 mg, 2.0 mmol), and dry DMF (1.0 ml) was heated at 60 °C for 5 h. To the reaction mixture was added EtOAc (20 ml) followed by water (10 ml), and the organic layer was separated. The aqueous layer was extracted with EtOAc (20 ml×2), and the combined organic layer and extracts were washed with water, brine, dried over Na₂SO₄, and concentrated under reduced pressure. The residue was purified by flash column chromatography [n-hexane-EtOAc (2:1)] to give 15 and 16 as an inseparable mixture (97 mg, 0.33 mmol, 67%) in a ratio of 2.2:1 in favor of 15. Colorless crystals. Melting point (mp) 168-171 °C (chloroform). ¹H-NMR [dimethyl sulfoxide (DMSO)- d_6] δ : 1.10 (0.94H, t, J= 7.2 Hz, 16), 1.11 (2.06H, t, J=7.2 Hz, 15), 2.21 (2.06H, s, 15), 2.26 (0.94H, s, 16), 2.27 (2.06H, s, 15), 2.38 (0.94H, s, 16), 3.94-4.05 (2H, m, 15+16), 5.24 (0.31H, d, J=3.0 Hz, 16), 5.50 (0.69H, s, 15), 7.15-7.35 (5H, m, 15+16), 9.06 (0.31H, d, J=3.0 Hz, 16), 9.58 (0.69H, s, 15). Structural as-

signment was made unambiguously by NOESY experiment: With the major component, the significant NOE was observed between 1-NH proton (δ 9.58) and 6-methyl protons (δ 2.21) and as such, its structure was determined to be 15 (Fig. 2). With the minor component, the significant NOE was observed between 1-NH proton (δ 9.06) and 6-proton (δ 5.24) and as such, its structure was determined to be 16 (Fig. 2). ¹³C-NMR (150 MHz, DMSO d_6) δ : 12.7 (unresolved), 14.3 (unresolved), 17.7, 23.3, 53.3, 59.0, 59.2, 59.3, 98.4, 103.4, 126.4, 126.7, 126.8, 127.7, 128.3, 128.6, 145.0, 145.9, 146.2, 150.6, 155.2, 160.9, 166.36, 166.43. The unresolved signals, δ 12.7 and δ 14.3, were further analyzed by heteronuclear single-quantum coherence (HSQC) experiment, which revealed that the signal at $\delta_{\rm C}$ 12.7 was due to the S-CH₃ of 15 and 16 because of its correlation to the signals at $\delta_{\rm H}$ 2.27 (s, 15) and $\delta_{\rm H}$ 2.38 (s, 16) and that $\delta_{\rm C}$ 14.3 was due to the OCH₂CH₃ of 15 and 16 because of its correlation to attached to the signals $\delta_{\rm H}$ 1.11 (t, 15) and $\delta_{\rm H}$ 1.10 (t, 16) (Fig. 2). IR (KBr) cm⁻¹: 3320, 1657, 1470, 1281, 1155, 1109. Electron ionization (EI)-MS *m/z*: 290.1083 (Calcd for C₁₅H₁₈N₂O₂S: 290.1089). MS m/z: 290 (M⁺), 261, 213, 185.

Ethyl 2-Acetyl-3-methylbut-2-enoate (17) According to the procedures reported by Lernert,^{4,5)} a solution of ethyl acetoacetate (**10**; 5.2 g, 40 mmol), acetone (**9**; 5.8 ml, 80 mmol), and pyridine (12.6 g, 160 mmol) in dry THF (15 ml) was added to an ice-cooled solution of TiCl₄ (4.4 ml, *d* 1.73, 40 mmol) in dry THF (25 ml) at 0 °C under an atmosphere of argon. The reaction mixture was stirred at room temperature for 12 h, and EtOAc (150 ml) and water (30 ml) were added. The organic layer was separated, and the aqueous layer was extracted with EtOAc (50 ml×2). The combined or-



Fig. 3. NOE and HSQC Observed with 12 and 13

ganic layer and extracts were washed with saturated NaHCO₃ aqueous solution, water, brine, dried over Na₂SO₄, and concentrated under reduced pressure. The residue was purified by flash column chromatography [*n*-hexane–EtOAc (20:1)] to give **17** (3.0 g, 17.6 mmol, 44%), the ¹H-NMR spectral data of which were identical to those reported ones.²⁵

2-(gem-Disubstituted)methylene-3-oxoester (4a—h) The 2-(gem-disubstituted)methylene-3-oxoesters **4a—h**, were all assembled under the conditions similar to those described for the synthesis of **17**,^{4,5)} and below recorded are their isolated yield and their physicochemical and spectral data while the spectral data of **4c** (yield: 17%) and **4f** (yield: 15%) were identical to those reported ones.²⁵⁾ Analysis and identification of the unresolved signal in the ¹³C-NMR spectrum of **4h** were made based on HSQC experiment.

Methyl 2-Acetyl-3-methylbut-2-enoate (4a): Colorless oil. Yield: 47%. ¹H-NMR (CDCl₃) δ: 1.95 (3H, s), 2.11 (3H, s), 2.29 (3H, s), 3.77 (3H, s). ¹³C-NMR (CDCl₃) δ: 22.9, 23.2, 30.6, 51.7, 131.8, 153.6, 166.0, 200.6. IR (neat) cm⁻¹: 2953, 1727, 1698, 1632, 1233. Chemical ionization (CI)-MS *m/z*: 156.1764 (Calcd for C₈H₁₂O₃: 156.0787). MS *m/z*: 156 (M⁺), 125.

Benzyl 2-Acetyl-3-methylbut-2-enoate (4b): Colorless oil. Yield: 37%. ¹H-NMR (CDCl₃) δ : 1.95 (3H, s), 2.10 (3H, s), 2.23 (3H, s), 5.22 (2H, s), 7.30—7.42 (5H, m). ¹³C-NMR (CDCl₃) δ : 23.0, 23.3, 30.7, 66.5, 128.2, 128.3, 128.5, 131.8, 135.3, 154.0, 165.4, 200.3. IR (neat) cm⁻¹: 1723, 1698, 1224, 1198. CI-MS *m/z*: 233.1189 (Calcd for C₁₄H₁₇O₃: 233.1177). MS *m/z*: 233 (M⁺+H), 125, 91.

Ethyl 2-Acetyl-3-methyloct-2-enoate (4d): Colorless oil. Yield: 16%. An inseparable (2.5 : 1) mixture of the *E/Z* isomers with theie structural assignment being impossible by the ordinary spectroscopic methods. ¹H-NMR (CDCl₃) δ : 0.89 (0.86H, t, *J*=7.2 Hz), 0.90 (2.14H, t, *J*=7.2 Hz), 1.25—1.37 (7H, m), 1.45—1.53 (2H, m), 1.95 (2.14H, s), 2.09 (0.86H, s), 2.18 (0.57H, t, *J*=7.8 Hz), 2.28 (2.14H, s), 2.29 (0.86H, s), 2.38 (1.43H, t, *J*=7.8 Hz), 4.23 (0.57H, q, *J*=7.2 Hz), 4.24 (1.43H, q, *J*=7.2 Hz). ¹³C-NMR (CDCl₃) δ : 13.8, 13.9, 14.0, 20.5, 21.1, 22.3, 22.4, 27.65, 27.67, 30.5, 30.9, 31.75, 31.83, 36.5, 36.8, 60.6, 60.7, 131.86, 131.89, 157.1, 165.6, 166.0, 2000, 200.7. IR (neat) cm⁻¹: 2958, 2933, 1725, 1703, 1626, 1227. EI-MS *m/z*: 226.1578 (Calcd for C₁₃H₂₉O₃: 226.1569). MS *m/z*: 226 (M⁺), 151, 43.

Ethyl 2-Acetyl-3-propylhex-2-enoate (4e): Colorless oil. Yield: 11%. ¹H-NMR (CDCl₃) δ : 0.94 (3H, t, *J*=7.2 Hz), 0.96 (3H, t, *J*=7.2 Hz), 1.30 (3H, t, *J*=7.2 Hz), 1.45—1.55 (6H, m), 2.17 (2H, t, *J*=7.8 Hz), 2.28 (3H, s), 2.38 (2H, t, *J*=7.8 Hz), 4.23 (2H, q, *J*=7.2 Hz). ¹³C-NMR (CDCl₃) δ : 14.1, 14.3, 14.4, 21.86, 21.93, 30.9, 35.7, 36.4, 60.7, 132.0, 160.8, 165.8, 200.5. IR (neat) cm⁻¹: 2964, 1724, 1698, 1615, 1205. CI-MS *m/z*: 227.1638 (Calcd for C₁₃H₂₃O₃: 227.1647). MS *m/z*: 227 (M⁺+H), 185, 153.

Ethyl 2-Acetyl-3-phenylpent-2-enoate (4g): Colorless oil. Yield: 7%. A single geometric isomer with its configuration being unidentified. ¹H-NMR (CDCl₃) δ: 1.01 (3H, t, *J*=7.2 Hz), 1.32 (3H, t, *J*=7.2 Hz), 1.83 (3H, s), 2.73 (2H, q, *J*=7.2 Hz), 4.28 (2H, t, *J*=7.2 Hz), 7.17—7.21 (2H, m), 7.35—7.40 (3H, m). ¹³C-NMR (CDCl₃) δ: 12.4, 14.0, 29.5, 30.8, 61.0, 127.8, 128.5, 128.7, 133.7, 139.2, 157.3, 165.5, 201.5. IR (neat) cm⁻¹: 2979, 1723, 1703, 1233, 1204. EI-MS *m/z*: 246.1255 (Calcd for C₁₅H₁₈O₃: 246.1256). MS *m/z*: 246 (M⁺), 200, 134, 105, 77.

Ethyl 2-Fluoren-9-ylidene-3-oxobutanoate (4h): Yellow crystals. Yield: 7%. mp 62—63 °C (*n*-hexane). ¹H-NMR (CDCl₃) δ: 1.40 (3H, t, *J*=7.2 Hz), 2.62 (3H, s), 4.44 (2H, q, *J*=7.2 Hz), 7.19 (1H, dt, *J*=1.2, 7.8 Hz), 7.23 (1H, dt, *J*=1.2, 7.8 Hz), 7.38 (1H, dt, *J*=1.2, 7.8 Hz), 7.39 (1H, dt, *J*=1.2, 7.8 Hz), 7.49 (1H, d, *J*=7.8 Hz), 7.631 (1H, d, *J*=7.8 Hz), 7.632 (1H, d, *J*=7.8 Hz), 7.83 (1H, d, *J*=7.8 Hz). ¹³C-NMR (CDCl₃) δ: 14.0, 31.1, 62.2, 119.7, 119.9, 125.4, 125.7, 127.6, 127.7, 130.6 (unresolved), 131.8, 135.45, 135.50, 140.1, 141.4, 141.5, 165.2, 201.2. The unresolved signal, δ 1306 was further analyzed by HSQC experiment, which revealed that its correlation to the signals at $\delta_{\rm H}$ 7.38 (dt) and $\delta_{\rm H}$ 7.39 (dt). IR (KBr) cm⁻¹: 1721, 1688, 1584, 1226, 1210. EI-MS *m/z*: 292.1092 (Calcd for C₁₉H₁₆O₃: 292.1100). MS *m/z*: 292 (M⁺), 263, 205, 176.



Ethyl 1,4-Dihydro-4,4,6-trimethyl-2-methylthio Pyrimidine-5-carboxylate (12) and Ethyl 1,6-Dihydro-4,6,6-trimethyl-2-methylthio Pyrimidine-5-carboxylate (13) Under an atmosphere of argon, a mixture of S-methylisothiourea hemisulfate (3; 84 mg, 0.6 mmol), 17 (85 mg, 0.5 mmol), and NaHCO $_3$ (168 mg, 2.0 mmol) in dry DMF (1.0 ml) was heated at 65 °C for 12 h. To the reaction mixture was added EtOAc (20 ml) followed by water (10 ml), and the organic layer was separated. The aqueous layer was extracted with EtOAc ($20 \text{ ml} \times 2$), and the combined organic layer and extracts were washed with water, brine, dried over Na2SO4, and concentrated under reduced pressure. The residue was purified by flash column chromatography [n-hexane-EtOAc (4:1)] to give 12 and 13 as an inseparable mixture (108 mg, 0.45 mmol, 89%) in a ratio of 1.5:1 in favor of 12. Colorless crystals. mp 93—94 °C (n-hexane–EtOAc). ¹H-NMR (DMSO-d₆) δ: 1.18 (1.8H, t, J=7.2 Hz, 12), 1.19 (1.2H, t, J=7.2 Hz, 13), 1.26 (3.6H, s, 12), 1.28 (2.4H, s, 13), 1.90 (1.8H, s, 12), 1.94 (1.2H, s, 13), 2.24 (1.8H, s, 12), 2.32 (1.2H, s, 13), 4.05 (1.2H, q, J=7.2 Hz, 12), 4.07 (0.8H, q, J=7.2 Hz, 13), 8.07 (0.4H, s, 13), 9.13 (0.6H, s, 12). Structural assignment was made unambiguously by NOESY experiment: With the major component, the significant NOE was observed between 1-NH proton (δ 9.13) and 6-methyl protons (δ 1.90) and as such, its structure was determined to be 12 (Fig. 3). With the minor component, the significant NOE was observed between 1-NH (δ 8.07) and 6,6-dimethyl protons (δ 1.28) and as such, its structure was determined to be 13 (Fig. 3). ¹³C-NMR (DMSO- d_6) δ : 12.4, 12.5, 14.3 (unresolved), 17.8, 23.2, 29.1, 31.0, 53.3, 55.8, 59.27, 59.33, 105.0, 110.2, 141.7, 147.4, 150.2, 159.7, 167.0, 167.2. The unresolved signal, δ 14.3, was further analyzed by HSQC experiment, which revealed that the signal at δ_{c} 14.3 was due to the OCH₂CH₂ of **12** and **13** because of its correlation to the signals at $\delta_{\rm H}$ 1.18 (t, 12) and $\delta_{\rm H}$ 1.19 (S, 13) (Fig. 3). IR (KBr) cm⁻¹: 3154, 2929, 1698, 1643, 1609, 1485, 1311, 1171. EI-MS m/z: 242.1088 (Calcd for C₁₁H₁₈N₂O₂S: 242.1089). MS *m/z*: 242 (M⁺), 227, 199.

1,4-Dihydro-2-methylthio-4,4,6-trisubstituted Pyrimidine-5-carboxylates and 1,4-Dihydro-2-methylthio-4,6,6-trisubstituted Pyrimidine-5carboxylates (1a—h/2a—h) A tautomeric mixture of 1a—h/2a—h were synthesized according to the same procedures as described for the synthesis of the mixture of 12 and 13; below listed are their isolated yield and their physicochemical and spectral data for 1a—h/2a—h. Analysis and identification of the unresolved signals in the ¹³C-NMR spectra of 1a—b/2a—b, 1d—e/2d—e and 1g/2g were made based on HSQC experiment.

Methyl 1,4-Dihydro-4,4,6-trimethyl-2-methylthio Pyrimidine-5-carboxylate (1a) and Methyl 1,6-Dihydro-4,6,6-trimethyl-2-methylthio Pyrimidine-5-carboxylate (2a): Colorless crystals. Yield: 84%. 1a/2a= 1.5:1. mp 85—86 °C (*n*-hexane). ¹H-NMR (DMSO- d_6) δ : 1.26 (3.6H, s, 1a), 1.28 (2.4H, s, 2a), 1.90 (1.8H, s, 1a), 1.95 (1.2H, s, 2a), 2.25 (1.8H, s, 1a), 2.33 (1.2H, s, 2a), 3.59 (1.8H, s, 1a), 3.60 (1.2H, s, 2a), 8.10 (0.4H, s, 2a), 9.16 (0.6H, s, 1a). Exact structural assignment was made using NOESY experiment: With the major component, the significant NOE was observed between 1-NH proton (δ 9.16) and 6-methyl protons (δ 1.90) and as such, its structure was determined to be 1a (Fig. 4). With the minor component, the significant NOE was observed between 1-NH (δ 8.10) and 6,6-dimethyl protons (δ 1.28) and as such, its structure was determined to be 2a (Fig. 4). ¹³C-NMR (DMSO- d_6) δ : 12.5 (unresolved), 17.8, 23.3, 29.1, 31.0, 50.7 (unresolved), 53.3, 55.8, 104.8, 110.0, 142.0, 147.3, 150.5, 159.9, 167.7. The unresolved signals, δ 12.5 and δ 50.7, were further analyzed by HSQC experiment, which revealed that the signal at $\delta_{\rm C}$ 12.5 was due to the S-CH₃ of 1a and 2a because of its correlation to the signals at $\delta_{\rm H}$ 2.25 (s, 1a) and $\delta_{\rm H}$ 2.33 (s, 2a) and that $\delta_{\rm C}$ 50.7 was due to the OCH₃ of 1a and 2a because of its correlation to attached to the signals $\delta_{\rm H}$ 3.59 (s, 1a) and $\delta_{\rm H}$ 3.60 (s, 2a) (Fig. 4). IR (KBr) cm⁻¹: 2929, 1708, 1649, 1609, 1486, 1316, 1173. EI-MS *m/z*: 228.0939 (Calcd for C₁₀H₁₆N₂O₂S: 228.0933). MS *m/z*: 228 (M⁺), 213, 181.

Benzyl 1,4-Dihydro-4,4,6-trimethyl-2-methylthio Pyrimidine-5-car-



Fig. 4. NOE and HSQC Observed with 1a and 2a



Fig. 5. NOE and HSQC Observed with 1b and 2b



Fig. 6. NOE Observed with 1c

boxylate (1b) and Benzyl 1,6-Dihydro-4,6,6-trimethyl-2-methylthio Pyrimidine-5-carboxylate (2b): Colorless oil. Yield: 68%. 1b/2b=1.6:1. ¹H-NMR (DMSO- d_6) δ : 1.27 (3.69H, s, **1b**), 1.28 (2.31H, s, **2b**), 1.91 (1.85H, s, 1b), 1.96 (1.15H, s, 2b), 2.25 (1.85H, s, 1b), 2.33 (1.15H, s, 2b), 5.09 (1.23H, s, 1b), 5.11 (0.77H, s, 2b), 7.28-7.41 (5H, m, 1b+2b), 8.13 (0.38H, s, 2b), 9.20 (0.62H, s, 1b). Exact structural assignment was made using NOESY experiment: with the major component, the significant NOE was observed between 1-NH proton (δ 9.20) and 6-methyl protons (δ 1.91) and as such, its structure was determined to be 1b (Fig. 5). With the minor component, the significant NOE was observed between 1-NH (δ 8.13) and 6,6-dimethyl protons (δ 1.28) and as such, its structure was determined to be **2b** (Fig. 5). ¹³C-NMR (DMSO- d_6) δ : 12.5 (unresolved), 18.0, 23.4, 29.2, 31.1, 53.4, 55.9, 65.4 (unresolved), 104.6, 109.9, 128.2, 128.4, 128.7, 136.6, 142.6, 147.4, 151.1, 160.1, 167.0, 167.1. The unresolved signals, δ 12.5 and δ 65.4, were further analyzed by HSQC experiment, which revealed that the signal at $\delta_{\rm C}$ 12.5 was due to the S-CH₃ of **1b** and **2b** because of its correlation to the signals at $\delta_{\rm H}$ 2.25 (s, **1b**) and $\delta_{\rm H}$ 2.33 (s, **2b**) and that $\delta_{\rm C}$ 65.4 was due to the OCH₂Ph of 1b and 2b because of its correlation to attached to the signals $\delta_{\rm H}$ 5.09 (s, 1b) and $\delta_{\rm H}$ 5.11 (s, 2b) (Fig. 5). IR (neat) cm⁻¹: 3322, 1685, 1522, 1489, 1313, 1131. EI-MS m/z: 304.1249 (Calcd for C₁₆H₂₀N₂O₂S: 304.1245). MS *m/z*: 304 (M⁺), 289, 91.

Ethyl 4-Methyl-2-methylthio-1,3-diazaspiro[5,5]undeca-2,5-diene 5-**Carboxylate (1c) and Ethyl 4-Methyl-2-methylthio-1,3-diazaspiro**[5,5] undeca-2,4-diene 5-Carboxylate (2c): Colorless crystals. Yield: 63%. **1c/2c**=8.4:1. mp 111—112 °C (*n*-hexane). ¹H-NMR (DMSO-*d*₆) δ: 1.10—1.90 (13.32H, m, **1c+2c**), 1.83 (2.68H, s, **1c**), 2.31 (2.68H, s, **1c**), 2.32 (0.32H, s, **2c**), 4.06 (1.79H, q, *J*=7.2 Hz, **1c**), 4.09 (0.21H, q, *J*=7.2 Hz, **2c**), 7.75 (0.11H, s, **2c**), 9.10 (0.89H, s, **1c**). Exact structural assignment was made using NOESY experiment: with the major component, the significant NOE was observed between 1-NH proton (δ 9.10) and 6-methyl protons (δ 1.83) and as such, its structure was determined to be **1c** (Fig. 6). ¹³C-NMR (DMSO-*d*₆) δ: 12.6, 12.8, 14.3, 17.5, 19.8, 21.0, 22.3, 25.0, 25.9, 34.4, 36.9, 54.9, 57.6, 59.4, 59.6, 106.1, 112.0, 140.3, 146.9, 147.2, 159.4, 167.5. IR (KBr) cm⁻¹: 3293, 2944, 2922, 1686, 1656, 1618, 1482, 1141. EI-MS *m/z*: 282.1408 (Calcd for $C_{14}H_{22}N_2O_2S$: 282.1402). MS *m/z*: 282 (M⁺), 239, 209, 154.

Ethyl 1,4-Dihydro-4,6-dimethyl-2-methylthio-4-pentyl Pyrimidine-5-





carboxylate (1d) and Ethyl 1,6-Dihydro-4,6-dimethyl-2-methylthio-6pentvl Pyrimidine-5-carboxylate (2d): Colorless oil. Yield: 80%. 1d/2d = 1.7 : 1. ¹H-NMR (DMSO- d_6) δ : 0.81 (1.89H, t, J=7.2 Hz, 1d), 0.82 (1.11H, t, J=7.2 Hz, 2d), 1.00-1.38 (7H, m, 1d+2d), 1.175 (1.89H, t, J=7.2 Hz, 1d), 1.178 (1.11H, t, J=7.2 Hz, 2d), 1.22 (1.89H, s, 1d), 1.25 (1.11H, s, 2d), 1.82-1.94 (1H, m, 1d+2d), 1.92 (1.89H, s, 1d), 1.97 (1.11H, s, 2d), 2.24 (1.89H, s, 1d), 2.32 (1.11H, s, 2d), 4.00-4.09 (2H, m, 1d+2d), 7.93 (0.37H, s, 2d), 9.03 (0.63H, s, 1d). Exact structural assignment was made using NOESY experiment: with the major component, the significant NOE was observed between 1-NH proton (δ 9.03) and 6-methyl protons (δ 1.92) and as such, its structure was determined to be 1d (Fig. 7). With the minor component, the significant NOE was observed between 1-NH (δ 7.93) and 6-methyl protons (δ 1.25) and as such, its structure was determined to be 2d (Fig. 7). ¹³C-NMR (DMSO- d_{δ}) δ : 12.4, 12.5, 14.1 (unresolved), 14.3 (unresolved), 17.9, 22.3, 23.5, 24.2, 24.7, 29.6, 30.9, 31.7, 31.8, 40.9, 42.5, 56.9, 59.1, 59.2, 59.3, 102.9, 107.8, 142.8, 147.3, 151.5, 160.0, 167.2, 167.4. The unresolved signals, δ 14.1 and δ 14.3, were further analyzed by HSQC experiment, which revealed that the signal at δ_c 14.1 was due to the CH₂CH₂CH₂CH₂CH₃ of 1d and 2d because of its correlation to the signals at $\delta_{\rm H}$ 0.81 (t, 1d) and $\delta_{\rm H}$ 0.82 (t, 2d) and that $\delta_{\rm C}$ 14.3 was due to the OCH₂CH₃ of 1d and 2d because of its correlation to attached to the signals $\delta_{\rm H}$ 1.175 (t, 1d) and $\delta_{\rm H}$ 1.178 (t, 2d) (Fig. 7). IR (neat) cm⁻¹: 3313, 2927, 1674, 1487, 1138. EI-MS m/z: 298.1720 (Calcd for C15H26N2O2S: 298.1715). MS m/z: 298 (M⁺), 227, 199.

Ethyl 1,4-Dihydro-6-methyl-2-methylthio-4,4-dipropyl Pyrimidine-5carboxylate (1e) and Ethyl 1,6-Dihydro-4-methyl-2-methylthio-6,6dipropyl Pyrimidine-5-carboxylate (2e): Colorless oil. Yield: 63%. 1e/2e=1.8:1. ¹H-NMR (DMSO- d_6) δ : 0.79 (3.86H, t, J=7.2 Hz, 1e), 0.82 (2.14H, t, J=7.2 Hz, 2e), 1.09–1.34 (9H, m, 1e+2e), 1.75–1.87 (2H, m, 1e+2e), 1.93 (1.93H, s, 1e), 1.99 (1.07H, s, 2e), 2.24 (1.93H, s, 1e), 2.31 (1.07H, s, 2e), 4.04 (1.29H, q, J=7.2 Hz, 1e), 4.05 (0.71H, q, J=7.2 Hz, 2e), 7.77 (0.36H, s, 2e), 8.90 (0.64H, s, 1e). Exact structural assignment was made using NOESY experiment: with the major component, the significant NOE was observed between 1-NH proton (δ 8.90) and 6-methyl protons (δ 1.93) and as such, its structure was determined to be 1e (Fig. 8). With the minor component, the significant NOE was observed between 1-NH (δ 7.77) and 6-methylene protons (δ 1.75–1.87) and as such, its structure was determined to be 2e (Fig. 8). ¹³C-NMR (DMSO- d_6) δ : 12.2, 12.3, 14.3, 14.4, 14.6 (unresolved), 17.8, 18.01, 18.03, 23.8, 44.1, 45.3, 59.06, 59.13, 60.7, 63.0, 100.4, 105.0, 143.7, 147.2, 152.8, 160.3, 167.4, 167.5. The unresolved signal, δ 14.6, was further analyzed by HSQC experiment, which revealed that the signal at $\delta_{\rm C}$ 14.6 was due to the CH₂CH₂CH₃ of 1e and 2e because of its correlation to the signals at $\delta_{\rm H}$ 0.79 (t, 1e) and $\delta_{\rm H}$ 0.82 (t, 2e) (Fig. 8). IR (neat) cm⁻¹: 2958, 1689, 1668, 1658, 1499, 1279, 1139. EI-MS m/z: 298.1714 (Calcd for C₁₅H₂₆N₂O₂S: 298.1715). MS *m/z*: 298 (M⁺), 255, 227.

Ethyl 1,4-Dihydro-4,4-dimethyl-2-methylthio-6-phenyl Pyrimidine-5carboxylate (1f) and Ethyl 1,6-Dihydro-6,6-dimethyl-2-methylthio-4phenyl Pyrimidine-5-carboxylate (2f): Colorless crystals. Yield: 73%. $1f/2f=1:3.7. mp 132-134 \degree C$ (*n*-hexane–EtOAc). ¹H-NMR (DMSO- d_6) δ : 0.67 (0.64H, t, J=7.2 Hz, 1f), 0.77 (2.36H, t, J=7.2 Hz, 2f), 1.33 (1.28H, s,



Fig. 7. NOE and HSQC Observed with 1d and 2d



Fig. 8. NOE and HSQC Observed with 1e and 2e



Fig. 10. NOE and HSQC Observed with **1g** and **2g**

1f), 1.34 (4.72H, s, **2f**), 2.29 (0.64H, s, **1f**), 2.38 (2.36H, s, **2f**), 3.70 (0.43H, q, J=7.2 Hz, **1f**), 3.79 (1.57H, q, J=7.2 Hz, **2f**), 7.21 (0.43H, dd, J=1.8, 7.8 Hz, **1f**), 7.25—7.34 (3.93H, m, **2f**), 7.35—7.45 (0.64H, m, **1f**), 8.28 (0.79H, s, **2f**), 9.52 (0.21H, s, **1f**). Exact structural assignment was made using NOESY experiment: with the major component, the significant NOE was observed between 1-NH proton (δ 8.28) and 6,6-methyl protons (δ 1.34) and as such, its structure was determined to be **2f** (Fig. 9). With the minor component, the significant NOE was observed between 1-NH proton (δ 8.28) and 6,6-methyl protons (δ 1.34) and as such, its structure was determined to be **2f** (Fig. 9). With the minor component, the significant NOE was observed between 1-NH (δ 9.52) and 6-phenyl protons (δ 7.21) and as such, its structure was determined to be **1f** (Fig. 9). ¹³C-NMR (DMSO-*d*₆) δ : 12.7, 12.8, 13.46, 13.53, 28.6, 30.2, 53.2, 55.8, 59.3, 59.6, 107.0, 110.9, 127.7, 127.9, 128.0, 128.2, 128.3, 129.1, 135.3, 140.8, 141.4, 147.8, 148.9, 159.7, 167.5, 167.9, IR (KBr) cm⁻¹: 2976, 1702, 1535, 1499, 1326, 1194, 1051. EI-MS (EI) *m*/*z*: 304.1250 (Calcd for C₁₆H₂₀N₂O,S: 304.1245). MS (EI) *m*/*z*: 304 (M⁺), 289, 261.

Ethyl 4-Ethyl-1,4-dihydro-6-methyl-2-methylthio-4-phenyl Pyrimidine-5-carboxylate (1g) and Ethyl 6-Ethyl-1,6-dihydro-4-methyl-2methylthio-6-phenyl Pyrimidine-5-carboxylate (2g): Colorless crystals. Yield: 63%. 1g/2g=3.0:1. mp 149-150 °C (n-hexane-EtOAc). ¹H-NMR $(DMSO-d_6) \delta: 0.80 (2.25H, t, J=7.2 Hz, 1g), 0.90 (0.75H, t, J=7.2 Hz, 2g),$ 0.94 (2.25H, t, J=7.2 Hz, 1g), 0.95 (0.75H, t, J=7.2 Hz, 2g), 1.80 (0.25H, dq, J=13.8, 7.2 Hz, 2g), 1.87 (0.75H, dq, J=7.2, 13.2 Hz, 1g), 2.06 (2.25H, s, 1g), 2.13 (0.75H, s, 2g), 2.23 (2.25H, s, 1g), 2.34 (0.75H, s, 2g), 2.36-2.47 (1H, m, 1g+2g), 3.79-3.92 (2H, m, 1g+2g), 7.11 (0.75H, t, J=7.2 Hz, 1g), 7.19 (0.25H, t, J=7.2 Hz, 2g), 7.23 (1.5H, t, J=7.2 Hz, 1g), 7.30 (0.5H, t, J=7.2 Hz, 2g), 7.31 (1.5H, d, J=7.2 Hz, 1g), 7.37 (0.5H, d, J=7.2 Hz, 2g), 8.45 (0.25H, s, 2g), 9.31 (0.75H, s, 1g). Exact structural assignment was made using NOESY experiment: with the major component, the significant NOE was observed between 1-NH proton (δ 9.31) and 6methyl protons (δ 2.06) and as such, its structure was determined to be 1g (Fig. 10). With the minor component, the significant NOE was observed between 1-NH (δ 8.45) and 6-methylene protons (δ 1.80) and as such, its structure was determined to be 2g (Fig. 10). ¹³C-NMR (DMSO- d_6) δ : 9.8 (unresolved), 12.4, 12.5, 13.95, 13.98, 18.10, 24.0, 31.3, 32.9, 59.0, 59.1, 62.2, 64.9, 101.0, 105.6, 125.8, 126.4, 126.7, 126.8, 127.5, 127.9, 144.3,







Fig. 9. NOE Observed with 1f and 2f





Fig. 11. NOE Observed with 1h

147.9, 149.0, 150.3, 153.5, 159.8, 166.95, 167.0. The unresolved signal, δ 9.8, was further analyzed by HSQC experiment, which revealed that the signal at $\delta_{\rm C}$ 9.8 was due to the S-CH₃ of **1g** and **2g** because of its correlation to the signals at $\delta_{\rm H}$ 2.23 (s, **1g**) and $\delta_{\rm H}$ 2.34 (s, **2g**) (Fig. 10). IR (KBr) cm⁻¹: 3264, 2969, 1664, 1583, 1530, 1462, 1314, 1073. EI-MS *m/z*: 318.1401 (Calcd for C₁₇H₂₂N₂O₂S: 318.1400). MS *m/z*: 318 (M⁺), 289, 261.

Ethyl 6-Methyl-2-methylthio Spiro[9'*H*-fluorene-9',4-1,4-dihydropyrimidine] 5-Carboxylate (1h) and Ethyl 4-Methyl-2-methylthio Spiro[9'*H*-fluorene-9',6-1,6-dihydropyrimidine] 5-Carboxylate (2h): Yellow amorphous solid. Yield: 40%. 1h/2h=6.0:1. mp 57—63 °C. ¹H-NMR (DMSO- d_6) δ: 0.35 (2.57H, t, *J*=7.2 Hz, 1h), 0.36 (0.43H, t, *J*=7.2 Hz, 2h), 2.02 (2.57H, s, 1h), 2.23 (2.57H, s, 1h), 2.27 (0.43H, s, 2h), 2.40 (0.43H, s, 2h), 3.30 (1.71H, q, *J*=7.2 Hz, 1h), 3.33 (0.29H, q, *J*=7.2 Hz, 2h), 7.19— 7.26 (3.42H, m, 1h), 7.26—7.29 (0.29H, m, 2h), 7.30 (1.71H, dt, *J*=1.2, 7.2 Hz, 1h), 7.36 (0.29H, dt, *J*=1.2, 7.2 Hz, 2h), 7.40 (0.29H, d, *J*=7.8 Hz, 2h), 7.71 (1.71H, d, *J*=7.2 Hz, 1h). Exact structural assignment was made using NOESY experiment: with the major component, the significant NOE was observed between 1-NH proton (δ 9.73) and 6-methyl protons (δ 2.23) and as such, its structure was determined to be 1h (Fig. 11). ¹³C-NMR Acknowledgements Y. N. thanks Prof. Hidetsura Cho, Graduate School of Pharmaceutical Sciences, Graduate School of Sciences, Tohoku University for a helpful discussion in a seminal stage of this work.

References and Notes

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