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Efficient synthesis of 4-phosphinoyl-4,5,6,7-tetrahydro-2H-indazol-3-ylamines

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ABSTRACT

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We described in this paper an efficient synthesis of variously substituted 4-phosphinoyl-4,5,6,7tetrahydro-2*H*-indazol-3-ylamines involving nucleophilic addition of hydrazine on β-ketothioamides as key-step.

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1. Introduction

Because of their chemical and/or biological significance, nitrogen containing heterocycles represent an important and attractive area of organic chemistry. Among various series of heterocycles, we focused our attention on indazolyl moieties. As mentioned in literature, numerous examples of indazolyl heterocycles exhibit biological properties as well as anticoagulating agent,¹ topoisomerase bacterium II inhibitor,² or for treatment of hypercholesterol,³ tuberculosis,⁴ ... (Fig. 1).



Fig. 1. Examples of imidazolyl heterocycles with inhibiting biological properties.

Given that phosphonate-containing azaderivatives are of biological interest,⁵ it seems interesting to combine indazolyl ring and phosphinoyl group in the same skeleton to enhance the potential biological properties. In connection with our interest in

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synthesizing azaheterocyclic derivatives,^{6,7} we then investigated the preparation of various phosphinoyltetrahydroindazolyl amines.

2. Results and discussion

In our knowledge, the most current strategy for the preparation of tetrahydroindazolyl or pyrazolyl rings consists in a cyclocondensation of hydrazines with 1.3-dicarbonyl compounds, unsaturated ketones. β -ketoesters or thioamides.^{8–12} Consequently. we envisioned to build 1,2-diazole cycle using a such cyclocondensation step between N-substituted hydrazines and desired phosphinoyl β-ketothiamides to success in the preparation of the expected amino tetrahydroindazoles (Scheme 1).



Scheme 1. Retrosynthetic analysis for the elaboration of indazolylamines 4.

The expected 3-phosphinoylcyclohexanones 2a-g were prepared in good yields (58-81%) by reaction between cyclohex-2enones **1a**–**g** and *P*-chlorodiphenylphosphine (1 equiv) (Table 1). at room temperature. A mechanism proposal might envision introduction of phosphinoyl group according to Michael addition of



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the in situ prepared diphenylphosphinyl acetate, followed by [1,2] oxaphospholane formation and then ring opening by acidic hydrolysis to give **2a**–**g**.

Table 1

Preparation of the 3-phosphinoylcyclohexanones 2a-h and mechanism proposal



^a Isolated yield after silica gel chromatography.

In a second step, isothiocyanates were used to introduce the expected thioamide units (Table 2). The sequence was realized within THF by action of potassium *tert*-butoxide (1 equiv) on 3-phosphinoylcyclohexanones **2a,b**, for 2 h, followed by the addition of isothiocyanate (1 equiv). The reaction was performed at room temperature overnight. The expected 2-phosphinoyl-6-oxocyclohexanecarbothioamides **3a**–**d** were then isolated in good yields (66–81%).

Table 2

Preparation of the 2-phosphinoyl-6-oxocyclohexanecarbothioamides 3a-d

| R ₅ R ₄ 2a-h | R ₂ P(Ph) ₂ U | i) <i>t</i> BuOK (1eq), ii) R ₆ N=C=S (1e iii) Hydrolysis | THF, rt, 2h aq), THF, rt, 16h | R ₅ R ₄ 3a- | $ \begin{array}{c} S \\ R_2 \\ P(Ph)_2 \\ 0 \\ d \end{array} $ |
|--|---|--|----------------------------------|---|--|
| Compound | R ₂ | R ₄ | R ₅ | R ₆ | Yields ^a (%) |
| 3a | Н | Н | Н | Ph | 81 |
| 3b | Н | Н | Н | CH ₂ Ph | 77 |
| 3c | Н | Н | Н | c-Hex | 72 |
| 3d | CH ₃ | CH ₃ | CH ₃ | Ph | 66 |

^a Isolated yield after silica gel chromatography.

The following stage consisted in adding hydrazines (1.1 equiv) on β -ketothioamides **3a**–**d** to lead to ring closure of the targeted 4-phosphinoyl-4,5,6,7-tetrahydro-2*H*-indazol-3-ylamines **4a**–**i** (Table 3).

By considering the possible mechanism (Scheme 2), such a cyclisation can lead to the formation of the two isomers **4** and **5**, according to the nucleophilic attack of the primary or the secondary amino site in hydrazine.

In a general way, it is to be noted that only isomers **4a** and **4c**–**i** were generated in the reaction mixture in good yields (60–76%).

Table 3

Preparation of the 4-phosphinoyl-4,5,6,7-tetrahydro-2H-indazol-3-ylamines 4a-i



| Compound | R ₆ | R ₇ | Yields ^a (%) |
|----------|--------------------|------------------------------------|-------------------------|
| 4a | Ph | Н | 74 |
| 4b | Ph | CH₃ | 51 ^{b,c} |
| 4c | Ph | CH ₂ CH ₂ CN | 68 |
| 4d | CH ₂ Ph | Н | 76 |
| 4e | CH ₂ Ph | CH₃ | 68 |
| 4f | CH ₂ Ph | CH ₂ CH ₂ CN | 74 |
| 4g | c-Hex | Н | 60 |
| 4h | c-Hex | CH ₃ | 61 |
| 4i | c-Hex | CH ₂ CH ₂ CN | 70 |

^a Isolated yield after silica gel chromatography.

^b Isomer **4b** was obtained in mixture with **5b**.

^c Yield of **4b** after purification by silica gel chromatography, 23% of **5b** were also obtained as by-product.



Scheme 2. Proposed mechanism for the reaction of 3a-c with hydrazine.

Surprisingly, in the case of the addition of methylhydrazine on β-ketothioamide **3a**, a mixture of isomers **4b** and **5b** was isolated. Isomers **4b** and **5b** were able to be separated by chromatography on silica gel (respectively, 51 and 23% yields). Structures of 4b and 5b were determined after a detailed two-dimensional NMR study (see Supplementary data). Potential explanations for these results involve electronic considerations. General behavior could be explained by a nucleophilic addition of the most reactive nitrogen atom present in hydrazine reagents. Due to the overall electronwithdrawing effect of the cyano group, only primary amino group of the propionitrile-hydrazine could react as nucleophile center conducting to the formation of derivatives 4c, 4f, and 4i. In contrast with methylhydrazine, the methyl group slightly increased the electron-density on the central nitrogen atom, and consequently its nucleophile character compared to the terminal one. However, starting from thioamides **3b** and **c**, only isomers **4e** and **4h** were afforded in 68 and 61% yields, respectively, no trace of isomers 5e and **5h** was detected, electronic effect induced by the methyl group did not affect the reactivity. On the other hand, starting from *N*-phenylthioamide **3a**, mixture of isomers **4b** and **5b** was obtained, nucleophile character of methylamino group was displayed in this case. Considering the proposed mechanism (Scheme 3), the formation of the indazolyl ring may be realized by nucleophilic addition of the amino group of **[I**] on thioamide substituent; as the result various intermediates **[I-5b]** to **[III-5b]** might be envisioned. **[III-5b]**, which presents a strong density of charge on nitrogen atom of the amine, might be favored only in the presence of phenyl as an electronic stabilizing effect substituent. Subsequent H₂S elimination allowed the formation of the compound **5b**.



Scheme 3. Proposed mechanism for the formation of 4-phosphinoyl-4,5,6,7-tetrahydro-1*H*-indazol-3-ylamine **5b**.

In the second part of our work, we tried to apply the strategy described above for **4a**–**i** to reach pyrazoles **8** starting from acyclic β -ketophosphonates **6a**,**b**^{13–16} (Table 4). The expected ketothioamides **7a**–**d** were easily prepared by addition of potassium enolates derived from **6a**,**b** on phenyl-, benzyl- or *c*-hexylisothiocyanates (77–87% yields). Nevertheless, concerning the second step, in spite of numerous realized experiments, no trace of the expected azaheterocyclic derivatives **8** was observed. Due to the nearness of phosphoryl unit in α -position as electron-withdrawing group,

Table 4

Aliphatic series: preparation of compounds 7a-d and 9b,c,e



^a Isolated yield after silica gel chromatography.

a cleavage of the carbonyl moieties occurred easily and led to the interesting phosphinoylthioamides **9b**, **c**, and **e** in 77, 72, and 76% yields, respectively.

This deacetylation (or debenzoylation for **7a**) sequence, using hydrazine as reagent and at room temperature as condition, turns out an interesting alternative leading to an efficient and versatile preparation of β -phosphorylthioamides from easily accessible substrates.

3. Conclusion

We succeeded in the synthesis of various 4-phosphinoyl-4,5,6,7-tetrahydro-2*H*-indazol-3-ylamines with moderate to good over yields (33–47%); an extension of this reaction sequence can be envisioned to allow the preparation of compounds bearing appropriate substituents, which can confer a potential biological activity to these heterocycles.

4. Experimental section

4.1. General

¹H, ¹³C, and ³¹P NMR spectra were recorded at 400 or 250 or 200, 100 or 63 or 50 and 300 Mhz, respectively, with CDCl₃ as solvent and TMS (for ¹H NMR) or H₃PO₄ (³¹P NMR) as internal standards. HRMS spectra were recorded on a BRUKER micrOTOF-Q spectrometer. MS were recorded on an SHIMADZU GCMS-QP2010 spectrometer. Melting temperatures are uncorrected.

Cyclohex-2-enones **1a**–**g** were commercially available or prepared according to the procedure described in literature.^{16–18}

4.2. General procedure for the preparation of 3-(diphenylphosphinoyl)cyclohexanones (2a–h)

To a solution of cyclohexenones 1a-g (5.0 mmol, 1.0 equiv) in glacial acetic acid (50 mL) was added dropwise *P*-chlorodiphenylphosphine (5.0 mmol, 1.0 equiv) at room temperature, under nitrogen atmosphere. After addition of H₂O (1.5 mL), the reactional mixture was refluxed during a reaction time (t_R). After extraction with chloroforme (CHCl₃), the combined organic layers were washed with an aqueous saturated NaCl solution, then with an aqueous saturated NaHCO₃ solution. After drying (MgSO₄), filtration, and solvent evaporation, the crude product was purified by recrystallisation in diethyl ether (Et₂O).

4.2.1. 3-(*Diphenylphosphinoyl*)*cyclohexanone* (**2a**). Compound **2a** $t_{\rm R}$: 2 h; solid; yield=81%; mp 144–146 °C; ¹H NMR $\delta_{\rm H}$ (ppm) 1.70–2.50 (m, 9H, CH, CH₂), 7.40–7.90 (m, 10H, CH_{ar}); ¹³C NMR $\delta_{\rm C}$ (ppm) 22.6, 25.5 (*J*_{CP} 15.9 Hz), 37.1 (*J*_{CP} 75.7 Hz), 38.6, 40.3, 128.0–132.0, 208.6 (*J*_{CP} 14.6 Hz); ³¹P NMR $\delta_{\rm P}$ (ppm) 31.46; HRMS calcd for C₁₈H₁₉NaO₂P (M+Na)⁺: 321.1015; found: 321.1014.

4.2.2. 3-(Diphenylphosphinoyl)-3,5,5-trimethylcyclo hexanone (**2b**). Compound **2b** $t_{\rm R}$: 5 h; solid; yield=79%; mp 116–118 °C; ¹H NMR $\delta_{\rm H}$ (ppm) 1.20–3.10 (m, 6H, CH₂), 1.01 (s, 3H, CH₃), 1.04 (s, 3H, CH₃), 1.42 (d, 3H, ³ $J_{\rm PH}$ 17.9 Hz, CH₃), 7.20–8.10 (m, 10H, CH_{ar}); ¹³C NMR $\delta_{\rm C}$ (ppm) 22.1, 28.8, 33.9, 37.0 ($J_{\rm CP}$ 13.7 Hz), 41.3, 42.1 ($J_{\rm CP}$ 68.9 Hz), 45.0, 53.5, 128.0–133.0, 210.0; ³¹P NMR $\delta_{\rm P}$ (ppm) 36.74.

4.2.3. 3-(*Diphenylphosphinoyl*)-3-*methylcyclohexanone* (**2c**). Compound **2c** $t_{\rm R}$: 2 h; solid; yield=76%; mp 131–133 °C; ¹H NMR $\delta_{\rm H}$ (ppm) 1.32 (s, 3H, ³ $J_{\rm PH}$ 18.0 Hz, CH₃), 1.00–2.90 (m, 8H, CH₂), 7.30–7.80 (m, 10H, CH_{ar}); ¹³C NMR $\delta_{\rm C}$ (ppm) 25.1 ($J_{\rm CP}$ 13.9 Hz), 27.8, 29.6, 32.5 (J_{CP} 69.6 Hz), 44.2, 44.9, 127.0–135.0, 211.5 (J_{CP} 13.2 Hz); ³¹P NMR δ_P (ppm) 31.56.

4.2.4. 3-(*Diphenylphosphinoyl*)-5-methylcyclohexanone (**2d**). Mixture of diastereoisomers: $t_{\rm R}$: 3 h; solid; yield=72%; mp 104–106C; ¹H NMR $\delta_{\rm H}$ (ppm) 1.02 (d, 3H, ³*J*_{HH} 6.4 Hz, CH₃), 1.10–2.80 (m, 8H, CH, CH₂), 7.20–8.00 (m, 10H, CH_{ar}); ¹³C NMR $\delta_{\rm C}$ (ppm) 21.7, 21.7, 26.5, 29.8 (*J*_{CP} 13.2 Hz), 37.4 (*J*_{CP} 75.6 Hz), 41.9, 42.1, 128.0–132.0, 199.2 and 208.2 (*J*_{CP} 11.7 Hz); ³¹P NMR $\delta_{\rm P}$ (ppm) 32.13 (63%) and 34.47 (38%).

4.2.5. 3-(*Diphenylphosphinoyl*)-4-*isopropylcyclo* hexanone (**2e**). Mixture of diastereoisomers: $t_{\rm R}$: 6 h; solid; yield=58%; mp 178–180 °C; ¹H NMR $\delta_{\rm H}$ (ppm) 0.44 (d, 3H, ³ $J_{\rm HH}$ 6.7 Hz, CH₃), 0.75 (d, 3H, ³ $J_{\rm HH}$ 6.7 Hz, CH₃), 1.40–2.80 (m, 9H, CH, CH₂), 7.20–8.00 (m, 10H, CH_{ar}); ¹³C NMR $\delta_{\rm C}$ (ppm) 15.9, 16.3, 28.8, 32.4 ($J_{\rm CP}$ 15.4 Hz), 33.2 ($J_{\rm CP}$ 73.1 Hz), 42.3, 43.6, 44.2, 129.0–134.0, 210.3 ($J_{\rm CP}$ 11.2 Hz); ³¹P NMR $\delta_{\rm P}$ (ppm) 33.60 (51%) and 31.92 (49%).

4.2.6. (5R)-3-(Diphenylphosphinoyl)-5-isopropenyl-2methylcyclohexanone (**2f**). Mixture of diastereoisomers: t_R : 6 h; solid; yield=64%; mp 167–169 °C; ¹H NMR δ_H (ppm) 0.97 (m, 3H, ³J_{HH} 6.6 Hz, CH₃), 1.05 (s, 3H, CH₃), 1.20–3.10 (m, 7H, CH, CH₂), 4.76–4.79 (m, 2H, CH₂=C), 7.30–8.00 (m, 10H, CH_{ar}); ¹³C NMR δ_C (ppm) 15.6 and 16.3 and 17.9 and 14.0, 19.8, 26.8, 35.2, 39.2 (J_{CP} 71.9 Hz), 40.4, 44.4, 128.0–135.0, 199.6 (J_{CP} 9.2 Hz); ³¹P NMR δ_P (ppm) 31.45 (66%) and 30.15 (19%) and 34.79 (8%) and 33.39 (7%).

4.2.7. (55)-3-(Diphenylphosphinoyl)-5-isopropenyl-2methylcyclohexanone (**2g**). Mixture of diastereoisomers: t_R : 6 h; solid; yield=62%; mp 167–168 °C; ¹H NMR δ_H (ppm) 1.00 (m, 3H, ³J_{HH} 6.9 Hz, CH₃), 1.08 (s, 3H, CH₃), 1.20–3.20 (m, 7H, CH, CH₂), 4.78–4.81 (m, 2H, CH₂=C), 7.50–8.00 (m, 10H, CH_{ar}); ¹³C NMR δ_C (ppm) 15.7 and 12.5, 20.3, 26.8, 35.2, 39.2 (J_{CP} 71.7 Hz), 40.5, 40.6, 120.0–135.0, 199.5 and 210.1 (J_{CP} 9.3 Hz); ³¹P NMR δ_P (ppm) 31.44 (41%), 33.43 (26%), 34.76 (23%), 31.53 (10%).

4.3. General procedure for the preparation of 2-phosphinoyl-6-oxocyclohexanecarbothioamides (3a–d)

To a solution of potassium *tert*-butoxide (0.01 mol, 1.0 equiv) in anhydrous THF (10 mL) was added dropwise 3-phosphinoylcyclohexanones **2** (0.01 mol, 1.0 equiv) in solution in anhydrous THF (20 mL) at room temperature, under nitrogen atmosphere. After 2 h of stirring, isothiocyanate (0.01 mol, 1.0 equiv) in solution in anhydrous THF (5 mL) was added to the reactional mixture. After 16 h of stirring at room temperature, the hydrolysis was performed with an aqueous saturated NH₄Cl solution. The aqueous layer was then extracted twice with CHCl₃ (2×50 mL). After drying (MgSO₄), filtration and solvent evaporation, the crude product was purified by column chromatography on silica gel (0.063–0.200 mm, eluent: 95/5 Et₂O/EtOH).

4.3.1. 2-(Diphenylphosphinoyl)-6-oxocyclohexane carbothioic acid phenylamide (**3a**). Solid; yield=81%; mp 202–204 °C; ¹H NMR $\delta_{\rm H}$ (ppm) 1.87–1.95 (m, 2H, CH₂), 2.18–2.25 (m, 2H, CH₂), 2.62–2.69 (m, 2H, CH₂), 4.22–4.29 (m, 1H, CHP(O)), 4.53–4.61 (m, 1H, CHC(S)), 7.02–7.83 (m, 15H, CH_{ar}), 11.58 (s, 1H, NH); ¹³C NMR $\delta_{\rm C}$ (ppm) 21.9 (J_{CP} 13.6 Hz), 22.5, 38.8, 39.6 (J_{CP} 68.6 Hz), 61.8, 123.0–139.0, 196.8 (J_{CP} 2.3 Hz), 205.9 (J_{CP} 8.3 Hz); ³¹P NMR $\delta_{\rm P}$ (ppm) 35.0; HRMS calcd for C₂₅H₂₅NO₂PS (M+H)⁺: 434.1338; found: 434.1366.

4.3.2. 2-(Diphenylphosphinoyl)-6-oxocyclohexane carbothioic acid benzylamide (**3b**). Solid; yield=77%; mp 195–197 °C; ¹H NMR $\delta_{\rm H}$ (ppm) 1.71–1.77 (m, 2H, CH₂), 1.92–1.96 (m, 2H, CH₂), 2.14–2.22 (m, 2H, CH₂), 3.92–3.99 (m, 1H, CHP(O)), 4.15–4.19 (m, 2H, CH₂N),

4.61–4.66 (m, 1H, CHC(S)), 7.03–7.78 (m, 15H, CH_{ar}), 9.67 (s, 1H, NH); ¹³C NMR δ_{C} (ppm) 21.9 (J_{CP} 12.8 Hz), 22.2, 38.7, 39.3 (J_{CP} 56.5 Hz), 39.6, 60.4, 127.5–135.7, 197.9 (J_{CP} 3.0 Hz), 205.9 (J_{CP} 7.5 Hz); ³¹P NMR δ_{P} (ppm) 35.1; HRMS calcd for C₂₆H₂₆NNaO₂PS (M+Na)⁺: 470.1314; found: 470.1337.

4.3.3. 2-(Diphenylphosphinoyl)-6-oxocyclohexane carbothioic acid cyclohexylamide (**3c**). Solid; yield=72%; mp 213–215 °C; ¹H NMR $\delta_{\rm H}$ (ppm) 1.29–1.77 (m, 12H, CH₂), 2.02–2.44 (m, 4H, CH₂), 3.63–3.65 (m, 1H, CHN), 3.81–3.89 (m, 1H, CHP(O)), 4.06–4.11 (m, 1H, CHC(S)), 7.14–7.61 (m, 10H, CH_{ar}), 8.93 (s, 1H, NH); ¹³C NMR $\delta_{\rm C}$ (ppm) 21.7, 22.1, 23.3, 24.6, 30.6, 37.7 ($J_{\rm CP}$ 59.6 Hz), 41.0, 54.7, 60.4, 128.1–132.4, 195.4 ($J_{\rm CP}$ 4.5 Hz), 205.6 ($J_{\rm CP}$ 13.3 Hz); ³¹P NMR $\delta_{\rm P}$ (ppm) 35.4; HRMS calcd for C₂₅H₃₁NO₂PS (M+H)⁺: 440.1808; found: 440.1812.

4.3.4. 2-(Diphenylphosphinoyl)-2,4,4-trimethyl-6oxocyclohexanecarbothioic acid phenylamide (**3d**). Solid; yield=66%; mp 198–200 °C; ¹H NMR $\delta_{\rm H}$ (ppm) 1.29–3.14 (m, 4H, CH₂), 1.11 (s, 3H, CH₃), 1.16 (s, 3H, CH₃), 1.49 (d, 3H, ³J_{PH} 18.0 Hz, CH₃), 4.31–4.35 (m, 1H, CHC(S)), 7.29–7.96 (m, 15H, CH_{ar}), 10.28 (s, 1H, NH); ¹³C NMR $\delta_{\rm C}$ (ppm) 22.1, 27.2, 34.0, 37.1 (J_{CP} 13.6 Hz), 41.2, 42.1 (J_{CP} 68.7 Hz), 45.1, 65.8, 128.4–132.6, 194.2 (J_{CP} 3.0 Hz), 210.4 (J_{CP} 13.6 Hz); ³¹P NMR $\delta_{\rm P}$ (ppm) 38.7.

4.4. General procedure for the preparation of 4-phosphinoyl-4,5,6,7-tetrahydro-2*H*-indazol-3-ylamines (4a–i)

Hydrazine (0.0011 mol, 1.1 equiv) was added to a solution of β ketothioamides **3a**–**c** (0.001 mol, 1.0 equiv) in CHCl₃. After 24 h of stirring at room temperature, the residue was dried (MgSO₄), solvent was evaporated and the crude product was purified by column chromatography on silica gel (0.063–0.200 mm, eluent: 90/10 Et₂O/EtOH).

4.4.1. [4-(Diphenylphosphinoyl)-4,5,6,7-tetrahydro-2H-indazol-3-yl] phenylamine (**4a**). Solid; yield=74%; mp 216–218 °C; ¹H NMR $\delta_{\rm H}$ (ppm) 1.66–1.73 (m, 2H, CH₂), 1.86–1.98 (m, 2H, CH₂), 2.26–2.40 (m, 2H, CH₂), 3.49–3.55 (m, 1H, CHP(O)), 6.59 (s, 1H, HN–N), 7.07–7.70 (m, 15H, CH_{ar}), 7.74 (s, 1H, NHPh); ¹³C NMR $\delta_{\rm C}$ (ppm) 24.1, 25.6, 26.3, 34.8 (*J*_{CP} 78.1 Hz), 99.8 (*J*_{CP} 6.8 Hz), 128.4–132.1, 144.8; ³¹P NMR $\delta_{\rm P}$ (ppm) 39.4.

4.4.2. [4-(Diphenylphosphinoyl)-2-methyl-4,5,6,7-tetrahydro-2H-indazol-3-yl]phenylamine (**4b**). Solid; yield=51%; mp 223–225 °C; ¹H NMR $\delta_{\rm H}$ (ppm) 1.50–1.79 (m, 2H, CH₂), 1.85–2.00 (m, 2H, CH₂), 2.30–2.55 (m, 2H, CH₂), 3.65 (s, 3H, CH₃), 3.78–3.90 (m, 1H, CHP(O)), 6.70–7.87 (m, 15H, CH_{ar}), 8.62 (s, 1H, NHPh); ¹³C NMR $\delta_{\rm C}$ (ppm) 21.5, 21.6, 25.1, 35.0, 35.8 (*J*_{CP} 64.1 Hz), 97.3 (*J*_{CP} 5.3 Hz), 116.2–144.1, 149.2; ³¹P NMR $\delta_{\rm P}$ (ppm) 38.6; HRMS calcd for C₂₆H₂₆N₃NaOP (M+Na)⁺: 450.1706; found: 450.1703.

4.4.3. [4-(Diphenylphosphinoyl)-1-methyl-4,5,6,7-tetrahydro-1H-indazol-3-yl]phenylamine (**5b**). Solid; yield=23%; mp 222–224 °C; ¹H NMR $\delta_{\rm H}$ (ppm) 1.41–1.57 (m, 2H, CH₂), 1.78–2.04 (m, 2H, CH₂), 2.30–2.65 (m, 2H, CH₂), 3.63 (s, 3H, CH₃), 3.71–3.85 (m, 1H, CHP(O)), 6.52–7.86 (m, 15H, CH_{ar}), 8.08 (s, 1H, NHPh); ¹³C NMR $\delta_{\rm C}$ (ppm) 20.2, 22.2, 23.3, 31.5 (*J*_{CP} 48.3 Hz), 34.7, 99.9 (*J*_{CP} 6.8 Hz), 113.9–144.0, 147.8; ³¹P NMR $\delta_{\rm P}$ (ppm) 40.1; HRMS calcd for C₂₆H₂₇N₃OP (M+H)⁺: 428.1886; found: 428.1896.

4.4.4. 3-[4-(Diphenylphosphinoyl)-3-phenylamino-4,5,6,7tetrahydro-2H-indazol-2-yl]propionitrile (**4c**). Solid; yield=68%; mp 228–230 °C; ¹H NMR $\delta_{\rm H}$ (ppm) 1.45–1.49 (m, 2H, CH₂), 1.81–1.93 (m, 2H, CH₂), 2.23–2.52 (m, 2H, CH₂), 2.76–2.88 (m, 2H, CH₂), 3.67–3.76 (m, 1H, CHP(O)), 4.03–4.09 (m, 2H, CH₂CN), 6.51–7.74 (m, 15H, CH_{ar}), 7.89 (s, 1H, NHPh); ¹³C NMR δ_{C} (ppm) 15.2, 18.1, 20.9, 24.1, 33.1 (J_{CP} 70.1 Hz), 43.6, 102.0 (J_{CP} 6.8 Hz), 114.7, 119.9–144.9, 150.5; ³¹P NMR δ_{P} (ppm) 40.3.

4.4.5. Benzyl-[4-(diphenylphosphinoyl)-4,5,6,7,-tetrahydro-2H-indazol-3-yl]amine (**4d**). Solid; yield=76%; mp 219–211 °C; ¹H NMR $\delta_{\rm H}$ (ppm) 1.33–1.41 (m, 2H, CH₂), 1.61–1.68 (m, 2H, CH₂), 2.43–2.49 (m, 2H, CH₂), 3.60–3.97 (m, 3H, CHP(O), CH₂N), 5.13 (s, 1H, HN–N), 6.33 (s, 1H, NHBn), 7.25–7.52 (m, 15H, CH_{ar}); ¹³C NMR $\delta_{\rm C}$ (ppm) 21.0, 23.5, 25.1, 34.1 (J_{CP} 115.5 Hz), 52.8, 100.4 (J_{CP} 6.9 Hz), 128.0–143.8, 146.8; ³¹P NMR $\delta_{\rm P}$ (ppm) 39.6; HRMS calcd for C₂₆H₂₇N₃OP (M+H)⁺: 428.1886; found: 428.1876.

4.4.6. Benzyl-[4-(diphenylphosphinoyl)-2-methyl-4,5,6,7tetrahydro-2H-indazol-3-yl]amine (**4e**). Solid; yield=68%; mp 227–229 °C; ¹H NMR $\delta_{\rm H}$ (ppm) 1.26–1.39 (m, 2H, CH₂), 1.51–1.64 (m, 2H, CH₂), 2.36–2.43 (m, 2H, CH₂), 3.66 (s, 3H, CH₃), 3.70–4.00 (m, 3H, CHP(O), CH₂N), 6.13 (s, 1H, NHBn), 7.15–7.42 (m, 15H, CH_{ar}); ¹³C NMR $\delta_{\rm C}$ (ppm) 21.6, 23.5, 24.7, 34.2, 35.1 (J_{CP} 145.5 Hz), 52.7, 100.0 (J_{CP} 6.8 Hz), 127.0–145.8, 147.8; ³¹P NMR $\delta_{\rm P}$ (ppm) 39.1; HRMS calcd for C₂₇H₂₉N₃OP (M+H)⁺: 442.2043; found: 442.2039.

4.4.7. 3-[3-Benzylamino-4-(diphenylphosphinoyl)-4,5,6,7tetrahydro-2H-indazol-2-yl]propionitrile (**4f**). Solid; yield=74%; mp 233–235 °C; ¹H NMR $\delta_{\rm H}$ (ppm) 1.29–1.38 (m, 2H, CH₂), 1.50–1.59 (m, 2H, CH₂), 2.34–2.41 (m, 2H, CH₂), 2.75–2.81 (m, 2H, CH₂), 3.72–4.00 (m, 3H, CHP(O), CH₂N), 4.06–4.10 (m, 2H, CH₂CN), 7.01 (s, 1H, NHBn), 7.15–7.52 (m, 15H, CH_{ar}); ¹³C NMR $\delta_{\rm C}$ (ppm) 18.1, 21.6, 22.7, 24.3, 35.1 (J_{CP} 90.5 Hz), 42.9, 52.7, 102.1 (J_{CP} 6.3 Hz), 128.3–144.6, 148.9; ³¹P NMR $\delta_{\rm P}$ (ppm) 39.8; HRMS calcd for C₂₉H₂₉N₄NaOP (M+Na)⁺: 503.1971; found: 503.1986.

4.4.8. Cyclohexyl-[4-(diphenylphosphinoyl)-4,5,6,7-tetrahydro-2Hindazol-3-yl]amine (**4g**). Solid; yield=60%; mp 218–220 °C; ¹H NMR $\delta_{\rm H}$ (ppm) 1.09–1.25 (m, 6H, CH₂), 1.45–1.88 (m, 8H, CH₂), 2.22–2.43 (m, 2H, CH₂), 3.18–3.26 (m, 1H, CHN), 3.58–3.67 (m, 1H, CHP(O)), 4.90 (s, 1H, HN–N), 6.10 (s, 1H, NH–cHex), 7.34–7.72 (m, 10H, CH_ar); ¹³C NMR $\delta_{\rm C}$ (ppm) 18.4, 20.9, 21.8, 24.9, 25.4, 32.2 (J_{CP} 113.1 Hz), 35.3, 51.9, 94.1 (J_{CP} 5.9 Hz), 128.3–132.1, 152.2; ³¹P NMR $\delta_{\rm P}$ (ppm) 37.3; HRMS calcd for C₂₅H₃₁N₃OP (M+H)⁺: 420.2199; found: 420.2197.

4.4.9. *Cyclohexyl-[4-(diphenylphosphinoyl)-2-methyl-4,5,6,7tetrahydro-2H-indazol-3-yl]amine* (**4h**). Solid; yield=61%; mp 220–222 °C; ¹H NMR $\delta_{\rm H}$ (ppm) 1.09–1.28 (m, 6H, CH₂), 1.42–1.78 (m, 8H, CH₂), 2.26–2.39 (m, 2H, CH₂), 3.23–3.31 (m, 1H, CHN), 3.56 (s, 3H, CH₃), 3.53–3.62 (m, 1H, CHP(O)), 6.23 (s, 1H, NH–*c*Hex), 7.31–7.70 (m, 10H, CH_{ar}); ¹³C NMR $\delta_{\rm C}$ (ppm) 18.9, 20.2, 21.8, 23.9, 25.7, 32.8 (*J*_{CP} 113.1 Hz), 34.7, 35.3, 52.6, 97.4 (*J*_{CP} 7.0 Hz), 127.3–135.4, 151.2; ³¹P NMR $\delta_{\rm P}$ (ppm) 38.3; HRMS calcd for C₂₆H₃₃N₃OP (M+H)⁺: 434.2356; found: 434.2347.

4.4.10. 3-[3-Cyclohexylamino-4-(diphenylphosphinoyl)-4,5,6,7tetrahydro-2H-indazol-2-yl] propionitrile (**4i**). Solid; yield=70%; mp 224–226 °C; ¹H NMR $\delta_{\rm H}$ (ppm) 1.14–1.28 (m, 6H, CH₂), 1.39–1.93 (m, 8H, CH₂), 2.30–2.41 (m, 2H, CH₂), 2.79–2.85 (m, 2H, CH₂), 3.22–3.31 (m, 1H, CHN), 3.58–3.69 (m, 1H, CHP(O)), 4.07–4.12 (m, 2H, CH₂CN), 6.80 (s, 1H, NH–cHex), 7.24–7.78 (m, 10H, CH_ar); ¹³C NMR $\delta_{\rm C}$ (ppm) 16.9, 18.4, 20.7, 21.8, 24.7, 26.0, 32.2 ($J_{\rm CP}$ 103.2 Hz), 35.7, 44.6, 51.8, 98.3 ($J_{\rm CP}$ 7.2 Hz), 127.3–137.1, 150.4; ³¹P NMR $\delta_{\rm P}$ (ppm) 37.3.

4.5. General procedure for the preparation of the derivatives $7a\!-\!d$

To a solution of potassium *tert*-butoxide (0.01 mol, 1.0 equiv) in anhydrous THF (10 mL) was added dropwise β -phosphorylketones **6a**,**b**¹³⁻¹⁶ (0.01 mol, 1.0 equiv) in solution in anhydrous THF (20 mL)

at room temperature, under nitrogen atmosphere. After 2 h of stirring, isothiocyanate (0.01 mol, 1.0 equiv) in solution in anhydrous THF (5 mL) was added to the reactional mixture. After 16 h of stirring at room temperature, the hydrolysis was performed with an aqueous saturated NH₄Cl solution. The aqueous layer was then extracted twice with CHCl₃ (2 x 50 mL). After drying (MgSO₄), filtration and solvent evaporation, the crude product was purified by column chromatography on silica gel (0.063–0.200 mm, eluent: 98/2 Et₂O/EtOH).

4.5.1. 2-(Diphenylphosphinoyl)-3-oxo-3, N-diphenyl thiopropionamide (**7a**). Solid; yield=87%; mp 184–186 °C; ¹H NMR $\delta_{\rm H}$ (ppm) 6.45 (d, 1H, ²J_{PH} 21.0 Hz, CHP(O)), 7.22–8.02 (m, 20H, CH_{ar}), 11.35 (s, 1H, NH); ¹³C NMR $\delta_{\rm C}$ (ppm) 67.3 (J_{CP} 44.5 Hz), 125.1–139.0, 186.4 (J_{CP} 6.0 Hz), 190.6 (J_{CP} 4.5 Hz); ³¹P NMR $\delta_{\rm P}$ (ppm) 28.3; HRMS calcd for C₂₇H₂₂NNaO₂PS (M+Na)⁺: 478.1001; found: 478.0994.

4.5.2. (2-Oxo-1-phenylthiocarbamoylpropyl) phosphonic acid diethyl ester (**7b**). Solid; yield=86%; mp 170–172 °C; ¹H NMR $\delta_{\rm H}$ (ppm) 1.35 (t, 6H, ³*J*_{HH} 6.0 Hz, *CH*₃CH₂O), 2.48 (s, 3H, CH₃), 4.22 (qp, 4H, ³*J*_{HH} 6.0 Hz, ³*J*_{PH} 6.0 Hz, CH₂O), 5.24 (d, 1H, ²*J*_{PH} 24.0 Hz, CHP(O)), 7.23–7.42 (m, 5H, CH_{ar}), 10.65 (s, 1H, NH); ¹³C NMR $\delta_{\rm C}$ (ppm) 16.2, 32.2, 63.3, 71.6 (*J*_{CP} 113.9 Hz), 123.0–139.1, 186.0 (*J*_{CP} 8.3 Hz), 198.9 (*J*_{CP} 5.3 Hz); ³¹P NMR $\delta_{\rm P}$ (ppm) 14.1.

4.5.3. 1-(Benzylthiocarbamoyl-2-oxopropyl) phosphonic acid diethyl ester (**7c**). Solid; yield=77%; mp 175–177 °C; ¹H NMR $\delta_{\rm H}$ (ppm) 1.28 (t, 6H, ³J_{HH} 6.0 Hz, *CH*₃CH₂O), 2.43 (s, 3H, CH₃), 4.17 (qp, 4H, ³J_{HH} 6.0 Hz, ³J_{PH} 6.0 Hz, CH₂O), 4.73–4.79 (m, 2H, CH₂N), 5.20 (d, 1H, ²J_{PH} 24.0 Hz, CHP(O)), 7.33–7.36 (m, 5H, CH_{ar}), 9.29 (s, 1H, NH); ¹³C NMR $\delta_{\rm C}$ (ppm) 16.1, 32.2, 50.7, 62.2, 68.4 (*J*_{CP} 86.7 Hz), 127.6–135.8, 188.0, 198.4 (*J*_{CP} 5.3 Hz); ³¹P NMR $\delta_{\rm P}$ (ppm) 14.4.

4.5.4. 1-(Cyclohexylthiocarbamoyl-2-oxopropyl) phosphonic acid diethyl ester (**7d**). Solid; yield=81%; mp 172–174 °C; ¹H NMR $\delta_{\rm H}$ (ppm) 1.37 (t, 6H, ³*J*_{HH} 6.0 Hz, CH₃CH₂O), 1.60–2.01 (m, 10H, CH₂), 2.43 (s, 3H, CH₃), 4.17 (qp, 4H, ³*J*_{HH} 6.0 Hz, ³*J*_{PH} 6.0 Hz, CH₂O), 4.30–4.36 (m, 1H, CHN), 5.11 (d, 1H, ²*J*_{PH} 24.0 Hz, CHP(O)), 9.02 (s, 1H, NH); ¹³C NMR $\delta_{\rm C}$ (ppm) 16.2, 24.1, 25.4, 30.7, 32.1, 54.7, 63.7, 69.0 (*J*_{CP} 114.3 Hz), 185.9 (*J*_{CP} 7.5 Hz), 198.4 (*J*_{CP} 5.3 Hz); ³¹P NMR $\delta_{\rm P}$ (ppm) 14.2.

4.6. General procedure for the preparation of the derivatives **9b**, **c** and **e**

Hydrazine (0.0011 mol, 1.1 equiv) was added to a solution of β ketothioamides **7a–d** (0.001 mol, 1.0 equiv) in ethanol. After 24 h of stirring at room temperature, the residue was dried (MgSO₄), solvent was evaporated and the crude product was purified by column chromatography on silica gel (0.063–0.200 mm, eluent: AcOEt).

4.6.1. 2-(Diphenylphosphinoyl)-N-phenylthio acetamide (**9b**). Gummy liquid; yield=77%; ¹H NMR $\delta_{\rm H}$ (ppm) 4.09 (d, 2H, ²J_{PH} 12.0 Hz, CH₂P), 7.12–7.81 (m, 15H, CH_{ar}), 11.47 (s, 1H, NH); ¹³C NMR $\delta_{\rm C}$ (ppm) 49.2 (J_{CP} 55.0 Hz), 123.2–139.2, 191.4 (J_{CP} 7.4 Hz); ³¹P NMR $\delta_{\rm P}$ (ppm) 33.1; HRMS calcd for C₂₀H₁₈NNaOPS (M+Na)⁺: 374.0739; found: 374.0742.

4.6.2. (Benzylthiocarbamoylmethyl)phosphonic acid diethyl ester (**9c**). Gummy liquid; yield=72%; ¹H NMR $\delta_{\rm H}$ (ppm) 1.18 (t, 6H, ³J_{HH} 6.0 Hz, *CH*₃CH₂O), 3.36 (d, 2H, ²J_{PH} 24.0 Hz, CH₂P), 3.96 (qp, 4H, ³J_{HH} 6.0 Hz, ³J_{PH} 6.0 Hz, CH₂O), 4.72–4.74 (m, 2H, CH₂N), 7.17–7.27 (m, 5H, CH_ar), 9.39 (s, 1H, NH); ¹³C NMR $\delta_{\rm C}$ (ppm) 15.3, 43.0 (*J*_{CP} 126.7 Hz), 49.3, 62.0, 124.4–135.2, 191.3 (*J*_{CP} 6.0 Hz); ³¹P NMR $\delta_{\rm P}$ (ppm) 21.1.

4.6.3. *Phenylthiocarbamoylmethylphosphonic acid diethyl ester* (**9***e*). Gummy liquid; yield=76%; ¹H NMR $\delta_{\rm H}$ (ppm) 1.34 (t, 6H, ³J_{HH} 6.0 Hz, *CH*₃CH₂O), 3.65 (d, 2H, ²J_{PH} 24.0 Hz, CH₂P), 4.19 (qp, 4H, ³J_{HH}

6.0 Hz, ${}^{3}J_{PH}$ 6.0 Hz, CH₂O), 7.11–7.72 (m, 5H, CH_{ar}), 11.09 (s, 1H, NH); ${}^{13}C$ NMR δ_{C} (ppm) 16.3, 46.3 (J_{CP} 126.7 Hz), 63.4, 122.8–139.2, 190.5 (J_{CP} 6.8 Hz); ${}^{31}P$ NMR δ_{P} (ppm) 21.2.

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Supplementary data

Supplementary data associated with this article can be found, in the online version, at doi:10.1016/j.tet.2011.09.093.

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