## Heavyweight "R-SMS-Phos" Ligands in the Olefins' Hydrogenation Arena

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## **ABSTRACT**

 $[Rh((R,R)-t-Bu-sms-phos)(MeOH)_2]BF_4$ 

A series of enantiopure P-stereogenic 1,2-bis[(o-RO-phenyl)(phenyl)phosphino]ethane (R-SMS-Phos) ligands wherein R= i-Pr, i-Bu, t-Bu, 3-Pen, and CH<sub>2</sub>TMS was assessed in the Rh(I)-catalyzed hydrogenation of an indicative set of olefins. The best performing t-Bu-SMS-Phos ligand was screened against a wide range of representative classes of standard and new olefinic substrates such as dehydroamido esters, dehydro- $\alpha$ -amido-phosphonates, enamides, itaconates, acrylates, enol acetates,  $\alpha$ -phosphonovinyl benzoates,  $\alpha$ -(2-pyridyl N-oxide)styrenes, and  $\alpha$ -(1-hydroxyliminoethyl)styrenes. Excellent enantioselectivities and high TOFs were attained under mild conditions.

Since the conception of the Rh(I)-DiPAMP (DiPAMP= 1,2-bis[(o-anisyl)(phenyl)phosphino]ethane) catalyst for the asymmetric hydrogenation of olefins more than four decades ago, stiff competition to attain higher catalyst performances is ever-increasing. In particular, P-stereogenic diphosphines have further advanced this field and are making a comeback. Innovation through optimization of chiral ligand designs with a proven track record is enticing for improved or even new industrial applications. A comprehensive survey of the literature reveals a number of study cases whereby diversification of a given diphosphine by judicious alterations was undertaken.<sup>2,3</sup>

In our ongoing research focus on P-stereogenic ligands,  $^{3n,o}$  we present herein our exploratory optimization results of our recently introduced 1,2-bis[(o-isopropoxyphenyl)(phenyl)phosphino]ethane (i-Pr-SMS-Phos) ligand  $^{3o}$  for the Rh(I)-mediated hydrogenation of olefins. Higher homologues at the level of the branched alkoxy groups were prepared.

Isolated in 50–55% overall yields following the Jugé-Stephan route<sup>4</sup> or following a straightforward functionalization—decomplexation sequence from the crystalline 1,2-bis[(o-hydroxyphenyl)(phenyl)phosphino-P-borane]ethane,<sup>5</sup> the R-SMS-Phos (1,2-bis[(o-RO-phenyl)(phenyl)phosphino]ethane) series (R= *i*-Pr, *i*-Bu, *t*-Bu, 3-Pen, and CH<sub>2</sub>TMS) was screened under mild conditions in the asymmetric

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**Table 1.**  $[Rh((R,R)-R-sms-phos)(MeOH)_2]BF_4$ -Catalyzed Hydrogenation of  $S1-S7^a$ 

.1.6		R= i-Pr		R=i	R= i-Bu		R=t-Bu		R= 3	R= 3-Pen		R= CH <sub>2</sub> TMS	
olefin		t, min	ee, %	t, min	ee, %		t, min	ee, %	t, min	ee, %	t, min	ee, %	
CO₂Me	<b>S1</b> (MAA)	6	99.4	6	99.4		4	99.9	5	99.8	6	99.7	
NHAc CO₂Me	<b>S2</b> (MAC)	4	99.7	4	99.7		2	99.8	4	99.4	4	99.7	
Ph NHAc AcNH CO₂Me	<b>S3</b> (Z-MAB)	20	82.1	30	70.2		7	80.1	15	84.4	30	71.4	
Me CO <sub>2</sub> Me	<b>S4</b> ( <i>E</i> -MAB)	90	93.0	90	93.4		60	97.3	90	94.9	120	95.1	
AcNH NHAc ────────────────────────────────────	<b>S5</b> (AS)	3	97.8	5	96.9		2	99.3	4	98.5	8	98.1	
$CO_2Me$ $CO_2Me$	<b>S6</b> (DMI)	5	98.1	5	98.5		2	99.8	4	98.6	7	99.1	
$\rightleftharpoons^{CO_2H}_{Ph}$	<b>S7</b> (AA) <sup>b</sup>	120	88.0	120 °	86.0		120	94.7	120	92.9	120 <sup>d</sup>	86.8	

<sup>a</sup> The catalyst was prepared in situ from [Rh(nbd)<sub>2</sub>]BF<sub>4</sub>. Runs were carried out under 1 bar of H<sub>2</sub> (10 bar for S7) at rt in MeOH (0.5 mmol of substrate in 7.5 mL MeOH) with a S/C = 100 (S/C = 1000 for S1) for the time indicated (100% conversion) if not stated otherwise and are unoptimized. Typical isolated yields were >90%. Ee's were determined by chiral GC (prior to analysis the carboxylic group of hydrogenation product of S7 was esterified with TMSCHN<sub>2</sub>). With (R,R)-R-SMS-Phos, S-configured products were obtained except with S6. <sup>b</sup> In the presence of Et<sub>3</sub>N (1.1 equiv). <sup>c</sup> 78% conversion. <sup>d</sup> 59% conversion.

hydrogenation of an indicative set of olefinic reference substrates S1-S7 (Table 1). Within the adopted systematic bulkiness modification of the R groups, valuable changes in reactivity and enantioselectivity of the Rh(I)-(R-SMS-Phos) catalysts were noticeable. Operating with a S/C 100 in methanol at rt under 1 bar of  $H_2$ , methyl  $\alpha$ -acetamidoacrylate (S1: MAA) and cinnamate (S2: MAC) were hydrogenated invariably with >99% ee's within minutes. However, the best hydrogenation results of methyl (Z)-3-acetamidobut-2-enoate (S3: (Z)-MAB) and its (E)-isomer (S4: (E)-MAB) were achieved with 3-Pen-SMS-Phos and t-Bu-SMS-Phos furnishing 84.4% and 97.3% ee, respectively. Further on, the hydrogenation of  $\alpha$ -acetamidostyrene (S5: AS) and dimethyl itaconate (S6: DMI) also proceeded smoothly within minutes with an incremental

increase in the ee, reaching, respectively, the maxima of 99.3% and 99.8% with *t*-Bu-SMS-Phos. Interestingly enough, the bulkiest 3-Pen-SMS-Phos and *t*-Bu-SMS-Phos designs afforded hydratropic acid from atropic acid (**S7**: AA) with reasonably good ee's, with up to 94.7% ee being attained with *t*-Bu-SMS-Phos. Hydratropic acid constitutes the basic model of nonsteroidal antiarthritics. Thus, among the screened ligand set, the *t*-Bu-SMS-Phos ligand emerged as being superior. Hence, the Rh(I)-(*t*-Bu-SMS-Phos) catalyst was screened under mild hydrogenation conditions against a selection of a broad diversity of conventional, more challenging benchmark and new classes of olefins **S8**–**S19** (Table 2).

High reaction rates coupled with excellent ee's were reached in the hydrogenation of virtually all of the considered various olefin groups.<sup>7</sup> In particular, the representative standard test substrate MAC (**S2**) was hydrogenated (100% conversion) in 99.8% ee within 5.5 h using a S/C 30000.  $\beta$ , $\beta$ -Disubstituted dehydro-(N-acetylalninates (**S8** and **S9**) were hydrogenated equally well under 3 bar of H<sub>2</sub> in >99% ee within 2 h using a S/C

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Table 2. [Rh((R,R)-t-Bu-sms-phos)(MeOH)<sub>2</sub>]BF<sub>4</sub>-Catalyzed Hydrogenation of Miscellaneous Classes of Olefins<sup>a</sup>

olefin		S/C	t, min	ee, % (config.)	_	olefin		S/C	t, min	ee, % (config.)
CO <sub>2</sub> Me NHAc	S2	1000 10000 30000	13 3 h 5.5 h	99.8 (S) 99.8 99.8	-	$=$ $CO_2H$ $CONH_2$	S13	1000	5	99.9 (R)
CO <sub>2</sub> Me NHAc	S8	1000	2 h	99.2 (S)		$\stackrel{CO_2H}{=\!\!\!\!=\!\!\!\!=\!\!\!\!=\!\!\!\!=\!\!\!\!\!\!\!\!\!\!\!\!\!\!\!$	<b>S</b> 7	$100^{\ b}$ $100^{\ c}$	2 h 2 h	94.7 (S) 95.6
CO <sub>2</sub> Me NHAc	<b>S9</b>	1000	2 h	99.6 (S)		→OAc Ph	S14	100 1000	30 5 h	98.7 (S) 98.7
PO(OEt) <sub>2</sub> NHAc	S10	1000	6	99.9 (R)		$\stackrel{OAc}{=\!\!\!\!=\!\!\!\!=\!\!\!\!\!=}}$ $CF_3$	S15	100 1000	10 90	>99 (S) >99
$=$ $CO_2Et$ $NPhth$	S11	100	30	94.6 (R)		CO <sub>2</sub> Et Ph OAc	<b>S16</b>	100	2 h	99.9 (S)
⇒(NHAc Ph	S5	1000 10000	20 4 h	99.4 (S) 99.3		$=$ $\begin{array}{c} PO(OMe)_2 \\ OBz \end{array}$	S17	100 1000	3 25	99.6 (R) 99.6
$=$ $CO_2Me$ $CO_2Me$	S6	1000 10000 30000	15 3 h 6 h	99.7 ( <i>R</i> ) 99.4 99.4		0+N Ph	S18	100 1000	7 60	99.2 (S) 99.2
$=$ $CO_2H$ $CO_2H$	S12	1000	6	99.7 (R)		= $N$ OH	S19	100	4 h	94.0 ( <i>Z</i> , +)

"The catalyst was prepared in situ from [Rh(nbd)<sub>2</sub>]BF<sub>4</sub> and (R,R)-t-Bu-SMS-Phos. Runs were carried out under 1 bar of H<sub>2</sub> (3 bar for **S8**, **S9** and 10 bar for **S7**, **S19**) at rt (50 °C for **S18**) in MeOH (0.5 mmol of substrate in 7.5 mL MeOH with a S/C = 100 or 1000; 10 mmol of substrate in 7.5 mL MeOH with a S/C = 10000; 30 mmol of substrate in 20 mL MeOH with a S/C = 30000) for the time indicated (100% conversion) and are unoptimized. Typical isolated yields were >90%. Ee's were determined by: chiral GC for **S2**, **S5-S10**, and **S12-S14** (prior to analysis the carboxylic groups of hydrogenation products of **S7**, **S12**, and **S13** were esterified with TMSCHN<sub>2</sub>); chiral HPLC for **S11** and **S16-S19**; <sup>1</sup>H NMR (in the presence of (+)-Pr(hfc)<sub>3</sub>) for hydrogenation product of **S15**.  $^b$  In the presence of Et<sub>3</sub>N (0.05 equiv).  $^c$  In the presence of Cy<sub>2</sub>NH (0.05 equiv).

1000, and  $\alpha$ -acetamido-vinylphosphonate (**S10**) led to 99.9% ee within minutes. The latter result presents the highest ee ever reported for the hydrogenation of the corresponding substrate. Up to 94.6% ee was achieved in the hydrogenation of ethyl  $\alpha$ -(phthalimidomethyl)acrylate (**S11**) which also constitutes the highest ee attained with this substrate under the given reaction conditions. Moreover, AS (**S5**) underwent hydrogenation using a S/C 10000 affording 99.3% ee within 4 h.

While a series of itaconates (**S6**, **S12**, **S13**) was hydrogenated with >99.7% ee within minutes using a S/C 1000, >99% ee was maintained with full conversion within 6 h for DMI (**S6**) using a S/C 30000.

A preliminary investigation on the variation of the reaction parameters toward hydratropic acid revealed that an incremental increase in the ee was feasible. Thus, the use of the bulkier  $\text{Cy}_2\text{NH}$  amine (5 mol %) further upgraded the ee to 95.6%.

In a similar vein, a variety of enol acetates (S14-S16) and a  $\alpha$ -benzoyloxy-vinylphosphonate (S17) were hydrogenated reasonably fast with exceptionally high ee's.

Namely,  $\alpha$ -acetoxystyrene (**S14**), which presents a somewhat difficult challenge, was hydrogenated in up to 98.7% ee within 5 h using a S/C 1000. Here again, these results obtained under the given mild condtions are to the best of our knowledge the highest reported ones with these substrates. <sup>3e,m,o,9</sup>

Finally, olefins **S18** and **S19**,  $^{10}$  which possess a "C-N-O" motif at the  $\alpha$ -position, were hydrogenated in 99.2% and 94.0% ee, respectively.

In conclusion, the hydrogenation under mild conditions with excellent ee's and high TOFs of a wide spectrum of representative classes of olefins catalyzed by [Rh(*t*-Bu-sms-phos)(MeOH)<sub>2</sub>] catalyst represents the advantages of this novel catalytic system. The overall results obtained with this catalyst are among the best ever reported in Rh-phosphine catalyzed hydrogenation.

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**Supporting Information Available:** Experimental procedures and characterization data for all new compounds. This material is available free of charge via the Internet at http://pubs.acs.org.

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