

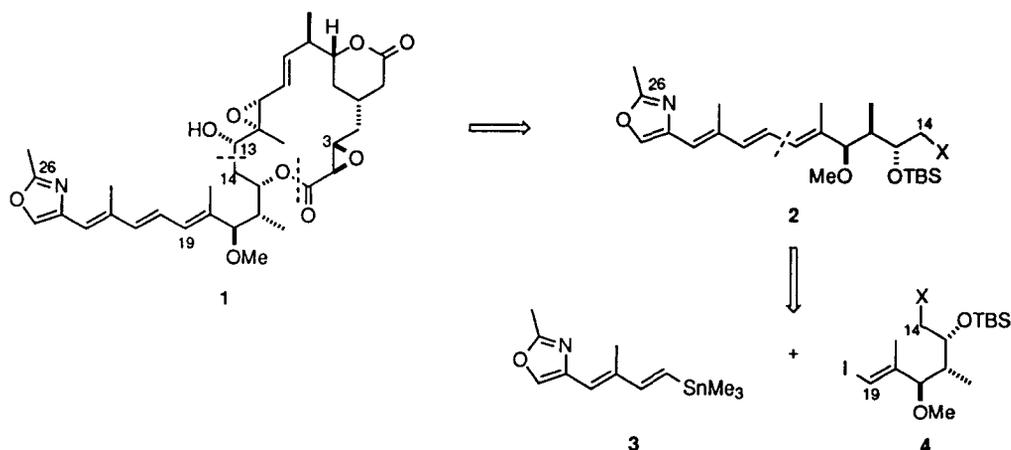
Studies on the Total Synthesis of the Macrolide Antitumor Agent Rhizoxin. 2. Synthesis of the C14-C26 Segment.

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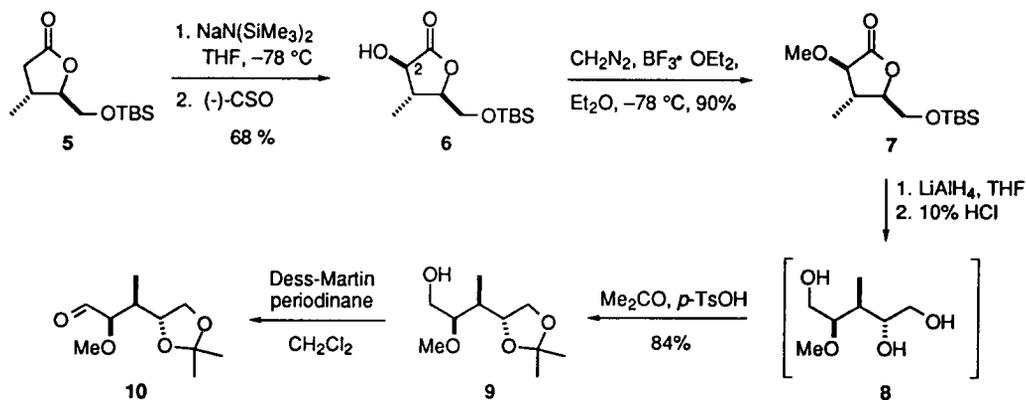
Abstract: An asymmetric synthesis of the C14-C26 segment of rhizoxin is described in which the three stereogenic centers are derived from a γ -lactone; stannylcupration-methylation of a terminal alkyne is used to generate an (*E*)-iodoalkene for Stille coupling with a dienylstannane that produces the conjugated (*E,E,E*)-triene unit of rhizoxin. © 1997 Elsevier Science Ltd.

The promising antitumor agent rhizoxin (**1**)¹ has been the subject of much synthetic interest.²⁻⁴ In the preceding Letter we outlined a convergent approach to the synthesis of this 16-membered macrolide and we described a stereocontrolled route to a segment representing C3-C13 of **1**.⁵ Herein, we report an efficient synthesis of the C14-C26 portion of rhizoxin, to which the previously prepared subunit will be attached. Construction of the trienyloxazole moiety **2** was envisioned through Stille coupling⁶ of the vinyltin species **3** with vinyl iodide **4**, a strategy for which good precedent already exists.³

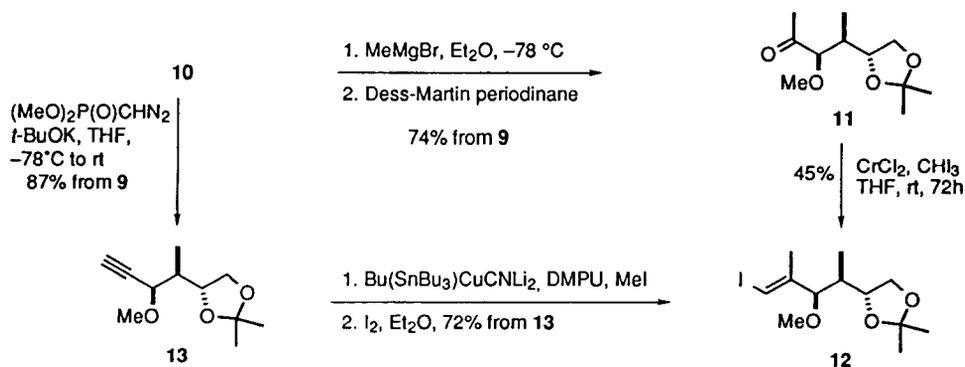


Synthesis of the C14-C19 subunit **4** began from the known γ -lactone **5**,⁷ readily prepared from D-glutamic acid.⁸ The functionality and stereochemistry present in **5** makes this template ideally suited to construction of the three contiguous asymmetric centers of **4** by stereocontrolled hydroxylation. Thus,

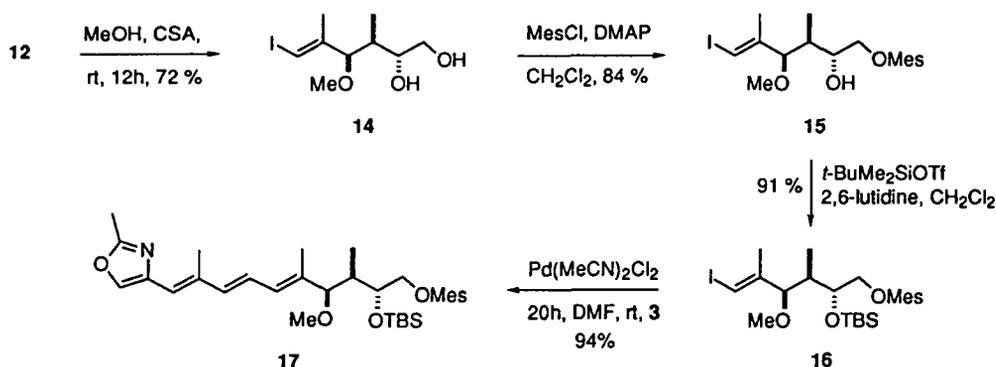
exposure of the *sodium* enolate of **5** to (1*R*)-(-)-(10-camphorsulfonyl)oxaziridine (CSO)⁹ afforded (2*R*) alcohol **6** in 30:1 excess over the (2*S*) stereoisomer.¹⁰ Methylation of this sterically hindered alcohol proved difficult but was accomplished by successive addition of diazomethane to **6** in the presence of boron trifluoride. The resultant lactone **7** was reduced to the water-soluble triol **8** which was converted without purification to its acetonide **9**. The latter underwent oxidation with Dess-Martin periodinane¹¹ to aldehyde **10**.



Our initial plan for elaboration of the trisubstituted alkene moiety of **4** required ketone **11**, which was readily prepared by a Grignard reaction of **10** with methylmagnesium bromide, followed by Dess-Martin oxidation of the resulting secondary alcohol. However, a Takai reaction¹² of **11** with iodoform in the presence of chromous chloride furnished a low yield of iodoalkenes as a 2:1 mixture of **12** and its (*Z*) isomer, respectively. Fortunately, an improved and completely stereoselective pathway to **12** was realized via the alkyne **13**, prepared from **10** by reaction with dimethyl diazomethylphosphonate.¹³ Although zirconium-catalyzed carboalumination-iodination of **13** under Negishi's conditions¹⁴ was unsuccessful, stannylcupration-methylation¹⁵ of this alkyne along lines reported by Kocienski,¹⁶ followed by iodination, furnished pure **12** in good yield.



Selective functionalization of the primary alcohol terminus of diol **14**, obtained after acidic methanolysis of **12**, could not be accomplished by tosylation. However, this selectivity was readily achieved by reaction of **14** with the more sterically demanding mesitylenesulfonyl (Mes) chloride, and the resultant alcohol **15** was protected as its silyl ether **16**. Stille coupling of **16** with the known stannane **3**,³ prepared by reaction of the corresponding iodoalkene with (Me₃Sn)₂ in the presence of Pd(PPh₃)₂Cl₂ as catalyst, afforded the (*E,E,E*) triene **17**¹⁷ in excellent yield. This material is now available in quantity for coupling with the C3-C13 portion of rhizoxin already in hand.



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References

1. a) Iwasaki, S.; Kobayashi, H.; Furukawa, J.; Namikoshi, M.; Okuda, S.; Sato, Z.; Matsuda, I.; Noda, T. *J. Antibiot.* **1984**, *37*, 354. b) Iwasaki, S.; Namikoshi, M.; Kobayashi, H.; Furukawa, J.; Okuda, S.; Itai, A.; Kasuya, A.; Iitaka, Y.; Sato, Z. *J. Antibiot.* **1986**, *39*, 424. c) Hendriks, H. R.; Plowman, J.; Berger, D.P.; Paull, K.D.; Fiebig, H.H.; Fodstad, O.; Dreef-van der Meulen, H.C.; Henrar, R.E.C.; Pinedo, H.M.; Schwartzmann, G. *Ann. Oncol.* **1992**, *3*, 755.
2. For the total synthesis of rhizoxin, see a) Nakada, M.; Kobayashi, S.; Iwasaki, S.; Ohno, M. *Tetrahedron Lett.* **1993**, *34*, 1035. b) Nakada, M.; Kobayashi, S.; Shibasaki, M.; Iwasaki, S.; Ohno, M. *Tetrahedron Lett.* **1993**, *34*, 1039.
3. For the total synthesis of didesepoxyrhizoxin, a naturally occurring congener of **1**, see: Kende, A.S.; Blass, B.E.; Henry, J.R. *Tetrahedron Lett.* **1995**, *36*, 4741.
4. For approaches to rhizoxin, see a) Keck, G.E.; Savin, K.A.; Weglarz, M.A.; Cressman, E.N.K. *Tetrahedron Lett.* **1996**, *37*, 3291. b) Lafontaine, J.A.; Leahy, J.W. *Tetrahedron Lett.* **1995**, *36*, 6029. c) Provencal, D.P.; Gardelli, C.; Lafontaine, J.A.; Leahy, J.W. *Tetrahedron Lett.* **1995**, *36*, 6033. d)

- Nakada, M.J. *Synth. Org. Chem. Jpn.* **1995**, *53*, 122. e) Rao, S. P.; Murthy, V.S.; Rama Rao, A.V. *Indian J. Chem. Sect. B* **1994**, *33*, 820. f) Keck, G.E.; Park, M.; Krishnamurthy, D. *J. Org. Chem.* **1993**, *58*, 3787. g) Rama Rao, A.V.; Bhanu, M.N.; Sharma, G.V.M. *Tetrahedron Lett.* **1993**, *34*, 707. h) Rama Rao, A.V.; Sharma, G.V.M.; Bhanu, M.N. *Tetrahedron Lett.* **1992**, *33*, 3907. i) Boger, D.L.; Curran, T.T. *J. Org. Chem.* **1992**, *57*, 2235.
- White, J.D.; Nylund, C.S.; Green, N.J. *Tetrahedron Lett.* **1997**, *38*, 7329.
 - Farina, V.; Krishnamurthy, V.; Scott, W.J. *Org. React.* **1997**, *50*, 1.
 - Mitome, H.; Miyaoka, H.; Nakano, M.; Yamada, Y. *Tetrahedron Lett.* **1995**, *36*, 8231.
 - a) Ravid, U.; Silverstein, R.M.; Smith, L.R. *Tetrahedron* **1978**, *34*, 1449. b) Hanessian, S.; Murray, P. J. *Tetrahedron* **1987**, *43*, 5055.
 - Davis, F. A.; Chen, B. *Chem. Rev.* **1992**, *92*, 919.
 - By contrast, the potassium enolate of **5** with CSO gave a (2*R*:2*S*) ratio of 14:1. Poor stereoselectivity was observed in hydroxylation of **5** with MoOPH. Assignment of configuration to **6** was made from the coupling constant (10.2 Hz) between H2 and H3.
 - Dess, D. B.; Martin, J. C. J. *Am. Chem. Soc.* **1991**, *113*, 7277.
 - Takai, K.; Nitta, K.; Utimoto, K. *J. Am. Chem. Soc.* **1986**, *108*, 7408.
 - Gilbert, J. C.; Weerasooriya, U. *J. Org. Chem.* **1979**, *44*, 4997.
 - Rand, C.L.; van Horn, D.E.; Moore, M.W.; Negishi, E. *J. Org. Chem.* **1981**, *46*, 4096.
 - Lipshutz, B. H.; Ellsworth, E. L.; Dimock, S. H.; Reuter, D. C. *Tetrahedron Lett.* **1989**, *30*, 2065.
 - Harris, L.; Jarowicki, K.; Kocienski, P.; Bell, R. *Synlett*, **1996**, 903. We are grateful to Professor Kocienski for providing experimental details.
 - Spectral data for compound **2**: ¹H NMR (CDCl₃, 300 MHz): δ 7.57 (s, 1H), 6.96 (s, 2H), 6.59 (dd, J = 10.8, 15 Hz, 1H), 6.37 (d, J = 15.3 Hz, 1H), 6.30 (s, 1H), 6.03 (d, J = 10.6 Hz, 1H), 3.85-4.05 (m, 2H), 3.80 (m, 1H), 3.50 (m, 1H), 3.16 (s, 3H), 2.61 (s, 6H), 2.53 (s, 3H), 2.32 (s, 3H), 2.15 (s, 3H), 1.87 (m, 1H), 1.73 (s, 3H), 0.88 (m, 12H), 0.05 (s, 3H), 0.0 (s, 3H). ¹³C NMR (CDCl₃, 100 MHz): δ 160.93, 143.09, 139.93, 138.88, 137.44, 137.01, 135.93, 135.71, 131.70, 130.85, 128.03, 124.09, 120.43, 86.66, 71.92, 70.64, 56.47, 41.09, 29.72, 25.82, 22.64, 21.05, 18.04, 14.43, 13.86, 13.27, 9.49, -4.34, -4.97.

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