

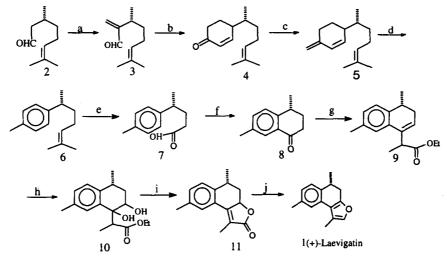
Enantiospecific total synthesis of (+)-laevigatin

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Abstract: The first enantiospecific, total synthesis of (+)-laevigatin from (+)-citronellal is described. © 1997 Published by Elsevier Science Ltd

Laevigatin 1, a naturally occurring terpene having an unusual skeleton, was isolated in optically active form from *Eupatorium laevigatum*¹ So far there have been only two syntheses of (\pm) -1 published,² but there has been no report of a synthesis of enantiomerically pure laevigatin. As a part of our interest in synthesis of Heritol, Heritonin and related compounds, two convenient and efficient methodologies to generate butenolides have been developed: (1) *via* osmylation of of β , γ -unsaturated esters³ and (2) direct oxidative conversion of β , γ -unsaturated acids to butenolides mediated by CAN at room temperature.⁴ Earlier^{2c} based on this strategy we have synthesized (±)-1.



a: HCHO, Piperdine acetate reflux b: Methylacetoacetate, MeOH, MeONa, reflux c: (Ph)₃PCH₃I, BuLi, THF 0°C d: Sulfur, DMF reflux e: OsO₄ (cat), Jones reagent, R.T. f: TFA, TFAA, 0°C g: ethylbromopropionate, Zn, H⁺ h: OsO₄, NMO, CH₃CN i: TsOH, C₆H₆, reflux j: DIBAL-H, THF, -40°C

We now wish to report the first enantiospecific synthesis of (+)-1, using the above protocol for the synthesis of the butenolide and its conversion into naturally occurring laevigatin. The key intermediate, optically active tetralone 8 was obtained from commercially readily available citronellal 2. (+)-Citronellal 2 was converted to enone 4 following a reported procedure.⁵ Wittig methylenation⁶ of 4 gave triene 5 in good yield (80%). The aromatization⁷ of 5 was achieved by refluxing it in DMF in the presence of sulfur to furnish the aromatic compound 6 in 70% yield. One pot oxidative cleavage of the double bond⁸ via the corresponding diol to acid 7 was achieved in 84% yield. Cyclisation of acid 7 by trifluoroacetic anhydride⁹ furnished enantiomerically pure tetralone 8 (80%).

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Having achieved the synthesis of chiral tetralone, the next task was the straightforward transformation of **8** to butenolide **11** by the protocol developed in this laboratory and its further conversion to (+)-laevigatin **1**. Accordingly, the Reformatsky reaction³ on tetralone **8** afforded the β , γ -unsaturated ester **9** in 78% yield. Dihydroxylation of **9** furnished the corresponding diol **10** as a mixture of diastereoisomers in 80% yield. Since the stereochemistry at the newly created centres was of no consequence as far the synthesis of (+)-laevigatin is concerned as they would be destroyed in the subsequent steps, no attempt was made to separate the isomers. The acid catalyzed cyclization of the diol **10** with a catalytic amount of p-TsOH in refluxing benzene furnished the corresponding butenolide **11** in 73% yield. Reduction of butenolide **11** with DIBAL-H¹⁰ in THF at -40°C provided (+)-laevigatin **1** in 77% yield. The spectroscopic and physical properties of (+)-laevigatin thus obtained were found to be identical in all respects with the reported values for (+)-1 obtained from natural sources. m.p. 66°C, lit¹ 65–66°C, specific rotation $[\alpha]_D=+89$ (c=2.3, CHCl₃). Lit.¹ $[\alpha]_D=+88$.

Thus, the first enantiospecific synthesis of (+)-laevigatin has been achieved from readily available (+)-citronellal. All reaction steps are very simple, mild and proceed with excellent yields and are easy to perform.

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References

- (a) Bohlmann. F. and Zdero, C. Chem. Ber. 1977, 110, 4687 (b) Filho, B.; Bacha, C. T. M.; Bauer, L.; de Silva, G. A. A. B. and Sequira, N. C. S. Tetrahedron Lett. 1978, 2653.
- (a)Kano, S.; Ebata, J. and Shibuya, S. *Heterocycles*, **1980**, *14*, 43. (b) Juo, R. R. and Herz, W., J. Org. Chem., **1985**, 50, 700. (c) Chavan, S. P.; Rao. Y. T. S.; Govande, C. A. and Zubaidha P. K. IUPAC, Bangalore, 1994.
- 3. Chavan S. P.; Zubaidha, P. K. and Ayyangar, N. R. Tetrahedron Lett. 1992, 33, 4605.
- 4. Chavan S. P.; Zubaidha, P. K.; Govande, C. A. and Rao, Y. T. S. J. Chem. Soc. Chem. Commun., 1994, 1101.
- 5. Takano, S.; Inomata, K.; Samizu, K.; Tomita, S.; Yanase, M.; Suzuki, M.; Iwabuchi, Y.; Sugihara, T. and Ogasawara, K. Chem. Lett. 1989, 1283.
- 6. Sondheimer, F. and Mechnoulam, R. J. Am. Chem. Soc. 1957, 79, 5029.
- 7. Wynberg, J. J. Am. Chem. Soc., 1958, 80, 364.
- 8. James, R. H. and Weinreb, S. M. J. Org. Chem. 1993, 58, 4745.
- 9. Bourne, E. J.; Stacy, M.; Tattlow, J. C and Teddar, J. M. J. Chem. Soc. 1951, 718.
- 10. Kitagawa, I.; Tsujii, S.; Nishikawa, F. and Shibuya, H. Chem. Pharm. Bull. 1983, 31, 2639.

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