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Highly Efficient Construction of Trifluoromethylated Heterocycles; [3+2] Annulation of N,N'-Cyclic or C,N-Cyclic Azomethine Imines with Trifluoromethyl-Containing Electron-Deficient Olefins

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A highly regio- and stereoselective synthesis of pyrazolidine analogues by the use of N,N'-cyclic or C,N-cyclic azomethine imines with two different trifluoromethyl-containing ole-fins has been developed. The method affords highly func-

Pyrazolidine is an important structural motif that exists in a broad range of biologically and pharmaceutically relevant compounds.^[1] For example, bicyclic pyrazolidinones (LY) have potential use as antibiotics^[2a,2b,2c] and their isoquinoline derivatives show antimicrobial activities (Figure 1).^[2d] Synthetic methods to access these molecules include [3+2] cycloaddition reactions of azomethine imines, diazoalkanes, and nitrile imines with activated alkenes.^[3] Among those, azomethine imines, as a class of 1,3-dipoles, have been widely used in [3+2] annulations with alkenes or alkynes,^[4] and some effective asymmetric variants by using different chiral catalysts have been also reported.^[5]

Recently, significant research efforts have focused on the strategic introduction of trifluoromethyl groups into druglike molecules in medicinal and agricultural chemistry due to the unique physical properties of this group.^[6] There are two general methods to achieve trifluoromethyl-containing compounds: one is direct introduction of the trifluoromethyl group through nucleophilic, electrophilic, or radical reactants; the second uses a building-block strategy involving the assembly of readily accessible trifluoromethyl-containing organic compounds as starting materials.^[7–9] Both approaches have their own particular advantages; whereas the latter approach is well-suited for the valorization of readily available trifluoromethyl-containing building blocks, 2,2,2-trifluoroethylidene malonates are good substrates to contionalized trifluoromethyl-containing pyrazolidine analogues in excellent yields with high diastereoselectivities under mild conditions.



Figure 1. Examples of biologically active pyrazolidine and pyrazoline derivatives.

struct stereogenic tertiary carbon centers bearing a trifluoromethyl group. In this research field, Lu and coworkers applied 2,2,2-trifluoroethylidene malonates to asymmetric Michael addition and Friedel–Crafts alkylation to give the desired products in good yields.^[10] Fang and coworkers have synthesized aryl-substituted α -trifluoromethyl- α , β -unsaturated carboxylic acids (AFUCA), which are one of the most valuable building blocks for the synthesis of CF₃-containing molecules.^[11]

The annulation of azomethine imines with trifluoromethyl-containing olefins is fascinating because it not only provides a method with which to construct five-membered heterocycles, but also leads to the generation of products containing the trifluoromethyl group. Great efforts have focused on [3+2] cycloaddition of azomethine imines with different olefins,^[4,5] however, the reactivities of azomethine

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imines with trifluoromethyl-containing olefins has rarely been reported. Only Shibata's group reported one example of [3+2] annulation using N,N'-cyclic azomethine imines with α -trifluoromethyl acrylate (MAF-TBE),^[12a] giving the products in 79–97% yields along with between 59:41 and 75:25 diastereoselectivities.^[12b]

In this paper, we wish to report a highly regio- and stereoselective synthesis of trifluoromethyl-containing pyrazolidine analogues by the use of N,N'-cyclic or C,N-cyclic azomethine imines with two different trifluoromethyl-containing olefins under mild conditions.^[13]

Initial examination was carried out using N,N'-cyclic azomethine imine 1a (0.2 mmol) and trifluoroethylidene malonate 2a (0.24 mmol) as the substrates in tetrahydrofuran (THF) at room temperature for 24 h (Table 1, entry 1). We found that the desired cycloadduct 3a was obtained in 55% yield and more than 99:1 diastereoselectivity. Subsequently, we attempted to optimize the reaction conditions by screening several solvents; the results are summarized in Table 1 (entries 2-6). As can be seen, in acetonitrile (CH₃CN), the desired product **3a** could be obtained in 65%vield with 99:1 diastereoselectivity, and in toluene or ethyl acetate (EtOAc), 3a was formed in 60 and 80% yields, respectively, with excellent diastereoselectivities (Table 1, entries 3 and 4). Using 1,2-dichloroethane (DCE) or dichloromethane (CH₂Cl₂) as solvent, the yield of 3a increased to 92 and more than 99%, respectively, also with excellent diastereoselectivities (Table 1, entries 5 and 6). Thus, CH₂Cl₂ was found to be the most suitable solvent for this reaction. Lowering the reaction temperature to 0 °C or -20 °C in CH₂Cl₂ led to a significant decrease in the yield of 3a to 30% or barely no product, respectively, even upon prolonging the reaction time to 48 h (Table 1, entries 7 and 8). Upon reducing the reaction time to 2 h, the reaction also proceeded smoothly to give 3a in more than 99% yield

Table 1. Optimization of the reaction conditions for [3+2] annulation.

	∋ MeO₂C + ≏Ph a	CO ₂ Me CF ₃ 2a	solvent, T Ph CF_3 CO_2Me				
Entry ^[a]	Solvent	<i>T</i> (°C)	<i>t</i> [h]	$dr^{[b]}$	Yield [%] ^[c]		
1	THF	r.t.	24	>99:1	55		
2	CH ₃ CN	r.t.	24	>99:1	65		
3	toluene	r.t.	24	>99:1	60		
4	EtOAc	r.t.	24	>99:1	80		
5	DCE	r.t.	24	>99:1	92		
6	CH_2Cl_2	r.t.	24	>99:1	>99		
7	CH_2Cl_2	0	48	>99:1	30		
8	CH_2Cl_2	-20	48	_	trace		
9	CH ₂ Cl ₂	rt	2	>99.1	>99		

[a] Reaction conditions: **1a** (0.2 mmol), **2a** (0.24 mmol), solvent (1.0 mL). [b] Determined by ¹H and ¹⁹F NMR spectroscopic analysis. [c] Isolated yield.

(Table 1, entry 9). The relative configuration of **3a** was assigned on the basis of X-ray diffraction studies. The OR-TEP drawing is shown in Figure 2 and the CIF data are summarized in the Supporting Information.



Figure 2. ORTEP drawing of product 3a.

With the identification of the optimal reaction conditions, the generality of this catalyst-free [3+2] cycloaddition was examined by using a variety of azomethine imines 1 and trifluoroethylidene malonates 2; the results are summarized in Table 2. All reactions proceeded smoothly to give the corresponding products 3 in good yields and moderate to excellent diastereoselectivities under the optimal conditions (Table 2). Excellent yields and diastereoselectivities were obtained when utilizing diethyl 2-(2,2,2-trifluoroethylidene)malonate (2b) or dimethyl 2-(2,2,3,3,3-pentafluoropropylidene)malonate (2c) instead of 2-(2,2,2-trifluoroethylidene)malonate (2a; Table 2, entries 2 and 3). Regardless of whether R¹ was an electron-rich or -deficient aromatic ring, the reactions proceeded smoothly to give the corresponding bicyclic pyrazolidinone products 3d-g in good yields and good dr values (Table 2, entries 4-7). Only in the case of p- BrC_6H_4 azomethine imine 1b, was the corresponding adduct 3d obtained in a slightly lower yield (87%), but still with more than 99:1 dr; in this case the reaction time was lengthened to 24 h because 1b is not very soluble in CH₂Cl₂ (Table 2, entry 4). Multisubstituted azomethine imine 1f was also investigated in this reaction, affording the corresponding product **3h** in more than 99:1 dr and 89% yield (Table 2, entry 8). When \mathbb{R}^1 was a heteroaromatic group (\mathbb{R}^1 = 2-furan) or a sterically more bulky 2-naphthalene moiety, the reactions also proceeded efficiently to afford the corresponding products 3i-j in 77 and 82% yields with high dr values, respectively (Table 2, entries 9 and 10). Changing the aromatic group to an aliphatic group provided the corresponding product 3k in 94% yield with >99:1 dr (Table 2, entry 11). When R^2 was a methyl group, the desired bicyclic product 31 was obtained in 88% yield with 2:1 dr (Table 2, entry 12). When R³ was a methyl group, almost no reaction occurred (Table 2, entry 13).

		$ \begin{array}{c} $	R ⁴ 0 ⊦	² C CO ₂ R ⁴ R ^f 2		<i>I</i> , r.t., 2 h ►	R ² R ³ R ³ 3	$ \begin{array}{c} O \\ N \\ - N \\ - N \\ - CO_2 R^4 \\ R^1 \end{array} $		
Entry ^[a]	1	P 1	-	- 2	2	- 1	D ²	<i>dr</i> ^[b]	Product	Yield [%] ^[c]
		R	R^2	R ³		R4	R			
1	1a	Ph	Н	Н	2a	Me	CF ₃	>99:1	3a	>99
2	1a	Ph	Н	Н	2b	Et	CF_3	>99:1	3b	96
3	1a	Ph	Н	Н	2c	Et	CF_2CF_3	>99:1	3c	90
4 ^[d]	1b	$4-BrC_6H_4$	Н	Н	2a	Me	CF ₃	>99:1	3d	87
5	1c	$4-FC_6H_4$	Н	Н	2a	Me	CF_3	>99:1	3e	98
6	1d	$4-CH_3C_6H_4$	Н	Н	2a	Me	CF_3	>99:1	3f	>99
7	1e	$4-CH_3OC_6H_4$	Н	Н	2a	Me	CF_3	>99:1	3g	92
8	1f	3,4,5-(CH ₃ O) ₃ C ₆ H ₄	Н	Н	2a	Me	CF_3	>99:1	3h	89
9 ^[d]	1g	2-furyl	Н	Н	2a	Me	CF_3	>99:1	3i	77
10	1h	2-naphthyl	Н	Н	2a	Me	CF_3	>99:1	3j	82
11	1i	n-Octyl	Н	Н	2a	Me	CF_3	>99:1	3k	94
12	1j	Ph	CH_3	Н	2a	Me	CF ₃	2:1	31	88
13	1k	Ph	Н	CH ₃	2a	Me	CF_3	_	_	trace

Table 2. Substrate scope of [3+2] annulation of azomethine imine 1 with trifluoroethylidene malonate 2.

[a] Reaction conditions: 1 (0.2 mmol), 2 (0.3 mmol), CH_2Cl_2 (2.0 mL), r.t., 2 h. [b] Determined by ¹⁹F NMR spectroscopic analysis. [c] Isolated yield. [d] The reaction was carried out for 24 h and monitored by TLC.

Encouraged by the [3+2] annulation of N,N'-cyclic azomethine imine with trifluoroethylidene malonate, we next attempted to use C,N-cyclic azomethine imine **4** instead of N,N'-cyclic azomethine imine **1** to examine the reaction outcome. We found that the reactions also proceeded smoothly to give the annulation products in high yields and high diastereoselectivities; the results are summarized in Table 3. The reaction was tolerant of either electron-deficient or -rich aromatic rings on the azomethine imines **4a**– **c** ($\mathbb{R}^5 = \mathbb{H}$, Br, or Me) with various trifluoroethylidene malonates **2a–c**, providing a series of the corresponding products **5a–g** in 81–99% yields with excellent diastereoselectivities (Table 3, entries 1–7). The relative configuration of **5a**

Table 3. Substrate scope of [3+2] annulation of azomethine imine 4 with trifluoroethylidene malonate 2.

$\mathbb{R}^{5} \xrightarrow{(\mathbb{R}^{4})} \mathbb{N}_{NBz}^{\mathbb{R}^{4}} \xrightarrow{\mathbb{R}^{4}O_{2}C} \xrightarrow{(\mathbb{C}O_{2}\mathbb{R}^{4})} \mathbb{C}_{N, r.t., r.t.,$								
	4				2		R	5 5
Entry ^[a]	4	\mathbb{R}^5	2	\mathbb{R}^4	\mathbf{R}^{f}	dr ^[b]	Product	Yield [%] ^[c]
1	4a	Н	2a	Me	CF ₃	>99:1	5a	91
2	4a	Н	2b	Et	CF ₃	>99:1	5b	82
3	4a	Η	2c	Me	CF_2CF_3	>99:1	5c	96
4 ^[d]	4b	Br	2a	Me	CF ₃	>99:1	5d	94
5 ^[d]	4b	Br	2b	Et	CF ₃	>99:1	5e	81
6 ^[d]	4b	Br	2c	Me	CF_2CF_3	>99:1	5 f	>99
7	4c	CH_3	2a	Me	CF ₃	>99:1	5g	>99

[a] Reaction conditions: **4** (0.2 mmol), **2** (0.24 mmol), CH_2Cl_2 (2.0 mL), r.t., 2 h. [b] Determined by ¹⁹F NMR spectroscopic analysis. [c] Isolated yield. [d] The reaction was carried out in 4–8 h and monitored by TLC.

was assigned on the basis of X-ray diffraction studies. The ORTEP drawing of **5a** is shown in Figure 3 and the CIF data are summarized in the Supporting Information.



Figure 3. ORTEP drawing of product 5a.

To extend this synthetic methodology to other trifluoromethyl-containing olefins or other 1,3-dipoles, we next attempted the reactions of 1a with 6 or 7 under the standard conditions. However, none of the desired cycloadducts were formed [Scheme 1, Equations (1) and (2)]. Using compound 8 instead of 1a to react with 2a, no reaction occurred, probably because 8 is a relatively unreactive 1,3-dipole; see Scheme 1, Equation (3).

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Scheme 1. Attempted reactions using other 1,3-dipoles and CF_3 -containing olefins.

Because anyl-substituted α -trifluoromethyl α , β -unsaturated carboxylic acids (AFUCA)^[11] are one of the most valuable building blocks for the synthesis of CF₃-containing molecules, we also prepared AFUCA esters 6a-c and applied them to this [3+2] annulation reaction; the results are summarized in Scheme 2. Regardless of whether R^1 or R^6 were electron-rich or electron-deficient aromatic rings, the reactions proceeded smoothly to give the [3+2] annulation products in 58-99% yields along with 17:1-20:1 dr values (Scheme 2). Switching to azomethine imine 4a, the desired annulation product 10 could also be obtained in 66% yield along with high diastereoselectivity. The relative configurations of 9c and 10 were assigned on the basis of X-ray diffraction analysis. The ORTEP drawing of 9c and 10 are shown in Figure 4 and the CIF data are summarized in the Supporting Information. Increasing the scale of the reaction (0.5 mmol) afforded **3a** in 98% yield and >99:1 dr under the standard conditions (Scheme 3).



Scheme 2. Substrate scope of [3+2] annulation of azomethine imine 1 or 4 with 6.





Figure 4. ORTEP drawing of products 9c and 10.



Scheme 3. The [3+2] annulation of **1a** and **2a** performed on a larger scale.

An asymmetric variant of these cyclizations has also been investigated by using a variety of chiral catalysts, but the cyclization products were obtained with low *ee* values (see the Supporting Information for details).

In conclusion, we have developed a novel [3+2] annulation of N,N'-cyclic or C,N-cyclic azomethine imines with CF₃-containing olefins, affording the corresponding trifluoromethyl-containing pyrazolidine analogues in excellent yields (up to >99%) with high diastereoselectivities (up to 99:1) under mild conditions. Current efforts focusing on developing an asymmetric version of this reaction and applying this methodology to synthesize biologically active products are in progress.

Experimental Section

General Procedure: Under an argon atmosphere, CF_3 -containing olefin (0.25 mmol) was added to a solution of azomethine imine **1** or **4** (0.20 mmol) in CH_2Cl_2 (2.0 mL) and the resulting reaction mixture was stirred at room temperature for 2–48 h. The solvent was removed under reduced pressure and residue was purified by

chromatography on silica gel (petroleum ether/EtOAc, $6:1 \rightarrow 4:1$) to provide the desired product.

Dimethyl 5-Oxo-1-phenyl-3-(trifluoromethyl)tetrahydropyrazolo[1,2*a***]pyrazole-2,2(1***H***)-dicarboxylate (3a):** Yield 78 mg (>99%); colorless solid; m.p. 172–173 °C. ¹H NMR (400 MHz, CDCl₃, TMS): δ = 7.37–7.34 (m, 3 H), 7.18–7.15 (m, 2 H), 5.51 (q, *J* = 7.2 Hz, 1 H), 4.86 (s, 1 H), 3.81 (s, 3 H), 3.40–3.32 (m, 4 H), 3.17–3.11 (m, 1 H), 2.37 (ddd, *J*₁ = 2.4, *J*₂ = 10.0, *J*₃ = 17.2 Hz, 1 H), 1.96 (dt, *J*₁ = 10.0, *J*₂ = 17.2 Hz, 1 H) ppm. ¹³C NMR (100 MHz, CDCl₃): δ = 172.6, 166.5, 166.0, 132.4, 129.3, 129.0, 128.9, 122.9 (q, *J* = 280.4 Hz), 72.0, 69.4, 59.2 (q, *J* = 33.8 Hz), 53.6, 53.1, 44.3, 32.6 ppm. ¹⁹F NMR (376 MHz, CFCl₃): δ = -69.6 (d, *J* = 7.5 Hz) ppm. IR (neat): \tilde{v} = 2952, 2362, 1740, 1716, 1401, 1270, 1178, 1142, 933, 704 cm⁻¹. MS (ESI): *m/z* (%) = 387.1 (100) [M + H⁺]. HRMS: Calcd. for C₁₇H₁₈F₃N₂O₅⁺ [M + H⁺] 387.1168; found 387.1162.

CCDC-866876 (for **3a**), -875407 (for **5a**), -876533 (for **9c**), and -895560 (for **10**) contain the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/ data_request/cif.

Supporting Information (see footnote on the first page of this article): Spectroscopic data of the compounds shown in Tables 1, 2, and 3, and Schemes 1 and 3, detailed descriptions of experimental procedures and the crystal structure data of 3a, 5a, 9c, and 10.

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