

# Glycosyl Phenyl Sulfoxides as a Source of Glycosyl Carbanions: Stereoselective Synthesis of C-Fucosides<sup>†</sup>

Carlos Jaramillo,<sup>§</sup> Guillermo Corrales, and Alfonso Fernández-Mayoralas\*

*Departamento de Química Orgánica Biológica, Instituto de Química Orgánica (CSIC), Juan de la Cierva 3, 28006 Madrid (Spain)*

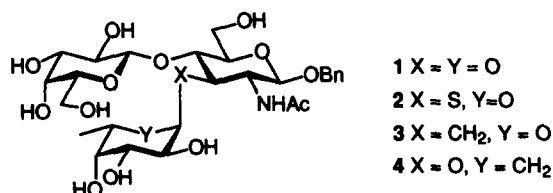
Received 25 June 1998; accepted 11 August 1998

## Abstract

Phenylsulfinyl-lithium exchange on glycosyl phenyl sulfoxides leads to configurationally stable anomeric carbanions which can react stereoselectively with electrophiles. Thus, the reaction of 3,4-*O*-isopropylidene- $\alpha$ -L-fucopyranosyl phenyl sulfoxide with *t*BuLi followed by treatment with isobutyraldehyde led to the  $\alpha$ -configured C-glycoside; the  $\beta$ -anomer furnished the corresponding  $\beta$ -C-glycoside. © 1998 Elsevier Science Ltd. All rights reserved.

**Keywords:** Carbanions; Carbohydrates; C-Glycosides; Addition reactions.

C-Glycosides are regarded as non-hydrolyzable analogues of O-glycosides, and the study of their synthesis [1] and structural properties [2,3] is subject of current interest. Among this group of substances, C-disaccharides have emerged as mimics [4] of oligosaccharide fragments. Within this context, we were interested in the preparation of analogues of the biologically important [5] Lewis X trisaccharide (1) in which one of the anomeric oxygens of the fucosyl moiety has been replaced by a sulfur atom (2) [6,7] or a methylene group (compounds 3 and 4 [8]). The preparation of the C-disaccharide fragment of 3 was envisaged by using a fucosyl anomeric carbanion and a 3-C-branched glucosamine electrophile.



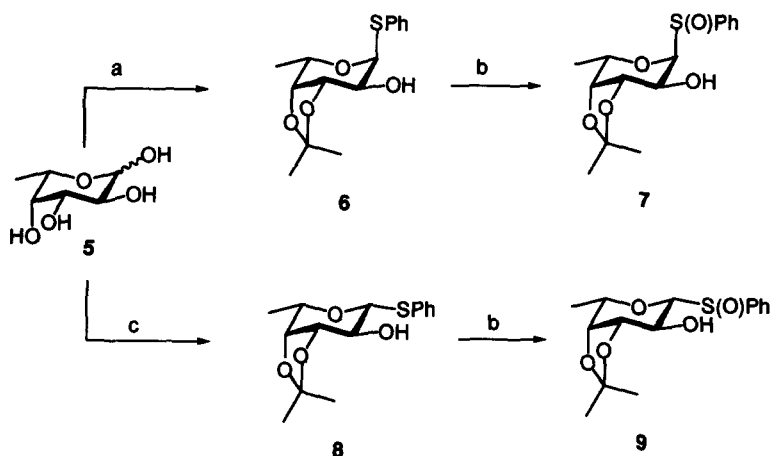
Non-stabilized, anomeric carbanions bearing oxygenated substituents at position 2 have been previously generated by sequential two electron transfer with either lithium naphthalenide [9] (from glycosyl chlorides) or samarium diiodide (from glycosyl sulfones [10,11], phosphates [12] or chlorides [13]), or by tin-lithium exchange with *n*-butyl lithium [14,15] (from glycosyl stannanes). In this communication we describe a new method to

<sup>†</sup> Dedicated to Prof. M. Martín-Lomas

<sup>§</sup> Present address: Lilly S.A., Avda. de la Industria 30, 28108 Alcobendas (Madrid), Spain.

generate anomeric carbanions from easily accessible glycosyl sulfoxides, and their stereoselective reaction with electrophiles with retention of the configuration at the anomeric centre.

Fucopyranosyl phenyl sulfoxides **7** and **9** were stereoselectively prepared from L-fucose **5** as outlined in Scheme 1. The  $\alpha$ -anomeric sulfoxide **7** was obtained through an original route whose key step, the formation of phenyl  $\alpha$ -thioglycoside **6**, was performed by a modified procedure for the synthesis of  $\alpha$ -fucopyranosides [16].

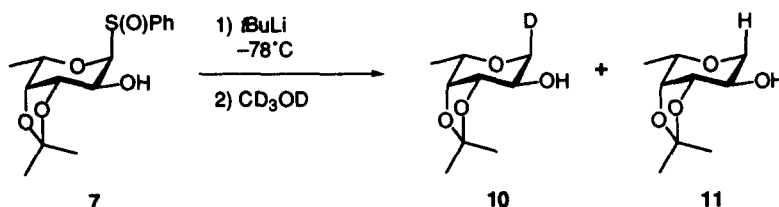


**Scheme 1.** Reagents and conditions. (a) 1.  $\text{Et}_3\text{N}$ , TMSCl, DMF,  $25^\circ\text{C}$ ; 2. TMSI,  $\text{CH}_2\text{Cl}_2$ ,  $25^\circ\text{C}$ ; 3. 2,6-di-*tert*-butyl-4-methylpyridine, PhSH,  $\text{CH}_2\text{Cl}_2$ ,  $5^\circ\text{C}$ ; 4. MeOH,  $\text{CH}_2\text{Cl}_2$ ,  $25^\circ\text{C}$ ; 5. 2,2-dimethoxypropane, *p*-TsOH,  $\text{Me}_2\text{CO}$ ,  $25^\circ\text{C}$ , 79% (5 steps); (b)  $\text{NaHCO}_3$ , *m*-CPBA,  $\text{CH}_2\text{Cl}_2$ ,  $-78^\circ\text{C}$ , 86% for **7** and 92% for **9**. (c) 1.  $\text{Ac}_2\text{O}$ , Py,  $25^\circ\text{C}$ ; 2. PhSH,  $\text{SnCl}_4$ ,  $\text{CH}_2\text{Cl}_2$ ,  $-20$  to  $0^\circ\text{C}$ ; 3. NaOMe, MeOH,  $25^\circ\text{C}$ ; 4. 2,2-dimethoxypropane, *p*-TsOH,  $\text{Me}_2\text{CO}$ ,  $25^\circ\text{C}$ , 69% (4 steps)

Phenylsulfinyl-lithium exchange [17,18] was tried on **7**, quenching the anomeric carbanion with deuterated methanol, which led to a mixture of the corresponding deuterated and protonated 1,5-anhydrofucitols **10** and **11**, respectively, plus recovered starting material (Table 1). Among the solvents tested ( $\text{C}_6\text{H}_6$ , toluene, DME, THF, and  $\text{Et}_2\text{O}$ ) the best yields of metallation were obtained in THF and  $\text{Et}_2\text{O}$ . The reaction was tried varying the number of equivalents of *t*BuLi and  $\text{CD}_3\text{OD}$ , the time of metallation and deuteration steps, and in the presence of MeLi·LiBr prior to *t*BuLi treatment [19] (Table 1). The best results in terms of yield and ratio of **10**:**11**, were obtained in  $\text{Et}_2\text{O}$  and adding one equivalent of MeLi·LiBr (Exp. 5 in Table 1).  $^1\text{H-N.m.r.}$  experiments of the crude mixtures showed, in all experiments, the only presence of the above-mentioned products. Hence, the anomeric carbanion seems to be configurationally stable, leading to a stereospecific reaction, since no  $\beta$ -deuterated 1,5-anhydrofucitol could be detected. This was further supported by the results obtained from reaction of the fucosyl lithium so generated with isobutyraldehyde, which led (Scheme 2) to a diastereomeric mixture of the  $\alpha$ -configured C-glycosides **12**.<sup>1</sup> Again, no  $\beta$ -C-glycosides

<sup>1</sup> Experimental Procedure: Under argon a 0.033 M solution of **7** in dry  $\text{Et}_2\text{O}$  at  $-78^\circ\text{C}$  was treated with 1.1 equivalents of MeLi·LiBr (1.5 M in  $\text{Et}_2\text{O}$ ) followed by slow addition of 5 equivalents of *t*-BuLi (1.64 M in hexanes). After 25 min, 3 equivalents of *i*-PrCHO were added, and the mixture was stirred for 90 min at  $-78^\circ\text{C}$  and then quenched with saturated aqueous solution of  $\text{NH}_4\text{Cl}$ . After partitioning between water and dichloromethane, the organic layer was dried over  $\text{Na}_2\text{SO}_4$  and concentrated to give a residue containing a diastereomeric mixture of **12**, which were separated by flash chromatography (eluent: hexane/ethyl acetate 6:1) and characterized as their diisopropylidene derivatives.

could be detected by  $^1\text{H}$ -n.m.r. The structures of diastereoisomers **12** were confirmed by their transformation into the diacetals **13** and **14**, which gave also further information about the new chiral center created in the C-glycosylation.<sup>2</sup>



**Table 1.**

Phenylsulfinyl-lithium exchange of **7**<sup>a</sup>

Exp.	Solvent	eq/t [min] <sup>b</sup>	eq/t [min] <sup>c</sup>	Yield[%]	10:11
1	THF	5/0.5	3/0.5	61	67:33
2	THF	5/5	6/5	80	65:35
3	Et <sub>2</sub> O	5/5	6/5	63	77:23
4 <sup>d</sup>	Et <sub>2</sub> O	5/5	6/5	54	89:11
5 <sup>d</sup>	Et <sub>2</sub> O	5/20	6/5	77	87:13

<sup>a</sup> Yields and ratio of **10:11** were determined by  $^1\text{H}$  N.m.r. spectroscopy.

<sup>b</sup> Eq: Equivalents of *t*BuLi; t: time of metallation step.

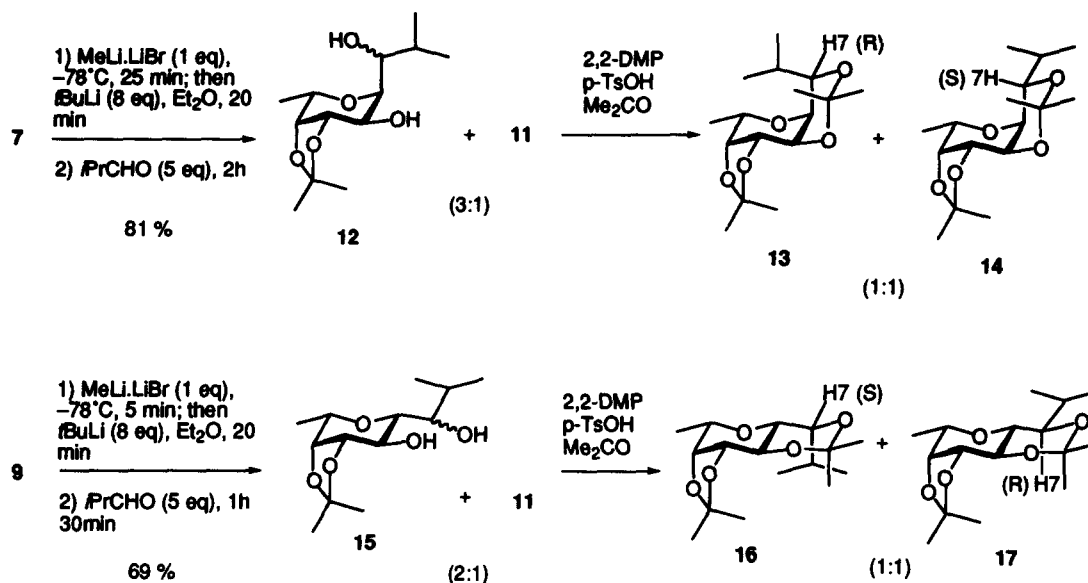
<sup>c</sup> Eq: Equivalents of CD<sub>3</sub>OD; t: time of deuteration step.

<sup>d</sup> 1 eq of MeLi·LiBr was added prior to *t*BuLi treatment [19].

Additional evidence for the stereospecificity of the process was obtained from the reaction of fucosyl phenyl sulfoxide **9**, prepared (Scheme 1) analogously to **7** from phenyl thioglycoside **8** [20]. In this case, phenylsulfinyl-lithium exchange was slower, and the corresponding  $\beta$ -configured fucosyl lithium species proved to be less efficient in the C-glycosylation, although it afforded only the corresponding  $\beta$ -C-glycosides **15** (Scheme 2), whose structure was again secured after acetalation.

In summary, a new method for the preparation of C-glycosides has been described, which makes use of a stereospecific phenylsulfinyl-lithium exchange. The required fucosyl phenyl sulfoxides have been efficiently prepared in a highly stereoselective manner. The identification of other and more complex electrophiles, as well as its use for the preparation of Lewis X analogues, is currently underway.

<sup>2</sup> All new compounds gave satisfactory elemental analyses and their expected  $^1\text{H}$ -NMR spectra. Selected  $^1\text{H}$ -NMR data (200 MHz, CDCl<sub>3</sub>) for **13**:  $\delta$  3.98 (t, 1H,  $J$  = 2.7 Hz, H-2), 3.73 (dd, 1H,  $J$  = 2.5 and 1.4 Hz, H-1), 3.16 (dd, 1H,  $J$  = 1.4 and 9.4 Hz, H-7). For **14**:  $\delta$  4.13 (dd, 1H,  $J$  = 5.3 and 2.7 Hz, H-2), 4.08 (dd, 1H,  $J$  = 5.3 and 8.4 Hz, H-1), 3.34 (dd, 1H,  $J$  = 8.4 and 5.4 Hz, H-7). For **16**:  $\delta$  3.66 (dd, 1H,  $J$  = 8.6 and 9.6 Hz, H-2), 3.46 (dd, 1H,  $J$  = 6.3 and 9.5 Hz, H-7), 3.19 (dd, 1H,  $J$  = 6.3 and 9.6 Hz, H-1). For **17**:  $\delta$  3.71 (dd, 1H,  $J$  = 7.0 and 9.7 Hz, H-2), 3.60 (dd, 1H,  $J$  = 3.4 and 9.4 Hz, H-7), 2.89 (t, 1H,  $J$  = 9.6 Hz, H-1).



Scheme 2

**Acknowledgment.** Financial support by DGES (PB96-0828) is gratefully acknowledged.

## References

- [1] Levy DE, Tang PC. The Chemistry of C-Glycosides. Exeter: Pergamon, 1995.
- [2] Espinosa JF, Cañada FJ, Asensio JL, Martín-Pastor M, Dietrich H, Martín-Lomas M, Schmidt RR, Jiménez-Barbero J. J. Am. Chem. Soc. 1996;118:10862-10871.
- [3] Wei A, Boy KM, Kishi Y, J. Am. Chem. Soc 1995;117:9432-9436.
- [4] Carbohydrate Mimics. Concepts and Methods. Chapleur Y, Editor. Weinheim: Wiley-VCH, 1998.
- [5] Feizi T. Curr. Opin. Struct. Biol. 1993;3:701-710.
- [6] Aguilera B, Fernández-Mayoralas A. Chem. Commun. 1996:127-128.
- [7] Aguilera B, Fernández-Mayoralas A. J. Org. Chem. 1998;63:2719-2723.
- [8] Carpintero M, Fernández-Mayoralas A, Jaramillo C. J. Org. Chem. 1997;62:1916-1917.
- [9] Wittman V, Kessler H. Angew. Chem; Int. Ed. Engl. 1993;32:1091-1093.
- [10] Mazéas D, Skrydstrup T, Beau J-M. Angew. Chem; Int. Ed. Engl. 1995;34:909-912.
- [11] de Pouilly P, Chénédé A, Mallet J-M, Sinaÿ P. Bull. Soc. Chim. Fr. 1993;130:256-265.
- [12] Hung S-C, Wong C-H. Angew. Chem; Int. Ed. Engl. 1996;35:2671-2674.
- [13] Hung S-C, Wong C-H. Tetrahedron Lett. 1996;37:4903-4906.
- [14] Frey O, Hoffmann M, Wittmann V, Kessler H, Uhlmann P, Vasella A. Helv. Chim. Acta. 1994;77:2060-2069.
- [15] Lesimple P, Beau J-M. Bioorg. Med. Chem. 1994;2:1319-1330.
- [16] Uchiyama T, Hindsgraul O. Synlett 1996:499-501.
- [17] Satoh T, Horiguchi K. Tetrahedron Lett. 1995;36:8235-8238.
- [18] Casillas M, Gómez AM, López JC, Valverde S. Synlett 1996:628-630.
- [19] Hoffmann M, Kessler H. Tetrahedron Lett. 1997;38:1903-1906.
- [20] For the preparation of the enantiomer of **8**, see: Paulsen H, Rutz V, Brockhausen I. Liebigs Ann. Chem. 1992:747-758.