

THE USE OF 2,6-ANHYDRO-2-THIO GLYCOPYRANOSYL FLUORIDE FOR A HIGHLY α -STEREOSELECTIVE GLYCOSYLATION

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Summary: The glycosylation of 2, 6-anhydro-1-fluoro-2-thioglycoside **1** with alcohols under various conditions proceeded smoothly at low temperature to give the corresponding α -glycosides in high yields, and the selectivity was highly independent on the conditions examined.

The selective creation of O-glycosidic linkage of 2,6-dideoxy sugar in glycosylation has been a long standing problem in organic chemistry and still one of the most challenging task for organic chemists.¹⁾ In previous paper, we demonstrated a novel and efficient synthesis of 2,6-dideoxy- α -glycosides by use of phenyl 2,6-anhydro-1,2-dithio-D-altropyranosides as glycosyl donors to illustrate a highly stereocontrolled glycosylation.²⁾ Recently, glycopyranosyl fluoride³⁾ has been paid considerable attention as an efficient glycosyl donor as well as phenylthio glycoside. In this communication, we wish to report that 3,4-di-O-acetyl-2,6-anhydro-1-fluoro-2-thio-D-altropyranoside (**1**) was smoothly glycosylated with alcohols at low temperature by a variety of methods to give the corresponding α -glycosides which were easily converted into the 2,6-dideoxy- α -glycosides²⁾ in high yields.

The key glycosyl donor **1**⁴⁾ was readily prepared from **2**²⁾ (1.3 equiv NBS, 1.5 equiv DAST,⁵⁾ -25°C, 10min) in 84% yield. We first examined the glycosylation of the anomeric mixture of **1** by using cyclohexylmethanol (**3**) as the glycosyl acceptor with many kinds of reagents such as SnCl₂-AgClO₄^{3a)}, SnCl₂-ZnCl₂, SnCl₂^{3b)}, TMSOTf^{3c)}, Cp₂MCl₂-AgClO₄ or AgBF₄ (M=Zr, Hf)^{3d, 3e)}. The results summarized as entries 1-6 in Table 1 showed that these reactions proceeded at low temperature to give the α -glycoside with high stereocontrol in high to excellent yields even by the method which was not originally developed for α -glycosides (entry 5). Remarkably, stereoselectivity of the glycosylations was highly independent on the conditions examined. Further **1** was found to glycosylate with even hindered alcohols **5** and **6** by using only SnCl₂ as an activator to afford the corresponding α -glycosides in high yields (entries 8 and 9).

These results indicated that 2,6-anhydro-1-fluoro-2-thio sugar should find wide application in the synthesis of 2,6-dideoxy- α -glycosides.

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References and Notes:

- 1) See references and notes cited in ref. 2.
- 2) K. Toshima, S. Mukaiyama, T. Ishiyama, and K. Tatsuta, *Tetrahedron Lett.*, **31**, 3339 (1990).

Table 1. Glycosylations of 1 with alcohols by several conditions

1: X=F
2: X=SPh

α β

Entry	Alcohol ^{a)} (R-OH)	Activators (equiv)	Solvent	Temp.	Time	Product ²⁾ (α/β^c)	Yield (%) ^{b)}
1		SnCl ₂ (1.1)-AgClO ₄ (1.1)	Et ₂ O	-10°C	90min	97/3	98
2	3	SnCl ₂ (1.1)-ZnCl ₂ (1.1)	Et ₂ O	-10°C	90min	98/2	89
3		SnCl ₂ (1.1)	Et ₂ O	-10°C	90min	100/0	91
4 ⁶⁾		TMSOTf(1.0)	Et ₂ O	-10°C	90min	92/8	90
5		Cp ₂ HfCl ₂ (5.0)-AgClO ₄ (5.0)	CH ₂ Cl ₂	-10→25°C	90min	93/7	95
6		Cp ₂ ZnCl ₂ (0.6)-AgBF ₄ (1.2)	CH ₂ Cl ₂	-20→10°C	120min	98/2	92
7		SnCl ₂ (1.1)	Et ₂ O	-10°C	90min	98/2	94
8		SnCl ₂ (1.1)	Et ₂ O	-10°C	140min	98/2	76
9		SnCl ₂ (1.1)	Et ₂ O	-10°C	90min	96/4	81

a) All reactions were carried out by use of 2.0 equiv. of alcohol to the glycosyl donor.

b) Isolated yields after purification by column chromatography.

c) α/β Ratios were determined by ¹H-NMR spectroscopy and /or isolation of pure isomers.

- 3) (a) T. Mukaiyama, Y. Murai, and S. Shoda, *Chem. Lett.*, **1981**, 431; (b) K. C. Nicolaou, T. Ladduwahetty, J. L. Randall, and A. Chucholowski, *J. Am. Chem. Soc.*, **108**, 2466 (1986); (c) S. Hashimoto, M. Hayashi, and R. Noyori, *Tetrahedron Lett.*, **25**, 1379 (1984); (d) T. Matsumoto, H. Maeta, K. Suzuki, and late G. Tsuchihashi, *Tetrahedron Lett.*, **29**, 3567, 3571, 3575 (1988); (e) K. Suzuki, H. Maeta, T. Suzuki, and T. Matsumoto, *Tetrahedron Lett.*, **30**, 6879 (1989).
- 4) ¹H-NMR(CDCl₃, 270MHz) spectra [δ (TMS), J(Hz)] are shown only for anomeric proton for 1: 5.92 (23/25H, ddd, J=69.0, 2.1 and 1.6 Hz, H-1(α)), 5.94 (2/25H, dd, J=68.6 and 3.8Hz, H-1(β)).
- 5) K. C. Nicolaou, R. E. Dolle, D. P. Papahatjis, and J. L. Randall, *J. Am. Chem. Soc.*, **106**, 4189 (1984).
- 6) Only when CH₃CN was used as a solvent in the glycosylation, β -anomer was predominantly obtained in high yield (88%, α/β =5/95).