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> SHORT COMMUNICATIONS

Synthesis of 5-Amino-3*H*-pyrrole-3,4-dicarbonitriles from 4-Aryl-4-oxobutane-1,1,2,2-tetracarbonitriles

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It is known that 4-oxoalkane-1,1,2,2-tetracarbonitriles react with nitrogen-centered nucleophiles, such as ammonia and amines, to give 3-amino-7-oxo-4,6-diazabicyclo[3.2.1]oct-2-ene-1,2-dicarbonitriles [1], 3-amidinio-2-aminopyridine-4-carboxylates [2], diethylammonium 3,4-dicyano-5,6,7,8-tetrahydroquinolin-2-olates [3], and ammonium 4-aryl-4-oxo-1,1,2-tricyanobut-2-en-1-ides [4]. However, these results do not allow us to draw a definite conclusion on the direction of attack by N-nucleophiles on the reaction centers in polyelectrophilic 4-oxoalkane-1,1,2,2-tetracarbonitriles.

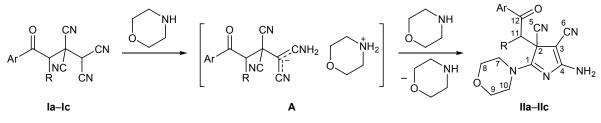
While continuing studies on reactions of tetracyanoethylated ketones with N-nucleophiles, we found that 4-aryl-4-oxobutane-1,1,2,2-tetracarbonitriles **Ia–Ic** react with morpholine to produce 80–93% of previously unknown 5-amino-3-(2-aryl-2-oxoethyl)-2-morpholino-3*H*-pyrrole-3,4-dicarbonitriles **IIa–IIc**.

Presumably, the initial step in this reaction is formation of morpholinium salt **A** with CH acid **Ia–Ic**. Analogous salts containing metal and ammonium ions as cationic species have been reported [4–6]. Salts **A** undergo further transformations by the action of excess morpholine. These transformations involve primarily the β -cyano group, as follows from the presence of morpholine fragment in position 2 of the 3*H*-pyrrole ring in **Ha–Hc**. Attack on the β -cyano group is determined by reduced electrophilicity of the terminal cyano groups due to delocalization of the negative charge in the anion of salt **A**. 3*H*-Pyrrole **Hc** possesses two asymmetric carbon atoms (C², C¹¹), but this compound was isolated as a single diastereoisomer.

The structure of pyrroles **IIa–IIc** was confirmed by their IR, ¹H and ¹³C NMR, and mass spectra and elemental analyses. The structure of compound **IIa** was unambiguously proved by the X-ray diffraction data.

General procedure for the synthesis of compounds IIa–IIc. A solution of 0.5 mmol of 4-aryl-4oxobutane-1,1,2,2-tetracarbonitrile Ia–Ic in 3 ml of anhydrous ethyl acetate was cooled to -10 to -15° C, 0.087 g (1 mmol) of morpholine was added under vigorous stirring, and the resulting yellow–orange solution was left to stand at -10 to -15° C in a closed vessel. After 2–3 days, the yellowish precipitate was filtered off, washed on a filter with cold ethyl acetate and diethyl ether, and dried in air.

5-Amino-2-morpholino-3-(2-oxo-2-phenylethyl)-3H-pyrrole-3,4-dicarbonitrile (IIa). Yield 0.156 g (93%), mp 208–209°C (decomp.). IR spectrum, v, cm⁻¹: 3184–3311 (NH₂); 2232, 2182 (C \equiv N); 1691 (C=O). ¹H NMR spectrum, δ, ppm (*J*, Hz): 3.62–3.80 m (8H, NCH₂CH₂O), 3.92 d (1H, CH₂, *J* = 17.8),



Ar = Ph, R = H (a); Ar = 4-ClC₆H₄, R = H (b); Ar = 4-MeOC₆H₄, R = Me (c).

4.06 d (1H, CH₂, J = 17.8), 7.27 s (2H, NH₂), 7.53– 7.57 m (2H, H_{arom}), 7.60–7.70 m (1H, H_{arom}), 7.98– 8.01 m (2H, H_{arom}). ¹³C NMR spectrum, $\delta_{\rm C}$, ppm: 42.73 (C¹¹), 47.45 (C², C⁷, C¹⁰), 57.54 (C³), 65.98 (C⁸, C⁹), 117.56, 118.54 (C⁵, C⁶), 128.66, 129.17, 134.24, 136.17 (C₆H₅); 170.66, 170.94 (C¹, C⁴); 194.10 (C¹²). Mass spectrum, *m/z* (*I*_{rel}, %): 335 (43) [*M*]⁺, 216 (100). Found, %: C 64.57; H 5.02; N 20.97. C₁₈H₁₇N₅O₂. Calculated, %: C 64.47; H 5.11; N 20.88. *M* 335.36.

5-Amino-3-[2-(4-chlorophenyl)-2-oxoethyl]-2morpholino-3H-pyrrole-3,4-dicarbonitrile (IIb). Yield 0.157 g (85%), mp 220–221°C (decomp.). IR spectrum, v, cm⁻¹: 3157–3403 (NH₂); 2233, 2172 (C=N); 1667 (C=O). ¹H NMR spectrum, δ , ppm (*J*, Hz): 3.66–3.78 m (8H, NCH₂CH₂O), 3.90 d and 4.06 d (1H each, CH₂CO, *J* = 17.7), 7.26 s (2H, NH₂), 7.61–7.64 m (2H, H_{arom}), 7.99–8.03 m (2H, H_{arom}). Mass spectrum, *m/z* (*I*_{rel}, %): 369 (3) [*M*]⁺, 216 (100). Found, %: C 58.53; H 4.31; N 19.02. C₁₈H₁₆ClN₅O₂. Calculated, %: C 58.46; H 4.36; N 18.94. *M* 369.80.

5-Amino-3-[2-(4-methoxyphenyl)-1-methyl-2oxoethyl]-2-morpholino-3*H*-pyrrole-3,4-dicarbonitrile (IIc). Yield 0.152 g (80%), mp 211–212°C (decomp.). IR spectrum, v, cm⁻¹: 3149–3347 (NH₂); 2233, 2169 (C=N); 1672 (C=O). ¹H NMR spectrum, δ , ppm (*J*, Hz): 0.98 d (3H, CH₃, *J* = 6.9), 3.68–3.76 m (8H, NCH₂CH₂O), 3.86 s (3H, OCH₃), 4.26 q (1H, CHCH₃, *J* = 6.9), 7.08 d (2H, H_{arom}, *J* = 8.9), 7.28 s (2H, NH₂), 8.00 d (2H, H_{arom}, *J* = 8.9). Mass spectrum, *m*/*z* (*I*_{rel}, %): 379 (3) [*M*]⁺, 216 (100). Found, %: C 63.33; H 5.51; N 18.52. C₂₀H₂₁N₅O₃. Calculated, %: C 63.31; H 5.58; N 18.46. *M* 379.41.

The progress of reactions and the purity of products were monitored by TLC on Silufol UV-254 plates (spots were visualized by UV irradiation, treatment with iodine vapor, or thermal decomposition). The IR spectra were recorded on an FSM-1202 spectrometer with Fourier transform from samples dispersed in mineral oil. The ¹H and ¹³C NMR spectra were measured on a Bruker DRX-500 spectrometer at 500.13 and 125.76 MHz, respectively, using DMSO- d_6 as solvent and tetramethylsilane as internal reference. The elemental compositions were determined on a Laboratorni Přistroje instrument. The mass spectra (electron impact, 70 eV) were obtained on a Finnigan MAT INCOS-50 spectrometer.

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