COMMUNICATIC

# *exo*-Glycals from glycosyl cyanides. First generation of *C*-glycosylmethylene carbenes from 2,5- and 2,6-anhydroaldose tosylhydrazones

### Marietta Tóth and László Somsák\*

Published on 18 April 2001 on http://pubs.rsc.org | doi:10.1039/B102963K

Downloaded by North Carolina State University on 15 January 2013

Department of Organic Chemistry, University of Debrecen, POB 20, H-4010, Debrecen, Hungary. E-mail: somsak@tigris.klte.hu; Fax: +36-52-435-836; Tel.: +36-52-512-900/2348

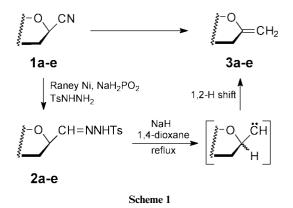
Received (in Cambridge, UK) 3rd January 2001, Accepted 2nd April 2001 First published as an Advance Article on the web 18th April 2001

In a one-pot reaction acylated glycosyl cyanides were transformed with Raney nickel-sodium hypophosphite into 2,5- and 2,6-anhydroaldose tosylhydrazones, which gave *exo*-glycals under Bamford–Stevens conditions.

*exo*-Glycals (2,5- or 2,6-anhydro-1-deoxy-hex- or -hept-1enitols) are useful carbohydrate derivatives exemplified by their applications *e.g.* in syntheses <sup>1</sup> and biological investigations.<sup>2</sup>

Several synthetic protocols are known for the preparation of such compounds: olefination of sugar lactones by using the Tebbe reagent,<sup>3</sup> dimethyltitanocene,<sup>1a</sup> or the Wittig methodology;<sup>4</sup> hydrogen halide eliminations from *C*-glycosyliodomethanes;<sup>5</sup> other methods using glycosyl carbanions as intermediates have been reviewed recently.<sup>6</sup>

Here we report a novel method for the preparation of *exo*-glycals based on the generation of *C*-glycosylmethylene carbenes by application of the aprotic Bamford–Stevens reaction<sup>7</sup> to tosylhydrazones of 2,5- and 2,6-anhydroaldose derivatives followed by their spontaneous rearrangement (Scheme 1). The reaction of acylated glycosyl cyanides<sup>8</sup> (1) with Raney nickel–sodium hypophosphite in the presence of



1.2–1.7 eq. of tosylhydrazine<sup>9</sup> in acetic acid–water–pyridine solvent mixture proved to provide easy access to the previously unknown 2,5- and 2,6-anhydroaldose tosylhydrazones 2 (Table 1). With the exception of 2e, one isomer was formed in

 Table 1
 Synthesis and selected data of 2,5- and 2,6-anhydroaldose tosylhydrazones 2 and exo-glycals 3

		2						3			
		Yield (%)	$[a]_{D}{}^{b}(c)$	NMR <sup>a</sup>					NMR <sup>a</sup>		
	1			H-1 (J <sub>1,2</sub> ) C-1	H-2 $(J_{2,3})$	NH	Yield (%)	$[a]_{\mathbf{D}}{}^{b}(c)$	H-1, H-1' <i>C-1</i>	C-2	
a	BzO OBz OBz OBz	64	+4 (1.00)	7.16 (6.3) <i>144.1</i>	4.34 (9.5)	8.12	72	+43 (0.98)	4.70, 4.96 97.3	153.3	
b		87	+6 (0.97)	7.00 <i>°</i> (6.7) 144.1	4.14 <sup>c</sup> (9.8)	11.6°	82	+74 (1.45) Lit. <sup>12</sup> +69 (1.00)	4.51, 4.82 96.1	154.1	
c		73	-35 (2.03)	7.04 (6.8) <i>144.3</i>	3.89 (9.5)	9.07	86	-94 (0.94)	4.49, 4.69° 96.9	154.5	
d	AcO NPhth	58	+65 (1.01)	7.10° (5.5) 144.0	4.87 <sup>c</sup> (10.4)	11.48 <sup>c</sup>	74	+29 (1.08)	4.24, 4.80 95.7	151.9	
e	BZO CN BZO OBZ	Iı	Inseparable mixture (probably two diastereomers)				26 <sup><i>d</i>,<i>e</i></sup>	+24 (0.90)	4.44, 4.67 <i>87.1</i>	157.7	

<sup>*a*</sup> For CDCl<sub>3</sub> solutions unless otherwise stated ( $\delta$ /ppm, *J*/Hz). <sup>*b*</sup> Measured in CHCl<sub>3</sub> at room temperature. [*a*]<sub>D</sub> Values are given in 10<sup>-1</sup> deg cm<sup>2</sup> g<sup>-1</sup>. <sup>*c*</sup> For DMSO-*d*<sub>6</sub> solutions. <sup>*d*</sup> Overall yield for the two steps. <sup>*e*</sup> For deprotonation 2 eq. of NaH was used.

these transformations. These compounds are new representatives of the rather sparsely populated class of 2,5- and 2,6anhydroaldimine derivatives.<sup>10</sup>

The tosylhydrazone salt formation and thermolysis were performed in one operation as suggested in the literature:<sup>7a</sup> a solution of **2** in 1,4-dioxane was added dropwise to boiling 1,4-dioxane containing 10 eq. of sodium hydride. *exo*-Glycals **3** were isolated by column chromatography. The reactions were rather clean and highly selective in so far as no products of 1,2-C or 1,2-O shifts could be detected in the mixtures. This was the case even with the furanoid derivative **2e**, in contrast to the ring enlargement observed with tetrahydrofuran-2-yl-methylenes.<sup>11</sup>

In summary, we have described a new route to *exo*-glycals from readily available glycosyl cyanides *via* 2,5- and 2,6- anhydroaldose tosylhydrazones. This method can be a useful alternative to the known procedures, especially for the preparation of acyl-protected derivatives.

## Experimental

#### General procedure for the synthesis of 2,5- and 2,6-anhydroaldose tosylhydrazones (2)

Raney nickel (1.5 g, from an aqueous suspension, Merck) was added at room temperature to a vigorously stirred solution of pyridine (5.7 mL), acetic acid (3.4 mL), and water (3.4 mL). Then sodium hypophosphite (0.74 g, 8.4 mmol), tosylhydrazine (0.22-0.32 g, 1.2-1.7 mmol), and the corresponding glycosyl cyanide 1 (1 mmol) were added to the mixture. When the reaction was complete (TLC, eluent: ethyl acetate-hexane 1:1) the insoluble materials were filtered off with suction, and washed with dichloromethane (10 mL). The organic layer of the filtrate was separated, washed sequentially with water (3 mL), 10% aqueous hydrogen chloride solution ( $2 \times 3$  mL), cold, saturated sodium hydrogen carbonate solution  $(2 \times 3 \text{ mL})$  and water (3 mL), and then dried on anhydrous magnesium sulfate. The solution was concentrated under reduced pressure, and traces of pyridine were removed by repeated co-evaporations with toluene. The residue was purified by column chromatography (eluent : ethyl acetate-hexane 1:1 or 1:2) to give syrupy 2a-d.

#### General method for the synthesis of exo-glycals (3)

Sodium hydride (0.24 g, 10 mmol) was added to dry 1,4dioxane (25 mL). The suspension was stirred and heated to reflux, and then a solution of a tosylhydrazone 2 (1 mmol) in dry 1,4-dioxane (25 mL) was added dropwise. When the reaction was complete (TLC, eluent: ethyl acetate-hexane 1 : 1), the mixture was cooled and the insoluble material filtered off. The solvent was removed under reduced pressure, and the residue was purified by column chromatography (eluent : gradient of ethyl acetate-hexane 1:2 to 1:1) to give *exo*-glycals **3a**-e.

#### Acknowledgements

This work was supported by the Hungarian Scientific Research Fund (Grant: OTKA T 32124).

#### Notes and references

- See as leading references: (a) C. R. Johnson and B. A. Johns, Synlett, 1997, 1406; (b) T. Vidal, A. Haudrechy and Y. Langlois, Tetrahedron Lett., 1999, 40, 5677; (c) A. D. Campbell, D. E. Paterson, T. M. Raynham and R. J. K. Taylor, Chem. Commun., 1999, 1599; (d) I. P. Smoliakova, Curr. Org. Chem., 2000, 4, 589.
- 2 Leading references: G. Legler, Adv. Carbohydr. Chem. Biochem., 1990, 48, 319; M. H. D. Postema, C-Glycoside Synthesis, CRC Press, Boca Raton, 1995, pp. 353–355.
- 3 C. S. Wilcox, G. W. Long and H. Suh, *Tetrahedron Lett.*, 1984, **25**, 395; T. V. RajanBabu and G. S. Reddy, *J. Org. Chem.*, 1986, **51**, 5458.
- 4 Y. Lakhrissi, C. Taillefumier, M. Lakhrissi and Y. Chapleur, *Tetrahedron: Asymmetry*, 2000, **11**, 417.
- M. Brockhaus and J. Lehmann, *Carbohydr. Res.*, 1977, **53**, 21;
   J. Lehmann and B. Schwesinger, *Carbohydr. Res.*, 1982, **107**, 43;
   O. R. Martin and F. Xie, *Carbohydr. Res.*, 1994, **264**, 141; R. D. Guthrie, I. D. Jenkins, J. J. Watters, M. W. Wright and R. Yamasaki, *Aust. J. Chem.*, 1982, **35**, 2169.
- 6 L. Somsák, Chem. Rev., 2001, 101, 81.
- 7 (a) K.-P. Zeller and H. Gugel, in *Methoden der organischen Chemie* (*Houben-Weyl*), Vol. E19b, ed. M. Regitz, Thieme, Stuttgart, 1989, pp. 225–243; (b) K. Maruoka and H. Yamamoto, in *Comprehensive Organic Synthesis*, Vol. 6, ed. B. M. Trost and I. Fleming, Pergamon, Oxford, 1991, pp. 776–779.
- 8 Z. Györgydeák and I. Pelyvás, Monosaccharide Sugars—Chemical Synthesis by Chain Elongation, Degradation and Epimerization, Academic Press, San Diego, 1998, pp. 316–331.
- 9 This method is a modification of a protocol used for the preparation of 2,5- and 2,6-anhydroaldoses from glycosyl cyanides in the presence of 1,2-dianilinoethane: H. P. Albrecht, D. B. Repke and J. G. Moffatt, J. Org. Chem., 1973, 38, 1836; H.-M. Dettinger, G. Kurz and J. Lehmann, Carbohydr. Res., 1979, 74, 301.
- 10 Leading references for 2,5- and 2,6-anhydroaldoximes: S. Kim, I. Y. Lee, J.-Y. Yoon and D. H. Oh, J. Am. Chem. Soc., 1996, 118, 5138; D.-P. Pham-Huu, M. Petrušová, J. N. BeMiller and L. Petruš, Synlett, 1998, 1319; D.-P. Pham-Huu, M. Petrušová, J. N. BeMiller and L. Petruš, J. Carbohydr. Chem., 2000, 19, 93; for nitrones of 2,5- and 2,6-anhydroaldoses: A. Dondoni, F. Junquera, F. L. Merchán, P. Merino, M.-C. Scherrmann and T. Tejero, J. Org. Chem., 1997, 62, 5484; for silyl nitronates of 2,6-anhydroaldoses: O. R. Martin, F. E. Khamis and S. P. Rao, Tetrahedron Lett., 1989, 30, 6143.
- 11 S. Kim, J.-Y. Yoon and C. M. Cho, *Chem. Commun.*, 1996, 909; S. Kim and J.-Y. Yoon, *Synthesis*, 2000, 1622.
- 12 H. Fritz, J. Lehmann and P. Schlesselmann, *Carbohydr. Res.*, 1983, 113, 71.