

Synthesis of C5-Allylindoles through an Iridium-Catalyzed Asymmetric Allylic Substitution/Oxidation Reaction Sequence of N-Alkyl Indolines

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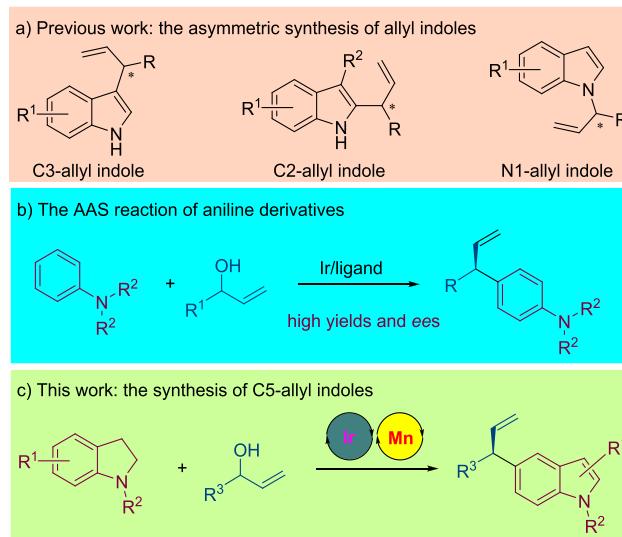
Supporting Information



ABSTRACT: Iridium/Brønsted acid cooperative catalyzed asymmetric allylic substitution reactions at the C5 position of indolines have been reported for the first time. The highly efficient protocol allows rapid access to various C5-allylated products in good to high yields (48–97%) and enantioselectivities (82% to >99% ee) with wide functional group tolerance. The transformations allow not only the formation of C5-allylindoline derivatives but also the synthesis of C5-allylindole analogues in good yields and excellent stereoselectivities via an allylation/oxidation reaction sequence.

Optically active indole fragments are ubiquitous features of numerous bioactive natural products and bioactive pharmaceuticals.¹ Thus, the enantioselective functionalization of the indole structure has attracted a great deal of attention,² and the transition metal-catalyzed asymmetric allylic substitution (AAS) reaction is particularly useful.³ There are six C–H bonds and one N–H bond in the indole structure that could be functionalized. It has been proven that the C3 position has priority in the intermolecular AAS reactions, and the first AAS reaction of indoles with allylic acetate was reported by Kocovský and co-workers in 1999.^{4a} Since then, several successful examples have been reported by different research groups.⁴ In addition, the branched selective N1-allylation reaction of indoles has also been described by Hartwig,⁵ You,⁶ Krische,⁷ and others.⁸ Furthermore, asymmetric C2-allylation has also been investigated. To overcome the regioselectivity issues, intramolecular asymmetric indole C2-allylation procedures were disclosed, which allowed the synthesis of annulated chiral C2-allylated indoles stereoselectively.⁹ As an exception, Tambar and co-workers developed a chiral phosphoric acid-catalyzed intramolecular asymmetric aza-Claisen rearrangement approach to form chiral C2-allylated 3-amino indoles.¹⁰ Very recently, Zheng, Taylor, Unsworth, and You reported an unprecedented enantioselective intermolecular C2-allylation of 3-substituted indoles, generating the C2-allylated products in excellent enantioselectivities with excellent regiocontrol.¹¹ However, the methods for the synthesis of 5-allylindoles are still a challenge due to the innate reactivity of the indole skeleton (Scheme 1a).

Scheme 1. Previous Studies and Approaches for Constructing Chiral C5-Allylindoles



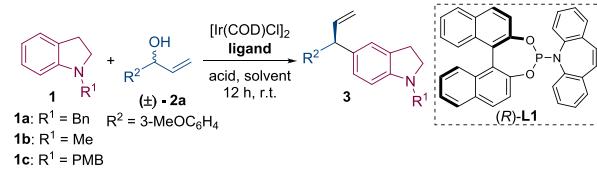
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Chiral C5-substituted indoles are key building blocks in many biologically active compounds, including the antitumor alkaloid drug vinblastine,¹² the haplophytine alkaloid,¹³ murrafoline D,¹⁴ the raputindole family,¹⁵ etc., and have attracted a great deal of attention from synthetic chemists. Most indole C5-functionalization reactions involve C–H activation,^{2e,16} coupling,¹⁷ and semireduction¹⁸ strategies. However, the dependency on directing groups and the formation of achiral products are obvious drawbacks. Aniline, which is always used as a nucleophile to react with electrophiles due to its high electron density at the *para* position,¹⁹ was successfully applied in the AAS reaction by Fu's research group (**Scheme 1b**).²⁰ As a continuation of our research interest in AAS reactions,²¹ herein we report the highly stereoselective synthesis of C5-allylindoles through an iridium-catalyzed AAS reaction of *N*-alkyl indolines with allylic alcohols followed by an oxidation reaction in a one-pot reaction (**Scheme 1c**). This method tolerates a broad range of indolines and 1,2,3,4-tetrahydroquinoline analogues with various functional groups. The transformation could be easily scaled up to gram scale, and the formed products could be applied in the synthesis of bioactive chiral C5-functionalized indoles.

We initially selected the AAS reaction of *N*-benzylindoline (**1a**) with 1-(3-methoxyphenyl)prop-2-en-1-ol (**2a**) as the model substrate (**Table 1**). The combination of commercially

Table 1. Optimization of Reaction Conditions^a



entry	R	acid	solvent	yield (%) ^b	ee (%) ^c
1	Bn	BF ₃ ·Et ₂ O	THF	67	96
2	Bn	BF ₃ ·Et ₂ O	DCM	74	81
3	Bn	BF ₃ ·Et ₂ O	Et ₂ O	57	95
4	Bn	BF ₃ ·Et ₂ O	toluene	60	96
5	Bn	BF ₃ ·Et ₂ O	CH ₃ CN	62	98
6	Bn	(PhO) ₂ POOH	CH ₃ CN	71	99
7	Bn	Sc(OTf) ₃	CH ₃ CN	48	84
8	Bn ^d	(PhO) ₂ POOH ^e	CH ₃ CN ^f	93	>99
9 ^g	Me	(PhO) ₂ POOH	CH ₃ CN	64	>99
10	PMB	La(OTf) ₃	CH ₃ CN	82	>99
11 ^g	Ac	La(OTf) ₃	CH ₃ CN	—	—

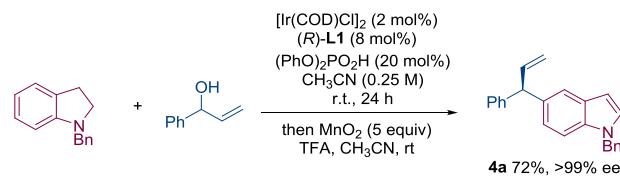
^aReaction conditions: 2 mol % [Ir(COD)Cl]₂, 8 mol % ligand, 20 mol % (PhO)₂POOH, 0.2 mmol of **1a**, and 0.4 mmol of **2a** in solvent at room temperature for 12 h. ^bIsolated yields of **3a**. ^cDetermined by HPLC analysis on a chiral stationary phase. ^dWith 2.5 equiv of **2a**. ^eWith 30 mol % (PhO)₂POOH. ^fReaction concentration of 0.25 M. ^gFor 24 h.

available chiral phosphoramidite ligands [e.g., (*R*)-L2, (*R*)-L3, and (*R*)-L4] with an iridium catalyst did not promote the reaction (entries 2–4). In contrast, a mixture of [Ir(COD)Cl]₂/(*R*)-L1 and BF₃·Et₂O afforded the corresponding AAS product **3a** in 67% yield and 96% ee (entry 1). The reaction could not proceed without acid. The solvent plays an important role in the reaction, and CH₃CN was found to be the best, which gave rise to an ee value of 98% [entries 5–8 (for more details, see **Table S2**)]. Upon investigation of other

acid additives, it was found that the desired product could be afforded in 71% yield and 99% ee in the presence of (PhO)₂POOH [entries 9 and 10 (for more details, see **Table S3**)]. Further investigation of the equivalents of **2a**, acid additive, catalytic loading, and reaction concentration shows that **3a** could be achieved in 93% yield and >99% ee by using 2 mol % [Ir(COD)Cl]₂, 8 mol % (*R*)-L2, and 30 mol % (PhO)₂POOH in CH₃CN (0.25 M) at room temperature for 12 h.

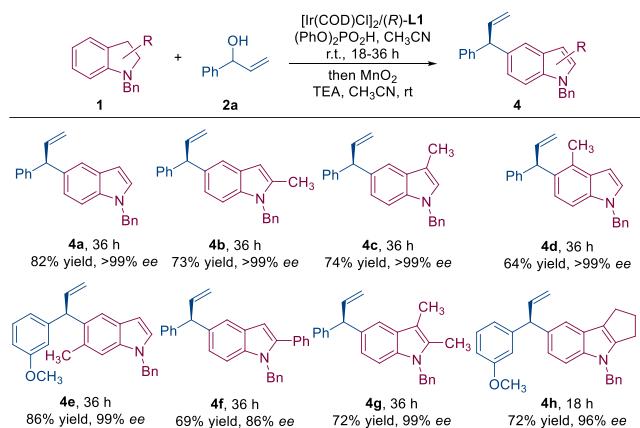
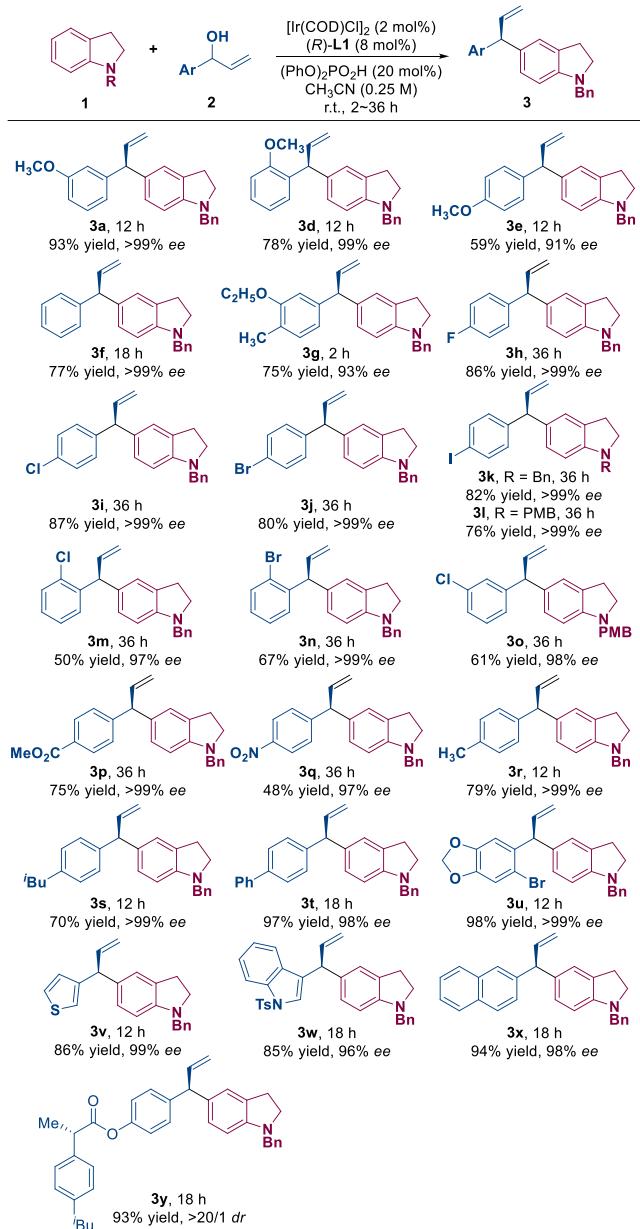
Notably, the methyl- and benzyl-protected indolines were also successful in the reaction, giving the corresponding products in excellent enantioselectivities; however, the yields are lower (entries 12 and 13). Unfortunately, the indolines with electron-withdrawing protecting group, such as acetyl, Boc, alloc, and Ts groups, could not be promoted in the reaction, probably due to the lower nucleophilicity of these substrates (entry 14). Furthermore, the synthesis of C5-allylic indole was tested, and 5 equiv of MnO₂ was added after the completion of the AAS process. Fortunately, desired C5-allylic indole **4a** was obtained in 72% yield without a loss of enantiomeric purity (**Scheme 2**).

Scheme 2. One-Pot Synthesis of C5-Allylindole



To assess the generality of the C5-allylation of indolines, the thus obtained optimal reaction conditions were first applied to diverse substrates. Beyond the parent allylic alcohol, electronically diverse allylic alcohols substituted at the *ortho*, *meta*, or *para* position of the aromatic ring each proceeded smoothly in the AAS reaction, delivering the respective C5-regioselective allylated products with good yields and stereoselectivities (**Figure 1**). It was found that allylic alcohols bearing electron-donating groups on the phenyl ring (**3a**–**3e**, **3g**, and **3r**–**3u**) gave rise to the products in moderate to good yields (59–98%) with excellent enantioselectivities (92% to >99% ee) in less time than with electron-withdrawing substituents (**3h**–**3q**, 48–87% yields and 97% to >99% ee). In addition, the synthetically important ester group and nitro group were compatible in the reaction, providing the corresponding products in excellent enantioselectivities (97% to >99% ee); however, the yields were relatively lower. Furthermore, heteroaromatic ring-substituted allylic alcohols were also well tolerated, providing the desired products (**3v** and **3w**) with good results. When the naphthyl-substituted allylic alcohol was used, the desired **3x** was afforded in 94% yield and 99% ee.

Next, a broad range of *N*-Bn indolines were examined in the AAS reaction followed by oxidation using MnO₂ as the oxidant (**Figure 2**). Indolines bearing electron-neutral and electron-donating substituents, including methyl (**4b**–**4e**), phenyl (**4f**), and methoxy (**4g**) groups at positions C2–C6, underwent facile transformations and afforded the corresponding C5-indole allylation products in excellent yields and enantioselectivities. C2 and C3 disubstituted indoles (**1g** and **1h**) also delivered allylation/oxidation products **4g** and **4h**, respectively, with good results.



Then, the AAS reaction was explored with a series of 1-benzyl-1,2,3,4-tetrahydroquinoline and its analogues (Figure 3). Conveniently, the conditions demonstrated to be optimal

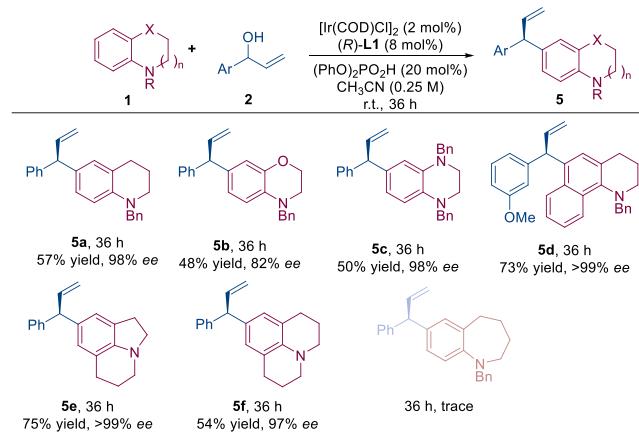
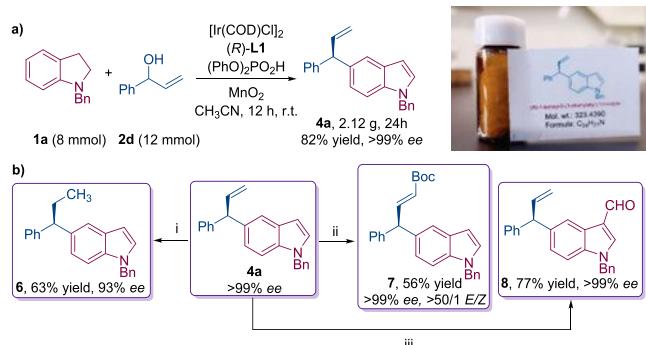


Figure 3. Substrate scope of 1,2,3,4-tetrahydroquinoline analogues.

for N-benzyl indoline extended well to these other nucleophiles, which led to a diverse series of allylation products. For example, when 1-benzyl-1,2,3,4-tetrahydroquinoline, 4-benzyl-3,4-dihydro-2H-benzo[b][1,4]oxazine, 1,4-di-benzyl-1,2,3,4-tetrahydroquinoxaline, and 1-benzyl-1,2,3,4-tetrahydrobenzo[h]quinoline were used, products 5a–5d, respectively, were isolated in moderate to good yields.

To further prove the reliability and practicability of our approach, a scaled-up reaction was carried out by using 8.0 mmol of **1a** as the starting material, affording 2.12 g of **4a** in 82% yield and >99% ee (Scheme 3a). The product thus

Scheme 3. (a) Large-Scale Reaction and (b) Versatile Transformations of C5-Allylindole^a



^aConditions: (i) H₂, Pd/C, MeOH, reflux; (ii) CH₂=CHCO₂tBu, Hoveyda-Grubbs II catalyst, DCM, reflux; (iii) POCl₃, DMF, 0 °C to rt.

obtained could be easily elaborated in different ways. The vinyl group of **4a** could be reduced by Pd/C under a hydrogen atmosphere to give corresponding product **6** in 63% yield, with the enantiomeric excess of the product remaining. Compound **4a** could also undergo an olefin metathesis reaction to form **7** in 56% yield with excellent stereoselectivities. Further transformation of **4a** was also successful through a Vilsmeier–Haack formylation, and the respective aldehyde **8** was afforded in 77% yield without erosion of the optical purity (Scheme 3b). Then, the stereodivergent syntheses of

diallylated products were carried out by using **4a** and *ent*-**4a** as the starting materials; all four stereoisomers (**9a–9d**) were afforded in 83–85% yields in >20:1 dr, and the ee values were all >99% (see Scheme S5). Finally, the AAS/oxidation strategy could be applied in the enantioselective synthesis of dopamine transporter (DAT) inhibitor 5-indolyl aryl propylamine intermediate through a three-step process, and key intermediate **11** could be obtained in 91% yield and 99% ee (see Scheme S6).

In conclusion, we have developed a highly enantioselective protocol for the synthesis of chiral C5-allylindolines by an iridium-catalyzed AAS reaction of allylic alcohols with various N-protected indolines and their analogues. More importantly, various enantioenriched 5-allylindoles could be achieved through a one-pot strategy involving the Ir-catalyzed asymmetric allylic alkylation and subsequent oxidation of *N*-alkyl indolines. This strategy features mild reaction conditions, readily available starting materials, a wide substrate scope, and excellent stereoselectivities and is easily scaled up. Furthermore, the C5-allylindole products can be used as effective linchpins for the streamlined preparation of various synthetically useful building blocks, which makes this protocol potentially useful in organic synthesis.

■ ASSOCIATED CONTENT

SI Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/acs.orglett.1c00810>.

Experimental details, analytical data of products, NMR and HPLC spectra, and plausible transition state structure (PDF)

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Notes

The authors declare no competing financial interest.

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