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Evaluation of a series of BACE 1 inhibitors containing novel heteroaryl-fused-piperazine amidine warheads

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KEYWORDS: BACE inhibitor, β-secretase 1, Alzheimer's disease, cyclic sulfamidates, pKa.

ABSTRACT: Despite several years of research, only a handful of BACE 1 inhibitors have entered clinical trials as potential therapeutics against Alzheimer's disease. The intrinsic basic nature of low MW, amidine-containing BACE 1 inhibitors makes them far from optimal as CNS drugs. Herein we present a set of novel heteroaryl-fused piperazine amidine inhibitors designed to lower the basicity of the key, enzyme binding, amidine functionality. This study resulted in the identification of highly potent (IC₅₀ \leq 10 nM), permeable lead compounds with a reduced propensity to suffer from Pgp-mediated efflux.



Genetics has suggested β -secretase 1 (BACE 1) as a key target to attenuate the formation of amyloid beta $(A\beta)$ in Alzheimer's disease (AD).¹ Since the discovery of BACE 1 in 1999,² many potent BACE 1 inhibitors have been described,³ but relatively few successfully combine in vitro potency and the necessary pharmacokinetic (PK) properties/parameters to achieve in vivo efficacy. The amidine class of BACE 1 inhibitors have become the main focus.^{3,4} and contain a protonated amidine warhead necessary for an optimal interaction with the aspartyl catalytic dyad of BACE 1.3 The high intrinsic basicity of amidine heterocycles presents multiple liabilities for their development, including high Pgp efflux and hERG inhibition, and low permeability. Several compounds have unfortunately failed clinical trials, including MK-8931 $(1)^5$ and LY-2886721 (2)(Figure 1).⁶

Figure 1. Representative BACE 1 inhibitors.



Our interest in amidine BACE 1 inhibitors can be traced back to early reports of aminodihydroquinazolines.⁷ Since then we have been dedicated to identifying novel warheads with adequate substitution to control the pKa of the guanidine/amidine and as such balance the liabilities with in vivo efficacy. Thus, the challenge of BACE 1 inhibition is revealed: liabilities are inherent to the amidine, yet it is crucial for key binding motifs. To be capable of identifying BACE 1 clinical candidates, there is a need to efficiently synthesize and evaluate many novel but synthetically challenging amidine containing heterocycles. Here we describe a convergent approach to meet this goal.

We have previously described our approach to assess new candidate warheads with computational approaches leading to series of cyclic amidines,⁸, acylguanidines,⁹, and aminopiperazinones such as **3** (Figure 1).¹⁰ The aminopiperazinones suffered from poor brain penetration which halted their progress. However, they served as inspiration for further research, such as a series of aminomorpholine containing BACE 1 inhibitors. The morpholine series, exemplified by JNJ-50138803, (4) (Figure 1) was also supported by *in-silico* work, and their optimization has been described.^{11,12} The initial prototypes, displayed high pKa values (9 < pKa < 9.6) and a strong Pgp-mediated efflux liability.¹¹ To attenuate the issues attributed to the elevated pKa, the amidine warhead was carefully modified with fluorine containing electron withdrawing substituents at position C-2 of the morpholine ring. This optimization focused on achieving a pKa within the range of 6.5-7.5. The evolution of the series ultimately culminated with the identification of 4 (Figure 1), a recently disclosed clinical candidate.12

Besides the exploration of the morpholine series, here we described a parallel effort using bioisosteric replacements of the amide in the aminopiperazinone warhead (3). We focused on replacing this amide functional group by nitrogen-containing fused five membered heteroaromatic rings, where a bridge head nitrogen was maintained.

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To validate our approach, we initially synthesized the imidazolopiperazine warhead (5) with identical decoration to the reported aminopiperazinone (Table 1). Similar potency in both the BACE 1 cellular and enzymatic assays was observed giving an immediate indication this could offer a promising avenue for deeper exploration. Therefore, an efficient and versatile synthetic route based on the ring opening of a cyclic sulfamidate intermediate (8) was developed, enabling the synthesis of the bicyclic warheads X1 - X8, shown in Figure 2.

 Table 1. BACE 1 cellular data for aminopiperazinone 3 and imidazolopiperazine 5.



Figure 2. Proposed aminopiperazinone bisosteric replacements X1-X8 from common sulfamidate intermediate.



Chemistry: Cyclic sulfamidates have emerged as important and versatile intermediates in organic synthesis because of their established ability to undergo highly selective reactions with O-, S-, N-, C-, P- and F-based nucleophiles.¹³ We reasoned that nitrogen-containing heteroaromatics could be suitable nucleophiles to ring open adequately substituted sulfamidates leading to intermediates which could then be further elaborated to yield the targeted warheads (Figure 2).

Key cyclic sulfamidate 8 was by Boc protection of the β -amino alcohol 6,11 followed by cyclisation with thionyl chloride in acetonitrile at -40 °C afforded the corresponding sulfamidite 7 in 93% yield over 2 steps. This was then oxidized to the target sulfamidate **8** using NaIO₄ and a catalytic amount of RuCl₃ in acetonitrile/water with excellent yield (Scheme 1).¹⁴

Scheme 1. Synthesis of cyclic sulfamidate 8.



Reagents and conditions: (a) $(Boc)_2O$, NaHCO₃ aq, THF, rt, 2.5 d; (b) SOCl₂, MeCN, 2 h, rt (93% yield over 2 steps); (c) RuCl₃, NaIO₄, CAN, MeCN/H₂O, 2 h, rt; (95% yield).

With sulfamidate 8 in hand, we studied its nucleophile induced ring opening reaction with several nitrogen-containing heteroaromatics. Sulfamidate 8 was successfully ring opened using methyl 1H-pyrrole-2-carboxylate in the presence of cesium carbonate in acetonitrile, to afford 9 in 84% yield (scheme 2). The resulting intermediate 9 was then deprotected and cyclized by sequential treatment with HCl in dioxane, followed by sodium methoxide to afford lactam 10, in 22% yield for the two steps. Lactam 10 was then elaborated to the targeted warhead by initial alkylation to form an imino-ether intermediate, and subsequent reaction with ammonia in EtOH to form the desired amidine 11, in 33% yield over the two transformations. While there are many techniques to convert arylbromides to the corresponding anilines, and alternatives are described vide infra, on this occasion we used a Büchwald coupling using the benzophenone imine; which was subsequently hydrolyzed to the corresponding aniline 12. Subsequent chemo-selective coupling of the carboxylic acid was mediated by 4-(4,6-dimethoxy-1,3,5-triazin-2-yl)-4methylmorpholinium chloride (DMTMM).¹⁵ This reagent is designed to activate the carboxylic acid and concomitantly protonate the most basic center, subsequently promoting the chemo-selective amide formation on the least basic center, giving 13, in 50% yield.¹⁵

Scheme 2. Synthesis of pyrrolopiperazine BACE 1 inhibitor 13.¹⁶



Reagents and conditions: (a) Cs_2CO_3 , MeCN, 130 °C MW, 30 min, (84% yield); (b) HCl (4 N in 1,4-dioxane), rt, 18 h; (c) MeONa, MeOH, 60 °C, 16 h, (22% yield over 2 steps); (d) OMe₃BF₄, DCM, rt, 4 d; (e) NH₄Cl, NH₃ (2 M in MeOH), EtOH, 80 °C, 48 h (33% yield from **10**); (f) Pd₂(dba)₃, BINAP, Na'BuO, benzophenone-imine, 100 °C, 2 h; (g) HCl (6 N in iPrOH) 1 h, rt; (h) 5-cyanopyridine-2-carboxylic acid, DMTMM, MeOH, 0 °C to rt, 3 h, (50% yield from **11**).

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The regioisomeric imidazolepiperazine warheads 5, 17 and 21 were also accessed using cyclic sulfamidate 8. Reaction of 2cyanoimidazole with 8 in the presence of DBU afforded 14 in 97% yield Scheme 3). HCl promoted cleavage of the Boc group in 14 occurred with concomitant cyclisation to amidine 15, in 71% yield. The aryl bromide was then converted to aniline 16 using sodium azide and copper iodide in a solution of dimethylene ethylene diamine and DMSO in 75% yield. The regioselective amide formation was again achieved using DMTMM to afford the final products 5 and 17 in 45% and 57% yield, respectively.

Scheme 3. Synthesis of imidazolopiperazine BACE 1 inhibitors 5 and 17.



Reagents and conditions: (a) DBU, ACN, 90 °C, 16 h, (97% yield); (b) HCl (4 N in 1,4-dioxane), 70 °C, 4 h, (71% yield); (c) NaN₃, DMEDA, CuI, Na₂CO₃, DMSO, 110 °C, 24 h, (75% yield); (d) (**5**) 5-Chloropyridine-2-carboxylic acid or 5-cyanopyridine-2-carboxylic acid, DMTMM, MeOH, 0 °C to rt, 24 h, (45-57% yield).

For the preparation of **21**, due to the lack of symmetry and difference in reactivity, ethylimidazole-4-carboxylate could not be used to open the cyclic sulfamidate. Instead, reaction of **8** with symmetrical diethyl_1H-imidazole-4,5-dicarboxylate yielded the expected product in 88% yield (Scheme 4). Subsequent Boc deprotection, followed by sodium methoxide mediated intramolecular cyclisation afforded lactam **18** (55% yield over 3 steps). Decarboxylation by treatment with NaOH and heating in DMSO led to **19**. Treatment of **19** with phosphorus pentasulfide to the thio-lactam and subsequent S-methylation followed by treatment with ammonia yielded amidine **20**. Target compound **21** was obtained from **20** following analogous conditions to those described in Schemes 2 and 3.¹⁷

Scheme 4. Synthesis of imidazolopiperazine BACE 1 inhibitor 21.



Reagents and conditions: (a) DBU, ACN, 90 °C, 5 h, (88% yield); (b) HCl (4 M in 1,4-dioxane), 1,4-dioxane, 70°C, 15 h; (c) NaOMe,

MeOH, 55 °C, 5 h, (55% yield over 3 steps); (d) 1 N NaOH, MeOH, rt, 1 h; (e) DMSO, 120 °C, 2 h (62% yield over 3 steps); (f) P_2S_5 , pyridine, 110 °C, 16 h; (g) MeI, K_2CO_3 , acetone, 16 h, rt; (h) NH₄Cl, NH₃ (7 M in MeOH), 80 °C, 48 h, (35% yield over 3 steps); (i) NaN₃, DMEDA, CuI, Na₂CO₃, DMSO, 110 °C, 4 h; (j) 5-cyanopyridine-2-carboxylic acid, DMTMM, MeOH, 0 °C to rt, 24 h, (33% yield over 2 steps).

Pyrazolopiperazine analog 26 could be prepared by reaction of cyclic sulfamidate 22^{18} with 4-bromopyrazole; this was achieved in good yield using cesium carbonate in dimethyl sulfoxide to yield 23 (Scheme 5). Cyclic sulfamidate 22 was synthesized in an analogous way to cyclic sulfamidate 8, previously described in Scheme 1, albeit racemically, thus requiring that the final compound undergo chiral separation, which in the case of 26 was achieved using chiral SFC. Intermediate 23 was deprotonated with LDA and reacted with ethylchloroformate to form the lactam precursor. This was then cyclized, after TFA-mediated Boc deprotection, by treatment with KOAc in refluxing ethanol. Lactam 24 was elaborated to the amidine via generation of the thioamide and subsequent reaction with ammonia. The aromatic nitro group in 24 was reduced to the corresponding aniline using heterogeneous catalysis which went with concomitant bromide removal to afford 25 in excellent yields over three reaction steps (94%). The chemoselective amide formation was facilitated by DMTMM to afford the desired amide 26 in 10% yield after chiral separation.

Scheme 5. Synthesis of pyrazolopiperazine BACE 1 inhibitor 26.



Reagents and conditions: (a) Na_2CO_3 , DMSO, 130 °C, 2 h, (79% yield); (b) ethylchloroformate, LDA, THF, -78 °C to -50 °C, 2 h (66% yield); (c) TFA, DCM, rt, 3 h; (d) KOAc, EtOH, reflux, 4 h, (87% yield over 2 steps); (e) P_2S_5 , pyridine, 110 °C, 24 h; (f) NH₄Cl, NH₃ (7 M in MeOH), 100 °C, 16 h; (g) Zn, AcOH, EtOH, reflux, 24 h, (37% yield over 3 steps); (h) 5-cyanopyridine-2-carboxylic acid, DMTMM, MeOH, 0 °C to rt, 4 h; (i) Chiral SFC purification, (10% yield over 2 steps).

To access 1,2,4-triazolopiperazine **30**, reaction of cyclic sulfamidate **8** with 1,2,4-triazole-3-carbonitrile afforded **27** in 92% yield (Scheme 6). This was then elaborated to the final compound following a very similar sequence to that previously described in Scheme 4 for the imidazolopiperazine analog **21**. This started with formic acid-mediated Boc cleavage and subsequent intramolecular lactamization facilitated by trimethyl aluminum. Then, amination of the bromophenyl ring in **28** by treatment with sodium azide and CuI afforded **29** in 92% yield. Regioselective amide formation led to the desired BACE 1 inhibitor **30** in 52% yield.

Scheme 6. Synthesis of 1,2,4-triazolopiperizine BACE 1 inhibitor 30.



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Reagents and conditions: (a) KOtBu, 60 °C, 1,4-dioxane, 16 h, (99% yield); (b) HCO₂H, rt, 2 h; (c) AlMe₃, toluene, 60 °C, 1 h, (53% yield over 3 steps) (d) NaN₃, DMEDA, Na₂CO₃, CuI, DMSO, 100 °C, 2 h, (50% yield); (e) 5-cyano-pyridine-2-carboxylic acid, EDCI, HCl (6 N in iPrOH), MeOH, rt, 2 h, (52% yield).

1,3,4-Triazolopiperazine 35 was accessed via ring opening of cyclic sulfamidate 8 with sodium azide to afford alkylazide 31, which, after subsequent hydrogenation, afforded diamine 32, with concomitant and undesired bromide removal (Scheme 7). Intermediate 32 then treated was with N.N'bis(dimethylaminomethylene)hydrazine in the presence of a catalytic amount of p-TSA to afford the desired 1,3,4-triazole 33. The introduction of a nitrile functional group in the heteroaromatic ring was achieved in a 4-step sequential sequence. First, a hydroxymethyl substituent was introduced with formaldehyde in the presence of triethylamine; this primary alcohol was then selectively oxidized to the corresponding aldehyde using manganese dioxide. Finally, aldehyde condensation with hydroxylamine followed by dehydration of the intermediate oxime using oxalyl chloride afforded the desired nitrile containing intermediate 34. Boc deprotection of 34 under acidic conditions and concomitant cyclisation led to the expected cyclic amidine intermediate. Nitration of the aromatic ring under standard conditions, followed by hydrogenation of the nitro group resulted in the corresponding aniline intermediate. Final amide coupling with 5-cyanoyridine-2-carboxylic acid afforded BACE 1 inhibitor 35 in 58% yield from nitrile containing intermediate 34.

Scheme 7. Synthesis of 1,3,4-triazolo-piperizine BACE 1 inhibitor 35.



Reagents and conditions: (a) NaN₃, DMF, 80 °C, 2 h; (quantitative); (b) H₂, Pd/C, TEA, MeOH, rt, 16 h; (87% yield) (c) p-TSA, toluene, 130 °C, 16 h, (47% yield) (d) CH₂O, TEA, 100 °C, 2 h; (e) MnO₂, CHCl₃, rt, 1 h; (f) NH₂OH·HCl, NaOAc, EtOH, 80 °C, 1.5 h; (g) CO₂Cl₂, TEA, EtOAc, 0 °C to rt, 2 h, (31% yield over 4 steps); (h) HCO₂H, 50 °C, 1 h; (i) HNO₃, H₂SO₄-10 °C, 30 min; (j) H₂, Pd/C, MeOH, rt, 4 h; (k) 5-cyanopyridine-2-carboxylic acid, EDCl, HCl (6 N in iPrOH), MeOH, rt, 16 h, (58% yield over 4 steps).

The 1,2,3-triazolopiperazine analogue **39** was prepared from intermediate **31** as outlined in Scheme 8. Boc deprotection of the amine group in **31** followed by coupling with propiolic acid led to amide **36**; the resulting intermediate was double cyclized under thermal conditions to afford **37**. Lactam **37** was elaborated to amidine **38** following a similar procedure as described in Scheme 5. Subsequent chemoselective amide coupling with 5-cyano-pyridine-2-carboxylic acid led to BACE 1 inhibitor **39**.¹⁹

Scheme 8. Synthesis of 1,2,3-triazolo-piperizine BACE 1 inhibitor 39.



Reagents and conditions: (a) HCl (4 N in 1,4-dioxane), rt, 24 h; (b) propiolic acid, DCC, DCM, 0 $^{\circ}$ C, 2 h; (c) toluene, 70 $^{\circ}$ C, 18 h; (78% yield over 3 steps); (d) P₂S₅, 1,4-dioxane, 95 $^{\circ}$ C, 2 h; (e) NH₃ (7 M in MeOH) 75 $^{\circ}$ C, 2 h; (f) NaN₃, DMEDA, Na₂CO₃, CuI, DMSO, 110 $^{\circ}$ C, 5 h; (g) 5-cyanopyridine-2-carboxylic acid, EDCI, HCl (6 N in iPrOH), MeOH, rt, 3 h, (39% yield over 4 steps).

Compound 43 containing a fused tetrazole ring was again generated from intermediate 31, as shown in Scheme 9. Intermediate 31 initially underwent a 1,3-dipolar cyclisation with ethyl cyanoformate to afford tetrazole 40. Acid-mediated Boc deprotection, followed by cyclisation gave lactam 41. Lactam 41 was converted to the corresponding amidine 42, as described in Schemes 5 and 7. Bromide 42 was converted to the aniline, which underwent chemoselective amide formation to yield BACE 1 inhibitor 43.

Scheme 9. Synthesis of tetrazolo-piperizine BACE 1 inhibitor 43.



Reagents and conditions: (a) Ethylcyanoformate, 120 °C, 20 h; (b) TFA, DCM, rt, 2 h; (c) K_2CO_3 , EtOH, reflux, 20 h, (75% yield over 3 steps); (d) P_2S_5 , 1,4-dioxane, 95 °C, 2 h; (e) NH₃, (7 M in MeOH), 70 °C, 18 h; (f) NaN₃, DMEDA, Na₂CO₃, CuI, DMSO, 110 °C, 2 h; (g) 5-cyanopyridine-2-carboxylic acid, EDCI, HCl (6 N, iProH), MeOH, rt, 2h, (5% yield over 4 steps).

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The resulting eight bicyclic heteroaryl-fused aminopiperazine BACE 1 inhibitors **13**, **17**, **21**, **26**, **30**, **35**, **39**, **43** provide an excellent opportunity for the side-by-side comparison of the effect of the fused heterocycle. They were profiled for their BACE 1 and 2 inhibitory activity in both our enzymatic assay and also cellular BACE1 assay, Table 2. Cathepsin D was not measured for this series of compound as it known that the expansion to the S3 pocket via the amide linked heteroaromatic removes the Cathepsin D liability.²⁰

Most compounds showed high cellular potency in the range of pIC_{50} 7.7-8.9. The low activity observed for the tetrazole derivative **43** (hAβ42 cell pIC_{50} = 7.01) was attributed to the low basicity of the amidine function (pKa = 5.1). In general, we have observed a dependency for BACE 1 cellular potency with pKa. This is rationalized by the BACE 1 cellular location in the endosomes, an acidic compartment that favors basic compound distribution.²¹

17 Our working hypothesis is that the optimal pKa for a BACE 1 18 inhibitor should be in the range of 6.5-8.5. Higher pKa's are 19 tolerated and, in some cases, result in increased cellular 20 potency. As demonstrated with compounds 13 and 43, 21 compound 13 has a pKa of 10.3 and this is translated into a cellular potency of pIC_{50} of 8.88, which compound 43 has the 22 lowest pKa of 5.1 and has a cellular potency of pIC₅₀ of 7.1. 23 However, the higher basicity can also be responsible for off-24 target related complications, such as hERG and Pgp-efflux 25 liabilities, this is not always the case and Pgp liability is not 26 resultant on one factor and multiple properties of the molecule 27 can affect this value. The measured pKa for compounds 13 (pKa 28 = 10.3) and 21 (pKa = 9.0) was found to be higher than the 29 optimal range. This high basicity could contribute to the high 30 Pgp efflux observed for both compounds (13, AB+/AB = 4.6and 21, AB+/AB = 5.3), but a clear outlier is compound 39 that 31 has an ideal pKa and suffers from significant Pgp efflux. We 32 did see in this series of compounds a clear correlation between 33 hERG and pKa, with compound 13 with the highest pKa having 34 a significant hERG liability and compound 43 not showing any 35 hERG inhibition. Therefore, further work around these warhead 36 modifications should aim at lowering the high pKa of the 37 amidine function and modulating the vitro and vivo profile. 38

39 The crystal structure of 13 with BACE 1 was solved at 2 Å 40 resolution (PDB code 6OD6, Figure 3). The result confirmed the binding mode to be similar to those seen for analogous 41 amidine containing BACE 1 inhibitors.^{11,22} This includes the 42 key interactions between the protonated amidine and the 43 catalytic dyad (Asp32 and Asp228). The quaternary center is 44 optimal for the orientation of the aryl ring and the methyl group, 45 allowing the methyl to protrude into the S2' pocket without 46 clashing with the protein. The quaternary center favorably 47 positions the F-aryl ring in an axial orientation to occupy the S1 48 pocket. The amide linker then optimally orientates the pyridine 49 ring in the S3 pocket of BACE 1, while generating a favorable 50 H-bond interaction with Gly230. Most of the compounds have a similar enzymatic activity with pIC₅₀ approx. 8-8.5, with 51 examples 17 and 39 showing the biggest difference: pIC₅₀ 7.54 52 and 8.70 respectively. Comparing pairs 13 and 17, 21 and 35, 53 21 and 35, 26 and 30, 39 and 43, all demonstrate that 54 incorporation of a nitrogen at position 2' led to a loss of activity. 55 The crystal structure suggests that this nitrogen points towards 56 the oxygen of Thr231, and is therefore electrostatically 57

unfavorable explaining the lower activity. The rest of the fused cycle is solvent exposed. Although not attempted here, additional substituents to the fused bicycle may allow further modulation of the potency by forming additional interactions with the receptor.

Figure 3. Crystal structure of 13 in BACE 1 showing key interactions.



Compounds **17**, **26**, **30**, **35** and **39** possess pKas in an almost optimal range between 6.4-8.5 and show good to excellent intrinsic permeability.

Unfortunately, despite their good pKa's, the Pgp-efflux liability measured for **35** (AB+/AB- = 4.8) and **39** (AB+/AB- = 9.8) precluded further investigation around these two chemotypes. On the other hand, compounds **17**, **26** and **30** exhibit the desired in vitro profile combining potency, with good intrinsic permeability a reduced Pgp-efflux and hERG liability and the heteroaryl-fused aminopiperazine warheads they represent have been selected for further profiling and optimization.





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Compound #	13	17	21	26	30	35	39	43
х	F	F	F	Н	F	F	F	F
BACE1 enz pIC ₅₀	8.21	7.54	8.40	8.34	8.22	7.97	8.70	8.04
hAβ42 cell pIC ₅₀ nM	8.88	8.20	7.74	8.66	8.01	8.18	8.92	7.01
BACE2 enz pIC ₅₀	7.60	6.85	7.72	7.58	7.43	7.25	8.26	7.57
hERG IC50 µM	0.49	4.17	3.72	1.78	11.75	>10	7.24	>50
Pgp								
AB+	11.1	17.2	1.5	15.6	24.8	5.9	11.6	22.1
AB+/AB-	4.6	2.0	5.3	3.6	2.0	4.8	9.8	1.9
рКа	10.3	8.2	9.0	8.5	6.5	6.4	7.1	5.1

See supplementary information for assay details

In summary, an efficient evaluation of multiple novel BACE 1 inhibitors has been achieved. Exploiting the reactivity of the

cyclic sulfamidates we have developed a convergent synthesis to explore the replacement of the aminopiperazinone warhead by bioisosteric replacements in the form of fused heteroaromatics. Several new series of BACE 1 inhibitors have been identified, and efforts are currently focused on the optimization of properties and progression of key compounds. These BACE 1 inhibitors represent a unique opportunity, as the fused 5-membered heterocyclic ring offers additional handles for modification to enhance their overall profile.

ASSOCIATED CONTENT.

Supporting Information.

The Supporting Information is available free of charge on the ACS Publications website.

Experimental procedures for the synthesis and the analytical characterization of all intermediates and final compounds are included.

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The manuscript was written through contributions of all authors. All authors have given approval to the final version of the manuscript.

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