

Intramolecular Mizoroki–Heck Reaction in the Regioselective Synthesis of 4-Alkylidene-tetrahydroquinolines

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The Mizoroki–Heck reaction of *N*-alkenyl-substituted 2haloanilines is an effective protocol for the synthesis of substituted 4-alkylidene-tetrahydroquinoline derivatives, avoiding isomerization and oxidation. When non-substituted alkenes are used, the regioselectivity of the reaction can be directed towards the formation of an exocyclic or endocyclic

Introduction

Quinoline derivatives are an important class of compounds in the pharmaceutical and agrochemical industries, as well as building blocks for the total synthesis of natural products.^[1] For example, several 2-aryl-tetrahydroquinolines and their aromatic counterparts have shown antifungal and anticancer activity,^[2] whereas 1,2-dihydro derivatives as ethoxyquin are known by their antioxidant effects.^[3] 4-Alkylidene-tetrahydroquinoline derivatives have attracted much attention as in vivo potent antagonists of the glycine binding site associated with the N-methyl-D-aspartic acid receptor, a target for innovative drugs to treat chronic neuropathic pain. The presence of an $exo \alpha,\beta$ -unsaturated amide moiety at the C-4 position of these compounds was found to play a pivotal role in their bioactivities.^[4] In addition, many naturally occurring alkaloids^[5] contain the tetrahydroquinoline or quinoline unit with the exocyclic double bond (Figure 1).^[6]

Classical methods for quinoline synthesis often do not allow adequate diversity and substitution on the quinoline ring system.^[7] Therefore, considerable efforts have been dedicated to developing new methods for the preparation of these pharmacophores: from acid promoted cycloaddition reactions to ruthenium or palladium-catalyzed methodologies.^[8] Within that last category, we have reported that intermolecular palladium-catalyzed arylation followed by

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double bond by choosing an adequate catalytic system. When the double bond is substituted by an amide moiety, the exocyclic double bond of *E* geometry is obtained selectively. Thus, this protocol efficiently synthesizes a series of 2-substituted tetrahydroquinolines with an $exo \alpha_{n}\beta$ -unsaturated amide moiety at the C-4 position.

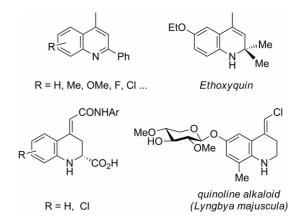


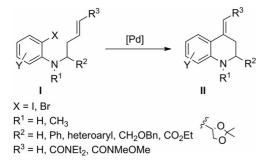
Figure 1. Selected bioactive quinoline derivatives.

Grignard addition to imines, and ring-closing metathesis provides an efficient approach to 2-phenyl-substituted quinoline, benzazocine, and benzazepine derivatives.^[9] 2,4-Substituted tetrahydroquinolines can be obtained through intramolecular carbolithiation of *N*-alkenyl-substituted 2iodoanilines, even enantioselectively.^[10]

However, the high reactivity associated with organolithium intermediates narrows the functional group tolerance for this procedure. Thus, although a variety of synthetic methods for preparing quinoline derivatives are available, it is still highly desirable to develop efficient procedures to obtain useful polyfunctionalized quinolines. In this context, palladium-catalyzed coupling reactions^[11] and, in particular, the intramolecular Mizoroki–Heck reaction^[12] is of special relevance. These reactions have found wide application in synthesis for preparing complex organic structures,^[13] even asymmetrically,^[14] and are also applied in the chemical and pharmaceutical industries.^[15]



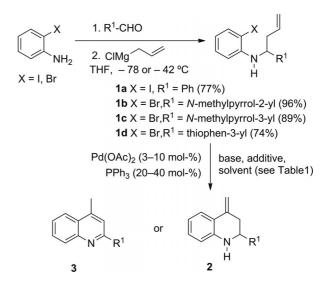
With these precedents, we envisioned that the Mizoroki-Heck reaction of N-alkenyl-substituted 2-haloanilines could lead to an effective protocol for the synthesis of substituted 4-alkylidene-tetrahydroquinoline derivatives II. Thus, we prepared a series of butenylanilines I, with different substitution patterns in the alkene and α to the nitrogen atom (Scheme 1), that would lead to tetrahydroquinolines II. This kind of approach has been studied before, but over-oxidation reactions or formation of regioisomeric mixtures have been observed. Pd-catalyzed cyclization of I (Y, R^1 , R^2 , $R^3 = H$) was previously reported by Larock^[16] and 4methylquinoline was obtained through 6-exo cyclization followed by double bond migration and oxidation. Starting from the same substrate, in the presence of Pd/N-heterocyclic carbene catalyst, Caddick^[17] observed the formation of a mixture of isomers through 6-exo or 7-endo cyclization and double bond migration. Thus, the main issue would be to obtain a regioselective procedure, avoiding over-oxidation and double-bond isomerization reactions that may occur in this type of reaction. Here we present a full account of our investigations.



Scheme 1. Mizoroki-Heck approach to 4-alkylidene-substituted quinolines.

Results and Discussion

It has been shown that the presence of a heterocyclic ring substituent (\mathbf{R}^2) is one of the main structural requirements for the biological activity of homoallylamines with general structure I and their quinoline or tetrahydroquinoline derivatives (II).^[18] For this reason we started our study preparing a series of secondary butenylanilines 1a-d, which carry a phenyl or heteroaryl group α to the nitrogen atom, in high overall yields, by condensation of *o*-iodo or *o*-bromoaniline and the corresponding aldehyde, followed by addition of allylmagnesium chloride (Scheme 2).^[19] When 1a was treated under phosphane-free Jeffery conditions,^[20] (Table 1, Entry 1), 4-methyl-2-phenylquinoline 3a was obtained in moderate yield after 6-exo cyclization, isomerization and oxidation, as has been described for o-iodobutenylaniline under analogous conditions.^[16] Various reaction conditions were tested to avoid this isomerization and oxidation. A change in the solvent to acetonitrile (Table 1, Entry 2) gave a sluggish reaction, obtaining also 3a as the main product, in moderate yield. However, in the presence of PPh₃, 4-methylene-trahydroquinoline 2a was obtained (Table 1, Entry 3).



Scheme 2. Synthetic route to compounds 2 and 3.

Table 1. Synthesis of 2-aryl-substituted 4-methylene-quinolines 2a–d.

	Substrate	Base, additive	Solvent	2 or 3 [%]
1	1a ^[a]	Na ₂ CO ₃ , Bu ₄ NBr	DMF	3a , 50
2	1a ^[a]	Na ₂ CO ₃ , Bu ₄ NBr	CH ₃ CN	3a , 33
3	1a ^[b]	Na ₂ CO ₃ , Bu ₄ NCl	CH ₃ CN	2a , 54
4	1b ^[a]	Na_2CO_3 , Bu_4NCl	DMF	[d]
5	1b ^[b]	Na ₂ CO ₃ , Bu ₄ NCl	CH ₃ CN	2b , 71
6	1b ^[c]	Bu ₄ NOAc	DMSO	2b , 49
7	1c ^[b]	Na ₂ CO ₃ , Bu ₄ NCl	CH ₃ CN	[e]
8	1c ^[c]	Bu ₄ NOAc	DMSO	2c , 55
9	1d ^[b]	Na ₂ CO ₃ , Bu ₄ NCl	CH ₃ CN	2d , 18
				3d , 30
10	1d ^[c]	Bu ₄ NOAc	DMSO	2d , 14

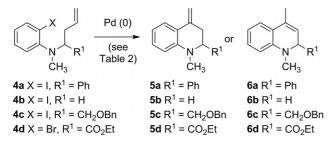
[a] 3 mol-% Pd(OAc)₂, no PPh₃ was added. [b] 3 mol-% Pd(OAc)₂, 20 mol-% PPh₃. [c] 10 mol-% Pd(OAc)₂, 40 mol-% PPh₃. [d] Starting material was recovered. [e] Mixture of products **2c** and **3c** were detected by NMR spectroscopy.

These conditions were applied to α -heteroaryl-substituted amines **1b–d**. Although **1b** gave expected product **2b** in good yield (Table 1, Entry 5), oxidation could not be avoided with **1c** and **1d**, leading to low isolated yields of tetrahydroquinoline **2d**, or a mixture of products in the case of **1c** (Table 1, Entries 7 and 9). However, the use of Bu₄NOAc and dimethyl sulfoxide (DMSO) with a larger amount of phosphane, improved the results with **1c** (Table 1, Entry 8). These conditions would favor a cationic pathway for the alkene insertion, and have been shown to favor the direct arylation reaction of the pyrrole nucleus.^[21] However, no arylation products of the pyrrole or the thiophene ring were observed with **1b–d**.

We next focused our attention on the reaction of tertiary anilines. For this purpose, we chose butenyl anilines 4a-d, that incorporate different functionality on the α position to nitrogen (Scheme 3). By using a similar approach, amines 4a, 4c, and 4d were prepared by condensation with the corresponding aldehyde, followed by allylation of the imine,

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and subsequent methylation. Amine **4b** was obtained by sequential alkylation reactions.^[19]



Scheme 3. Synthesis of tetrahydroquinolines 5 and 1,2-dihydroquinolines 6.

Under Pd⁰ conditions, the substitution on the nitrogen atom on amines **4a–d** would avoid the formation of aromatic quinolines, but the regioselectivity of the formation of the double bond would still be an issue (Scheme 3). Thus, 1,2-dihydroquinolines **6**, with a more stable endocyclic double bond, could be formed by isomerization of the exocyclic double bond of tetrahydroquinolines **5** by reinsertion of the hydropalladium species formed after β -elimination, followed by a second elimination. Optimization of the reaction conditions led to the selective formation of each regioisomer, as shown in Table 2.

When butenylanilines **4a–d** were treated with $Pd(OAc)_2$ in the presence of a base and tetrabutylammonium halide, under various conditions, the reaction was non selective, obtaining mixtures of 4-methylenetetrahydroquinolines **5** together with isomerized **6** (Table 2, Entries 1, 6, 10 and 14). Phosphane-free conditions led to sluggish reactions, and a mixture of isomers (Table 2, Entries 2, 11 and 15). However, isomerization could be avoided by the addition of silver salts.^[22] In the presence of silver carbonate, tetrahydroquinolines **5a–d** were formed regioselectively in good yields (Table 2, Entries 3, 7, 12 and 16). We also found that quinolines **5a** and **5b** could be selectively obtained without the use of silver additives, by using $Pd(dba)_2/Ph_3P$ in the pres-

Table 2. Mizoroki-Heck reactions of 4a-d.

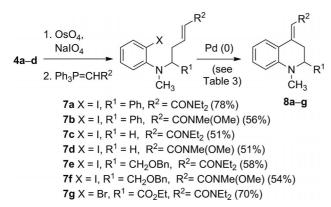
ence of tetrabutylammonium acetate and allyltrimethylsilane, with lower catalyst loading (3 mol-%; Table 2, Entries 4 and 8). The presence of allylsilane is relevant because the reaction does not proceed without this additive. In fact, the presence of dibenzylideneacetone has been shown to decrease the rate of oxidative addition of aryl iodides when Pd(dba)₂/PPh₃ systems are employed.^[23] Although the formation of palladium complexes with allysilanes is known,^[24] the use of allyltrimethylsilane as an additive in Mizoroki-Heck reactions has not been described to the best of our knowledge, and its role has not been studied in detail. Dihydroquinolines 6a-d were selectively obtained in good yields by using Pd(PPh₃)₄/Et₃N as catalytic system (Table 2, Entries 5, 9, 13 and 17), though the reaction requires high catalyst loadings to reach reasonable yields. Thus, the Mizoroki-Heck reaction allowed the selective synthesis of quinolines with exocyclic or endocyclic double bonds, and it is compatible functional groups on C-2, like an ester (5c and 6c) or a protected alcohol (5d and 6d).

To access tetrahydroquinolines with an *exo* α,β -unsaturated amide moiety at the C-4 position, a series of tertiary butenylanilines with *trans*-substituted double bonds were prepared. Thus, **7a**–g were obtained in good yields diastereoselectivity from **4a–d** through oxidative cleavage of the double bond followed by a Wittig reaction in a one-pot procedure,^[4b,19] as shown in Scheme 4.

When optimized reaction conditions [Method A: $Pd(OAc)_2/PPh_3$, AgCO₃, DMF, 90 °C] were applied to **7a** and **7b** the corresponding 2-phenyl-substituted quinolines **8a,b** were obtained regioselectivity in good yield (Table 3, Entries 1 and 2). No formation of the *endo* double bond was detected, and the tetrahydroquinolines were obtained as single *E* isomers. The geometry of the double bond, which was confirmed by NOESY experiments, is consistent with a *syn* β -hydride elimination. These conditions were extended to the synthesis of **8c**–**g**, with consistently good yields (Table 3, Entries 3–7). The same regioisomers were obtained when Pd(PPh_3)_4/Et_3N was used as catalytic system

	Substrate	Pd	PPh ₃ [mol-%]	Base, additive	Solvent	Temperature	5 [%]	6 [%]
1	4 a	Pd(OAc) ₂ ^[a]	20	Na ₂ CO ₃ , Bu ₄ NBr	DMF	90 °C	46	4
2	4a	$Pd(OAc)_2^{[a]}$	-	Na_2CO_3 , Bu_4NBr	DMF	90 °C	35	4
3	4a	$Pd(OAc)_2^{[b]}$	30	Ag_2CO_3	DMF	90 °C	78	_
4	4a	$Pd(dba)_2^{[a]}$	14	Bu ₄ NOAc, allylTMS	DMF	50 °C	75	_
5	4a	$Pd(PPh_3)_4^{[c]}$	-	Et ₃ N	toluene	reflux	_	76
6	4 b	$Pd(OAc)_2^{[b]}$	30	NaHCO ₃ , Bu ₄ NCl	DMF	90 °C	[d]	
7	4 b	$Pd(OAc)_2^{[b]}$	30	Ag_2CO_3	DMF	90 °C	69	_
8	4 b	$Pd(dba)_2^{[a]}$	14	Bu ₄ NOAc, allylTMS	DMF	50 °C	71	
9	4 b	$Pd(PPh_3)_4^{[c]}$	-	Et ₃ N	toluene	reflux	_	55 ^[e]
10	4 c	$Pd(OAc)_2^{[b]}$	30	NaHCO ₃ , Bu ₄ NCl	CH ₃ CN	reflux	22	51
11	4 c	$Pd(OAc)_2^{[b]}$	_	NaHCO ₃ , Bu ₄ NCl	CH ₃ CN	reflux	17	52
12	4 c	$Pd(OAc)_2^{[b]}$	30	Ag_2CO_3	DMF	90 °C	78	_
13	4 c	$Pd(PPh_3)_4^{[c]}$	-	Et ₃ N	toluene	reflux	_	69
14	4d	$Pd(OAc)_2^{[b]}$	30	NaHCO ₃ , Bu ₄ NCl	CH ₃ CN	reflux	45	15
15	4d	$Pd(OAc)_2^{[b]}$	-	NaHCO ₃ , Bu ₄ NCl	CH ₃ CN	reflux	52	10
16	4d	$Pd(OAc)_2^{[b]}$	30	Ag ₂ CO ₃	DMF	90 °C	53	_
17	4d	$Pd(PPh_3)_4^{[c]}$	_	Et ₃ N	toluene	reflux	_	60

[a] 3 mol-% of precatalyst was used. [b] 10 mol-% Pd(OAc)₂. [c] 30 mol-% Pd(PPh₃)₄. [d] Mixture of products. [e] Reduced tetrahydroquinoline was obtained.



Scheme 4. Synthesis of 4-alkylidene-substituted quinolines 8a-g.

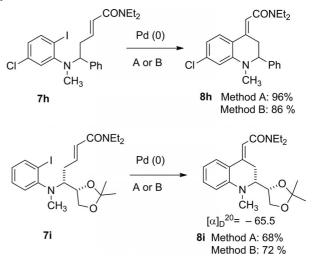
Table 3. Synthesis of 4-alkylidene-substituted quinolines 8a-g.

	Substrate	Method ^[a]	Product	Yield [%]
1	7a	А	8a	79
2	7b	А	8b	72
3	7c	А	8c	67
4	7d	А	8d	67
5	7e	А	8e	57
6	7f	А	8f	73
7	7g	А	8g	61
8	7a	В	8a	85
9	7b	В	8b	77
10	7c	В	8c	76
11	7d	В	8d	67
12	7e	В	8e	61
13	7f	В	8f	81
14	7g	В	8g	67
15	7a	С	8a	74
16	7b	С	8b	60
17	7c	С	8c	61
18	7d	С	8d	60
19	7e	С	8e	40
20	7f	С	8f	68
21	7g	С	8g	82

[a] Method A: $Pd(OAc)_2$ (10 mol-%), PPh_3 (30 mol-%), $AgCO_3$ (1.5 equiv.), dimethylformamide (DMF), 100 °C, 6 h; Method B: $Pd(PPh_3)_4$ (30 mol-%), Et_3N (2.2 equiv.), toluene, reflux, 6 h; Method C: $Pd(OAc)_2$ (15 mol-%), $NaHCO_3$ (2.2 equiv.), Bu_4NCl (1.3 equiv.) CH_3CN , reflux, 6 h.

(Table 3, Entries 8-14, Method B), probably owing to the stabilization of the exo double bond by the amide group that precludes isomerization. These conditions gave higher yields of the quinolines but, once again, required high catalyst loadings to reach completion in reasonable reaction times. Finally, the Mizoroki-Heck reaction could also be performed under phosphane-free conditions (Table 3, Entries 15–21, Method C) obtaining E-quinolines, though in generally lower yields, with the exception of 8g (Table 3, Entry 21). As can be seen in Table 3, the use of diethyl amides or Weinreb amides led to comparable yields of corresponding tetrahydroquinolines 8. Some biologically active quinolines bear chlorine atoms, mainly in the C-5 and/or C-7 position.^[2,4] To access this type of 4-alkylidene-quinolines, butenylaniline 7h was prepared starting from 5-chloro-2-iodoaniline,^[19] and was submitted to Mizoroki-Heck

conditions previously tested (Scheme 5). Tetrahydroquinoline 8h was obtained in excellent yields, as a single E isomer, with no isomerization of the double bond. Finally, this procedure could be successfully applied to the synthesis of enantiomerically pure tetrahydroquinoline 8i. Enantiomerically pure butenylaniline 7i was prepared following the same synthetic strategy, staring from o-iodoaniline and (R)-2,2-dimethyl-1,3-dioxolane-4-carbaldehyde.^[19] Treatment of 7i with both catalytic systems afforded tetrahydroquinoline 8i as a single isomer, with no loss of optical purity. The synthesis of enantiopure tetrahydroquinolines with an *exo* α , β -unsaturated amide moiety at the C-4 position had been reported by Di Fabio, in an approach that used an (R)-(+)tert-butyl lactate as chiral auxiliary.^[4b] The regioselectivity could also be controlled by the catalytic system, and the best results for the formation of the exocyclic double bond were obtained when the Heck reaction was carried out with Pd(PPh₃)₄/Et₃N in toluene. Thus, the results obtained in this work are in agreement with these previous results, and expand the scope of this reaction for the synthesis of heterocycles.



Scheme 5. Synthesis of (E)-alkylidene-quinolines 8.

Conclusions

An intramolecular Heck reaction provides mild and regioselective access to 2-substituted 4-alkylidene-tetrahydroquinoline derivatives, similar to other methods described in the literature.^[25] In all cases a 6-exo-trig process occurs obtaining quinoline derivatives in good yields. When non-substituted alkenes are used, the regioselectivity of the reaction can be directed towards the formation of an exocyclic or endocyclic double bond by choosing a suitable catalytic system. When the double bond is substituted by an amide moiety, intramolecular cyclization is favored, obtaining selectively the exocyclic double bond of *E* geometry. The formation of this isomer does not depend on the precatalyst used.^[26] Optimal conditions have been found for the efficient preparation of a variety of 2-substituted tetrahydroquinolines with an *exo* α , β -unsaturated amide moiety at the C-4 position, showing high functional group tolerance. This procedure can also be applied to chiral substrates, thus achieving a synthesis of enantiopure quinolines with the configuration of known glycine antagonists.

Experimental Section

General Methods: Melting points were determined in unsealed capillary tubes. IR spectra were recorded as films over NaCl pellets, on KBr pellets or with an ATR. NMR spectra were recorded at 20-25 °C, at 300 MHz for ¹H and 75.5 MHz for ¹³C or at 500 MHz for ¹H and 125.7 MHz for ¹³C in CDCl₃ solutions. Assignments of individual ¹³C and ¹H resonances are supported by DEPT experiments and 2D correlation experiments (COSY, HSQCed or HMBC). Selective NOE or NOESY experiments were performed when necessary. Mass spectra were recorded with electron impact (EI) at 70 eV, under chemical ionization (CI) at 230 eV or with electrospray ionization (ESI). HRMS data was recorded with a TOF detector. TLC was carried out with 0.2 mm thick silica gel plates. Visualization was accomplished by UV light. Flash column chromatography^[27] was performed on silica gel (230-400 mesh) or on alumina (70-230 mesh). All solvents used in reactions were anhydrous and purified according to standard procedures.^[28] All airor moisture-sensitive reactions were performed under argon; the glassware was dried (130 °C) and purged with argon.

Synthesis of 2-Aryl-4-methylenequinolines 2. General Procedure: $Pd(OAc)_2$ (3 mol-%) and PPh₃ (20 mol-%) were added to a mixture of butenylamine 1 (1 mmol), Na₂CO₃ (2.5 mmol) and *n*Bu₄NCl (1.5 mmol) in dry CH₃CN. The mixture was heat to reflux until TLC showed complete reaction. The reaction mixture was then diluted with AcOEt (50 mL) and filtered through a Celite pad. The resulting filtrate was washed with saturated NH₄Cl (3 × 30 mL) and brine (3 × 30 mL), dried with Na₂SO₄, filtered and concentrated. The crude oil was purified by flash column chromatography (silica gel, hexane/AcOEt) to give corresponding tetrahydroquinolines 2.

4-Methylene-2-phenyl-1,2,3,4-tetrahydroquinoline (2a): (Table 1, Entry 3) According to the general procedure, amine 1a (111 mg, 0.318 mmol) was treated with Pd(OAc)₂ (2 mg, 0.009 mmol) and PPh₃ (17 mg, 0.064 mmol), Na₂CO₃ (90 mg, 0.80 mmol) and nBu₄NCl (124 mg, 0.47 mmol) in CH₃CN (10 mL). The reaction mixture was heated to reflux for 12 h. After work-up, flash column chromatography (silica gel, 2% hexane/AcOEt) afforded 2a (38 mg, 54%) as an oil. ¹H NMR (300 MHz, CDCl₃): $\delta = 2.74-2.78$ (m, 2 H, H³), 4.16 (s, 1 H, NH), 4.43–4.49 (m, 1 H, H²), 4.81 (s, 1 H, =CH_a), 5.46 (s, 1 H, =CH_b), 6.58 (d, J = 8.0 Hz, 1 H, H⁸), 6.72 (t, J = 8.0 Hz, 1 H, H⁶), 7.10 (t, J = 8.0 Hz, 1 H, H⁷), 7.31–7.53 (m, 5 H, Ph), 7.56 (d, J = 8.0 Hz, 1 H, H⁵) ppm. ¹³C NMR (75.5 MHz, $CDCl_3$): $\delta = 40.8 (C^3)$, 57.2 (C²), 106.7 (C= CH_2), 115.0 (C⁸), 117.8 (C⁶), 119.8 (C^{4a}), 124.7, 126.6, 127.6, 128.7, 129.2 (C⁵, C⁷, C-Harom), 139.6 (C8a), 143.4 (C-Carom), 144.6 (C4) ppm. MS (70 eV, EI): m/z (%) = 221 (2) [M⁺], 219 (100), 218 (37), 204 (57), 109 (6), 77 (5)

2-(1-Methyl-1*H***-pyrrol-2-yl)-4-methylene-1,2,3,4-tetrahydroquinoline (2b):** (Table 1, Entry 5) According to the general procedure, amine **1b** (158 mg, 0.52 mmol) was treated with Pd(OAc)₂ (4 mg, 0.016 mmol) and PPh₃ (26 mg, 0.11 mmol), Na₂CO₃ (137 mg, 1.30 mmol) and *n*Bu₄NCl (173 mg, 0.62 mmol) in CH₃CN (20 mL). The reaction mixture was heated to reflux for 18 h. After work-up, flash column chromatography (silica gel, 10% hexane/Ac-OEt) afforded **2b** (83 mg, 71%) as an oil. IR (ATR): $\tilde{v} = 3375$ cm⁻¹.



¹H NMR (300 MHz, CDCl₃): δ = 2.68–2.96 (m, 2 H, H³), 3.69 (s, 3 H, CH₃), 4.07 (s, 1 H, NH), 4.54 (dd, *J* = 11.2, 3.3 Hz, 1 H, H²), 4.86 (s, 1 H, =CH_a), 5.50 (s, 1 H, =CH_b), 6.08–6.16 (m, 1 H, H^{4'}), 6.16–6.24 (m, 1 H, H^{3'}), 6.52–6.65 (m, 2 H, H⁸, H^{5'}), 6.73 (t, *J* = 7.5 Hz, 1 H, H⁵) ppm. ¹³C NMR (75.5 MHz, CDCl₃): δ = 34.1 (CH₃), 38.8 (C³), 49.8 (C²), 106.4 (C^{3'}), 106.8 (C=CH₂), 107.1 (C^{4'}), 115.3 (C⁸), 118.0 (C⁶), 120.0 (C^{4a}), 122.7 (C^{5'}), 124.7 (C⁵), 129.1 (C⁷), 133.5 (C^{2'}), 139.7 (C^{8a}), 144.3 (C⁴) ppm. MS (70 eV, EI): *m/z* (%) = 224 (100) [M⁺], 223 (87), 222 (39), 221 (85), 209 (32), 208 (11), 207 (15), 194 (11), 180 (10), 143 (99). 142 (25), 115 (39), 81 (45), 77 (10), 65 (6). HRMS (ESI⁺): calcd. for C₁₅H₁₇N₂ [MH⁺] 225.1392; found 225.1392.

4-Methylene-2-(thiophen-3-yl)-1,2,3,4-tetrahydroquinoline (2d): (Table 1, Entry 9) According to the general procedure, amine 1d (385 mg, 1.25 mmol) was treated with $Pd(OAc)_2$ (9 mg, 0.037 mmol) and PPh₃ (66 mg, 0.25 mmol), Na₂CO₃ (332 mg, 3.13 mmol) and *n*Bu₄NCl (417 mg, 1.8 mmol) in CH₃CN (20 mL). The reaction mixture was heated to reflux for 12 h. After workup, flash column chromatography (silica gel, 1% hexane/AcOEt) afforded **2d** (50 mg, 18%) as an oil. IR (ATR): $\tilde{v} = 3385 \text{ cm}^{-1}$. ¹H NMR (300 MHz, CDCl₃): $\delta = 2.76-2.83$ (m, 2 H, H³), 4.22 (s, 1 H, NH), 4.51–4.67 (m, 1 H, H²), 4.84 (s, 1 H, =CH_a), 5.48 (s, 1 H, =CH_b), 6.58 (d, J = 7.5 Hz, 1 H, H⁸), 6.73 (t, J = 7.5 Hz, 1 H, H⁶), 7.09 (t, J = 7.5 Hz, 1 H, H⁷), 7.14 (dd, J = 5.0, 1.0 Hz, 1 H, H⁴), 7.24–7.26 (m, 1 H, $H^{2'}$), 7.33 (dd, $J = 5.0, 3.0 \text{ Hz}, 1 \text{ H}, H^{5'}$), 7.57 (d, J = 7.5 Hz, 1 H, H⁵) ppm. ¹³C NMR (75.5 MHz, CDCl₃): $\delta =$ 40.1 (C³), 52.8 (C²), 106.9 (C=CH₂), 115.1 (C⁸), 118.0 (C⁶), 120.1 (C^{4a}) , 121.1 $(C^{2'})$, 124.7 (C^{5}) , 126.1, 126.5 $(C_{4'}, C^{5'})$, 129.1 (C^{7}) , 139.3 (C^{3'}), 144.2 (C^{8a}), 144.7 (C⁴) ppm. MS (230 eV, CI): *m/z* (%) $= 228 (100) [MH^+], 227 (72), 226 (30), 225 (7), 212 (6), 144 (12).$ HRMS (CI): calcd. for C14H14NS [MH⁺] 228.0847; found 228.0839.

4-Methyl-2-(thiophen-3-yl)quinoline (3d):^[29] (Table 1, Entry 9; 85 mg, 30%). IR (ATR): $\hat{v} = 1600 \text{ cm}^{-1}$. ¹H NMR (300 MHz, CDCl₃): $\delta = 2.74$ (s, 3 H, CH₃), 7.44 (dd, J = 5.1, 3.0 Hz, 1 H, H⁴), 7.52 (t, J = 7.5 Hz, 1 H, H⁷), 7.62 (s, 1 H, H³), 7.70 (t, J = 7.5 Hz, 1 H, H⁶), 7.87 (dd, J = 5.1, 1.1 Hz, 1 H, H⁵), 7.97 (d, J = 7.5 Hz, 1 H, H⁸), 8.03 (dd, J = 3.0, 1.1 Hz, 1 H, H²), 8.12 (d, J = 7.5 Hz, 1 H, H⁵) ppm. ¹³C NMR (75.5 MHz, CDCl₃): $\delta = 18.9$ (CH₃), 119.7 (C³), 123.6 (C⁸), 124.4 (C^{2'}), 125.8 (C⁷), 126.2 (C^{4'}), 126.8 (C^{5'}), 127.2 (C^{4a}), 129.3 (C⁶), 130.0 (C⁵), 142.7 (C^{3'}), 144.7 (C⁴), 148.1 (C^{8a}), 153.0 (C²) ppm. MS (70 eV, EI): *m/z* (%) = 226.1 (19) [M⁺ + 1], 225.1 (100) [M⁺], 224.1 (40), 210 (11), 199 (5), 181.1 (6), 180.1 (6), 168.1 (3), 115 (5), 77 (1), 65 (1).

4-Methylene-2-(1-methyl-1H-pyrrol-3-yl)-1,2,3,4-tetrahydroquinoline (2c): (Table 1, Entry 8) Pd(OAc)₂ (11 mg, 0.049 mmol) and PPh₃ (52 mg, 0.20 mmol) were added to a mixture of amine 1c (151 mg, 0.49 mmol) and nBu₄NOAc (231 mg, 0.74 mmol) in dry DMSO (10 mL), and the mixture was heated at 60 °C for 24 h. The reaction mixture was washed with saturated NH₄Cl (2×30 mL) and brine $(2 \times 30 \text{ mL})$, dried with Na₂SO₄, filtered and concentrated. The crude oil was purified by flash column chromatography (silica gel, 10% hexane/AcOEt) to afford 2c (60 mg, 55%) as an oil. IR (ATR): $\tilde{v} = 3385 \text{ cm}^{-1}$. ¹H NMR (300 MHz, CDCl₃): $\delta =$ 2.69-2.92 (m, 2 H, H³), 3.66 (s, 3 H, CH₃), 4.13 (s, 1 H, NH), 4.43 $(dd, J = 8.9, 5.3 Hz, 1 H, H^2), 4.85 (s, 1 H, =CH_a), 5.49 (s, 1 H, -CH_a)$ =CH_b), 6.20 (s, 1 H, H^{4'}), 6.52–6.67 (m, 3 H, H₈, H_{2'}, H^{5'}), 6.68– 6.78 (m, 1 H, H⁶), 7.04–7.15 (m, 1 H, H⁷), 7.57 (d, J = 7.8 Hz, 1 H, H⁵) ppm. ¹³C NMR (75.5 MHz, CDCl₃): δ = 36.1 (CH₃), 40.6 (C^3) , 50.6 (C^2) , 106.1 $(C^{4'})$, 106.5 $(C=CH_2)$, 115.0 (C^8) , 117.5 (C^6) , 118.9, 121.9 ($C_{2'}$, $C^{5'}$), 120.0 ($C^{3'}$), 124.6 (C^{5}), 126.7 (C^{4a}), 129.0

(C⁷), 140.3 (C^{8a}), 144.8 (C⁴) ppm. MS (70 eV, EI): m/z (%) = 224 (100) [M⁺], 223 (77), 209 (51), 208 (8), 207 (10), 194 (11), 180 (10), 143 (25), 115 (17), 81 (28), 77 (7), 65 (4). HRMS (ESI⁺): calcd. for $C_{15}H_{17}N_2$ [MH⁺] 225.1392; found 225.1392.

4-Methyl-2-phenylquinoline (3a):^[30] (Table 1, Entry 1). Pd(OAc)₂ (8 mg, 0.036 mmol) was added to a mixture of butenylamine 1a (432 mg, 1.2 mmol), Na₂CO₃ (341 mg, 3.2 mmol) and *n*Bu₄NBr (498 mg, 1.5 mmol) in dry DMF (20 mL). The mixture was heated at 60 °C for 18 h. The reaction mixture was then diluted with of AcOEt (20 mL) and filtered through a Celite pad. The resulting filtrate was washed with saturated NH₄Cl (3×20 mL) and brine $(3 \times 20 \text{ mL})$, dried with Na₂SO₄ filtered and concentrated. The crude oil was purified by flash column chromatography (silica gel, 5% hexane/AcOEt) to give quinoline 3a (130 mg, 50%) whose data are coincidental with those reported.^[30] IR (NaCl): $\tilde{\nu} = 1600 \text{ cm}^{-1}$. ¹H NMR (300 MHz, CDCl₃): δ = 2.75 (s, 3 H, CH₃), 7.47–7.58 (m, 4 H, H³, H^{3'}, H^{4'}, H^{5'}), 7.70–7.76 (m, 2 H, H^{2'}, H^{6'}), 7.99 (d, J =8.3 Hz, 1 H, H⁸), 8.15–8.22 (m, 3 H, H⁶, H⁷, H⁵) ppm. ¹³C NMR $(75.5 \text{ MHz}, \text{CDCl}_3)$: $\delta = 18.9 (\text{CH}_3), 119.7 (\text{C}^3), 123.5 (\text{C}^8), 125.9$ (C⁷), 127.1 (C^{4a}), 127.4, 128.7, 129.1, 129.2 (C⁶, C-H_{arom}), 130.1 (C⁵), 139.7 (C_{arom}-C), 144.7 (C⁴), 148.0 (C^{8a}), 156.9 (C²) ppm. MS $(70 \text{ eV}, \text{EI}): m/z \ (\%) = 220 \ (18) \ [\text{M}^+ + 1], 219 \ (100) \ [\text{M}^+], 218 \ (35),$ 204 (63), 109 (13), 102 (14), 95 (11), 77 (8).

Synthesis of 4-Methylenequinolines 5. General Procedure A: Ag₂CO₃ (1.5 mmol), PPh₃ (30 mol-%) and Pd(OAc)₂ (10 mol-%), were added to a solution of amine 4 (1 mmol) in dry DMF (30 mL). The mixture was stirred at 90 °C until TLC showed complete reaction. The reaction mixture was then diluted with Et₂O (20 mL), and filtered through a Celite pad. The resulting filtrate was washed with brine (3×30 mL), dried with Na₂SO₄, filtered and concentrated. The crude oil was purified by flash column chromatography (silica gel, hexane/AcOEt) to give corresponding tetrahydroquinolines 5.

1-Methyl-4-methylene-2-phenyl-1,2,3,4-tetrahydroquinoline (5a): (Table 2, Entry 3) According to general procedure A, amine 4a (135 mg, 0.37 mmol) was treated with Pd(OAc)₂ (8.3 mg, 0.037 mmol) PPh₃ (28.8 mg, 0.11 mmol) and Ag₂CO₃ (153 mg, 0.55 mmol) in DMF (10 mL). The reaction mixture was stirred at 90 °C for 12 h. After work-up, flash column chromatography (silica gel, hexane) afforded 5a (68 mg, 78%) as a white solid. M.p. (hexane) 64–66 °C. IR (KBr): $\tilde{v} = 890 \text{ cm}^{-1}$. ¹H NMR (300 MHz, CDCl₃): $\delta = 2.62-2.73$ (m, 1 H, H³), 2.88 (s, 3 H, NCH₃), 2.97-3.04 (m, 1 H, H³), 4.50 (t, J = 5.2 Hz, 1 H, H²), 4.61 (s, 1 H, =CH_a), 5.32 (s, 1 H, =CH_b), 6.67–6.73 (m, 2 H, H⁶, H⁸), 7.12–7.49 (m, 7 H, H⁵, H⁷, Ph) ppm. ¹³C NMR (75.5 MHz, CDCl₃): δ = 38.0 (NCH₃), 39.4 (C³), 64.1 (C²), 107.7 (C=CH₂), 111.1 (C⁸), 116.0 (C⁶), 121.4 (C^{4a}), 124.6, 126.4, 127.0, 128.3, 129.6 (C⁵, C⁷, C-H_{arom}), 138.2 (C-C_{arom}), 142.9 (C^{8a}), 145.4 (C⁴) ppm. MS (70 eV, EI): m/z (%) = 235 (41) [M⁺], 234 (16), 159 (15), 158 (100), 143 (10), 115 (13), 91 (11), 77 (11). C₁₇H₁₇N (235.33): calcd. C 86.77, H 7.28, N 5.95; found C 86.56, H 7.02, N 5.72.

1-Methyl-4-methylene-1,2,3,4-tetrahydroquinoline (5b): (Table 2, Entry 7) According to general procedure A, amine **4b** (120 mg, 0.43 mmol) was treated with Pd(OAc)₂ (10 mg, 0.043 mmol) and PPh₃ (33.8 mg, 0.129 mmol) and Ag₂CO₃ (178 mg, 0.64 mmol) in DMF (10 mL). The reaction mixture was stirred at 90 °C for 12 h. After work-up, flash column chromatography (silica gel, hexanes) afforded **5b** (47 mg, 69%) as an oil. IR (NaCl): $\tilde{v} = 1595$ cm⁻¹. ¹H NMR (300 MHz, CDCl₃): $\delta = 2.69$ (t, J = 6.0 Hz, 2 H, H³), 2.91 (s, 3 H, NCH₃), 3.26 (t, J = 6.0 Hz, 2 H, H²), 4.77 (s, 1 H, =CH₃), 5.38 (s, 1 H, =CH_b), 6.64–6.69 (m, 2 H, H⁶, H⁸), 7.16 (td, J = 7.7, 1.3 Hz, 1 H, H⁷), 7.51 (dd, J = 7.7, 1.3 Hz, 1 H, H⁵) ppm. ¹³C NMR (75.5 MHz, CDCl₃): $\delta = 32.1$ (C³), 39.6 (NCH₃), 51.8 (C²), 105.7 (C=*C*H₂), 112.1 (C⁸), 116.7 (C⁶), 121.4 (C^{4a}), 124.6 (C⁵), 129.3 (C⁷), 140.6 (C^{8a}), 146.3 (C⁴) ppm. C₁₁H₁₃N (159.23): calcd. C 82.97, H 8.23, N 8.80; found C 82.71, H 7.93, N 8.77.

2-Benzyloxymethyl-1-methyl-4-methylene-1,2,3,4-tetrahydroquinoline (5c): (Table 2, Entry 12) According to general procedure A, amine 4c (180 mg, 0.45 mmol) was treated with $Pd(OAc)_2$ (10 mg, 0.04 mmol), PPh₃ (36 mg, 0.13 mmol) and Ag₂CO₃ (176 mg, 0.67 mmol) in DMF (20 mL). The reaction mixture was stirred at 90 °C for 5 h. After work-up, flash column chromatography (silica gel, 5% hexane/AcOEt) afforded 5c (96 mg, 78%) as an oil. IR (NaCl): $\tilde{v} = 875 \text{ cm}^{-1}$. ¹H NMR (300 MHz, CDCl₃): δ = 2.71 (dd, J = 13.8, 2.3 Hz, 1 H, H³), 2.78–2.82 (m, 1 H, H³), 3.07 (s, 3 H, NCH₃), 3.39-3.46 (m, 1 H, CH₂O), 3.54-3.60 (m, 1 H, CH₂O), 3.63–3.72 (m, 1 H, H²), 4.52–4.54 (m, 2 H, PhCH₂), 4.85 $(s, 1 H, =CH_a), 5.49 (s, 1 H, =CH_b), 6.60 (d, J = 7.5 Hz, 1 H, H^8),$ 6.70 (t, J = 7.5 Hz, 1 H, H⁶), 7.20 (t, J = 7.5 Hz, 1 H, H⁷), 7.33– 7.43 (m, 5 H, Ph), 7.53 (d, J = 7.5 Hz, 1 H, H⁵) ppm. ¹³C NMR $(75.5 \text{ MHz}, \text{ CDCl}_3): \delta = 32.6 (\text{C}^3), 38.7 (\text{NCH}_3), 58.8 (\text{C}^2), 69.7$ (CH_2O) , 73.4 $(PhCH_2)$, 107.8 $(C=CH_2)$, 111.0 (C^8) , 115.6 (C^6) , 120.3 (C^{4a}), 124.5 (C⁵), 127.4, 127.5, 128.3 (C-H_{arom}), 128.4 (C-Carom), 129.5 (C7), 138.2 (C8a), 143.8 (C4) ppm. MS (70 eV, EI): m/z (%) = 279 (3) [M⁺], 158 (100), 143 (11), 115 (10), 91 (12), 83 (18). HRMS (EI): calcd. for C₁₉H₂₁NO 279.1623; found 279.1631. C₁₉H₂₁NO (279.38): calcd. C 81.68, H 7.58, N 5.01; found C 81.45, H 7.32, N 5.22.

Ethyl 1-Methyl-4-methylene-1,2,3,4-tetrahydroquinoline-2-carboxylate (5d): (Table 2, Entry 16) According to general procedure A, amine 4d (156 mg, 0.50 mmol) was treated with Pd(OAc)₂ (11 mg, 0.05 mmol), PPh₃ (40 mg, 0.15 mmol) and Ag₂CO₃ (210 mg, 0.75 mmol) in DMF (15 mL). The reaction mixture was stirred at 90 °C for 5 h. After work-up, flash column chromatography (silica gel, 10% hexane/AcOEt) afforded 5d (59 mg, 53%) as an oil. IR (NaCl): $\tilde{v} = 1735$, 875 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): $\delta = 1.18$ $(t, J = 7.1 \text{ Hz}, 3 \text{ H}, \text{CH}_2\text{C}H_3), 2.91 \text{ (d}, J = 4.0 \text{ Hz}, 2 \text{ H}, \text{H}^3), 2.99$ (s, 3 H, NCH₃), 4.02–4.18 (m, 3 H, H², CH₂CH₃), 4.78 (d, J =1.1 Hz, 1 H, =CH_a), 5.38 (s, 1 H, =CH_b), 6.63–6.70 (m, 2 H, H⁶, H⁸), 7.18 (t, J = 7.6 Hz, 1 H, H⁷), 7.44 (dd, J = 7.7, 1.5 Hz, 1 H, H⁵) ppm. ¹³C NMR (75.5 MHz, CDCl₃): δ = 14.2 (CH₂CH₃), 34.1 (C³), 39.1 (NCH₃), 60.8 (*C*H₂CH₃), 63.1 (C²), 107.9 (C=*C*H₂), 111.5 (C⁸), 116.8 (C⁶), 120.8 (C⁴a), 124.4 (C⁵), 129.5 (C⁷), 137.4 (C^{8a}) , 144.4 (C⁴), 172.0 (CO) ppm. MS (70 eV, EI): m/z (%) = 231(<1) [M⁺] 158 (56), 97 (15), 85 (40), 71 (39), 57 (100). HRMS (EI): calcd. for C₁₄H₁₇NO₂ 231.1259; found 231.1258. C₁₄H₁₇NO₂ (231.29): calcd. C 72.70, H 7.41, N 6.06; found C 72.32, H 7.18, N 5.80.

Synthesis of 1,2-Dihydroquinolines 6. General Procedure B: $Pd(PPh_3)_4$ (30 mol-%) and Et_3N (2 mmol) were added to a solution of amine 4 (1 mmol) in dry toluene (30 mL). The mixture was heated to reflux for 5–12 h. The reaction mixture was then diluted with AcOEt (20 mL), and filtered through a Celite pad. The resulting filtrate was washed with brine (3 × 30 mL), dried with Na₂SO₄, filtered and concentrated. The crude oil was purified by flash column chromatography (silica gel, hexane/AcOEt) to give corresponding 1,2-dihydroquinolines 6.

1,4-Dimethyl-2-phenyl-1,2-dihydroquinoline (6a): (Table 2, Entry 5) According to general procedure B, amine **4a** (201 mg, 0.55 mmol) was treated with $Pd(PPh_3)_4$ (192 mg, 0.17 mmol) and Et_3N (0.15 mL, 1.10 mmol) in dry toluene (15 mL), and the mixture was heated to reflux for 12 h. After work-up, flash column chromatog-



raphy (silica gel, hexanes) afforded **6a** (98 mg, 76%) as an oil. IR (NaCl): $\tilde{v} = 815 \text{ cm}^{-1}$. ¹H NMR (300 MHz, CDCl₃): $\delta = 2.01$ (s, 3 H, CH₃), 2.70 (s, 3 H, NCH₃), 5.07 (d, J = 4.7 Hz, 1 H, H²), 5.53 (d, J = 4.7 Hz, 1 H, H³), 6.44 (d, J = 8.3 Hz, 1 H, H⁸), 6.65 (t, J = 7.3 Hz, 1 H H⁶), 7.09–7.28 (m, 7 H, H⁵, H⁷, Ph) ppm. ¹³C NMR (75.5 MHz, CDCl₃): $\delta = 18.7$ (CH₃), 36.2 (NCH₃), 65.5 (C²), 109.4 (C⁸), 116.0 (C³), 122.2 (C⁴), 123.1, 123.6 (C⁵, C⁶), 126.3, 127.5, 128.6, 129.6 (C⁷, C–H_{arom}), 142.8, 144.8 (C^{4a}, C^{8a}, C–C_{arom}) ppm. MS (70 eV, EI): *m/z* (%) = 235 (13) [M⁺], 221 (13), 220 (29), 158 (100), 115 (11), 77 (13).

2-Benzyloxymethyl-1,4-dimethyl-1,2-dihydroquinoline (6c): (Table 2, Entry 13) According to general procedure B, amine 4c (230 mg, 0.56 mmol) was treated with Pd(PPh₃)₄ (192 mg, 0.17 mmol) and Et₃N (0.16 mL, 1.13 mmol) in dry toluene (15 mL), and the mixture was heated to reflux for 5 h. After work-up, flash column chromatography (silica gel, hexanes) afforded 6c (107 mg, 69%) as an oil. IR (NaCl): $\tilde{v} = 1655$, 740 cm⁻¹. ¹H NMR (300 MHz, $CDCl_3$): $\delta = 2.05$ (s, 3 H, CH_3), 3.03 (s, 3 H, NCH_3), 3.37 (dd, J = 9.7, 4.3 Hz, 1 H, CHHO), 3.56 (dd, J = 9.7, 7.0 Hz, 1 H, CHHO), 4.12-4.22 (m, 1 H, H²), 4.45 (s, 2 H, PhCH₂), 5.54 (d, J = 5.8 Hz, 1 H, H³), 6.54 (d, J = 8.0 Hz, 1 H, H⁸), 6.68 (t, J = 7.4 Hz, 1 H, H⁶), 7.10–7.19 (m, 2 H, H⁵, H⁷), 7.27–7.37 (m, 5 H, Ph) ppm. ¹³C NMR (75.5 MHz, CDCl₃): δ = 18.8 (CH₃), 38.2 (NCH₃), 60.1 (C²), 72.7 (CH₂O), 73.1 (Ph*C*H₂), 110.7 (C⁸), 116.0 (C⁶), 119.5 (C³), 122.9 (C⁴), 123.6 (C⁵), 127.3, 127.4, 128.3, 128.8 (C⁷, C-H_{arom}), 131.8 (C^{4a}), 138.3 (C-C_{arom}), 144.4 (C^{8a}) ppm. MS (70 eV, EI): *m*/*z* (%) = 279 (1) [M⁺], 173 (17), 82 (100), 77 (29). $C_{19}H_{21}NO$ (279.38): calcd. C 81.68, H 7.58, N 5.01; found C 81.39, H 7.45, N 5.34.

Ethyl 1,4-Dimethyl-1,2-dihydroquinoline-2-carboxylate (6d): (Table 2, Entry 17) According to general procedure B, amine 4d (131 mg, 0.42 mmol) was treated with $Pd(PPh_3)_4$ (150 mg, 0.13 mmol) and Et₃N (0.12 mL, 0.84 mmol) in dry toluene (15 mL), and the mixture was heated to reflux for 5 h. After workup, flash column chromatography (silica gel, 5% hexane/AcOEt) afforded **6d** (58 mg, 60%) as an oil. IR (NaCl): $\tilde{v} = 1740 \text{ cm}^{-1}$. ¹H NMR (300 MHz, CDCl₃): δ = 1.23 (t, *J* = 7.1 Hz, 3 H, CH₂CH₃), 2.03 (s, 3 H, CH₃), 2.92 (s, 3 H, NCH₃), 4.12 (q, J = 7.1 Hz, 2 H, CH_2CH_3), 4.61 (d, J = 5.9 Hz, 1 H, H²), 5.60 (d, J = 5.9 Hz, 1 H, H^{3}), 6.57 (d, J = 8.0 Hz, 1 H, H^{8}), 6.68 (t, J = 7.5 Hz, 1 H, H^{6}), 7.10 (dd, J = 7.5, 1.4 Hz, 1 H, H⁷), 7.16 (td, J = 7.5, 1.4 Hz, 1 H, H⁵) ppm. ¹³C NMR (75.5 MHz, CDCl₃): δ = 14.1 (CH₂CH₃), 18.7 (CH₃), 37.3 (NCH₃), 60.7 (CH₂CH₃), 63.4 (C²), 110.2 (C⁸), 116.3, 116.8 (C³, C⁶), 122.1 (C⁴), 123.8 (C⁵), 129.3 (C⁷), 132.8 (C^{4a}), 144.5 (C^{8a}), 170.9 (CO) ppm. MS (70 eV, EI): m/z (%) = 231 (3) [M⁺], 158 (67), 97 (11), 85 (42), 83 (100), 71 (23), 57 (62). HRMS (EI): calcd. for C₁₄H₁₇NO₂ 231.1259; found 231.1253.

(E)-N,N-Diethyl-2-[1-methyl-2-phenyl-2,3-dihydroquinolin-4(1H)ylidene]acetamide (8a): (Table 3, Entry 8) According to general procedure B, amine 7a (148 mg, 0.32 mmol) was treated with Pd-(PPh₃)₄ (112 mg, 0.09 mmol) and Et₃N (0.09 mL, 0.64 mmol) in dry toluene (15 mL), and the mixture was heated to reflux for 6 h. After work-up, flash column chromatography (silica gel, 40% hexane/AcOEt) afforded 8a (86 mg, 85%) as an oil. IR (NaCl): \tilde{v} = 1600 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): $\delta = 0.83$ (t, J = 7.1 Hz, 3 H, CH_2CH_3), 1.01 (t, J = 7.1 Hz, 3 H, CH_2CH_3), 2.69–2.85 (m, 2 H, CH₂CH₃), 2.93 (s, 3 H, NCH₃), 3.03-3.14 (m, 2 H, H³, $CHHCH_3$), 3.31 (dd, J = 14.0, 4.2 Hz, 1 H, H³), 3.41–3.53 (m, 1 H, CHHCH₃), 4.58 (t, J = 4.2 Hz, 1 H, H²), 6.22 (s, 1 H, =CH), 6.64-6.70 (m, 2 H, H⁶, H⁸), 7.00-7.07 (m, 2 H, Ph₂, H⁷), 7.16-7.29 (m, 4 H, Ph), 7.35 (dd, J = 7.6, 1.3 Hz, 1 H, H⁵) ppm. ¹³C NMR $(75.5 \text{ MHz}, \text{CDCl}_3): \delta = 12.8, 13.8 (\text{CH}_2\text{CH}_3), 33.8 (\text{C}^3), 37.9$ (NCH₃), 39.2, 41.6 (CH₂CH₃), 63.2 (C²), 110.7 (C⁸), 115.2

(C=*C*H), 115.7 (C⁶), 121.0 (C^{4a}), 125.0 (C⁵), 126.1, 126.9, 128.3, 130.6 (C⁷, C–H_{arom}), 141.1 (C–C_{arom}), 142.1 (C^{8a}), 145.8 (C⁴), 167.0 (CO) ppm. MS (70 eV, EI): m/z (%) = 334 (56) [M⁺], 260 (60), 234 (45), 220 (47), 219 (35), 184 (100), 158 (20), 130 (13), 69 (49). HRMS (EI): calcd. for C₂₂H₂₆N₂O 334.2045; found 334.2041. C₂₂H₂₆N₂O (334.46): calcd. C 79.00, H 7.84, N 8.38; found C 78.81, H 7.83, N 8.13.

(E)-N-Methoxy-N-methyl-2-[1-methyl-2-phenyl-2,3-dihydroquinolin-4(1H)-ylideneJacetamide (8b): (Table 3, Entry 9) According to general procedure B, amine 7b (134 mg, 0.30 mmol) was treated with Pd(PPh₃)₄ (112 mg, 0.09 mmol) and Et₃N (0.08 mL, 0.60 mmol) in dry toluene (10 mL), and the mixture was heated to reflux for 6 h. After work-up, flash column chromatography (silica gel, 30% hexane/AcOEt) afforded 8b (72 mg, 77%) as an oil. IR (NaCl): \tilde{v} = 1640 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ = 2.93 (s, 3 H, NCH₃), 3.10 (s, 3 H, NCH₃), 3.24–3.32 (m, 1 H, H³), 3.30 (s, 3 H, OCH₃), 3.75 (dd, J = 14.7, 4.5 Hz, 1 H, H³), 4.56 (t, J = 4.5 Hz, 1 H, H²), 6.58 (s, 1 H, =CH), 6.65-6.72 (m, 2 H, H⁶, H⁸), 7.09-7.32 (m, 6 H, H⁷, Ph), 7.43 (dd, J = 7.8, 1.4 Hz, 1 H, H⁵) ppm. ¹³C NMR $(75.5 \text{ MHz}, \text{CDCl}_3): \delta = 32.2 (\text{NCH}_3), 33.1 (\text{C}^3), 38.0 (\text{NCH}_3), 61.2$ (OCH₃), 63.3 (C²), 110.3 (C=CH), 111.2 (C⁸), 115.9 (C⁶), 121.0 (C^{4a}), 125.3 (C⁵), 126.4, 126.9, 128.3 (C-H_{arom}), 131.3 (C⁷), 142.0 (C-C_{arom}), 146.3, 146.5 (C⁴, C^{8a}), 167.5 (CO) ppm. MS (70 eV, EI): m/z (%) = 322 (6) [M⁺], 292 (26), 281 (15), 262 (68), 234 (28), 215 (11), 184 (81), 126 (10), 84 (62), 72 (93), 69 (31), 59 (100), 55 (60). HRMS (EI): calcd. for C₂₀H₂₂N₂O₂ 322.1681; found 322.1693. C₂₀H₂₂N₂O₂ (322.41): calcd. C 74.51, H 6.88, N 8.69; found C 74.26, H 6.77, N 8.38.

(E)-N,N-Diethyl-2-[1-methyl-2,3-dihydroquinolin-4(1H)-ylidene]acetamide (8c): (Table 3, Entry 3) According to general procedure A, amine 7c (90 mg, 0.23 mmol) was treated with Pd(OAc)₂ (8 mg, 0.023 mmol) PPh₃ (20 mg, 0.07 mmol) and Ag₂CO₃ (100 mg, 0.35 mmol) in DMF (10 mL). The reaction mixture was stirred at 90 °C for 5 h. After work-up, flash column chromatography (silica gel, 20% hexane/AcOEt) afforded 8c (40 mg, 67%) as an oil. IR (NaCl): $\tilde{v} = 1625 \text{ cm}^{-1}$. ¹H NMR (300 MHz, CDCl₃): $\delta = 1.16$ – 1.21 (m, 6 H, 2×CH₂CH₃), 2.91 (s, 3 H, NCH₃), 3.05–3.12 (m, 2 H, H³), 3.21–3.27 (m, 2 H, H²), 3.38–3.50 (m, 4 H, 2×CH₂CH₃), 6.39 (s, 1 H, =CH), 6.64–6.68 (m, 2 H, H⁶, H⁸), 7.20 (td, J = 7.4, 1.4 Hz, 1 H, H⁷), 7.45 (dd, J = 7.4, 1.4 Hz, 1 H, H⁵) ppm. ¹³C NMR (75.5 MHz, CDCl₃): δ = 13.3, 14.5 (2×NCH₂CH₃), 27.2 (C^3) , 39.4 (NCH₃), 39.9, 42.6 (2×NCH₂CH₃), 51.0 (C²), 111.7 (C=CH), 112.3 (C^8) , 116.7 (C^6) , 121.0 (C^{4a}) , 124.7 (C^5) , 130.4 (C^7) , 144.2 (C^{8a}), 147.5 (C⁴), 167.4 (CO) ppm. MS (70 eV, EI): m/z (%) = 258 (52) [M⁺], 184 (56), 158 (100), 143 (21). HRMS (EI): calcd. for C₁₆H₂₂N₂O 258.1732; found 258.1738. C₁₆H₂₂N₂O (258.36): calcd. C 74.38, H 8.58, N 10.84; found C 74.86, H 8.77, N 10.96.

(*E*)-*N*-Methoxy-*N*-methyl-2-[1-methyl-2,3-dihydroquinolin-4(1*H*)ylidene]acetamide (8d): (Table 3, Entry 11) According to general procedure B, amine 7d (95 mg, 0.25 mmol) was treated with Pd(PPh₃)₄ (89 mg, 0.08 mmol) and Et₃N (0.08 mL, 0.50 mmol) in dry toluene (10 mL), and the mixture was heated to reflux for 6 h. After work-up, flash column chromatography (silica gel, 40% hexane/AcOEt) afforded 8d (40 mg, 67%) as an oil. IR (NaCl): $\tilde{v} =$ 1650 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): $\delta = 2.91$ (s, 3 H, NCH₃), 3.21–3.23 (m, 2 H, H³), 3.23 (s, 3 H, NCH₃), 3.37–3.39 (m, 2 H, H²), 3.72 (s, 3 H, OCH₃), 6.64–6.72 (m, 3 H, C=CH, H⁶, H⁸), 7.22 (t, *J* = 7.1 Hz, 1 H, H⁷), 7.55 (d, *J* = 7.9 Hz, 1 H, H⁵) ppm. ¹³C NMR (75.5 MHz, CDCl₃): $\delta = 26.7$ (C³), 32.3 (NCH₃), 39.3 (NCH₃), 50.8 (C²), 61.5 (OCH₃), 106.9 (C=CH), 112.4 (C⁸), 116.6 (C⁶), 121.0 (C^{4a}), 124.9 (C⁵), 131.0 (C⁷), 148.1 (C^{8a}), 149.4 (C⁴), 168.1 (CO) ppm. MS (230 eV, CI): *m/z* (%) = 247 (34) [MH⁺], 246

(21) [M⁺], 217 (82), 216 (81), 186 (100), 158 (30), 144 (17). HRMS (CI): calcd. for $C_{14}H_{19}N_2O_2$ 247.1447; found 247.1436.

(E)-2-[2-Benzyloxymethyl-1-methyl-2,3-dihydroquinolin-4(1H)ylidene]-N,N-diethylacetamide (8e): (Table 3, Entry 12) According to general procedure B, amine 7e (132 mg, 0.26 mmol) was treated with Pd(PPh₃)₄ (91 mg, 0.08 mmol) and Et₃N (0.08 mL, 0.52 mmol) in dry toluene (10 mL), and the mixture was heated to reflux for 6 h. After work-up, flash column chromatography (silica gel, 40% hexane/AcOEt) afforded 8e (60 mg, 61%) as an oil. IR (NaCl): $\tilde{v} = 1630 \text{ cm}^{-1}$. ¹H NMR (300 MHz, CDCl₃): $\delta = 1.16$ – 1.21 (m, 6 H, $2 \times CH_2CH_3$), 2.71 (dd, J = 14.6, 5.5 Hz, 1 H, H³), 3.07 (s, 3 H, NCH₃), 3.35–3.50 (m, 5 H, CHHO, $2 \times CH_2CH_3$), 3.55-3.60 (m, 2 H, H³, CHHO), 3.67-3.72 (m, 1 H, H²), 4.48 (d, J = 11.9 Hz, 2 H, CH_2Ph), 6.48 (s, 1 H, =CH), 6.59 (d, J = 8.0 Hz, 1 H, H⁸), 6.60–6.70 (m, 1 H, H⁶), 7.21–7.36 (m, 6 H, H⁷, Ph), 7.44 (dd, *J* = 8.0, 1.3 Hz, 1 H, H⁵) ppm. ¹³C NMR (75.5 MHz, CDCl₃): $\delta = 13.2, 14.4 \ (2 \times CH_2CH_3), 27.8 \ (C^3), 38.9 \ (NCH_3), 39.8, 42.6$ $(2 \times CH_2CH_3)$, 58.7 (C²), 71.0 (CH₂O), 73.4 (CH₂Ph), 111.5 (C⁸), 113.7 (C=CH), 115.4 (C⁶), 119.7 (C^{4a}), 124.6 (C⁵), 127.4, 127.5, 128.4, 130.7 (C⁷, C–H_{arom}), 138.4 (C–C_{arom}), 142.1 (C^{8a}), 144.9 (C⁴), 167.2 (CO) ppm. MS (70 eV, EI): m/z (%) = 257 (15) [M⁺ – CH₂OBn], 184 (100), 91 (8). HRMS (EI): calcd. for $C_{24}H_{30}N_2O_2$ 378.2307; found 378.2328. C₂₄H₃₀N₂O₂ (378.51): calcd. C 76.16, H 7.99, N 7.40; found C 75.98, H 8.01, N 6.99.

(E)-2-[2-Benzyloxymethyl-1-methyl-2,3-dihydroquinolin-4(1H)ylidene]-N-methoxy-N-methylacetamide (8f): (Table 3, Entry 13) According to general procedure B, amine 7f (176 mg, 0.35 mmol) was treated with Pd(PPh₃)₄ (120 mg, 0.11 mmol) and Et₃N (0.10 mL, 0.71 mmol) in dry toluene (10 mL), and the mixture was heated to reflux for 6 h. After work-up, flash column chromatography (silica gel, 40% hexane/AcOEt) afforded 8f (101 mg, 81%) as an oil. IR (NaCl): $\tilde{v} = 1640 \text{ cm}^{-1}$. ¹H NMR (300 MHz, CDCl₃): $\delta = 2.70-2.77$ (m, 1 H, H³), 3.06 (s, 3 H, NCH₃), 3.24 (s, 3 H, NCH₃), 3.41–3.46 (m, 1 H, CHHO), 3.54–3.59 (m, 1 H, CHHO), 3.69 (broad s, 4 H, OCH₃, H²), 4.11 (dd, J = 15.1, 2.3 Hz, 1 H, H³), 4.44 (s, 2 H, CH₂Ph), 6.57–6.66 (m, 2 H, H⁶, H⁸), 6.76 (s, 1 H, =C*H*), 7.22–7.34 (m, 6 H, H⁷, H_{arom}), 7.50 (dd, J = 7.8, 1.4 Hz, 1 H, H⁵) ppm. ¹³C NMR (75.5 MHz, CDCl₃): δ = 27.2 (C³), 32.4 (NCH₃), 38.9 (NCH₃), 58.6 (C²), 61.5 (OCH₃), 70.8 (CH₂O), 73.3 (CH₂Ph), 109.1 (C=CH), 111.7 (C⁸), 115.5 (C⁶), 119.7 (C^{4a}), 124.9 (C⁵), 127.2, 127.4, 128.3, 131.4 (C⁷, C-H_{arom}), 138.3 (C-H_{arom}), 145.4 (C^{8a}), 147.2 (C⁴), 168.1 (CO) ppm. MS (70 eV, EI): m/z (%) $= 366 (9) [M^+], 306 (39), 245 (45), 215 (82), 184 (100), 157 (19),$ 144 (12), 136 (11), 122 (12), 91 (52), 72 (24), 59 (32), 55 (28). HRMS (EI): calcd. for C₂₂H₂₆N₂O₃ 366.1943; found 366.1936. C₂₂H₂₆N₂O₃ (366.46): calcd. C 72.11, H 7.15, N 7.64; found C 72.03, H 7.16, N 7.53.

Ethyl (*E*)-4-[2-(Diethylamino)-2-oxoethylidene]-1-methyl-1,2,3,4tetrahydroquinoline-2-carboxylate (8g): (Table 3, Entry 21) Pd-(OAc)₂ (16 mg, 0.07 mmol), NaHCO₃ (88 mg, 1.05 mmol) and *n*Bu₄NCl (176 mg, 0.632 mmol) were added to a solution of butenylamine 7g (197 mg, 0.48 mmol) and in dry CH₃CN (10 mL), and the mixture was heated to reflux for 6 h. The reaction mixture was then diluted with Et₂O (10 mL), and filtered through a Celite pad. The resulting filtrate was washed with saturated NH₄Cl (2 × 10 mL) and brine (2 × 10 mL), dried with Na₂SO₄, filtered and concentrated. The crude oil was purified by flash column chromatography (silica gel, 50% hexane/AcOEt) to afford 8g (130 mg, 82%) as an oil. IR (NaCl): $\tilde{v} = 1735$, 1680 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): $\delta = 1.17$ (t, J = 7.1, Hz, 9 H, CH₂CH₃), 2.92 (dd, J = 15.1, 3.0 Hz, 1 H, H³), 2.98 (s, 3 H, NCH₃), 3.26–3.54 (m, 4 H, CH₂CH₃), 3.85 (dd, J = 15.1, 3.0 Hz, 1 H, H³), 4.03–4.11 (m, 3 H, H², CH₂CH₃), 6.41 (s, 1 H, =CH), 6.64–6.70 (m, 2 H, H⁶, H⁸), 7.22 (td, J = 7.7, 1.5 Hz, 1 H, H⁷), 7.38 (dd, J = 7.8, 1.5 Hz, 1 H, H⁵) ppm. ¹³C NMR (75.5 MHz, CDCl₃): $\delta = 13.2$, 14.1, 14.3 (3 × CH₃), 28.8 (C³), 38.8 (NCH₃), 39.8, 42.5 (2 × CH₂CH₃), 60.8 (OCH₂CH₃), 62.3 (C²), 111.8 (C⁶), 113.6 (C=CH), 116.8 (C⁸), 120.3 (C^{4a}), 124.5 (C⁵), 130.7 (C⁷), 141.8 (C^{8a}), 145.4 (C⁴), 166.6 (CO), 171.6 (CO) ppm. MS (70 eV, EI): m/z (%) = 330 (7) [M⁺], 257 (42), 184 (92), 84 (100), 72 (12), 59 (15). HRMS (EI): calcd. for C₁₉H₂₆N₂O₃ 330.1943; found 330.1940.

(E)-2-[7-Chloro-1-methyl-2-phenyl-2,3-dihydroquinolin-4(1H)ylidene]-N,N-diethylacetamide (8h): According to general procedure A, amine 7h (120 mg, 0.24 mmol) was treated with Pd(OAc)₂ (8 mg, 0.024 mmol) PPh₃ (20 mg, 0.07 mmol) and Ag₂CO₃ (100 mg, 0.36 mmol) in DMF (10 mL). The reaction mixture was stirred at 90 °C for 5 h. After work-up, flash column chromatography (silica gel, hexane) afforded 8h (85 mg, 96%) as an oil. ¹H NMR (300 MHz, CDCl₃): $\delta = 0.81$ (t, J = 7.1 Hz, 3 H, CH₂CH₃), 0.99 $(t, J = 7.1 \text{ Hz}, 3 \text{ H}, \text{CH}_2\text{CH}_3), 2.65-2.82 \text{ (m}, 2 \text{ H}, \text{CH}_2\text{CH}_3), 2.92$ (s, 3 H, NCH₃), 2.99–3.11 (m, 2 H, H³, CHHCH₃), 3.32 (dd, J = 14.1, 3.7 Hz, 1 H, H³), 3.38–3.54 (m, 1 H, CHHCH₃), 4.53–4.63 (m, 1 H, H²), 6.19 (d, J = 1.3 Hz, 1 H, C=CH), 6.56–6.69 (m, 2 H, H⁶, H⁸), 6.95–7.05 (m, 2 H, Ph, H⁵), 7.12–7.29 (m, 4 H, Ph) ppm. ¹³C NMR (75.5 MHz, CDCl₃): δ = 12.8, 13.7 (2× CH₂CH₃), 33.6 (C^3) , 38.0 (NCH₃), 39.2, 41.6 (2 × CH₂CH₃), 63.0 (C²), 110.3 (C⁸), 115.6 (C=CH), 115.7 (C⁶), 119.4 (C^{4a}), 127.0 (C⁵), 125.9, 128.5 (C- H_{arom}), 136.21 (C⁷), 139.9 (C– C_{arom}), 141.5 (C^{8a}), 146.5 (C⁴), 166.6 (CO) ppm. MS (230 eV, CI): m/z (%) = 371 (31) [MH⁺ + 2], 369 (100) [MH⁺], 333 (37), 296 (17). HRMS (CI): calcd. for C₂₂H₂₆ClN₂O [MH⁺] 369.1734; found 369.1736.

(-)-(E)-2-{(R)-2-[(S)-2,2-Dimethyl-1,3-dioxolan-4-yl]-1-methyl-2,3-dihydroquinolin-4(1H)-ylidene}-N,N-diethylacetamide (8i): According to general procedure A, amine 7i (160 mg, 0.32 mmol) was treated with Pd(OAc)₂ (7 mg, 0.032 mmol) PPh₃ (23 mg, 0.09 mmol) and Ag₂CO₃ (132 mg, 0.48 mmol) in DMF (10 mL). The reaction mixture was stirred at 90 °C for 5 h. After work-up, flash column chromatography (silica gel, 20% hexane/AcOEt) afforded 8i (80 mg, 68%) as an oil. $[a]_{D}^{20} = -65.5$ (c = 5, CH₂Cl₂). IR (NaCl): $\tilde{v} = 1630 \text{ cm}^{-1}$. ¹H NMR (300 MHz, CDCl₃): $\delta = 1.00$ (t, J =7.1 Hz, 3 H, CH_2CH_3), 1.17 (t, J = 7.1 Hz, 3 H, CH_2CH_3), 1.30 (s, $3 H, CH_3$, 1.41 (s, $3 H, CH_3$), 2.62 (dd, J = 13.3, 2.2 Hz, 1 H, H³), 2.73 (ddd, J = 13.3, 5.1, 2.2 Hz, 1 H, H³), 3.03 (s, 3 H, NCH₃), 3.20-3.40 (m, 2 H, CH₂CH₃), 3.43-3.48 (m, 1 H, H²), 3.54-3.66 (m, 2 H, CH₂CH₃), 3.77 (t, J = 7.5 Hz, 1 H, OCHH), 3.92 (dd, J = 13.1, 7.5 *Hz*, 1 H, OC*H*), 4.02 (dd, *J* = 7.5, 5.6 *Hz*, 1 H, OCH*H*), 5.69 (s, 1 H, =CH), 6.50–6.58 (m, 2 H, H⁸, H⁶), 7.12–7.18 (m, 1 H, H⁷), 7.43 (dd, *J* = 7.8, 1.4 Hz, 1 H, H⁵) ppm. ¹³C NMR (75.5 MHz, CDCl₃): δ = 12.4, 14.1 (CH₂CH₃), 25.7, 26.8 (CH₃), 33.0 (C³), 39.0 (*C*H₂CH₃), 40.0 (NCH₃), 42.6 (*C*H₂CH₃), 62.0 (C²), 67.6 (OCH), 76.7 (OCH₂), 108.5 (OOC), 111.2 (C⁸), 115.9 (C⁶), 118.6 (C^{4a}), 118.9 (C=CH), 127.8 (C⁵), 130.3 (C⁷), 134.0 (C^{8a}), 144.3 (C⁴), 168.9 (CO) ppm. C₂₁H₃₀N₂O₃ (358.48): calcd. C 70.36, H 8.44, N 7.81; found C 70.27, H 8.37, N 7.61.

Supporting Information (see footnote on the first page of this article): Synthesis and characterization of precursors 1a–d, 4a–d, and 7a–i. Copies of ¹H and ¹³C NMR spectra of compounds described

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- For selected recent reviews, see: a) K. Kaur, M. Jain, R. P. Reddy, R. Jain, *Eur. J. Med. Chem.* 2010, 45, 3245–3264; b)
 V. R. Solomon, H. Lee, *Curr. Med. Chem.* 2011, 18, 1488–1508; c) A. Garrido Montalbán, in: *Heterocycles in Natural Products Synthesis* (Eds.: K. C. Majumdar, S. K. Chattopadhyay), Wiley-VCH, Weinheim, Germany, 2011, pp. 299–339; d) V. Sridharan, P. A. Suryavanshi, J. C. Menéndez, *Chem. Rev.* 2011, 111, 7157–7259.
- [2] For recent examples, see: a) V. V. Kouznetsov, C. M. Meléndez Gómez, M. G. Derita, L. Svetaz, E. del Olmo, S. A. Zacchino, *Bioorg. Med. Chem.* 2012, 20, 6506–6512; b) V. V. Kouznetsov, F. A. Ruiz, L. Y. Vargas, M. P. Gupta, *Lett. Drug Des. Discovery* 2012, 9, 680–686, and references cited therein.
- [3] For example, see: a) W. A. Pryor, T. Strickland, D. F. Church, J. Am. Chem. Soc. 1988, 110, 2224–2229; b) A. J. de Koning, Int. J. Food Prop. 2002, 5, 451–461; c) A. Blaszczyck, J. Skolimowski, Chem.-Biol. Interact. 2006, 162, 70–80.
- [4] a) R. Di Fabio, G. Alvaro, B. Bertani, S. Giacobbe, Can. J. Chem. 2000, 78, 808–815; b) R. Di Fabio, G. Alvaro, B. Bertani, D. Donati, S. Giacobbe, C. Marchioro, C. Palma, S. M. Lynn, J. Org. Chem. 2002, 62, 7319–7328; c) R. Di Fabio, E. Tranquillini, B. Bertani, G. Alvaro, F. Micheli, F. Sabbatini, M. D. Pizzi, G. Pentassuglia, A. Pasquarello, T. Messeri, D. Donati, E. Rattti, R. Arban, G. Dal Forno, A. Reggiani, R. J. Barnaby, Bioorg. Med. Chem. Lett. 2003, 13, 3863–3866; d) R. Di Fabio, G. Alvaro, B. Bertani, D. Donati, D. M. Pizzi, G. Gentile, G. Pentassuglia, S. Giacobbe, S. Spada, E. Ratti, M. Corsi, M. Quartaroli, R. J. Barnaby, G. Vitulli, Bioorg. Med. Chem. Lett. 2007, 17, 1176–1180.
- [5] For example, see: a) M. F. Fernandes da Silva, M. S. Soares,
 J. B. Fernandes, P. C. Vieria, *Alkaloids* 2007, 64, 139–214; b)
 J. P. Michael, *Nat. Prod. Rep.* 2008, 25, 166–187.
- [6] L. M. Nogle, W. H. Gerwick, J. Nat. Prod. 2003, 66, 217-220.
- [7] For recent reviews, see: a) V. Kouznetsov, L. Y. Vargas, C. M. Meléndez Gómez, *Curr. Org. Chem.* 2005, *9*, 141–161; b) J. Barluenga, F. Rodriguez, F. J. Fañanás, *Chem. Asian J.* 2009, *4*, 1036–104; c) P. E. Alford, *Prog. Heterocycl. Chem.* 2011, *23*, 329–369; d) R. Alajarín, C. Burgos, in: *Modern Heterocyclic Chemistry* (Eds.: J. Alvarez-Builla, J. J. Vaquero, J. Barluenga), Wiley-VCH, Weinheim, Germany, 2011, vol. 3, pp. 1527–1629.
- [8] For a review on palladium-mediated quinoline Synthesis see: N. M. Ahmad, in: *Palladium in Heterocyclic Chemistry* (Eds.: J. J. Li, G. W. Gribble), Elsevier, Amsterdam, 2007, 2nd edn., pp. 511–539.
- [9] U. Martínez-Estíbalez, N. Sotomayor, E. Lete, *Tetrahedron Lett.* 2007, 48, 2919–2922.
- [10] U. Martínez-Estíbalez, N. Sotomayor, E. Lete, Org. Lett. 2009, 11, 1237–1240.
- [11] For selected reviews, see: a) A. de Meijere, S. Bräse, in: Metal Catalyzed Cross-Coupling Reactions (Eds.: F. Diederich, A. de Meijere), Wiley-VCH, Weinheim, Germany, 2nd ed., 2004;
 b) A. Roglans, A. Pla-Quintana, M. Moreno-Mañas, Chem. Rev. 2006, 106, 4622–4643; c) G. C. Fu, Acc. Chem. Res. 2008, 41, 1555–1564; d) A. T. Lindhardt, T. Skrydstrup, Chem. Eur. J. 2008, 14, 8756–8766; e) The Mizoroki-Heck reaction (Ed.: M. Oestreich), Wiley, Chichester, UK, 2009; f) X.-F. Wu, P. Anbarasan, H. Neumann, M. Beller, Angew. Chem. 2010, 122, 9231–9234; Angew. Chem. Int. Ed. 2010, 49, 9047–9050.
- [12] For selected reviews on the application of the Mizorki–Heck reaction to the synthesis of heterocycles, see: a) G. Zeni, R. C. Larock, *Chem. Rev.* 2006, 106, 4644–4680; b) *Palladium in Heterocyclic Chemistry* (Eds.: J. J. Li, G. W. Gribble), Elsevier, Am-

sterdam, **2007**; c) T. Muller, S. Bräse, in: *The Mizoroki-Heck reaction* (Ed.: M. Oestreich), Wiley, Chichester, UK, **2009**, pp. 215–258; d) K. C. Majumdar, S. Samanta, B. Sinha, *Synthesis* **2012**, *44*, 817–847.

- [13] See, for instance: a) K. C. Nicolaou, P. G. Bulger, D. Sarlah, Angew. Chem. 2005, 117, 4516–4563; Angew. Chem. Int. Ed. 2005, 44, 4442–4489; b) J.-P. Corbet, G. Mignani, Chem. Rev. 2006, 106, 2651–2710; c) A. B. Dounay, L. E. Overman, in: The Mizoroki–Heck reaction (Ed.: M. Oestreich), Wiley, Chichester, UK, 2009, pp. 533–568; d) E. A. Anderson, Org. Biomol. Chem. 2011, 9, 3997–4006.
- [14] For selected reviews on the asymmetric Heck reaction, see: a)
 M. Shibasaki, E. M. Vogl, T. Ohshima, Adv. Synth. Catal.
 2004, 346, 1533–1552; b) L. F. Tietze, H. Ila, H. P. Bell, Chem. Rev. 2004, 104, 3453–3516; c) L. F. Tietze, F. Lotz, in: Asymmetric Heck and other palladium-catalyzed reactions (Eds.: M. Christmann, S. Braese), Wiley-VCH, Weinheim, Germany,
 2007, pp. 147–152; d) T. Muller, S. Bräse, in: The Mizoroki-Heck reaction (Ed.: M. Oestreich), Wiley, Chichester, UK,
 2009, pp. 433–462; e) D. McCartney, P. J. Guiry, Chem. Soc. Rev. 2011, 40, 5122–5150.
- [15] a) J. S. Carey, D. Laffan, C. Thomson, M. T. Williamson, Org. Biomol. Chem. 2006, 4, 2337–2347; b) H. Doucet, J.-C. Hierso, Curr. Op. Drug Discov. Dev. 2007, 10, 672–690; c) C. Torborg, M. Beller, Adv. Synth. Catal. 2009, 351, 3027–3043; d) K. Suzuki, Y. Hori, Y. Nakayama, T. Kobayashi, J. Synth. Org. Chem. Jpn. 2011, 69, 1231–1240.
- [16] R. C. Larock, S. Babu, Tetrahedron Lett. 1987, 28, 5291-5294.
- [17] S. Caddick, W. Kofe, Tetrahedron Lett. 2002, 43, 9347–9350.
- [18] a) F. D. Suivre, M. A. Sortino, V. V. Kouznetzov, L. Y. Vargas, S. Zacchino, U. M. Cruz, R. D. Enriz, *Bioorg. Med. Chem.* 2006, 14, 1851–1862; b) C. M. Meléndez Gómez, V. V. Kouznetsov, M. A. Sortino, S. L. Álvarez, S. A. Zacchino, *Bioorg. Med. Chem.* 2008, 16, 7908–7920; c) V. V. Kouznetsov, L. Y. Vargas, M. A. Sortino, Y. Vázquez, M. P. Gupta, M. Frele, R. D. Enriz, *Bioorg. Med. Chem.* 2008, 16, 794–809.
- [19] See Supporting Information for experimental and characterization data.
- [20] a) T. Jeffery, J. Chem. Soc., Chem. Commun. 1984, 1287–1289;
 b) T. Jeffery, Tetrahedron 1996, 52, 10113–10130.
- [21] S. Lage, U. Martínez-Estibalez, N. Sotomayor, E. Lete, Adv. Synth. Catal. 2009, 351, 2460–2468.
- [22] M. W. Abelman, T. Oh, L. E. Overman, J. Org. Chem. 1987, 52, 4130–4133.
- [23] J. P. Knowles, A. Whiting, Org. Biomol. Chem. 2007, 5, 31–44, and references cited therein.
- [24] For selected examples, see: a) A. C. Albéniz, P. Espinet, R. López-Fernández, Organometallics 2006, 25, 5449–5455; b) S. Ogoshi, W. Yoshida, K. Ohe, S. Murai, Organometallics 1993, 12, 578–579.
- [25] Selected examples, Pd^{II}-catalyzed ene-type cyclization: a) M. Hatano, K. Mikami, J. Am. Chem. Soc. 2003, 125, 4703–4705. Friedel–Crafts reaction: b) T. Ishikawa, S. Marabe, T. Aikawa, T. Kudo, S. Saito, Org. Lett. 2004, 6, 2361–2364. Gold(I)-catalyzed alkyne hydroarylation: c) C. Gronnier, Y. Odabachian, F. Gagosz, Chem. Commun. 2011, 47, 218–220. Wittig reactions with quinol-4-ones: d) O. Venier, C. Pascal, A. Braun, C. Namane, P. Mougenot, O. Crespin, F. Pacquet, C. Mougenot, C. Monseau, B. Onofri, R. Dadji-Faïhun, C. Leger, M. Ben-Hassine, T. Van-Pham, J.-L. Ragot, C. Philippo, S. Güssregen, C. Engel, G. Farjot, L. Noah, K. Maniani, E. Nicolaï, Biorg. Med. Chem. 2011, 21, 2244–2251. Povarov reaction/carbene generation sequence: e) A. M. Jadhav, V. V. Pagar, R.-S. Liu, Angew. Chem. Int. Ed. 2012, 51, 11809–11813; see also ref.^[4]
- [26] F. Sanchez-Sancho, E. Mann, B. Herradón, Adv. Synth. Catal. 2001, 343, 360–368.
- [27] W. C. Still, H. Kann, A. J. Miltra, J. Org. Chem. 1978, 43, 2923–2925.

3021

- [28] a) D. D. Perrin, W. L. F Armarego, *Purification of Laboratory Chemicals*, Pergamon Press, Oxford, UK, 4th ed. **1997**; b) B. G. Williams, M. Lawton, *J. Org. Chem.* **2010**, *75*, 8351–8354.
- [29] C. A. Fleckenstein, H. Plenio, J. Org. Chem. 2008, 73, 3236-3244.
- [30] a) H. Li, C. Wang, H. Huang, X. Xu, Y. Li, *Tetrahedron Lett.* **2011**, *52*, 1108–1111; b) F. Xiao, W. Chen, Y. Liao, G.-J. Deng, Org. Biomol. Chem. **2012**, *10*, 8593–8596.

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