Synthesis of α-Acylamino-β-oxo Acid Esters¹

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 α -Acylamino- β -oxo acid esters are prepared by reaction of carboxylic acid chlorides with the dilithium derivative of alkyl hydrogen (acylamino)malonate.

The title α -acylamino- β -oxo acid esters **6** can be hydrogenated via dynamic kinetic resolution using ruthenium BINAP catalysts to yield $erythro-\alpha$ -amino- β -hydroxy esters with excellent enantio- and diastereoselectivities. The configurations of the two new stereogenic centres are determined by the configuration of the ligand in the homogeneous catalyst.

Suitably protected α -acylamino- β -oxo acid esters are usually prepared by a reaction sequence comprising Claisen condensation, oximation of the β -oxo acid esters by nitrous acid, and reduction with concomitant acylation. Few published methods concern the acylation of CH-acidic glycine esters, i.e. of dilithiated methyl hippurate³ or of isocyano acetate.⁴ In the course of the total synthesis of biphenomycin A,⁵ however, we were not able to achieve reduction of the oxime 1. Therefore, we have developed a smooth, one-step synthesis of the α -amino- β -oxo acid ester 3 by reaction of the carboxylic acid chloride 2 with the dilithium derivative of alkyl hydrogen (tert-butoxycarbonylamino)malonate.⁵ We now describe further examples of this methodology.

Alkyl hydrogen (acylamino)malonate 5 is easily accessible from the corresponding commercially available dialkyl aminomalonate hydrochloride, and is metalated to the dianion by treatment with 2 equivalents of butyllithium. The use of more than two equivalents of the lithium derivative is visualized with benzylidenebenzylamine as the indicator.

O
$$CO_2H$$
 $\frac{2 \text{ BUL}}{-70^{\circ}\text{C}, \text{ hexane}}$ $\frac{1}{70-85\%}$ NHCOR² $\frac{1}{70-85\%}$ R^1 $\frac{1}{100}$ CO₂R³ $\frac{1}{100}$ CO₂R³ $\frac{1}{100}$ CO₂R³ $\frac{1}{100}$ CO₂R³

4,6	R ¹	4,6	R ¹	5,6	R ²	R ³
а	n-C5H11	С	Bn	е	BnO	Me
b	Ph	d	(CH ₂) ₂ Ph	f	t-BuO	Εt

Acylation of the dilithium derivative was best achieved with carboxylic acid chlorides; other active carboxylic acid derivatives were found to be less suitable. Yields of up to 70% were realized with 1.2 equivalents of the dilithium compound and in excess of 80% with 2 equivalents of the dilithium derivative. The acylation of magnesium and lithium salts of malonyl monoesters and α -alkyl-substituted malonyl monoesters have been reported repeatedly.⁶

The ¹H NMR spectra were recorded on a Bruker AC-F (250 MHz). Melting points (Reichert microscope) are uncorrected. TLC was

Table. α-Amino-β-oxo Esters Prepared via Aminomalonic Acid Monoester

Prod- uct	Yield ⁵ (%)	mp (°C)	t _R (min)	1 H NMR (250 MHz, TMS) δ , J (Hz)
6ae	70	oil	4.10 ^a	0.84 (t, 3 H, $J = 6.8$), 1.1–1.35 (m, 4 H), 1.35–1.55 (m, 2 H), 2.59 (t, 2 H, $J = 7.2$), 3.33 (s, 3 H), 5.05 (d, 1 H, 8.2), 5.07 (s, 2 H), 7.36 (s, 5 H), 8.15 (d, 1 H, $J = 8.2$)°
6be	80	oil	5.77ª	3.68 (s, 3 H), 5.07 (s, 2 H), 6.00 (d, 1 H, $J = 8.3$), 7.33 (s, 5 H), 7.50–7.75 (m, 3 H), 7.96 (d, 2 H, $J = 7.2$), 8.42 (d, 1 H, $J = 8.3$)°
6bf	85	70-73	2.90ª	1.13 (t, 3H, $J = 7.1$), 1.38 (s, 9H), 4.13 (dq, 2H, $J = 1.5$, 7.1), 5.85 (d, 1H, $J = 8.5$), 7.50–7.75 (m, 3H), 7.82 (d, 1H, $J = 8.3$), 7.95 (d, 2H, $J = 7.4$)°
6cf	82	oil	3.18ª	1.19 (t, 3H, J = 7.1), 1.41 (s, 9H), 3.95 (d, 2H, 2.7), 4.14 (dq, 2H, J = 2.4, 7.1), 5.02 (d, 1H, J = 8.2), 7.15–7.40 (m, 5H), 7.7 (d, 1H, J = 8.2) $^{\circ}$
6df	87	oil	2.82ª	1.17 (t, 3 H, $J = 7.1$), 1.39 (s, 9 H), 2.78 (t, 2 H, $J = 6.8$), 2.93 (t, 2 H, $J = 6.8$), 4.11 (dq, 2 H, $J = 2.1$, 7.1), 4.95 (d, 1 H, $J = 8.1$), 7.10–7.35 (m, 5 H), 7.59 (d, 1 H, $J = 8.1$)°
32	82	foam	5.50 ^b	1.43 (s, 9 H), 1.95 (m, 1H), 2.27 (m, 1H), 3.59 (s, 3 H), 4.00 (dt, 2 H, J = 2.3, 12.0), 4.26 (dd, 2 H, J = 5.0, 10.9), 5.14 (s, 2 H), 5.26 (s, 2 H), 5.83 (d, 1 H, J = 7.8), 5.95 (s, 1 H), 6.08 (d, 1 H, J = 8.0), 6.96 (d, 1, J = 8.6), 6.98 (d, 1 H, J = 8.6), 7.25–7.50 (m, 11 H), 7.65 (dd, 1 H, J = 2.4, 8.6), 7.83 (d, 1 H, J = 2.4), 7.93 (d, 1 H, J = 2.4) ^d

^a Merck Si 60 [LiChrosorb (7 μ), 250-4, 2,0 mL/min]; eluent: hexane/EtOAc (8:2).

b Merck Si 60 [LiChrosorb (7 μ), 250-4, 2,0 mL/min]; eluent: hexane/EtOAc (7:3).

^c Solvent DMSO-d₆.

d Solvent CDCl3.

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done on silica gel (Merck Silica 60 $F_{2.54}$ sheets) and MPLC using Merck LiChroprep Si 60 (15–25 μ). HPLC was done with a LKB Instrument and a silica gel column (Merck LiChroCART 250-4 mm), LiChroSorb Si 60 7 μ . Satisfactory microanalyses were obtained for 6ae, be, bf, cf, df, 3, 5 e,f: $C \pm 0.24$, $H \pm 0.13$, $N \pm 0.19$.

Methyl Hydrogen (Benzyloxycarbonylamino)malonate (5e); Typical Procedure:

To a stirred solution of dimethyl aminomalonate hydrochloride (10 g, 54.5 mmol) in 2 M aq KHCO₃ (70 mL) and dioxane (70 mL) benzyl chloroformate (9.1 mL, 64.5 mmol) was added at 0 °C over a period of 10 min. The mixture was stirred for 24 h at r.t., dioxane was evaporated in vacuo and the aqueous layer was extracted with EtOAc (3 × 50 mL). The combined organic layers were dried (MgSO₄) and evaporated. The residue was recrystallized from EtOAc/hexane to give dimethyl (benzyloxycarbonylamino)malonate (12.2 g). The solution of crude dimethyl (benzyloxycarbonylamino)malonate (12.2 g, 45.7 mmol) in dioxane (50 mL) was treated with 1 M aq NaOH (45.7 mL) over a period of 3 h. The mixture was stirred for 15 h at r.t. and dioxane was evaporated in vacuo. The aqueous residue was washed with $\rm Et_2O$ (50 mL), then diluted with $\rm EtOAc$ (50 mL), acidified with 1 M $\rm H_2SO_4$ (25 mL) at 0 °C (stirring), separated and extracted with EtOAc (2 × 50 mL). The combined organic layers were dried (MgSO₄), concentrated and dried in vacuo (0.001 mbar); yield: 10.9 g (75% from dimethyl aminomalonate hydrochloride); mp 83-85°C.

 $^1\text{H NMR}$ (250 MHz, DMSO- d_6/TMS): $\delta=3.69$ (s, 3 H); 4.85 (d, 1 H, J=8.1 Hz); 5.06 (s, 2 H); 7.36 (s, 5 H); 8.17 (d, 1 H, J=8.1 Hz); 13.45 (br s, 1 H).

Ethyl Hydrogen (tert-Butoxycarbonylamino)malonate (5f): Compound 5f was prepared analogously from diethyl aminomalonate hydrochloride using di-tert-butyl dicarbonate; yield⁵: 10.8 g (80 %); mp 98–99 °C.

¹H NMR (250 MHz, DMSO- d_6 /TMS): $\delta = 1.19$ (t, 3 H, J = 7.1 Hz); 1.38 (s, 9 H); 4.14 (dq, 2 H, J = 1.2, 7.1 Hz); [4.64 (br) and 4.71 (d, J = 8.1 Hz) 1 H, ratio: 1:7]; [6.86 (br) and 7.50 (d, J = 8.1 Hz) 1 H ratio: 1:7]; 14.4 (br s, 1 H).

α-Acylamino-β-oxo Acid Esters 6; General Procedure:

To a solution of N-protected alkyl hydrogen aminomalonate (2 mmol) in THF (10 mL) at $-80\,^{\circ}$ C, BuLi (2.5 mL, 4 mmol; 1.6 M solution in hexane) and the acid chloride (1 mmol) were added successively over a period of 10 min. The mixture was stirred for a further 2 h during which the temperature reached $-50\,^{\circ}$ C. The reaction was quenched with sat. aq NH₄Cl (10 mL), stirred at r. t. for another 10 min and extracted with EtOAc (3 × 10 mL). The combined organic layers were dried (MgSO₄) and concentrated in vacuo. The residue was purified by MPLC (eluent hexane/EtOAc) to give the pure α -amino- β -oxo ester.

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- Amino acids and peptids 92; part 91: Schmidt, U.; Kleefeldt, A.; Mangold, R. J. Chem. Soc., Chem. Commun., in press.
- (2) Noyori, R.; Ikeda, T.; Onkuma, T.; Widhalm, M.; Kitamura, M.; Takaya, H.; Akutagawa, S.; Sayo, N.; Saito, T.; Taketoni, T.; Kumobayashi, H. J. Am. Chem. Soc. 1989, 111, 9134. Mashima, K.; Kusano, K.; Ohta, T.; Noyori, R.; Takaya, H. J. Chem. Soc., Chem. Commun. 1989, 1208.
- (3) Harding, K. E.; Moreno, L. N.; Nace, V. M. J. Org. Chem. 1981, 46, 2809.
- (4) Suzuki, M.; Jwasaki, T.; Matsumotu, K.; Okumura, K. Synth. Commun. 1972, 2, 237.
- (5) Schmidt, U.; Leitenberger, V.; Griesser, H.; Schmidt, J.; Meyer, R. Synthesis, 1993, in press.
- (6) Ireland, R. E.; Marshall, J. A. J. Am. Chem. Soc. 1959, 81, 2907.
 Bram, G.; Vilkas, M. Bull. Soc. Chim. Fr. 1964, 945.
 Clezy, P. S.; Fookes, C. J. R. Aust. J. Chem. 1977, 30, 1799.
 Pollet, P.; Gelin, S. Synthesis 1978, 142.