

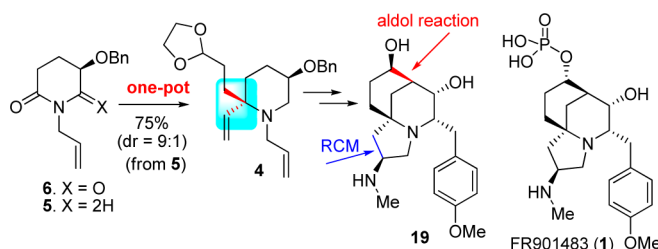
A Formal Enantioselective Total Synthesis
of FR901483Hao-Hua Huo,[†] Hong-Kui Zhang,[†] Xiao-Er Xia,[†] and Pei-Qiang Huang^{*,†,‡}

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ABSTRACT



A formal enantioselective total synthesis of the potent immunosuppressant FR901483 (**1**) has been accomplished. Our approach features the use of chiron **6** as the starting material, the application of the one-pot amide reductive bisalkylation method to construct the chiral aza-quaternary center (dr = 9:1), regio- and diastereoselective intramolecular aldol reaction to build the bridged ring, and ring closing metathesis to form the 3-pyrrolin-2-one ring.

FR901483 (**1**, Figure 1) is a potent immunosuppressant isolated from the fermentation broth of the *Cladobotryum* sp. No. 11231 by a research group at the Japan Fujisawa Pharmaceutical Company in 1996.¹ Due to its promising

biological activity and challenging structure, this molecule has attracted the attention of many research groups. To date, six enantioselective total syntheses/formal total syntheses^{2,3} and numerous synthetic studies^{4,5} toward FR901483 (**1**) have been reported. As a continuation

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of our endeavor to develop step-economical⁶ and 3-benzyloxyglutarimide⁷-based synthetic methodologies for the asymmetric synthesis of piperidine ring-containing alkaloids,⁸ we have embarked on the enantioselective total synthesis of FR901483. We report herein a formal enantioselective total synthesis of FR901483 (**1**).

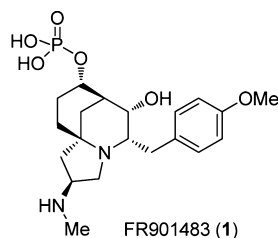
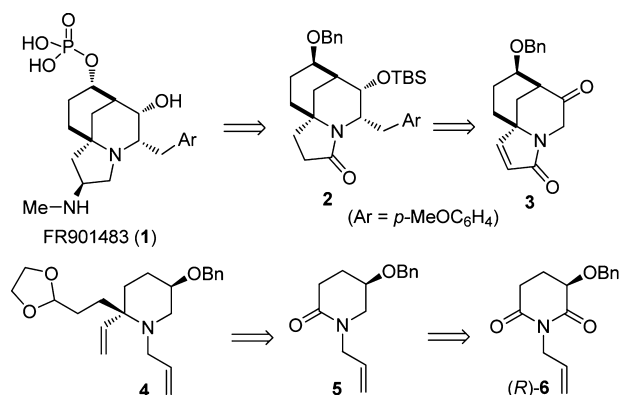


Figure 1. Potent immunosuppressant FR901483 (**1**).

As illustrated retrosynthetically in Scheme 1, our synthetic plan is based on two key synthetic methodologies developed from our laboratory, namely, the use of chiron **6**⁸ as the starting point and the amide reductive bisalkylation method^{6b} for a one-pot conversion of piperidin-2-one **5** to piperidine **4** with the formation of two C–C bonds.

Scheme 1. Retrosynthetic Analysis of FR901483 (**1**)

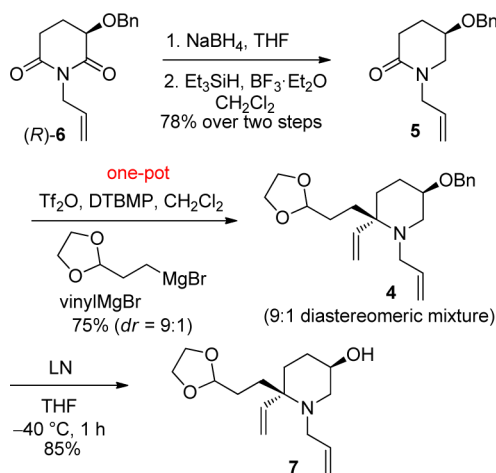


The synthesis commenced with the partial and regioselective reduction^{8b} of (3*R*)-benzyloxyglutarimide **6** with NaBH₄ followed by reductive dehydroxylation (BF₃•OEt₂, Et₃SiH, CH₂Cl₂) via an iminium ion intermediate^{8a,c} (Scheme 2). The subsequent bisalkylation of lactam **5** (Tf₂O, DTBMP, CH₂Cl₂; then successive addition of two Grignard reagents) proceeded as planned to produce the desired lactam **4** in 75% yield. Remarkably, the reaction proceeded with high diastereoselectivity (*dr* = 9:1,

determined by ¹H NMR of the crude sample). The stereochemistry of the major diastereomer was deduced to be *trans* on the basis of the mechanistic consideration that the vinyl group approaches the iminium ion intermediate from the α-side opposing the benzyloxy group, which was confirmed at a latter stage. It is worth mentioning that, among various synthetic approaches for the synthesis of FR901483 (**1**), this is the first example utilizing the one-pot amide reductive bisalkylation method to construct the chiral aza-quaternary center.

The selective *O*-debenzylation of compound **4** was achieved by treatment of piperidine **4** (diastereomeric mixture) with lithium naphthalenide (LN)⁹ in THF at –40 °C for 1 h, which gave, after chromatographic separation, the piperidin-3-ol **7** as a pure diastereomer in 85% yield.

Scheme 2. Stereoselective Synthesis of the Key Intermediate **7**



Next, the piperidin-3-ol **7** needed to be oxidized to the ketone **8** before the key intramolecular aldol ring closure reaction could be commenced. After unsuccessful trials with the Dess-Martin periodinane, and partial success with the Ley oxidation (35% yield), the Swern oxidation was attempted. It was found that by the use of an extreme excess of triethylamine, the desired ketone could be obtained in 90% yield [(COCl)₂ 6 equiv, DMSO 12 equiv, CH₂Cl₂, –78 °C, 1 h; and then NEt₃ 25 equiv]. After deacetalization of compound **8** with a 4 *N* HCl solution, the resulting keto-aldehyde, in the form of its hydrochloride salt, was heated with ethylene glycol (4.0 equiv)¹⁰ and CSA (0.3 equiv) in toluene at 90 °C. The desired regioselective intramolecular aldol reaction took place smoothly with concomitant acetalization of the ketone to give compound **9** as the solely observable regio- and diastereomer in 53% yield (Scheme 3). The relative stereochemistry of **9** was established on the basis of the observed

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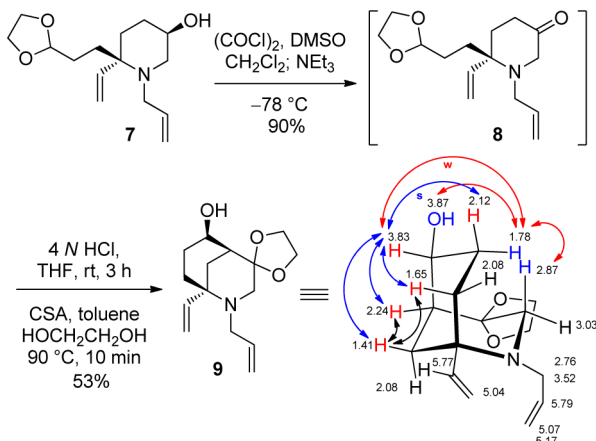
(8) For recent examples, see: (a) Fu, R.; Ye, J.-L.; Dai, X.-J.; Ruan, Y.-P.; Huang, P.-Q. *J. Org. Chem.* **2010**, *75*, 4230–4243. (b) Yang, R.-F.; Huang, P.-Q. *Chem.—Eur. J.* **2010**, *16*, 10319–10322. (c) Tuo, S.-C.; Ye, J.-L.; Wang, A.-E.; Huang, S.-Y.; Huang, P.-Q. *Org. Lett.* **2011**, *13*, 5270–5273.

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(10) If the hydrochloride salt was neutralized and the mixture was heated in the absence of ethylene glycol, a regiomer aldol product was obtained in 65% yield as a single diastereomer.

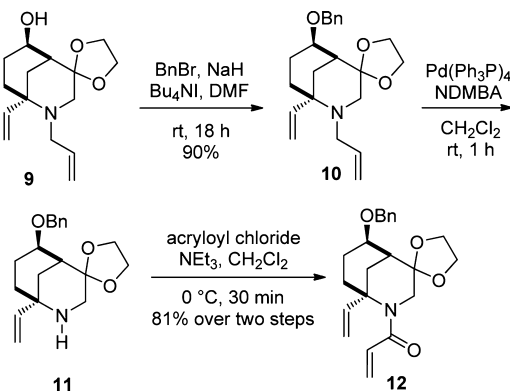
strong correlation between the methine proton (δ 3.83) and the H_{eq} (δ 2.12), and a smaller correlation between the methine proton (δ 3.83) and the H_{ax} (δ 1.78) in its NOESY spectrum (Scheme 3).

Scheme 3. Synthesis and the Observed NOE Correlations of Compound **9**



After securing access to the bicyclic core **9**, the conversion of the latter to a precursor for the ring closing metathesis (RCM) reaction was investigated. Thus, the hydroxyl group was protected as its benzyl ether **10** (BnBr, NaH, Bu₄NI, DMF, 90% yield) (Scheme 4). Compound **10** was subjected to Pd-catalyzed *N*-deallylation [Pd(Ph₃P)₄ (0.01 equiv), NDMBA (3 equiv), CH₂Cl₂, rt, 1 h],¹¹ and acryloylation (acryloyl chloride, NEt₃, CH₂Cl₂, 0 °C, 30 min), which gave acrylamide **12** in 81% yield over two steps.

Scheme 4. Synthesis of the Acrylamide Derivative **12**

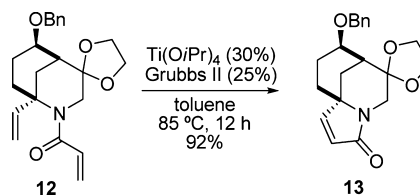


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We next investigated the RCM reaction of **12**. Formation of α,β -unsaturated lactams by the RCM reaction of acrylamides has been well documented using either Grubbs' first or second generation catalyst.¹² The reaction generally proceeded in CH₂Cl₂.^{5f,12} However in our case all attempts to perform the RCM reaction of **12** by the use of the Grubbs first or second generation catalyst in CH₂Cl₂ were unsuccessful. Considering the steric hindrance of the substrate, it was envisioned that a higher reaction temperature would favor the reaction. Indeed, when a toluene solution of acrylamide **12** and the Grubbs second generation catalyst (25%) was heated to 85 °C for 12 h, the desired cyclized product **13** was obtained in 30% yield, along with 60% of the recovered starting material. Remarkably, when the reaction (Grubbs II 25%, toluene, 85 °C) was run in the presence of 30% molar equiv of Ti(O*i*Pr)₄,¹³ the desired cyclic core **13** was obtained in 92% yield (Scheme 5).

Scheme 5. Construction of the Tricyclic Core **13** by the RCM Reaction



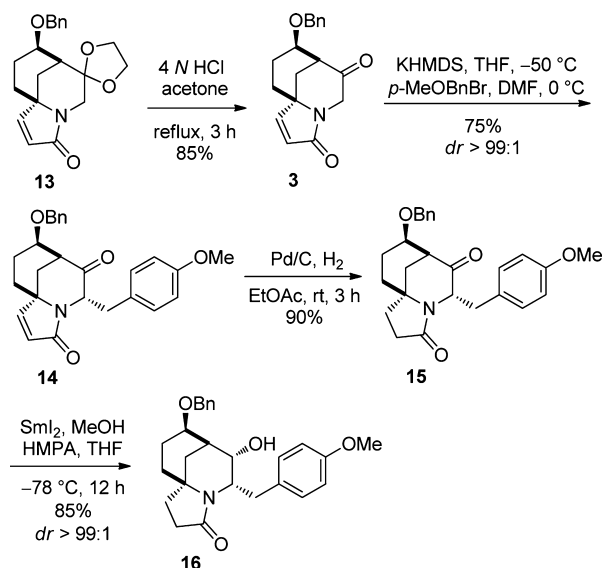
Now we were in a position to undertake the functional group exchange and the functionalization of the core structure. Thus, acetal **13** was cleaved by treatment with a solution of 4 *N* HCl in acetone at reflux for 3 h to give the keto-lactam **3** in 85% yield (Scheme 6). Inversed addition^{3a} of the enolate, generated from ketone **3** by deprotonation with KHMDS (1 equiv) in THF at –50 °C, with *p*-methoxybenzyl bromide (3.0 equiv) in DMF at 0 °C for 20 min produced compound **14** as the only observable regio- and diastereomer in 75% yield. Pd/C-catalyzed selective hydrogenation of **14** (Pd/C 30 wt %, H₂, 1 atm, EtOAc) produced lactam **15** in 90% yield. Subjection of ketone-lactam **15** to the SmI₂-mediated reduction^{3b,14} (SmI₂ 5 equiv, HMPA 25 equiv, MeOH 10 equiv, THF, –78 °C, 12 h) yielded the desired *cis*-diastereomer **16** as the only observable diastereomer in 85% yield. The spectral data of our synthetic **16** matched those reported for the racemic **16**,^{5d} which confirmed the relative stereochemistry of our synthetic product.

We next turned our attention to the stereoselective α -amination of lactam **16**. Thus, the hydroxyl group in **16** was first protected (TBSOTf, NEt₃, CH₂Cl₂, 93%) as TBS

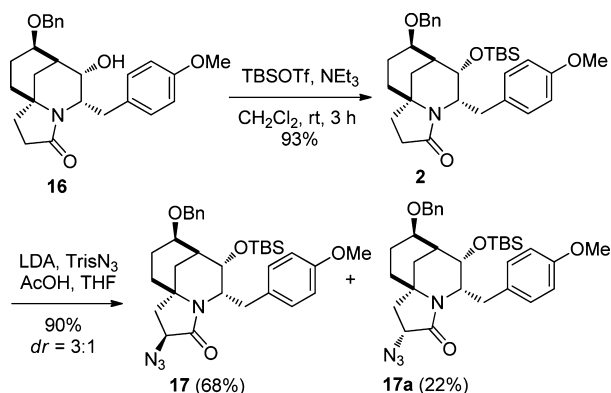
(13) Ghosh, A. K.; Cappiello, J.; Shin, D. *Tetrahedron Lett.* **1998**, *39*, 4651–4654.

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Scheme 6. Diastereoselective Synthesis of Compound 16



Scheme 7. Synthesis of Compound 17

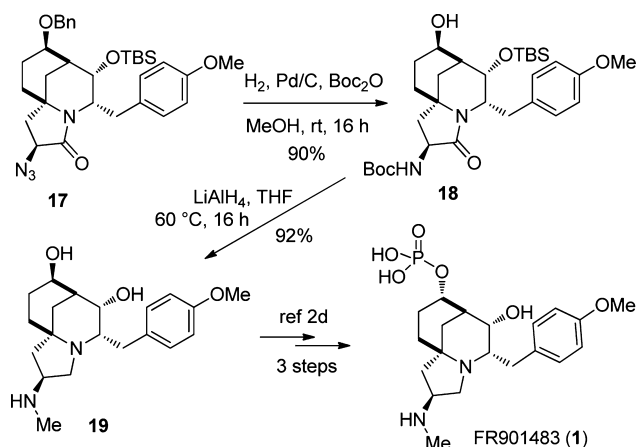


ether **2** (Scheme 7). Successive treatment of lactam **2** with LDA (3.0 equiv) and TrisN₃ (3 equiv) at $-78\text{ }^{\circ}\text{C}$ for 5 min, followed by addition of HOAc, led to the formation of the desired azide **17** as a separable diastereomeric mixture in a 3:1 ratio with a combined yield of 90%. The stereochemistry of the major diastereomer **17**, which was assumed to be formed by approaching the electrophile from the less hindered β -face, was determined by NOESY experiments.

To convert the azido group to the *N*-methylamino group, azide **17** was hydrogenated (10% Pd/C 30 wt %, H₂ 1 atm) in the presence of Boc₂O (1.2 equiv), which gave

the concomitantly protected compound **18** in 90% yield (Scheme 8). Treatment of compound **18** with a large excess of LiAlH₄ (30 equiv) in THF at $60\text{ }^{\circ}\text{C}$ for 16 h afforded the amino-diol **19** in 92% yield. It is worth noting that, in this scenario, *N*-Boc reduction to give *N*-Me group, lactam reduction to give pyrrolidine, and *O*-desilylation occurred sequentially. The ¹H and ¹³C NMR spectral data of compound **19** are in agreement with those reported by Ciufolini.^{2d} Since compound **19** has been converted into FR901483 (**1**) in three steps,^{2d} our synthetic approach thus constitutes a formal total synthesis of (–)-FR901483 (**1**).

Scheme 8. Synthesis of the Known Precursor (19) of FR901483



In summary, starting from the known chiron **6**, we have achieved the synthesis of the advanced intermediate **19** in 18 steps with an overall yield of 4.1%, which constitutes a formal enantioselective total synthesis of FR901483 (**1**).

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Supporting Information Available. Full experimental procedures; ¹H and ¹³C NMR spectra of all new compounds; NOESY spectra of compounds **9** and **17**. This material is available free of charge via the Internet at <http://pubs.acs.org>.

The authors declare no competing financial interest.