The synthesis of 3-methyleneindol-2(3*H*)-ones related to mitomycins *via* 5-*exo-dig* aryl radical cyclisation

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Introduction

The mitomycins are a group of antibiotics with activity against Gram-positive and Gram-negative bacteria and also against several kinds of tumours.¹ They all contain the common structure **1**.



We have previously described two routes to the pyrroloindole skeleton² starting from an oxindole and involving either an intra- or intermolecular organolithium addition to the oxindole carbonyl.³

Having developed a new approach to the core ring system and prepared a suitably substituted oxindole,⁴ we turned our attention to the remaining problems. One of the obvious drawbacks of the oxindole-based approach is that the aryl radical cyclisation chemistry we have developed leads to simple nonfunctionalised substituents at the oxindole C-3 position. In order to develop an approach to mitomycins we require a C_1 -substituent at this position containing suitable functionality to be readily converted into the mitomycin C-9 carbamate. One possibility involves the 5-*exo-dig* cyclisation of an aryl radical such as **2** (Scheme 1) to produce a 3-methylene oxindole **3**.



Conversion of the exomethylene group into the desired carbamate could be envisaged by a variety of reactions.

Although there are now many examples of 5-exo-trig cyclisations of aryl radicals⁵ there are few examples of 5-exo-dig cyclisations. Boger and Coleman⁶ used such cyclisations in the total synthesis of the antitumour antibiotic CC-1065 (Scheme 2) and Dittami and Ramanathan⁷ used this reaction to



Scheme 2





We now wish to report our attempts at 5-*exo-dig* aryl radical cyclisations to produce oxindoles which contain an exocyclic double bond.

Results

Initially, the alkynyl amide 8a was synthesised in order to study the cyclisation of the derived aryl radical. Addition of 2bromoaniline to propiolic acid using dicyclohexylcarbodiimide (DCC) gave 7a in 75% yield (Scheme 4). All attempts to prepare this amide using the usual acid chloride route failed. It is necessary to N-alkylate the amide first in order to convert the amide into the correct rotamer for the radical cyclisation to occur. There has been extensive work in our group on the conformational aspects of radical cyclisations for the formation of oxindoles.⁸ Reaction of 7a with sodium hydride and methyl iodide gave the tertiary amide 8a in 70% yield. Cyclisation of 8a was attempted using 1.1 equivalents of tributyltin hydride (TBTH) and 10 mol% of azobisisobutyronitrile (AIBN) added to a solution of the amide 8a in toluene under reflux. This resulted in a mixture of three products (Scheme 5) in 42% yield which were tentatively identified as 11 arising from hydrostannylation⁹ of the acetylenic bond, 12 formed by hydrostannylation of the triple bond followed by cyclisation and 13 formed by replacement of the acetylenic hydrogen by SnBu₃. The major product appeared to be 13.

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Reaction of the alkyne with TBTH was clearly a major problem. In order to explore this reaction further, a range of amides 8 and 9 were synthesised. The obvious way to slow down this unwanted reaction was to introduce a substituent onto the terminus of the alkyne. The simplest substituent to introduce was a methyl group although this would not be suitable for the formation of the C-9 carbamate of the mitomycins. 2-Bromoaniline was reacted with but-2-ynoic acid using DCC to give butynamide 7b in 21% yield. N-Methylation under the usual conditions gave 8b in 72% yield. To fulfill the long-term aim of the project, a removable group was required on the terminus of the alkyne and a silvl group was chosen.¹⁰ Initial studies using the trimethylsilyl group proved unsuccessful at the radical cyclisation stage and so attention was turned to the bulkier tert-butyldimethylsilyl (TBDMS) group. In addition, we introduced an N-allyl group into the cyclisation precursors to provide a basis for the synthesis of the third ring of the mitomycins. Reaction of 7a with allyl bromide using sodium hydride as base gave the desired N-allyl amide 8c in a poor 18% yield. This was improved to 68% by using allyl triflate in place of the bromide. Silvlation of 8a and 8c was achieved in 98% and 99% yields respectively using lithium hexamethyldisilazide at -78 °C with addition of tert-butyldimethylsilyl chloride and warming to room temperature. With reliable syn-

 Table 1 Ratio of 3-methyleneindol-2(3H)-one isomers produced in the radical cyclisation in Scheme 7

R ¹	R ²	Ratio $E:Z$	Combined yield (%)
Me	TBDMS	1:3	62
Allyl	TBDMS	1:4.7	59
CH ₂ -Dioxolane	TBDMS	1:3.5	67

theses of **8b**, **9a** and **9b** completed, the radical cyclisation was studied.

Radical cyclisation of 8b under the normal radical conditions gave an inseparable mixture of three products which were tentatively assigned as the Z- and E-isomers of 10a along with starting material in combined 40% yield. Evidence for the cyclised products was obtained from the mass spectrum which showed M^+ , 173. The presence of unreacted starting material is indicative of a poor radical chain presumably owing to the relatively slow addition of the nucleophilic aryl radical to the carbon-carbon triple bond. Consequently, in the reactions of 9a and 9b, 10 mol% of AIBN was added every hour for 3 hours to overcome this problem. In this way reasonable yields of cyclised products were obtained which consisted of separable mixtures of E- and Z-isomers of the desired methylene oxindoles 10. The results are summarised in Table 1. The isomers were assigned on the basis of NOE experiments which showed an enhancement between the olefinic proton and an aromatic proton in the Z-isomers. Simultaneously a cyclisation substrate was prepared which we envisaged would lead to the synthesis of the pyrroloindole skeleton of mitomycin. This involved aryl radical cyclisation of an anilide carrying a protected acetaldehyde unit on the nitrogen (Scheme 6). The cyclisation precursor 14 was prepared by ozonolysis of 9b followed by protection of the aldehyde using ethane-1,2-diol. Radical cyclisation then afforded the E- and Z-isomers of oxindole 15 in a combined 67% vield (see Table 1).



In all the cyclisations studied, the Z-isomer of the 3methylene oxindole was the major product. The intermediate vinyl radical is not configurationally stable,¹¹ therefore a mixture of E- and Z-isomers around the double bond was expected. It seems likely that an unfavourable steric interaction between R^2 and the C-4 hydrogen leads to preferential formation of the less hindered Z-isomer (Scheme 7). The



bulky TBDMS group would be expected to be particularly discriminating in this respect.

In order to introduce the C-9 carbamate group found in the mitomycins, oxygenation of the exocyclic double bond of the 3-methylene oxindoles is required. The reactivity of this alkene is not clear as it is conjugated to the carbonyl group but carries an electron-donating silyl group. Initial attempts to epoxidise the Z-isomer of **10b** using base-catalysis and either hydrogen peroxide or *tert*-butyl hydroperoxide failed. However reaction with excess *m*-chloroperoxybenzoic acid (MCPBA) gave the epoxide **16a** in 43% yield along with recovered starting material (Scheme 8). A clear illustration of the reactivity of this alkene



was obtained when the Z-isomer of **10c** carrying an N-allyl group was epoxidised in the same manner to give epoxide **16b** in 51% yield with 40% recovered starting material.

In conclusion we have shown that the facile 5-exo-dig aryl radical cyclisation provides a concise route to novel silyl-substituted methylene indol-2-ones. The epoxidation of these systems can be achieved using MCPBA.

Experimental

General conditions

¹H and ¹³C NMR spectra were recorded on a Bruker AM360 spectrometer at 360 and 90 MHz respectively or on a Bruker AM400 spectrometer at 400 and 100 MHz respectively using CDCl₃ as solvent with SiMe₄ as an internal standard, unless otherwise stated. J-Values are given in Hz. IR spectra were recorded on a Perkin-Elmer Paragon 1000 Infrared spectrometer, using Nujol mulls or dichloromethane solutions unless otherwise stated. All values in cm⁻¹. Data are presented where w, m, s, represent weak, medium and strong absorptions respectively. Mass spectral data was recorded on a JEOL AX505W with complement data system. Samples were ionised electronically at 70 eV with typical accelerating voltage of 6 kV. Melting points were determined using a Kofler hot plate apparatus and are uncorrected. All column chromatography was carred out using the flash chromatography technique of Still, using Merck 60 (230-400 mesh) silica gel. Analytical TLC was carried out on Merck plastic backed TLC plates, coated with silica gel 60 F-254. Plates were visualised using ultraviolet light, unless otherwise stated. Eluting solvent systems are stated where appropriate. All dry reactions were performed in an inert argon atmosphere using a vacuum-argon manifold for the exclusion of water. Stirring was by internal magnetic bead. All syringes, needles and glassware were pre-dried at 110 °C and cooled in an anhydrous atmosphere before use. Diethyl ether, THF, and toluene were pre-dried over Na wire and refluxed over Na under Ar with benzophenone as an indicator in the reaction vessel. Dichloromethane was refluxed under Ar, over CaH₂ and distilled directly into the reaction vessel.

Tri-*n***-butylstannane.** This was prepared by the method of Szammer and Otvos.¹² Bis(tri-*n*-butyltin)oxide (50 g, 0.084 mol) was dissolved in absolute ethanol (250 ml) at 0 °C under argon. Sodium borohydride (2.22 g, 0.059 mol) was added and the reaction allowed to warm to room temperature then stirred for one hour. The ethanol was removed *in vacuo* and the resultant white slurry was dissolved in hexane, washed with water (3 × 50 ml), dried (MgSO₄) and evaporated under reduced pressure. The resultant liquid was rapidly distilled under reduced pressure. Yield: 90%. *R*_f(hexane) 0.83; $\delta_{\rm H}(360 \text{ MHz}; \text{CDCl}_3) 0.92$ (9H, t, *J* 7.3, 3 × CH₃), 1.27–1.41 (12H, m, 3 × H-1 + 3 × H-3), 1.60–1.69 (6H, m, 3 × H-2), 5.29 (1H, t, *J* 1.0, Sn-H).

Preparation of amides: general procedure

2-Bromoaniline was added to a solution of the acid in dichloromethane. A solution of DCC in dichloromethane was then added dropwise at 0 °C. The resulting solution was stirred for 30 minutes after which time a white precipitate formed. The reaction mixture was stirred for a further 3 hours at room temperature, then cooled in ice and filtered, the residue was washed with dichloromethane. The filtrate was washed with 2 M HCl (3 × 200 ml), then with sodium hydrogen carbonate solution (3 × 200 ml) and water (3 × 200 ml). The organic layer was dried (MgSO₄) and evaporated under reduced pressure.

N-(2'-Bromophenyl)propynamide, 7a. 2-Bromoaniline (40.5 g, 0.235 mol) and propiolic acid (15 g, 0.214 mol) gave the amide 7a as a white solid (36 g, 75%) after chromatography (SiO₂, 7% EtOAc–hexane); mp 96–98 °C (Found: C, 47.89; H, 2.59; N, 6.09%. C₉H₆BrNO requires C, 48.25; H, 2.70; N, 6.25%); $R_{\rm f}(30\%$ EtOAc–hexane) 0.33; $v_{\rm max}/{\rm cm}^{-1}$ 3288 (m, acetylenic C–H), 3199 (m, secondary amide N–H), 2110 (s, acetylenic C=C), 1633 (s, amide C=O); $\delta_{\rm H}(360$ MHz; CDCl₃) 8.29 (1H, d,

J 8.0, H-6'), 7.96 (1H, br s, N–H), 7.56 (1H, d, J 8.0, H-3'), 7.33 (1H, t, J 8.0, H-5'), 7.02 (1H, t, J 8.0, H-4'), 3.01 (1H, s, H-3); $\delta_{\rm C}({\rm CDCl}_3)$ 149.5 (C=O), 134.9 (C-1'), 132.4 (C-6'), 128.6 (C-3'), 126.1 (C-5'), 122.3 (C-4'), 113.2 (C-2'), 81.5 (HC=C), 74.8 (HC=C); *m*/*z* 225 (63.3%, ⁸¹Br, M⁺), 223 (63.3, ⁷⁹Br, M⁺) 173 (62.4, M⁺ – C₃O, ⁸¹Br), 171 (65.2, M⁺ – C₃O, ⁷⁹Br), 144 (94.8, M⁺ – Br), 64 (100) (Found: M⁺ (⁷⁹Br) 222.9632. C₉H₆⁷⁹BrNO requires *M* 222.9633).

N-(2'-Bromophenyl)but-2-ynamide, 7b. 2-Bromoaniline (4.5 g, 26 mmol) and butynoic acid (2 g, 23 mmol) gave a white solid which was recrystallised from EtOAc and hexane to give 7b (1.2 g, 21%), mp 78–79 °C (Found: C, 50.29; H, 3.22; N, 5.72%. C₁₀H₈BrNO requires C, 50.45; H, 3.39; N, 5.88%); *R*_f(30% EtOAc–hexane) 0.26; *v*_{max}/cm⁻¹ 2235 (C≡C) 1741 (secondary amide C=O); *δ*_H(360 MHz; CDCl₃) 8.31 (1H, d, *J* 8.5, H-6'), 7.85 (1H, br s, NH), 7.53 (1H, dd, *J* 8.5, 1.3, H-3'), 7.31 (1H, td, *J* 8.5, 1.3, H-5'), 6.99 (1H, t, *J* 8.5, H-4'), 2.03 (3H, s, Me); *δ*_c(CDCl₃) 150.9 (C=O), 135.3 (C-1'), 132.3 (C-6'), 128.4 (C-3'), 125.6 (C-5'), 122.1 (C-4'), 112.9 (C-2'), 85.3 (CH₃), 81.5 (C≡C-CH₃), 75.2 (*C*≡C-CH₃); *m*/*z* 237 (8.3%, M⁺, ⁷⁹Br), 239 (6.6, M⁺, ⁸¹Br), 158 (60.4, M⁺ – Br), 144 (41.7, M⁺ – BrN), 131 (97, M⁺ – CHBrN), 119 (100, M⁺ – C₂HBrN) (Found M⁺ (⁷⁹Br) 236.9793. C₁₀H₈⁷⁹BrNO requires *M* 236.9789).

Preparation of N-alkyl amides: general procedure

The amide in THF was added to a stirred suspension of sodium hydride (1.1 equivalents) in THF at 0 °C under argon. The reaction was allowed to stir for 1 hour after which time hydrogen evolution had ceased. The alkyl halide was then added and the reaction allowed to stir at room temperature overnight. The THF was removed *in vacuo* and the resultant solid dissolved in ether (200 ml), then washed with water (3×200 ml). The ethereal layer was dried (MgSO₄) and evaporated under reduced pressure.

N-(2'-Bromophenyl)-*N*-methylpropynamide, 8a. Amide 7a (9 g, 40 mmol) and methyl iodide (5.7 g, 40 mmol) gave a white solid 8a (6.7 g, 70%) after chromatography (SiO₂, 15% EtOAc–hexane), mp 88–89 °C (Found: C, 50.33; H, 3.08; N, 5.74%. C₁₀H₈BrNO requires C, 50.45; H, 3.39; N, 5.88%); *R*_f(30% EtOAc–hexane) 0.22; v_{max} /cm⁻¹ 3216 (s, acetylenic C–H), 2103 (s, acetylenic C=C) 1639 (s, C=O); $\delta_{\rm H}$ (360 MHz; CDCl₃) 7.68 (1H, dd, *J* 7.9, 1.4, H-3'), 7.21–7.43 (3H, m, H-4', H-5', H-6'), 3.26 (3H, s, NMe), 2.75 (1H, s, H-3); $\delta_{\rm C}$ (CDCl₃) 153.2 (C=O), 141.4 (C-1'), 133.6 (C-3'), 130.3 (C-5', C-6'), 128.6 (C-4'), 123.6 (C-2'), 79.9 (HC=C), 75.9 (HC=C), 35.2 (N-Me); *m*/*z* 239, (0.6%, ⁸¹Br, M⁺), 237 (2.3, ⁷⁹Br, M⁺), 186 (53.6%, M⁺ – C₃HO, ⁸¹Br), 184 (30.1, M⁺ – C₃HO, ⁸¹Br), 157 (17.6, M⁺ – C₄H₄NO, ⁸¹Br) 155 (22.7, M⁺ – C₄H₄NO, ⁷⁹Br) (Found: M⁺ (⁷⁹Br) 236.9878. C₁₀H₈⁷⁹BrNO requires *M* 236.9789).

N-(2'-Bromophenyl)-N-methylbutynamide, 8b. Amide 7b (1.2 g, 5.1 mmol) and methyl iodide (0.78 g, 5.5 mmol) gave 8b as a white solid, (0.92 g, 72%), mp 61-63 °C (Found: C, 52.36; H, 3.77; N, 5.46%. C₁₁H₁₀BrNO requires C, 52.41; H, 3.96; N, 5.56%); $R_{\rm f}(30\%$ EtOAc–hexane) 0.31; $v_{\rm max}/{\rm cm}^{-1}$ 2233 (C=C), 1642 (tertiary amide C=O); $\delta_{\rm H}$ (360 MHz; CDCl₃) 7.67 (1H, dd, J 8.0, 1.5, H-3'), 7.39 (1H, td, J 7.8, 1.4, H-5'), 7.26 (2H, m, H-4' + H-6'), 3.48 (s, NMe minor rotamer), 3.23 (3H, s, NMe), 2.09 (s, CH₃, minor rotamer), 1.70 (3H, s, CH₃); $\delta_{\rm C}$ (CDCl₃) 154.4 (C=O), 142.0 (C-1'), 133.7 (C-3' minor rotamer), 133.5 (C-3'), 130.4 (C-5'), 129.9 (C-6'), 129.5 (minor rotamer), 129.3 (minor rotamer) 128.7 (minor rotamer), 128.5 (C-4'), 123.6 (C-2'), 89.4 (C≡C-CH₃), 73.7 (C≡C-CH₃) 38.9 (NCH₃ minor rotamer), 35.1 (NCH₃), 4.2 (C=CCH₃ minor rotamer), 3.8 (C=CCH₃); m/z 253 (3.2%, M⁺, ⁸¹Br), 251 (1.0, M⁺, ⁷⁹Br), 172 (35.6, M⁺ - Br), 83.8 (100, M⁺ - C₆H₄BrN) (Found: M⁺ (⁷⁹Br) 250.9951. C₁₁H₁₀⁷⁹BrNO requires M 250.9945).

N-(2'-Bromophenyl)-*N*-prop-2-enylpropynamide, 8c. Amide 7a (0.95 g, 4.3 mmol) and allyl triflate (2 g, 10.5 mmol) gave 8c as a brown oil (0.77 g, 68%); $R_f(30\%$ EtOAc–hexane) 0.33; v_{max}/cm^{-1} 3289 (acetylenic C–H) 2107 (C=C), 1650 (tertiary amide C=O); $\delta_H(360$ MHz; CDCl₃) 7.68 (1H, dd, *J* 7.3, 0.9, H-6'), 7.36 (1H, dd, *J* 7.5, 1.3, H-3'), 7.28 (2H, m, H-4', H-5'), 5.86 (1H, m, H-2"), 5.16, (1H, dd, *J* 10.0, 1.0, H-3"), 5.09 (1H, dd, *J* 16.9, 1.0, H-3"), 4.77 (1H, dd, *J* 14.7, 5.7, H-1"), 3.83 (1H, dd, *J* 14.7, 7.6, H-1"), 2.74 (1H, s, H-3); *m/z* 265 (2.2%, M⁺, ⁸¹Br), 263 (1.8, M⁺, ⁷⁹Br), 184 (58.6, M⁺ − Br), 84 (100) (Found: M⁺ (⁷⁹Br) 262.9946. C₁₂H₁₀⁷⁹BrNO requires *M* 262.9946).

Silylation of alkynes: general procedure

To a stirred solution of the *N*-alkylated amide in THF at -78 °C under argon was added lithium bis(trimethylsilyl)amide (1.2 equivalents). After 30 minutes *tert*-butyldimethylsilyl chloride was added. The reaction was stirred for 30 minutes then slowly warmed to room temperature over 60 minutes. The reaction mixture was quenched with ammonium chloride and THF was evaporated under reduced pressure. The residue was diluted with ether (100 ml) and washed with water (3 × 100 ml), dried (MgSO₄) and evaporated under reduced pressure.

N-(2'-Bromophenyl)-N-methyl-3-tert-butyldimethylsilyl-

propynamide, 9a. Amide **8a** (6 g, 25.2 mmol) and *tert*butyldimethylsilyl chloride (4.6 g, 30.3 mmol) gave **9a** (8.7 g, 98%) as a white solid, mp 49–51 °C (Found: C, 54.49; H, 6.20; N, 3.92%. C₁₆H₂₂BrNOSi requires C, 54.54; H, 6.29; N, 3.98%); $R_{\rm f}(30\%$ EtOAc–hexane) 0.41; $v_{\rm max}$ /cm⁻¹ 2196 (C=C) 1650 (C=O) 1250 (Si–Me₂); $\delta_{\rm H}(360$ MHz; CDCl₃) 7.53 (1H, dd, *J* 8.0, 1.4, H-3'), 7.29 (2H, m, H-5', H-6'), 7.24 (1H, m, H-4'), 3.25 (3H, s, NMe), 0.67 (9H, s, Si–'Bu), -0.054 (3H, s, SiMe), -0.085 (3H, s, SiMe); $\delta_{\rm C}$ (CDCl₃) 153.8 (C=O), 142.0 (C-1'), 133.6 (C-3'), 130.5 (C-5'), 130.1 (C-6'), 128.6 (C-4'), 123.9 (C-2'), 96.9 (HC=C), 96.4 (HC=C), 35.2 (Si–'Bu), 26.8 (N-Me), 16.2 (*C*-Me₃), -5.4 (Si-Me₂); *m*/*z* 353 (0.5%, ⁸¹Br, M⁺), 351 (0.1, ⁷⁹Br, M⁺), 272 (100, M⁺ – Br) (Found: M⁺ (⁷⁹Br) 351.0677. C₁₆H₂₂⁷⁹BrNOSi requires *M* 351.0654).

N-(2'-Bromophenyl)-N-prop-2-enyl-3-tert-butyldimethyl-

silylpropynamide, 9b. Amide 8c (0.73 g, 2.65 mmol) and *tert*butyldimethylsilyl chloride (0.5 g, 3.18 mmol) gave 9b (1.0 g, 99%) as a brown oil; $R_f(30\%$ EtOAc–hexane) 0.59; v_{max}/cm^{-1} 1655 (s, tertiary amide C=O), 1252 (s, Si–C); $\delta_H(360$ MHz; CDCl₃) 7.65 (1H, dd, J 7.9, 1.3, H-3'), 7.34 (1H, td, J 7.2, 1.4, H-5'), 7.24 (2H, m, H-4' + H-6'), 5.85 (1H, m, H-2''), 5.11 (1H, dd, J 6.9, 1.0, *trans* H-3'') 5.09 (1H, dd, J 14.5, 1.3, *cis* H-3'), 4.75 (1H, ddt, J 14.5, 5.7, 1.5, H-1''), 3.82 (1H, ddt, J 14.5, 7.5, 1.0, H-1''), 0.67 (9H, s, SiC(CH₃)₃), -0.05 (3H, s, SiCH₃), -0.09 (3H, s, SiCH₃); δ_C (CDCl₃) 153.4 (C=O), 140.3 (C-1'), 133.6 (C-3'), 131.8 (C-5' + C-2''), 130.0 (C-6'), 129.7 (C-4'), 124.5 (C-2'), 119.3 (C-3''), 97.1 (C-3), 96.4 (C-2), 50.4 (C-1''), 25.8 (C(CH₃)₃), 16.2 (C(CH₃)₃), -5.38 (2 × CH₃); *m/z* 379 (0.2%, M⁺, ⁸¹Br), 377 (0.2, M⁺, ⁷⁹Br), 298 (10.1, M⁺ - Br), 84 (100) (Found: M⁺ (⁷⁹Br) 377.0825. C₁₈H₂₄⁷⁹BrNOSi requires *M* 377.0811).

N-(2'-Bromophenyl)-*N*-(1,3-dioxolan-2-ylmethyl)-3-*tert*-butyldimethylsilylpropynamide, 14. Amide 9b (3.5 g, 9 mmol) was dissolved in a mixture of dichloromethane and methanol (5:1, 200 ml) and cooled to -78 °C. A steady stream of ozone was passed through the solution until a blue colour persisted and then for a further 60 minutes. Dimethyl sulfide (1.35 ml, 18 mmol) was added and the solution stirred at room temperature for 16 hours. The solvents were evaporated *in vacuo* and then rapidly chromatographed on silica gel (1:3 ethyl actetate–hexane) to give the aldehyde (2.1 g, 62%) as an unstable oil which was immediately reacted with ethane-1,2-diol (0.36 ml, 6.5 mmol), toluene-p-sulfonic acid (1 crystal) and toluene (50 ml) in a Dean-Stark apparatus until no more water was collected. The toluene was removed in vacuo and the residue chromatographed on silica gel (1:5 ethyl acetatehexane) to give the title compound, 14 as a colourless oil (1.7 g, 85%); $R_{\rm f}$ (30% EtOAc-hexane) 0.4; $v_{\rm max}$ /cm⁻¹ 2927 (m, C-H), 1652 (s, C=O); δ_H(360 MHz; CDCl₃) 7.64 (1H, dd, J 8, 1.5, H-3), 7.48 (1H, dd, J 8, 1.5, H-6), 7.35 (1H, td, J 8, 1.5, H-4), 7.21 (1H, td, J 8, 1.5, H-5), 5.18 (1H, dd, J 5.3, 4.4, H-2"), 4.38 (1H, dd, J 14.2, 4.4, H-1"), 3.90 (4H, m, OCH₂CH₂O), 3.25 (1H, dd, J 14.2, 5.3, H-1"), 0.67 (9H, s, C(CH₃)₃), -0.07 (3H, s, Si-CH₃), -0.09 (3H, s, Si-CH₃); $\delta_{\rm C}$ (CDCl₃) 154.0 (C=O), 141.0 (C-1), 133.3 (C-3), 132.2 (C-6), 130.0 (C-5), 128.1 (C-4), 124.1 (C-2), 101.0 (C-2"), 97.0 (C-2'), 96.8 (C-3'), 64.9, 64.8 (OCH₂CH₂O), 49.8 (C-1"), 25.7 (C-(CH₃)₃), 16.2 (C-(CH₃)₃), -5.4 (Si-Me₂); m/z 425 (29%, M⁺, ⁸¹Br), 423 (29%, M⁺, ⁷⁹Br), 167 (35), 73 (100) (Found: M⁺ (⁷⁹Br) 423.0886. C₁₉H₂₆⁷⁹BrNO₃Si requires M 423.0865).

Radical cyclisations: general procedure

The amide in toluene was heated at 80 °C under argon. AIBN (0.1 equivalents) and tributyltin hydride were then added simultaneously, dropwise. AIBN (0.1 equivalents) was then added every hour for 3 hours. The reaction was then allowed to stir overnight at 80 °C. Toluene was removed in vacuo. The residue was dissolved in ethyl acetate and washed with 20% ammonia solution (5 \times 100 ml). The organic layer was dried (MgSO₄) and the solvent removed in vacuo.

N-Methyl-3-tert-butyldimethylsilylmethylideneindol-2(3H)one, 10a. Amide 9a (0.2 g, 0.57 mmol) and tributyltin hydride (0.17 ml, 0.63 mmol) gave 10a as a mixture of E- and Z-isomers which were separable by chromatography (SiO₂, 2% EtOAchexane).

Z-Isomer. A green solid (71 mg, 46%), mp 84-86 °C; R_f(30% EtOAc-hexane) 0.56; ν_{max}/cm^{-1} 2925 (m, C-H), 1703 (s, C=O), 1611 (s, C=C), 1469 (s, benzene ring); $\delta_{\rm H}$ (360 MHz; CDCl₃) 7.43 (1H, dd, J7.5, 0.6, H-4), 7.27 (1H, td, J7.5, 1.1, H-6), 7.03 (1H, s, H-1'), 7.01 (1H, td, J 7.5, 0.8, H-5), 6.76 (1H, d, J 7.5, H-7), 3.21 (3H, s, NMe), 1.00 (9H, s, C(CH₃)₃), 0.31 (6H, s, 2(CH₃)); $\delta_{\rm C}({\rm CDCl}_3)$ 167.1 (C=O), 143.5 (C-7a), 142.3 (C-3a), 138.0 (C-4), 129.8 (C-6), 123.2 (C-3), 121.9 (C-5), 120.0 (C-1'), 107.9 (C-7), 26.5 (C-(CH₃)₃), 25.8 (N-CH₃), 17.1 (C-(CH₃)₃), -6.2 (Si-Me₂); m/z 273 (12.2%, M⁺), 258 (52.3, M⁺ – CH₃), 216 (100, $M^+ - C(CH_3)_3$), 201 (54.4), 186 (70.4), 158 (35) (Found: M⁺, 273.1541. C₁₆H₂₃NOSi requires *M* 273.1549).

E-Isomer. A green solid (25 mg, 16%), $R_f(30\%$ EtOAchexane) 0.50; v_{max}/cm^{-1} 2929 (C-H), 1708 (s, C=O), 1611 (s, C=C), 1470 (s, benzene ring), 1265 (s, Si-C); $\delta_{\rm H}$ (360 MHz; CDCl₃) 7.57 (1H, d, J 7.6, H-4), 7.30 (1H, s, H-1'), 7.29 (1H, t, J 7.6, H-6), 7.02 (1H, t, J 7.6, H-5), 6.81 (1H, d, J 7.6, H-7), 3.24 (3H, s, NMe), 1.01 (9H, s, C(CH₃)₃), 0.32 (6H, s, 2(CH₃)); $\delta_{\rm C}({\rm CDCl}_3)$ 167.1 (C=O), 145.0 (C-7a), 141.8 (C-3a), 137.5 (C-4), 129.9 (C-6), 123.7 (C-3), 122.6 (C-1'), 121.8 (C-5), 108.5 (C-7), 29.7 (N-Me), 26.9 (C-(CH₃)₃), 17.4 (C-(CH₃)₃), -5.4 (Si-Me₂); m/z 273 (35.6%, M⁺), 258 (13.9, M⁺ - CH₃), 216 (85.6, $M^+ - C(CH_3)_3$), 201 (7.8), 186 (41.5), 158 (4.4), 28.0 (100) (Found: M⁺, 273.1544. C₁₆H₂₃NOSi requires M 273.1549).

N-(Prop-2-enyl)-3-tert-butyldimethylsilylmethylideneindole-2-(3H)-one, 10b. Amide 9b (0.9 g, 2.38 mmol) and tributyltin hydride (0.75 g, 2.61 mmol) gave 10b as a mixture of E- and Z-isomers which were separable by chromatography (SiO₂, 3%EtOAc-hexane).

Z-Isomer. A green oil (0.35 g, 49%), Rf(30% EtOAc-hexane) 0.56; v_{max}/cm⁻¹ 2927 (m, C–H), 1705 (s, C=O), 1610 (s, C=C), 1470 (s, benzene ring); $\delta_{\rm H}$ (360 MHz; CDCl₃) 7.45 (1H, dd, J 7.5, 0.7, H-4), 7.24 (1H, td, J 7.5, 1.2, H-6), 7.06 (1H, s, H-1'), 7.01 (1H, td, J 7.5, 1, H-5), 6.77 (1H, d, J 7.5, H-7), 5.85 (1H, m, H-2"), 5.21 (2H, m, H-3"), 4.36 (2H, dt, J 5.3, 1.7, H-1"), 1.00 (9H, s, SiC(CH₃)₃), 0.31 (6H, s, Si(CH₃)₂); δ_C(CDCl₃) 166.7 (C=O), 142.9 (C-7a), 142.2 (C-3a), 138.3 (C-4), 131.8 (C-2"), 129.7 (C-6), 123.4 (C-3), 121.9 (C-5), 120.1 (C-1'), 117.3 (C-1"), 108.8 (C-7), 42.1 (2C, CH₂, C-3"), 26.5 (SiC(CH₃)₃), 17.2 (SiC(CH₃)₃), -6.1 (Si(CH₃)₂); m/z 299 (7.4%, M⁺), 242 (100, $M^+ - C_4H_9$), 186 (75%) (Found: M^+ , 299.1695. $C_{18}H_{25}NOSi$ requires M 299.1705).

E-Isomer. A green oil (74 mg, 10%), $R_f(30\%$ EtOAc–hexane) 0.49; v_{max}/cm⁻¹ 2927 (m, C-H), 1703 (s, C=O) 1610 (s, C=C), 1468 (s, benzene ring); $\delta_{\rm H}$ (360 MHz; CDCl₃) 7.58 (1H, d, J 7.6, H-4), 7.31 (1H, s, H-1'), 7.25 (1H, td, J 7.6, 1.1, H-6), 7.01 (1H, td, J 7.6, 1, H-5), 6.81 (1H, d, J 7.6, H-7), 5.84 (1H, m, H-2"), 5.23 (2H, m, H-3"), 4.39 (2H, dt, J 5.2, 1.6, H-1"), 1.02 (9H, s, SiC(CH₃)₃), 0.33 (6H, s, Si(CH₃)₂); δ_C(CDCl₃) 166.0 (C=O), 148.0 (C-7a), 144 (C-3a), 137.8 (C-4), 131.6 (C-2"), 129.8 (C-6), 123.8 (C-1'), 122.5 (C-3), 121.8 (C-5), 117.5 (C-1"), 109.1 (C-7), 42.4 (2C, CH₂, C-3"), 26.3 (SiC(CH₃)₃), 18.0 $(SiC(CH_3)_3)$, -5.4 $(Si(CH_3)_2)$; m/z 299 (9%, M⁺), 242 (100, $M^+ - C_4H_9$) (Found: M^+ , 299.1697. $C_{18}H_{25}NOSi$ requires M 299.1705).

N-(1,3-Dioxolan-2-ylmethyl)-3-tert-butyldimethylsilylmethylideneindol-2(3H)-one, 15. Amide 14 (1.7 g, 3.99 mmol) and tributyltin hydride (1.28 g, 4.38 mmol) gave 15 as a mixture of E- and Z-isomers separable by chromatography (SiO₂, 3% EtOAc-hexane).

Z-Isomer. A green oil (0.62 g, 45%), $R_f(30\%$ EtOAc-hexane) 0.67; v_{max}/cm⁻¹ 3054 (olefinic C-H), 2926 (C-H), 1652 (tertiary amide C=O), 1456 (benzene ring), 1265 (C-O); $\delta_{\rm H}$ (360 MHz; CDCl₃), 7.43 (1H, d, J 7.6, H-4), 7.26 (1H, td, J 7.6, 1.0, H-6), 7.04 (1H, s, H-1'), 7.02 (1H, d, J 7.6, H-5), 6.99 (1H, t, J 7.6, H-7), 5.15 (1H, t, J 4.1, H-2"), 3.97 (2H, m, H-4" or H-5"), 3.91 (2H, d, J 4.1, H-1"), 3.87 (2H, m, H-4" or H-5"), 0.99 (9H, s, Si(CH₃)₃), 0.30 (6H, s, $2 \times$ Si(CH₃)); δ_{C} (CDCl₃) 138.3 (C-4), 129.8 (C-6), 127.2 (C-3), 121.9 (C-5), 120.0 (C-1'), 109.3 (C-7), 101.9 (C-2"), 65.2 (CH₂), 42.9 (C-1"), 26.5 (SiC(CH₃)₃), 17.2 (SiC(CH₃)₃), -6.2 (Si(CH₃)₂); *m*/*z* 346 (33%, M⁺+1), 345 (100, M^+), 288 (65, $M^+ - C_4 H_9$) (Found: M^+ 345.1780. $C_{19}H_{27}NO_3Si$ requires M 345.1760).

E-Isomer. A green oil (0.3 g, 22%), R_f(30% EtOAc-hexane) 0.54; v_{max}/cm⁻¹ 3054 (olefinic C–H), 2926 (C–H), 1652 (tertiary amide C=O), 1456 (benzene ring), 1265 (C-O); $\delta_{\rm H}$ (360 MHz; CDCl₃), 7.56 (1H, d, J 7.3, H-4), 7.30 (1H, s, H-1'), 6.99-7.13 (3H, m, H-5, H-6, H-7), 5.13-5.17 (1H, m, H-2"), 3.97 (2H, m, H-4" or H-5"), 3.95 (2H, d, J 4.1, H-1"), 3.87 (1H, m, H-4" or H-5"), 1.01 (9H, s, Si(CH₃)₃), 0.32 (6H, s, 2 × Si(CH₃)); $\delta_{\rm C}({\rm CDCl}_3)$ 166.8 (C=O), 144.4 (C-7a), 141.5 (C-3a), 137.7 (C-4), 129.8 (C-6), 128.0 (C-3), 123.6 (C-1'), 121.8 (C-5), 109.4 (C-7), 101.7 (C-2"), 65.1 (CH₂), 42.9 (C-1"), 26.4 (SiC(CH₃)₃), 17.3 $(SiC(CH_3)_3)$, -5.4 $(Si(CH_3)_2)$; m/z 345 (17.6%, M⁺), 288 (100, $M^+ - C_4H_9$) (Found: M^+ 345.1783. $C_{19}H_{27}NO_3Si$ requires M 345.1760).

N-Methyl-3'-(tert-butyldimethylsilyl)spiro[indole-3,2'-

oxiran]-2-one, 16a. MCPBA (0.14 g, 50% w/v, 0.4 mmol) was added portionwise to a solution of Z-10a (0.1 g, 0.37 mmol) in dichloromethane (5 ml) at room temperature. The reaction mixture was stirred for 16 hours after which time a further quantity of MCPBA (0.14 g) was added. The reaction mixture was stirred for a further 2 hours and then sodium sulfite solution (10 ml, 10% w/v) was added to quench the reaction. The organic layer was washed with saturated sodium bicarbonate solution $(3 \times 5 \text{ ml})$, water $(3 \times 5 \text{ ml})$ and finally brine $(3 \times 5 \text{ ml})$. The organic layer was dried and evaporated under reduced pressure. Chromatography (SiO₂, 15% EtOAchexane) gave the *title compound*, **16a** as a pale yellow oil (0.045 g, 43%); $R_{\rm f}$ (3:7 EtOAc-hexane) 0.52; $v_{\rm max}$ /cm⁻¹ 2928 (C-H),

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1709 (C=O), 1253 (epoxide C–O); $\delta_{\rm H}(360 \text{ MHz}; \text{CDCl}_3)$, 7.59 (1H, dd, J 7.5, 1.2, H-4), 7.29 (1H, td, J 7.5, 1.2, H-6), 7.09 (1H, t, J 7.5, H-5), 6.86 (1H, d, J 7.5, H-7), 3.27 (3H, s, NCH₃), 3.02 (1H, s, OCHSi), 0.65 (9H, s, Si(CH₃)₃), 0.41 (3H, s, SiCH₃), 0.18 (3H, s, SiCH₃); $\delta_{\rm C}(\text{CDCl}_3)$ 172.31 (C=O), 141.55 (C-7a), 127.44 (C-4), 124.98 (C-6), 124.42 (C-3a), 122.38 (C-5), 107.9 (C-7), 66.74 (C-3), 61.1 (C-1'), 26.72 (SiC(CH₃)₃), 26.3 (NCH₃) 19.74 (SiC(CH₃)₃), -6.4 (SiCH₃), -6.6 (SiCH₃); *m*/z 289 (38.6%, M⁺), 232 (100, M⁺ – C₄H₉) (Found: M⁺ 289.1500. C₁₆H₂₃NO₂Si requires *M* 289.1498).

N-Prop-2-enyl-3'-(tert-butyldimethylsilyl)spiro[indole-3,2'-

oxiran]-2-one, 16b. Procedure as for 16a. Z-10b (0.3 g, 1 mmol) gave the title compound, 16b as a pale yellow oil (0.16 g, 51%) after chromatography (SiO₂, 15% EtOAc–hexane); $R_f(3:7)$ EtOAc-hexane) 0.49; ν_{max}/cm^{-1} 1703 (C=O), 1610 (C=C), 1250 (epoxide C–O); δ_H(360 MHz; CDCl₃), 7.31 (1H, dd, J 7.7, 1.6, H-4), 7.11 (1H, td, J 7.7, 1.6, H-6), 7.05 (1H, t, J 7.7, H-5), 6.88 (1H, d, J 7.7, H-7), 5.84 (1H, m, CH=CH₂), 5.25 (1H, dd, J 11.8, 1, cis-C=CH₂), 5.22 (1H, dd, J 5.0, 1, trans-C=CH₂), 4.45 (1H, ddt, J 16.3, 5.2, 1, N-CH₂), 4.29 (1H, ddt, J 16.3, 5.3, 1, N-CH₂), 3.00 (1H, s, OCHSi), 0.99 (9H, s, Si(CH₃)₃), 0.36 (3H, s, SiCH₃), 0.08 (3H, s, SiCH₃); δ_C(CDCl₃) 172.5 (C=O), 143.9 (C-7a), 131.1 (C-4), 129.8 (C-6), 125.1 (C-3a), 122.7 (C-5), 121.7 (C-7), 117.7 (C=CH₂), 109.4 (C=CH₂), 60.8 (O-C-Si), 60.3 (C-3), 42.6 (N-CH₂), 26.4 (SiC(CH₃)₃), 17.0 (SiC(CH₃)₃), -7.05 (SiCH₃), -7.1 (SiCH₃); m/z 200 (30%, M^+ – TBDMS), 73 (100). No M^+ was observed and no HRMS could be determined.

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