

# Ring Closing Metathesis Mediated Synthesis of 4a-Aryloxadecahydroisoquinolines, Intermediates in the Preparation of Novel Opiates

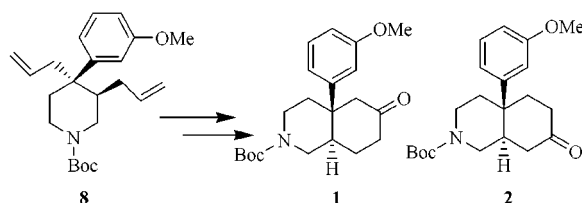
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## ABSTRACT

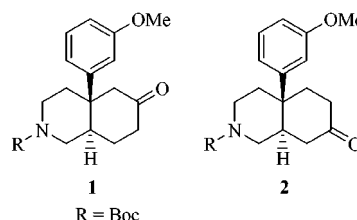


The concise syntheses of the 4a-aryldecahydroisoquinolines **1** and **2** through a uniform strategy starting from *N*-methyl-3-allyl-4-piperidinone are reported in this Letter. Key transformations include a ring closing metathesis reaction to prepare a *trans*-octahydroisoquinoline common intermediate and a regiocontrolled hydroboration–oxidation sequence.

The 4a-aryldecahydroisoquinolines **1** and **2** (Scheme 1) represent structural fragments of morphine with significant pharmacological activity. The compounds exhibit potent affinity for the  $\mu$  opioid receptor and possess antinociceptive properties consistent with those of  $\mu$  receptor agonists.<sup>3</sup> The pharmaceutical importance of these compounds is further enhanced by their utility as advanced intermediates for the synthesis of novel selective ligands for the  $\delta$  opioid receptor.<sup>4</sup> Synthetic efforts toward the 4a-aryldecahydroisoquinoline **1** have been previously reported.<sup>5</sup> In contrast, the synthesis of

compound **2** has been the subject of fewer reports.<sup>6</sup> As part of a program that aimed at the design and synthesis of novel  $\delta$  opioid receptor ligands, we sought to develop concise syntheses for substrates **1** and **2**. In this Letter we report a uniform approach to the targeted 4a-aryldecahydroisoquinolines through common intermediates highlighted by a ring closing metathesis reaction and a regiochemically controlled hydroboration–oxidation step.

Scheme 1



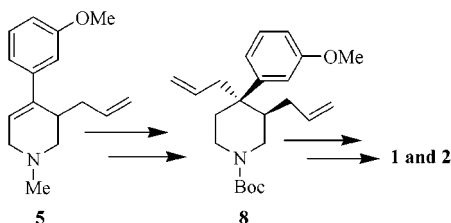
(1) CNS medicinal chemistry.

(2) Exploratory Medicinal Sciences, computational chemistry.

(3) For a review of the opioid ligands, see: Casy, A. F.; Parfitt, R. T. *Opioid Analgesics: Chemistry and Receptors*; Plenum Press: New York, 1986.

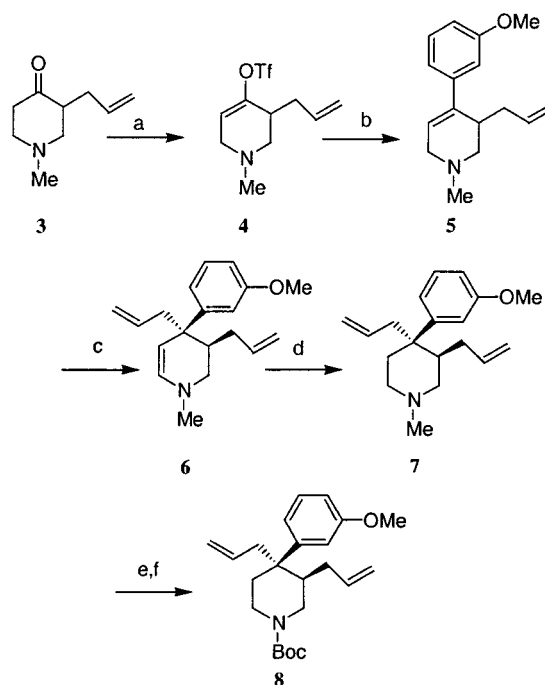
(4) For recent examples, see: (a) Dondio, G.; Ronzoni, S.; Eggleston, D. S.; Artico, M.; Petrillo, P.; Petrone, G.; Visentin, L.; Farina, C.; Vecchietti, V.; Clarke, G. D. *J. Med. Chem.* **1997**, *40*, 3192. (b) Nagase, H.; Kawai, K.; Hayakawa, J.; Wakita, H.; Mizusuna, A.; Matsuura, H.; Tajima, C.; Takezawa, Y.; Endoh, T. *Chem. Pharm. Bull.* **1998**, *46*(11), 1695.

Scheme 2



For the syntheses of both substrates, we envisioned an approach as described in Scheme 2. The diallyl piperidine **8**, precursor to the ring closing metathesis reaction, would be accessed following an allylation reaction of the metalloenamine generated from compound **5**. A ring closing metathesis of compound **8** would furnish a *trans* octahydroisoquinoline with a quaternary carbon center, which after a regioselective hydroboration–oxidation sequence would be converted to the target compounds **1** and **2**. The regiochemical preference of the hydroboration–oxidation step could be predicted on the basis of molecular orbital calculations.

The reduction of this plan to practice commenced with the multigram preparation of 3-allyl-*N*-methyl-4-piperidinone **3** according to literature procedures.<sup>7</sup> Related piperidinones have been converted to the 3-alkyl-4-aryltetrahydropyridines similar to **5** via aryl Grignard or aryllithium additions followed by regioselective dehydration.<sup>8</sup> As an alternative to this strategy, we envisioned the preparation of compound **5** from a precursor vinyl triflate utilizing organopalladium chemistry (Scheme 3). Thus, the piperidinone **3** was transformed with exclusive regiocontrol to the vinyl triflate **4** in 92% yield with LiHMDS and *N*-phenyl triflamide<sup>9</sup> in THF at  $-78^{\circ}\text{C}$ . The vinyl triflate was converted to the desired tetrahydropyridine **5** after Suzuki coupling with the commercially available 3-methoxyphenylboronic acid. The yields for the Suzuki coupling ranged from 48% to 94%. Optimum reaction conditions for this step were discovered when the vinyl triflate and boronic acid were treated with KBr,  $\text{K}_3\text{PO}_4$ , and  $\text{Pd}(\text{PPh}_3)_4$  as catalyst in dioxane solvent at  $85^{\circ}\text{C}$  to afford compound **5** in 94% yield.<sup>10</sup> With tetrahydropyridine **5** available, our attention was turned to the formation of diallylpiperidine **7**. The preparation of diallylpiperidine **7** was achieved via a metalloenamine generation–alkylation sequence. This sequence was first reported in the context of

Scheme 3<sup>a</sup>

<sup>a</sup> (a) LiHMDS, *N*-phenyltrifluoromethanesulfonimide, THF  $-78^{\circ}\text{C}$  to rt, 92%; (b) 3-methoxyphenylboronic acid, KBr,  $\text{K}_3\text{PO}_4$ ,  $\text{Pd}(\text{PPh}_3)_4$ , 1,4-dioxane,  $85^{\circ}\text{C}$ , 94%; (c) *sec*-BuLi, allyl bromide, THF,  $-45^{\circ}\text{C}$  to  $-78^{\circ}\text{C}$  to rt; (d)  $\text{NaBH}_4$ , MeOH, rt, 88% for steps c and d; (e) ACE-Cl, 1,2-dichloroethane, reflux, then methanol, reflux; (f) di-*tert*-butyl dicarbonate,  $\text{Et}_3\text{N}$ , dichloromethane, rt, 80% for steps e and f.

the synthesis of related morphinoid alkaloids.<sup>11</sup> In the event, treatment of compound **5** with *sec*-BuLi in THF at  $-45^{\circ}\text{C}$  followed by addition of allyl bromide produced enamine **6** in quantitative yield. The crude enamine was reduced with  $\text{NaBH}_4$  in methanol to afford piperidine **7** in 88% yield for the two steps. The *N*-methylpiperidine was subsequently converted to the *N*-Boc equivalent to ultimately produce a compound with an easily removable nitrogen-protecting group, suitable for analogue formation. Thus, after treatment of **7** with 1-chloroethyl chloroformate in dichloroethane at reflux, subsequent carbamate methanolysis and treatment of the crude hydrochloride salt with di-*tert*-butyl dicarbonate and triethylamine compound **8** was obtained in 80% overall yield.

With the synthesis of diallylpiperidines **7** and **8** completed, our attention was turned to the ring closing metathesis transformation (Scheme 4). The diallylpiperidine **8** was converted to the cyclic olefin **11** after treatment with 0.1 equiv of the Grubbs catalyst **9** in dichloroethane at  $60^{\circ}\text{C}$  in 93% yield. Similarly, the hydrochloride salt of compound **7** was converted to the corresponding olefin **10** in 88% yield.<sup>12</sup>

The regioselective functionalization of the olefin in compound **11** was predicted on the basis of molecular orbital interactions. Restricted Hartree–Fock calculations were carried out for the transition states corresponding to

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(6) See ref 5c.

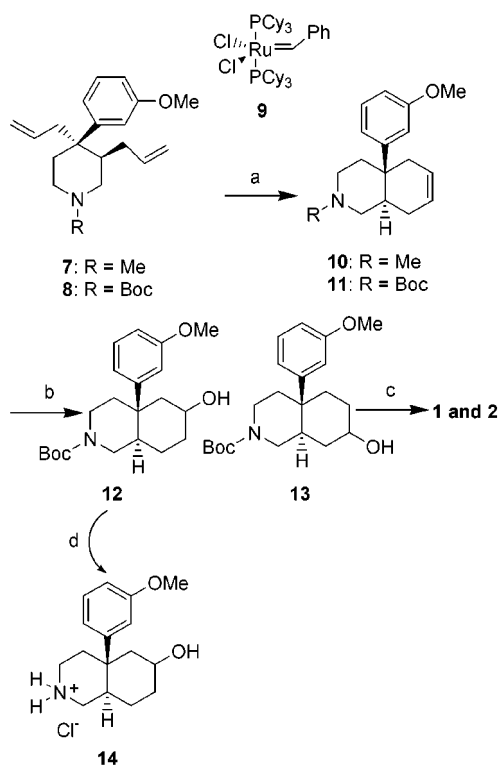
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Scheme 4<sup>a</sup>

<sup>a</sup> (a) Grubbs catalyst **9**, 1,2-dichloroethane, 60 °C, 93% for compound **11**, 88% for compound **10**; (b) 9-BBN, THF, reflux, 30% aqueous H<sub>2</sub>O<sub>2</sub>, EtOH, 6 N NaOH, quantitative yield for **12** and **13** (3:1 ratio); (c) TPAP, NMO, dichloromethane, rt, 88% for **1** and 84% for **2**; (d) HCl, EtOAc, rt, 85%.

the hydroboration of **11** with both 9-BBN and borane. The geometries for this series of molecules were fully optimized by means of analytical energy gradients<sup>13</sup> with the 6-31G-(d)-basis set<sup>14</sup> in the gas phase. The ab initio molecular orbital calculations were carried out with the Gaussian 94 series of programs on a Silicon Graphics computer.<sup>15</sup> Comparison of the transition state energies for the reaction of **11** with 9-BBN shows a preference for product **12** over **13** of 0.616 kcal/mol, corresponding to a ratio of 2.83:1. The reaction of **11** with borane is predicted to favor the regioisomer, **13**, by 0.767 kcal/mol (predicted ratio 1:3.65). Our findings are consistent with prior ab initio calculations that implicate

(12) For recent reviews regarding applications of the ring closing metathesis reaction, see: (a) Grubbs, R. H.; Miller, S. J.; Fu, G. C. *Acc. Chem. Res.* **1995**, 28, 446. (b) Pandit, U. K.; Overkleeft, H. S.; Borer, B. C.; Bieraugel, H. *Eur. J. Org. Chem.* **1999**, 5, 959. (c) Wright, D. L. *Curr. Org. Chem.* **1999**, 3, 211. (d) Phillips, A. J.; Abell, A. D. *Aldrichimica Acta* **1999**, 32, 75. (e) Maier, M. E. *Angew. Chem., Int. Ed.* **2000**, 39, 2073.

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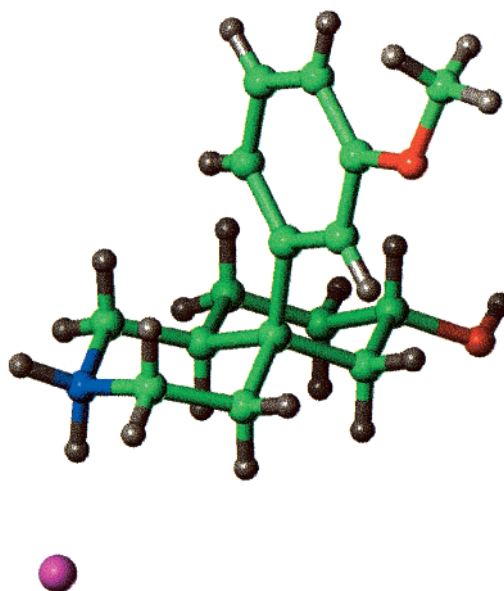


Figure 1.

both steric and electronic factors in the regioselectivity.<sup>16</sup> Experimental results supported the computational chemistry predictions. Thus, treatment of olefin **11** with a sterically hindered borane such as 9-BBN in refluxing THF, followed by oxidation of the organoborane species, yielded in quantitative yield a 3:1 ratio of the separable, isomeric alcohols **12** and **13**. Conversely, treatment of **11** with the smaller borane—dimethyl sulfide reagent complex yielded a 3:1 ratio of alcohols **13** and **12** in a not optimized 52% yield. The final step for the preparation of isoquinolines **1** and **2** was the oxidation of the secondary hydroxyl groups. Typically, Swern oxidation reactions had been employed for this transformation in good yields. In this instance, oxidation of alcohols **12** and **13** with TPAP and *N*-morpholine oxide afforded ketones **1** and **2** in 88% and 84% yields, respectively.

The stereochemistry of the major product **12** was confirmed from a crystal structure obtained from its hydrochloride salt **14** (Figure 1). Compound **14** was produced after removal of the nitrogen-protecting group with HCl in ethyl acetate at room temperature in 85% yield.

In conclusion, the efficient syntheses of the 4a-aryldecahydroisoquinolines **1** and **2** through a uniform strategy were reported in this Letter. The desired compounds exhibit significant pharmacological activity at opioid receptors and provide useful intermediates for the synthesis of novel and selective  $\delta$  opioid receptor ligands. Key synthetic steps in this approach include a highly efficient ring closing metathesis reaction to form a *trans* octahydroisoquinoline substrate and a regioselective hydroboration—oxidation reaction sequence.

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**Acknowledgment.** The authors thank Dr. Jon Bordner, small molecule crystallography, Pfizer Global Research and Development, Groton Laboratories, for the X-ray structure of compound **14** and Ms. Sharon Mellow for preparation of the document.

**Supporting Information Available:** Representative experimental procedures for the preparation of compounds **4**, **5**, **7**, **8**, **11**, **12**, **13**, **1**, and **2**. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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