## LETTERS 2013 Vol. 15, No. 3 554–557

ORGANIC

## Concise Asymmetric Synthesis of Orthogonally Protected *syn*- and *anti*-1,3-Aminoalcohols

## Jae Seung Lee, Dongeun Kim, Lucia Lozano, Suk Bin Kong, and Hyunsoo Han\*

Department of Chemistry, University of Texas at San Antonio, San Antonio, Texas 78249, United States

Hyunsoo.han@utsa.edu

## Received December 10, 2012



Novel chiral binfunctional reagents V and *ent*-V undergo asymmetric aldehyde allylation followed by Ir(I)-catalyzed enantioselective allylic amidation to give orthogonally protected *syn*- and *anti*-1,3-aminoalcohols with complete control of absolute and relative stereochemistry. The Mitsunobu reaction of the initial homoallylic alcohol products followed by Ir(I)-catalyzed enantioselective allylic amidation provides orthogonally protected *syn*- and *anti*-1,3-diamine derivatives in high yields and with excellent stereoselectivities.

The 1,3-amino alcohol motifs are common in many natural products and biologically active compounds.<sup>1</sup>

10.1021/ol303371u © 2013 American Chemical Society Published on Web 01/14/2013

1,3-Aminoalcohols have also been used as chiral auxiliaries and ligands for asymmetric catalysis.<sup>2</sup> A vast majority of methods for their synthesis to date rely on diastereoselective reduction/addition of optically enriched  $\beta$ -hydroxy imine and  $\beta$ -amino ketone derivatives,<sup>3</sup> as well as diastereoselective C–H amination.<sup>4</sup> To our knowledge, only a few enantioselective methods are currently available, which include rhodium-catalyzed asymmetric hydrogenation of  $\beta$ -ketoenamides,<sup>5</sup> a proline-catalyzed enantioselective Mannich reaction followed by tandem hydrogenation– enzymatic dynamic kinetic resolution,<sup>6</sup> and a strategy employing proline-catalyzed  $\alpha$ -aminoxylation and  $\alpha$ -amination of aldehydes for the asymmetric introduction of C–O/N

 <sup>(</sup>a) Shibahara, S.; Kondo, S.; Maeda, K.; Umezawa, H.; Ohno, M. J. Am. Chem. Soc. 1972, 94, 4353–4354. (b) Wang, Y.-F.; Izawa, T.; Kobayashi, S.; Ohno, M. J. Am. Chem. Soc. 1982, 104, 6465–6466. (c) Knapp, S. Chem. Rev. 1995, 95, 1859–1876. (d) Kempf, D. J.; Marsh, K. C.; Denissen, J. F.; McDonald, E.; Vasavanonda, S.; Flentge, C. A.; Green, B. E.; Fino, L.; Park, C. H.; Kong, X.-P.; Wideburg, N. E.; Saldivar, A.; Kati, W. M.; Sham, H. L.; Tobins, T.; Stewart, K. D.; Hsu, A.; Plattner, J. J.; Leonard, J. M.; Norbeck, D. W. Proc. Natl. Acad. Sci. U.S.A. 1995, 92, 2484–2488. (e) Sham, H. L.; Zhao, C.; Li, L.; Betebenner, D. A.; Saldivar, A.; Vasavanonda, S.; Kempf, D. J.; Plattner, J. J.; Norbeck, D. W. Bioorg. Med. Chem. Lett. 2002, 12, 3101–3103. (f) Carlier, P. R.; Lo, M. M.-C.; Lo, P. C.-K.; Richelson, E.; Tatsumi, M.; Reynolds, I. J.; Sharma, T. A. Bioorg. Med. Chem. Lett. 1998, 8, 487–492. (g) Shi, Z.; Harrison, B. A.; Verdine, G. L. Org. Lett. 2003, 5, 633–636. (h) Raju, B.; Mortell, K.; Anandan, S.; O'Dowd, H.; Gao, H.; Gomez, M.; Hackbarth, C.; Wu, C.; Wang, W.; Yuan, Z.; White, R.; Trias, J.; Patel, D. V. Bioorg. Med. Chem. Lett. 2003, 13, 2413–2418. (i) Benedetti, F.; Norbedo, S. Chem. Commun. 2001, 203– 204.

<sup>(2) (</sup>a) Lait, S. M.; Rankic, D. A.; Keay, B. A. Chem. Rev. 2007, 107, 767–796.

<sup>(3) (</sup>a) Kochi, T.; Tang, T. P.; Ellman, J. A. J. Am. Chem. Soc. 2002, 124, 6518–6591. (b) Kochi, T.; Tang, T. P.; Ellman, J. A. J. Am. Chem. Soc. 2003, 125, 11276–11282. (c) Williams, D. R.; Osterhout, M. H. J. Am. Chem. Soc. 1992, 114, 8750–8751. (d) Keck, G. E.; Truong, A. P. Org. Lett. 2002, 4, 3131–3134. (e) Menche, D.; Arikan, F.; Li, J.; Rudolph, S. Org. Lett. 2007, 9, 267–270. (f) Liu, Z.-J.; Mei, Y.-Q.; Liu, J.-T. Tetrahedron 2007, 63, 855–860. (g) Edupuganti, R.; Davis, F. A. Org. Biomol. Chem. 2012, 10, 5021–5031. (h) Kennedy, A.; Nelson, A.; Perry, A. Synlett 2004, 967–970. (i) Yamamoto, Y.; Komatsu, T.; Maruyama, K. J. Chem. Soc., Chem. Commun. 1985, 814–816.

<sup>(4) (</sup>a) Rice, G. T.; White, M. C. J. Am. Chem. Soc. 2009, 131, 11707–11711. (b) Milczek, E.; Boudet, N.; Blakey, S. Angew. Chem., Int. Ed. 2008, 47, 6825–6828. (c) Nahra, F.; Liron, F.; Prestat, G.; Mealli, C.; Messaoudi, A.; Poli, G. Chem.—Eur. J. 2009, 15, 11078–11082. (d) Zalatan, D. N.; Du Bois, J. J. Am. Chem. Soc. 2008, 130, 9220–9221. (e) Harvey, M. E.; Nusaev, D. G.; Du Bois, J. J. Am. Chem. Soc. 2011, 133, 17207–17216. (f) Paradine, S. M.; White, M. C. J. Am. Chem. Soc. 2012, 134, 2036–2039.

<sup>(5)</sup> Millet, R.; Träff, A. M.; Petrus, M. L.; Bäckvall, J. E. J. Am. Chem. Soc. 2010, 132, 15182–15184.

<sup>(6)</sup> Geng, H.; Zhang, W.; Chen, J.; Hou, G.; Zhou, L.; Zou, Y.; Wu, W.; Zhang, X. Angew. Chem., Int. Ed. **2009**, 48, 6052–6054.

<sup>(7)</sup> Jha, V.; Kondekar, N. B.; Kumar, P. Org. Lett. 2010, 12, 2762-2765.

bonds.<sup>7</sup> Although these methods can deliver 1,3-aminoalcohols in a stereoselective fashion, they require multiple steps and/or suffer from a limited substrate scope.<sup>8</sup> Furthermore, rhodium-catalyzed asymmetric hydrogenation of  $\beta$ -ketoenamides gives rise to only *anti*-1,3aminoalcohols. Remarkably, very efficient enantioselective methods to give orthogonally protected 1,3-aminoalcohols with complete control of their relative and absolute stereochemistry remain undeveloped. Herein we report a conceptually distinct method for the very efficient asymmetric synthesis of orthogonally protected *syn*- and *anti*-1,3-aminoalcohols, which utilizes newly developed air-stable chiral binfunctional allylation reagents V and *ent*-V (Scheme 1).

Scheme 1. Design Concept of Chiral Bifunctional Reagent V for the Asymmetric Synthesis of Orthogonally Protected *syn-* and *anti-*1,3-Aminoalcohols



Recently the Nokami group reported that, in the presence of an acid, enantioenriched tertiary homoallylic alcohol I could react with aldehydes through [3,3]-sigmatropic rearrangement to deliver homoallylic alcohols II with near-perfect chirality transfer and high E/Z-selectivity

(eq 1 in Scheme 1).<sup>9,10</sup> We and others recently disclosed Ir(I)-catalyzed enantioselective allylic amidation between ethyl allyl carbonates III and diacylamine nucleophiles to give protected allylic amines IV (eq 2 in Scheme 1).<sup>11,12</sup> A logical extension of these two observations would be to combine Nokami's allyl transfer reaction with the Ir(I)catalyzed allylic amidation reaction to create a new chiral bifunctional reagent V. As shown in Scheme 1, asymmetric allylation of aldehydes by V is expected to give allyl carbonates VI. Ir(I)-catalyzed allylic amidation of VI with diacylamine nucleophiles will initially generate the intermediates VII, which could undergo an intramolecular 1,5-acyl transfer reaction under the appropriate allylic amidation conditions to finally give orthogonally protected 1,3-aminoalcohols VIII and IX. If the stereochemistry of the amidation step is governed by the chiral ligands L\* or ent-L\* used, the presented two-step strategy can provide a most direct method for the asymmetric synthesis of orthogonally protected 1,3-aminoalcohols from the readily available aldehydes with complete control of their relative and absolute stereochemistry.

Scheme 2. Asymmetric Synthesis of Enantiomerically Pure Bifunctional Allyl Transfer Reagent V



Enantiomerically pure binfunctional reagents V and *ent*-V were conveniently prepared from commercial 3-methylbuten-2-en-1-ol in three steps (Scheme 2). Sharpless asymmetric epoxidation (AE) of 3-methylbuten-2-en-1-ol (1) by (+)-diethyl tartrate (DET) gave rise to the corresponding 2-epoxy alcohol, which was isolated as its *p*-nitrobenzoate ester 2.<sup>9</sup> Enantiomerically pure 2 (*ee* > 99%) was obtained by washing crude 2 with ethyl ether. Reaction of 2 with vinylMgCl in the presence of CuBr·SMe<sub>2</sub> at -20 °C furnished diol 3,<sup>9</sup> which upon treatment with ethyl chloroformate and pyridine transformed into V. The same reaction sequence using (–)-DET in the Sharpless AE step was used to prepare *ent*-V. Binfunctional reagents V and *ent*-V are stable at ambient temperature and can be stored without any special precautions.

<sup>(8)</sup> Reference 5 works only for the terminal methyl group and gives only an (*R*)-configuration at the oxygen-substituted carbon due to the substrate specificity of the enzyme CALB used; ref 6 requires preparation of  $\beta$ -ketoenamides from the corresponding ketones in two steps; ref 7 requires five to six steps to prepare 1,3-aminoalcohol derivatives.

<sup>(9)</sup> Shafi, S. M.; Chou, J.; Kataoka, K.; Nokami, J. Org. Lett. 2005, 7, 2957–2960.

<sup>(10)</sup> For other similar reports, see: (a) Nokami, J.; Yoshizane, K.; Matsuura, H.; Sumida, S.-I. J. Am. Chem. Soc. 1998, 120, 6609–6610.
(b) Malkov, A. V.; Kabeshov, M. A.; Barlog, M.; Kočovský, P. Chem.— Eur. J. 2009, 15, 1570–1573. (c) Hayashi, S.; Hirano, K.; Yorimitsu, H.; Oshima, K. Org. Lett. 2005, 7, 3577–3579. (d) Nokami, J.; Nomiyama, K.; Shafi, S. M.; Kataoka, K. Org. Lett. 2004, 6, 1261–1264. (e) Lee, C.-L. K.; Lee, C.-H. A.; Tan, K.-T.; Loh, T.-P. Org. Lett. 2004, 6, 1281– 1283. (f) Cheng, H.-S.; Loh, T.-P. J. Am. Chem. Soc. 2003, 125, 4990–4991.
(g) Nokami, J.; Nomiyama, K.; Matsuda, S.; Imai, N.; Kataoka, K. Angew. Chem., Intl. Ed. 2003, 42, 1273–1276. (h) Loh, T.-P.; Lee, C.-L. K.; Tan, K.-T. Org. Lett. 2002, 4, 2985–2987. (i) Nokami, J.; Ohga, M.; Nakamoto, H.; Matsubara, T.; Hussain, I.; Kataoka, K. J. Am. Chem. Soc. 2001, 123, 9168–9169. (j) Loh, T.-P.; Tan, K.-T.; Hu, Q.-Y. Angew. Chem., Int. Ed. 2001, 40, 2921–2922. (k) Nokami, J.; Anthony, L.; Sumida, S.-I. Chem.— Eur. J. 2000, 6, 2909–2913. (l) Sumida, S.-I.; Ohga, M.; Mitani, J.; Nokami, J. J. Am. Chem. Soc. 2000, 122, 1310–1313. (m) Loh, T.-P.; Hu, Q.-Y.; Ma, L.-T. Org. Lett. 2002, 4, 2389–2391.

<sup>(11) (</sup>a) Weihofen, R.; Tverskoy, O.; Helmchen, G. Angew. Chem., Int. Ed. 2006, 45, 5546–5549. (b) Pouy, M. J.; Leitner, A.; Weix, D. J.; Ueno, S.; Hartwig, J. F. Org. Lett. 2007, 9, 3949–3952. (c) Weix, D. J.; Markovic, D.; Ueda, M.; Hartwig, J. F. Org. Lett. 2009, 11, 2944–2947.
(d) Singh, O. V.; Han, H. J. Am. Chem. Soc. 2007, 129, 774–775.
(e) Singh, O. V.; Han, H. Org. Lett. 2007, 9, 4801–4804. (f) Singh, O. V.; Han, H. Tetrahedron Lett. 2007, 48, 7094–7098.

<sup>(12)</sup> For recent reviews on Ir(I)-catalyzed allylation: (a) Hartwig, J. F.; Pouy, M. J. *Top. Organomet. Chem.* **2011**, *34*, 169–208. (b) Helmchen, G.; Dahnz, A.; Dübon, P.; Schelwies, M.; Weihofen, R. *Chem. Commun.* **2007**, 675–691.

 Table 1. Asymmetric Allyl Transfer Reaction between

 Aldehydes 4 and Chiral Bifunctional Reagent V

0 .		TfOH (0.2 equiv)	QH	5 O
R <sup>∕</sup> H <sup>′</sup> ≷ 4	V O OET	CH <sub>2</sub> Cl <sub>2</sub> , -78 °C	R	>∕O <sup>//</sup> OE
entry	R-	product	yield [%] <sup>a</sup>	ee [%] <sup>b</sup>
1 🦯	~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	(S)- <b>5a</b>	90	99
2	- Jon	( <i>R</i> )- <b>5</b> b	78	98
3	- vr	( <i>R</i> )- <b>5c</b>	82	99
4	- r	(S)- <b>5d</b>	75	99
5	- r	( <i>R</i> )- <b>5e</b>	74	99
6	X	( <i>R</i> )-5f	76	98
7	BnO ~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	( <i>R</i> )- <b>5</b> g	86	99
<sup>a</sup> Isolated	yields. <sup>b</sup> Determ	ined by chiral HI	PLC.	

Acid-mediated asymmetric allylation of n-octanal by V was used to determine the optimal conditions. After screening various acids (TfOH, TMSOTf, trifluoromethanesulfonimide, 2,4-dinitrobenzenesulfonic acid, (1S)-(+)camphorsulfonic acid, p-TsOH·H2O, TiCl4, Sc(OTf)3, and  $BF_3 \cdot OEt_2$ ) and temperatures (-78, -50, -20, and 0 °C) in methylene chloride, the reaction conditions involving 0.2 equiv of TfOH at -78 °C were determined to be optimal to give the corresponding homoallylic ether 5a in 90% reaction yield and 99% ee. A selection of aldehydes were allowed to react with I under the determined reaction conditions, and the results are shown in Table 1. The asymmetric allyl transfer reaction tolerated wide structural variations in aldehydes involving linear alkyl (entry 1),  $\beta$ -branched alkyl (entry 2),  $\alpha$ -branched alkyl (entries 3 and 4), cycloalkyl (entry 5), tert-butyl (entry 6), and functionalized linear alkyl (entry 7) groups. The corresponding homoallylic alcohols 5 were obtained in good yields up to 90%, and chirality transfer was nearly perfect to give  $\geq 98\%$  ee's. In all cases studied, only E-5 formed as judged by NMR analysis; the E-geometry is necessary for Ir(I)-catalyzed enantioselective allylic amidation.

To study Ir(I)-catalyzed diastereoselective allylic amidation, **5a** was subjected to the catalytic conditions involving  $[Ir(COD)Cl]_2$  (2 mol %), L\* (4 mol %), DBU (20 mol %), and benzyl *tert*-butyl imidodicarboxylate as a nucleophile in THF.<sup>11f,13</sup> However, the desired allylic amidation reaction did not occur even under heating conditions with higher ligand loadings. Based on the premise that [Ir(dbcot)Cl]<sub>2</sub> could generate more robust catalysts than  $[Ir(COD)Cl]_{2}^{14}$  2 mol %  $[Ir(dbcot)Cl]_{2}$  was used at rt. The ethyl allyl carbonate 5a did not react at rt, but upon raising the temperature to 65 °C, the reaction proceeded cleanly to give O-Cbz, N-Boc protected 1,3-aminoalcohol **6a** in 85% yield and with > 20:1 diastereoselectivity. On the other hand, under the same conditions, ent-5a provided the corresponding diastereomer 7a in 88% vield and with >20:1 diastereoselectivity (Table 2, entries 1) and 2). Similarly, other structurally diverse 5 and ent-5 compounds (which were prepared by using ent-V and aldehydes 4 as in Table 1) provided the corresponding O-Cbz, N-Boc protected 1,3-aminoalcohols 6 and 7, respectively, in good yields and with excellent diastereoselectivities. These results indicate that the stereochemistry of the allylic amidation is predominantly controlled by the stereochemistry of the ligand/catalyst used, and 1,5-acyl transfer takes place under the amidation conditions.

 
 Table 2. Ir(I)-Catalyzed Diastereoselective Allylic Amidation of Homoallylic Alcohols 5 and *ent*-5



<sup>*a*</sup> Isolated yields. <sup>*b*</sup> Determined by the <sup>1</sup>H NMR spectrum of reaction mixtures.

To further demonstrate the synthetic utility of homoallylic alcohols **5** and *ent*-**5**, we chose to transform

<sup>(13) (</sup>Cbz)<sub>2</sub>NH,  $(Boc)_2$ NH, and phthalimide were tried, but the desired allylic amidation did not occur. AcNHBoc gave rise to a mixture of amidation product and 1,5-acyl transfer product.

<sup>(14)</sup> Spiess, S.; Welter, C.; Frank, G.; Taquet, J.-P.; Helmchen, G. Angew. Chem., Int. Ed. 2008, 47, 7652–7655.

ent-5b into the corresponding 1,3-diamine derivatives. 1,3-Diamine functionalities are a key structural element in numerous natural products and pharmacologically active compounds,<sup>15</sup> as well as chiral ligands for asymmetric catalysis.<sup>16</sup> Despite recent synthetic advancements, methods for the asymmetric synthesis of 1.3-diamines via enantioselective processes have been severely underdeveloped.<sup>17</sup> Scheme 3 describes the two-step asymmetric synthesis of orthogonally protected 1.3-diamines from ent-5b. The Mitsunobu reaction between homoallylic alcohol ent-5b (entry 4, Table 2) and phthalimide furnished the corresponding homoallylic amine,<sup>18,18b</sup> which upon Ir(I)catalyzed allylic amidation in the presence of L\* transformed into orthogonally protected 1,3-diamine 9 in 76% overall yield and with > 20:1 dr as judged by NMR analysis. The same reaction sequence using ent-L\* for the allylic amidation delivered the corresponding diastereomer 10 in 71% overall yield and with > 20:1 dr. Similarly, homoallylic alcohol ent-5b was converted into diastereomeric azides 11 and **12** by using diphenyl phosphoryl azide (DPPA)<sup>18c,d</sup> in the Mitsunobu reaction in 87% and 84% overall yields, respectively, and with > 20:1 dr.

In conclusion, novel chiral bifunctional reagents V and *ent*-V, which are air-stable and can be used in a stepeconomical fashion, have been developed. The reagents undergo asymmetric aldehyde allylation followed by Ir(I)catalyzed allylic amidation to deliver orthogonally protected *syn*- and *anti*-1,3-aminoalcohols in good yields and

(16) (a) Busscher, G. F.; Retjes, F. P. J. T.; van Delft, F. L. Chem. Rev. 2005, 105, 775. (b) Kizirian, J.-C. Chem. Rev. 2008, 108, 140-205. (17) (a) Morgen, M.; Bretzke, S.; Li, P.; Menche, D. Org. Lett. 2010, 12, 4494–4497. (b) Trost, B. M.; Malhotra, S.; Olson, D. E.; Maruniak, A.; Du Bois, J. J. Am. Chem. Soc. 2009, 131, 4190-4191. (c) Matsubara, R.; Nakamura, Y.; Kobayashi, S. Angew. Chem., Int. Ed. 2004, 43, 1679-1681. (d) Terada, M.; Machioka, K.; Sorimachi, K. Angew. Chem., Int. Ed. 2006, 45, 2254-2257. (e) Terada, M.; Machioka, K.; Sorimachi, K. Angew. Chem., Int. Ed. 2009, 48, 2553-2556. (f) Lu, S.-F.; Du, D.-M.; Xu, J.; Zhang, S.-W. J. Am. Chem. Soc. 2006, 128, 7418-7419. (g) Giampietro, N. C.; Wolfe, J. P. J. Am. Chem. Soc. 2008, 120, 1410-1419. (g) 12911. (h) Martjuga, M.; Shabashov, D.; Belakov, S.; Liepinsh, E.; Suna, E. J. Org. Chem. 2010, 75, 2357-2368. (i) Martjuga, M.; Belakov, S.; Liepinsh, E.; Suna, E. J. Org. Chem. 2011, 76, 2635–2647. (j) Weatherly, C. D.; Rigoli, J. W.; Schomaker, J. M. Org. Lett. 2012, 14, 1704–1707. (k) Vesely, J.; Ibrahem, I.; Rios, R.; Zhao, G.; Xu, Y.; Cordova, A. *Tetrahedron Lett.* **2007**, *48*, 2193–2198. (l) Lanter, J. C.; Chen, H.; Zhang, X.; Sui, Z. Org. Lett. 2005, 7, 5905-5907. (m) Kurokawa, T.; Kim, M.; DuBois, J. Angew. Chem., Int. Ed. 2009, 48, 2777-2779. (n) Wu, J.; Zhu, K.-C.; Yuan, P.-W.; Panek, J. S. Org. Lett. 2012, 14, 3624-3627.

Scheme 3. Asymmetric Synthesis of 1,3-Diamine Derivatives



with excellent stereoselectivities. The synthetic utility of V was further demonstrated in the three-step asymmetric synthesis of orthogonally protected 1,3-diamine derivatives. Considering the importance and prevalence of 1,3-aminoalcohol and 1,3-diamine motifs in numerous (bio)chemically active molecules, the developed strategies employing V and *ent*-V should find wide synthetic applications.

Acknowledgment. Support by the National Science Foundation (CHE 0911134) is greatly acknowledged.

**Supporting Information Available.** Experimental procedures and spectroscopic data for all new compounds. This material is available free of charge via the Internet at http://pubs.acs.org.

<sup>(15) (</sup>a) Dergeron, R. J.; Feng, Y.; Weimar, W. R.; McManis, J. S.; Dimova, H.; Porter, C.; Raisler, B.; Phanstiel, O. J. Med. Chem. 1997, 40, 1475–1494. (b) Arya, D. P.; Xue, L.; Willis, B. J. Am. Chem. Soc. 2003, 125, 10148–10149. (c) Kurosawa, W.; Kan, T.; Fukuyama, T. J. Am. Chem. Soc. 2003, 125, 8112–8113. (d) Welch, K. T.; Virga, K. G.; Whittemore, N. A.; Ozen, C.; Wright, E.; Brown, C. L.; Lee, R. E.; Serpersu, E. H. Bioorg. Med. Chem. 2005, 12, 6252–6263. (e) When, P. M.; Du Bois, J. J. Am. Chem. Soc. 2002, 124, 12950. (f) Franklin, A. S.; Ly, S. K.; Mackin, G. H.; Overman, L. E.; Shaka, A. J. J. Org. Chem. 1999, 64, 1512–1519.

<sup>(18)</sup> For a review, see: (a) Mitsunobu, O. Synthesis **1981**, 1–28. For the use of phthalimide in the Mitsunobu reaction, see: (b) Sen, S. E.; Roach, S. L. Synthesis **1995**, 756–758. For the use of dppa in the Mitsunobu reaction, see: (c) Cossy, J.; Willis, C.; Bellosta, V.; BouzBouz, S. J. Org. Chem. **2002**, 67, 1982–1992. (d) Chen, Q.; Huo, X.; Zheng, H.; She, X. Synlett **2012**, 23, 1349–1352.

The authors declare no competing financial interest.