Note

Subscriber access provided by University of South Dakota

Iridium-Catalyzed Hydrosilylation of Unactivated Alkenes: Scope, and Application to Late-Stage Functionalization

Xingze Xie, Xueyan Zhang, Haoyu Yang, Xin Ji, Jianing Li, and Shengtao Ding

J. Org. Chem., Just Accepted Manuscript • DOI: 10.1021/acs.joc.8b02838 • Publication Date (Web): 18 Dec 2018

Downloaded from http://pubs.acs.org on December 19, 2018

Just Accepted

"Just Accepted" manuscripts have been peer-reviewed and accepted for publication. They are posted online prior to technical editing, formatting for publication and author proofing. The American Chemical Society provides "Just Accepted" as a service to the research community to expedite the dissemination of scientific material as soon as possible after acceptance. "Just Accepted" manuscripts appear in full in PDF format accompanied by an HTML abstract. "Just Accepted" manuscripts have been fully peer reviewed, but should not be considered the official version of record. They are citable by the Digital Object Identifier (DOI®). "Just Accepted" is an optional service offered to authors. Therefore, the "Just Accepted" Web site may not include all articles that will be published in the journal. After a manuscript is technically edited and formatted, it will be removed from the "Just Accepted" Web site and published as an ASAP article. Note that technical editing may introduce minor changes to the manuscript text and/or graphics which could affect content, and all legal disclaimers and ethical guidelines that apply to the journal pertain. ACS cannot be held responsible for errors or consequences arising from the use of information contained in these "Just Accepted" manuscripts.



is published by the American Chemical Society. 1155 Sixteenth Street N.W., Washington, DC 20036

Published by American Chemical Society. Copyright © American Chemical Society. However, no copyright claim is made to original U.S. Government works, or works produced by employees of any Commonwealth realm Crown government in the course of their duties.

Iridium-Catalyzed Hydrosilylation of Unactivated Alkenes: Scope, and Application to Late-Stage Functionalization

Xingze Xie,[†] Xueyan Zhang,[†] Haoyu Yang,[†] Xin Ji,[†] Jianing Li,[†] Shengtao Ding^{*,†}

[†]State Key Laboratory of Organic-Inorganic Composites, College of Chemical Engineering, Beijing University of Chemical Technology, Beijing, 100029, China

KEYWORDS: Iridium, hydrosilylation, alkene, late-stage, catalysis

$R^1X \stackrel{f}{\underset{R^2}{\overset{h}{\underset{R^2}{\underset{R^2}{\overset{h}{R^2}{\overset{h}{\underset{R^2}{\overset{h}{\underset{R^2}{\overset{h}{\underset{R^2}{\underset{R^2}{\underset{R^2}{\underset{R^2}{\overset{h}{\underset{R^2}}{\underset{R^2}{\underset{R^2}{\underset{R^2}{\underset{R^2}{\underset{R^2}}{\underset{R^2}{\underset{R^2}{\underset{R^2}{\underset{R^2}{\underset{R^2}}{\underset{R^2}{\underset{R^2}{\underset{R^2}}{\underset{R^2}{\underset{R^2}{\underset{R^2}{\underset{R^2}{R^2}{\underset{R^2}{R^2}{\underset{R^2}{R}}{\underset{R^2}{\underset{R^2}{R}}{R^2}{R^2}{R}{R}}{R}}{R}}}}}}}}}}$	Si-H	R ¹ X h Si R ²	X = S, O, N, Br, B, Si, C, P(=O), SO ₂ , ester n = 0, 1, 2, R ¹ , R ² = H, alkyl, aryl, etc.	
54 examples, up to 99% yield Excellent <i>anti-</i> Markovnikov regioselectivity Late-stage hydrosilylation				

ABSTRACT: Highly efficient and general Ir-catalyzed hydrosilylation of unactivated alkenes with excellent *anti*-Markovnikov regioselectivity was described. A broad scope of hydrosilylated products were synthesized economically and conveniently from commercially or naturally available compounds, which provides versatile valuable precursors for organic and medicinal studies.

Organosilicon compounds are intriguing reagents in organic transformations because of their great advantages in stability, solubility, non-toxicity, easy-handling, accessibility, and selectivity (Figure 1, **a**).¹ Moreover, benefiting from the unique atomic properties of silicon, silicon analogues play important roles in drug discovery (Figure 1, **b**).² Alkene hydrosilylation is one major approach widely used in industry to provide commodity silicones after the Direct Process.³ However, application of this atom-economic strategy in fine chemical synthesis, such as the above-mentioned areas, is much less common (Scheme 1, **a**). Mild and efficient methods for construction of organosilicons with broad substrate scope and high selectivity are still in urgent desire.

Figure 1. Selected examples of organosilicon compounds' applications in organic transformations and pharmaceuticals.

(a) Transformation of aliphatic silyl groups (Ref. 1e & 1f)





Scheme 1. Hydrosilylation of alkenes.



Iridium complexes are powerful tools in silvlation of a wide variety of functional groups.⁴ Surprisingly, despite usually recognized as one main kind of precious metal catalysts in alkene hydrosilylation, reports on exploitation of iridium complexes for this process are rare. Till now, most of related works were published as patents, which mainly focused on simple terminal alkenes.^{5,6} Inspired by this research vacancy, a few works were released recently.7-10 Shimada reported the iridium-catalyzed hydrosilylation of allyl acetates8 and sulfur-containing olefins10. Kühn gave a mechanistic study on the iridium-catalyzed hydrosilylation of allyl compounds.9 Recently, one of us, together with Sun and Wu, developed iridium-catalyzed hydrosilylation of internal thioalkynes with excellent regio- and stereoselectivity.¹¹ Generation of intermediate Ir(I) hydride A (in Scheme 1, **b**) is presumptively crucial in this process, which we envisioned is probably one key factor in other iridiumcatalyzed hydrosilylation processes as well.^{10, 11} Meanwhile, existence of heteroatoms or functional groups that can coordinate to iridium center might facilitate these catalytic conversions. Herein, we demonstrate our research on the Ir-catalyzed hydrosilylation of unactivated alkenes with

ACS Paragon Plus Environment

59

60

excellent *anti*-Markovnikov regioselectivity and high efficiency, which provides complementary examples and additional insight to previous reports.^{5-10, 12} This catalytic system is suitable for a broad substrate scope, especially for those containing heteroatoms. More importantly, it works effectively on late-stage hydrosilylation, which is rarely reported before.

1

2

3

4

5

6

7

8

9

10

11

12

13

14

15

16

17

18

19

20

21

22

23 24

25

26

27

28

29 30

31

32

33

34

35

36

37

38

39

40

41

42

43

44

45

46

47

48

49

50

51

52

53

54

55

56

57 58 59

60

We initiated this study with phenyl allyl sulfide **1a** as the model substrate and triethoxysilane **2a** as the silylation reagent (Table 1). Several iridium complexes were investigated by using deuterated chloroform as the solvent, among which [Ir(COD)Cl]₂ was identified as the most efficient catalyst, affording linear product **3a** in 82% NMR yield with excellent regioselectivity (Entry 1-4). Chlorinated solvents, such as dichloromethane and dichloroethane, gave better results in this process (Entry 5-13). A slightly decreased yield of 82% was observed when the loading amount of the catalyst was reduced to 0.5 mol % (Entry 14). Furthermore, this reaction could be carried out in neat status as most industrial productions do, though providing a moderate yield (Entry 15).

Ph^SSi(OEt)₃

Table 1. Optimization of Reaction Conditions.

s .	Cat.	3a		
Ph 12	+ (EtO) ₃ SIH	→ Si ^{r.t.} ,S. 人	(OEt) ₃	
Ia	20	Ph ⁻ 3a'	Me	
Entry	Cat.	Solvent	Yield ^b	3a/3a' ^b
1	[Ir(COD)Cl] ₂	CDCl ₃	82	>99:1
2	[Ir(COD)(OMe)] ₂	CDCl ₃	48	>99:1
3	$[Ir(COE)_2Cl]_2$	CDCl ₃	17	
4	[Cp*IrCl ₂] ₂	CDCl ₃	<5	
5	[Ir(COD)Cl] ₂	DCM	93	>99:1
6	[Ir(COD)Cl] ₂	DCE	89	>99:1
7	[Ir(COD)Cl] ₂	THF	55	
8	[Ir(COD)Cl] ₂	MeCN	47	
9	[Ir(COD)Cl] ₂	DMF	10	
10	[Ir(COD)Cl] ₂	1,4-Dioxane	52	
11	[Ir(COD)Cl] ₂	Toluene	36	
12	[Ir(COD)Cl] ₂	EtOH	45	
13	[Ir(COD)Cl] ₂	H₂O	n.d.	
14 ^c	[Ir(COD)Cl] ₂	DCM	82	>99:1
15	[Ir(COD)Cl] ₂	neat	61	>99:1

^aReaction conditions: The mixture of **1a** (0.10 mmol, 1.0 equiv.), **2a** (0.15 mmol, 1.5 equiv.) and catalyst (0.002 mmol, 2 mol %) in solvent (0.50 mL) was stirred for 6 h under room temperature. ^bYield and ratio were determined by ¹H NMR spectroscopy using dimethyl sulfone as internal standard. ^c0.5 mol % of catalyst was used.

Table 2. Scope of alkenyl sulfides and silanes.



^aReaction conditions: The mixture of **1** (0.50 mmol, 1.0 equiv.), **2** (0.75 mmol, 1.5 equiv.) and catalyst (0.01 mmol, 2 mol %) in solvent (2.0 mL) was stirred for 6 h under room temperature. ^bIsolated yield. ^c3.0 Equivalent of silane was used. ^d4.5 Equivalent of silane was used. ^eRatio of allylic sulfide and silane is 3:1.

With the optimized condition in hand, a variety of alkenes containing sulfur atom were tested to react with different silanes (Table 2). Good to excellent yields were observed in the reactions of aryl allyl sulfides bearing diverse functional groups (3a to 3d). As Shimada has reported one similar work mainly covering simple aryl and aliphatic sulfurcontaining olefins,¹⁰ we turned our attention to more complexed substrates. N-Heterocycles are common structures in drugs. To probe the efficiency of this catalytic system, several substrates containing different types of N-heterocycles, including pyridine, pyrimidine, triazine, tetrazole, and benzo[*d*]thiazole, were investigated (**3e** to **3j**). Related products were afforded in moderate to high yields with excellent regioselectivity in all cases. This system is also applicable to various alkyl allyl sulfides (3k to 30). Both homoallylic and vinylic sulfides performed well under the mild condition (**3p & 3q**). Next, we evaluated the silane scope in this process, and found that several familiar silanes, including (MeO)₃SiH, (EtO)₂MeSiH, PhMe₂SiH, and BnMe₂SiH, could react with the allylic sulfides smoothly, providing the desired products 3r-3u with high efficiency and selectivity. Only a trace amount of product 3v was detected while Et₃SiH was used. The success of using disilane as reaction partner permits the potential application of this new developed catalytic system in linear polymer and porous organic polymer synthesis (3w).

We then turned our attention to oxygen-containing olefins (Table 3). This catalytic system was proved to be effective

2

3

4

5

6

7

8

9

10

11

12

13

14

15

16

17

18

19

20

21

22

23

24

25

26

27

28

29

30

31

32

33

34

35

36

37

38

39

40

41

42

43

44

45

46

47

48

49

50

51

52

53

54

55

56

57 58 59

60

in hydrosilylation of several selected representative aryl/alkyl allyl ethers, including bis(allyl) ethers derived from hydroquinone and BINOL (**5a** to **5f**). Vinyl ether could be hydrosilylated efficiently as well (**5g**). Encouraged by these positive results, more efforts were put on exploring latestage hydrosilylation of nature molecule analogues. Inspiringly, allylic ether derivatives from proline, menthol, glucose, and glucofuranose were tolerated well, giving corresponding *anti*-Markovnikov addition products in good to excellent yields (**5h** to **5k**). Allyl ether analogues of estrone, testosterone, and vitamin E were also successfully modified by using this hydrosilylation protocol (**5l** to **5n**).

Table 3. Scope of olefins containing oxygen atom.



^aReaction conditions: The mixture of **4** (0.50 mmol, 1.0 equiv.), **2** (0.75 mmol, 1.5 equiv.) and catalyst (0.01 mmol, 2 mol %) in solvent (2.0 mL) was stirred for 6 h under room temperature. ^bIsolated yield. ^c3.0 Equivalent of silane was used.

Having examined alkenes containing chalcogens, we took a further step to check the capacity of the Ir-catalyzed hydrosilylation processes for functional groups or other elements (Table 4). Allylic sulfone and ester were compatible with this method (7a & 7b). The reactions of allylic tertiary and secondary amines provided desired hydrosilylated product (7c & 7d). Allyl phosphine showed poor reactivity, which might be due to the coordination of phosphine atom with iridium which obstructs the formation of active catalyst intermediate A, as the reaction of allylic phosphonate with (EtO)₃SiH proceeded smoothly under this condition (7e). The investigation of this protocol for allyl bromide and allyl Bpin gave satisfied results with excellent yields (7f & 7g). Sterically bulky allylsilane and vinylsilane were converted into related products in good yields (7h & 7i). As hydrosilylation of alkyl/aryl olefins has been addressed

with numerous catalysts, we just listed several examples here (7j to 7l).





^aReaction conditions: The mixture of **6** (0.50 mmol, 1.0 equiv.), **2** (0.75 mmol, 1.5 equiv.) and catalyst (0.01 mmol, 2 mol %) in solvent (2.0 mL) was stirred for 6 h under room temperature. ^bIsolated yield. ^c3.0 Equivalent of silane was used.

Sterically hindered disubstituted terminal alkenes were investigated as well (Scheme 2). No desired product was observed from **8a** or **8b** (Eq. 1). **8c** could be hydrosilylated to afford **3x** in 74% yield, while isomerization of **8d** to **9** was the major phenomenon under the same condition (Eq. 2). Notably, both of the terminal alkenyl groups in myrcene (**8e**) were hydrosilylated (**7m**, Eq. 3).

Scheme 2. Hydrosilylation of disubstituted terminal alkenes.



Based on previous reports and our observation,^{10, 11} a plausible mechanism is given in Scheme 3. Heteroatoms or functional groups in the substrates is presumed to coordinate to iridium center and could help to stabilize possible intermediates **C** and **D**, which would facilitate the catalytic cycle. This coordination effect might be more critical towards disubstituted terminal alkenes. More detailed mechanism studies are underway.

Scheme 3. Proposed mechanism.



In summary, we demonstrated the efficiency and compatibility of one simple, mild iridium catalytic system in hydrosilylation of unactivated alkenes. A wide scope of *anti*-Markovnikov hydrosilylated products with excellent regioselectivity were afforded conveniently from commercially or naturally available compounds, providing versatile valuable blocks for organic and medicinal studies.

EXPERIMENTAL SECTION

1

2

3

4

5

6

7

8

9

10

11

12

13

14

15

16

17

18

19

20

21

22

23

24

25

26

27

28

29

30

31

32

33

34

35

36

37

38

39

40

41

42

43

44

45

46

47

48

49

50

51

52

53

54

55

56

57

58 59

60

General Information. All air or moisture sensitive reactions were conducted in oven-dried glassware under nitrogen atmosphere using dry solvents. Flash column chromatography was performed over silica gel (200-300 mesh) purchased from Qingdao Puke Co., China. Silanes and common organic chemicals were purchased from commercial suppliers, such as Sigma-Aldrich® and J&K® Scientific Ltd., and used as received. Iridium complexes were purchased from Strem® Chemicals, Inc. ¹H and ¹³C NMR spectra were collected on a Bruker AV 400 MHz NMR spectrometer using residue solvent peaks as an internal standard (¹H NMR: CDCl₃ at 7.26 ppm, ¹³C NMR: CDCl₃ at 77.0 ppm). Mass spectra were collected on an Thermo Scientific GC/MS ISQ7000 system, or a Xevo G2 Qtof mass spectrometer.

Safety Note. Triethoxysilane is a corrosive and flammable liquid. Methyldiethoxysilane is advised to be used for large-scale (> 1.0 g) reactions instead.

General Procedure for Hydrosilylation. In a glove box, to an oven-dried 5-mL vial was added the alkene (0.50 mmol), the silane (0.75 mmol), [Ir (COD)CI]₂ (0.008 mmol), and DCM (2.0 mL). The vial was capped and removed from the glove box. The reaction mixture was stirred at room temperature for 4 h, and then concentrated under reduced pressure. The residue was purified by silica gel flash column chromatography (eluent: 0-50% EtOAc in petroleum ether) to give the desired product. Compounds 3a,¹⁰ 3e,¹³ 3i,¹⁴ 3j,¹³ 3k,¹⁵ 3l,¹⁶ 30,¹⁷ 3g,¹⁸ 3r,¹⁰ 3s,¹⁰ 5a,¹⁹ 5c,²⁰ 5e,²¹ 5f,²² 5g,²³ 5i,²⁴ 5n,²⁵ 7a,²⁶ 7b,²⁷ 7d,²⁸ 7f,²⁹ 7g,³⁰ 7h,²⁸ 7j,³¹ 7k,²⁸ 7l³¹ are known compounds. Result of MS was given for known compounds.

Triethoxy(3-(*phenylthio*)*propyl*)*silane* (**3***a*) was prepared as pale yellow oil from **1a** (0.50 mmol, 75.1 mg) and (EtO)₃SiH (0.75 mmol, 125.8 mg), according to the General Procedure in 93% yield (146.2 mg, *l/b* > 99:1): ¹H NMR (400 MHz, CDCl₃) δ 7.37-7.33 (m, 2 H), 7.31-7.25 (m, 2 H), 7.20-7.14 (m, 1 H), 3.82 (q, *J* = 8.0 Hz, 6 H), 2.98 (t, *J* = 7.2 Hz, 2 H), 1.84-1.75 (m, 2 H), 1.23 (t, *J* = 8.0 Hz, 9 H), 0.81 (t, *J* = 8.0 Hz, 2 H); ¹³C {¹H} NMR (100 MHz, CDCl₃) δ 136.7, 129.0, 128.7, 125.6, 58.3, 36.4, 22.7, 18.2, 9.7; MS (ESI) *m/z* (relative intensity) 314.1 (5), 163.1 (100) [M]⁺. (3-((4-Chlorophenyl)thio)propyl)triethoxysilane (**3b**) was prepared as pale yellow oil from **1b** (0.50 mmol, 92.4 mg) and (EtO)₃SiH (0.75 mmol, 125.8 mg), according to the General Procedure in 86% yield (150.1 mg, l/b > 99:1): 'H NMR (400 MHz, CDCl₃) δ 7.26-7.20 (m, 4 H), 3.80 (q, *J* = 6.8 Hz, 6 H), 2.92 (t, *J* = 7.2 Hz, 2 H), 1.79-1.70 (m, 2 H), 1.20 (t, *J* = 6.8 Hz, 9 H), 0.79-0.74 (m, 2 H); ¹³C {'H} NMR (100 MHz, CDCl₃) δ 135.3, 131.5, 130.3, 128.8, 58.3, 36.5, 22.6, 18.2, 9.7; HRMS *m*/z (Cl) calcd. for C₁₅H₂₅ClO₃SSiNa (M+Na)⁺ 371.0880, found 371.0898.

Triethoxy(*3*-(*p*-tolylthio)*propy*])*silane* (*3***c**) was prepared as pale yellow oil from **1c** (0.50 mmol, 82.1 mg) and (EtO)₃SiH (0.75 mmol, 125.8 mg), according to the General Procedure in 84% yield (138.0 mg, l/b > 99:1): ¹H NMR (400 MHz, CDCl₃) δ 7.24 (d, *J* = 8.0 Hz, 2 H), 7.06 (d, *J* = 8.0 Hz, 2 H), 3.79 (q, *J* = 6.8 Hz, 6 H), 2.90 (t, *J* = 7.2 Hz, 2 H), 2.30 (s, 3 H), 1.77-1.69 (m, 2 H), 1.20 (t, *J* = 6.8 Hz, 9 H), 0.79-0.75 (m, 2 H); ¹³C {¹H} NMR (100 MHz, CDCl₃) δ 135.7, 132.8, 129.9, 129.5, 58.3, 37.1, 22.7, 20.8, 18.2, 9.7; HRMS *m*/*z* (CI) calcd. for C₁₆H₂₈O₃SSiNa (M+Na)⁺ 351.1426, found 351.1419.

Triethoxy(*3*-((*4*-*methoxyphenyl*)*thio*)*propyl*)*silane* (*3d*) was prepared as pale yellow oil from **1d** (0.50 mmol, 90.1 mg) and (EtO)₃SiH (0.75 mmol, 125.8 mg), according to the General Procedure in 76% yield (130.9 mg, *l/b* > 99:1): 'H NMR (400 MHz, CDCl₃) δ 7.28-7.24 (m, 2 H), 6.78-6.74 (m, 2 H), 3.72 (q, *J* = 6.8 Hz, 6 H), 3.71 (s, 3 H), 2.77 (t, *J* = 7.2 Hz, 2 H), 1.66-1.58 (m, 2 H), 1.13 (t, *J* = 6.8 Hz, 9 H), 0.70-0.66 (m, 2 H); ¹³C {¹H} NMR (100 MHz, CDCl₃) δ 158.7, 133.0, 126.6, 114.4, 58.3, 55.3, 38.7, 22.8, 18.2, 9.6; HRMS *m/z* (CI) calcd. for C₁₆H₂₈O₄SSiNa (M+Na)⁺ 367.1375, found 367.1365.

2-((*3*-(*Triethoxysilyl*)*propyl*)*thio*)*pyridine* (*3e*) was prepared as pale yellow oil from **1e** (0.50 mmol, 75.6 mg) and (EtO)₃SiH (0.75 mmol, 125.8 mg), according to the General Procedure in 53% yield (83.6 mg, *l/b* > 99:1): ¹H NMR (400 MHz, CDCl₃) δ 8.44-8.42 (m, 1 H), 7.50-7.45 (m, 1 H), 7.19 (d, *J* = 8.4 Hz, 1 H), 7.00-6.95 (m, 1 H), 3.83 (q, *J* = 7.2 Hz, 6 H), 3.21 (t, *J* = 7.2 Hz, 2 H), 1.90-1.81 (m, 2 H), 1.23 (t, *J* = 7.2 Hz, 9 H), 0.84 (t, *J* = 8.0 Hz, 2 H); ¹³C {¹H} NMR (100 MHz, CDCl₃) δ 159.4, 149.4, 135.8, 122.1, 119.1, 58.4, 33.1, 23.0, 18.3, 10.0; MS (ESI) *m/z* (relative intensity) 315.4 (5), 55.1 (100) [M]⁺.

2-((3-(*Triethoxysilyl*)*propyl*)*thio*)*pyrimidine* (**3***f*) was prepared as pale yellow oil from **1***f* (0.50 mmol, 76.1 mg) and (EtO)₃SiH (0.75 mmol, 125.8 mg), according to the General Procedure in 80% yield (126.6 mg, *l/b* > 99:1): ¹H NMR (400 MHz, CDCl₃) δ 8.47 (d, *J* = 5.2 Hz, 2 H), 6.92 (dd, *Ji* = *J2* = 5.2 Hz, 1 H), 3.80 (q, *J* = 6.8 Hz, 6 H), 3.16 (t, *J* = 7.6 Hz, 2 H), 1.89-1.80 (m, 2 H), 1.20 (t, *J* = 6.8 Hz, 9 H), o.80 (t, *J* = 8.0 Hz, 2 H); ¹³C [¹H] NMR (100 MHz, CDCl₃) δ 172.6, 157.0, 116.1, 58.2, 33.6, 22.7, 18.1, 9.8; HRMS *m/z* (CI) calcd. for $C_{13}H_{24}N_2O_3SSiNa$ (M+Na)⁺ 339.1175, found 339.1179.

2, 4-Bis((3-(triethoxysilyl)propyl)thio)pyrimidine (**3g**) was prepared as pale yellow oil from **1g** (0.50 mmol, 112.17 mg) and (EtO)₃SiH (1.50 mmol, 251.6 mg), according to the General Procedure in 85% yield (235.0 mg, l/b > 99:1): 'H NMR (400 MHz, CDCl₃) δ 8.07 (d, J = 5.2 Hz, 1 H), 6.78 (d, J = 5.2 Hz, 1 H), 3.85-3.79 (m, 12 H), 3.20-3.14 (m, 4 H), 1.91-1.80 (m, 4 H), 1.24-1.20 (m, 18 H), 0.83-0.76 (m, 4 H); ¹³C {¹H} NMR (100 MHz, CDCl₃) δ 171.6, 170.0, 154.1, 114.0, 58.23, 58.19, 33.6, 31.9, 22.8, 22.7, 18.1, 9.9; HRMS m/z (Cl) calcd. for C₂₂H₄₅N₂O₆S₂Si₂ (M+H)⁺ 553.2258, found 553.2272.

2, 4, 6-Tris((3-(triethoxysilyl)propyl)thio)-1, 3, 5-triazine (**3h**) was prepared as pale yellow oil from **3h** (0.50 mmol, 148.7 mg) and (EtO)₃SiH (2.25 mmol, 377.4 mg), according to the General Procedure in 65% yield (256.8 mg, l/b > 99:1): 'H NMR

2

3

4

5

6

7

8

9

10

11

12

23

24

25

26

27

28

29

30

31

32

33

34

35

36

37

38

39

40

41

42

43

44

45

46

57

58 59

60

(400 MHz, CDCl₃) δ 3.81 (q, *J* = 6.8 Hz, 18 H), 3.10 (t, *J* = 7.2 Hz, 6 H), 1.87-1.78 (m, 6 H), 1.21 (t, *J* = 6.8 Hz, 27 H), 0.79-0.74 (m, 6 H); ¹³C {¹H} NMR (100 MHz, CDCl₃) δ 179.4, 58.3, 33.1, 22.7, 18.2, 9.9; HRMS *m*/*z* (Cl) calcd. for C₃₀H₆₃N₃O₉S₃Si₃Na (M+Na)⁺ 812.2932, found 812.2925.

i-Methyl-5-((3-(triethoxysilyl)propyl)thio)-iH-tetrazole (*3i*) was prepared as pale yellow oil from **ii** (0.50 mmol, 78.1 mg) and (EtO)₃SiH (0.75 mmol, 125.8 mg), according to the General Procedure in 90% yield (144.2 mg, l/b > 99:1): ⁱH NMR (400 MHz, CDCl₃) δ 3.90 (s, 3 H), 3.79 (q, *J* = 7.2 Hz, 6 H), 3.35 (t, *J* = 7.2 Hz, 2 H), 1.94-1.85 (m, 2 H), 1.19 (t, *J* = 7.2 Hz, 9 H), 0.76 (t, *J* = 8.0 Hz, 2 H); ⁱ³C {ⁱH} NMR (100 MHz, CDCl₃) δ 154.2, 58.3, 35.9, 33.2, 23.1, 18.1, 9.7; MS (ESI) *m/z* (relative intensity) 320.1 (5), 163.1 (100) [M]⁺.

13 2-((3-(Triethoxysilvl)propyl)thio)benzo[d]thiazole (3i) was prepared as pale yellow oil from 1j (0.50 mmol, 103.7 mg) and 14 (EtO)₃SiH (0.75 mmol, 125.8 mg), according to the General 15 Procedure in 96% yield (178.4 mg, *l/b* > 99:1): ¹H NMR (400 16 MHz, CDCl₃) δ 7.87 (d, J = 8.4 Hz, 1 H), 7.77 (d, J = 8.4 Hz, 1 H), 17 7.45-7.40 (m, 1 H), 7.33-7.28 (m, 1 H), 3.85 (q, J = 7.2 Hz, 6 H),18 3.40 (t, J = 7.2 Hz, 2 H), 2.03-1.94 (m, 2 H), 1.24 (t, J = 7.2 Hz, 9 19 H), 0.88-0.83 (m, 2 H); ¹³C {¹H} NMR (100 MHz, CDC₁₃) δ 167.2, 20 153.4, 135.2, 125.9, 124.0, 121.4, 120.8, 58.4, 36.4, 23.0, 18.2, 9.9; 21 MS (ESI) *m/z* (relative intensity) 371.1 (5), 167.0 (100) [M]⁺. 22

Triethoxy(3-(*methylthio*)*propyl*)*silane* (**3***k*) was prepared as pale yellow oil from **1k** (0.50 mmol, 44.1 mg) and (EtO)₃SiH (0.75 mmol, 125.8 mg), according to the General Procedure in 66% yield (83.3 mg, *l/b* > 99:1): ¹H NMR (400 MHz, CDCl₃) δ 3.82 (q, *J* = 7.2 Hz, 6 H), 2.52 (t, *J* = 6.8 Hz, 2 H), 2.09 (s, 1 H), 1.78-1.69 (m, 2 H), 1.22 (t, *J* = 7.2 Hz, 9 H), 0.74 (t, *J* = 8.4 Hz, 2 H); ¹³C {¹H} NMR (100 MHz, CDCl₃) δ 58.4, 37.2, 22.6, 18.3, 18.3, 9.7; MS (ESI) *m/z* (relative intensity) 252.1 (7), 163.1 (100) [M]⁺.

(3-(Benzylthio)propyl)triethoxysilane (31) was prepared as pale yellow oil from 11 (0.50 mmol, 82.1 mg) and (EtO)₃SiH (0.75 mmol, 125.8 mg), according to the General Procedure in 73% yield (119.9 mg, l/b > 99:1): 'H NMR (400 MHz, CDCl₃) δ 7.32-7.29 (m, 4 H), 7.25-7.20 (m, 1 H), 3.80 (q, *J* = 6.8 Hz, 6 H), 3.70 (s, 2 H), 2.44 (t, *J* = 7.2 Hz, 2 H), 1.72-1.64 (m, 2 H), 1.22 (t, *J* = 6.8 Hz, 9 H), 0.72-0.68 (m, 2 H); '³C {'H} NMR (100 MHz, CDCl₃) δ 138.6, 128.8, 128.4, 126.8, 58.3, 36.1, 34.4, 22.7, 18.3, 9.9; MS (ESI) *m/z* (relative intensity) 328.1 (5), 91.0 (100) [M]⁺.

Triethoxy(*3*-(*tritylthio*)*propyl*)*silane* (*3m*) was prepared as white powder from **1m** (0.50 mmol, 158.2 mg) and (EtO)₃SiH (0.75 mmol, 125.8 mg), according to the General Procedure in 72% yield (173.1 mg, *l/b* > 99:1): 'H NMR (400 MHz, CDCl₃) δ 7.52-7.48 (m, 6 H), 7.36-7.32 (m, 6 H), 7.29-7.24 (m, 3 H), 3.83 (q, *J* = 7.2 Hz, 6 H), 2.28 (t, *J* = 7.2 Hz, 2H), 1.66-1.58 (m, 2H), 1.27 (t, *J* = 7.2 Hz, 9 H), 0.68 (t, *J* = 8.0 Hz, 2 H); '³C {'H} NMR (100 MHz, CDCl₃) δ 145.1, 129.6, 127.8, 126.5, 66.3, 58.3, 35.1, 22.3, 18.2, 10.3; HRMS *m/z* (CI) calcd. for C₂₈H₃₆O₃SSiNa (M+Na)⁺ 503.2052, found 503.2048.

Triethoxy(3-((furan-2-ylmethyl)thio)propyl)silane (3n) was 47 prepared as pale yellow oil from in (0.50 mmol, 77.1 mg) and 48 (EtO)₃SiH (0.75 mmol, 125.8 mg), according to the General 49 Procedure in 63% yield (100.3 mg, l/b > 99:1): ¹H NMR (400 50 MHz, CDCl₃) δ 7.36-7.34 (m, 1 H), 6.31-6.29 (m, 1 H), 6.18-6.16 51 (m, 1 H), 3.81 (q, J = 7.2 Hz, 6 H), 3.71 (s, 2 H), 2.53 (t, J = 7.6 Hz)52 2 H), 1.73-1.64 (m, 2 H), 1.22 (t, J = 7.2 Hz, 9 H), 0.74-0.69 (m, 53 2 H); ¹³C {¹H} NMR (100 MHz, CDCl₃) δ 151.9, 142.0, 110.3, 107.2, 54 58.4, 34.8, 28.1, 22.8, 18.3, 9.9; HRMS m/z (CI) calcd. for 55 $C_{14}H_{26}O_4SSiNa (M+Na)^+$ 341.1219, found 341.1228. 56

(3-(Cyclohexylthio)propyl)triethoxysilane (30) was prepared as pale yellow oil from 10 (0.50 mmol, 78.1 mg) and (EtO)₃SiH (0.75 mmol, 125.8 mg), according to the General Procedure in 99% yield (158.7 mg, l/b > 99:1): ¹H NMR (400 MHz, CDCl₃) δ 3.82 (q, J = 6.8 Hz, 6 H), 2.65-2.59 (m, 1 H), 2.55 (t, J = 7.2 Hz, 2 H), 1.98-1.95 (m, 2 H), 1.78-1.75 (m, 2 H), 1.73-1.65 (m, 2 H), 1.33-1.25 (m, 6 H), 1.23 (t, J = 6.8 Hz, 9 H), 0.77-0.72 (m, 2 H); ¹³C {¹H} NMR (100 MHz, CDCl₃) δ 58.3, 43.2, 33.7, 33.1, 26.1, 25.8, 23.6, 18.2, 9.9; MS (ESI) m/z (relative intensity) 320.1 (6), 163.1 (100 [M]⁺.

Triethoxy(*4*-(*p*-*tolylthio*)*butyl*)*silane* (**3***p*) was prepared as pale yellow oil from **1p** (0.50 mmol, 89.1 mg) and (EtO)₃SiH (0.75 mmol, 125.8 mg), according to the General Procedure in 80% yield (137.0 mg, *l/b* > 99:1): ¹H NMR (400 MHz, CDCl₃) δ 7.25-7.22 (m, 2 H), 7.09-7.05 (m, 2 H), 3.80 (q, *J* = 6.8 Hz, 6 H), 2.87 (t, *J* = 7.2 Hz, 2 H), 2.30 (s, 3 H), 1.70-1.62 (m, 2 H), 1.58-1.50 (m, 2 H), 1.20 (t, *J* = 6.8 Hz, 9 H), 0.65-0.61 (m, 2 H); ¹³C {¹H} NMR (100 MHz, CDCl₃) δ 135.7, 133.1, 129.7, 129.5, 58.2, 33.8, 32.4, 22.0, 20.8, 18.2, 9.9; HRMS *m/z* (CI) calcd. for C₁₇H₃₀O₃SSiNa (M+Na)⁺ 365.1583, found 365.1580.

Triethoxy(*2*-(*phenylthio*)*ethyl*)*silane* (*3q*) was prepared as pale yellow oil from **1q** (0.50 mmol, 68.1 mg) and (EtO)₃SiH (0.75 mmol, 125.8 mg), according to the General Procedure in 47% yield (70.6 mg, *l/b* > 99:1): 'H NMR (400 MHz, CDCl₃) δ 7.29-7.24 (m, 2 H), 7.23-7.19 (m, 2 H), 7.18-7.13 (m, 1 H), 3.82 (q, *J* = 6.8 Hz, 6 H), 2.77-2.73 (m, 2 H), 1.23 (t, *J* = 6.8 Hz, 9 H), 1.02-0.97 (m, 2 H); ¹³C {¹H} NMR (100 MHz, CDCl₃) δ 144.6, 128.2, 127.7, 125.6, 58.3, 28.8, 18.3, 12.5; MS (ESI) *m/z* (relative intensity) 300.1 (5), 163.1 (100) [M]⁺.

Trimethoxy(3-(*phenylthio*)*propyl*)*silane* (**3***r*) was prepared as pale yellow oil from **1a** (0.50 mmol, 75.1 mg) and (MeO)₃SiH (0.75 mmol, 91.6 mg), according to the General Procedure in 68% yield (92.6 mg, *l/b* > 99:1): 'H NMR (400 MHz, CDCl₃) δ 7.37-7.34 (m, 2 H), 7.31-7.25 (m, 2 H), 7.21-7.14 (m, 1 H), 3.57 (s, 9 H), 2.97 (t, *J* = 7.2 Hz, 2 H), 1.83-1.74 (m, 2 H), 0.85-0.81 (m, 2 H); ¹³C {¹H} NMR (100 MHz, CDCl₃) δ 136.6, 129.0, 128.8, 125.7, 50.4, 36.3, 22.5, 8.4; MS (ESI) *m/z* (relative intensity) 272.1 (12), 121.0 (100) [M]⁺.

Diethoxy(methyl)(3-(phenylthio)propyl)silane (**3s**) was prepared as pale yellow oil from **1a** (0.50 mmol, 75.1 mg) and , (EtO)₂MeSiH (0.75 mmol, 100.8 mg)according to the General Procedure in 79% yield (112.4 mg, l/b > 99:1): 'H NMR (400 MHz, CDCl₃) δ 7.25-7.21 (m, 2 H), 7.19-7.14 (m, 2 H), 7.08-7.03 (m, 1 H), 3.64 (q, *J* = 7.2 Hz, 6 H), 2.85 (t, *J* = 7.2 Hz, 2 H), 1.67-1.58 (m, 2 H), 1.10 (t, *J* = 7.2 Hz, 9 H), 0.69-0.64 (m, 2 H), 0.00 (s, 3 H); ¹³C {¹H} NMR (100 MHz, CDCl₃) δ 136.7, 129.0, 128.8, 125.7, 58.1, 36.7, 22.8, 18.3, 13.4, -4.9; MS (ESI) *m*/*z* (relative intensity) 284.09 (9), 133.1 (100) [M]⁺.

Dimethyl(phenyl)(3-(phenylthio)propyl)silane (**3t**) was prepared as pale yellow oil from **1a** (0.50 mmol, 75.1 mg) and PhMe₂SiH (0.75 mmol, 102.3 mg), according to the General Procedure in 85% yield (121.8 mg, l/b > 99:1): ¹H NMR (400 MHz, CDCl₃) δ 7.56-7.52 (m, 2 H), 7.42-7.39 (m, 2 H), 7.39-7.38 (m, 1 H), 7.34-7.27 (m, 4 H), 7.22-7.18 (m, 1 H), 2.96 (t, *J* = 7.2 Hz, 2 H), 1.76-1.68 (m, 2 H), 0.99-0.93 (m, 2 H), 0.31 (s, 6 H); ¹³C {¹H} NMR (100 MHz, CDCl₃) δ 138.9, 136.8, 133.5, 129.0, 128.9, 128.8, 127.8, 125.6, 37.1, 23.7, 15.2, -3.1; HRMS *m/z* (Cl) calcd. for C₁₇H₂₂SSiNa (M+Na)⁺ 309.1109, found 309.1132.

Benzyldimethyl(*3*-(*phenylthio*)*propyl*)*silane* (*3u*) was prepared as pale yellow oil from **1a** (0.50 mmol, 75.1 mg) and BnMe₂SiH (0.75 mmol, 112.8 mg), according to the General Procedure in 96% yield (144.3 mg, *l/b* > 99:1): 'H NMR (400 MHz, CDCl₃) δ 7.37-7.31 (m, 4 H), 7.26-7.22 (m, 2 H), 7.21-7.18 (m, 1 H), 7.13-7.07 (m, 1 H), 7.02-6.98 (m, 2 H), 2.94 (t, *J* = 7.2 Hz, 2 H), 2.11 (s, 2 H), 1.70-1.62 (m, 2 H), 0.74-0.65 (m, 2 H),

0.00 (s, 6 H); ¹³C {¹H} NMR (100 MHz, CDCl₃) δ 140.0, 136.8, 128.9, 128.8, 128.2, 128.0, 125.7, 123.9, 37.1, 25.4, 23.8, 14.4, -3.7; HRMS *m*/*z* (Cl) calcd. for C₁₈H₂₄SSiNa (M+Na)⁺ 323.1266, found 323.1260.

1,4-Bis(dimethyl(3-(phenylthio)propyl)silyl)benzene (3w) was prepared as pale yellow oil from 1a (1.00 mmol, 150.2 mg) and 1,4-bis (dimethylsilyl)benzene (0.50 mmol, 97.2 mg)according to the General Procedure in 67% yield (165.8 mg, l/b >99:1): 'H NMR (400 MHz, CDCl₃) δ 7.50-7.48 (m, 4 H), 7.35-7.22 (m, 8 H), 7.21-7.18 (m, 2 H), 2.96 (t, J = 7.2 Hz, 4 H), 1.79-1.57 (m, 4 H), 0.95 (t, J = 8.0 Hz, 4 H), 0.29 (s, 12H); ¹³C {¹H} NMR (100 MHz, CDCl₃) δ 139.6, 136.8, 132.9, 129.1, 128.8, 125.7, 37.2, 23.7, 15.2, -3.2; HRMS m/z (CI) calcd. for C₂₈H₃₈S₂Si₂Na (M+Na)⁺ 517.1850, found 517.1829.

Dimethyl(2-methyl-3-(p-tolylthio)propyl)(phenyl)silane (**3x**) was prepared as pale yellow oil from **8c** (0.50 mmol, 89.1 mg) and PhMe₂SiH (0.75 mmol, 102.3 mg), according to the General Procedure in 74% yield (116.3 mg, l/b > 99:1): ¹H NMR (400 MHz, CDCl₃) δ 7.61-7.58 (m, 2 H), 7.46-7.41 (m, 3 H), 7.30-7.24 (m, 2 H), 7.18-7.12 (m, 2 H), 2.94-2.79 (m, 2 H), 2.41 (s, 3 H), 2.05-1.90 (m, 1 H), 1.30-1.20 (m, 1 H), 1.10 (d, J = 6.8 Hz, 3 H), 0.90-0.80 (m, 1 H), 0.39 (s, 6 H); ¹³C {¹H} NMR (100 MHz, CDCl₃) δ 139.4, 135.6, 133.5, 129.8, 129.5, 128.8, 127.7, 45.2, 29.7, 23.5, 22.3, 20.9, -2.0, -2.3; HRMS m/z (Cl) calcd. for C₁₉H₃₀NSSi (M+NH4⁺)⁺ 332.1863, found 332.1882.

Triethoxy(*3-phenoxypropy*]*silane* (*5a*) was prepared as pale yellow oil from **4a** (0.50 mmol, 67.1 mg) and (EtO)₃SiH (0.75 mmol, 125.8 mg), according to the General Procedure in 83% yield (123.9 mg, l/b > 99:1): ¹H NMR (400 MHz, CDCl₃) δ 7.34-7.27 (m, 2 H), 7.00-6.91 (m, 3 H), 3.98 (t, *J* = 6.8 Hz, 2 H), 3.87 (q, *J* = 7.2 Hz, 6 H), 2.00-1.90 (m, 2 H), 1.27 (t, *J* = 7.2 Hz, 9 H), 0.81 (t, *J* = 8.0 Hz, 2 H); ¹³C {¹H} NMR (100 MHz, CDCl₃) δ 159.0, 129.3, 120.4, 114.5, 69.7, 58.4, 22.8, 18.2, 6.5; MS (ESI) *m/z* (relative intensity) 298.1 (5), 163.1 (100) [M]⁺.

(3-([*i*, *i'-Biphenyl]-2-yloxy)propyl)triethoxysilane* (**5b**) was prepared as pale yellow oil from **4b** (0.50 mmol, 105.1 mg) and (EtO)₃SiH (0.75 mmol, 125.8 mg), according to the General Procedure in 99% yield (185.4 mg, *l/b* > 99:1): ¹H NMR (400 MHz, CDCl₃) δ 7.49-7.44 (m, 2 H), 7.32-7.27 (m, 2 H), 7.27-7.18 (m, 3 H), 6.96-6.86 (m, 2 H), 3.86 (t, *J* = 6.4 Hz, 2 H), 3.70 (q, *J* = 7.2 Hz, 6 H), 1.79-1.71 (m, 2 H), 1.12 (t, *J* = 7.2 Hz, 9 H), 0.64-0.59 (m, 2 H); ¹³C {¹H} NMR (100 MHz, CDCl₃) δ 155.9, 138.6, 130.9, 130.8, 129.6, 128.5, 127.7, 126.6, 120.7, 112.5, 70.3, 58.3, 22.8, 18.2, 6.5; HRMS *m/z* (Cl) calcd. for C₂₁H₃₀O₄SiNa (M+Na)+ 397.1811, found 397.1810.

8, 8-diethoxy-2, 2, 3, 3-tetramethyl-4, 9-dioxa-3, 8-disilaundecane (**5c**) was prepared as pale yellow oil from **4c** (0.50 mmol, 95.1 mg) and (EtO)₃SiH (0.75 mmol, 125.8 mg), according to the General Procedure in 92% yield (238.6 mg, l/b > 99:1): ¹H NMR (400 MHz, CDCl₃) δ 3.81 (q, J = 7.2 Hz, 6 H), 3.56 (t, J =6.8 Hz, 2 H), 1.68-1.58 (m, 2 H), 1.21 (t, J = 7.2 Hz, 9 H), 0.88 (s, 9 H), 0.65-0.59 (m, 2 H), 0.03 (s, 6 H); ¹³C {¹H} NMR (100 MHz, CDCl₃) δ 65.4, 58.3, 26.1, 25.9, 18.3, 18.2, 6.1, -5.3; MS (ESI) m/z(relative intensity) 336.2 (5), 237.3 (100) [M]⁺.

1, 4-Bis(3-(triethoxysilyl)propoxy)benzene (**5d**) was prepared as pale yellow oil from **4d** (0.50 mmol, 95.1 mg) and (EtO)₃SiH (1.50 mmol, 251.6 mg), according to the General Procedure in 92% yield (238.6 mg, l/b > 99:1): 'H NMR (400 MHz, CDCl₃) δ 6.73 (s, 4 H), 3.81 (t, *J* = 6.8 Hz, 4 H), 3.75 (q, *J* = 6.8 Hz, 12 H), 1.83-1.76 (m, 4 H), 1.15 (t, *J* = 6.8 Hz, 18 H), 0.70-0.66 (m, 4 H); ¹³C {¹H} NMR (100 MHz, CDCl₃) δ 153.0, 115.3, 70.4, 58.3, 22.8, 18.2, 6.4. HRMS *m/z* (CI) calcd. for C₂₄H₄₆O₈Si₂Na (M+Na)⁺ 541.2629, found 541.2624. 2, 2'-Bis(3-(triethoxysilyl)propoxy)-1, 1'-binaphthalene (**5e**) was prepared as pale yellow oil from **4e** (0.50 mmol, 183.2 mg) and (EtO)₃SiH (1.50 mmol, 251.6 mg), according to the General Procedure in 88% yield (305.8 mg, l/b > 99:1): ¹H NMR (400 MHz, CDCl₃) δ 7.95-7.93 (m, 2 H), 7.88-7.84 (m, 2 H), 7.45-7.41 (m, 2 H), 7.34-7.29 (m, 2 H), 7.25-7.18 (m, 4 H), 4.01-3.91 (m, 4 H), 3.63 (q, J = 7.2 Hz, 12 H), 1.60-1.52 (m, 4 H), 1.15 (t, J = 7.2 Hz, 18 H), 0.38-0.24 (m, 4 H); ¹³C [¹H] NMR (100 MHz, CDCl₃) δ 154.4, 134.2, 129.2, 128.9, 127.7, 126.0, 125.4, 123.3, 120.3, 115.4, 71.4, 58.1, 22.8, 18.2, 5.7; MS (ESI) m/z (relative intensity) 694.8 (12), 205.3 (100) [M]⁺.

Benzyl(3-methoxypropyl)dimethylsilane (**5***f*) was prepared as pale yellow oil from **4***f* (0.50 mmol, 36.1 mg) and BnMe₂SiH (0.75 mmol, 112.8 mg), according to the General Procedure in 96% yield (106.8 mg, *l/b* > 99:1): ¹H NMR (400 MHz, CDCl₃) δ 7.29-7.23 (m, 2 H), 7.15-7.09 (m, 1 H), 7.07-7.03 (m, 2 H), 3.38 (s, 3 H), 3.37 (t, *J* = 6.8 Hz, 2 H), 2.15 (s, 2 H), 1.66-1.59 (m, 2 H), 0.60-0.50 (m, 2 H), 0.04 (s, 6 H); ¹³C {¹H} NMR (100 MHz, CDCl₃) δ 140.2, 128.1, 128.0, 123.8, 75.5, 58.4, 25.4, 23.8, 10.7, -3.8; MS (ESI) *m/z* (relative intensity) 222.1 (5), 149.1 (100) [M]⁺.

(*2-Butoxyethyl*)*triethoxysilane* (*5g*) was prepared as pale yellow oil from **4g** (0.50 mmol, 50.1 mg) and (EtO)₃SiH (0.75 mmol, 125.8 mg), according to the General Procedure in 92% yield (121.6 mg, *l/b* > 99:1): ¹H NMR (400 MHz, CDCl₃) δ 3.75 (q, *J* = 7.2 Hz, 6 H), 3.50-3.47 (m, 2 H), 3.33 (t, *J* = 6.8 Hz, 2 H), 1.52-1.43 (m, 2 H), 1.35-1.25 (m, 2 H), 1.16 (t, *J* = 7.2 Hz, 9 H), 1.01 (t, *J* = 8.0 Hz, 2 H), 0.85 (t, *J* = 7.2 Hz, 3 H); ¹³C {¹H} NMR (100 MHz, CDCl₃) δ 70.2, 66.3, 58.4, 31.9, 19.4, 18.2, 13.9, 12.9; MS (ESI) *m/z* (relative intensity) 264.1 (5), 163.1 (100) [M]⁺.

(*R*)-tert-Butyl 2-((3-(triethoxysilyl)propoxy)methyl)pyrrolidine-1-carboxylate (**5**h) was prepared as pale yellow oil from **4**h (0.50 mmol, 120.7 mg) and (EtO)₃SiH (0.75 mmol, 125.8 mg), according to the General Procedure in 80% yield (162.2 mg, l/b > 99:1): 'H NMR (400 MHz, CDCl₃) δ 4.00-3.80 (m, 2 H), 3.79 (q, *J* = 6.8 Hz, 6 H), 3.52-3.49 (m, 1 H), 3.44-3.32 (m, 2 H), 3.33-3.22 (m, 2 H), 1.92-1.85 (m, 2 H), 1.82-1.72 (m, 2 H), 1.70-1.51 (m, 2 H), 1.44 (s, 9 H), 1.20 (t, *J* = 6.8 Hz, 9 H), 0.63-0.58 (m, 2 H); ¹³C {¹H} NMR (100 MHz, CDCl₃) δ 154.4, 79.0, 73.4, 71.4, 70.8, 58.2, 56.4, 46.7, 46.3, 28.4, 23.0, 18.2, 6.4; HRMS *m*/*z* (CI) calcd. for C₁₉H₃₉NO₆SiNa (M+Na)⁺ 428.2445, found 428.2448.

Triethoxy(*3*-(((*iR*, 2*S*, 5*R*)-*2*-*isopropyl*-5-*methylcyclohexyl*)*oxy*)*propyl*)*silane* (*5***i**) was prepared as pale yellow oil from **4i** (0.50 mmol, 98.2 mg) and (EtO)₃SiH (0.75 mmol, 125.8 mg), according to the General Procedure in 89% yield (160.3 mg, *l/b* > 99:1): 'H NMR (400 MHz, CDCl₃) δ 3.80 (q, *J* = 7.2 Hz, 6 H), 3.58-3.53 (m, 1 H), 3.25-3.20 (m. 1 H), 3.05-2.92 (m, 1 H), 2.26-2.16 (m, 1 H), 2.10-2.03 (m, 1 H), 1.69-1.63 (m, 2 H), 1.61-1.55 (m, 2 H), 1.38-1.26 (m, 1 H), 1.20 (t, *J* = 7.2 Hz, 9 H), 0.99-0.90 (m, 2 H), 0.89 (d, *J* = 6.8 Hz, 3 H), 0.86 (s, 3 H), 0.84-0.78 (m, 2 H), 0.75 (d, *J* = 7.2 Hz, 3 H), 0.67-0.60 (m, 2 H); ¹³C {¹H} NMR (100 MHz, CDCl₃) δ 79.1, 70.9, 58.2, 48.3, 40.6, 34.6, 31.5, 25.5, 23.5, 23.3, 22.3, 20.9, 18.2, 16.2, 6.6; MS (ESI) *m/z* (relative intensity) 360.2 (5), 95.1 (100) [M]⁺.

Triethoxy(*3*-(((*2S*, *3R*, *4S*, *5R*, *6R*)-*3*, *4*, *5*-*tris*(*benzyloxy*)-*6*-((*benzyloxy*)*methyl*)*tetrahydro*-*2H*-*pyran*-*2*-*yl*)*oxy*)*propyl*)*silane* (*5j*) was prepared as pale yellow oil from *4j* (0.50 mmol, 290.4 mg) and (EtO)₃SiH (0.75 mmol, 125.8 mg), according to the General Procedure in 70% yield (260.7 mg, *l/b* > 99:1): ¹H NMR (400 MHz, CDCl₃) δ 7.46-7.30 (m, 18 H), 7.26-7.20 (m, 2 H), 5.10-5.03 (m, 1 H), 4.95-4.89 (m, 2 H), 4.89-4.82 (m, 2 H), 4.76-4.67 (m, 2 H), 4.58-4.49 (m, 2 H), 4.09 (t, J = 9.2 Hz, 1 H), 3.94-3.86 (m, 6 H), 3.84-3.80 (m, 1 H), 3.79-3.74 (m, 1

56

57

58 59

60

2

3

4

5

6

7

49

50

51

52

53

54

55

56

57

58 59

60

H), 3.74-3.68 (m, 2 H), 3.67-3.63 (m, 1 H), 3.61-3.48 (m, 2 H), 1.90-1.81 (m, 2 H), 1.33-1.28 (m, 9 H), 0.83-0.71 (m, 2 H); ${}^{13}C$ { ^{1}H } NMR (100 MHz, CDCl₃) δ 138.8, 138.5, 138.4, 138.3, 138.2, 138.1, 138.0, 137.9, 128.3, 128.2, 128.2, 128.0, 127.9, 127.8, 127.8, 127.8, 127.7, 127.6, 127.5, 127.4, 127.4, 103.5, 96.7, 84.6, 82.2, 82.0, 80.077.8, 77.6, 75.5, 74.9, 74.7, 73.3, 72.9, 72.1, 70.2, 70.0, 68.9, 68.4, 58.2, 23.1, 22.7, 18.2, 6.6, 6.5; HRMS *m/z* (CI) calcd. for C₄₃H₅₆O₉SiNa (M+Na)⁺ 767.3592, found 767.3584.

(3-(((3aR,5R,6S,6aR)-5-((R)-2,2-Dimethyl-1,3-dioxolan-4-8 yl)-2,2-dimethyltetrahydrofuro[2,3-d][1,3]dioxol-6-9 yl)oxy)propyl)triethoxysilane (5k) was prepared as pale yellow 10 oil from $4\mathbf{k}$ (0.50 mmol, 150.2 mg) and $(EtO)_3SiH$ (0.75 mmol, 11 125.8 mg), according to the General Procedure in 67% yield 12 (155.7 mg, 1/b > 99:1): ¹H NMR (400 MHz, CDCl₃) δ 5.87 (d, J = 13 3.6 Hz, 1 H), 4.54 (d, J = 3.6 Hz, 1 H), 4.32 (q, J = 6.4 Hz, 1 H), 4.14 (q, J = 3.2 Hz, 1 H), 4.10-4.06 (m, 2 H), 4.01-3.97 (m, 1 H), 14 3.82 (q, J = 7.2 Hz, 6 H), 3.62-3.46 (m, 2 H), 1.72-1.64 (m, 2 H), 15 1.50 (s, 3H), 1.43 (s, 3H), 1.35 (s, 3H), 1.32 (s, 3H), 1.23 (t, J = 7.2 16 Hz, 9 H), 0.68-0.63 (m, 2 H); ${}^{13}C$ { ${}^{1}H$ } NMR (100 MHz, CDCl₃) 17 $\delta \ {\it 111.7, \ 108.8, \ 105.2, \ 82.5, \ 82.0, \ 81.1, \ 72.7, \ 72.5, \ 67.1, \ 58.3, \ 26.8, }$ 18 26.7, 26.2, 25.3, 23.1, 18.3, 6.5; HRMS m/z (CI) calcd. for 19 C21H40O9SiNa (M+Na)+ 487.2340, found 487.2337. 20

(8R, 9S, 13S, 14S)-13-Methyl-3-(3-(triethoxysilyl)propoxy)-7, 21 8, 9, 11, 12, 13, 15, 16-octahydro-6H-cyclopenta[a]phenanthren-22 17(14H)-one (5l) was prepared as pale yellow oil from 4l (0.50 23 mmol, 155.2 mg) and (EtO)₃SiH (0.75 mmol, 125.8 mg), accord-24 ing to the General Procedure in 81% yield (192.3 mg, 1/b > 99:1): ¹H NMR (400 MHz, CDCl₃) δ 7.18 (d, J = 8.6 Hz, 1 H), 6.73-6.68 25 (m, 1 H), 6.65-6.62 (m, 1 H), 3.91 (t, J = 6.8 Hz, 2 H), 3.83 (q, J = 26 7.2 Hz, 6 H), 2.91-2.85 (m, 2 H), 2.54-2.46 (m, 1 H), 2.43-2.36 27 (m, 1 H), 2.29-2.20 (m, 1 H), 2.19-2.04 (m, 2 H), 2.03-1.93 (m, 2 28 H), 1.92-1.84 (m, 2 H), 1.66-1.57 (m, 2 H), 1.55-1.52 (m, 2 H), 1.49-29 1.35 (m, 2 H), 1.23 (t, J = 7.2 Hz, 9 H), 0.91 (s, 3 H), 0.78-0.73 (m, 2 H), 0.78-0.30 2 H); ¹³C {¹H} NMR (100 MHz, CDCl₃) δ 157.1, 137.6, 131.8, 126.2, 31 114.5, 112.1, 69.9, 58.4, 50.4, 48.0, 44.0, 38.4, 35.8, 31.6, 29.6, 56.5, 32 25.9, 22.8, 21.6, 18.3, 13.8, 6.5; HRMS m/z (CI) calcd. for 33 C₂₇H₄₂O₅SiNa (M+Na)⁺ 497.2700, found 497.2701.

34 (8R, 9S, 10R, 13S, 14S)-10, 13-Dimethyl-17-(3-(triethoxysi-35 lyl)propoxy)-6, 7, 8, 9, 10, 11, 12, 13, 14, 15, 16, 17-dodecahydro-1H-cyclopenta[a]phenanthren-3 (2H)-one (5m) was prepared 36 as pale yellow oil from 4m (0.50 mmol, 164.2 mg) and 37 (EtO)₃SiH (0.75 mmol, 125.8 mg), according to the General 38 Procedure in 57% yield (140.4 mg, 1/b > 99:1): ¹H NMR (400 39 MHz, CDCl₃) δ 5.66 (s, 1 H), 3.75 (q, J = 7.2 Hz, 6 H), 3.39-3.27 40 (m, 2 H), 3.22 (t, J = 8.4 Hz, 1 H), 2.37-2.26 (m, 2 H), 2.26-2.1641 (m, 1 H), 2.00-1.90 (m, 2 H), 1.90-1.84 (m, 1 H), 1.80-1.74 (m, 1 42 H), 1.67-1.61 (m, 2 H), 1.55-1.43 (m, 4 H), 1.43-1.34 (m, 2 H), 1.26-43 1.18 (m, 2 H), 1.16 (t, J = 7.2 Hz, 9 H), 1.12 (s, 3 H), 0.96-0.86 (m, 44 2 H), 0.86-0.78 (m, 2 H), 0.73 (s, 3 H), 0.60-0.54 (m, 2 H): ¹³C 45 ${}^{1}H$ NMR (100 MHz, CDCl₃) δ 199.6, 171.4, 123.8, 88.7, 72.3, 58.3, 46 53.9, 50.7, 42.8, 38.6, 37.7, 35.7, 35.5, 33.9, 32.8, 31.6, 28.1, 23.5, 47 23.3, 20.7, 18.3, 17.4, 11.6, 6.5; HRMS m/z (CI) calcd. for C₂₈H₄₉O₅Si (M+H)⁺ 493.3350, found 493.3348. 48

Triethoxy(3-(((R)-2,5,7,8-tetramethyl-2-((4R,8R)-4,8,12-trimethyltridecyl)chroman-6-yl)oxy)propyl)silane (5n) was prepared as pale yellow oil from 4n (0.50 mmol, 235.4 mg) and(EtO)₃SiH (0.75 mmol, 125.8 mg), according to the GeneralProcedure in 87% yield (276.3 mg,*l/b*> 99:1): 'H NMR (400 $MHz, CDCl₃) <math>\delta$ 3.85 (q, *J* = 7.2 Hz, 6 H), 3.61 (t, *J* = 6.8 Hz, 2 H), 2.56 (t, *J* = 6.8 Hz, 2 H), 2.16 (s, 3 H), 2.12 (s, 3 H), 2.07 (s, 3 H), 1.95-1.86 (m, 2 H), 1.84-1.70 (m, 2 H), 1.56 (s, 3 H), 1.54-1.50 (m, 2 H), 1.49-1.31 (m, 6 H), 1.26-1.22 (m, 15 H), 1.14 (t, *J* = 7.6 Hz, 2 H), 1.11-1.04 (m, 4 H), 0.88-0.82 (m, 13 H), 0.81-0.76 (m, 2 H); ¹³C {¹H} NMR (100 MHz, CDCl₃) δ 148.3, 147.6, 127.9, 125.8, 122.7, 117.4, 75.2, 74.7, 58.4, 40.1, 39.4, 37.5, 37.5, 37.4, 37.3, 32.8, 32.7, 31.3, 28.0, 24.8, 24.4, 23.9, 23.6, 22.7, 22.6, 21.0, 20.6, 19.7, 19.6, 18.3, 12.8, 11.9, 11.8, 6.7; MS (ESI) *m/z* (relative intensity) 634.8 (10), 430.7 (100) [M]⁺.

Triethoxy(*3*-(*phenylsulfonyl*)*propyl*)*silane* (*7a*) was prepared as pale yellow oil from **6a** (0.50 mmol, 91.1 mg) and (EtO)₃SiH (0.75 mmol, 125.8 mg), according to the General Procedure in 99% yield (171.5 mg, *l/b* > 99:1): ¹H NMR (400 MHz, CDCl₃) δ 7.88-7.85 (m, 2 H), 7.64-7.59 (m, 1 H), 7.55-7.51 (m, 2 H), 3.73 (q, *J* = 6.8 Hz, 6 H), 3.15-3.10 (m, 2 H), 1.84-1.75 (m, 2 H), 1.14 (t, *J* = 6.8 Hz, 9 H), 0.64 (t, *J* = 8.0 Hz, 2 H); ¹³C {¹H} NMR (100 MHz, CDCl₃) δ 139.0, 133.4, 129.0, 127.9, 58.3, 58.2, 18.0, 16.7, 9.1; MS (ESI) *m/z* (relative intensity) 346.1 (5), 91.0 (100) [M]⁺.

Bis(*3*-(*triethoxysilyl*)*propyl*)*phthalate* (*7b*) was prepared as pale yellow oil from **6b** (0.50 mmol, 123.1 mg) and (EtO)₃SiH (1.50 mmol, 251.6 mg), according to the General Procedure in 80% yield (229.9 mg, *l/b* > 99:1): ¹H NMR (400 MHz, CDCl₃) δ 7.74-7.70 (m, 2 H), 7.55-7.51 (m, 2 H), 4.31-4.26 (m, 4 H), 3.87-3.80 (m, 12 H), 1.88-1.81 (m, 4 H), 1.26-1.21 (m, 18 H), 0.74-0.68 (m, 4 H); ¹³C {¹H} NMR (100 MHz, CDCl₃) δ 167.5, 132.2, 130.8, 128.8, 67.6, 58.3, 22.1, 18.2, 6.5; MS (ESI) *m/z* (relative intensity) 574.4 (5), 207.3 (100) [M]⁺.

i-(*3*-(*Triethoxysilyl*)*propyl*)-*iH*-*indole* (*7***c**) was prepared as pale yellow oil from **6c** (0.50 mmol, 78.6 mg) and (EtO)₃SiH (0.75 mmol, 125.8 mg), according to the General Procedure in 70% yield (112.5 mg, *l/b* > 99:1): 'H NMR (400 MHz, CDCl₃) δ 7.63-7.59 (m, 1 H), 7.36-7.32 (m, 1 H), 7.21-7.14 (m, 1 H), 7.10-7.05 (m, 2 H), 6.48-6.45 (m, 1 H), 4.10 (t, *J* = 7.2 Hz, 2 H), 3.78 (q, *J* = 6.8 Hz, 6 H), 2.00-1.89 (m, 2 H), 1.20 (t, *J* = 6.8 Hz, 9 H), 0.60 (t, *J* = 8.0 Hz, 2 H); ¹³C {¹H} NMR (100 MHz, CDCl₃) δ 135.9, 128.5, 127.8, 121.2, 120.8, 119.1, 109.4, 100.7, 58.4, 48.6, 23.8, 18.2, 7.6; HRMS *m/z* (CI) calcd. for C₁₇H₂₈NO₃Si (M+H)+ 322.1839, found 322.1842.

N-(3-(*Triethoxysilyl*)*propyl*)*aniline* (*7d*) was prepared as colorless oil from **6d** (0.50 mmol, 66.6 mg) and (EtO)₃SiH (1.50 mmol, 251.6 mg), according to the General Procedure in 92% yield (136.8 mg, *l/b* > 99:1): ¹H NMR (400 MHz, CDCl₃) δ 7.21-7.18 (m, 2 H), 6.72-6.68 (m, 1 H), 6.66-6.63 (m, 2 H), 3.86 (q, *J* = 7.2 Hz, 6 H), 3.16 (t, *J* = 7.2 Hz, 2 H), 1.82-1.73 (m, 2 H), 1.26 (t, *J* = 7.2 Hz, 9 H), 0.77-0.72 (m, 2 H); ¹³C {¹H} NMR (100 MHz, CDCl₃) δ 148.2, 129.2, 117.1, 112.8, 58.4, 46.5, 22.7, 18.3, 7.8; MS (ESI) *m/z* (relative intensity) 297.2 (6), 84.3 (100) [M]⁺.

Diethyl(3-(triethoxysilyl)propyl)phosphine oxide (7e) was prepared as pale yellow oil from 6e (0.50 mmol, 73.1 mg) and (EtO)₃SiH (0.75 mmol, 125.8 mg), according to the General Procedure in 98% yield (152.1 mg, l/b > 99:1): ¹H NMR (400 MHz, CDCl₃) δ 4.12-3.98 (m, 4 H), 3.77 (q, J = 6.8 Hz, 6 H), 1.81-1.64 (m, 4 H), 1.27 (t, J = 7.2 Hz, 6 H), 1.18 (t, J = 6.8 Hz, 9 H), 0.69 (t, J = 8.0 Hz, 2 H); ¹³C {¹H} NMR (100 MHz, CDCl₃) δ 61.1, 58.2, 29.3, 27.9, 18.1, 16.3, 11.6; HRMS m/z (CI) calcd. for C₁₃H₃₂O₆PSi (M+H)⁺ 343.1707, found 343.1696.

Triethoxy(*3*-(*4*, *4*, *5*, *5*-*tetramethyl*-*1*, *3*, *2*-*dioxaborolan*-2*yl*)*propyl*)*silane* (*7f*) was prepared as pale yellow oil from **6f** (0.50 mmol, 67.5 mg) and BnMe₂SiH (0.75 mmol, 112.8 mg), according to the General Procedure in 93% yield (132.7 mg, *l/b* > 99:1): ¹H NMR (400 MHz, CDCl₃) δ 7.25-7.20 (m, 2 H), 7.11-7.06 (m, 1 H), 7.03-6.99 (m, 2 H), 3.41 (t, *J* = 6.8 Hz, 2 H), 2.1 (s, 2 H), 1.92-1.82 (m, 2 H), 1.49-1.41 (m, 2 H), 0.54-0.50 (m, 2 H), 0.0 (s, 6 H); ¹³C [¹H] NMR (100 MHz, CDCl₃) δ 140.2, 128.2, 128.0, 123.9, 36.2, 33.5, 25.5, 22.3, 13.8, -3.7; MS (ESI) *m/z* (relative intensity) 284.0 (6), 149.3 (100) [M]⁺. *Triethoxy*(3-(4, 4, 5, 5-*tetramethyl*-1, 3, 2-*dioxaborolan*-2y*l*)*propyl*)*silane* (**7g**) was prepared as pale yellow oil from **6g** (0.50 mmol, 84.0 mg) and (EtO)₃SiH (0.75 mmol, 125.8 mg), according to the General Procedure in 96% yield (159.5 mg, *l/b* > 99:1): 'H NMR (400 MHz, CDCl₃) δ 3.82 (q, *J* = 7.2 Hz, 6 H), 1.61-1.52 (m, 2 H), 1.25 (s, 12 H), 1.23 (t, *J* = 7.2 Hz, 9 H), 0.86 (t, *J* = 7.6 Hz, 2 H), 0.71-0.66 (m, 2 H); ¹³C {¹H} NMR (100 MHz, CDCl₃) δ 82.8, 58.2, 24.8, 18.2, 17.4, 13.3; MS (ESI) *m/z* (relative intensity) 332.2 (5), 163.1 (100) [M]⁺.

Triethoxy(*2*-(*trimethylsily*)*ethyl*)*silane* (**7***h*) was prepared as pale yellow oil from **6h** (0.50 mmol, 50.1 mg) and (EtO)₃SiH (0.75 mmol, 125.8 mg), according to the General Procedure in 70% yield (92.6 mg, *l/b* > 99:1): ¹H NMR (400 MHz, CDCl₃) δ 3.84 (q, *J* = 7.2 Hz, 6 H), 1.52-1.44 (m, 2 H), 1.25 (t, *J* = 7.2 Hz, 9 H), 0.76-0.70 (m, 2 H), 0.64-0.59 (m, 2 H), 0.0 (s, 9 H); ¹³C {¹H} NMR (100 MHz, CDCl₃) δ 58.2, 20.9, 18.2, 17.4, 14.8, -1.7; MS (ESI) *m/z* (relative intensity) 278.1 (5), 163.0 (100) [M]⁺.

6, 6-Diethoxy-3-methoxy-3-methyl-2, 7-dioxa-3, 6-disilanonane (7i) was prepared as pale yellow oil from **6i** (0.50 mmol, 66.1 mg) and (EtO)₃SiH (0.75 mmol, 125.8 mg), according to the General Procedure in 76% yield (112.7 mg, l/b > 99:1): ¹H NMR (400 MHz, CDCl₃) δ 3.71 (q, J = 7.2 Hz, 6 H), 3.40 (s, 6 H), 1.11 (t, J = 7.2 Hz, 9 H), 0.58-0.45 (m, 4 H), 0.0 (s, 3 H); ¹³C {¹H} NMR (100 MHz, CDCl₃) δ 58.4, 50.2, 18.3, 4.0, 1.4, -6.6; HRMS m/z (Cl) calcd. for C₁₁H₂₈O₅Si₂Na (M+H)⁺ 319.1373, found 319.1381.

Triethoxy(*4-phenylbutyl*)*silane* (*7j*) was prepared as pale yellow oil from **6j** (0.50 mmol, 59.1 mg) and (EtO)₃SiH (0.75 mmol, 125.8 mg), according to the General Procedure in 71% yield (100.3 mg, l/b > 99:1): ¹H NMR (400 MHz, CDCl₃) δ 7.28-7.23 (m, 2 H), 7.19-7.13 (m, 3 H), 3.80 (q, *J* = 6.8 Hz, 6 H), 2.64 (t, *J* = 7.6 Hz, 2 H), 1.79-1.71 (m, 2 H), 1.21 (t, *J* = 6.8 Hz, 9 H), 0.69-0.65 (m, 2 H); ¹³C {¹H} NMR (100 MHz, CDCl₃) δ 142.3, 128.5, 128.1, 125.6, 58.3, 39.2, 24.8, 18.2, 10.1; MS (ESI) *m/z* (relative intensity) 282.2 (5), 163.1 (100) [M]⁺.

Triethoxy(*4-phenylbutyl*)*silane* (*7k*) was prepared as pale yellow oil from 6k (0.50 mmol, 66.1 mg) and (EtO)₃SiH (0.75 mmol, 125.8 mg), according to the General Procedure in 83% yield (123.0 mg, *l/b* > 99:1): ¹H NMR (400 MHz, CDCl₃) δ 7.27-7.23 (m, 2 H), 7.17-7.13 (m, 3 H), 3.82 (q, *J* = 7.2 Hz, 6 H), 2.60 (t, *J* = 7.6 Hz, 2 H), 1.71-1.61 (m, 2 H), 1.51-1.43 (m, 2 H), 1.21 (t, *J* = 7.2 Hz, 9 H), 0.70-0.63 (m, 2 H); ¹³C {¹H} NMR (100 MHz, CDCl₃) δ 142.6, 128.3, 128.1, 125.5, 58.2, 35.5, 34.8, 22.4, 18.2, 10.2; MS (ESI) *m/z* (relative intensity) 296.1 (5), 250.1 (100) [M]⁺.

Triethoxy(phenethyl)silane (*γl*) was prepared as pale yellow oil from **61** (0.50 mmol, 52.1 mg) and (EtO)₃SiH (0.75 mmol, 125.8 mg), according to the General Procedure in 88% yield (118.1 mg, l/b > 99:1): ¹H NMR (400 MHz, CDCl₃) δ 7.29-7.24 (m, 2 H), 7.24-7.18 (m, 2 H), 7.18-7.13 (m, 1 H), 3.82 (q, *J* = 6.8 Hz, 6 H), 2.77-2.71 (m, 2 H), 1.23 (t, *J* = 6.8 Hz, 9 H), 1.02-0.97 (m, 2 H); ¹³C [¹H] NMR (100 MHz, CDCl₃) δ 144.6, 128.2, 127.7, 125.6, 58.3, 28.8, 18.3, 12.5; MS (ESI) *m/z* (relative intensity) 268.1 (20), 163.1 (100) [M]⁺.

4,4,9,9-*Tetraethoxy*-6-(4-*methylpent-3-en-1-yl*)-3,10-*dioxa*-4,9-*disiladodecane* (7*m*) was prepared as pale yellow oil from 6**m** (0.50 mmol, 68.1 mg) and (EtO)₃SiH (1.50 mmol, 251.6 mg), according to the General Procedure in 44% yield (102.3 mg, *l/b* > 99:1): ¹H NMR (400 MHz, CDCl₃) δ 5.14-5.10 (m, 1 H), 3.86-3.80 (m, 12 H), 1.97 (q, *J* = 7.6 Hz, 2 H), 1.69 (s, 3 H), 1.62 (s, 3 H), 1.50-1.43 (m, 2 H), 1.40-1.33 (m, 2 H), 1.28-1.26 (m, 1 H), 1.28-1.22 (m, 18 H), 0.66-0.59 (m, 4 H); ¹³C {¹H} NMR (100 MHz, CDCl₃) δ 130.9, 125.0, 58.3, 58.2, 35.3, 35.0, 28.4, 25.7, 25.1, 18.3, 18.2, 17.6, 14.9, 6.6; HRMS m/z (CI) calcd. for $C_{22}H_{48}O_6Si_2Na$ (M+Na)⁺ 487.2887, found 487.2885.

2,4-Bis(allylthio)pyrimidine (2q): NaOt-Bu (13.0 mmol) was added to ethanol (20.0 mL) in a reflux apparatus, and stirred until NaOtBu was totally dissolved. Then pyrimidine-2, 4-dithiol (5.0 mmol) was added slowly. The reaction mixture was refluxed for 1 h. Allyl bromide (13.0 mmol) was added and refluxed for another 12 h. The solution was allowed to cool to room temperature, and then water (10.0 mL) was added to quench the reaction. The mixture was extracted with EtOAc (10.0 mL x 3), and the combined organic layer was washed with brine and dried over anhydrous sodium sulfate, filtered and concentrated in vacuum. The crude was purified by flash column chromatography on silica gel to afford pure product 2g: ¹H NMR (400 MHz, CDCl₃) δ 8.12 (d, J = 5.6 Hz, 1 H), 6.81 (d, J = 5.2 Hz, 1 H), 6.05-5.80 (m, 2 H), 5.35-5.32 (m, 1 H), 5.30-5.28 (m, 1 H), 5.20-5.11 (m, 2 H), 3.86-3.80 (m, 4 H); ¹³C {¹H} NMR (100 MHz, CDCl₃) δ 171.1, 169.5, 154.4, 133.4, 132.7, 118.3, 117.7, 114.3, 33.6, 31.9; HRMS m/z (CI) calcd. for $C_{10}H_{12}N_2S_2Na$ (M+Na)⁺ 247.0340, found 247.0337.

ASSOCIATED CONTENT

Supporting Information.

The Supporting Information is available free of charge on the ACS Publications website at <u>http://pubs.acs.org</u>.

¹H and ¹³C NMR spectra of products (PDF).

AUTHOR INFORMATION

Corresponding Author

*stding@mail.buct.edu.cn.

ORCID

Shengtao Ding: 0000-0003-1515-2931 Notes

The authors declare no competing financial interest.

ACKNOWLEDGMENT

This work was financially supported by the National Natural Science Foundation of China (No.21702015), State Key Laboratory of Organic–Inorganic Composites, Beijing University of Chemical Technology (oic-20180101), the Fundamental Research Funds for the Central Universities (buctrc201719).

REFERENCES

(1) For recent reviews, see: (a) Komiyama, T.; Minami Y.; Hiyama, T. Recent Advances in Transition-Metal-Catalyzed Synthetic Transformations of Organosilicon Reagents. ACS Catal. 2017, 7, 631-651. For recent examples, see: (b) Shin, K.; Park, Y.; Baik, M.-H.; Chang, S. Iridium-Catalysed Arylation of C-H Bonds Enabled by Oxidatively Induced Reductive Elimination. Nat. Chem. 2018, 10, 218-224. (c) Shibata, M.; Ito H.; Itami, K. C-H Arylation of Phenanthrene with Trimethylphenylsilane by Pd/o-Chloranil Catalysis: Computational Studies on the Mechanism, Regioselectivity, and Role of <u>o</u>-Chloranil. J. Am. Chem. Soc. **2018**, 140, 2196-2205. (d) Komiyama, T.; Minami, Y.; Furuya Y.; Hiyama, T. Palladium/Copper Dual Catalysis for the Cross-Coupling of Aryl(trialkyl)silanes with Aryl Bromides. Angew. Chem., Int. Ed. 2018, 57, 1987-1990. (e) Xu, P.; Wang, F.; Fan, G.; Xu X.; Tang, P. Hypervalent Iodine(III)-Mediated Oxidative Fluorination of Alkylsilanes by Fluoride Ions. Angew. Chem., Int. Ed. 2017, 56, 1101-1104. (f) Wang, F.; Xu, P.; Cong F.; Tang, P. Silver-Mediated Oxidative Functionalization of Alkylsilanes. Chem. Sci. 2018, 9, DOI: 10.1039/c8sc03730b.

58 59 60

55

56

57

2

3

4

5

6

7

8

9

56

57

58 59

60

(2) For reviews, see: (a) Ramesh, R.; Reddy, D. S. Quest for Novel Chemical Entities through Incorporation of Silicon in Drug Scaffolds. J. Med. Chem. 2018, 61, 3779-3798. (b) Farkas, S. Silperisone: A Centrally Acting Muscle Relaxant. CNS Drug Rev. 2006, 12, 218-235. (c) Fujii, S.; Hashimoto, Y. Progress in the Medicinal Chemistry of Silicon: C/Si Exchange and Beyond. Future Med. Chem. 2017, 9, 485-505. (d) Franz, A. K.; Wilson, S. O. Organosilicon Molecules with Medicinal Applications. J. Med. Chem. 2013, 56, 388-405. (e) Mills, J. S.; Showell, G. A. Exploitation of Silicon Medicinal Chemistry in Drug Discovery. Expert Opin. Investig. Drugs 2004, 13, 1149-1157. (f) Showell, G. A.; Mills, J. S. Chemistry Challenges in Lead Optimization: Silicon Isosteres in Drug Discovery. 10 Drug Discov. Today 2003, 8, 551-556.

11 (3) For Direct Process, pleased refer related explanation in Wik-12 ipedia: https://en.wikipedia.org/wiki/Direct_process. For reviews 13 on alkene hydrosilylation, see: (a) Marciniec, B. Hydrosilylation: 14 A Comprehensive Review on Recent Advances, Springer, Berlin, 2009. (b) Marciniec, B.; Maciejewski, H.; Pawluć P. in Organosil-15 icon Compounds: Experiment, Physico-Chemical Studies and Ap-16 plications, (Eds.: V. Y. Lee), Elsevier, London, 2017, pp. 169-217. (c) 17 Nakajima, Y.; Shimada, S. Hydrosilylation Reaction of Olefins: Re-18 cent Advances and Perspectives. RSC Adv. 2015, 5, 20603-20616. (d) 19 Sun, J.; Deng, L. Cobalt Complex-Catalyzed Hydrosilylation of Al-20 kenes and Alkynes. ACS Catal. 2016, 6, 290-300. (e) Marciniec, B. Catalysis by Transition Metal Complexes of Alkene Silvlation-Re-21 cent Progress and Mechanistic Implications. Coordin. Chem. Rev. 22 2005, 249, 2374-2390. (f) Troegel, D.; Stohrer, J. Recent Advances 23 and Actual Challenges in Late Transition Metal Catalyzed Hydros-24 ilylation of Olefins from an Industrial Point of View. Coordin. 25 Chem. Rev. 2011, 255, 1440-1459. (g) Du, X.; Huang, Z. Advances in 26 Base-Metal-Catalyzed Alkene Hydrosilylation. ACS Catal. 2017, 7, 27 1227-1243. (h) Obligacion, J. V.; Chirik, P. J. Earth-Abundant Transition Metal Catalysts for Alkene Hydrosilylation and Hydrobora-28 tion. Nat. Rev. Chem. 2018, 2, 15-34. 29

(4) For reviews on iridium-catalyzed silylation reactions, see: (a) 30 Marciniec, B.; Kownacki, I. Iridium Complexes in Organic Synthe-31 sis, Wiley-VCH, Weinheim, 2009. (b) Malacea, R.: Poli, R.: 32 Manoury, E. Asymmetric Hydrosilylation, Transfer Hydrogena-33 tion and Hydrogenation of Ketones Catalyzed by Iridium Complexes. Coord. Chem. Rev. 2010, 254, 729-752. (c) Cheng, C.; Hart-34 wig, J. F. Catalytic Silylation of Unactivated C-H Bonds. Chem. Rev. 35 2015, 115, 8946-8975. 36

(5) Please see Chapter 14 in Ref. 4a and references cited therein. 37

(6) Perales, J. B.; Van Vranken, D. L. Thioether-Directed Plati-38 num-Catalyzed Hydrosilylation of Olefins. J. Org. Chem. 2001, 66, 39 7270-7274.

(7) Muchnij, J. A.; Kwaramba, F. B.; Rahaim, R. J. Sterically Di-40 rected Iridium-Catalyzed Hydrosilylation of Alkenes in the Pres-41 ence of Alkynes. Org. Lett. 2014, 16, 1330-1333. 42

(8) Igarashi, M.; Matsumoto, T.; Kobayashi, T.; Sato, K.; Ando, W.; 43 Shimada, S.; Hara, M.; Uchida, H. Ir-Catalyzed Hydrosilylation 44 Reaction of Allyl Acetate with Octakis(dimethylsiloxy)oc-45 tasilsesquioxane and Related Hydrosilanes. J. Organomet. Chem. 46 2014, 752, 141-146.

(9) Reiner, K.; Meister, T. K.; Gigler, P.; Herrmann, W. A.; Kühn, 47 F. E. Mechanistic Insights into the Iridium-Catalyzed Hydrosilyla-48 tion of Allyl Compounds. J. Catal. 2015, 331, 203-209. 49

(10) Srinivas, V.; Nakajima, Y.; Sato, K.; Shimada, S. Iridium-Cata-50 lyzed Hydrosilylation of Sulfur-Containing Olefins. Org. Lett. 51 2018, 20, 12-15.

52 (11) Song, L.-J.; Ding, S.; Wang, Y.; Zhang, X.; Wu, Y.-D.; Sun, J. Ir-Catalyzed Regio- and Stereoselective Hydrosilylation of Internal 53 Thioalkynes: A Combined Experimental and Computational 54 Study. J. Org. Chem. 2016, 81, 6157-6164. 55

(12) Voronkov, M. G.; Vlasova, N. N.; Bolshakova, S. A.; Kirpichenko, S. V. The Catalytic Reactions of Triethyl- and Triethoxy-silane with Unsaturated Sulphides. J. Organomet. Chem. 1980, 190, 335-341.

(13) del Hierro, I.; Sierra, I.; Perez-Quintanilla, D.; Carrillo-Hermosilla, F.; Lopez-Solera, I.; Fajardo, M. Synthesis of adducts from mercury(II) with N and S donor ligands as models of adsorbent materials for the retention of heavy metals X-ray crystal structure bis(3,5-dimethylpyrazol-1-yl)methane(dichloro)mercury(II). of Inorg. Chim. Acta 2003, 355, 347-353.

(14) Guerra, D. L.; Viana, R. R.; Airoldi, C. Adsorption of thorium cation on modified clays MTTZ derivative. J. Hazard. Mater. 2009, 168, 1504-1511.

(15) Obora, Y.; Lin, X.; Nakajima, Y.; Shimada, S.; Sato, K. Method for producing organosilicon compound by hydrosilylation with metallic-element-containing nanoparticles. WO2018131430, 2018. (16) Tucker-Schwartz, A. K.; Farrell, R. A.; Garrell, R. L. Thiol-ene Click Reaction as a General Route to Functional Trialkoxysilanes for Surface Coating Applications. J. Am. Chem. Soc. 2011, 133, 11026-11029.

(17) Zhang, L.; Li, J.; Wang, C. A 3-cyclohexyl thio-1-propyl-triethoxysilane and the preparation method and application. CN105367599, 2016.

(18) Seshadri, T.; Haupt, H. J. Silica-immobilized 2-[(2-(triethoxysily)ethyl)thio]aniline as a selective sorbent for the separation and preconcentration of palladium. Anal. Chem. 1988, 60, 47-52.

(19) Zha, L.; Hao, W.; Cai, M. A diphosphino-functionalized MCM-41-anchored platinum complex: an efficient and reusable catalyst for the hydrosilylation of olefins. J. Chem. Res. 2010, 34, 648-652.

(20) Tonomura, Y.; Kubota, T. Preparation of protected hydroxy group-containing organoxysilanes. JP2005255614A, 2005.

(21) Yu, H.; Yin, C.; Jia, C.; Jin, Y.; Ke, Y.; Liang, X. Evaluation of "Click" Binaphthyl Chiral Stationary Phases by Liquid Chromatography. *Chirality* **2012**, *24*, 391-399.

(22) Hudrlik, P. F.; Abdallah, Y. M.; Hudrlik, A. M. Generation of anionic intermediates by intramolecular nucleophilic attack at silicon. Tetrahedron Lett. 1992, 33, 6747-6750.

(23) Xu, Y.; Bai, Y.; Peng, J.; Li, J.; Xiao, W.; Lai, G. Hydrosilylation of alkenes catalyzed by rhodium with polyethylene glycol-based ionic liquids as ligands. J. Organomet. Chem. 2014, 765, 59-63.

(24) Andrianov, K. A.; Mamedov, A. A.; Volkova, L. M.; Klabunovskii, E. I. Optically active menthoxyethyl- and menthoxypropyl(alkyl)(alkoxy)(chloro)silanes. Izv. Akad. Nauk SSSR, Ser. Khim. 1968, 2, 356-360.

(25) Arkles, B. C.; Pan, Y.; Hollenberg, J. C. Preparation of siliconbased tocopherol derivatives for cosmetics. US20070253925A1, 2007.

(26) Shibayama, W.; Shigaki, S.; Nakajima, M.; Takeda, S.; Wakayama, H.; Sakamoto, R. Composition for forming dry etchable resist underlayer including silicon and having organic group containing aliphatic polycyclic structure, and photolithographic manufacture of semiconductor devices. WO2016009965, 2016.

(27) Kojima, K.; Iwabuchi, S.; Sangane, T.; Miyoshi, K.; Shinomura, K. Preparation and application of UV-curing silicone resins for hard coating agents. Chiba Daigaku Kogakubu Kenkyu Hokoku 1987, 38, 33-39.

(28) Liu, Y.; Deng, L.. Mode of Activation of Cobalt(II) Amides for Catalytic Hydrosilylation of Alkenes with Tertiary Silanes. J. Am. Chem. Soc. 2017, 139, 1798-1801.

(29) Maercker, A.; Stoetzel, R. Carbanion rearrangement through 1, ω -proton shift. 5. Carbanion rearrangements of ω -phenyl- ω -(trimethylsilyl)alkyllithium compounds: intramolecular reactions of benzyltrimethylsilanes with a carbon-lithium bond. Chem. Ber. 1987, 120, 1695-1706.

(30) Bismuto, A.; Cowley, M. J.; Thomas, S. P. Aluminum-Catalyzed Hydroboration of Alkenes. ACS Catal. 2018, 8, 2001-2005.

(31) Carney, J. R.; Dillon, B. R.; Campbell, L.; Thomas, S. P. Manganese-Catalyzed Hydrofunctionalization of Alkenes. *Angew. Chem., Int. Ed.* **2018**, *57*, 10620-10624.

ACS Paragon Plus Environment