

# New chalcone-3-*O*-glycoside derivatives: Synthesis and characterization

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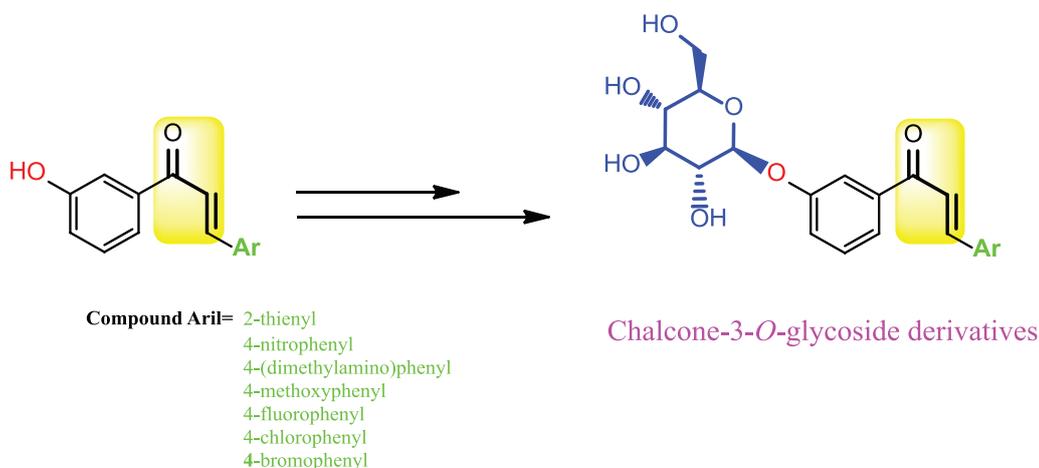
## Abstract

Seven novel carbohydrate conjugates of new chalcone-3-*O*-glycosides were synthesized and characterized. Starting from the substituted 3'-hydroxyarylmethylacetophenone derivatives (chalcones) with  $\alpha$ -acetobromoglucose in anhydrous acetone were synthesized 2,3,4,6-tetra-*O*-acetyl-3'-*O*- $\beta$ -D-glucopyranosyloxychalcones. Deblocking the latter with  $\text{CH}_3\text{ONa}$  in dry methanol results in substituted chalcone-3-*O*-glycosides (3'-*O*- $\beta$ -D-glucopyranosyloxychalcones). The structures of the newly synthesized chalcone-3-*O*-glycosides were characterized based on  $^1\text{H}$  nuclear magnetic resonance,  $^{13}\text{C}$  nuclear magnetic resonance, mass spectroscopy, and Fourier-transform infrared spectroscopy.

## Keywords

chalcone, chalcone-3-*O*-glycosides, characterization, nuclear magnetic resonance, synthesis

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## Introduction

Chalcones, compounds bearing the 1,3-diaryl-2-propen-1-one system, play a vital role in chemical compounds associated with diverse pharmacological activities.<sup>1</sup> Due to their flexible structures, chalcones can bind effectively to many kinds of enzymes or receptors and exhibit a broad spectrum of biological activities such as anticancer,<sup>2</sup> anti-HIV,<sup>3</sup> anti-inflammatory,<sup>4,5</sup> anti-invasive,<sup>6,7</sup> and antibacterial properties.<sup>8</sup>

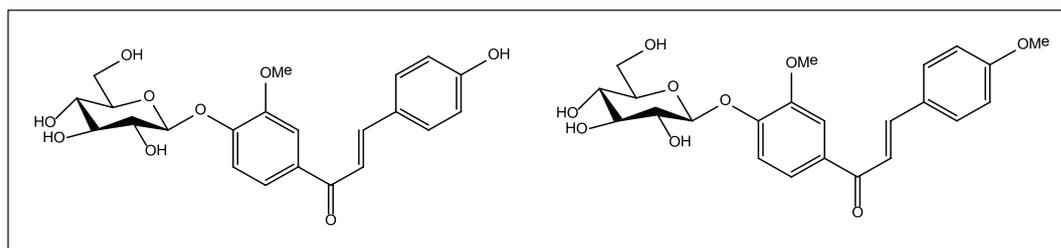
Glycosylation of small biologically active molecules, either of natural or synthetic origin, has an effect on their solubility, and bioactivity. Notably,  $\beta$ -glucosylation improves the drug's targeting to cells as well as their "solubility" in cell

membranes.<sup>9,10</sup> *O*-glycosides are a family of carbohydrates widely found in plants with carbon–oxygen bond formation employing activated sugar.<sup>11,12</sup> Phenolic hydroxyl groups occur in the free state or in combined form as *O*-glycosides in plants where they play many interesting biological activities,

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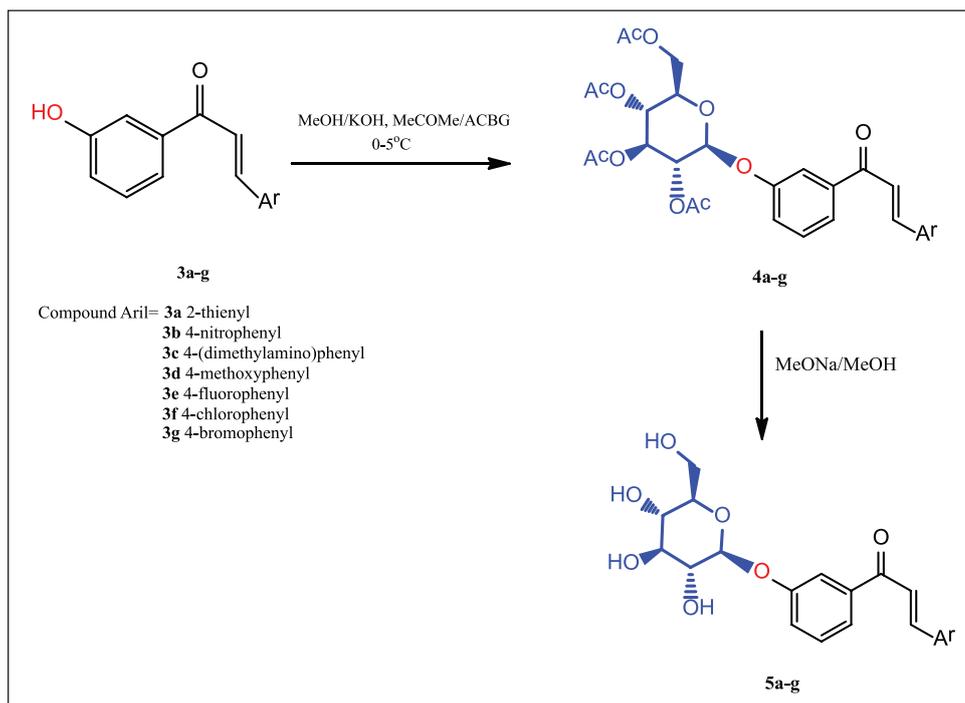
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**Scheme 1.** Structures of isolated chalcone glycosides from *Brassica rapa* L. “hidabeni.”

4'-O-β-D-Glucopyranosyl-4-hydroxy-3'-methoxychalcone (**1**).

4'-O-β-D-Glucopyranosyl-3',4-dimethoxychalcone (**2**).



**Scheme 2.** General route for the synthesis of chalcone-3-O-glycosides (**5a-g**).

such as antitumor activity and inhibitors of the metabolic process.<sup>13</sup>

Much attention has been paid to chalcones as potential biological agents after the discovery of the natural chalcone-*O*-glycosides which demonstrated a broad spectrum of antibiotic, antifungal, and anti-HIV activity.<sup>14</sup> The isolation of some chalcone glycosides (see Scheme 1) from the aerial parts of *Brassica rapa* L. “hidabeni” has been reported.<sup>15</sup> However, to the best of the author’s knowledge, few reports have been dedicated to the synthesis and inhibitory activity of chalcone derivatives containing an *O*-glycoside skeleton.

In this study, preliminary results in the organic synthesis of some new chalcone-3-*O*-glycosides will be reported. Scheme 2 shows the synthetic analysis of the target molecules. Target molecules in this study are chalcones with various sugar substitutions at 3-position.

## Results and discussion

First, substituted chalcones starting from 3-hydroxyacetophenone and different aldehydes were synthesized at room

temperature according to the Claisen–Schmidt condensation method, which is a single-step process and a practical method for the synthesis of substituted chalcone derivatives (Scheme 2).<sup>16</sup>

In the second step, *O*-glucosylation reactions were performed by condensing α-acetobromoglucose (ACBG) and 3-hydroxychalcones (**3a-g**) at 0–5 °C using 5% aqueous potassium hydroxide (KOH) solution in dry acetone under an inert atmosphere.<sup>13</sup> Deacetylation by methanolic MeONa solution gave, after purification, chalcone-3-*O*-glycosides (**5a-g**) in good yield in favor of the β-anomer from 2,3,4,6-tetra-*O*-acetyl-3'-*O*-β-D-glucopyranosyloxybenzylideneacetophenones **4a-g** see (Scheme 2).<sup>17</sup>

The synthesis of chalcone-3-*O*-glycosides was achieved in this second step. The structures of all the compounds were confirmed by nuclear magnetic resonance (NMR), liquid chromatography–mass spectrometry (LC/MS–MS), and Fourier-transform infrared spectroscopy (FTIR; attenuated total reflection (ATR)), spectroscopy. The results were consistent with the predicted structures for all the new chalcone-3-*O*-glycosides, as described in section “Experiment”.

The  $\beta$ -configuration of the glycosidic bond was confirmed from  $^1\text{H}$  NMR and  $^{13}\text{C}$  NMR. In the  $^1\text{H}$  NMR spectrum of **5a–g**, the anomeric proton appeared as a doublet at  $\sim\delta$  5.03 (d,  $\sim 7.6$  Hz) that confirmed the  $\beta$ -orientation of the sugar unit.<sup>18</sup> Characteristic pyranosyl ring protons were located at  $\delta$  3.40–3.96.

In the proton-decoupled  $^{13}\text{C}$  NMR, the anomeric carbon was located at  $\delta \sim 101.00$ , which is consistent with the formation of the  $\beta$ -glucoside **5a–g**. The mass spectrum of 3-((2*E*)-3-(2-thienyl)prop-2-enol-yl)phenyl hexopyranoside (**5a**) also confirmed the proposed structure of the chalcone-3-*O*-glycoside. The molecular ion peak was observed at  $m/z$  393.0825 as  $(\text{M}+\text{H})^+$ , (100; see Electronic Supplementary Information (ESI)).

The structure of 3-((2*E*)-3-(2-thienyl)prop-2-enol-yl)phenyl hexopyranoside (**5a**) was confirmed by the appearance of the  $\text{C}=\text{O}$  band at  $1651\text{ cm}^{-1}$  in the FTIR spectrum. The characteristic bands at  $3343$  and  $1651\text{ cm}^{-1}$  indicated the existence of  $\text{OH}$  and  $\text{C}=\text{O}$  groups, respectively.

In the  $^1\text{H}$  NMR spectra of the synthesized chalcone-3-*O*-glycoside derivatives (**5a–g**), the  $\alpha,\beta$ -protons resonated as doublet doublets (dd) between  $\delta$  7.74 and 7.59, which is characteristic for  $\alpha,\beta$ -unsaturated carbonyl compounds. The olefinic H-atoms of chalcone-3-*O*-glycoside (**5a–g**) were shown to be *trans* based on their  $J$  values of 15.4–15.7 Hz for  $\text{H}_\alpha$  and  $\text{H}_\beta$ .<sup>19</sup>

In the  $^{13}\text{C}$  NMR spectra of the compound **5e**, the characteristic carbon–fluorine signals appear at 131.32/131.29 ( $\text{C}_2''$ ), 115.74/115.52 ( $\text{C}_3''$ ), 165.44/162.95 ( $\text{C}_4''$ ), 131.32/131.29 ( $\text{C}_5''$ ), and 115.74/115.52 ( $\text{C}_6''$ ) (see ESI).<sup>20</sup>

The mass spectra of compounds **5a–g** showed molecular ion peaks at the appropriate  $m/z$  values confirming their molecular mass. In the  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectra of compounds **5a–g**, in particular, anomeric protons and anomeric carbons showed peaks at, respectively,  $\delta_{\text{H}}$  5.03 (1H, d) and  $\delta_{\text{C}}$  101.00 ppm; which are an indication of the  $\beta$ -configuration of the chalcone-3-*O*-glycosides (**5a–g**).

## Conclusion

The synthesis of seven new chalcone-3-*O*-glycosides, which can be used to make more glycosylated chalcone, is reported in this study. The structures of the synthesized chalcone and chalcone-3-*O*-glycoside have been confirmed by NMR, LC/MS–MS, and FTIR (ATR) spectroscopy. Chalcones are important  $\alpha,\beta$ -unsaturated ketones and constitute a class of naturally occurring substances, which are very attractive bioactive starting materials for the synthesis of substituted chalcone-*O*-glycosides.

## Experiment

$^1\text{H}$  and  $^{13}\text{C}$  NMR spectra were recorded on a 400 (100)-MHz Bruker Avance spectrometer in methanol- $d_4$ ;  $\delta$  is given in ppm relative to  $\text{Me}_4\text{Si}$  (tetramethylsilane (TMS)) as an internal standard. The following abbreviations are used: s=singlet, d=doublet, dd=doublet doublet, t=triplet, m=multiplet, bs=broad singlet, bd=broad doublet.

The high-resolution accurate masses were determined using Micromass Quattro LC–MS/MS. The IR spectra were recorded on a Perkin Elmer 1600 Fourier transform infrared (FTIR-ATR) spectrophotometer. Compounds **3a–g** were synthesized by the method described by Narender et al.<sup>21</sup> NMR, MS, IR data, and melting point of the known compounds were consistent with those reported previously.<sup>22–28</sup>

### Synthesis of chalcones (**3a–g**): general procedure

Compounds **3a–g** were synthesized according to the previously reported procedure.<sup>22–28</sup>

### Synthesis of 2,3,4,6-tetra-*O*-acetyl-3'-*O*- $\beta$ -glucopyranosyloxybenzlideneacetophenones (**4a–g**): general procedure

The 2,3,4,6-tetra-*O*-acetyl- $\alpha$ - $\text{D}$ -glucopyranosyl bromide (0.411 g, 1 mmol) was dissolved in dry acetone (20 mL) and cooled to 0–3 °C. To this, a solution of the potassium salt of 3'-hydroxybenzylideneacetophenones (1 mmol) in 5% methanolic KOH (10 mL) was added dropwise under a nitrogen atmosphere. The resulting mixture was stirred at 0–3 °C for 8 h and the reaction was allowed to proceed for an additional 10 h at room temperature. After stirring for 19 h, the resultant solution was partitioned with EtOAc, distilled water and brine, dried over anhydrous  $\text{Na}_2\text{SO}_4$ , and concentrated in vacuo. The residue was purified using silica gel 230–400 mesh eluting with 20% MeOH in  $\text{CHCl}_3$  to obtain compounds **4a–g** as orange syrups.

### The general method for 3'-*O*- $\beta$ -glucopyranosyloxybenzlideneacetophenones **5a–g**

A freshly prepared solution of  $\text{MeONa}$ –MeOH (3 mL, 0.05 M) was added to a solution of 2,3,4,6-tetra-*O*-acetyl-3'-*O*- $\beta$ -glucopyranosyloxybenzlideneacetophenones **4a–g** (0.1 g) in 20 mL dry methanol and kept at room temperature under a nitrogen atmosphere. The progress of the reaction was followed by silica gel thin-layer chromatography (TLC;  $\text{CHCl}_3$ –MeOH, 3:1). The reaction was stopped after 24 h. The mixture was extracted with EtOAc ( $3 \times 10$  mL), the combined organic phases were successively washed with  $\text{H}_2\text{O}$  ( $2 \times 10$  mL), dried over  $\text{Na}_2\text{SO}_4$  and concentrated. The syrupy residue was purified on silica gel using  $\text{CHCl}_3$ –MeOH (9/1 to 7/3, then 6/4, v/v).

3-[(2*E*)-3-(2-thienyl)prop-2-enol-yl]phenyl hexopyranoside (**5a**): Orange oil; yield 98 mg (25%);  $^1\text{H}$  NMR ( $\text{CD}_3\text{OD}$ ):  $\delta$  7.45 (d,  $J=15.6$  Hz, 1H), 7.94 (d,  $J=15.6$  Hz, 1H), 7.75 (bs, 1H), 7.40 (bd,  $J=7.8$  Hz, 1H), 7.49 (t,  $J=7.9$  Hz, 1H), 7.72 (d,  $J=7.8$  Hz, 1H), 7.63 (d,  $J=7.9$  Hz, 1H), 7.16 (dd,  $J=8.2, 4.6$  Hz, 1H), 7.54 (d,  $J=8.2$  Hz, 1H), 5.06 (d,  $J=7.6$  Hz, 1H), 3.60–3.40 (m, 4H), 3.72 (dd,  $J=12.4, 4.1$  Hz, 1H), 3.94 (dd,  $J=12.4, 2.1$  Hz, 1H);  $^{13}\text{C}$  NMR ( $\text{CD}_3\text{OD}$ ):  $\delta$  189.84, 122.13, 137.45, 139.21, 116.97, 158.03, 122.98, 129.60, 122.13, 140.01, 132.36, 128.24, 129.53, 101.00, 73.49, 76.84, 69.95, 76.52, 61.04; MS

(ESI)  $m/z$  for  $C_{19}H_{20}O_7S$   $[M+1]^+$ : 393 (90), 282 (50); IR (ATR) ( $\nu$ ,  $cm^{-1}$ ): 3343, 2962, 1651, 1574, 1251, 1071, 711.

3-[(2E)-3-(4-nitrophenyl)prop-2-enol-yl]phenyl hexopyranoside (**5b**): Orange oil; yield 87 mg (20%);  $^1H$  NMR ( $CD_3OD$ ):  $\delta$  7.83 (d,  $J=15.6$  Hz, 1H), 7.93 (d,  $J=15.6$  Hz, 1H), 7.80 (bs, 1H), 7.40 (dd,  $J=7.6$ , 3.2 Hz, 1H), 7.52 (t,  $J=7.6$  Hz, 1H), 7.84 (m, 1H), 8.00 (d,  $J=7.8$  Hz, 2H), 8.32 (d,  $J=7.8$  Hz, 2H), 5.06 (d,  $J=7.6$  Hz, 1H), 3.60–3.40 (m, 4H), 3.72 (dd,  $J=12.4$ , 4.1 Hz, 1H), 3.94 (dd,  $J=12.4$ , 2.1 Hz, 1H);  $^{13}C$  NMR ( $CD_3OD$ ):  $\delta$  189.57, 123.68, 141.56, 138.83, 115.84, 158.08, 121.73, 129.65, 122.43, 141.12, 129.27, 125.42, 148.59, 125.42, 129.27, 100.97, 74.34, 76.95, 70.22, 76.55, 61.17; MS (ESI)  $m/z$  for  $C_{21}H_{21}NO_9$   $[M+Na]^+$ : 454 (25), 282 (100); IR (ATR) ( $\nu$ ,  $cm^{-1}$ ): 3364, 2926, 1669, 1515, 1337, 1246, 1046, 754.

3-[(2E)-3-(4-dimethylaminophenyl)prop-2-enol-yl]phenyl hexopyranoside (**5c**): Orange oil; yield 94 mg (22%);  $^1H$  NMR ( $CD_3OD$ ):  $\delta$  7.72 (d,  $J=15.4$  Hz, 1H), 7.78 (d,  $J=15.4$  Hz, 1H), 7.42 (s, 1H), 7.34 (dd,  $J=9.4$ , 2.1 Hz, 1H), 7.48 (m, 1H), 7.50 (d,  $J=7.8$  Hz, 1H), 7.64 (d,  $J=7.8$  Hz, 2H), 6.78 (d,  $J=7.8$  Hz, 2H), 3.06 (s, 6H), 5.02 (d,  $J=7.7$  Hz, 1H), 3.60–3.40 (m, 4H), 3.74 (dd,  $J=12.2$ , 3.7 Hz, 1H), 3.94 (dd,  $J=12.2$ , 2.1 Hz, 1H);  $^{13}C$  NMR ( $CD_3OD$ ):  $\delta$  190.70, 122.27, 146.83, 140.11, 116.00, 157.97, 122.27, 129.41, 122.06, 122.27, 130.58, 111.60, 152.65, 111.60, 130.28, 58.09, 101.03, 73.51, 76.88, 70.17, 76.53, 61.09; MS (ESI)  $m/z$  for  $C_{23}H_{27}NO_7$   $[M+1]^+$ : 430 (95), 282 (15); IR (ATR) ( $\nu$ ,  $cm^{-1}$ ): 3376, 2923, 1644, 1571, 1243, 1076, 785.

3-[(2E)-3-(4-methoxyphenyl)prop-2-enol-yl]phenyl hexopyranoside (**5d**): Yellow oil; yield 124 mg (30%);  $^1H$  NMR ( $CD_3OD$ ):  $\delta$  7.60 (d,  $J=15.7$  Hz, 1H), 7.72 (d,  $J=15.7$  Hz, 1H), 7.74 (m, 1H), 7.38 (m, 1H), 7.49 (t,  $J=7.7$  Hz, 1H), 7.74 (m, 1H), 7.73 (d,  $J=7.6$  Hz, 2H), 7.00 (d,  $J=7.6$  Hz, 2H), 3.75 (s, 3H), 5.01 (d,  $J=7.8$  Hz, 1H), 3.60–3.40 (m, 4H), 3.74 (dd,  $J=12.2$ , 4.1 Hz, 1H), 3.94 (dd,  $J=12.3$ , 2.2 Hz, 1H);  $^{13}C$  NMR ( $CD_3OD$ ):  $\delta$  190.56, 123.33, 146.83, 140.14, 139.57, 116.52, 158.00, 122.21, 129.41, 122.60, 127.79, 131.42, 114.13, 162.16, 131.42, 114.13, 54.53, 101.00, 73.50, 76.88, 69.94, 76.52, 61.10; MS (ESI)  $m/z$  for  $C_{22}H_{24}O_8$   $[M+1]^+$ : 417 (100); IR (ATR) ( $\nu$ ,  $cm^{-1}$ ): 3290, 2924, 1656, 1585, 1250, 1022, 793.

3-[(2E)-3-(4-fluorophenyl)prop-2-enol-yl]phenyl hexopyranoside (**5e**): Yellow oil; yield 113 mg (28%);  $^1H$  NMR ( $CD_3OD$ ):  $\delta$  7.70 (d,  $J=15.6$  Hz, 1H), 7.82 (d,  $J=15.6$  Hz, 1H), 7.39 (m, 1H), 7.20 (d,  $J=7.7$  Hz, 1H), 7.48 (t,  $J=7.7$  Hz, 1H), 7.18 (d,  $J=7.8$  Hz, 1H), 7.86–7.75 (m, 4H), 5.03 (d,  $J=7.6$  Hz, 1H), 3.60–3.40 (m, 4H), 3.72 (dd,  $J=12.1$ , 4.0 Hz, 1H), 3.96 (bd,  $J=12.2$  Hz, 1H);  $^{13}C$  NMR ( $CD_3OD$ ):  $\delta$  190.22, 122.69, 143.66, 139.23, 116.47, 158.01, 122.33, 129.58, 122.69, 131.47, 131.32/131.29, 115.74/115.52, 165.44/162.95, 131.32/131.29, 115.74/115.52, 103.98, 74.73, 76.78, 70.70, 76.52, 61.13; MS (ESI)  $m/z$  for  $C_{21}H_{21}FO_7$   $[M+1]^+$ : 405 (100); IR (ATR) ( $\nu$ ,  $cm^{-1}$ ): 3310, 2926, 1672, 1580, 1225, 1045, 790.

3-[(2E)-3-(4-chlorophenyl)prop-2-enol-yl]phenyl hexopyranoside (**5f**): Yellow oil; yield 85 mg (20%);  $^1H$  NMR ( $CD_3OD$ ):  $\delta$  7.40 (d,  $J=15.6$  Hz, 1H), 7.52 (d,  $J=15.6$  Hz, 1H), 7.80 (m, 1H), 7.30 (m, 1H), 7.44 (m, 1H), 7.78 (m, 1H), 7.78 (d,  $J=7.8$  Hz, 2H), 7.46 (d,  $J=7.8$  Hz, 2H), 5.01

(d,  $J=7.7$  Hz, 1H), 3.60–3.40 (m, 4H), 3.74 (dd,  $J=12.3$ , 4.1 Hz, 1H), 3.94 (bd,  $J=12.3$  Hz, 1H);  $^{13}C$  NMR ( $CD_3OD$ ):  $\delta$  190.09, 122.33, 143.33, 139.16, 116.16, 158.05, 122.20, 129.88, 122.33, 133.61, 129.59, 129.38, 136.12, 129.38, 129.88, 100.99, 73.50, 76.91, 69.98, 76.54, 61.13; MS (ESI)  $m/z$  for  $C_{21}H_{21}ClO_7$   $[M+1]^+$ : 421 (90); IR (ATR) ( $\nu$ ,  $cm^{-1}$ ): 3348, 2919, 1658, 1579, 1245, 1010, 822, 782.

3-[(2E)-3-(4-bromophenyl)prop-2-enol-yl]phenyl hexopyranoside (**5g**): Yellow oil; yield 116 mg (25%);  $^1H$  NMR ( $CD_3OD$ ):  $\delta$  7.45 (d,  $J=15.4$  Hz, 1H), 7.48 (d,  $J=15.4$  Hz, 1H), 7.74 (bs, 1H), 7.33 (m, 1H), 7.40 (m, 1H), 7.78 (m, 1H), 7.60 (d,  $J=7.6$  Hz, 2H), 7.69 (d,  $J=7.6$  Hz, 2H), 5.03 (d,  $J=7.8$  Hz, 1H), 3.60–3.40 (m, 4H), 3.72 (dd,  $J=12.1$ , 3.9 Hz, 1H), 3.95 (dd,  $J=12.1$ , 1.3 Hz, 1H);  $^{13}C$  NMR ( $CD_3OD$ ):  $\delta$  190.09, 124.40, 143.42, 139.13, 116.48, 158.02, 122.24, 129.60, 122.73, 133.96, 129.60, 131.58, 124.42, 129.60, 131.58, 100.97, 74.19, 76.89, 69.97, 76.53, 61.13; MS (ESI)  $m/z$  for  $C_{21}H_{21}BrO_7$   $[M]^+$ : 465 (95); IR (ATR) ( $\nu$ ,  $cm^{-1}$ ): 3362, 2921, 1662, 1579, 1249, 1069, 776.

### Declaration of conflicting interests

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### Supplemental material

Supplemental material for this article is available online.

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