Concise total synthesis of (\pm) -aspidospermidine *via* an oxidative Hosomi–Sakurai process[†]

Cyrille Sabot, Kimiaka C. Guérard and Sylvain Canesi*

Received (in College Park, MD, USA) 19th February 2009, Accepted 14th April 2009 First published as an Advance Article on the web 27th April 2009 DOI: 10.1039/b903530c

A concise total synthesis of (\pm) -aspidospermidine has been achieved in twelve steps from inexpensive 4-ethylphenol, based on an oxidative Hosomi–Sakurai reaction *via* the concept of "aromatic ring umpolung".

Electron-rich aromatic compounds such as phenols, anilines, and their derivatives normally express nucleophilic reactivity. However, oxidative activation^{1–3} can convert these aromatics into reactive electrophilic intermediates, which may be intercepted with appropriate nucleophiles in synthetically useful yields. Thus, oxidative attack of a phenol⁴ with iodobenzene diacetate (DIB), an environmentally benign and inexpensive reagent, in the presence of allyltrimethylsilane promotes an oxidative Hosomi–Sakurai process⁵ that produces products **3** in 37–84% yield.^{5d} The erstwhile electron-rich aromatic substrate expresses electrophilic character in this reaction. The overall transformation may thus be thought of as involving "aromatic ring umpolung",⁶ Fig. 1.

This concept provides new strategic opportunities in synthetic chemistry. Indeed, this method has already allowed us to develop an innovative method directly applicable to the expeditious total syntheses of natural products.⁷ In this communication, we illustrate a new use of the concept by a direct application of a bimolecular oxidative Hosomi–Sakurai condition in connection with a fully stereocontrolled total synthesis of (\pm) -aspidospermidine, **4**. Compound **4** belongs to the family of aspidosperma alkaloids (Fig. 2).⁸ The complex architecture and biological activities of these products⁹ have



Fig. 1 Aromatic ring umpolung.



Fig. 2 Aspidosperma alkaloid family.

elicited substantial interest in the synthetic arena. Aspidospermidine and aspidospermine contain a pentacyclic ring system and five asymmetric centers. The usual strategy used to produce these compounds is generally based on the formation of a key intermediate **6**, first described by Stork and Dolfini.¹⁰ Since then, many other syntheses or formal syntheses have been reported.¹¹

Our approach, based on aromatic ring umpolung, starts with a polysubstituted phenol 7.¹² This compound can be rapidly converted into the corresponding dienone 8 via a bimolecular oxidative Hosomi–Sakurai process in 56% yield by DIB oxidation in the presence of allyltrimethysilane and a perfluorinated solvent such as trifluoroethanol (TFE) or hexafluoroisopropanol (HFIP). The use of a polysubstituted phenol such as 7, containing removable directing groups, is very important in order to force the subsequent allyltrimethyl-silane to locate exclusively at position 4. The corresponding dienone 8 is treated under hydroboration conditions to produce alcohol 9 in 72% yield. The latter is activated with mesyl chloride to generate compound 10 in 97% yield, Scheme 1.

Compound 10 is thus produced rapidly from 7 *via* the oxidative allylation strategy. At this point the nitrogen moiety is efficiently introduced by an S_N^2 reaction between 10 and the corresponding anion of 11^{13} to afford the sulfonamide 12 in 89% yield. This product contains protected secondary amine and alcohol groups which represents a good precursor of the main tricyclic core of aspidospermidine. The nosylamide or Fukuyama's sulfonamide¹⁴ was chosen because it is known to be easily cleaved under mild reaction conditions, Scheme 2.

Indeed, the elaboration of aspidospermidine required a method for the stereoselective construction of the main



Scheme 1 Oxidative Hosomi-Sakurai reaction.



Scheme 2 Alkylation of sulfonamide 11.

Laboratoire de Méthodologie et Synthèse de Produits Naturels, Université du Québec à Montréal, C.P. 8888, Succ. Centre-Ville, Montréal, Québec, Canada H3C 3P8.

E-mail: canesi.sylvain@uqam.ca

[†] Electronic supplementary information (ESI) available: Experimental procedures and characterization data for all new compounds along with copies of ¹H and ¹³C NMR spectra. See DOI: 10.1039/b903530c

tricyclic core. We envisaged, by means of Fukuyama's deprotection, generating the free secondary amine, which could cyclize (6-exo-trig), to produce the first heterocycle. Intriguingly, addition of thiophenolate, under these conditions, leads to compound **16** *via* a substitution of the bromide by a thioether. This transformation, occurring in 85% yield, takes place *via* four successive distinct steps: (1) nucleophilic aromatic substitution (S_NAr) deprotection leads to the corresponding free amine **13**; (2) addition of thiophenol leads to compound **14**; (3) substitution of the bromide by an S_N2 process generates **15**; and (4) a final retro-Michael reaction produces **16**. At this point, the addition of the free amine on the dienone is not observed, Scheme 3.

Treatment of compound **16** with TBAF produces the free alcohol and bicyclic ring **19** in 87% yield. Once again, different transformations occurred in the same pot: (1) desilylation of the TBDMS moiety produces **17**; (2) the free amine reacts *via* a Michael process to afford the bicyclic compound **18**; (3) desilylation of the TMS group occurs to produce **19**, Scheme 4.

The action of mesyl chloride on compound **19** results in the formation of the corresponding chloride **21** in 79% yield. Compound **21** is obtained probably due to the formation of the corresponding aziridinium **20** in the medium, which is opened by nucleophilic attack by the released chloride. Treatment of **21** with a base such as potassium *tert*-butoxide leads to the desired tricyclic compound **22** in 84% yield. At this point, treatment of compound **22** with RANEY[®] nickel in ethanol produces **6** in 85% yield. This sequence of reactions, including a Fukuyama and Michael–retro-Michael tandem process, offers an efficient access to key intermediate **6**, known as formal synthesis of aspidospermine, aspidospermidine and quebrachamine,^{11m} Scheme 5.



Scheme 3 Fukuyama and Michael-retro-Michael tandem process.



Scheme 4 Formation of the bicyclic core 19.



Scheme 5 Formal total syntheses of (\pm) -aspidospermine, (\pm) -aspidospermidine and (\pm) -quebrachamine.



Scheme 6 Synthesis of (\pm) -aspidospermidine.

To complete the total synthesis, the known tricycle **6** is converted to (\pm) -aspidospermidine *via* a Fischer indole synthesis,¹⁵ as demonstrated first by Stork and Dolfini.¹⁰ Indeed, treatment of compound **6** with phenylhydrazine at reflux leads to the hydrazone **23**, which is converted to imine **24** in acetic acid. The latter is reduced in the same pot by LiAlH₄ to produce (\pm) -aspidospermidine, **4**, in 43% yield overall, Scheme 6.

In summary, a concise and fully diastereoselective synthesis of (\pm) -aspidospermidine is accomplished in ten steps from phenol 7 in 5.4% yield overall. This novel synthesis is based on a first application of a bimolecular oxidative Hosomi–Sakurai reaction. In addition, an efficient method to produce the bicyclic ring of this compound *via* a Fukuyama–Michael tandem process has been developed. Formal total syntheses of (\pm) -aspidospermine and (\pm) -quebrachamine have also been presented. This work demonstrates the potential of "aromatic ring umpolung", further applications of which are under study in our laboratories.

The authors thank the Donors of the Petroleum Research Fund (ACS PRF), administered by the American Chemical Society, for support of this research. We are also very grateful to the Natural Sciences and Engineering Research Council of Canada (NSERC) and to the provincial government of Quebec (FQRNT).

Notes and references

 (a) Y. Tamura, T. Yakura, J. Haruta and Y. Kita, J. Org. Chem., 1987, 52, 3927; (b) Y. Kita, H. Tohma, K. Hatanaka, T. Takada, S. Fujita, S. Mitoh, H. Sakurai and S. Oka, J. Am. Chem. Soc., 1994, 116, 3684; (c) Y. Kita, T. Takada, M. Gyoten, H. Tohma, M. H. Zenk and J. Eichhorn, J. Org. Chem., 1996, 61, 5854; (d) Y. Kita, M. Gyoten, M. Ohtsubo, H. Tohma and T. Takada, *Chem. Commun.*, 1996, 1481; (e) T. Takada, M. Arisawa, M. Gyoten, R. Hamada, H. Tohma and Y. Kita, J. Org. Chem., 1998, 63, 7698; (f) M. Arisawa, S. Utsumi, M. Nakajima, N. G. Ramesh, H. Tohma and Y. Kita, Chem. Commun., 1999, Published on 27 April 2009. Downloaded by Monash University on 29/06/2013 14:49:44.

469; (g) T. Dohi, A. Maruyama, N. Takenaga, K. Senami, Y. Minamitsuji, H. Fujioka, S. B. Caemmerer and Y. Kita, *Angew. Chem.*, *Int. Ed.*, 2008, **47**, 3787.

- 2 (a) A. Pelter and R. A. Drake, Tetrahedron Lett., 1988, 29, 4181;
 (b) S. Quideau, M. A. Looney and L. Pouységu, J. Org. Chem., 1998, 63, 9597; (c) A. Ozanne-Beaudenon and S. Quideau, Angew. Chem., Int. Ed., 2005, 44, 7065; (d) M. A. Ciufolini, S. Canesi, M. Ousmer and N. A. Braun, Tetrahedron, 2006, 62, 5318;
 (e) M. A. Ciufolini, N. A. Braun, S. Canesi, M. Ousmer, J. Chang and D. Chai, Synthesis, 2007, 24, 3759; (f) D. Bérard, A. Jean and S. Canesi, Tetrahedron Lett., 2007, 48, 8238;
 (g) D. Bérard, L. Racicot, C. Sabot and S. Canesi, Synlett, 2008, 1076; (h) L. Pouységu, S. Chassaing, D. Dejugnac, A. M. Lamidey, K. Miqueu, J. M. Sotiropoulos and S. Quideau, Angew. Chem., Int. Ed., 2008, 47, 3552; (i) D. Bérard, M. A. Giroux, L. Racicot, C. Sabot and S. Canesi, Tetrahedron, 2008, 64, 7537.
- 3 (a) Akai, N. Kawashita, N. Morita, Y. Nakamura, K. Iio and Y. Kita, *Heterocycles*, 2002, 58, 75; (b) A. Jean, J. Cantat, D. Bérard, D. Bouchu and S. Canesi, *Org. Lett.*, 2007, 9, 2553.
 4 (a) B. D. Gates, P. Dalidowicz, A. Tebben, S. Wang and
- J. S. Swenton, J. Org. Chem., 1992, 57, 2135; (b) N. A. Braun, M. A. Ciufolini, K. Peters and E. M. Peters, Tetrahedron Lett., 1998, 39, 4667; (c) N. A. Braun, J. Brav, M. Ousmer, K. Peters, E. M. Peters, D. Bouchu and M. A. Ciufolini, J. Org. Chem., 2000, 65, 4397; (d) G. Scheffler, H. Seike and E. J. Sorensen, Angew. Chem., Int. Ed., 2000, 39, 4593; (e) M. Ousmer, N. A. Braun, C. Bavoux, M. Perrin and M. A. Ciufolini, J. Am. Chem. Soc., 2001, 123, 7534; (f) S. Quideau, in Modern Arene Chemistry, ed. D. Astruc, Wiley-VCH, Weinheim, Germany, 2002, p. 539; (g) S. Canesi, P. Belmont, D. Bouchu, L. Rousset and M. A. Ciufolini, Tetrahedron Lett., 2002, 43, 5193; (h) I. Drutu, J. T. Njardarson and J. L. Wood, Org. Lett., 2002, 4, 493; (i) S. Canesi, D. Bouchu and M. A. Ciufolini, Angew. Chem., Int. Ed., 2004, **43**, 4336; (j) S. Quideau, L. Pouységu and D. Deffieux, Curr. Org. Chem., 2004, **8**, 113; (k) S. Canesi, D. Bouchu and M. A. Ciufolini, Org. Lett., 2005, 7, 175; (1) S. Quideau, L. Pouységu and D. Deffieux, Synlett, 2008, 467; (m) H. Liang and M. A. Ciufolini, J. Org. Chem., 2008, 73, 4299.
- 5 (a) K. C. Nicolaou, D. J. Edmonds, A. Li and G. S. Tria, Angew. Chem., 2007, 119, 4016; (b) S. Quideau, M. A. Looney and L. Pouységu, Org. Lett., 1999, 1, 1651; (c) S. Quideau, L. Pouységu, M. Oxoby and M. A. Looney, Tetrahedron, 2001, 57, 319; (d) C. Sabot, B. Commare, S. Nahi, M. A. Duceppe, K. C. Guérard and S. Canesi, Synlett, 2008, 3226.
- 6 (a) The concept of "umpolung" chemistry was discovered by E. J. Corey and D. Seebach: D. Seebach and E. J. Corey, J. Org. Chem.,

1975, **40**, 231; (b) D. Seebach, Angew. Chem., Int. Ed. Engl., 1979, **18**, 239.

- 7 (a) C. Sabot, D. Bérard and S. Canesi, Org. Lett., 2008, 10, 4629;
 (b) K. C. Guérard, C. Sabot, L. Racicot and S. Canesi, J. Org. Chem., 2009, 74, 2039.
- 8 (a) G. A. Cordell, in *The Alkaloids*, ed. R. H. F. Manske and R. G. A. Rodrigo, Academic Press, New York, 1979, vol. 17, p. 199; (b) J. E. Saxton, *Nat. Prod. Rep.*, 1993, **10**, 349; J. E. Saxton, *Nat. Prod. Rep.*, 1994, **11**, 493.
- 9 Antitumor Bisindole Alkaloids from Carentheus roseus, in The Alkaloids, ed. A. Brossi and M. Suffness, Academic Press, San Diego, 1990, vol. 37.
- 10 G. Stork and J. E. Dolfini, J. Am. Chem. Soc., 1963, 85, 2872.
- 11 (a) O. Callaghan, C. Lampard, A. R. Kennedy and J. A. Murphy, J. Chem. Soc., Perkin Trans. 1, 1999, 995; (b) M. A. Toczko and C. H. Heathcock, J. Org. Chem., 2000, 65, 2642; (c) S. A. Kozmin, T. Iwama, Y. Huang and V. H. Rawal, J. Am. Chem. Soc., 2002, 124, 4628; (d) J. P. Marino, M. B. Rubio, G. Cao and A. de Dios, J. Am. Chem. Soc., 2002, 124, 13398; (e) Y. Fukuda, M. Shindo and K. Shishido, Org. Lett., 2003, 5, 749; (f) D. Gnecco, E. Vazquez, A. Galindo, J. L. Teran, L. Orea, S. Bernes and R. G. Enriquez, ARKIVOC, 2003, 185; (g) H. Tanino, K. Fukuishi, M. Ushiyama and K. Okada, Tetrahedron, 2004, 60, 3273; (h) M. G. D. Banwell and W. Lupton, Org. Biomol. Chem., 2005, 3, 213; (i) M. G. Banwell, D. W. Lupton and A. C. Willis, Aust. J. Chem., 2005, 58, 722; (j) R. Iyengar, K. Schildknegt, M. Morton and J. Aubé, J. Org. Chem., 2005, 70, 10645; (k) L. A. Sharp and S. Z. Zard, Org. Lett., 2006, 8, 831; (1) W. H. Pearsona and A. Aponick, Org. Lett., 2006, 8, 1661; (m) I. Coldham, A. J. M. Burrell, L. E. White, H. Adams and N. Oram, Angew. Chem., Int. Ed., 2007, 46, 6159; (n) A. C. Callier-Dublanchet, J. Cassayre, F. Gagosz, B. Quiclet-Sire, L. A. Sharp and S. Z. Zard, Tetrahedron, 2008, 64, 4803; (o) T. Ishikawa, K. Kudo, K. Kuroyabu, S. Uchida, T. Kudoh and S. Saito, J. Org. Chem., 2008, 73, 7498; (p) M. Suzuki, Y. Kawamoto, T. Sakai, Y. Yamamoto and K. Tomioka, Org. Lett., 2009, 11, 653; (q) A. J. M. Burrell, I. Coldham, L. Watson, N. Oram, C. D. Pilgram and N. G. Martin, J. Org. Chem., 2009, 74, 2290.
- 12 Details of the preparation of 7 are provided as ESI[†]. This compound has been prepared in two steps from inexpensive 4-ethyl phenol in 86% overall yield.
- 13 Details of the preparation of **11** are provided as ESI[†]. This compound has been prepared in two steps from inexpensive 2-aminoethanol in 89% overall yield.
- 14 Review: T. Kan and T. Fukuyama, Chem. Commun., 2004, 105.
- 15 E. Fischer and F. Jourdan, Ber. Dtsch. Chem. Ges., 1883, 16, 2241.