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A Convenient Synthesis of (\pm) -4-Prenylpterocarpin

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Coupling of 7-methoxy-8-(3-methyl-2-butenyl)-2H-1-benzopyran (4a) with 2-chloromercurio-4,5-methylenedioxyphenol (5) yields (\pm) -6a,12a-cis-dihydro-3-methoxy-4-(3-methyl-2-butenyl)-6H-[1,3]dioxolo[5,6]benzofuro[3,2-c][1]benzopyran (6; \pm -4-prenylpterocarpin) in an efficient manner. An improved procedure for the preparation of chromenes is reported.

Pterocarpans are an important group of naturally occurring products. Many of these compounds act as phytoalexins. For example, phaseollidin and erybraedins A, B and C³ possess relevant antibiotic activities. Further, Nakanishi and co-workers demonstrated that cabenegrins A-I and A-II are the active components of a brazilian folk medicine used against snake venoms. These compounds are active in dogs against the venom of Bothrops atrox. More recently researchers at this center demonstrated that wedelolactone, a closely related isoflavonoid, is the main active component of Eclipta prostata, a plant also used in Brazil as venom neutralizer. 5

A common structural feature of many relevant pterocarpans is the presence of a prenyl substituent which seems essential for their biological activity.

The most efficient approaches for the synthesis of pterocarpans have used chromenes as the key intermediates.⁶ However, the obtention of pterocarpans with prenyl groups in the A-ring, remains problematic,⁷ as the necessary prenylated chromenes have not been reported.

As part of our program in the synthesis of isoflavonoids, we now wish to report the synthesis of 4-prenylpterocarpin 6 where a prenylated chromene, 4a, is the key intermediate. This latter compound was obtained from the resorcinol derivative 1, as shown in Scheme 1. Thus, compound 1 was regioselectively lithiated with butyllithium in refluxing cyclohexane, followed by trapping the formed intermediate by the addition of prenyl bromide in diethylether under reflux. The tetrahydropyranyl group (THP) was removed in situ by addition of 5% aqueous oxalic acid giving, after chromatography on silica gel, the prenylated resorcinol 2a in 75% yield. The high regioselectivity of this process was confirmed by the complete absence of signals in the ¹H NMR and ¹³C NMR spectra that could be attributed to isomers.

The transformation of 2a into the chromene 4a is based in the two-step sequence described by Schuda and Phillips. Deven though for simple chromenes the yields reported are acceptable, if the brevity of the procedure is considered, we felt that for the prenylated intermediates the procedure should be further developed. Thus, we have found that the use of 3-iodopropional dehyde dimethyl acetal as the alkylation agent (instead of 2-[2-chloroethyl]-1,3-dioxolane) leads to the desired ethers in excellent yields and the time of the reaction is shortened from 3-4 days to only 15 hours. Further, the use of dioxane in the cyclization step (instead of benzene), results in the

1. KOH/THF r.t., 30 min OMe 2. 1 OMe reflux, 15h 85-95% 2 a-d 2 a-d 2 a-d A a-d R10 OMe OMe OMe A a-d

2-4	R ¹	R²	Yield (%)	
			3ª	4 ^a
a	Me	CH ₂ CHC(CH ₃) ₂	95	80
b	Me	CH ₂ CHCH ₂	92	75
c	Bn	H Ž	90	85 ^b 75 ^{b, c}
d	Me	Н	85	75 ^{b, c}

^a Yields after purification.

b Chromenes 4c and 4d were reported by Schuda¹⁰ in 50 and 44% yield, respectively. The use of dioxane does not change the high degree of regioselectivity demonstrated in the cyclization step.

Attempt to cyclize the ethyleneglycol acetal analog of 3d in dioxane led to chromene 4d in only 40% yield.

Scheme 2

Scheme 1

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formation of chromenes in much higher yields and shorter reaction times are observed. A summary of our results is shown in Scheme 2.

Chemoselective coupling of 4a with o-(chloromercurio)phenol 5 under the conditions described by Horino and Inoue¹¹ led to (\pm) -4-prenylpterocarpin 6 almost exclusively. While this work was under way, Iyer and co-workers, ¹² demonstrated that bischromenes having two conjugated double bonds undergo chemoselective Heck arylation in the less sterically hindered olefin. However, the reported yields are much lower than in our case where the second olefn is not conjugated. Work is now in progress in our laboratory to evaluate the scope of this procedure.

Dioxane, $\rm Et_2O$ and cyclohexane were distilled from Na/benzophenone. Prenyl bromide was distilled from $\rm CaH_2$ prior to use. Acetone was dried ($\rm K_2CO_3$), distilled and stored over 4 Å molecular sieves. p-TsOH was dried for 4 h, in vacuo ~ 1 Torr) prior to use. 2-Chloromercurio-4,5-methylenedioxyphenol (5), 6a 1-methoxy-3-(2-tetrahydropyranyloxy)benzene (1), 13 3-benzyloxyphenol (2c) 14 and 3-iodopropionaldehyde dimethyl acetal 15 were prepared according to literature procedures.

¹H NMR spectra were recorded on a Varian Gemini-200 (200 MHz) instrument using TMS as standard and CDCl₃ as solvent. ¹³C NMR spectra were obtained at 50 MHz. The coupling constants (*J*) are in hertz (Hz). Mass spectra were recorded on a Micromass MM 12 F and a VG Autospec spectrometer.

3-Methoxy-2-(3-methyl-2-butenyl)phenol (2a); Typical Procedure:

To a cooled (0 °C) solution of 1-methoxy-3-(2-tetrahydropyranyloxy)benzene (1; 300 mg, 1.48 mmol), in cyclohexane (1.0 mL), BuLi (1.35 mL, 2.16 mmol, 1.6 M) was added under N_2 atmosphere. The mixture was warmed to r. t., refluxed for 2 h, and cooled (r. t.). After this time a solution of prenyl bromide (0.48 mL, 3.92 mmol) was added and the resulting mixture refluxed for 2 h. The reaction was quenched by addition of sat. aq NH₄Cl (20 mL), and extracted with Et₂O (3 × 30 mL). The organic layer was washed with 10 % solution of NaOH (3 × 30 mL), brine, H₂O, dried (Na₂SO₄), filtered and concentrated. The crude product was immediately treated with 5 % aq. oxalic acid solution (1.0 mL) in MeOH (10 mL) over 1 h (r. t.). Solid NaHCO₃ was added and the mixture was extracted with EtOAc. The organic layer was dried (Na₂SO₄), filtered and concentrated. The resulting product was purified by column chromatography on silica gel giving 2a as a yellow oil (208 mg, 75 %).

IR (neat): v = 3546, 2924; 1220; 787 cm⁻¹.

¹H NMR: δ = 7.1 (t, 1 H, J = 8); 6.5 (d, 2 H, J = 8); 5.4 (br, 1 H); 5.3 (m, 1 H); 3.8 (s, 3 H); 3.4 (d, 2 H, J = 7); 1.8 (s, 3 H); 1.7 (s, 3 H). ¹³C NMR: δ = 17.6 (q); 22.1 (t); 25.6 (q), 55.6 (q); 103.1 (d); 108.7 (d); 115.3 (s); 122.0 (d); 126.9 (d); 133.7 (s); 155.1 (s); 157.9 (s). MS (70 eV): m/z (%) = 192 (M⁺, 58); 175 (30); 137 (100); 77 (30). HRMS (70 eV): m/z, $C_{12}H_{16}O_{2}$, calc.: 192.1150; found: 192.1184. 2-Allyl-3-methoxyphenol (2b):

This compound was obtained in similar manner to 2a in 80 % yield. IR (neat): v = 3597; 2985; 1587; 1449 cm⁻¹.

¹H NMR: $\delta = 7.0$ (t, 1 H, J = 8); 6.5 (d, 2 H, J = 8); 5.9 (m, 1 H); 5.1 (m, 2 H); 3.8 (s, 3 H); 3.4 (d, 2 H, J = 7).

MS (70 eV): m/z (%) = 164 (M⁺, 100); 149 (30); 135 (30); 107 (30).

3-[3-Methoxy-2-(3-methyl-2-butenyl)phenoxylpropionaldehyde Dimethyl Acetal (3a); Typical Procedure:

A mixture of KOH (112.2 mg, 2.0 mmol) and 3-methoxy-2-(3-methyl-2-butenyl)phenol (2a; 325 mg, 1.71 mmol) in THF (20 mL), was stirred for 30 min at r.t. Then 3-iodopropionaldehyde dimethyl acetal (470 mg, 2.0 mmol) in THF (20 mL) was added slowly and the resulting mixture refluxed for 15 h. The reaction was cooled, washed with aq. 10 % NaOH (3×20 mL), brine and H₂O. The organic layer

was dried (Na₂SO₄), filtered and concentrated. This crude product was purified by distillation in a bulb-to-bulb apparatus, yielding 3a as a light yellow oil (473 mg, 95%).

IR (neat): v = 2920; 1590; 925; 725 cm⁻¹.

¹H NMR: δ = 7.1 (t, 1 H, J = 8); 6.5 (d, 2 H, J = 8); 5.2 (m, 1 H); 4.7 (t, 1 H, J = 6); 4.0 (t, 2 H, J = 6); 3.8 (s, 3 H); 3.4 (br, 8 H); 2.1 (q, 2 H, J = 6); 1.8 (s, 3 H); 1.7 (s, 3 H).

¹³C NMR: δ = 17.6 (q); 22.1 (t); 25.6 (q); 32.0 (t); 53.1 (q); 55.5 (q); 64.0 (t); 102.0 (d); 103.6 (d); 104.4 (d); 118.2 (s); 123.0 (d); 126.5 (d); 130.4 (s); 157.1 (s); 158.0 (s).

MS (70 eV): m/z (%) = 294 (M⁺, 1.6); 262 (60); 230 (75); 215 (60); 163 (60); 75 (70); 71 (100).

HRMS (70 eV): m/z, $C_{17}H_{26}O_4$, calc.: 294.1831; found: 294.1833.

In a similar procedure 3b, 3c and 3d were obtained from the corresponding phenols 2:

3-(2-Allyl-3-methoxyphenoxy)propionaldehyde Dimethyl Acetal (3b):

IR (neat): v = 3063; 2920; 1645; 1593 cm⁻¹.

¹H NMR: δ = 7.2 (t, 1 H, J = 8); 6.5 (d, 2 H, J = 8); 6.0 (m, 1 H); 4.9 (m, 2 H); 4.7 (t, 1 H, J = 6); 4.0 (t, 2 H, J = 6); 3.8 (s, 3 H); 3.4 (dt, 2 H, J = 1.3, 6.2); 3.3 (s, 6 H); 2.1 (q, 2 H, J = 6).

MS (70 eV): m/z (%) = 234 (20); 203 (40); 161 (22); 75 (100).

3-(3-Benzyloxyphenoxy)propionaldehyde Dimethyl Acetal (3c): IR (neat): v = 3050; 2930; 1610 cm⁻¹.

¹H NMR: δ = 7.4 (m, 5 H); 7.2 (t, 1 H, J = 8); 6.6 (m, 3 H); 5.0 (s, 2 H); 4.6 (t, 1 H, J = 5.5); 4.0 (t, 2 H, J = 5.5); 3.4 (s, 3 H); 2.1 (q, 2 H, J = 5.5).

MS (70 eV): m/z (%) = 302 (M⁺, 23); 91 (100); 75 (24); 71 (30).

3-(3-Methoxyphenoxy) propional dehyde Dimethyl Acetal (3d):

IR (neat): v = 2930; 2820; 1595 cm⁻¹.

¹H NMR: $\delta = 7.2$ (t, 1 H, J = 8); 6.5 (m, 3 H); 4.6 (t, 1 H, J = 6); 4.0 (t, 2 H, J = 6); 3.8 (s, 3 H); 3.4 (s, 6 H); 2.1 (q, 2 H, J = 6). MS (70 eV): m/z = 226 (M⁺, 20); 161 (92); 75 (100).

7-Methoxy-8-(3-methyl-2-butenyl)-2*H*-1-benzopyran (4a); Typical Procedure:

To a solution of p-TsOH (10 mg) in dioxane (10 mL), 3-[3-meth-oxy-2-(3-methyl-2-butenyl)phenoxy]propionaldehyde dimethyl acetal (3a; 200 mg) in dioxane (10 mL) was added. The mixture was refluxed for 6 h under N₂. After this time 10% aq NaOH (10 mL) was added and the mixture was partitioned with Et₂O, washed with NaOH (3 × 10 mL), brine, H₂O and dried (Na₂SO₄). After concentration and distillation in a bulb-to-bulb apparatus compound 4a was obtained as a light yellow oil (125 mg, 80%).

IR (neat): v = 2900; 2830; 1635; 1575 cm⁻¹.

¹H NMR: $\delta = 6.8$ (d, 1 H, J = 8); 6.4 (d, 1 H, J = 8); 6.4 (dt, 1 H, J = 10 and 2); 5.7 (dt, 1 H, J = 10, and 3.8); 5.2 (m, 1 H), 4.7 (dd, 2 H, J = 3.8, and 2); 3.8 (s, 3 H); 3.3 (d, 2 H, J = 7); 1.8 (s, 3 H); 1.7 (s, 3 H).

¹³C NMR: δ = 17.6 (q); 22.0 (t); 25.7 (q); 55.5 (q); 65.3 (t); 103.1 (d); 116.0 (s); 117.4 (s); 119.0 (d); 122.5 (d); 124.1 (d); 124.6 (d); 131.1 (s); 152.5 (s); 158.0 (s).

MS (70 eV): m/z (%) = 230 (M⁺, 90); 215 (100); 187 (55); 115 (30). HRMS (70 eV): m/z, $C_{15}H_{18}O_2$, calc.: 230.1307; found: 230.1302.

In a similar procedure 4b, 4c and 4d were obtained from the corresponding ethers 3:

8-Allyl-7-methoxy-2H-1-benzopyran (4b):

IR (neat): $v = 2858, 1640, 1610, 1590 \text{ cm}^{-1}$.

¹H NMR: $\delta = 6.8$ (d, 1 H, J = 8); 6.4 (d, 1 H, J = 8); 6.4 (dt, 1 H, J = 2, and 10); 6.0 (m, 1 H); 5.7 (dt, 1 H, J = 3.8, 10); 5.0 (m, 2 H); 4.7 (dd, 2 H, J = 2.0, 3.8); 3.9 (s, 3 H); 3.4 (dt, 2 H, J = 1.7, 6.2).

MS (70 eV): m/z (%) = 202 (M⁺, 100); 201 (77); 187 (22); 171 (25).

7-Benzyloxy-2H-1-benzopyran (4c):

IR (neat): v = 3045; 2850; 1645; 1605 cm⁻¹.

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¹H NMR: δ = 7.4 (m, 5H); 6.9 (d, 1H, J = 9.8); 6.5 (dd, 2H, J = 9.8, 3.0); 6.5 (d, 1H, J = 3), 6.4 (dt, 1H, J = 13, 3.2); 5.6 (dt, 1H, J = 13, 4.9); 5.0 (s, 2H); 4.8 (dd, 2H, J = 4.9, 3.2).

7-Methoxy-2H-1-benzopyran (4d).

IR (neat): v = 2960; 2840; 1615; 1280 cm⁻¹.

¹H NMR: $\delta = 6.8$ (d, 1 H, J = 8); 6.4 (m, 3 H); 5.6 (dt, 1 H, J = 3.5, 9.8); 4.8 (dd, 2 H, J = 3.5, 2.0); 3.7 (s, 3 H).

MS (70 eV): m/z (%) = 162 (M⁺, 60); 161 (100); 147 (30).

(\pm)-6a,12a-cis-Dihydro-3-methoxy-4-(3-methyl-2-butenyl)6H-[1,3]-dioxolo[5,6]benzofuro[3,2-c][1]benzopyran (4-prenylpterocarpin) (6): To a mixture of PdCl₂ (57 mg, 0.32 mmol) and LiCl (28 mg, 0.32 mmol) in acetone (10 mL) was added 7-methoxy-8-(3-methyl-2-butenyl)-2H-chromene 4a (74 mg, 0.32 mmol) in dry acetone (10 mL). This mixture was stirred for 15 min and 2-chloromercurio-4,5-methylenedioxyphenol (5; 120 mg, 0.32 mmol) in acetone (10 mL) was added. The suspension thus obtained was stirred for 5 h at r. t. After this time, brine (20 mL) was added and the mixture was extracted with benzene, dried (Na₂SO₄), filtered and concentrated. The product was purified by column chromatography leading to 6 as a colorless solid. (65 mg, 55%). mp = 185-186°C.

IR (KBr): v = 3050; 1615; 1486; 1250 cm⁻¹.

¹H NMR: δ = 7.3 (d, 1 H, J = 8); 6.7 (s, 1 H); 6.6 (d, 1 H, J = 8); 6.4 (s, 1 H); 5.9 (2 d, 2 H); 5.5 (d, 1 H, J = 6.7); 5.2 (m, 1 H); 4.2 (dd, 1 H, J = 11, 4.4); 3.8 (s, 3 H); 3.6 (t, 1 H, J = 11); 3.5 (m, 1 H); 3.3 (d, 2 H, J = 7); 1.8 (s, 3 H); 1.7 (s, 3 H).

¹³C NMR: δ = 17.6 (q); 22.2 (t); 26.7 (q); 40.1 (d); 55.7 (d); 66.6 (t); 79.1 (d); 93.6 (d); 101.1 (t); 104.6 (d); 112.8 (d); 118.0 (s); 118.1 (s); 122.4 (d); 128.5 (d); 131.3 (s); 141.5 (s); 147.7 (s); 153.8 (s); 154.2 (s); 158.2 (s).

MS (70 eV): m/z (%) = 366 (M⁺, 100); 310 (40); 175 (10); 149 (25). HRMS (70 eV): m/z, $C_{22}H_{22}O_5$, calc.: 366.1467; found: 366.1444.

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