



Total synthesis of antifungal gamahonolide A



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ARTICLE INFO

Article history:

Received 17 February 2014

Revised 8 April 2014

Accepted 8 April 2014

Available online 18 April 2014

Keywords:

Aminoxylation

Keck allylation

Ring-closing metathesis

Natural product

Antifungal

ABSTRACT

The first stereoselective total synthesis of gamahonolide A (**1**) has been accomplished using aminoxylation, Keck allylation, and ring-closing metathesis (RCM) reactions as key steps.

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6-Substituted α,β -unsaturated δ -lactone moiety containing natural products attracted the attention of synthetic chemists due to their interesting structure and potential biological activities. The activities include antiviral, antifungal, antibacterial, growth inhibitory, antitumor, and antileukemic.^{1,2} Gamahonolide A³ (**1**) is a δ -lactone, isolated from stromata of phytopathogenic fungus, *Epichloe typhina* on *Phleum pratense*. *E. typhina* causes choke disease to the pasture grass, timothy, *P. pratense*. Gamahonolide structure was determined by spectroscopic methods and the absolute configuration by its ORD spectrum and ¹H NMR shift difference between the diastereomeric pair of its *O*-methylmandelates. To the best of our knowledge, the synthesis of gamahonolide A (**1**) has not been reported to date. In continuation of our program on the total synthesis of bioactive α,β -unsaturated δ -lactone containing natural products,⁴ herein we report the first total synthesis of gamahonolide A (**1**) (Fig. 1) by a simple synthetic strategy.

Retrosynthetic analysis of **1** is outlined in Scheme 1. Ring-closing metathesis reaction and removal of the functional group in compound **2** would provide the target molecule, whereas, RCM precursor **2** could be made available from **3** by oxidation followed by Keck allylation and acryloylation. This in turn could be made from octane-1,8-diol via compound **4** by adopting α -aminoxylation reaction as a key step.

To begin the synthesis, octane-1,8-diol was selectively mono-protected as its PMB ether **5** and the remaining hydroxyl group was oxidized to the corresponding aldehyde. The aldehyde without isolation was subjected to the MacMillan α -hydroxylation⁵ using

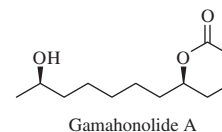
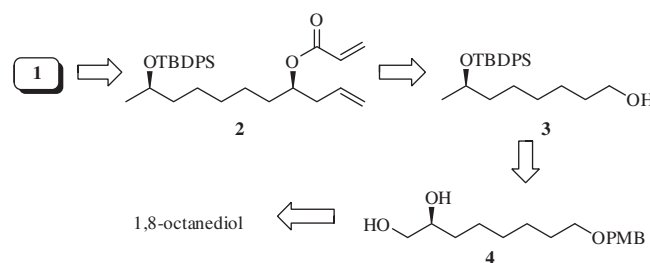


Figure 1. Structure of Gamahonolide A.

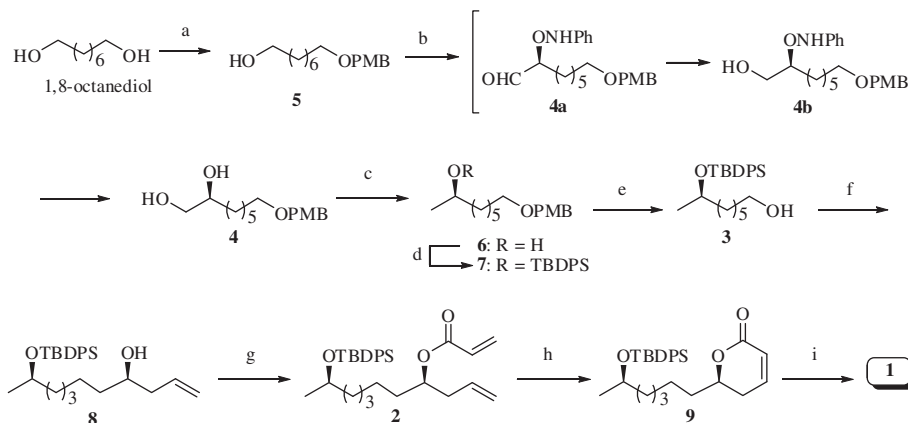


Scheme 1. Retrosynthetic analysis.

nitroso benzene and 40 mol % of D-proline in CHCl₃, followed by rapid reduction with sodium borohydride to furnish the unstable anilinoxy compound which was further treated with Zn in acetic acid at room temperature to cleave the O–N bond providing the diol **4** with high enantioselectivity (98.6% ee)⁶ (Scheme 2). This resulting 1,2-diol **4** on treatment with TsCl, and Et₃N in the presence of dibutyltin oxide was monosilylated which on further reduction with LAH in THF provided terminal methyl compound

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Scheme 2. Reagents and conditions: (a) NaH, PMB-Br THF, 0 °C–rt, 80%; (b) (i) IBX, DMSO, CH₂Cl₂, 3 h; (ii) PhNO, D-proline, CHCl₃, 0 °C, 2 h then NaBH₄, EtOH, 0 °C, 2 h then AcOH, Zn, rt, 12 h, 55% for 3 steps; (c) (i) TsCl, Et₃N, dibutyltin oxide (cat), dry CH₂Cl₂, 0 °C to rt, 12 h; (ii) LiAlH₄, THF, reflux, 3 h, 85% (for 2 steps); (d) TBDPSCI, imidazole, CH₂Cl₂, 0 °C, 3 h, 95%; (e) DDQ, CH₂Cl₂:H₂O (9:1), 0 °C, 2 h, 90%; (f) (i) IBX, DMSO, CH₂Cl₂, 3 h; (ii) (S)-BINOL, 4 Å MS, Ti(Oi-Pr)₄, Allyl-tributylstannane, CH₂Cl₂, –78 °C to –20 °C, 24 h, 83% (for 2 steps); (g) acryloyl chloride, NaH, THF, 0 °C, 87%; (h) Grubbs' 2nd generation catalyst (G-II, 10 mol %), CH₂Cl₂, reflux, 1 h, 72%; (i) TBAF, THF, 0 °C–rt, 85%.

6. The secondary alcohol silylated using TBDPSCI and imidazole to silyl ether **7** followed by removal of the PMB group resulted in **3**. IBX oxidation of alcohol provided aldehyde which was allylated following Keck's protocol⁷ to give homoallyl alcohol **8** with high diastereoselectivity (83%, 98% de).⁸

Acrylation of **8** was achieved by treatment with acryloyl chloride and NaH in THF to obtain diene **2** in 87% yield. Ring-closing metathesis⁹ of **2** proceeded well with 10 mol % of Grubbs-II to produce lactone **9** in 72% yield. Finally, desilylation with TBAF in THF afforded the target gamahonolide A (**1**) in 85% yield. This compound is identical in all respects to the reported natural product including NMR, optical rotation.

In conclusion, we have accomplished the first total synthesis of gamahonolide A by a simple strategy involving α -aminoxylation, Keck allylation, and ring closing-metathesis as the key steps.

Acknowledgements

The authors, K.P. thanks CSIR and S.P. thanks UGC New Delhi, INDIA, for the financial support.

Supplementary data

Supplementary data (copies of ¹H NMR, ¹³C NMR spectra available) associated with this article can be found, in the online version, at <http://dx.doi.org/10.1016/j.tetlet.2014.04.026>.

References and notes

- (a) Sam, T. W.; Yeu, C. S.; Jodynis-Liebert, J.; Murias, M.; Bloszyk, E. *Planta Med.* **2000**, *66*, 199; (b) Murga, J.; Falomoir, E.; Garcia-Fortanet, J.; Carda, M.; Marco, J. *A. Org. Lett.* **2002**, *4*, 3447.
- (a) Hagen, S. E.; Vara Prasad, J. V. N.; Tait, B. D. *Adv. Med. Chem.* **2000**, *5*, 159; (b) Inayat-Hussain, S. H.; Annuar, B. O.; Din, L. B.; Ali, A. M.; Ross, D. *Toxicol. In Vitro* **2003**, *17*, 433; (c) Kikuchi, H.; Sasaki, K.; Sekiya, J.; Maeda, Y.; Amagai, A.; Kubohara, Y.; Ohsima, Y. *Bioorg. Med. Chem.* **2004**, *12*, 3203.
- Koshino, H.; Yoshihara, T.; Okuno, M.; Sakamura, S.; Tajimi, A.; Shimanuki, T. *BioSci., Biotechnol., Biochem.* **1992**, *56*, 1096–1099.
- (a) Sabitha, G.; Reddy, C. N.; Raju, A.; Yadav, J. S. *Tetrahedron: Asymmetry* **2011**, *22*, 493–498; (b) Sabitha, G.; Reddy, C. N.; Gopal, P.; Yadav, J. S. *Tetrahedron Lett.* **2010**, *51*, 5736–5739; (c) Sabitha, G.; Gopal, P.; Reddy, C. N.; Yadav, J. S.

- Tetrahedron Lett.* **2009**, *50*, 6298–6302; (d) Sabitha, G.; Rao, A. S.; Yadav, J. S. *Tetrahedron: Asymmetry* **2011**, *22*, 866–871; (e) Sabitha, G.; Reddy, S. S. S.; Yadav, J. S. *Tetrahedron Lett.* **2010**, *51*, 6259–6261; (f) Sabitha, G.; Reddy, S. S. S.; Yadav, J. S. *Tetrahedron Lett.* **2011**, *52*, 2407–2409; (g) Sabitha, G.; Shankaraiah, K.; Yadav, J. S. *Eur. J. Org. Chem.* **2013**, 4870–4878.
 - (a) Mangion, I. K.; MacMillan, D. W. C. *J. Am. Chem. Soc.* **2005**, *127*, 3696–3697; (b) Varseev, G. N.; Maier, M. E. *Org. Lett.* **2007**, *9*, 1461–1464; (c) Brown, S. P.; Brochu, M. P.; Sinz, C. J.; MacMillan, D. W. C. *J. Am. Chem. Soc.* **2003**, *125*, 10808–10809; For other reports on proline catalyzed oxidation of aldehydes, see: (d) Zhong, G. *Angew. Chem., Int. Ed.* **2003**, *42*, 4247–4250; (e) Hayashi, Y.; Yamaguchi, J.; Hibino, K.; Shoji, M. *Tetrahedron Lett.* **2003**, *44*, 8293–8296.
 - The enantiomeric excess was determined by chiral HPLC column: (CHIRAL PAK OD-H: 250 × 4.6 mm, 5 μ) mobile phase: 10% IPA in Hexane, flow rate: 1 mL/min, detection: 210 nm, retention time: 13.054 min, 98.6% ee.
 - Keck, G. E.; Tarbet, K. H.; Geraci, L. S. *J. Am. Chem. Soc.* **1993**, *115*, 8467.
 - The diastereomeric excess of the product was determined using a Shimadzu high-performance liquid-chromatography (HPLC) system equipped with a chiral HPLC column and a UV detector at an absorbance of 210 nm. Atlantis C18, 4.6 × 150 MM, 5 μ) (column) and a solvent system of 90% acetonitrile in water at a flow rate of 1.0 ml/min were used. *t_R*: 6.7 and 9.9 min.
 - For a recent review on ring-closing metathesis reactions, see: (a) Frustner, A. *Angew. Chem., Int. Ed.* **2000**, *39*, 3012–3043; (b) Trnka, T.; Grubbs, R. H. *Acc. Chem. Res.* **2001**, *34*, 18–29; (c) Connon, S. J.; Blechert, S. *Angew. Chem., Int. Ed.* **2003**, *42*, 1900–1923; (d) Hoveyda, A. H.; Zhugralin, A. R. *Nature* **2007**, *450*, 243–251; (e) Garber, S. B.; Kingsbury, J. S.; Gray, B. L.; Hoveyda, A. H. *J. Am. Chem. Soc.* **2000**, *122*, 8168–8179; (f) Reddy, M. V. R.; Rearick, J. P.; Hock, N.; Ramachandran, P. V. *Org. Lett.* **2001**, *3*, 19–20; (g) Ramachandran, P. V.; Liu, H.; Reddy, M. V. R.; Brown, H. C. *Org. Lett.* **2003**, *5*, 3755–3757.
- Spectral data for selected compounds:
Compound 4: [α]_D²⁵ +9.0 (c = 0.5, CHCl₃). IR (neat): 3243, 2933, 2855, 1613, 1514, 1249, 1033, 816. ¹H NMR (CDCl₃, 500 MHz): δ 7.26 (d, *J* = 8.5 Hz, 2H), 6.88 (d, *J* = 8.6 Hz, 2H), 4.43 (s, 2H), 3.81 (s, 3H), 3.73–3.62 (m, 2H), 3.45–3.40 (m, 3H), 1.64–1.56 (m, 4H), 1.48–1.30 (m, 6H). ¹³C NMR (CDCl₃, 75 MHz): δ 158.7, 130.3, 128.9, 113.4, 72.1, 69.8, 66.3, 54.9, 32.7, 29.3, 29.2, 25.8, 25.6. ESI-MS: *m/z* 305 [M+Na]⁺. HRMS (ESI): calc. 305.1723 C₁₆H₂₆O₄Na, found 305.1719; **Compound 8**: [α]_D²⁵ –0.20 (c = 0.1, CHCl₃). IR (neat): 3416, 2930, 2857, 1637, 1108, 703. ¹H NMR (CDCl₃, 500 MHz): δ 7.70–7.65 (m, 4H), 7.44–7.34 (m, 6H), 5.87–5.74 (m, 1H), 5.16–5.08 (m, 2H), 3.87–3.79 (m, 1H), 3.64–3.52 (m, 1H), 2.16–2.03 (m, 2H), 1.52–1.15 (m, 10H), 1.07 (d, *J* = 6.1 Hz, 3H), 1.05 (s, 9H). ¹³C NMR (CDCl₃, 75 MHz): δ 135.8, 134.8, 134.5, 129.4, 129.3, 127.4, 127.3, 118.0, 70.4, 69.3, 41.8, 39.2, 36.6, 26.9, 23.1, 21.2. ESI-MS: *m/z* 447 [M+Na]⁺. HRMS (ESI): calc. 447.2689 C₂₇H₄₀O₂NaSi, found 447.2694 [M+Na]⁺; **Compound 1**: [α]_D²⁵ –67.5 (c = 0.2, CHCl₃). IR (neat): 3449, 2925, 2855, 1713, 1649, 1256, 1093, 769. ¹H NMR (CDCl₃, 500 MHz): δ 6.91 (ddd, *J* = 9.7, 5.4, 3.8 Hz, 1H), 6.02 (ddd, *J* = 10.0, 1.8, 1.5 Hz, 1H), 4.47–4.40 (m, 1H), 3.82 (ddq, *J* = 6.2, 6.1, 5.9 Hz, 1H), 2.36–2.31 (m, 2H), 1.85–1.75 (m, 1H), 1.73–1.56 (m, 2H), 1.52–1.38 (m, 7H), 1.20 (d, *J* = 6.2 Hz, 3H). ¹³C NMR (CDCl₃, 125 MHz): δ 164.4, 144.9, 121.4, 77.9, 68.0, 39.1, 34.7, 29.4, 29.3, 25.5, 24.8, 23.5. ESI-MS: *m/z* 235 [M+Na]⁺. HRMS (ESI): calc. 235.13047 C₁₂H₂₀O₃Na, found 235.13106 [M+Na]⁺.