

Asymmetric Synthesis of Heterocyclic Analogues of a CGRP Receptor Antagonist for Treating Migraine

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Supporting Information

ABSTRACT: An asymmetric synthesis of novel heterocyclic analogue of the CGRP receptor antagonist rimegepant (BMS-927711, 3) is reported. The cycloheptane ring was constructed by an intramolecular Heck reaction. The application of Hayashi—Miyaura and Ellman reactions furnished the aryl and the amine chiral centers, while the separable diastereomeric third chiral center alcohols led to both carbamate and urea analogues. This synthetic approach was applicable to both 6- and 5-membered heterocycles as exemplified by pyrazine and thiazole derivatives.

igraine is a painful, incapacitating disease that affects a Llarge portion (12%) of the adult population and imposes a substantial economic burden on society (estimated to be \$13 billion per year). Calcitonin gene-related peptide (CGRP) is thought to play a causal role in migraine and may act via multiple mechanisms including pain transmission, neurogenic inflammation, and/or neurogenic vasodilation.² As current standard of care, the triptan class of 5-HT_{1B/1D} receptor agonists actively constrict the dilated cranial arteries associated with a migraine and relieve migraine coincident with a reduction in CGRP levels.³ However, triptans are also associated with a number of unpleasant, and potentially dangerous, cardiovascular side effects due to their nonselective smooth muscle vasoconstriction.⁴ Because CGRP receptor antagonists block cranial vessel dilation, they are devoid of these undesirable cardiovascular effects of triptans and are emerging as new therapeutics for the effective treatment of migraine.

An oral CGRP receptor antagonist, telcagepant (MK-0974)⁶ (1, Figure 1), showed positive results in several phase II and phase III trials but was discontinued following a migraine-preventative study.⁷ A recent publication from our laboratory disclosed the potent, oral CGRP receptor antagonist, BMS-846372, containing a cyclohepta[b]pyridine core (2, Figure 1), that was an attractive preclinical lead.⁸ Poor aqueous solubility of 2 ($<2 \mu g/mL$) led us to install a primary amine in 3 (BMS-927711/rimegepant, Figure 1), which had even better potency and much improved aqueous solubility ($50 \mu g/mL$).⁹ In a phase II clinical trial, compound 3 showed efficacy comparable to sumatriptan (Imitrex, 100 mg), but without the significant cardiovascular side effects associated with sumatriptan.¹⁰ We wanted to further explore this core structure with other heterocycles such as pyrazine 4a and thiazole 4b, keeping the

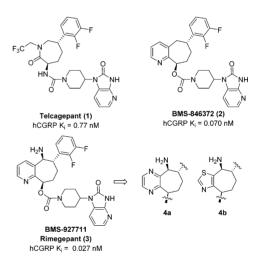


Figure 1. CGRP receptor antagonists.

spatial relationship of the pyridine sp₂ nitrogen, which is a critical pharmacophore for CGRP receptor affinity (Figure 1).

Attempts to prepare 4a in a manner analogous to the synthesis of 2 and 3^{8,9,11} failed. Indeed, heterocycle-fused cycloheptane ring systems are rare in the literature. ¹² A retrosynthetic analysis of target 4 is shown in Scheme 1 using the precedented carbamate formation from the alcohol 5.^{8,9,11,13} For the synthesis of 5, we envisioned a key Heck cyclization at the position next to the heterocyclic nitrogen to afford 6, which left a double bond for generation of the alcohol through the ketone intermediate. The Heck substrate 7, a chiral amine intermediate, could be generated in a diastereoselective

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Scheme 1. Retrosynthetic Analysis for 4

fashion from a chiral sulfinamide intermediate 9. The counterpart anion nucleophile 8 could be generated from a halogenated heterocycle. The sulfinamide 9 came from the α -aryl aldehyde 10 which had to be prepared enantioselectively and was expected to be prone to epimerization.

One proven approach to establish a chiral aryl center is the Hayashi-Miyaura Rh-catalyzed arylboronic acid addition to a nitroalkene. 14 In the asymmetric synthesis of 1, 2,3difluorophenylboronic acid was added successfully to an α unsubstituted nitroalkene with good diastereoselectivity. 15 We were able to apply the reported conditions 15 to our synthesis of 12 as shown in Scheme 2. Nitroalkene 11 was prepared in 72% yield from 5-pentenaldehyde with nitromethane using tetramethylguanidine (TMG) as catalyst and followed by addition of methansulfonyl chloride and triethylamine to effect the elimination of nitro alcohol. Compound 12 was then prepared in 96% yield using the reported conditions. 15 The Nef reaction 16 was successfully optimized to convert the nitroalkane to the desired aldehyde 10, which was directly used in the next reaction without noticeable epimerization. 17 Enantiomerically pure (R)-(+)-2-methyl-2-propanesulfinamide reacted with 10 to afford 9 in 73% yield from 12 with good diastereomeric ratio (>93:7 dr) as evidenced by ¹H NMR. Diastereoselective reaction of 9 with lithiated 2-bromopyrazine 19 went smoothly to afford mostly 7a in 62% yield (~9:1 dr by ¹H NMR²⁰). ^{18b} Several attempts at Heck cyclization of 7a under mild conditions gave no product, while temperatures higher than 120 °C²¹ caused decomposition of the sulfinyl group. Consequently, a protecting group swap with Boc was carried out using HCl in dioxane²² and Boc anhydride, giving 13 in 92% yield. The minor diastereomer was easily removed by flash column chromatography at this stage. With 13, a series of conditions for Heck cyclization were screened. Under the conditions shown in Scheme 2,23 the desired product 6a was

obtained in 20% yield. However, higher reaction temperatures intended to effect better conversion gave **14** as the main product instead. For example, when the reaction was run at 100 °C for 20 h, 49% of **14** and only 13% of **6a** were obtained. Further optimization failed to improve the yield of **6a** for enough material to proceed.

An alternative route for the Heck reaction to work is shown in Scheme 3. Compound 13 was converted to the α , β -unsaturated ester 15 under standard reaction conditions with Grubbs II catalyst. With the ester group in place, Heck cyclization could be run at higher temperatures (160 °C under microwave heating, 2 h) to effect the best conversion to the desired 17 (major product in 54% yield) with a smaller amount of the isomerized product 16 (24% yield). Ozonolysis of 17 led to the ketone 18 in 66% yield. Reduction of 18 by NaBH₄ afforded the two easily separable diastereomeric alcohols 5a and 19 in good yield. Coupling of 5a with compound 20 under the previously reported conditions, ¹¹ followed by TFA-mediated Boc removal, gave the desired carbamate 4a.

We also decided to prepare the corresponding urea analogue. Thus, alcohol 19 was converted to the key intermediate amine 21 in 77% overall yield as shown in Scheme 4 through Mitsunobu reaction with phthalimide, followed by treatment with hydrazine. However, 21 failed to react with 20 or our previous activated carbamate. 8,13 In the end, the carbamoyl chloride 24 was freshly prepared from 23, which was in turn synthesized from 22 in three steps. Under these conditions, formation of the urea proceeded smoothly in 61% yield. A minor side product in which the SEM group transferred to the oxygen was also generated in 13% yield. 25 Removal of both Boc and SEM groups by TFA afforded the urea analogue 25 in 55% yield.

With this route successfully established, we applied it to the synthesis of the thiazole analogue 4b as shown in Scheme 5. Starting with 2,4-dibromothiazole, we hoped that the 2-Br substituent could be retained to the end of the sequence as a handle for further functionalization. However, the 2-Br derivative of 27 was not well-behaved in the Heck cyclization. Hence, 2-Br was removed in one pot by lithium-halogen exchange after the lithiation of 2,4-dibromothiazole by LDA²⁶ and addition of 9 to generate compound 7b as the major diasteeromer. Removal of the sulfinyl group followed by Boc protection afforded compound 26, which was converted to acrylate 27 as described above. Under the conditions used for the pyrazine 15, Heck cyclization yielded a majority of the undesired endocyclic alkene 28 (2:1 over the desired isomer 29). For this synthesis, we discovered that the aldehyde 10 that had been used to prepare 9 had partially epimerized during

Scheme 2. Enantioselective Synthesis of 13 and Attempted Heck Cyclization

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Scheme 3. Heck Cyclization and Synthesis of 4a

Scheme 4. Synthesis of 25

Scheme 5. Asymmetric Synthesis of 4b

storage (~25% wrong diastereomer in 9). Consequently, compounds 7b, 26, 27, 28, and 29 each consisted of two inseparable diastereomers in a 3:1 ratio, as evidenced by ¹H NMR analysis.²⁵ After ozonolysis of 29, the undesired minor diastereomer was finally removed by flash column chromatography from ketone 30 (58% yield plus 20% diasteromeric ketone). After NaBH₄ reduction, the desired major alcohol 5b was obtained in 58% yield, which was carried on to 4b in good yield. Using a recently reported trifluoromethylation reaction,²⁷ we were able to convert 5b to 31 in 24% yield, which led to analogue 4c as shown in Scheme 6.

Binding affinities for the new CGRP receptor antagonists were determined by inhibition of ¹²⁵I-CGRP binding to SK-N-MC cell membranes, which endogenously express the receptor. ²⁸ Binding data for compounds tested is shown in Table 1. Urethanes 4a, 4b, and 4c all showed CGRP receptor binding activities that were comparable to those of 2 (BMS-846372) and rimegepant 3. The urea analogue 25, compared to 4a, had 10-fold lower potency, which, however, was comparable

Scheme 6. Synthesis of 4c

to telcagepant 1. Further profiling of these new compounds showed them to be inferior to 3 in terms of metabolic stability.

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Table 1. hCGRP K, Data

compd	$hCGRP\ K_i\ (nM)$
rimegepant	0.027 ± 0.001
4a	0.056 ± 0.011
25	0.55 ± 0.094
4b	0.048 ± 0.011
4c	0.065 ± 0.011

In summary, we have developed a novel asymmetric synthesis of select heterocyclic analogues of the CGRP receptor antagonist rimegepant 3. These showed binding affinity against the CGRP receptor that was comparable to the rimegepant. Our approach to the novel core structures featured a 7-membered ring intramolecular Heck cyclization in which the critical location of the double bond was achieved by addition of an ester group. The aryl chiral center was constructed by a variant of the Hayashi—Miyaura reaction, and the amine chiral center was obtained by Ellman chiral sulfinamide chemistry in a diastereoselective reaction with nucleophiles generated by lithiation of bromo heterocycles.

ASSOCIATED CONTENT

S Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.orglett.5b02921.

Experimental procedures and characterization of all intermediates and products (PDF)

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Notes

The authors declare no competing financial interest.

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