Tetrahedron Letters 53 (2012) 5891-5894

Contents lists available at SciVerse ScienceDirect

Tetrahedron Letters

journal homepage: www.elsevier.com/locate/tetlet



Synthesis of novel tetrazole C5-linked C₀- and C₂-ribonucleoside phosphoramidites using MePOM and POM groups for probing RNA catalysis

Shinya Harusawa ^{a,*}, Hiroki Yoneyama ^a, Daiki Fujisue ^a, Masayoshi Nishiura ^a, Mihoyo Fujitake ^a, Yoshihide Usami ^a, Zheng-yun Zhao ^b, Scott A. McPhee ^c, Timothy J. Wilson ^c, David M. J. Lilley ^c

^a Osaka University of Pharmaceutical Sciences, 4-20-1 Nasahara, Takatsuki, Osaka 569-1094, Japan
^b School of Chemistry, University of Birmingham, Edgbaston, Birmingham B15 2TT, UK
^c CR-UK Nucleic Acid Structure Group, MSI/WTB Complex, University of Dundee, Dundee DDI 5EH, UK

ARTICLE INFO

Article history: Received 1 June 2012 Revised 14 August 2012 Accepted 22 August 2012 Available online 29 August 2012

Keywords: Tetrazole Phosphoramidite Protecting group Chemogenetic approach Ribozyme

ABSTRACT

Novel C5-linked C_{0^-} and C_2 -tetrazole ribonucleoside phosphoramidites were designed and synthesized via tetrazole C-nucleosides. Pivaloyloxymethyl (POM) and methyl-substituted POM (MePOM) groups were introduced as N-protecting groups in the tetrazole ring that can be readily removed under mild basic conditions. The phosphoramidites were successfully incorporated into the VS ribozyme substrate and hence providing a chemogenetic approach to determine which nucleobases of ribozymes function as the acid or base, in the studies of ribozyme general acid and base catalysis.

© 2012 Elsevier Ltd. All rights reserved.

RNA catalysis occurs in cells for the processing of RNA molecules, the control of gene expression, and even the synthesis of proteins; yet the catalysis mechanisms are still unclear in most cases.¹ The Varkud satellite (VS) ribozymes are the largest of a group of nucleolytic ribozymes that include hammerhead, hairpin, HDV, and GlmS ribozymes.² While no crystal structure of VS ribozymes has been determined yet, a small-angle X-ray scattering-derived model places G638 and A756 in proximity to the scissile phosphate.^{2d}

We have recently developed a novel chemogenetic approach for the analysis of general acid–base catalysis by nucleobases in ribozymes.³ This involves substitution of a C4 linked-ribonucleoside with imidazole $(pK_a: 7.1)^4$ being a good donor and acceptor of proton in place of an important nucleobase in VS and hairpin ribozymes.⁵ In this study, *N*-pivaloyloxymethyl (POM) 2'-O*t*BDMS-imidazole C₀-phosphoramidite (PA) **2**^{5a,b} was first employed to yield an imidazole-substituted VS ribozyme by covalently placing imidazole as a pseudonucleobase at position 756 (replacing adenine) of the VS ribozyme (Fig. 1).^{3a} The imidazole ribozyme (A756Imz) catalyzed the almost complete cleavage of substrate stem-loops at the correct position; this confirms the direct role of the nucleobase at position 756 in the chemistry of the natural VS ribozyme.³ The overall yield of **2** tends to be rather limited due to the lack of selective introduction of the *t*BDMS group for 2'-hydroxy group as well as due to its acid-labile character.^{5a,b} Therefore, a new combination of protecting groups was introduced for the carbon-elongated homologs PAs **3** (n = 0-3): POM for imidazole-*N* and cyanoethyl (CE) groups for the 2'-hydroxy group (Fig. 1).^{5c,e} In addition, PAs **3a–d** are much more stable than PA **2**, making the purification steps easier. Further, it is of particular interest that the C₂-imidazole VS ribozyme (G638C₂Imz) incorporated from PA **3c** (n = 2) showed a 15-fold greater catalytic activity than the C₀-imidazole ribozyme (G638C₀Imz) ^{5c} These results definitely demonstrated that the key functionalities in the catalytic mechanism of VS ribozyme are G638 and A756 that concertedly act in general acid–base catalysis. However, it is impossible by this approach to determine which nucleobases specifically function as the acid or base when imidazole is employed as a nucleobase.

5-Aliphatic-1*H*-tetrazoles with pK_a of 4.5–4.9 are often used as metabolism-resistant isomeric replacements for carboxylic acids in medicinal chemistry.⁶ Given that, the synthesis of tetrazole C-nucleoside PAs may allow the incorporation of an acid surrogate at the G638 or A756 position to specify the role of the nucleobase as a general acid in the chemistry of natural VS ribozyme. However, the tetrazole-related C-nucleosides have not been extensively studied so far.⁷ We herein describe the synthesis of a novel C5-linked C₀- and C₂-tetrazole ribonucleoside PAs **1a** and **1b** (Fig. 1). The latter, with a flexible linker, may lead to greater catalytic activity at the G638 position of VS ribozyme. In this study,

^{*} Corresponding author. Tel./fax: +81 72 690 1086. E-mail address: harusawa@gly.oups.ac.jp (S. Harusawa).

^{0040-4039/\$ -} see front matter @ 2012 Elsevier Ltd. All rights reserved. http://dx.doi.org/10.1016/j.tetlet.2012.08.082



Figure 1. Structures of imidazole and tetrazole C_n-ribonucleoside PAs employed for solid-phase RNA synthesis.

POM and a novel methyl-substituted POM (MePOM) groups were used as protecting groups for the tetrazole-*N*. In addition, we demonstrate initial results about the efficient syntheses of RNA oligonucleotides using PAs **1a** and **1b**.

We previously reported an efficient synthesis of B-D-ribofuranosyl cyanide 4, in 56% overall yield from commercially available 2,3,5-tri-O-benzyl-D-ribose in two steps (Scheme 1).5c Tetrazole C-ribonucleoside **5** (n = 0) was prepared in quantitative yield by the treatment of nitrile 4 with sodium azide and triethylammonium chloride in DMF. This reaction proceeded sufficiently rapidly due to the activation of the neighboring electron-withdrawing oxygen in the sugar moiety. We first examined the use of the POM group as a protecting group for tetrazole-N of 5. The POM group was used in the synthesis of imidazole C_n-PAs and was removed in the final step of RNA synthesis by a mixture of ammonia and ethanol with other base-protecting groups.^{5a,b} However, the introduction of the POM group into tetrazole **5** resulted in a 1:1 inseparable mixture **11ab** (74%) of *N*-1 and *N*-2 POM-protected tetrazoles (Scheme 1, the bottom left). The subsequent synthetic processes from 11ab suffered a setback owing to the formation of further complicated mixtures. We then tried MePOM protection via 1-chloroethyl pivaloate (MePOMCl),⁸ with expectation of *N*-2 protection caused preferentially by stereo hindrance of the methyl group. Indeed, the reaction of tetrazole C-nucleoside 5 with MePOMCl in the presence of DMAP in DMF afforded N-1 and N-2 MePOM derivatives 6a (24%) and 6b (60%), which could be isolated as diastereoisomers by column chromatography because of the additional stereogenic center present in the MePOM group (which, of course, could easily be identified from NMR spectra). The location of the MePOM group on the tetrazole ring was assigned, since the ¹³C NMR signal from the only carbon in tetrazole was generally about 10 ppm deshielded in 2,5-disubstituted derivatives relative to the corresponding 1,5-disubstituted isomers [¹³C NMR shift of δ 5-C (ppm) in CDCl₃: **6a**, 153.4; **6b**, 165.0].^{6d} MePOM groups of **6a** and **6b** could be removed quantitatively by aqueous ammonia/ EtOH (MeOH) (1:3, v/v) at room temperature (rt) in 17 h to convert back into 5. Removal of the MePOM group from tetrazole-N proceeded somewhat slowly compared to the removal of the POM group at imidazole-N (rt, 3 h).^{5a,b} In contrast, stirring of **6b** in 2N HCl-EtOH (1:3, v/v) at rt for 17 h resulted in recovery of **6b** in 94% yield, indicating the stability of the *N*-MePOM group under an aqueous acidic condition. Subsequent treatment of 6b with Pd(OH)₂-C/cyclohexene in refluxing ethanol produced *N*-MePOM-tetrazole C-nucleoside **7** (98%). It should be noted that MePOM group tolerates the debenzylation-condition. In addition, to our knowledge, the MePOM group has not been employed as an *N*-protecting group. 3',5'-O-TIPDS-protection (1,1,3,3-tetraisopropyldisiloxanediyl) of **7** gave **8** (86%), and this allowed selective introduction of the 2'-hydroxy protecting group. Cyanoethylation of **8** with acrylonitrile gave fully protected intermediate **9** (62%).⁹ The TIPDS group of **9** was selectively removed by treatment with Et₃N·3HF to give 3',5'-O-unprotected ribonucleoside derivative **10** (95%). Standard 5'-O-dimethoxytritylation (80%) of **10** followed by phosphitylation of 3' hydroxy afforded tetrazole C₀-PA **1a** as a mixture of four diastereomers (67%).¹⁰

C₂-tetrazole ribonucleoside PA 1b was synthesized from aldehyde 12 that was prepared in two steps (92%) from 4 (Scheme 1).^{5c} Wittig olefination of **12** using (cvanomethyl)triphenvlphosphonium chloride gave the vinvlnitrile (73%) and then reduction of the formed double bond yielded the two-carbon-elongated nitrile 13 (94%). Conversion of the inactive alkylcyanide 13 into tetrazole 14 in the presence of sodium azide and triethylammonium chloride proceeded slowly (40 h, 130 °C, DMF) in a moderate yield (57%), while microwave (MW) irradiation (130 °C, DMF) led to both a shorter reaction time (2 h) and an improved yield (95%). The treatment of 14 with POMCI (DBU, 100 °C, 0.5 h, DMF, MW) yielded N-1 and N-2 POM derivatives 15a (40%) and 15b (55%), respectively, which could be easily separated by column chromatography. Meanwhile, MePOM-protection of 14 also gave almost the similar result. The N-2 POM isomer 15b was used for the subsequent synthetic process because it showed a clearer ¹H NMR peak pattern than 15a did. The base-labile POM 15a could be transformed back to tetrazole 14 quantitatively for reuse, while **15a** remained intact under 2N HCl/EtOH (1:2, v/v; rt, 5 h). There is little information regarding N-POM tetrazoles in the literature,¹¹ and their role as a protecting group has not yet been fully clarified. Debenzylation of 15b produced N-2-POM-tetrazole C₂-ribonucleoside 16 (91%) and subsequent 3',5'-O-DTBS-protection (88%, DTBS = di-tert-butylsilanediyl) afforded 17 (88%). Although 2'-O-TBDMS protection of 17 provided 2'-O-silvlated derivative 18 (98%), cyanoethylation of 17 afforded only a POM removed tetrazole 20 (5%) (Scheme 1). Alternatively, 3',5'-O-TIPDS protection of 16 gave 21 (quant), but the cyanoethylation of 21 to the corresponding 22 could not be obtained at all (Scheme 1). Selective removal (90%) of the DTBS group in 18, DM-tritylation (quant), and phosphitylation (87%) proceeded successfully to give the final 2'-O-TBDMS tetrazole C₂-PA **1b**. ¹²



Scheme 1. Synthesis of tetrazole C_0 and C_2 -PAs **1** and related intermediates. Reagents and conditions: (a) BuLi, (cyanomethyl)triphenylphosphonium chloride, toluene, $-14 \degree$ C, 30 min, then rt, 2 h; (b) 10% Pd/C (cat.), H₂, 2Kg/cm², AcOEt, 1 h; (c) for **4**: NaN₃ (3.0 equiv), Et₃N·HCl (3.0 equiv), DMF, 130 °C, 2 h; for **13**: NaN₃ (3.0 equiv), Et₃N·HCl (3.0 equiv), DMF, MW, 130 °C, 2 h; (d) for **5**: MePOMCl (4 equiv), DMP (4 equiv), DMF, 50 °C, 10 h; for **14**: POMCl (2 equiv), DBU (2 equiv), DMF, MW, 100 °C, 0.5 h; (e) 28% aq. NH₃/EtOH (MeOH) (1:3, v/v), rt, 14–17 h; (f) for **6b**: 20% Pd(OH)₂-C, cyclohexene , EtOH, reflux, 25 h; for **15b**: 20% Pd(OH)₂-C, cyclohexene , EtOH, MW, 80 °C, 0.5 h; (g) for **7**: TIPDSCl2, py, 0 °C then rt, 2 h; for **16**: 'Bu₂Si(OTf)₂, py., rt, 10 min; (h) for **8**: CH₂=CHCN, Cs₂CO₃, 'BuOH, 40 °C, 10 h; for **17**: TBDMSCl, imidazole, DMF, 50 °C 5 h; (i) for **9**: Et₃N·HF, Et₃N, THF, rt, 1.5 h; for **18**: pyHF, py, CH₂Cl₂, 0°C, 1h; (j) DMTCl, Et₃N, DMAP, py, rt, 2.5 - 3.0 h; (k) Et₃N·3HF, (¹Pr₂N)₂POCH₂CH₂CN, 4,5-DCl, ClCH₂CH₂Cl, 40 °C, 20–30 min.

Purification of crude PAs **1a** and **1b** could be easily carried out by medium-pressure flash column chromatography (MPLC) using a standard silica gel column (Fuji Silysia Chemical FL60D, pH 6.6). This suggests that **1a** and **1b** are considerably more stable than the acid-sensitive imidazole PAs. Base-protecting groups of RNA are conventionally removed in the final step of RNA synthesis by ammonia and ethanol mixture [28% aq NH₃/EtOH (3:1, v/v)] at 55 °C for 16 h. As the POM and MePOM groups can be removed under milder basic conditions at rt, it is particularly attractive to sensitive RNA such as Cy5 labeled RNA where deprotection should preferably be carried out at rt in order to minimize the destruction of cyanine dye. In addition, MS measurements of tetrazole C_n-PAs **1a** and **1b** were problematic owing to their labile properties, but we recently reported that the molecular weight of PAs may be accurately determined by using a novel matrix system [triethanol-amine (TEOA)–NaCl] on FABMS equipped with a double-focusing mass spectrometer.¹³

The present method successfully revealed the sodium adduct ions $[M+Na]^+$ of **1a** and **1b**, and thus, their composition formulas could be determined.^{10,12} The overall yields of **1a** is 16.0% in 8 steps from nitrile **4**, and that of **1b** is 22.0% in 10 steps from aldehyde **12**; making them feasible approaches for the scaling up of their synthesis.

Preliminary incorporation of these modified tetrazole PAs **1a** and **1b** into VS ribozyme substrate stands (25nt) was carried out







Figure 2. (A) Analysis of the products of RNA synthesis by gel electrophoresis.¹⁴ Lane 1 VS substrate wtG638; Lane 2 G638C0Tez; Lane 3 G638C2Tez. (B) HPLC profile of G638C₂Tez. Desired 25mer has a retention-time of 14 min.

with Applied Biosystems 394 synthesiser with stepwise coupling vield over 97%.¹⁴ Demonstrated here is the denaturing gel image for wild type substrate and C₀- and C₂-tetrazole substituted (G638C₀Tez and G638C₂Tez) strands at the same position of the sequence as well as the HPLC profile¹⁵ of G638C₂Tez, as shown in Figure 2.

Further investigation on the application of **1a** and **1b** in ribozyme studies is under way and will be published in due course.

Acknowledgment

This work was supported in part by a Grant-in-Aid for Scientific Research [Grant No.2159030 (to S.H.)] from JSPS, 2009-2013.

References and notes

- 1. Lilley, D. M. J.; Eckstein, F. In Ribozymes and RNA Catalysis; Royal Society of Chemistry: Cambridge, 2008.
- (a) Wilson, T. J.; McLeod, A. C.; Lilley, D. M. J. EMBO J. 2007, 26, 2489; (b) 2. Lafontaine, D. A.; Wilson, T. J.; Norman, D. G.; Lilley, D. M. J. J. Mol. Biol. 2001,

312, 663; (c) Lafontaine, D. A.; Wilson, T. J.; Zhao, Z.; Lilley, D. M. J. J. Mol. Biol. 2002, 323, 23; (d) Lipfert, J.; Ouellet, J.; Norman, D. G.; Doniach, S.; Lilley, D. M. J. Structure 2008, 16, 1357.

- 3 (a) Zhao, Z.; McLeod, A.; Harusawa, S.; Araki, L.; Yamaguchi, M.; Kurihara, T.; Lilley, D. M. J. *J. Am. Chem. Soc.* **2005**, *127*, 5026; (b) Wilson, T. J.; Ouellet, J.; Zhao, Z.; Harusawa, S.; Araki, L.; Kurihara, T.; Lilley, D. M. J. RNA **2006**, *12*, 980; (c) Lilley, D. M. J. Biol. Chem. 2007, 388, 699.
- 4. Perrotta, A. T.; Shih, I.; Been, M. D. Science 1999, 286, 123.
- (a) Araki, L.; Harusawa, S.; Yamaguchi, M.; Yonezawa, S.; Taniguchi, N.; Lilley, 5. D. M. J.; Zhao, Z.; Kurihara, T. Tetrahedron Lett. 2004, 45, 2657; (b) Araki, L.; Harusawa, S.; Yamaguchi, M.; Yonezawa, S.; Taniguchi, N.; Lilley, D. M. J.; Zhao, Z.; Kurihara, T. Tetrahedron 2005, 61, 11976; (c) Araki, L.; Morita, K.; Yamaguchi, M.; Zhao, Z.; Wilson, T. J.; Lilley, D. M. J.; Harusawa, S. J. Org. Chem. 2009, 74, 2350; (d) Araki, L.; Zhao, Z.; Lilley, D. M. J.; Harusawa, S. Heterocycles 2010, 81, 1861; (e) Harusawa, S.; Fujii, K.; Nishiura, M.; Araki, L.; Usami, Y.; Zhao, Z.; Lilley, D. M. J. Heterocycles 2011, 83, 2041.
- 6 For recent reviews on tetrazoles, see: (a) Bhatt, U. Five-membered Heterocycles with Four Heteroatoms Tetrazoles In Modern Heterocyclic Chemistry; Alvarez-Builla, J., Vaquero, J. J., Barluenga, J., Eds.; Wiley-VCH: Weinheim, 2011; Vol. 3, pp 1401-1430; (b) Herr, R. J. Bioorg. Med. Chem. 2002, 10, 3379; (c) Bräse, S.; Gil, C.; Knepper, K.; Zimmermann, V. Angew. Chem., Int. Ed. 2005, 44, 5188; (d) Butler, R. N. Tetrazoles In Comprehensive Heterocyclic Chemistry II; Katritzky, A. R., Rees, C. W., Scriven, E. F. V., Eds.; Pergamon: Oxford, 1996; Vol. 4, pp 621-678.
- 7. (a) Popsavin, M.; Torović, L.; Spaić, S.; Stankov, S.; Kapor, A.; Tomić, Z.; Popsavin, V. Tetrahedron 2002, 58, 569; (b) Kobe, J.; Prhavc, M.; Hohnjec, M.; Townsend, L. B. Nucleosides Nucleotides 1994, 13, 2209.
- Yoshimura, Y.; Hamaguchi, N.; Yashiki, T. J. Antibiot. 1986, 39, 1329.
- Saneyoshi, H.; Seio, K.; Sekine, M. J. Org. Chem. 2005, 70, 10453.
- **1a**: oil; ¹H NMR (500 MHz, CDCl₃) δ 1.01–1.26 (21H, m), 1.84 (1.05H, d, 10. J = 6.5 Hz), 1.85 (0.45H, d, J = 6.5 Hz), 1.88 (1.05H, d, J = 6.5 Hz), 1.90 (0.45H, d, J = 6.5 Hz), 2.35 (0.6H, t, J = 6.6 Hz), 2.60 (1.4H, t, J = 6.3 Hz), 2.56–2.68 (2H, m), 3.06–3.11 (1H, m), 3.37–3.47 (1H, m), 3.54–3.62 (2H, m), 3.64–3.86 (2H, m), 3.78 (4.2H, s), 3.79 (1.8H, s), 3.88-4.00 (2H, m), 4.28-4.30 (0.7H, m), 4.33-4.35 (0.3H, m), 4.45–4.52 (1H, m), 4.56–4.62 (1H, m), 5.26–5.30 (1H, m), 6.78–6.82 (4H, m), 7.16–7.44 (10H, m); $^{31}\mathrm{P}$ NMR (202 MHz, CDCl₃) δ 149.8, 149.9, 150.4, 150.5; HRMS (FABMS:TEOA+NaCl) m/z calcd for C46H60N7O9P+Na [M+Na] 908.4088, found 908.4085.
- (a) Heerding, J. M.; Lampe, J. W.; Darges, J. W.; Stamper, M. L. Bioorg. Med. Chem. 11. 1995, 5, 1839; (b) Mcdonald, I. M.; Black, J. M.; Buck, I. M.; Dunstone, D. J.; Griffin, E. P.; Harper, E. A.; Hull, R. A. D.; Kalindjian, S. B.; Lilley, E. J.; Linney, I. D.; Pether, M. J.; Roberts, S. P.; Shaxted, M. E.; Spencer, J.; Steel, K. I. M.; Sykes, D. A.; Walker, M. K.; Watt, G. F.; Wright, L. W.; Wright, P. T.; Xun, W. J. Med. Chem. 2007, 50, 3101.
- 12 **1b**: oil; ¹H NMR (400 MHz, CDCl₃) δ 0.11 (1.8H, s), 0.13 (2.1H, s), 0.14 (2.1H, s), 0.92 (8.4H, d, *J* = 7.2 Hz), 0.98 (3.6H, d, *J* = 6.8 Hz), 1.08–1.18 (9H, m), 1.19 (9H, s), 1.92–2.04 (1H, m), 2.18–2.30 (2H, m), 2.53–2.67 (1H, m), 2.98–3.14 (2H, m), 3.19–3.38 (2H, m); 3.48–3.64 (3H, m), 3.74–4.22 (5H, m), 3.78 (4.2H, s), 3.79 (1.8H, s), 6.40 (0.6H, s), 6.41 (1.4H, s), 6.80–6.86 (4H, m), 7.16–7.48 (9H, m); ³¹P NMR (120 MHz, CDCl₃) δ 148.6, 151.2; HRMS (FABMS: TEOA+NaCl) m/z calcd for C₅₀H₇₃N₆O₉PSi+Na [M+Na]⁺ 983.4844, found 983.4841.
- (a) Fujitake, M.; Harusawa, S.; Araki, L.; Yamaguchi, M.; Lilley, D. M. J.; Zhao, Z.; 13. Kurihara, T. *Tetrahedron* **2005**, *61*, 4689; (b) Fujitake, M.; Harusawa, S.; Zhao, Z.; Kurihara, T. Bull. Osaka Univ. Pharm. Sci. 2007, 1, 107; (c) Harusawa, S.; Fujitake, M.; Kurihara, T.; Zhao, Z.; Lilley, D. M. J. Mass Determination of Phosphoramidites. In *Current Protocols in Nucleic Acid Chemistry*; Beaucage, S. L., Bergstrom, D. E., Herdewijn, P., Matsuda, A., Eds.; John Wiley & Sons: New York, 2006; pp 10.11.1–10.11.16; (d) Fujitake, M.; Harusawa, S. Bull. Osaka Univ. Pharm. Sci. 2011. 5. 49.
- (a) Oligoribonucleotides were synthesized on an Applied Biosystems 394 14. instrument using 2'-O-tBDMS PA chemistry on Applied Biosystems LV 200 polystyrene 2'-deoxythymidine columns. A coupling time of 12 min was used. 0.5 M 5-ethylthio-1H-tetrazole in acetonitrile was used as an activator and 0.1 M iodine/water was used as an oxidiser.; (b) Oligoribonucleotides containing tetrazole modification were deprotected using 75% aqueous ammonia (35%) in ethanol for 17 h at rt and then evaporated to dryness. Oligoribonucleotides were redissolved in 0.3 ml of 1 M TBAF in THF to remove t-BDMS, and agitated at rt for 16 h prior to desalting by NAP-10 columns (GEHealthcare). Following ethanol precipitation, an aliquot (0.5 OD) of oligoribonucleotides was loaded on 20% denaturing polyacrylamide gel with $7\, \ensuremath{\widetilde{M}}$ urea and $90\, \ensuremath{\mathrm{mM}}$ TBE, and electrophoresed at $25\, \ensuremath{\mathrm{W}}$ for 2.5 h with bromophenyl blue (BPB) and xylene cyanol (XC) as makers. The gel was visualized by UV shadowing over a F254 TLC plate; (c) Sequences of oligoribonucleotides synthesized; Wild type: GCGCGAAGGCGUCGUCGC CCCGAdT; C_1Pez-strand: GCGCGAAGGCGUCGUCGCCCC(C_0Tet)AdT; C_2Tezstrand: GCGCGAAGGGCGUCGUCGCCCC(C₂Tez)AdT.
- 15 It was analyzed on an Ace C_{18} -300 column, eluted (1 mL/min) with buffer A (0.1 M TEAA, pH 7.0) and then eluted with a linear gradient from buffer A to 100% buffer B (MeCN) over a period of 60 min.