



Advanced Synthesis & Catalysis

Accepted Article

Title: Sn-Li Transmetalation of α -Aminoorganostannanes for the Stereoselective Synthesis of Substituted Dehydropiperidines and Dehydroazepanes

Authors: Alexandre Lumbroso, Isabelle Beaudet, Jean-Paul Quintard, Cécile Fraisse, Nicolas Galland, Loic Toupet, and Erwan Le Grogne

This manuscript has been accepted after peer review and appears as an Accepted Article online prior to editing, proofing, and formal publication of the final Version of Record (VoR). This work is currently citable by using the Digital Object Identifier (DOI) given below. The VoR will be published online in Early View as soon as possible and may be different to this Accepted Article as a result of editing. Readers should obtain the VoR from the journal website shown below when it is published to ensure accuracy of information. The authors are responsible for the content of this Accepted Article.

To be cited as: *Adv. Synth. Catal.* 10.1002/adsc.201900349

Link to VoR: <http://dx.doi.org/10.1002/adsc.201900349>

DOI: 10.1002/adsc.201((will be filled in by the editorial staff))

Sn-Li Transmetalation of α -Aminoorganostannanes for the Stereoselective Synthesis of Substituted Dehydropiperidines and Dehydroazepanes

Alexandre Lumbroso,^a Isabelle Beaudet,^a Jean-Paul Quintard,^a Cécile Fraisse,^a Nicolas Galland,^a Loïc Toupet,^b Erwan Le Grogne^{a*}

^a Université de Nantes, CNRS, CEISAM, UMR 6230, F-44000 Nantes, France.
e-mail: erwan.legrogne@univ-nantes.fr

^b Université de Rennes 1, CNRS UMR 6251, Institut de Physique de Rennes, Campus de Beaulieu, 35042 Rennes, France.

Received: ((will be filled in by the editorial staff))



Supporting information for this article is available on the WWW under <http://dx.doi.org/10.1002/adsc.201#####>.

Abstract. A highly diastereoselective synthesis of (*R,S*) or (*S,S*) 2,6-disubstituted dehydropiperidines and 2,7-disubstituted dehydroazepanes has been developed. The stereochemical preference for the (*R,S*) or the (*S,S*) isomer is governed by a tin-lithium exchange/electrophilic trapping sequence combined with a ring-closing metathesis. Their relative order was found to have a dramatic influence on the

interaction controlling the epimerization at the anionic center. In each case, the complete stereodivergence of the reactions sequence is explained by considering the stabilization brought by the lithium-carbonyl coordination and rationalized by DFT calculations. **Keywords:** Tin, Lithium, Sn-Li exchange, Chiral carbanion, DFT calculations, Ring-closing metathesis, Dehydropiperidines.

Introduction

Functionalized piperidines are widely present in alkaloids and bioactive compounds, therefore many methodologies have been developed to prepare these six-membered ring heterocycles.^[1] Moreover, polyhydroxylated piperidines (named iminosugars or iminocyclitols) constitute the most attractive carbohydrate mimetics reported to date and can act as glycosidase or glycosyltransferase inhibitors.^[2] For these applications, it is essential to control the stereocenter configuration of the piperidines and especially at the 2 and 6 positions since they define the series of sugars they mimic.^[3] In this context, the stereocontrolled synthesis of 2,6-disubstituted dehydropiperidines, which can be used as precursors of 2,6-disubstituted piperidines, presents a strong interest. Having in hand a series of 2-stannylated dehydropiperidines precursors,^[4] we have evaluated their interest for the stereoselective synthesis of 2,6-disubstituted-3,4-dehydropiperidines through a sequence involving a Sn-Li transmetalation and an electrophilic trapping. We also considered this strategy for the synthesis of dehydroazepanes, since

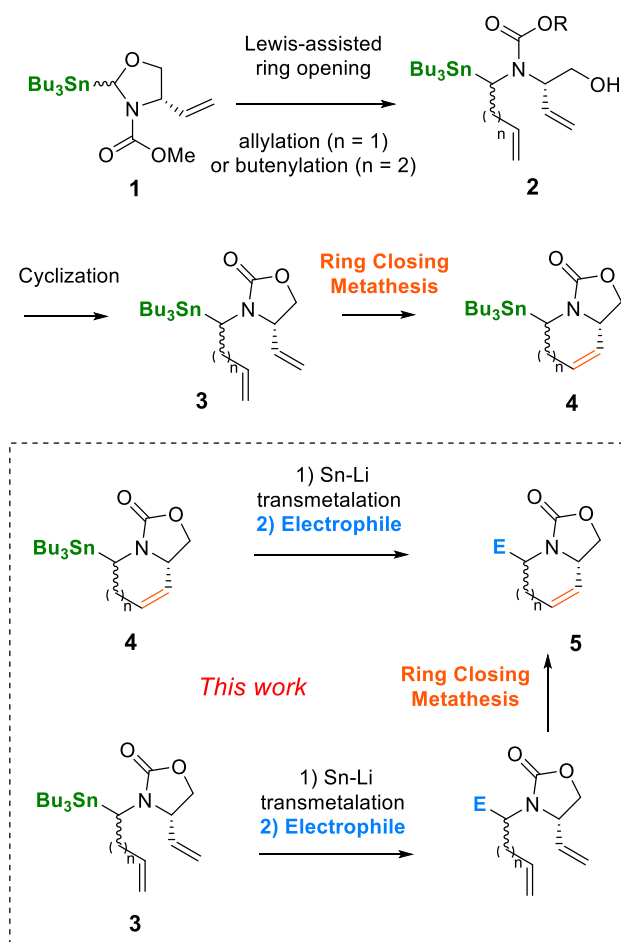
polyhydroxylated azepanes are also present in biological active products and have received some attention.^[5]

The Sn-Li transmetalation of organostannanes by organolithium reagents is an efficient reaction which has been applied to aryltins, heteroaryltins, vinyltins, allyltins or α -heterosubstituted organotins.^[6] When applied to α -oxygenated organostannanes to prepare α -oxygenated organolithium reagents, this reaction can be highly stereoselective when achieved at low temperature, although sensitive to steric hindrance.^[7] Similarly, α -nitrogenated organolithium reagents can be obtained by Sn-Li transmetalation from the corresponding organostannyl precursors. However, in this case the stereoselectivity is highly dependent on the temperature, the solvent and the nitrogen protecting group.^[8] Previous studies have shown that a *N*-carbamoyl protection allows a stabilization of the desired lithiated species.^[9] Moreover, the Sn-Li transmetalation of 2-organostannyl-1,3-oxazolidinones affords organolithium derivatives having a higher trend to epimerize into the isomer allowing the most stabilizing coordination.^[10] Based on these contributions, we assumed that the Sn-Li

transmetalation of 2-stannylated dehydropiperidines protected as oxazolidinones could afford the corresponding anionic species with a control of the stereoselectivity.

We have previously examined the ring opening of 2-tributylstannyl-1,3-oxazolidinones derived from (*S*)-vinylglycinol by soft nucleophiles. This procedure allows an efficient access to 4-tributylstannyl-3-azadienyl alcohols **2** when alkenylmetals are used in the presence of a Lewis acid.^[11] These latter can be subsequently converted into the corresponding oxazolidinones **3** leading to the stannylated dehydropiperidines (or dehydroazepanes) **4** by ring closing metathesis.^[4] Following a preliminary communication, the aim of the present work is to study the Sn-Li transmetalation reaction carried out on the stannylated oxazolidinones **3** and **4** by *n*-butyllithium in order to evaluate its diastereoselectivity and to rationalize the obtained results (Scheme 1).^[12]

We first examined the transmetalation reaction of stannylated oxazolidinones **4** before considering those of **3**.



Scheme 1. Preparation and transmetalation of stannylated oxazolidinones **3** and **4**.

Results

Transmetalation of stannylated oxazolidinones 4. Our investigations first focused on the Sn-Li

transmetalation reaction of enantiopure *N*-(α -tributylstannylorgano)oxazolidin-2-ones (*R,S*)-**4a** and (*S,S*)-**4a**. The reaction was conducted at -78°C in THF in the presence of *n*-BuLi (1.6 equiv.) and cyclohexanone was added after 20 min. Under these standard experimental conditions, the (*S,S*)-**4a** diastereomer afforded the **5a** product as a single diastereomer in 72% yield (Table 1, entry 1). Suitable monocrystals for X-ray diffraction analysis were obtained and the absolute configuration was found to be (*R,S*), indicating that the configuration of the stereocenter was preserved through the reactions.

Compound (*R,S*)-**5a** was also obtained when the reaction sequence was carried out with (*R,S*)-**4a**, corresponding in this case to an inversion of configuration (entry 2), while the yield was found to be much lower (30%). Starting from the (*S,S*)-**4b** diastereomer, the 7-membered ring analogue, product **5b** was selectively obtained as a single isomer in 79% yield (entry 3), and assigned to be (*R,S*)-**5b** by analogy to (*R,S*)-**5a**. This (*R,S*)-**5b** diastereomer was also selectively obtained in 58% yield when the reaction sequence was carried out starting from (*R,S*)-**4b** (entry 4). It indicates a high diastereoselectivity regardless of the configuration of the carbon bearing the tin moiety on the stannylated precursor and of the size of the bicyclic heterocycle. However, since higher yields were obtained after the transmetalation/trapping sequence when starting from (*S,S*)-**4** instead of (*R,S*)-**4**, the (*S,S*)-**4** isomer was systematically used in subsequent studies.

Under the same experimental conditions (-78°C , THF), the transmetalation/trapping sequence carried out on (*S,S*)-**4a** proceeded with a good yield and a very high diastereoselectivity when $\text{CO}_2(\text{g})$ was the electrophile.^[12] The Sn-Li exchange on (*S,S*)-**4a** followed by electrophilic trapping was also proved to be efficient with benzaldehyde, isovaleraldehyde and hexanal as electrophiles, leading selectively to (*R,S*) products, but as a couple of diastereomers due to the lack of diastereoselectivity on the aldehyde enantiotopic faces. The configuration of the major diastereomer of **7a** was determined by X-ray diffraction analysis of suitable monocrystals and was found to be (*S,R,S*). When the electrophile was benzaldehyde or isovaleraldehyde, the reaction temperature was lowered to -90°C to avoid an intramolecular acyl transfer reaction occurring on one of the two diastereomers.^[13] This transfer results from an intramolecular attack of the alcoholate on the oxazolidinone, as it was previously reported for similar compounds.^[14] When hexanal was the electrophile, a temperature of -100°C was required to prevent this side reaction. However, these experimental conditions dramatically impacted the yield (33%) due to the concomitant formation of the hydrolysis product **10** resulting of a “frozen” reactivity of the lithiated intermediate.^[13] The absolute configuration of the **8a** and **9a** diastereomers were deduced from the comparison of their NMR spectra and R_F values to those of **7a** (see supporting information).

Table 1. Sn-Li transmetalation followed by an electrophilic trapping of **4a** and **4b**.

4a , $n = 1$ 4b , $n = 2$			
5a , $n = 1$ 5b , $n = 2$			
A. With cyclohexanone as electrophile			
Entry	Stannylated oxazolidinone	Products	Yield (%) (<i>trans/cis</i>)
1	(S,S)- 4a	(R,S)- 5a (X-ray ^[12])	72 (0/100) ^[a,b]
2	(R,S)- 4a	(R,S)- 5a	30 (0/100) ^[b]
3	(S,S)- 4b	(R,S)- 5b	79 (0/100) ^[b]
4	(R,S)- 4b	(R,S)- 5b	58 (0/100) ^[b]
B. With other electrophiles starting from (S,S)-4a			

^[a] The (*R,S*) absolute configuration was confirmed by X-ray analysis.

^[b] dr was determined by GC on the crude products.

^[c] CO₂(g) was used as electrophile followed by an esterification (TMSCHN₂ (1.3 equiv.), C₆H₆:MeOH = 4:1).

^[d] Benzaldehyde was used as electrophile and the temperature was lowered to -90°C; the configuration of the major compound **7a** was determined to be (*S,R,S*) by X-ray diffraction analysis.

^[e] Isovaleraldehyde was used as electrophile and the temperature was lowered to -90°C.

^[f] Hexanal was used as electrophile and the temperature was lowered to -100°C; dr was determined by ¹H NMR analysis.

^[g] Products resulting of an intramolecular acyl transfer reaction obtained when the reaction conditions were not optimized^[13]

Transmetalation of stannylated oxazolidinones 3.

We then turned our attention to *N*-(α -tributylstannylorgano) oxazolidin-2-ones **3a**. A similar Sn-Li transmetalation/electrophilic trapping procedure was applied to these derivatives and the obtained results are presented in Table 2. With cyclohexanone as electrophile, diastereomer (*R,S*)-**3a** afforded product **11a** in 71% yield as a single diastereomer (entry 1). Suitable monocrystals from **11a** were obtained for X-ray diffraction analysis, allowing the determination of its (*S,S*) absolute configuration, meaning that the configuration was preserved under these experimental conditions. Starting from the (*S,S*)-**3a** diastereomer, the yield was found to be strongly lowered (entry 2, 47%), but (*S,S*)-**11a** was obtained once more as a unique isomer when the reaction temperature was maintained at -78°C (corresponding to a complete inversion of the configuration). The achievement of this reaction at -100°C afforded **11a** in good yield (77%), but as a mixture of diastereomers (entry 3, (*S,S*)/(*R,S*) = 26/74) due to a partial epimerization of the chiral anionic center. Since higher yields were obtained after the transmetalation/trapping sequence when starting from (*R,S*)-**3a** instead of (*S,S*)-**3a**, the (*R,S*)-**3** isomers were used in subsequent studies.

Starting from the (*R,S*)-**3b** diastereomer, bearing a butenyl group instead of an allyl group, product **11b** was also selectively obtained as a single isomer in good yield (entry 4, 82%). It was identified to be the (*S,S*) diastereomer by analogy to (*S,S*)-**11a**. When the transmetalation was performed on (*R,S*)-**3a** and the nonanone was used as the electrophilic trapping reagent, the (*S,S*)-**12a** product was isolated as a unique product but in 30% yield only, presumably because of the enolization reaction induced by the higher steric hindrance. The electrophilic trapping was also achieved with $\text{CO}_2(\text{g})$ and, after esterification under neutral conditions with TMSCHN_2 , **13a** was obtained in 76% yield with a strong (*S,S*) preference [(*S,S*)-**13a** / (*R,S*)-**13a** = 97/3].^[12]

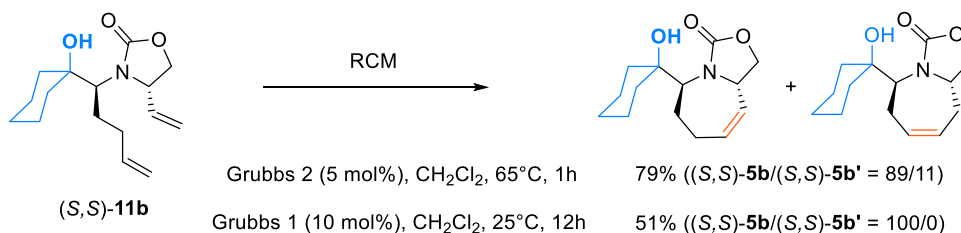
We then looked at the use of aldehydes as electrophiles and, taking into account the possible occurrence of an intramolecular acyl transfer, the reaction temperature was fixed to -90°C . Starting from (*R,S*)-**3a**, the reaction products were obtained with an exclusive (*S*)-configuration at the previous anionic center whatever the nature of the aldehyde (benzaldehyde, isovaleraldehyde or hexanal). In every cases, the (*S,S*)-products were obtained selectively in good yields (74–81%). Unfortunately, no

diastereoselectivity was observed for the addition on the enantiotopic faces of these aromatic or aliphatic aldehydes (dr = 58/42, 50/50 and 50/50, respectively). All these diastereomers were separated by column chromatography on silica gel. The absolute configuration of (*S,S,S*)-**14a** was based on that of (*S,S,S*)-**7a** (*vide infra*), and those of **15a** and **16a** were deduced from the comparison of their NMR spectra and their R_F with those of **14a**.

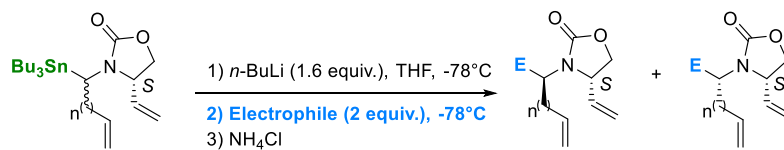
Finally, we have considered the introduction of an *n*-alkyl chain. While no reaction occurred when 1-iodohexane was used to trap the organolithium species, the desired coupling product **17a** was obtained with a moderate *R,S* selectivity if the organolithium intermediate was *in situ* converted into the corresponding organocopper reagent. The use of $\text{CuCN} \cdot 2 \text{LiCl}$ afforded **17a** in 71% yield with a (*R,S*)/(*S,S*) selectivity of 65/35, while CuCN or pentynyl copper furnished low yields and minor stereoselectivity improvement (70/30 and 87/13 ratios, respectively, see supporting information).

Then, we focused on the ring-closing metathesis reaction of dienyl oxazolidinones **11–17** to prepare the dehydropiperidines (or dehydroazepanes) similar to those obtained previously from **4a** (or **4b**) but with a different stereochemistry. Both Grubbs 1 and Grubbs 2 catalysts were considered for this reaction and the results are presented in Table 3.

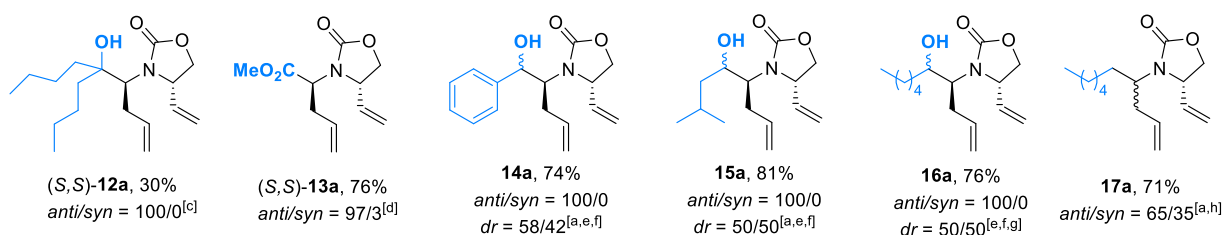
The absolute configuration of all products listed in Table 3 were determined thanks to X-ray diffraction of suitable monocrystals of (*S,S*)-**5a**, (*S,S*)-**5b** and (*S,S,S*)-**7a**. These reactions were monitored by TLC and conversion rates over 90% were observed for the formation of the six membered rings with isolated yields in the range of 80–95% using Grubbs 1 catalyst at 25°C for 1 to 3h or Grubbs 2 catalyst at 40°C for 1 to 6h. For the ring-closing metathesis of (*S,S*)-**11b** with Grubbs 2 catalyst at 40°C , the formation of a minor side product (identified as a regioisomer resulting from the migration of the double bond in the ring) was observed along with the formation of the expected one (Scheme 2). The (*S,S*)-**5b**/(*S,S*)-**5b'** ratio was found to be 89/11 by ^1H NMR. The attempts to separate (*S,S*)-**5b'** from (*S,S*)-**5b** were unsuccessful. However, by using Grubbs 1 (10 mol %) at room temperature for 12h, the (*S,S*)-**5b** product was obtained exclusively in 51% isolated yield (Scheme 2). Such an isomerization has been previously observed and explained by a reversible addition of a ruthenium hydride complex on the double bond.^[4,15]



Scheme 2. Ring-closing metathesis of (*S,S*)-**11b** using Grubbs 1 or Grubbs 2 catalysts.

Table 2. Sn-Li transmetalation followed by an electrophilic trapping of **3a** and **3b**.**3a**, *n* = 1; **3b**, *n* = 2**A. With cyclohexanone as electrophile**

Entry	Stannylated oxazolidinone	Products	Yield (%) (<i>anti</i> / <i>syn</i>) ^[a]
1		(X-ray ^[12])	71 (100/0)
2			47 (100/0)
3			77 (26/74) ^[b]
4			82 (100/0)

B. With other electrophiles starting from (R,S)-3a

^[a] The ratio *anti*/*syn* was determined by GC on the crude; *anti*/*syn* are defined by the relative position of the allyl (or butenyl) and the vinyl moieties; the configuration of the *anti* isomer is (*S,S*) if electrophile is a ketone, an aldehyde or CO₂ and (*R,S*) with a *n*-alkyl iodide.

^[b] The reaction was carried out at -100°C.

^[c] The reaction temperature was increased from -78°C to -50°C for dissolving nonanone.

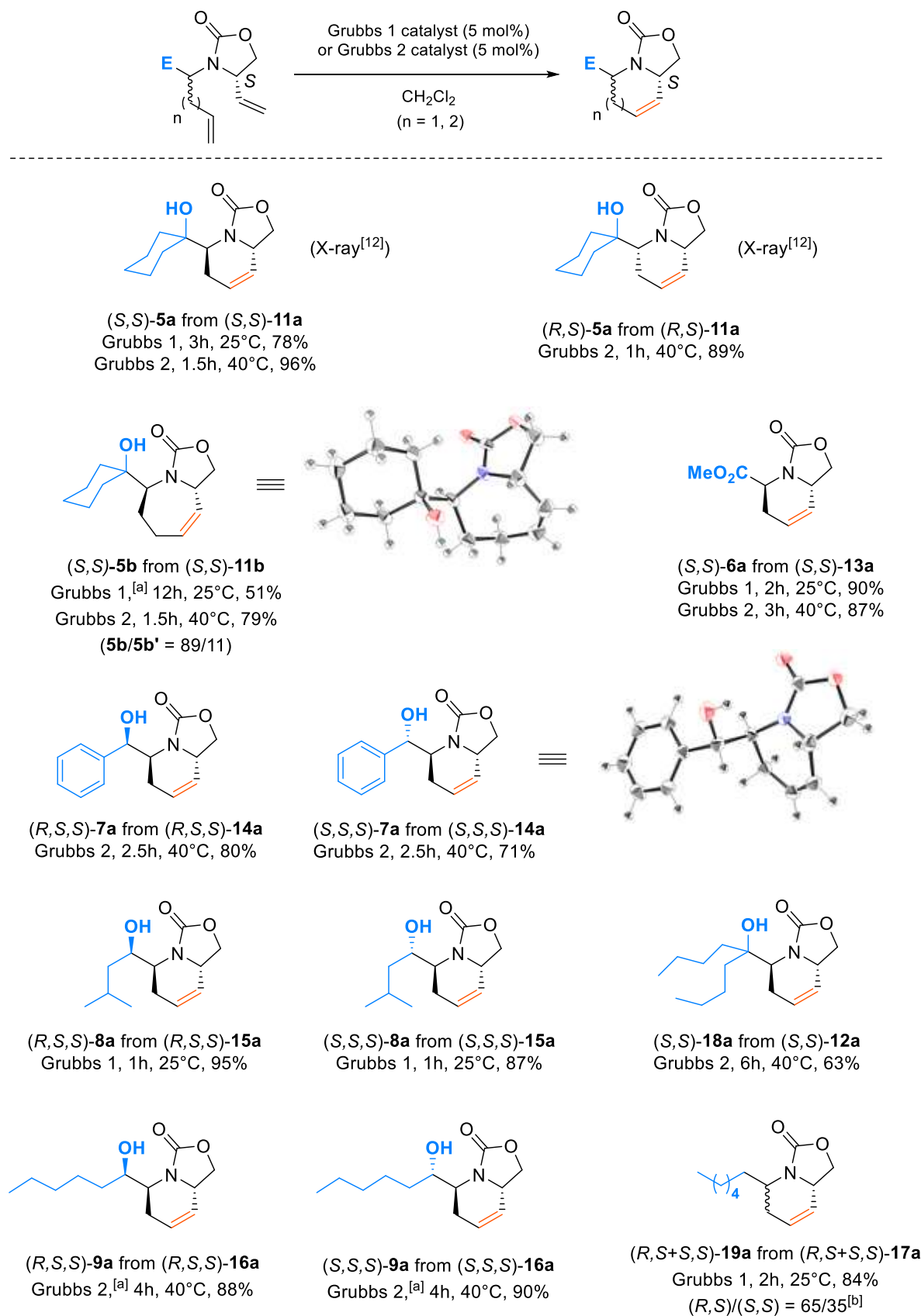
^[d] CO₂(g) was used and the crude was treated with TMSCHN₂ (1.3 equiv.), C₆H₆:MeOH=4:1.

^[e] The reaction temperature was -90°C to avoid the intramolecular acyl transfer reaction.

^[f] Diastereomers were separated by column chromatography on silica gel. The absolute configuration of (*S,S,S*)-**14a** (the minor one) was deduced from that of (*S,S,S*)-**7a** determined by X-ray diffraction. The absolute configurations of **15a** and **16a** were deduced from the comparison of their NMR data and *R_F*.

^[g] Determined by NMR on the crude.

^[h] Unseparated mixture of diastereomers; CuCN.2LiCl (0.5 equiv.) and 1-iodohexane (1.8 equiv.) were used; the configuration of **17a** was deduced from that of **19a**.

Table 3. Ring-closing metathesis of dienyloxazolidinones **11-17**.^[a] 10 mol% of Grubbs catalyst used.^[b] The diastereomers **(R,S)-19a** and **(S,S)-19a** were separated by chromatography. Their absolute configurations were confirmed by NOESY NMR experiments.

Discussion

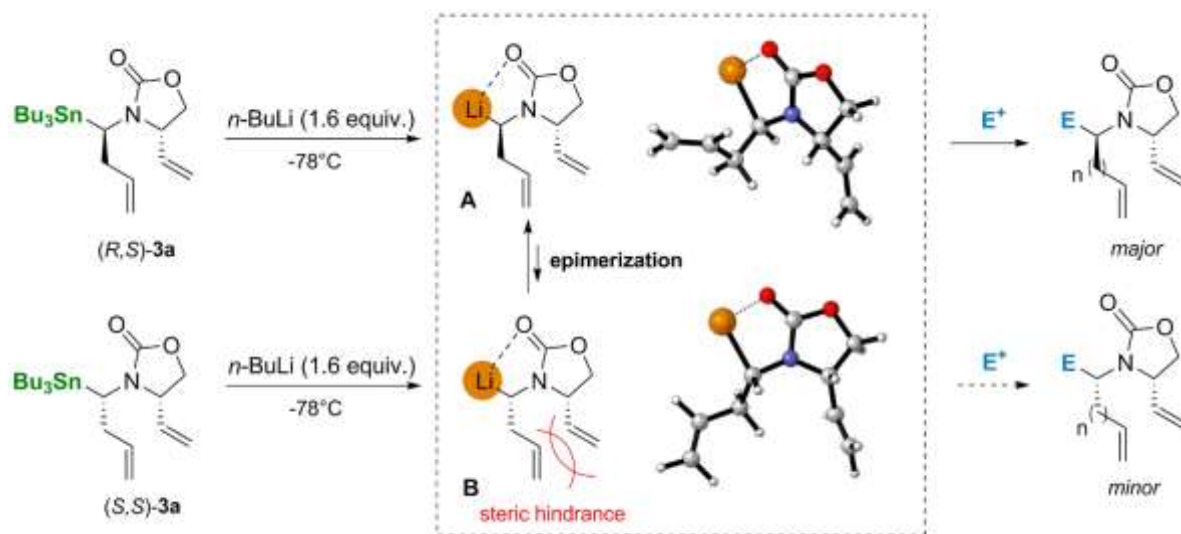
The results displayed in Table 2 indicate that the sequence involving a Sn-Li transmetalation reaction with *n*-BuLi followed by an electrophilic trapping applied to **3a** proceeds with a very high *anti* selectivity regardless of the configuration of **3a**. When the reaction is carried out at -78°C on diastereomer (*R,S*)-**3a**, the reaction leads selectively to the *anti* (*S,S*)-product. Starting from diastereomer (*S,S*)-**3a**, the selectivity is found to be temperature dependent. At -78°C or at -90°C, the *anti* (*S,S*)-product is also selectively obtained, while a mixture of *syn* and *anti* products in a 3/1 ratio was obtained when the reaction is performed at -100°C. This high diastereoselectivity can be explained by the occurrence of a thermodynamic resolution. Under the experimental conditions, the configurational stability of intermediates **A** and **B** is dependent on the relative position of the allyl (or butenyl) and the vinyl moieties (Scheme 3). When the reaction is carried out at -78°C on (*R,S*)-**3a**, the initially generated organolithium intermediate **A** is the more stable one and does not epimerize significantly into **B**. Therefore, intermediate **A** is trapped by the electrophile with complete retention of configuration. A similar reasoning can be held for (*R,S*)-**3b**. On the other hand, when starting from (*S,S*)-**3a**, the more sterically hindered organolithium intermediate **B** is initially generated. At -78°C, it epimerizes and leads to **A** along with side reaction products before the electrophilic trapping. The epimerization of the stereogenic centre is fast and the ratio of the product is a reflection of the ratio of the diastereomeric organolithium complexes.^[16] When the reaction is carried out at a lower temperature, the epimerization is slowed down and can be partially avoided, affording mixture of the two diastereomers. This qualitative mechanism based on the experimental results is fully supported by DFT calculations at the BMK/cc-pVDZ level of theory. Based on computed free energies at -78°C, and taking into account the

solvent effects, the intermediate **B** was found 1.5 kcal/mol less stable than **A** (Scheme 3). It means that under thermodynamic control, the occurrence of an equilibrium between the *syn* and *anti*-organolithium intermediates will result in a high prevalence of the *anti*-isomer (**A**) which would represent about 99 % of the total population. When the reaction is carried out at -100°C, the reaction remains partially under kinetic control affording a 3/1 mixture of diastereomers.

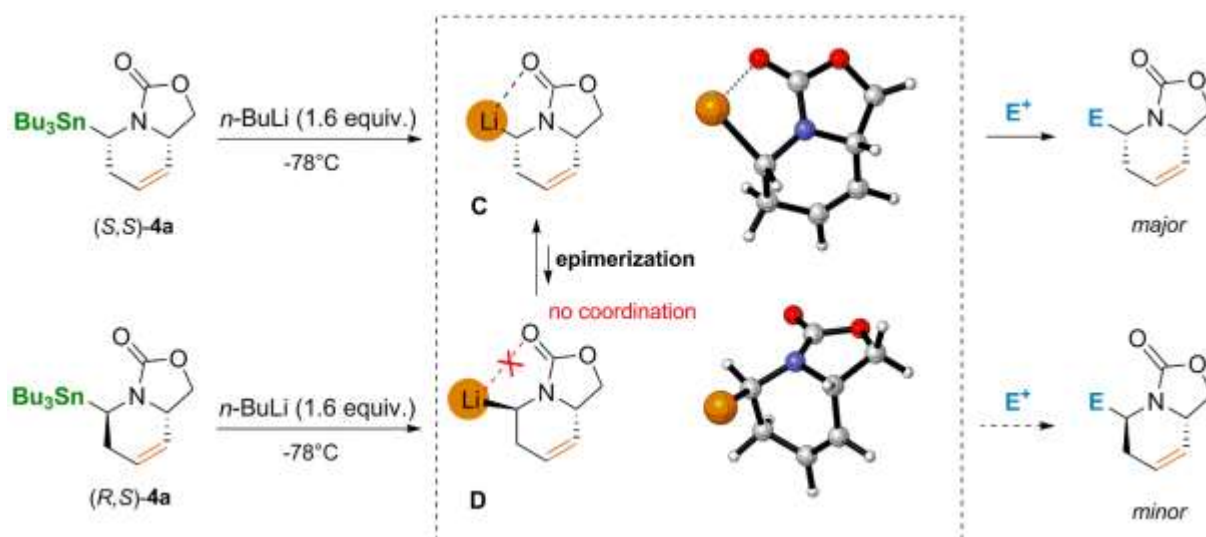
When 1-iodohexane was used as electrophile (Table 2 and supporting information), the conversion of the organolithium intermediate into an organocopper reagent is required. It induces a dramatic effect on the diastereoselectivity of the reaction, presumably due to a modification of the reaction mechanism of the copper intermediate with the electrophile, which may proceed by a single electron transfer.^{[17],[18]}

When considering the transmetalation of the fused ring systems tributylstannyl dehydropiperidine / oxazolidinone **4a**, an enhanced selectivity is observed. It leads to the *cis* (*R,S*)-product as unique diastereomer regardless of the configuration of the stannylated precursor, but with a drop in the yield when starting from the *trans* stannylated precursor (Table 1). In these cases, no trace of *trans* product can be observed by achieving the transmetalation of *trans* (*R,S*)-**4a** at temperatures as low as -100°C, indicating a strong preference for the *cis*-configuration in the anionic species. The situation is similar in dehydroazepanes series. The *cis* isomer is selectively obtained regardless of the configuration of **4b**, with a drop in the yields when starting from the *trans* precursor (Table 1, entries 3,4).

This high stereoselectivity can be explained by considering the structure of the organolithium intermediate. The *cis*-isomer furnished an intermediate which adopts a “ridge tile” shape and places the lithium in an adequate position for an intramolecular coordination with the oxazolidinone carbonyl.



Scheme 3. Equilibration of organolithium species generated from (*R,S*)-**3a** and (*S,S*)-**3a**, and their computed structures at the BMK/cc-pVDZ level of theory.



Scheme 4. Equilibration of organolithium species generated from (*R,S*)-**4a** and (*S,S*)-**4a**, and their computed structures at the BMK/cc-pVDZ level of theory

Indeed, the quantum mechanical calculations predict a short Li...O distance, 1.896 Å (Scheme 4, intermediate **C**), away below the sum of the van der Waals radii of lithium and oxygen (3.33 Å).

In the lithiated species obtained from the *trans*-isomer, the relative orthogonal positions of lithium and oxazolidinone carbonyl does not allow such stabilizing coordination (Scheme 4, intermediate **D**). According to the free energies calculated at the BMK/cc-pVDZ level of theory, the intermediate **D** is less stable than **C** by 14.7 kcal/mol in THF at -78°C. Therefore, while *cis* configuration is completely maintained when starting from *cis*-**4a**, a complete epimerization arises when starting from *trans*-**4a**. Here again, the epimerization reaction occurs very fast under the experimental conditions used.

The last part of the study (Table 3) points out the possibility to convert efficiently adducts **11a-17a** and **11b** into the corresponding dehydropiperidines and dehydroazepanes through ring-closing metathesis. Therefore, since adducts **11a-16a** and **11b** were obtained selectively as *anti* isomers, this strategy opens the route to *trans* dehydropiperidines.

Conclusion

The control of the configurational stability of the organolithium species obtained by transmetalation of stannylated oxazolidinones **3** and **4** by *n*-butyllithium allowed a highly diastereoselective synthesis of 2,6-disubstituted dehydropiperidines (and 2,7-disubstituted dehydroazepanes). The stereochemical trends are rationalized by considering the difference of stabilization brought by the lithium-carbonyl coordination, which was estimated by DFT calculations as 1.5 kcal/mol for the organolithium species derived from **3a** and as 14.7 kcal/mol for the organolithium species derived from **4a**.

Thus, an efficient stereodivergent synthesis of (*S,S*)-*trans* or (*R,S*)-*cis* 2,6-disubstituted dehydropiperidines has been achieved starting from **3a**. Using the reaction sequence: Sn-Li transmetalation/electrophilic trapping

followed by a ring-closing metathesis, the (*S,S*)-*trans* 2,6-disubstituted dehydropiperidines are exclusively obtained, while the inversion of the reaction sequence affords (*R,S*)-*cis* 2,6-disubstituted dehydropiperidines. The same strategy remains valid in dehydroazepanes series. Application of this methodology to the synthesis of bioactive polysubstituted piperidines and iminosugars C-glycosides is currently under investigation.

Experimental Section

The general procedures are fully described in supporting information.

Preparation and Characterization of Compounds

Organotin precursor **1** was obtained by a transacetalisation reaction^[11a,19] between diethoxymethyltributyltin and (*S*)-vinylglycinol protected as *N*-CO₂Me.^[20] Details of the procedure have been previously reported^[4] together with the identification of *cis* and *trans* isomers which was established on the basis of their NMR data.^[21] Compound **2** was obtained by ring opening of **1** by soft alkenylmetals in the presence of Lewis acids and converted into oxazolidinone **3** before being submitted to ring-closing metathesis to afford **4**. Compounds **2-4** have been already prepared and fully characterized.^[4]

General procedure for Sn-Li transmetalation and electrophilic trapping of **3** or **4**.

In a Schlenk tube, stannyl oxazolidinone (**1** mmol) was dissolved in THF (10 mL) and the solution was cooled to -78°C and kept under argon before slow dropwise addition of *n*-butyllithium (1 mL, 1.6 M in hexanes, 1.6 mmol) to the solution which turned to a slight yellow colour. Once this addition was complete, the solution was allowed to stir for 20 min at -78°C before dropwise addition of electrophile (2 mmol). The reaction was quenched at -78°C by addition of saturated aqueous NH₄Cl after 20 min. The aqueous phase was extracted with diethyl ether (3 x 20 mL) and the combined organic extracts were dried (MgSO₄),

filtered, and concentrated under vacuum. Purification was carried out by flash chromatography on silica gel.

General procedure for ring closing metathesis of dienyloxazolidinones 11-17 into bicyclic oxazolidinones 5-9 and 18-19.

To a solution of dienyloxazolidinones (0.65 mmol) in dichloromethane ($c = 0.005 \text{ mol.L}^{-1}$) was added Grubbs catalyst (0.032 mmol) and the reactive mixture was stirred at 25°C for Grubbs 1 catalyst and 40°C for Grubbs 2 catalyst. After complete conversion, DMSO (0.11 mL, 1.62 mmol) was added and the solution was stirred overnight. The reaction mixture was concentrated under reduced pressure and the purification was carried out by flash chromatography on silica gel.

All the characterization data and NMR spectra of new compounds are given in supplementary information.

CCDC 718901, 723326 and 736227 contain the supplementary crystallographic data for this paper. These data can be obtained free of charge via www.ccdc.cam.ac.uk/data_request/cif, or by emailing data_request@ccdc.cam.ac.uk, or by contacting The Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, UK; fax: +44 1223 336033.

Acknowledgements

We gratefully acknowledge the Université de Nantes, the "Centre National de la Recherche Scientifique" (CNRS). A.L. acknowledges the "Institut de Chimie des Substances Naturelles, ICSN" for a PhD grant.

References

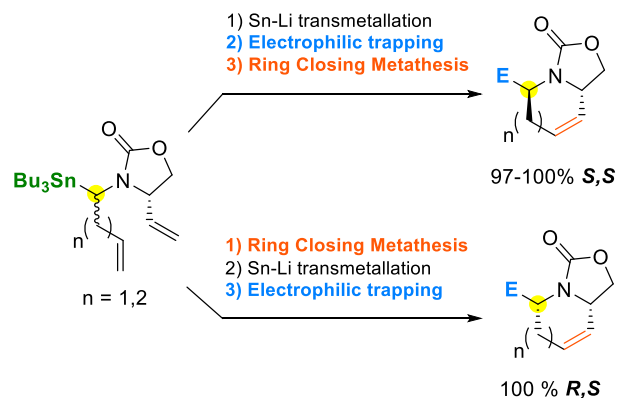
- [1] a) F.-X. Felpin, J. Lebreton, *Curr. Org. Chem.* **2004**, *1*, 83-109; b) J. Cossy, *Chem. Rec.* **2005**, *5*, 70-80; c) N. Toyooka, in *Asymmetric Synthesis of Nitrogen Heterocycles* (Ed.: J. Royer), Wiley, Weinheim, **2009**, pp. 95-138; d) R. Vardanyan, in *Piperidine-Based Drug Discovery* (Ed.: R. Vardanyan), Elsevier, **2017**, pp. 299-332; e) P. Goel, O. Alam, M. J. Naim, F. Nawaz, M. Iqbal, M. I. Alam, *Eur. J. Med. Chem.* **2018**, *157*, 480-502.
- [2] a) N. Asano, R. J. Nash, R. J. Molyneux, G. W. J. Fleet, *Tetrahedron Asymmetry* **2000**, *11*, 1645-1680; b) V. H. Lillielund, H. H. Jensen, X. Liang, M. Bols, *Chem. Rev.* **2002**, *102*, 515-554; c) P. Compain, O. R. Martin, *Iminosugars: Past, Present and Future*, John Wiley & Sons, Ltd, **2008**.
- [3] a) S. Mix, S. Blechert, *Adv. Synth. Catal.* **2007**, *349*, 157-160; b) G. Horne, F. X. Wilson, J. Tinsley, D. H. Williams, R. Storer, *Drug Discov. Today* **2011**, *16*, 107-118; c) P. Compain, *Molecules* **2018**, *23*, 1658.
- [4] A. Lumbroso, V. Coeffard, D. Gagneau, S. Stecko, I. Beaudet, J.-P. Quintard, E. Le Grogne, *Eur. J. Org. Chem.* **2016**, 5146-5159.
- [5] a) Y. Troin, M.-E. Sinibaldi, in *Asymmetric Synthesis of Nitrogen Heterocycles* (Ed.: J. Royer), Wiley, Weinheim, **2009**, pp. 139-186; b) A. G. Meyer, J. A. Smith, C. Hyland, C. C. Williams, A. C. Bissember, T. P. Nicholls, in *Progress in Heterocyclic Chemistry*, Vol. 28 (Eds.: G. W. Gribble, J. A. Joule), Elsevier, **2016**, pp. 579-622.
- [6] a) M. Pereyre, J.-P. Quintard, A. Rahm, *Tin in Organic Synthesis*, Butterworths, London, **1987**; b) C. Rim, D. Y. Son, *Arkivoc* **2006**, 265-291.
- [7] a) W. C. Still, C. Sreekumar, *J. Am. Chem. Soc.* **1980**, *102*, 1201-1202; b) J. S. Sawyer, T. L. Macdonald, G. J. McGarvey, *J. Am. Chem. Soc.* **1984**, *106*, 3376-3377; c) J. S. Sawyer, A. Kucero, T. L. Macdonald, G. J. McGarvey, *J. Am. Chem. Soc.* **1988**, *110*, 842-853; d) P. Lesimple, J.-M. Beau, P. Sinay, *J. C. S. Chem. Commun.* **1985**, 894-895; e) D. S. Matteson, P. B. Tripathy, A. Sarkar, K. M. Sadhu, *J. Am. Chem. Soc.* **1989**, *111*, 4399-4402; f) K. Tomooka, T. Igarashi, M. Watanabe, T. Nakai, *Tetrahedron Lett.* **1992**, *33*, 5795-5798; g) K. Tomooka, N. Komine, T. Nakai, *Tetrahedron Lett.* **1997**, *38*, 8939-8942.
- [8] a) R. E. Gawley, Q. Zhang, *J. Am. Chem. Soc.* **1993**, *115*, 7515-7516; b) R. E. Gawley, Q. Zhang, *Tetrahedron* **1994**, *50*, 6077-6088; c) I. Coldham, R. Hufton, D. J. Snowden, *J. Am. Chem. Soc.* **1996**, *118*, 5322-5323; d) J. M. Chong, S. B. Park, *J. Org. Chem.* **1992**, *57*, 2220-2222; e) V. Coeffard, I. Beaudet, J.-P. Quintard, E. Le Grogne, *Chirality* **2010**, *22*, 864-869; f) J.-P. Quintard, E. Le Grogne, in *Comprehensive Chirality*, Vol. 3 (Eds.: E. M. Carreira, H. Yamamoto), Elsevier, Amsterdam, **2012**, pp. 751-779.
- [9] P. Beak, A. Basu, D. J. Gallagher, Y. S. Park, S. Thayumanavan, *Acc. Chem. Res.* **1996**, *29*, 552-560.
- [10] a) W. H. Pearson, A. C. Lindbeck, *J. Am. Chem. Soc.* **1991**, *113*, 8546-8548; b) W. H. Pearson, A. C. Lindbeck, J. W. Kampf, *J. Am. Chem. Soc.* **1993**, *115*, 2622-2636; c) K. Brückner, *Chem. Ber.* **1993**, *126*, 1227; d) V. Coeffard, E. Le Grogne, I. Beaudet, M. Lepeltier, V. Léat-Crest, J.-P. Quintard, *Synthesis* **2006**, 4151-4158.
- [11] a) V. Coeffard, J.-C. Cintrat, E. Le Grogne, I. Beaudet, J.-P. Quintard, *J. Organomet. Chem.* **2006**, *691*, 1488-1497; b) V. Coeffard, I. Beaudet, M. Evain, E. Le Grogne, J.-P. Quintard, *Eur. J. Org. Chem.* **2008**, 3344-3351; c) V. Coeffard, E. Le Grogne, I. Beaudet, M. Evain, J.-P. Quintard, *J. Org. Chem.* **2009**, *74*, 5822-5838.
- [12] A. Lumbroso, I. Beaudet, L. Toupet, E. Le Grogne, J.-P. Quintard, *Org. Lett.* **2013**, *15*, 160-163.
- [13] Our investigations on the influence of the temperature and on the absolute configuration of

- the compounds resulting from the acyl transfer reaction are reported in supporting information.
- [14] a) D. A. Evans, M. M. Morrissey, R. L. Dorow, *J. Am. Chem. Soc.* **1985**, *107*, 4346-4348; b) C. Aciro, S. G. Davies, A. C. Garner, Y. Ishii, M.-S. Key, K. B. Ling, R. S. Prasad, P. M. Roberts, H. Rodriguez-Solla, C. O'Leary-Steele, A. J. Russell, H. J. Sanganee, E. D. Savory, A. D. Smith, J. E. Thomson, *Tetrahedron* **2008**, *64*, 9320-9344.
- [15] a) B. Schmidt, *J. Org. Chem.* **2004**, *69*, 7672-7687; b) S. Fustero, M. Sánchez-Roselló, D. Jiménez, J. F. Sanz-Cervera, C. del Pozo, J. L. Aceña, *J. Org. Chem.* **2006**, *71*, 2706-2714.
- [16] a) P. Beak, D. R. Anderson, M. D. Curtis, J. M. Laumer, D. J. Pippel, G. A. Weisenburger, *Acc. Chem. Res.* **2000**, *33*, 715-727; b) I. Coldham, S. Dufour, T. F. N. Haxell, J. J. Patel, G. Sanchez-Jimenez, *J. Am. Chem. Soc.* **2006**, *128*, 10943-10951; c) I. Coldham, J. J. Patel, S. Raimbault, D. T. E. Whittaker, *Chem. Commun.* **2007**, 4534-4536; d) I. Coldham, N. S. Sheikh, *Top. Stereochem.* **2010**, *26*, 253-293.
- [17] E. C. Ashby, D. Coleman, *J. Org. Chem.* **1987**, *52*, 4554-4565.
- [18] S. Mori, E. Nakamura, in *Modern Organocopper Chemistry* (Ed.: N. Krause), Wiley-VCH Verlag GmbH, **2002**, pp. 315-346.
- [19] a) J.-C. Cintrat, V. Léat, J.-L. Parrain, E. Le Grogne, I. Beaudet, L. Toupet, J.-P. Quintard, *Organometallics* **2004**, *23*, 943-945; b) J.-C. Cintrat, V. Léat-Crest, J.-L. Parrain, E. Le Grogne, I. Beaudet, J.-P. Quintard, *Eur. J. Org. Chem.* **2004**, 4251-4267.
- [20] A. Lumbroso, V. Coeffard, E. Le Grogne, I. Beaudet, J.-P. Quintard, *Tetrahedron Lett.* **2010**, *51*, 3226-3228.
- [21] a) J.-C. Cintrat, V. Léat-Crest, J.-L. Parrain, E. Le Grogne, I. Beaudet, L. Toupet, J.-P. Quintard, *Eur. J. Org. Chem.* **2004**, 4268-4279; b) G. Casella, F. Ferrante, G. Saielli, *Org. Biomol. Chem.* **2010**, *8*, 2711-2718.

FULL PAPER

Sn-Li Transmetalation of α -Aminoorganostannanes for the Stereoselective Synthesis of Substituted Dehydropiperidines and Dehydroazepanes*Adv. Synth. Catal.* **Year**, *Volume*, Page – Page

Alexandre Lombroso,^a Isabelle Beaudet,^a Jean-Paul Quintard,^a Cécile Fraisse,^a Nicolas Galland,^b Loïc Toupet,^b Erwan Le Grogne^a *



Accepted Manuscript