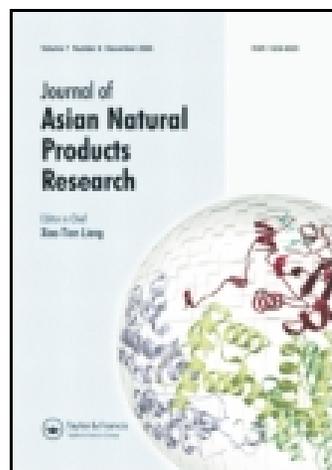


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Three new triterpenoid saponins from *Ilex pubescens*

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Three new triterpene saponins, ilexosaponins D–F (**1–3**), were isolated from the roots of *Ilex pubescens*. Their structures were elucidated as 3-*O*- β -D-glucopyranosyl(1 \rightarrow 3)- α -L-arabinopyranosyl urs-12,18-diene-24,28-dioic acid 28-*O*- β -D-glucopyranoside (**1**), 3-*O*- β -D-glucopyranosyl(1 \rightarrow 3)- α -L-arabinopyranosyl urs-12,18-diene-24,28-dioic acid (**2**), and 3-*O*- β -D-glucopyranosyl(1 \rightarrow 3)- α -L-arabinopyranosyl-30-hydroxyurs-12,19-diene-24,28-dioic acid 28-*O*- β -D-glucopyranoside (**3**) on the basis of the chemical and spectroscopic evidence.

Keywords: *Ilex pubescens*; triterpene saponin; ilexosaponins D–F

1. Introduction

The roots of *Ilex pubescens* Hook et Arn. (native name, Mao-Dong-Qing) are a traditional Chinese medicine (TCM) commonly used in South China for the treatment of cardiovascular disease and hypercholesterolemia [1]. Previous phytochemical investigation of its roots and leaves led to the isolation of glaberrides, triterpene saponins, and phenolics, specifically vomifoliol [2–7]. As a part of our program searching for bioactive saponins from TCMs, we present in this report the isolation and structural elucidation of three new triterpene saponins (**1–3**) (Figure 1) from the roots of *I. pubescens*.

2. Results and discussion

Compound **1** was obtained as a white amorphous powder. It showed positive Liebermann–Burchard and Molish reactions. The molecular formula was assigned as C₄₇H₇₂O₁₉ on the basis of the negative-ion HR-ESI-MS at *m/z* 939.4598 [M–H][–], confirmed by its ¹H and ¹³C NMR spectral

data (Table 1). The ¹H NMR spectrum indicated six methyl group signals at δ_{H} 1.72, 1.17, 1.21, 1.08, 1.73 (s, 3H each), and 1.06 (3H, d, *J* = 7.0 Hz) correlated with the carbons at δ_{C} 24.8, 14.6, 18.7, 22.0, 19.5, and 18.6 in the HSQC spectrum, respectively. There were two carboxyl carbons at δ_{C} 174.7 and 180.6 and two pairs of olefinic carbon signals at δ_{C} 126.7, 138.8, 133.8, and 136.0 in the ¹³C NMR spectrum. An olefinic proton at δ_{H} 5.71 (brt) correlated with the carbon at δ_{C} 126.7 in the HSQC spectrum (see Table 1) indicated the presence of an ursane-type triterpene aglycon. The UV absorbance at 225 nm suggested that compound **1** contains a heteroannular-conjugated diene system [8]. The second double bond located at C-18/C-19 was assigned by the interactions of the olefinic proton at δ_{H} 5.71 (brt, *J* = 3.0 Hz), methyl protons H-29 (δ_{H} 1.73) with the carbon signals C-18 (δ_{C} 133.8), and methyl protons H-29 (δ_{H} 1.73), H-30 (δ_{H} 1.06) with C-19 (δ_{C} 136.0) in the HMBC spectrum. In addition, the HMBC correlations of H-23 (δ_{H} 1.72), H-3

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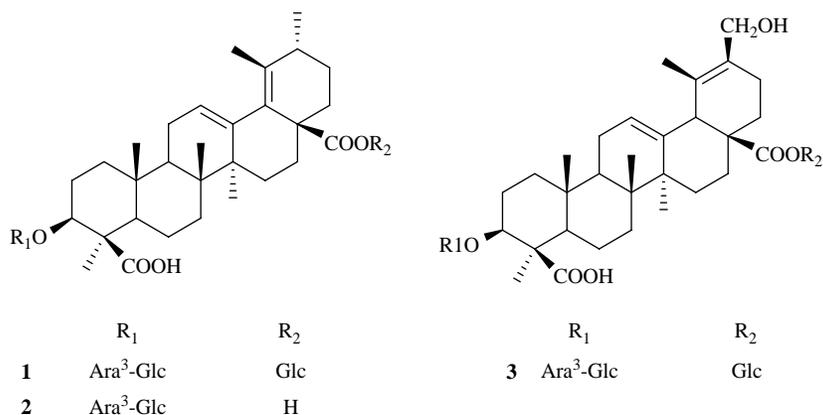


Figure 1. The structures of compounds 1–3.

(δ_{H} 3.34) with C-24 (δ_{C} 180.6) showed that the carboxyl carbon at δ_{C} 180.6 was located at C-24, therefore the other carboxyl carbon at δ_{C} 174.7 was attributed to C-28 (see Figure 2). After extensive NMR (^1H NMR, ^{13}C NMR, HMBC, and NOESY) spectral analysis (Table 1), the aglycon was established to be a 3 β -hydroxyurs-12, 18-dien-24, 28-dioic acid [9]. ESI-MS showed the ion peaks at m/z 777 $[\text{M}-\text{H}-162]^-$, 483 $[\text{M}-\text{H}-162-132-162]^-$, attributable to the sequential losses of a hexose, pentose, and hexose residues. Acid hydrolysis and gas chromatography (GC) experiment of **1** showed that the sugar moieties are L-arabinose and D-glucose [10]. H-1' (δ_{H} 4.95) of the L-arabinose was correlated with the C-3 (δ_{C} 88.9), and H-1'' (δ_{H} 5.16) of the terminal glucose was correlated with C-3' (δ_{C} 81.1) of L-arabinosyl in the HMBC spectrum. H-1''' (δ_{H} 6.28) of the D-glucose was correlated with the C-28 (δ_{C} 174.7) in the HMBC spectrum. The anomeric configuration of the L-arabinosyl and D-glucosyl moieties were α and β , respectively, from the $^3J_{1-2}$ coupling constant and NOESY experiment [10]. On the basis of this evidence, compound **1** was characterized as 3-*O*- β -D-glucopyranosyl(1 \rightarrow 3)- α -L-arabinopyranosyl urs-12,18-diene-24,28-dioic acid 28-*O*- β -D-glucopyranoside, named ilexsaponin D.

Compound **2** was obtained as a white amorphous powder. It showed positive Liebermann–Burchard and Molish reactions. The molecular formula was assigned as $\text{C}_{41}\text{H}_{62}\text{O}_{14}$ on the basis of the negative-ion HR-ESI-MS at m/z 777.4033 $[\text{M}-\text{H}]^-$. Acid hydrolysis of **2** afforded L-arabinose, D-glucose, and the aglycon, which is the same as those of **1**. Comparison of the NMR spectral data for **2** and **1** revealed that the two compounds are similar except that signals due to the D-glucopyranoside linked to C-28 in **1** were absent for **2**. This observation was supported by a relative downfield shift of C-28 (δ_{C} 180.6) of **2** in the ^{13}C NMR spectrum. Thus, the structure of **2** was determined as 3-*O*- β -D-glucopyranosyl (1 \rightarrow 3)- α -L-arabinopyranosyl urs-12,18-diene-24,28-dioic acid, named ilexsaponin E.

Compound **3** was obtained as a white amorphous powder. It showed positive Liebermann–Burchard and Molish reactions. The molecular formula was assigned as $\text{C}_{47}\text{H}_{72}\text{O}_{20}$ on the basis of the negative-ion HR-ESI-MS at m/z 955.4539 $[\text{M}-\text{H}]^-$. In the ^1H NMR spectrum, the aglycon part of **3** showed five methyl proton signals δ_{H} 1.67, 1.11, 1.19, 1.12, and 1.83 (s, 3H each), and one olefinic proton signal at δ_{H} 5.72 (brt). It also had two pairs of olefinic carbon signals at δ_{C} 128.1, 129.7, 131.1, and 137.6

Table 1. ^1H and ^{13}C NMR spectral data for 1–3 (500 MHz for ^1H and 125 MHz for ^{13}C , in $\text{C}_5\text{D}_5\text{N}$, δ in ppm, J in Hz).^a

Position	1		2		3	
	δ_{H}	δ_{C}	δ_{H}	δ_{C}	δ_{H}	δ_{C}
1	39.2	1.03 m 1.81 m	39.1	1.00 m 1.82 m	39.7	1.06 m 1.70 m
2	28.3	1.21 m 2.00 m	28.7	1.20 m 2.00 m	28.2	1.98 m
3	88.9	3.34 dd (14.0, 4.0)	88.9	3.33 dd (14.0, 4.0)	88.7	3.31 dd (14.0, 4.0)
4	48.2		48.7		48.0	
5	57.1	1.04 m	57.0	1.07 m	57.0	1.06 m
6	20.8	2.00 m 2.14 m	20.8	1.98 m 2.10 m	20.7	1.99 m 2.09 m
7	35.3	1.52 m 2.58 m	35.3	1.55 m 2.44 m	34.5	1.53 m
8	39.4		39.5		39.7	
9	47.6	1.52 m	47.4	1.55 m	47.8	1.57 m
10	37.7		37.7		37.8	
11	23.8	2.00 m	23.8	2.02 m	24.0	1.98 m
12	126.7	5.71 brt (3.0)	126.3	5.73 brt (3.0)	128.1	5.72 brt (3.0)
13	138.8		138.9		137.6	
14	45.0		45.1		43.7	
15	28.9	1.25 m 2.42 m	28.9	1.24 m 2.41 m	28.8	1.23 m 2.22 m
16	35.4	1.50 m 2.55 m	35.6	1.49 m 2.55 m	23.6	2.03 m
17	49.9		49.9		47.3	
18	133.8		133.7		51.8	3.63 s
19	136.0		136.0		131.1	
20	34.6	2.14 m	34.7	2.14 m	129.7	
21	26.8	1.25 m 2.07 m	26.8	1.23 m 2.07 m	24.4	2.56 m
22	31.0	1.68 m 2.14 m	31.2	1.68 m 2.13 m	33.0	1.85 m 2.14 m
23	24.8	1.72 s	24.9	1.71 s	24.7	1.67 s
24	180.6		180.4		180.9	
25	14.6	1.17 s	14.5	1.18 s	14.4	1.11 s
26	18.7	1.21 s	18.8	1.21 s	18.3	1.19 s
27	22.0	1.08 s	22.1	1.09 s	22.2	1.12 s
28	174.7		180.6		176.2	
29	19.5	1.73 s	19.5	1.76 s	16.8	1.83 s
30	18.6	1.06 d (7.0)	18.7	1.07 d (7.0)	63.0	4.39, 4.46 d (6.5)
3- <i>O</i> -sugar-Ara						
1'	104.8	4.95 d (5.5)	105.0	4.96 d (5.5)	104.8	4.93 d (5.5)
2'	73.4	4.57 m	73.4	4.55 m	73.4	4.55 m
3'	81.1	4.23 m	81.3	4.23 m	81.1	4.20 m
4'	68.3	4.34 m	68.5	4.33 m	68.3	4.35 m
5'	64.9	3.76 m 4.26 m	64.7	3.76 m 4.25 m	64.9	3.74 m 4.27 m
Glc						
1''	106.0	5.16 d (7.5)	106.2	5.20 d (7.5)	106.0	5.11 d (7.5)
2''	76.4	4.07 m	76.7	4.07 m	76.4	4.05 m
3''	78.1	4.17 m	78.3	4.16 m	78.1	4.16 m
4''	71.5	4.31 m	71.7	4.33 m	71.5	4.32 m
5''	78.1	3.80 m	78.3	3.83 m	78.1	3.77 m

Table 1 – continued

Position	1		2		3	
	δ_{H}	δ_{C}	δ_{H}	δ_{C}	δ_{H}	δ_{C}
6''	62.5	4.38 m 4.41 m	62.6	4.38 m 4.42 m	62.5	4.37 m 4.42 m
28- <i>O</i> -sugar-Glc						
1'''	95.8	6.28 d (8.0)			95.8	6.35 d (8.0)
2'''	74.1	4.18 t (8.5)			74.1	4.22 m
3'''	78.8	4.25 t (8.9)			78.9	4.30 m
4'''	71.1	4.31 t (9.2)			71.3	4.42 m
5'''	79.2	4.00 m			79.0	4.12 m
6'''	62.2	4.37 m 4.42 m			62.4	4.41 m 4.54 m

^aThe assignments were based upon ¹H NMR, ¹³C NMR, HSQC, HMBC, and NOESY spectra.

in the ¹³C NMR spectrum. UV absorbance at 203 nm of **3** suggested that there was no conjugation between the two double bonds. The double bond was located at C-12/C-13 (δ_{C} 128.1 and 137.6) according to the characteristic proton signal at δ_{H} 3.63 (1H, s, H-18) and carbon signal δ_{C} 51.8 (C-18) [9]. The correlations of H-29 (δ_{H} 1.83) with C-19 (δ_{C} 131.1) and C-20 (δ_{C} 129.7) in the HMBC experiment suggested the second double bond was located at C-19/C-20. The hydroxymethyl C-30 was attached to C-20 based on the following HMBC correlations of H-30 (δ_{H} 4.39 and 4.46) with C-19 (δ_{C} 131.1), C-20 (δ_{C} 129.7), and C-21 (δ_{C} 24.4). In addition, the two carbonyls at C-24 and C-28 were assigned by the HMBC

correlations of H-23 (δ_{H} 1.67), H-3 (δ_{H} 3.31) with C-24 (δ_{C} 180.9) and H-18 (δ_{H} 3.63) with C-28 (δ_{C} 176.2). The aglycon moiety was further confirmed to be olean-12, 19-diene-24, 28-dioic acid by comparison with the literature data [9]. Acid hydrolysis of **3** also afforded L-arabinose and D-glucose. Detailed NMR spectral comparison of the sugar moiety for **3** with **1** (see Table 1) demonstrated that both compounds possess same sugar sequence. Consequently, compound **3** was elucidated to be 3-*O*- β -D-glucopyranosyl(1 \rightarrow 3)- α -L-arabinopyranosyl-30-hydroxyurs-12,19-diene-24,28-dioic acid 28-*O*- β -D-glucopyranoside, named ilexsaponin F.

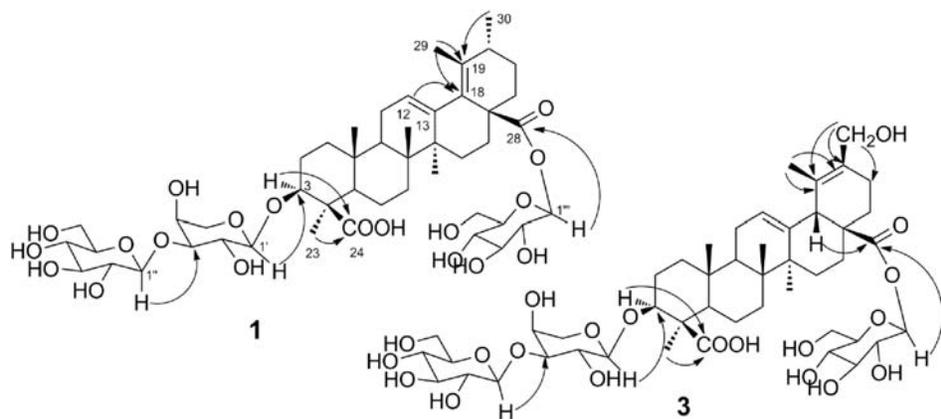


Figure 2. Selected HMBC correlations for compounds **1** and **3**.

3. Experimental

3.1 General experimental procedures

Optical rotations were measured on a JASCO P-1020 digital polarimeter (JASCO Corporation, Tokyo, Japan). Spots were visualized by spraying 10% H₂SO₄-EtOH followed by heating. UV spectra were obtained on a TU-1901 spectrometer (Beijing Purkinje General Instrument Co., Ltd, Beijing, China). IR spectra were obtained on a Perkin-Elmer 577 spectrometer (PerkinElmer, Waltham, MA, USA). NMR spectra were recorded on an Inova 500 spectrometer (Varian, Palo Alto, CA, USA), operating at 500 MHz for ¹H and 125 MHz for ¹³C. ESI-MS data were recorded on an MS Agilent 1100 series LC/MSD Trap Mass spectrometer (Agilent Technologies Inc., Santa Clara, CA, USA) and HR-ESI-MS data were obtained on a G1969A TOF-MS instrument (Agilent Technologies Inc.). Precoated silica gel GF254 plates (Qingdao Haiyang Chemical Co., Qingdao, China) were employed for thin layer chromatography. Column chromatography was carried out on silica gel (Qingdao Haiyang Chemical Co.) and macroporous resin D₁₀₁ (pore size B 13–14 nm, 26–60 mesh; Tianjin, China). High performance liquid chromatography (HPLC) separation was carried out on an ODS column (YMC-pack ODS-A, 250 × 10 mm, i.d. 5 μm, YMC, Kyoto, Japan) with an Alltech evaporative light scattering detector (Alltech, Dearfield, IL, USA). GC analysis was carried out on an Agilent 6890N gas chromatograph (Agilent, Waldbronn, Germany) using a HP-5 capillary column (28 m × 0.32 mm i.d.), with an FID detector (detector temperature: 260°C; column temperature: 180°C; carrier gas: N₂; flow rate: 40 ml/min).

3.2 Plant material

The roots of *I. pubescens* were collected in Guangzhou, Guangdong Province, China, in June 2010, and identified by Prof. Y.-X.

He, Xihua University. A voucher specimen (No. 20100810) has been deposited in the herbarium of Bioengineering College of Xihua University.

3.3 Extraction and isolation

The roots of *I. pubescens* (5 kg) were refluxed two times with 70% (v/v) EtOH, each for 2 h. After concentrated *in vacuo*, the residue was suspended in water, and partitioned with ethyl acetate and *n*-butanol successively. The *n*-butanol fraction (114 g) was further chromatographed over a macroporous resin D₁₀₁ column eluted initially with water, then with 30%, 50%, and 70% EtOH to give fractions A–C. Fraction B (34 g) was subjected to a silica gel column using CHCl₃-MeOH-H₂O (90:10:0.1–70:30:5) as an eluent to give fractions B₁–B₉. Fraction B₃ (2.9 g) was subjected to ODS open column chromatography (MeOH-H₂O 40:60 to 80:20) to afford fractions B_{3.1}–B_{3.4}. Fraction B_{3.1} (600 mg) was subjected to prep-HPLC (MeOH-H₂O, 40:60, UV detection at 210 nm, affording **1** (18 mg, t_R 16.7 min), **2** (12 mg, 20.7 min), and **3** (5 mg, 18.5 min), respectively.

3.3.1 *Ilexsaponin D* (**1**)

White amorphous powder; $[\alpha]_D^{22} + 32$ ($c = 0.12$, MeOH); UV (MeOH) λ_{\max} 225 nm; IR (KBr) ν_{\max} (cm⁻¹): 3432, 1724, 1681. ¹H NMR (C₅D₅N, 500 MHz) and ¹³C NMR (C₅D₅N, 125 MHz) spectral data, see Table 1; ESI-MS: m/z 939 [M-H]⁻; HR-ESI-MS m/z 939.4598 [M-H]⁻ (calcd for C₄₇H₇₁O₁₉, 939.4595).

3.3.2 *Ilexsaponin E* (**2**)

White amorphous powder; $[\alpha]_D^{30} + 34$ ($c = 0.12$, MeOH); UV (MeOH) λ_{\max} 225 nm; IR (KBr) ν_{\max} (cm⁻¹): 3432, 1725, 1680. ¹H NMR(C₅D₅N, 500 MHz) and ¹³C NMR(C₅D₅N, 125 MHz) spectral

data, see Table 1; ESI-MS: m/z 777 $[M-H]^-$; HR-ESI-MS m/z 777.4033 $[M-H]^-$ (calcd for $C_{41}H_{61}O_{14}$, 777.4066).

3.3.3 Ilexsaponin F (3)

White amorphous powder; $[\alpha]_D^{30} + 40$ ($c = 0.12$, MeOH). UV (MeOH) λ_{max} 203 nm; IR (KBr) ν_{max} (cm^{-1}): 3432, 1725, 1683. 1H NMR (C_5D_5N , 500 MHz) and ^{13}C NMR (C_5D_5N , 125 MHz) spectral data, see Table 1; ESI-MS: m/z 955 $[M-H]^-$; HR-ESI-MS m/z 955.4539 $[M-H]^-$ (calcd for $C_{47}H_{71}O_{20}$, 955.4544).

3.4 Acid hydrolysis of 1–3

Each saponin (5 mg) was heated in 5 ml of 10% HCl–dioxane (1:1) at 80°C for 4 h. After the dioxane was removed, the solution was extracted with EtOAc (3 ml \times 3) to yield the aglycon and the sugar, respectively. The sugar components in the aqueous layer obtained after acid hydrolysis of 1–3 were analyzed by GC analysis of the methyl sugar peracetates. The aqueous layer was evaporated and dissolved in anhydrous pyridine (100 ml), 0.1 M L-cysteinemethyl ester hydrochloride (200 ml) was added, and the resultant was warmed at 60°C for 1 h. The trimethylsilylation reagent HMDS–TMCS (hexamethyldisilazane–trimethylchlorosilane–pyridine, 2:1:10; Acros Organics, Geel, Belgium) was added and warmed at 60°C for 30 min. To the mixture, the thiazolidine derivatives were analyzed by GC for sugar identification. The retention times of L-arabinose (t_R , 5.32 min) and D-glucose (t_R , 12.39 min)

were confirmed by comparison with those of authentic standards [11].

Acknowledgments

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