

Preparation of *N, N'*-Disubstituted Piperazines from Bis(2-anilinoethyl) Phenylphosphonites

Oyo MITSUNOBU,^{*,*1} Takashi OHASHI,^{**} Motokazu KIKUCHI^{**}
and Teruaki MUKAIYAMA^{*}

^{*} *Laboratory of Organic Chemistry, Tokyo Institute of Technology, Ookayama, Meguro-ku, Tokyo*

^{**} *Technical Research Department, Bridgestone Tire Co., Ltd., Kodairashi, Tokyo*

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Preceding paper¹⁾ describes that bis(2-anilinoethyl) phenylphosphonite (I) is prepared by the reaction of 2, 3-diphenyl-1, 3, 2-oxazaphospholidine (II, R=C₆H₅) with an equimolar amount of 2-

anilinoethanol. A small amount of *N, N'*-diphenylpiperazine was always formed by the above reaction. The formation of the piperazine suggests that the phosphonite decomposes thermally into the piperazine. In this paper, the formation of *N, N'*-disubstituted piperazines by the thermal decomposition of various bis(2-aminoethyl) phenylphosphonites were investigated.

When bis(2-anilinoethyl) phenylphosphonite was

^{*1} Present address: College of Science and Engineering, Aoyama Gakuin University, Megurisawa-cho, Setagaya-ku, Tokyo.

1) O. Mitsunobu, T. Ohashi, M. Kikuchi and T. Mukaiyama, *This Bulletin*, **39**, 214 (1966).

TABLE 1. PREPARATION OF *N,N'*-DISUBSTITUTED PIPERAZINE

$\begin{array}{c} \text{HO}-\text{CH}_2 \\ \\ \text{HN}-\text{CH}_2 \\ \\ \text{R}' \end{array}$ $\begin{array}{c} \text{C}_6\text{H}_5\text{P} < \text{O} > \text{N} \\ \quad \quad \\ \text{R} \quad \quad \text{R}' \end{array}$	Temp. °C	Time hr	$\begin{array}{c} \text{R} \end{array}$	$\begin{array}{c} \text{R}' \end{array}$	$\begin{array}{c} \text{RN} \end{array}$	$\begin{array}{c} \text{NR}' \end{array}$	Yield %	Product			
								Mp °C or Bp °C/mmHg	Anal, %		
C_6H_5	210	3	C_6H_5	C_6H_5	C_6H_5		80*	163—165 (THF)	$\begin{Bmatrix} \text{F} \\ \text{C} \end{Bmatrix}$	$\begin{Bmatrix} \text{C} \\ \text{C} \end{Bmatrix}$	$\begin{Bmatrix} \text{H} \\ \text{N} \end{Bmatrix}$
$p\text{-CH}_3\text{C}_6\text{H}_4$	180	2	$p\text{-CH}_3\text{C}_6\text{H}_4$ <td>$p\text{-CH}_3\text{C}_6\text{H}_4$<td>$p\text{-CH}_3\text{C}_6\text{H}_4$</td><td></td><td>73</td><td>186—188 (toluene)</td><td>$\begin{Bmatrix} \text{F} \\ \text{C} \end{Bmatrix}$</td><td>$\begin{Bmatrix} \text{C} \\ \text{C} \end{Bmatrix}$</td><td>$\begin{Bmatrix} \text{H} \\ \text{N} \end{Bmatrix}$</td></td>	$p\text{-CH}_3\text{C}_6\text{H}_4$ <td>$p\text{-CH}_3\text{C}_6\text{H}_4$</td> <td></td> <td>73</td> <td>186—188 (toluene)</td> <td>$\begin{Bmatrix} \text{F} \\ \text{C} \end{Bmatrix}$</td> <td>$\begin{Bmatrix} \text{C} \\ \text{C} \end{Bmatrix}$</td> <td>$\begin{Bmatrix} \text{H} \\ \text{N} \end{Bmatrix}$</td>	$p\text{-CH}_3\text{C}_6\text{H}_4$		73	186—188 (toluene)	$\begin{Bmatrix} \text{F} \\ \text{C} \end{Bmatrix}$	$\begin{Bmatrix} \text{C} \\ \text{C} \end{Bmatrix}$	$\begin{Bmatrix} \text{H} \\ \text{N} \end{Bmatrix}$
$p\text{-CH}_3\text{OC}_6\text{H}_4$	180	1	$p\text{-CH}_3\text{OC}_6\text{H}_4$ <td>$p\text{-CH}_3\text{OC}_6\text{H}_4$<td>$p\text{-CH}_3\text{OC}_6\text{H}_4$</td><td></td><td>82</td><td>242—244 (anisole)</td><td>$\begin{Bmatrix} \text{F} \\ \text{C} \end{Bmatrix}$</td><td>$\begin{Bmatrix} \text{C} \\ \text{C} \end{Bmatrix}$</td><td>$\begin{Bmatrix} \text{H} \\ \text{N} \end{Bmatrix}$</td></td>	$p\text{-CH}_3\text{OC}_6\text{H}_4$ <td>$p\text{-CH}_3\text{OC}_6\text{H}_4$</td> <td></td> <td>82</td> <td>242—244 (anisole)</td> <td>$\begin{Bmatrix} \text{F} \\ \text{C} \end{Bmatrix}$</td> <td>$\begin{Bmatrix} \text{C} \\ \text{C} \end{Bmatrix}$</td> <td>$\begin{Bmatrix} \text{H} \\ \text{N} \end{Bmatrix}$</td>	$p\text{-CH}_3\text{OC}_6\text{H}_4$		82	242—244 (anisole)	$\begin{Bmatrix} \text{F} \\ \text{C} \end{Bmatrix}$	$\begin{Bmatrix} \text{C} \\ \text{C} \end{Bmatrix}$	$\begin{Bmatrix} \text{H} \\ \text{N} \end{Bmatrix}$
$p\text{-ClC}_6\text{H}_4$	180—200	1	$p\text{-ClC}_6\text{H}_4$ <td>$p\text{-ClC}_6\text{H}_4$<td>$p\text{-ClC}_6\text{H}_4$</td><td></td><td>57</td><td>238—240 (toluene)</td><td>$\begin{Bmatrix} \text{F} \\ \text{C} \end{Bmatrix}$</td><td>$\begin{Bmatrix} \text{C} \\ \text{C} \end{Bmatrix}$</td><td>$\begin{Bmatrix} \text{H} \\ \text{N} \end{Bmatrix}$</td></td>	$p\text{-ClC}_6\text{H}_4$ <td>$p\text{-ClC}_6\text{H}_4$</td> <td></td> <td>57</td> <td>238—240 (toluene)</td> <td>$\begin{Bmatrix} \text{F} \\ \text{C} \end{Bmatrix}$</td> <td>$\begin{Bmatrix} \text{C} \\ \text{C} \end{Bmatrix}$</td> <td>$\begin{Bmatrix} \text{H} \\ \text{N} \end{Bmatrix}$</td>	$p\text{-ClC}_6\text{H}_4$		57	238—240 (toluene)	$\begin{Bmatrix} \text{F} \\ \text{C} \end{Bmatrix}$	$\begin{Bmatrix} \text{C} \\ \text{C} \end{Bmatrix}$	$\begin{Bmatrix} \text{H} \\ \text{N} \end{Bmatrix}$
C_6H_5	180—190	3	C_6H_5 <td>CH_3<td>CH_3</td><td></td><td>25**</td><td>105—107/5</td><td>$\begin{Bmatrix} \text{F} \\ \text{C} \end{Bmatrix}$</td><td>$\begin{Bmatrix} \text{C} \\ \text{C} \end{Bmatrix}$</td><td>$\begin{Bmatrix} \text{H} \\ \text{N} \end{Bmatrix}$</td></td>	CH_3 <td>CH_3</td> <td></td> <td>25**</td> <td>105—107/5</td> <td>$\begin{Bmatrix} \text{F} \\ \text{C} \end{Bmatrix}$</td> <td>$\begin{Bmatrix} \text{C} \\ \text{C} \end{Bmatrix}$</td> <td>$\begin{Bmatrix} \text{H} \\ \text{N} \end{Bmatrix}$</td>	CH_3		25**	105—107/5	$\begin{Bmatrix} \text{F} \\ \text{C} \end{Bmatrix}$	$\begin{Bmatrix} \text{C} \\ \text{C} \end{Bmatrix}$	$\begin{Bmatrix} \text{H} \\ \text{N} \end{Bmatrix}$
C_2H_5	200	1	C_2H_5 <td>C_2H_5<td>C_2H_5</td><td></td><td>43</td><td>80—83/35—40</td><td>$\begin{Bmatrix} \text{F} \\ \text{C} \end{Bmatrix}$</td><td>$\begin{Bmatrix} \text{C} \\ \text{C} \end{Bmatrix}$</td><td>$\begin{Bmatrix} \text{H} \\ \text{N} \end{Bmatrix}$</td></td>	C_2H_5 <td>C_2H_5</td> <td></td> <td>43</td> <td>80—83/35—40</td> <td>$\begin{Bmatrix} \text{F} \\ \text{C} \end{Bmatrix}$</td> <td>$\begin{Bmatrix} \text{C} \\ \text{C} \end{Bmatrix}$</td> <td>$\begin{Bmatrix} \text{H} \\ \text{N} \end{Bmatrix}$</td>	C_2H_5		43	80—83/35—40	$\begin{Bmatrix} \text{F} \\ \text{C} \end{Bmatrix}$	$\begin{Bmatrix} \text{C} \\ \text{C} \end{Bmatrix}$	$\begin{Bmatrix} \text{H} \\ \text{N} \end{Bmatrix}$

F = Found, C = Calcd (); Solvent of recrystallization. * 36% of phenylphosphinic acid was obtained. ** 31% of diphenylpiperazine was obtained.

TABLE 2. PREPARATION OF 1,3,2-OXAZAPHOSPHOLININES

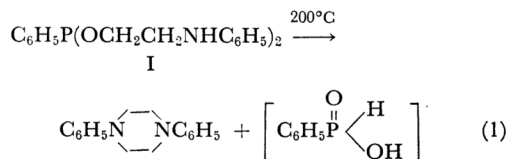
$\begin{array}{c} \text{H} \\ \\ \text{HOCH}_2\text{CH}_2\text{NR}' \end{array}$	Temp. °C	Time hr	$\begin{array}{c} \text{R} \\ \\ \text{RP} < \text{O} > \text{N} \\ \quad \quad \\ \text{R}' \quad \quad \text{R}' \end{array}$	Yield %	Bp °C/mmHg (Mp °C)	Anal, N%	
						Calcd	Found
C_6H_5	150—155	1	C_6H_5	H	115—118/4—5	8.38	8.62
C_6H_5	127	2.5	C_6H_5	CH_3	88—91/0.02	7.74	7.83
C_6H_5	150—170	3	C_6H_5	C_2H_5	79—81/1—2	7.18	7.29
C_6H_5	*	2.5	C_6H_5	C_6H_5	130—132/0.03 (75—76)	5.76	5.88
C_6H_5	150—170	4	C_6H_5	$p\text{-CH}_3\text{C}_6\text{H}_4$	163—165/0.1 (62—64)	5.46	5.69
C_6H_5	150—170	4	C_6H_5	$p\text{-CH}_3\text{OC}_6\text{H}_4$	158—162/0.05 (56—58)	5.13	5.15
C_6H_5	150—170	3	C_6H_5	$p\text{-ClC}_6\text{H}_4$	154—162/0.08 (53—55)	5.04	5.03
$\text{C}_2\text{H}_5\text{O}$	120	3	$\text{C}_2\text{H}_5\text{O}$	C_6H_5	88—91/0.02	6.64	6.79
***	**	5	$\text{C}_2\text{H}_5\text{O}$	C_6H_5	91—92/0.02	6.64	6.70

* The reaction was carried out in refluxing xylene.

** The reaction was carried out in refluxing toluene.

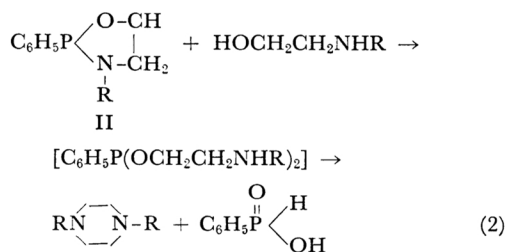
*** Ethyl *N*-phenyliminophosphite ($\text{C}_2\text{H}_5\text{O}-\text{P}=\text{NC}_6\text{H}_5$) was used.

heated at 200°C for 1 hr, *N, N'*-diphenylpiperazine was obtained in a 78% yield.



As shown in the previous paper,^{1,2} bis(2-anilinoethyl) phenylphosphonite is formed by the reaction of 2, 3-diphenyl-1, 3, 2-oxazaphospholidine with 2-anilinoethanol. Thus the direct method for the preparation of piperazines from 1, 3, 2-oxazaphospholidines and 2-aminoethanols were attempted.

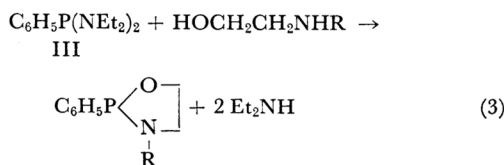
The reaction of equimolar amounts of 2, 3-diphenyl-1, 3, 2-oxazaphospholidine and 2-anilinoethanol at 210°C for 3 hr resulted in the formation of *N, N'*-diphenylpiperazine and phenyl phosphinic acid in 80% and 36% yields respectively (Eq. (2), R = C₆H₅).



Similarly, various *N, N'*-disubstituted piperazines were prepared and the results are summarized in Table 1.

It must be noted that when 2, 3-diphenyl-1, 3, 2-oxazaphospholidine and 2-methylaminoethanol were heated at 190°C for 3 hr, an unsymmetrical piperazine, *N*-phenyl-*N'*-methylpiperazine was obtained in 25% yield along with *N, N'*-diphenylpiperazine.

The starting materials, 3-substituted 2-phenyl-1, 3, 2-oxazaphospholidines were synthesized by the reaction of bis(diethylamino) phenylphosphine (III) with the corresponding *N*-substituted 2-aminoethanol as shown in the following equation (Eq. (3)). The results are summarized in Table 2.



Experimental

Preparation of 2-Phenyl-3-*p*-methoxyphenyl-1, 3, 2-oxazaphospholidine. A mixture of bis(diethylamino) phenylphosphine (III, 5.04 g, 0.02 mol) and 2-*p*-anisidinoethanol (3.34 g, 0.02 mol) was heated at 150–170°C for 4 hr. The mixture was distilled to give 2-phenyl-3-*p*-methoxyphenyl-1, 3, 2-oxazaphospholidine (5.00 g, 91.3%, bp 174–178°C/0.1–0.15 mmHg) which was solidified on standing. Recrystallization from ether gave mp 41–44°C.

Found: N, 5.15%. Calcd for C₁₅H₁₆NO₂P: N, 5.13%.

Similarly, several 1, 3, 2-oxazaphospholidines were prepared. The results are summarized in Table 2.

Preparation of *N, N'*-Diphenylpiperazine. A mixture of 2, 3-diphenyl-1, 3, 2-oxazaphospholidine (2.43 g, 0.01 mol) and 2-anilinoethanol (1.37 g, 0.01 mol) was heated at 180°C for 3 hr. Benzene (10 ml) was added to the mixture and *N, N'*-diphenylpiperazine (1.5 g) was removed by filtration. Aqueous sodium hydroxide (5%) was added to the filtrate and the solution was concentrated to give *N, N'*-diphenylpiperazine (0.41 g). Total yield of *N, N'*-diphenylpiperazine was 80% (1.91 g), mp 160–163°C. To the aqueous layer, concentrated hydrochloric was added and the solution was concentrated to give phenylphosphinic acid (36%). Recrystallization from water gave mp 71–72°C.

Similarly, some *N, N'*-diarylpiperazine were obtained as shown in Table 1.

Preparation of *N, N'*-Diethylpiperazine. A mixture of 2-phenyl-3-ethyl-1, 3, 2-oxazaphospholidine (2.48 g, 0.0127 mol) and 2-ethylaminoethanol (1.14 g, 0.0127 mol) was heated at 200°C for 1 hr. Ether and aqueous sodium hydroxide were added to the mixture. The ether layer was dried and distilled to give *N, N'*-diethylpiperazine (0.78 g, 43%, bp 80–83°C/35–40 mmHg, *n*_D²⁵ 1.4541. Found: C, 67.71; H, 13.00; N, 19.75%).

Preparation of *N*-Phenyl-*N'*-methylpiperazine. A mixture of 2, 3-diphenyl-1, 3, 2-oxazaphospholidine (4.6 g, 0.03 mol) and 2-methylaminoethanol (2.25 g, 0.03 mol) was heated at 180–190°C for 3 hr. Benzene (10 ml) was added to the mixture and *N, N'*-diphenylpiperazine (0.07 g, mp 160–163°C) was removed by filtration. Aqueous sodium hydroxide (1.2 g. NaOH in 10 ml of H₂O) was added to the filtrate. Benzene layer was dried and distilled to give *N*-phenyl-*N'*-methylpiperazine (1.31 g, 25%, bp 100–110°C/4–6 mmHg. Redistillation gave an analytical sample, bp 105–107°C/5 mmHg. Found: C, 74.23; H, 9.08; N, 15.41%). From the residue of the distillation, *N, N'*-diphenylpiperazine was obtained (0.16 g, mp 160–163°C).

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