AN UNUSUAL SYNTHESIS

OF 1,1,4,4-TETRAARYLBUTATRIENES

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1,1,4,4-Tetraarylbutatrienes are obtained from acetylenic 1,4-diols or their ethers under reduction conditions by the action of $SnCl_2$ in hydrochloric or sulfuric acid, hydriodic acid in acetic acid, KI in sulfuric acid and alcohol, and PX_3 (X=Cl, I) in the presence of bases [1]. We are the first to report the formation of cumulenes (IIIa) and (IIIb) by the action of 18% ethanolic H_2SO_4 on 1,1,4,4-tetraaryl-2-butyne-1,4-diols (Ia) and (Ib). The electron donor required for reduction of the dicarbocation to the cumulene is presumably SO_2 formed upon heating ethanol with H_2SO_4 [2, 3].

$$\begin{array}{c} Ar \\ C-C \equiv C-C \\ Ar' \quad OH \\ Ar' \quad HO \\ Ar' \\ Ar$$

 $Ar = Ar' = Ph (I) - (IIIa); Ar = Ph, Ar' = p-MeC_6H_4(I) - (IIIb).$

1.1,4,4-Tetraphenylbutatriene (IIIa). A sample of 2 g glycol (Ia) was heated with 18% ethanolic $\rm H_2SO_4$ for 10 h at 60-70°C. White crystals of (IIa) began to precipitate after 3-5 min, mp 116-117°C [4]. PMR spectrum in $\rm CCl_4$ (δ , ppm): 1.17 m (Me), 3.48 q (OCH₂), 7.28 and 7.47 m (Ph). The crystals gradually yellowed upon heating. The crystals were filtered off, washed with aqueous ammonia and ethanol to give 1 g (80%) tetraphenylbutatriene (IIIa), mp 235°C [1]. Found: C, 94.37; H, 5.75%. Calculated for $\rm C_{28}H_{20}$: C, 94.38; H, 5.61%. PMR spectrum in $\rm CS_2$ and $\rm CDCl_3$ (δ , ppm): 7.32 and 7.51 m (Ph). UV spectrum in $\rm CHCl_3$: $\lambda_{\rm max}$ 420 nm. Mass spectrum: m/z 357.

LITERATURE CITED

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