An efficient one-pot synthesis of 3,4-dihydropyrimidin-2-(*1H*)-ones/thiones catalysed by 2-thienylboronic acid

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A simple efficient one-pot three-component condensation for synthesis of 3,4-dihydropyrimidin-2-(*1H*)-ones/thiones in good to excellent yields using 2-thienylboronic acid as a catalyst was described.

Keywords: three-component condensation, 2-thienylboronic acid, Biginelli reaction, 3,4-dihydropyrimidin-2-(1H)-ones/thiones

Recently, the synthesis of 3,4-dihydropyrimi-din-2(*1H*)-ones/ thiones (DHPMs) has attracted interest of organic chemists. This interest can be attributed to 3,4-dihydropyrimidin-2(1*H*)ones/thiones showing potent therapeutical and pharmacological properties, such as anti-inflammatory, antitumor, and antibacterial activity. Some have been successfully used as calcium channel blockers¹, antihypertensive agents², and a_{1a} -antagonists³. Several alkaloids containing the dihydropyrimidine core unit, exhibiting anticancer activity and antiviral activities, have been studied. The DHPM monastrol (**1**, Fig. 1) as a novel cell-permeable molecule can block normal bipolar spindle assembly in mammalian cells causing cell cycle arrest^{4,5}. Thus the synthesis and investigation of dihydropyrimidines are of much current importance.

In the past, the Biginelli reaction had not attracted much attention. In 1978, Khanna first discovered the cardiovascular activity of Biginelli compounds arousing interest. This reaction suffers from long reaction times, acidic reaction conditions and low-yields of the products. A variety of catalysts have been reported to improve the efficiency of the Biginelli reaction, such as NiCl₂·6H₂O,⁷*p*-TsOH,⁸LaCl₃·7H₂O,⁹ Cu(OTf)₂,¹⁰ FeCl₃,¹¹ ZrCl₄,¹² Bi(OTf)₃,¹³ [bmim][FeCl₄],¹⁴ TCICA,¹⁵ NH₂SO₃H,¹⁶ CuI,¹⁷ CuCl₂·H₂O,¹⁸ BiCl₃,¹⁹ Mn(OAc)₃·2H₂O,²⁰ InBr₃,²¹ BF₃·OEt₂,²² LiClO₄,²³ YbOTf,²⁴ NH₄Cl,²⁵ InCl₃²⁶ or MTSA²⁷ and so on. Asymmetric syntheses using such as CeCl₃/InCl₃,²⁸ Yb(OTf)₃²⁹ attached to chiral ligands were recently reported. Other studies using microwave irradiation,³⁰ solid phase reagents,³¹ and polymer-supported catalysts,³² have been also reported.

Boronic acids are effective catalysts for the Biginelli reaction,^{33,34} but to our knowledge, heterocyclic boronic acid, ferrocenic boronic acid or substituted phenylboronic acids have not been tested as catalysts in the Biginelli reaction. We now report our research on arylboronic acids as catalysts for the Biginelli reaction. We found that 2-thienylboronic acid as a simple efficient catalyst for the synthesis of 3,4-dihydropyrimidin-2-(*1H*)-ones/thiones, the yield was up to 97%. A variety of new dihydropyrimidines were synthesised using 2-thiophenylboric acid as a catalyst in 5 hours.

Results and discussion

We optimised the reaction conditions in terms of catalyst, solvent and reaction time. 4-Chloro-benzaldehyde was selected as the model substrate and acetonitrile which is less toxic, and



Fig. 1 3,4-dihydropyrimidin-2(1*H*)-ones/thiones.

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is commonly used in Biginelli reaction, was chosen as solvent. The results are summarised in table 1. Initially, the combination of 4 (0.2 mol%), 5 (0.3 mol%), 6 (0.2 mol%) and phenylboronic acid(0.02 mol%) as catalyst were selected for study, and the desired 1a was isolated in 54% yield (entry 1). Arylboronic acids with electron-donating group showed less improvement (entries 2 and 3), while having electronwithdrawing group exhibited good activity, furnishing 1a in 74-95% yield (entries 4-6). The above results indicated that the structure of the acid was crucial for the present reaction. At the same time, we examined naphthalen-1-ylboronic acid (2g) and 1,1'-ferrocenediboronic acid (2h), yet both were not as efficient as 3,4-difluorophenylboronic acid (entry 7 and 8). When we tested 2-thienylboronic acid (2i), we were surprised that the yield rose to 97% (entry 9). Then the effects of the solvent on the reaction were studied and acetonitrile was found to be the best. Finally, we studied the reaction times and found the perfect reaction time was 5 hours (entry 9). On the basis of these investigations, the preferred reaction condition is ethyl acetoacetate:aldehydes:urea/thiourea:2-thienylboronic acid = 1:1:1.5:0.1 in CH₃CN (5 mL) 85 °C for 5 hours.

With the optimised conditions in hand, a variety of new dihydropyrimidines were synthesised in good to excellent yields, as show in Table 2. We found that aromatic aldehydes all gave higher yields than heteroaromatic aldehydes.

In conclusion, we have developed a simple and efficient method for a one-pot three-component synthesis of 3,4-dihy-dropyrimidin-2(1H)-ones/thiones using 2-thienylboronic acid as the catalyst. This process may be useful in the pharmaceutical and biochemical fields.

Experimental

¹H spectra and ¹³C NMR spectra were recorded on a Bruker Avance III 500 MHz spectrometer and using *d*-DMSO (500 MHz for ¹H or 125 MHz for ¹³C, respectively) as solvent. IR spectra (film) were recorded on a NEXUS FI/IR spectrometer. Melting points were taken on Büchi M-560. Mass spectra (MS) were carried out on Varian1200 and measured by the EI method. The reaction mixture was monitored by TLC on silica gel plates (60 F-254). Microanalyses were performed on a LECO CHNS-932 Elemental Analyser. All boronic acids were purchased from Fluka (Buchs, Switzerland) chemical company.

Synthesis of 3,4-dihydropyrimi din-2(1H)-ones/thiones (1a–n); general procedure

The mixture of ethyl acetoacetate (2 mmol), aromatic aldehyde (2 mmol), urea/thiourea (3 mmol), CH_3CN (5 mL) and 2-thienylboronic acid (0.1 mmol) was refluxed for 5 hours. The reaction mixture was monitored by TLC. After completion, the reaction mixture was poured into crushed ice, filtrated washed with 95% ethanol, dried, and recrystallised from 95% ethanol. The structures and yields of the products are given in Table 2.

5-*Ethoxycarbonyl*-6-*methyl*-4(4-*hydroxy*-3-*nitrophenyl*)-3, 4*dihydropyrimidin*-2(1*H*)-*one* (**1e**): Yellow solid; IR (film) 1697, 1643 cm⁻¹; ¹H NMR (500 MHz, d-DMSO) δ 1.09 (t, J = 7.0 Hz, 3H), 2.25 (s, 3H), 3.99 (m, 2H), 5.14 (s, 1H), 7.12 (d, J = 8.6 Hz, 1H), 7.41 (dd, J = 2.2, 8.6 Hz, 1H), 7.71 (d, J = 2.2 Hz, 1H), 7.77 (d, J = 3.4 Hz, 1H), 9.27 (s, 1H), 10.97 (s, 1H); ¹³C NMR (125 MHz, d-DMSO) δ 165.1, 151.7, 151.3, 148.7, 136.2, 136.0, 133.3, 122.7, 119.4, 98.6, 59.3,

Table 1 Optimisation for synthesis of 3,4-dihydropyrimidin-2(1H)-ones/thiones (DHPMs)^a



Entry	Catalyst	Solvent	Time/h	Yield ^b /%
1	2a	CH₃CN	5	54
2	2b	CH ₃ CN	5	21
3	2c	CH ₃ CN	5	10
4	2d	CH ₃ CN	5	76
5	2e	CH ₃ CN	5	95
6	2f	CH ₃ CN	5	75
7	2g	CH ₃ CN	5	Trace
8°	2ĥ	CH ₃ CN	5	Trace
9	2i	CH ₃ CN	5	97
10	2i	CH ₂ Cl ₂	5	23
11	2i	MeOH	5	34
12	2i	EtOH	5	77
13	2i	Toluene	5	54
14	2i	CH ₃ CN	1	30
15	2i	CH ₃ CN	2.5	82
16	2i	CHICN	3	95

2h

2i

^aReaction conditions: **4** (0.2 mol%), **5** (0.3 mol%), **6** (0.2 mol%) catalyst (0.02 mol%), Solvent (5 mL), reflux. blsolated yield. °Reaction conditions: 4 (0.2 mol%), 5 (0.3 mol%), 6 (0.2 mol%) catalyst (0.01 mol%), Solvent (5 mL), reflux.

53.0, 17.8, 14.0; MS m/z 321 (M⁺); Anal. Calcd for C₁₄H₁₅N₃O₆: C, 52.32; H, 4.72; N, 13.08. Found: C, 52.28; H, 4.64; N, 12.97%.

Ethyl-4-(4-hydroxy-3-nitrophenyl)-6-methyl-2-thioxo-1,2,3,4tetrahydropyrimidine-5-carboxylate (1f): Yellow solid; IR (film) 1709, 1654 cm⁻¹; ¹H NMR (500 MHz, d-DMSO) δ 1.11 (t, J = 7.0 Hz, 3H), 2.30 (s, 3H), 4.02 (m, 2H), 5.16 (s, 1H), 7.14 (d, J = 8.6 Hz, 1H), 7.38 (dd, J = 2.2, 8.6 Hz, 1H), 7.70 (d, J = 2.2 Hz, 1H), 9.65 (d, J = 3.2 Hz, 1H), 10.40 (s, 1H) ¹³C NMR (125 MHz, d-DMSO) δ 174.6, 165.3, 152.1, 145.9, 136.6, 135.1, 133.7, 123.4, 120.0, 100.6, 60.1, 53.4, 17.6, 14.4; MS *m/z* 337 (M⁺); Anal. Calcd for C₁₄H₁₅N₃O₅S: C, 49.84; H, 4.49; N, 12.46. Found: C, 49.72; H, 4.31; N, 12.38%.

Ethyl-4-(3,5-dibromo-4-hydroxyphenyl)-6-methyl-2-oxo-1,2,3,4tetrahydropyrimidine-5-carboxylate (1g): White solid; IR (film) 3104, 1684 cm^{-1} ; ¹H NMR (500 MHz, d-DMSO) δ 1.12 (t, J = 7.0 Hz, 3H), 2.26 (s, 3H), 4.01 (m, 2H), 5.08 (s, 1H), 7.33 (s, 2H), 7.74 (s, 1H), 9.26 (s, 1H); ^{13}C NMR (125 MHz, d-DMSO) δ 165.6, 152.2, 150.3, 149.4, 139.6, 139.6, 130.5, 112.2, 99.0, 59.8, 53.2, 18.3, 18.2, 14.5; MS m/z 432 (M⁺); Anal. Calcd for C₁₄H₁₄Br₂N₂O₄: C, 38.90; H, 3.27; N, 6.48. Found: C, 38.80; H, 3.18; N, 6.32%.

Ethyl-4-(3,5-dibromo-4-hydroxyphenyl)-6-methyl-2-thioxo-1,2,3,4tetrahydropyrimid ine-5-carboxylate (1h): White solid; IR (film) 3307, 1667 cm⁻¹; ¹H NMR (500 MHz, d-DMSO) δ 1.12 (t, J = 7.0 Hz, Table 2 Synthesis of 3,4-dihydropyrimidin-2(1H)-ones/thiones via 2-thiophenylboric acid cascade process^a



^aReaction conditions: **2i** (0.02 mol%), **5** (0.3 mol%), **6** (0.2 mol%), 7 (0.2 mol%), CH₃CN (5 mL), reflux, 5 h. ^bMelting point measured. ° Melting point from ref.27.

3H), 2.30 (s, 1H), 4.03 (m, 2H), 5.11 (s, 1H), 7.32 (s, 1H), 9.62 (s, 1H), 10.40 (s, 1H); ¹³C NMR (125 MHz, d-DMSO) δ 174.7, 165.4, 150.7, 146.0, 138.2, 130.7, 112.3, 100.5, 60.2, 53.2, 53.1, 17.7, 17.6, 14.5; MS m/z 448 (M⁺); Anal. Calcd for C₁₄H₁₄Br₂N₂O₃S: C, 37.51; H, 3.16; N, 6.25. Found: C, 37.40; H, 3.02; N, 6.10%.

Ethyl-6-methyl-2-oxo-4-(4-(trifluoromethyl) phenyl)-1,2,3,4-tetrahydropyrimidine-5-carboxylate (**1i**): White solid; IR (film) 3120, 1707 cm⁻¹; ¹H NMR (500 MHz, d-DMSO) δ 1.09 (t, J = 7.0 Hz, 3H), 2.27 (s, 3H), 3.99 (q, J = 7.0, J = 7.0 Hz, 2H), 5.24 (s, 1H), 7.46 (d, J = 8.0 Hz, 2H), 7.71 (d, J = 8.1 Hz, 2H), 7.83 (d, J = 3.1 Hz, 2H), 9.30 (s, 1H); ¹³C NMR (125 MHz, d-DMSO) δ 165.6, 152.4, 149.7, 149.5, 128.6, 128.2, 127.6, 126.0, 125.9, 123.3, 99.0, 59.8, 54.1, 18.2, 14.5; MS *m/z* 328 (M⁺); Anal. Calcd for C₁₅H₁₅F₃N₂O₃: C, 54.86; H, 4.62; N, 8.53. Found: C, 54.71; H, 4.46; N, 8.47%.

Ethyl-6-methyl-2-oxo-4-(3-(trifluoromethyl) phenyl)-1,2,3,4-tetrahydropyrimidine-5-carboxylate (**1j**): White solid; IR (film) 2982, 1691 cm⁻¹; ¹H NMR (500 MHz, d-DMSO) δ 1.08 (t, J = 7.0 Hz, H), 2.27 (s, 3H), 3.99 (m, 2H), 5.26 (d, J = 2.7 Hz, 1H), 7.59 (m, 4H), 7.83 (d, J = 3.2 Hz, 1H), 9.31 (s, 1H); ¹³C NMR (125 MHz, d-DMSO) δ 165.6, 152.3, 149.5, 146.7, 130.8, 129.6, 128.9, 124.6, 120.6, 99.0, 59.7, 54.2, 18.2, 14.4; MS *m*/z 328 (M⁺); Anal. Calcd for C₁₅H₁₅F₃N₂O₂S: C, 52.31; H, 4.40; N, 8.14. Found: C, 52.23; H, 4.23; N, 7.09%.

Ethyl-4-(4-bromothiophen-2-yl)-6-methyl-2-oxo-1,2,3,4-tetrahydropyrimidine-5-carboxylate (**1k**): White solid; IR (film) 3108, 1718 cm⁻¹; ¹H NMR (500 MHz, d-DMSO) δ 1.17 (t, J = 7.1 Hz, 3H), 2.23 (s, 3H), 4.08 (q, J = 7.0 Hz, 2H), 5.39 (s, 1H), 6.86 (s, 1H), 7.51 (s, 1H), 7.95 (d, J = 3.3 Hz, 1H), 9.39 (s, 1H); ¹³C NMR (125 MHz, d-DMSO) δ 165.3, 152.4, 151.0, 126.3, 122.9, 108.3, 99.3, 60.0, 18.1, 14.6; MS *m/z* 344 (M⁺); Anal. Calcd for C₁₂H₁₃BrN₂O₃S: C, 41.86; H, 3.82; N, 8.14. Found: C, 41.75; H, 3.67; N, 7.08%.

Ethyl-6-methyl-2-oxo-4-(thiophen-3-yl)-1,2,3,4-tetrahydropyrimidine-5-carboxylate (**1**): White solid; IR (film) 2980, 1649cm⁻¹; ¹H NMR (500 MHz, d-DMSO) δ 1.15 (t, J = 7.0 Hz, 3H), 2.22 (s, 3H), 4.05 (q, J = 7.0 Hz, 2H), 5.21 (s, 1H), 6.98 (d, J = 4.9 Hz, 1H), 7.15 (s, 1H), 7.45 (m, 1H), 7.74 (d, J = 2.9 Hz, 1H), 9.17 (s, 1H); ¹³C NMR (125 MHz, d-DMSO) δ 165.8, 153.0, 148.8, 146.2, 127.1, 126.6, 121.2, 99.9, 59.7, 49.9, 18.1, 14.6; MS *m*/z 266 (M⁺); Anal. Calcd for C₁₂H₁₄N₂O₃S: C, 54.12; H, 5.31; N, 10.52. Found: C, 54.06; H, 5.16; N, 10.43%.

Ethyl-6-methyl-4-(thiophen-3-yl)-2-thioxo-1,2,3,4-tetrahydropyrimidine-5-carboxylate (**1m**): Yellow solid; IR (film) 2980, 1665 cm⁻¹; ¹H NMR (500 MHz, d-DMSO) δ 1.15 (t, *J* = 7.0 Hz, 3H), 2.27 (s, 3H), 4.07 (m, 2H), 5.24 (s, 1H), 6.97 (d, *J* = 5.0 Hz, 1H), 7.16 (d, *J* = 2.3 Hz, 1H), 7.49 (dd, *J* = 2.9, 4.9 Hz, 1H); ¹³C NMR (125 MHz, d-DMSO) δ 175.2, 165.5, 145.6, 144.8, 127.5, 126.5, 122.0, 101.3, 60.1, 50.0, 17.6, 14.6; MS *m/z* 282 (M⁺); Anal. Calcd for C₁₂H₁₄N₂O₂S₂: C, 51.05; H, 5.01; N, 9.93. Found: C, 50.97; H, 4.96; N, 9.80%.

Ethyl-6-methyl-4-(3-methylthiophen-2-yl)-2-oxo-1,2,3,4-tetrahydro-pyrimidine-5-carboxylate (**1n**): Grey solid; IR (film) 2980, 1699 cm⁻¹; ¹H NMR (500 MHz, d-DMSO) δ 1.08 (t, *J* = 7.0 Hz, 3H), 2.21 (s, 3H), 2.24 (s, 3H), 3.96 (m, 2H), 5.45 (s, 1H), 6.75 (d, *J* = 5.0 Hz, 1H), 7.21 (d, *J* = 5.0 Hz, 1H), 7.76 (d, *J* = 2.4 Hz, 1H), 9.28 (s, 1H); ¹³C NMR (125 MHz, d-DMSO) δ 165.5, 152.1, 143.0, 133.3, 130.2, 126.1, 123.4, 100.5, 59.8, 48.0, 14.5, 18.1, 13.7; MS *m/z* 280 (M⁺); Anal. Calcd for C₁₃H₁₆N₂O₃S: C, 55.70; H, 5.77; N, 10.00. Found: C, 55.66; H, 5.58; N, 9.92%.

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